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Subject: perchlorate

Attachments: Clewell etal (2019) CIO4 BBDR Review RegToxPharm.pdf

This came my way today. Haven't read yet, but thought you may be interested.

\*

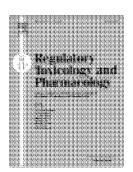
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# **Accepted Manuscript**

An evaluation of the USEPA Proposed Approaches for applying a biologically based dose-response model in a risk assessment for perchlorate in drinking water

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- 1 An Evaluation of the USEPA Proposed Approaches for Applying a Biologically Based
- 2 Dose-Response Model in a Risk Assessment for Perchlorate in Drinking Water
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### **ABSTRACT**

The United States Environmental Protection Agency's (USEPA) 2017 report, "Draft Report: Proposed
Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking
Water", proposes novel approaches for deriving a Maximum Contaminant Level Goal (MCLG) for
perchlorate using a biologically-based dose-response (BBDR) model. The USEPA (2017) BBDR model
extends previously peer-reviewed perchlorate models to describe the relationship between perchlorate
exposure and thyroid hormone levels during early pregnancy. Our evaluation focuses on two key
elements of the USEPA (2017) report: the plausibility of BBDR model revisions to describe control of
thyroid hormone production in early pregnancy and the basis for linking BBDR model results to
neurodevelopmental outcomes While the USEPA (2017) BBDR model represents a valuable research
tool, the lack of supporting data for many of the model assumptions and parameters calls into
question the fitness of the extended BBDR model to support quantitative analyses for regulatory
decisions on perchlorate in drinking water. Until more data can be developed to address uncertainties
in the current BBDR model, USEPA should continue to rely on the RfD recommended by the NAS
(USEPA 2005) when considering further regulatory action.

Keywords: perchlorate, risk assessment, MCLG, BBDR model

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From a regulatory perspective, the critical effect of concern from exposure to perchlorate is disruption of thyroid function and the potential for thyroid-hormone-related effects on neurodevelopment in gestation; these effects represent downstream events resulting from competitive inhibition of iodide uptake by the perchlorate ion (USEPA 2002). Based on an analysis of the mode of action for perchlorate, the United States Environmental Protection Agency (USEPA) (2002) determined that inhibition of thyroid iodide uptake could be used as an obligatory precursor for these critical effects in a harmonized cancer/noncancer risk assessment for perchlorate (Figure 1). This mode-of-action directed risk assessment approach was used in the derivation of the current Reference Dose (RfD) for perchlorate of 0.0007 mg/kg/day (USEPA 2005). Following the recommendations of the National Academy of Sciences National Research Council (NRC) (2005), the point of departure (POD) for this RfD was a reported No Observed Effect Level (NOEL): a non-statistically significant mean of 1.8% (standard error of the mean 8.3%) decline in radioactive iodine uptake (RAIU) in healthy adults following two weeks exposure to a daily perchlorate dose of 0.007 mg/kg/day (Greer et al. 2002). An intraspecies uncertainty factor of 10 was applied to protect the most sensitive population, the fetuses of pregnant women who might have hypothyroidism or iodide deficiency.

Subsequently, the USEPA Office of Drinking Water (2008) published an Interim Health Advisory Level for perchlorate of 15 µg/L, based on the USEPA (2005) RfD of 0.7 µg/kg/day, as recommended by the NRC (2005). Determination of this Interim Health Advisory Level considered Physiologically-Based Pharmacokinetic (PBPK) Modeling (Clewell et al. 2007) to estimate the potential effect of perchlorate on iodide uptake in several sensitive subgroups, including the pregnant woman and fetus, the lactating woman and neonate, and the young child. Despite widespread scientific acceptance of iodide inhibition as an obligatory precursor to downstream toxicity endpoints, there was remaining concern regarding the level of protection for the population perceived to have the greatest susceptibility – the fetuses of hypothyroid mothers.

Over the next several years, the focus of research on perchlorate shifted to the development of a biologically based dose-response (BBDR) model of the hypothalamic-pituitary-thyroid (HPT) axis that could be linked with the PBPK model of perchlorate and iodide to predict dose-dependent interactions

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55	of perchlorate with iodine hormone homeostasis as a function of iodide intake in an effort to more
56	quantitatively account for the effects of low dietary iodide intake and hypothyroidism in pregnant
57	women on fetal development (McLanahan et al. 2008, 2009; Fisher et al. 2012; Lumen and George
58	2017a, 2017b; Lumen et al. 2013, 2015).
59	The USEPA Science Advisory Board (SAB) (2013) report on perchlorate in drinking water supported
60	the utility of BBDR modeling to help characterize the potential for neurological effects from perchlorate
61	exposure:
62	"As perchlorate research continues, studies in animals may provide important insights into the
63	neurobehavioral consequences of perchlorate exposure. A physiologically-based
64	pharmacokinetic/pharmacodynamic framework is well suited to help place these findings in the
65	context of human perchlorate exposure."
66	The USEPA SAB (2013) identified a number of areas for improvement or modification of the existing
67	models. However, they also noted that "Models can always be improved, but the goal is to have a
68	model that is fit for the intended purpose.", apparently cautioning against perpetual model refinement
69	at the expense of implementation, echoing the concern of the renowned statistician, George E.P. Box,
70	who famously used to say: "All models are wrong but some are useful" (Box 1976).
71	Recently, the USEPA's Office of Ground Water and Drinking Water (USEPA 2017) responded to the
72	Science Advisory Board recommendations and proposed novel approaches to inform the derivation of
73	a Maximum Contaminant Level Goal (MCLG) for perchlorate, including the use of BBDR modeling in
74	their report entitled "Draft Report: Proposed Approaches to Inform the Derivation of a Maximum
75	Contaminant Level Goal for Perchlorate in Drinking Water". This MCLG approach (USEPA 2017)
76	includes revisions to a previously developed and peer reviewed BBDR model (McLanahan et al. 2008,
77	2009; Fisher et al. 2012, Lumen and George 2017a, 2017b; Lumen et al. 2013, 2015) that was
78	extended to predict the relationship between perchlorate exposure and thyroid hormone levels in
79	sensitive life stages. These revisions aim to address suggestions by the USEPA SAB (2013), including
80	the following:
81	$_{\infty}$ Derivation of a perchlorate MCLG that addresses sensitive life stages through PBPK/PD
82	modeling;

$\infty$	Expansion of the modeling approach to account for thyroid hormone perturbations and
	potential adverse neurodevelopmental outcomes from perchlorate exposure;

- □ Utilization of a mode of action framework for developing the MCLG that links the steps in the
   proposed mechanism leading from perchlorate exposure through iodide uptake inhibition to
   thyroid hormone changes and finally neurodevelopmental impacts; and
- Extension of "the [BBDR] model expeditiously to...provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (USEPA SAB 2013, p. 19).

Model revisions presented in the USEPA (2017) report include: incorporating a description of the physiology of early pregnancy, biological feedback control of hormone production via thyroid-stimulating hormone (TSH) and human chorionic gonadotropin (hCG), and a description of the response to lower levels of iodide nutrition. In addition, an attempt was made to calibrate the model's behavior for upper and lower percentiles of the population, in addition to the population median, for thyroid hormone production. The report also included an uncertainty analysis for key BBDR model parameters.

For the development of the MCLG, USEPA (2017) proposed a two-stage approach linking the revised BBDR model results ("Stage 1") with quantitative information on neurodevelopmental outcomes from epidemiological studies ("Stage 2"). Stage 1 describes the thyroidal hormone levels in women of childbearing age with low to adequate iodide intake. In this stage, the revised BBDR model is applied to predict the relationship between perchlorate exposure and changes in thyroid hormone levels in early pregnancy. Data for Stage 2 of the approach is provided from epidemiological studies evaluating maternal thyroid hormone levels in early pregnancy and the relationship between changes in these levels and the observation of neurodevelopmental outcomes. The USEPA (2017) report also described development of a novel population-based approach that uses the revised BBDR model to estimate changes in levels of selected thyroid hormones, specifically free tetraiodothyronine (fT4) and TSH, resulting from perchlorate exposure that may result in an increase in the prevalence of hypothyroxinemia in pregnant women. Hypothyroxinemia (low circulating concentrations of fT4) is

111	often associated with hypothyroidism (low concentrations of fT4 despite increased concentrations of
112	TSH).
113	The evaluation and (potential) application of the perchlorate BBDR model will serve as an important

precedent for future consideration of such models by the agency, as it is only the second such model to be seriously evaluated by USEPA and subjected to external peer-review. The first BBDR model to be considered, formaldehyde nasal carcinogenicity developed by Conolly and colleagues (2003, 2004), has been under consideration by the agency for more than a decade. Interest in the use of BBDR modeling in risk assessment peaked in the 1990s when the draft USEPA (2003) Cancer Guidelines identified these models as the preferred option for performing a cancer dose-response. However, since that time, work in this area has waned, possibly due to the perceived difficulty of gaining regulatory acceptance. By their nature, BBDR models are descriptions of complex biological systems that necessarily include significant uncertainty. The challenge going forward will be to develop approaches for characterizing that uncertainty in a risk assessment context and ensuring that these complex models are fit for their intended purpose. It is with this consideration in mind that we have performed a focused evaluation of the proposed USEPA (2017) approaches.

Our critical review focused on two key areas of importance for determining whether the current BBDR model is fit for the purpose of supporting regulatory decisions based on predicted effect of perchlorate exposure on human fetal development:

- 1. Evaluation of USEPA (2017) model revisions to the peer reviewed BBDR models, including extending the model to early pregnancy, incorporating biological feedback control of hormone production via thyroid stimulating hormone (TSH) and human chorionic gonadotropin (hCG) signaling, calibration of the model for thyroid hormone effects, and uncertainty analysis for key parameters. This evaluation included comparison of model output to results from key human studies identified in previous assessments (Greer et al. 2002, Braverman et al. 2006, Téllez et al. (2005a, 2005b), as well as in the USEPA (2017) document (Steinmaus et al. 2016);
- Evaluation of USEPA (2017) approaches for linking BBDR results to neurodevelopmental outcomes and identification of published literature to develop the quantitative relationship between thyroid hormone levels and neurodevelopmental outcomes; and

140	Aft	er describing the results of this evaluation, we present a comparison of the results from the
141	US	EPA (2017) approach with results from previous USEPA assessments, in order to put the
142	un	certainties in the BBDR approach in perspective against the potential impact of the new
143	apı	proach on the existing regulatory guidelines for perchlorate USEPA (2005, 2008).
144	METH	IODS
145	Evalua	ting Stage 1 of USEPA MCLG approach: Stage 1 of USEPA's MCLG approach relies upon the
146	applica	ation of the BBDR model to predict the effect of perchlorate on the thyroid hormone in pregnant
147	womer	n at different iodine nutrition levels, with the goal of predicting fT4 hormone reduction in
148	pregna	ant women with low dietary iodide. To evaluate the utility of the proposed model to support
149	such p	redictions, we independently ran the model and tested model predictions against data from
150	multipl	le studies. These exercises attempted to both duplicate BBDR model results for datasets that
151	were u	sed by USEPA (2017) to calibrate the model and to test the ability of the BBDR model to predict
152	the we	ell-described precursor event inhibition of iodide uptake, which was successfully described with
153 154		us versions of the perchlorate PBPK models (Merrill et al. 2003; Clewell et al. 2007). These tions included:
155	∞	Steinmaus et al. 2016 – cross-sectional epidemiological study evaluation of serum and urine in
156		pregnant women in California: used in USEPA (2017) to evaluate BBDR model predictions of
157		perchlorate effects on fT4 and TSH
158	∞	Greer et al. 2002 – 14-day controlled perchlorate dose study in male and female adults in the
159		US: used in USEPA (2017) to estimate urinary clearance parameters in BBDR model
160	∞	Braverman et al. 2006 – 6-month controlled perchlorate dose study in male and female adults:
161		not used in USEPA (2017)
162	∞	Téllez Téllez et al. 2005a, 2005b – longitudinal epidemiological study in pregnant and lactating
163 164		women in Chile: used in USEPA (2017) to estimate urinary clearance parameters in BBDR model
165	In our	efforts to produce these simulations, it was noted that instructions provided in the USEPA
166	docum	entation for running the model for different scenarios, and documentation of the rationale for

the model parameter values associated with them, are often inadequate and lack transparency; this

deficiency is exacerbated by the number of permutations of parameter settings used in the scripts that

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generate the results in the document. The complexity of the BBDR model makes it difficult to perform

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literature and evidence integration.

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170	this evaluation, even though it has been conducted by experienced modelers.
171	Evaluating Stage 2 of USEPA MCLG approach: Stage 2 of USEPA's approach involved evaluating the
172	published epidemiological literature to identify publications that would define quantitative relationships
173	between thyroid hormone levels and neurodevelopmental effects. The USEPA approach was focused
174	on the identification of studies that provided information on levels of fT4 in pregnant mothers during
175	early gestation and the potential for changes in neurodevelopmental outcomes in their offspring.
L76	Through targeted literature searching and recommendations from the Science Advisory Board (SAB), a
177	total of 55 studies were identified by USEPA to provide information on altered maternal thyroid
178	hormone levels and offspring development. These studies were divided into three groups to facilitate
179	evaluation:
180	$_{\infty}$ Group 1 – studies that may be able to quantitatively describe a relationship between
181	incremental alterations in maternal thyroid hormone levels and alternations in offspring
182	development;
183	$\infty$ Group 2 – studies that do not have data from which to derive a quantitative relationship
184	between maternal hormones and offspring neurodevelopment, but instead present only a
185	categorical analysis with thyroid hormones below and above a defined cut point and adverse
186	neurodevelopmental outcomes; and
187	$_{\infty}$ Group 3 – studies that present an analysis that is not directly compatible with BBDR output.

Of the 55 studies, 15 were identified as Group 1, 14 were identified as Group 2, 26 were identified as

Group 3. The 15 Group 1 studies were then evaluated further and only 5 were deemed useful by the

USEPA for further quantitative analysis to attempt to connect alterations in thyroid hormone levels to

alterations in neurodevelopment. in our evaluation, we performed a critical review of the USEPA

recommendations from the National Research Council (NRC 2014) on systematic review of the

Stage 2 approach and the study summaries provided in USEPA (2017), considering the most recent

### **RESULTS**

### **Evaluation of the Perchlorate BBDR Model for Early Pregnancy**

The draft MCLG approach (USEPA 2017) is based on a hypothesized mode of action (Figure 1) for neurodevelopmental outcomes resulting from development of hypothyroxinemia from perchlorate-induced inhibition of iodide uptake in the thyroid. As noted in USEPA (2017):

"Thyroid hormones are essential for the development and differentiation of the developing brain. The brain and spinal cord begin development in the first half of the first trimester. fT4 passes through the blood-brain barrier via multiple, specific transporter proteins. Next, T4 is converted to T3 by the developing glial cells and then transported to neurons. T3 then interacts with nuclear receptors to tightly regulate gene expression so that neurogenesis, synaptogenesis, neuronal migration, cell differentiation, and myelination are developmentally appropriate. Deficiencies in thyroid hormones through iodine deficiency, congenital hypothyroidism, or maternal hypothyroidism/hypothyroxinemia can result in neurological impairments and intellectual deficits (Morreale de Escobar, Obregón, & Escobar del Ray 2000)."

As recommended by the USEPA SAB (2013), the USEPA extended a published BBDR model for perchlorate induced hypothyroxinemia in late gestation (Lumen et al. 2013; Lumen and George 2017a, 2017b) to address the sensitive population of concern for exposure to perchlorate: the fetuses of hypothyroxinemic women during early pregnancy (Figure 2). These concerns were motivated by new studies (Steinmaus et al. 2016), suggesting an association between perchlorate exposure and decreased levels of free thyroxine (fT4) in pregnant women. Because the fetus is entirely dependent on maternal thyroid hormones for neurodevelopment in early gestation (Clewell et al. 2007; Howdeshell 2002), the endpoint of interest was defined as reduction in maternal fT4 in early pregnancy and the perchlorate BBDR models were extended to describe hormone homeostasis during gestation. Expansion of the original models of perchlorate and iodide (Clewell et al. 2007) to predict the impact of perchlorate exposure on fT4 during early pregnancy, however, is complicated by the significant variability in the levels of fT4 in the general population and the challenges in measuring fT4, as well as the dynamics of changing hormones through the course of gestation and the uncertainty in identifying the level of alteration that may lead to hypothyroidism and fetal effects.

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225	According to the "American Thyroid Association Task force on Thyroid Disease During Pregnancy and
226	Postpartum", isolated hypothyroxinemia is defined as a normal maternal TSH concentration in
227	conjunction with fT4 concentrations in the lower 5th or 10th percentile of the reference range
228	(Stagnaro-Green et al. 2011). USEPA (2017) has also focused on selected percentiles of the reference
229	range; however, reference ranges can vary from population to population according to the 2017
230	Guidelines of the American Thyroid Association for the Diagnosis and Management of Thyroid Disease
231	During Pregnancy and Postpartum (Alexander et al. 2017). Even within US populations and across
232	ethnic groups, the 2.5th percentile can vary by up to 2 pmol/L or approximately 20% (9.3-11.4
233	pmol/L as reported by Alexander et al. 2017).
234	The variation in fT4 reported in the published literature during early pregnancy is provided in USEPA
235	(2017), Appendix A, Figure A-33 and reproduced in Figure 3. The levels of fT4 during early
236	pregnancy, based on the studies identified by USEPA (2017), appear to range from approximately 13-
237	17 pmol/L. This range is consistent with the range of baseline fT4 means reported in the Greer et al.
238	(2002) study of approximately 1.1 – 1.3 ng/dL (14 - 17 pmol/L). However, the 50th percentile BBDR
239	model predictions at zero dose perchlorate and 170 $\mu g/day$ iodine intake are approximately 10 pmol/L
240	at gestation weeks 12, 13, and 16, considerably below these reported values.
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242	Measuring fT4 in the presence of high concentrations of bound T4 is challenging, especially in
243	conditions where binding proteins are altered such as during pregnancy (Alexander et al. 2017).
244	Measurement techniques are prone to inaccuracy during pregnancy due to disruption of the original
245	equilibrium. The 95% fT4 reference intervals decrease gradually with advancing gestational age: from
246	$1.08-\ 1.82\ ng/dL$ (approximately $13.9-\ 23.5\ pmol/L$ ) in week 14 to $0.86-1.53\ ng/dL$ (approximately
247	11.1 – 19.8 pmol/L) in week 20 (Alexander et al. 2017).
248	Extending the thyroid BBDR model to address early gestation is particularly challenging due to
249	the complex interaction between thyroid homeostasis and gestational development.
250	Considering the addition of TSH feedback dynamics, and an adjustment factor to match
251	specific population percentiles, there is reason for concern regarding the uncertainty of the

252	revised model predictions under low iodide intake conditions. Some of these concerns are
253	highlighted below:
254	<u>Description of hCG dynamics</u> : Human chorionic gonadotropin (hCG) levels rise in early pregnancy
255	and this in turn increases both sodium-iodide symporter (NIS) uptake activity and T4 production.
256	hCG is structurally similar to TSH and, like TSH, increases thyroidal iodide uptake and thyroid
257	hormone synthesis by binding to the thyroid-stimulating hormone receptor (TSHR) (Hoermann et
258	al. 1994). In the model, hCG levels are calculated as a function of gestational age, using an
259	equation for the parameter HCGREG (Figure 4, purple curve), and these changing levels are used
260	to increase the rate of T3 and T4 production as a function of the hCG concentration:
261	HCGreg = 1 + 0.00354*hCG
262	The variation of hCG over the duration of gestation is based on direct measurements of hCG in
263	pregnant women (Korevaar et al. 2015). However, the concurrent increase in thyroidal iodine
264	uptake is described in the model based on an empirical relationship between gestational age in
265	weeks (GW) and radioactive iodide uptake, using an equation for the parameter VCHNG (Figure 4,
266	green curve):
267	$VCHNG(GW) = 1 + 0.076 \cdot GW - 0.0025 \cdot GW^{2}$
268	Thus the model does not correctly attribute the gestational control of NIS uptake to hCG, when in
269	fact both uptake and hormone synthesis respond to the same changes in hCG (Pesce and Kopp
270	2014). By using different equations for the time-dependence of hCG-stimulated uptake and
271	hormone production (Figure 4) the model decouples processes that are fundamentally linked by
272	their biology. Figure 4 depicts the time-course of the parameters controlling changes in iodide
273	uptake (VCHNG) and hCG hormone levels (HCGreg) over the course of gestation. While the
274	biology indicates a proportional relationship between the two parameters, the equations used in
275	the model are not parallel. Elucidating the impact of this decoupling is challenging, and is beyond
276	the scope of this review, since it would have to be investigated at a large number of time-points
277	throughout pregnancy and under different conditions of iodine intake, but the disparity between
278	the model description and the underlying biology justifies caution regarding its predictions of T4

and TSH at different gestational ages, as these parameters govern hormone production and

release. We address the impact of the discrepancy between the time-courses for HCGREG and VCHNG in a later section.

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<u>Damping of TSH response</u>: The USEPA (2017) BBDR model includes a parameter, pTSH (power to which the ratio of current TSH to the TSH set-point is raised), that reduces the response of the thyroid to increases in TSH:

 $TSHreg = (TSH/TSH_0)^{pTSH}$ 

Using this equation, a pTSH exponent of 0 would represent no control of thyroid function by TSH and an exponent of 1 would represent a linear response of thyroid function to changes in TSH. In their calculations of the effect of perchlorate on the prevalence of hypothyroxinemic pregnant women, the USEPA (2017) use a pTSH exponent of 0.398, which results in a response to TSH that is substantially less than linear, an assumption that is inconsistent with the fundamental biological relationship between TSH and thyroid hormones (production and release of T3 and T4), effectively decoupling a relationship that has been well established in the medical, pharmacological and toxicological communities. USEPA (2017) describes the rationale for this parameter: "The NHANES data do not show a clear correlation between TSH and fT4, so within that data set they vary independently. One could assume, therefore, that individuals with an average fT4 and high TSH have that combination because their thyroid has a weak response to TSH, and vice-versa." To address this concern, USEPA (2017) estimated a lower and upper bound for pTSH as (median TSH)/(97.5th percentile TSH) = 0.398 and (median TSH)/(2.5th percentile TSH) = 3.09, respectively, with a median value of pTSH = 1. Thus, this parameter is used to attempt to represent disease states where the individual's thyroid is either exquisitely sensitive or insensitive to TSH stimulation. At lower values, this parameter reduces the impact of TSH on the Vmax for thyroid iodide uptake as well as the rate constants for T4 and T3 production in the thyroid. However, the USEPA (2017) also states that: "The coefficient, pTSH, is included...to allow for tuning of the strength of the TSH feedback, but in practice model simulations versus data appear quite adequate with pTSH=1." Concerns about this parameter are two-fold. First, the complexity of the model and various runtime scripts makes it nearly impossible to determine the use of this

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308	parameter during some of the model assessment and risk assessment simulations presented in
309	USEPA (2017). Second, using point estimate population level data to define the quantitative
310	temporal relationship between two fundamentally linked processes at the individual level is
311	scientifically inappropriate. To understand the biological feedback within a single individual (i.e. to
312	determine the relationship of TSH to T3/T4 and Vmax for a hypothyroid or hyperthyroid
313	individual), matched samples would be needed for TSH, T3 and T4. This information – to our
314	knowledge – is not available from NHANES. Thus, the epidemiological point estimate data are
315	being used well beyond its domain of applicability to predict the quantitative outcome of disease
316	states.
317	<u>Calibration of hormone production rates:</u> The model uses a baseline first-order constant calibrated
318	to NHANES 2007-2012 median, 10th, or 90th percentile non-pregnant data (fT4, fT3, T4 and T3
319	concentrations). The model parameter for the rate of production of T4 (KProdT4F) for the median
320	NHANES calibration used in USEPA (2017) is $6.25*10-7$ /hr/kg $^{0.75}$ (their Table A-2), which is 4-fold
321	lower than the value of 2.45 x 10-6 estimated for the published model (Lumen et al. 2013), which
322	was based on the data of Nicoloff et al. (1972). However, the use of a T4 production rate that is
323	lower than the published value is not adequately justified, given the importance of this parameter,
324	which has a direct impact on predictions of fT4 changes, the intended application of the model.
325	This baseline value is then scaled in pregnancy through GW 16 (peak occurring ~ GW 9) based
326	upon placental hCG increase over this time, according to the linear relationship from Glinoer
327	(1997): $hCGreg = 1 + 0.00354 \times hCG$ .
328	Affinity of NIS Iodine uptake: The model uses a Km for perchlorate binding to the NIS (KmNIS_P)
329	that is 3-fold lower than the value estimated by Lumen et al. (2013) (i.e. a 3-fold higher affinity).
330	Specifically, the new Km represents the 2.5th percentile lower confidence limit of the population
331	median based upon the USEPA (2017) reanalysis of Greer et al. (2002). The median value (50th
332	percentile = $0.73 \mu M$ ) is similar to that obtained from a re-analysis of in vitro binding data, 0.59
333	$\mu M$ (Schlosser 2016); the use of a value of KmNIS_P = 0.489 $\mu M$ makes perchlorate 3 times more
334	effective at competitive inhibition of NIS compared to the model of Lumen et al. (2013). This
335	revision to the Km in USEPA (2017) necessitated revisions to the Vmax (VmaxNISF_thy_P) and
336	urinary excretion parameters (CLFUP) (Table 1 of USEPA 2017), further affecting the model's

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337	sensitivity to changes in perchlorate dose, particularly under conditions of low iodide. Thus, The
338	USEPA (2017) BBDR model predicts much greater effects of perchlorate on iodide uptake than any
339	previous version of the model, without justification for re-estimating these parameters rather than
340	using the published values.
341	Assumptions regarding thyroidal iodide storage: Plots of NHANES 2007-2012 data for non-
342	pregnant women demonstrated little relationship between iodine intake and fT4, even at iodide
343	intake levels below 75 $\mu g/day$ (Figure A-54 of USEPA 2017; reproduced in Figure 5a). USEPA
344	(2017) used data on the relationship between thyroidal iodide stores (mg) and iodine intake from
345	Delange (2000), which assumes depletion of fT4 at iodide intake levels below 100 $\mu g/day$ . As is
346	clear from Figure 5b, this assumed model behavior at concentrations below 100 $\mu g/day$ , which
347	drives model predictions at low intakes, is inconsistent with the NHANES data and could result in
348	overprediction of fT4 responses at moderately low intakes of iodide, including the ranges simulated
349	in the USEPA report. This possibility was investigated in this evaluation and the results are
350	discussed in the next section.
351	Evaluation of BBDR Model Behavior
352	Comparison to the Steinmaus et al. 2016 Results
353	In Appendix B of USEPA (2017), a comparison of the predicted changes in both fT4 and TSH from the
354	BBDR model were compared to the results reported by Steinmaus et al. (2016). The Steinmaus et al.
355	(2016) study was conducted to evaluate the potential for perchlorate exposure to impact thyroid
356	hormone levels in pregnant women (any trimester) in San Diego. They reported an effect of
357	perchlorate on fT4 levels to be similar among women with both low iodine (<100 $\mu g/day$ ) and normal
358	(100-300 μg/day), with a greater effect of perchlorate observed among pregnant women in the high
359	iodine intake group ( $>300~\mu\text{g}/\text{day}$ ). They further noted that this result is in contrast to some previous
360	results from NHANES (Blount et al. 2006) and may be due to the overall iodine sufficiency in the
361	studied population or the fairly long time between urine iodine and serum thyroid hormone sample
362	collection (about 9 weeks).
363	The comparison of the predicted fT4 changes from the BBDR model and the Steinmaus et al. (2016)
364	results associated with changes in perchlorate dose are reported in Figure B-1 of Appendix B of USEPA
365	(2017) and reproduced in Figure 6. This comparison, which we were able to reproduce using the

366	USEPA (2017) BBDR model, clearly highlights the differences between the model predictions and the
367	published human data. The USEPA (2017) BBDR model simulations with normal iodine intake (170
368	$\mu g/day)$ demonstrate no change in fT4, which is consistent with other studies in which no impact on
369	fT4 has been observed at doses up to 7 $\mu$ g/kg/day perchlorate (Greer et al. 2002; Braverman et al.
370	2006). The USEPA (2017) BBDR model greatly under-predicts the changes in fT4, even in the
371	scenario with low dietary iodine intake (75 $\mu g/day$ ), in comparison to the changes reported by
372	Steinmaus et al. (2016). This discrepancy raises concerns about the ability of the USEPA (2017)
373	BBDR model to predict changes in fT4 associated with chronic perchlorate exposure during pregnancy.
374	
375	Greer et al. 2002 – 14 day human controlled perchlorate dosing study
376	The Greer et al. (2002) study was conducted to establish the dose-response in humans for perchlorate
377	inhibition of thyroidal iodide uptake and any short-term effects on thyroid hormones following
378	exposure for male and female volunteers to perchlorate in drinking water at doses of 7, 20, 100 or
379	500 μg/kg/day for 14 days. The results of this study have previously been relied upon by the USEPA
380	(2005) to derive a reference dose (RfD) and to determine health reference levels (HRLs). The results
381	of this study indicate a decrease in iodide uptake following exposure to a dose of 20 µg/kg/day, but no
382	effect on hormone levels, including fT4 and TSH, at the highest dose tested. A No Observed Effect
383	Level (NOEL) of 7 μg/kg/day was determined based on these results, and an RfD of 0.7 μg/kg/day
384	was adopted, based on NAS recommendations, with the application of an uncertainty factor of 10 for
385	intraspecies variability or sensitive subpopulations.
386	Consistent with the results of the study, our simulations of the adult exposures reported in Greer et al.
387	(2002) with the BBDR model (Table 1) indicated no significant change in fT4 at doses up to 500
388	μg/kg/day. However, predicted concentrations of fT4 are lower than those measured by Greer et al.
389	(2002). The model simulation reported in Table 1 was run with an iodine intake of 90 μg/day, as this
390	was the value USEPA (2017) used in the Greer_test.m script provided with the BBDR model code.
391	However, 90 $\mu$ g/day is not consistent with the 170 $\mu$ g/day value USEPA (2017) reports as
392	representing a sufficient intake and USEPA's (2017) documentation does not indicate why a lower
393	value was used for the individuals in the Greer study. Simulation of iodide uptake inhibition (RAIU)
394	appears to over-predict the reduction in uptake compared to measured values, though the qualitative

increasing trend of inhibition with dose behaves appropriately. This discrepancy may result from the low iodine intake chosen by USEPA (2017), or a number of other decisions made in the model revisions, including the reduced Km parameter value. It is unclear why the parameters governing iodide inhibition were altered from previous models that successfully predicted inhibition of iodide in human subjects (Clewell et al. 2007; Merrill et al. 2003; Lumen et al. 2013). Given that iodide inhibition is the obligatory precursor to all downstream effects in the USEPA's proposed mode of action for perchlorate, it would be expected that any changes to the model that lead to reduced accuracy in the prediction of iodide inhibition would be accompanied by substantial support. However, no such support is provided in USEPA (2017) for the changes in the key parameters and the resulting effect on iodide inhibition predictions.

Table 1. Simulation of the Greer et al. (2002) Perchlorate Study					
Dose	RAIU	J (%)	fT4 (pM)		
(µg/kg/d)	Simulated	Measured	Simulated	Measured	
0	100	100	10.33	-	
7	89	98.2	10.33	-	
20	74	83.6	10.32	16.09	
100	37	55.3	10.31	15.26	
500	11	32.9	10.30	15.44	

### Braverman et al. 2006 – 6 month human controlled perchlorate dosing study

The Braverman et al. (2006) study was conducted to determine whether prolonged exposure (6 months) of adults to low levels of perchlorate (0.5, 1.0 or 3.0 mg/day) would perturb thyroid function. The study included a small number of individuals (n=13); however, iodine levels were comparable with those of the general population. The authors noted the limitations of the small sample size, but concluded that the results suggested that healthy, euthyroid individuals, with normal levels of iodine intake, can tolerate chronic exposure to perchlorate at doses of up to 3 mg/day (approximately 40 µg/kg/day) without any effects on thyroid function, including inhibition of iodine uptake.

The Braverman et al. (2006) study was simulated as part of the current evaluation using the BBDR model and predicted T3 and TSH levels were compared to the reported measurements (Table 2). fT4

was not compared because it was not clear how to convert the T4 index reported in the study to a concentration and vice versa. As with the Greer et al. (2002) simulation, 90  $\mu$ g/day was used for iodine intake. Baseline T3 and TSH are similar to the measured values. But, as was seen with fT4, the model fails to predict the observed changes in hormone levels in the adult subjects.

Table 2. Simulation of Braverman et al. (2006) perchlorate study.				
Dose	ТЗ (	(nM)	TSH (ı	mIU/L)
(µg/kg/d)	Simulated	Measured	Simulated	Measured
0	2.63	2.49	1.51	1.20
7	2.63	2.51	1.52	1.60
43	2.62	1.77	1.53	2.60

Téllez Téllez et al. 2005a, 2005b - Chilean epidemiological study in pregnant women

Téllez Téllez et al. (2005a, 2005b) reports the results of a longitudinal epidemiological study among pregnant women from three cities in Chile exposed to concentrations of perchlorate as high as 114 µg/L in the public drinking water. The focus of the study was to evaluate maternal thyroid function during pregnancy, neonatal thyroid function and developmental status at birth, and breast milk iodine and perchlorate levels during lactation. The National Academy of Sciences (2005) has reviewed this study in the context of health implications for perchlorate ingestion and concluded this study should be considered in the evaluation of the US experience with perchlorate in drinking water. The total iodine nutrition among this cohort was also noted to be similar to that of US pregnant women (Téllez Téllez et al. 2005a); therefore, this study should be a key consideration in evaluating the relationship between perchlorate exposure, changes in fT4 in pregnant women and developmental status; however, it was not considered in Stage 2 of the USEPA (2017) assessment because it pre-dated the cutoff used by USEPA in their review (2010).

Results from this study indicated no effect on thyroid levels in early pregnancy, late pregnancy, or neonates at birth related to perchlorate in drinking water at concentrations up to 114  $\mu$ g/L. Given these findings, this study provides a reasonable dataset for validating the impact of high perchlorate exposure concentrations in drinking water on potential changes in fT4 or TSH.

We also ran the (USEPA 2017) BBDR model to simulate the Téllez Téllez et al. (2005a, 2005b) drinking water study (Table 3). The BBDR model predictions of fT4 for GW 13-16 are consistent with the negative results of the study, though the predicted concentrations are lower than those observed. This is not a strong validation of the model given the weak trend of changes in hormone levels seen in comparisons to other studies.

Table 3. Simulation of the Téllez Téllez et al. 2005a, 2005b study of pregnant women exposed to perchlorate via drinking water.

Dose	fT4	(pM)
(µg/kg/d)	Simulated	Measured
0.01	9.74	12.5
0.08	9.73	12.2
2	9.69	12.7

Summary: Evaluation of Model Behavior

Our simulations of the Greer et al. (2002) and Braverman et al. (2006) studies with the BBDR model indicate that thyroid hormone levels are relatively insensitive to inhibition of thyroid iodine uptake by perchlorate exposures as high as 7  $\mu$ g/kg/day. Moreover, our simulations of the Téllez Téllez et al. (2005a, 2005b) study with the BBDR model do not predict an effect on fT4 from exposures to perchlorate at up to 2  $\mu$ g/kg/d, consistent with the fact that the exposures were demonstrated to be without effect to pregnant women in the study. However, the USEPA (2017) BBDR modeling analysis (Table 4, taken from USEPA 2017) predicted population-level changes in fT4 deficiency during the first trimester at perchlorate exposures nearly an order of magnitude lower (0.3  $\mu$ g/kg/d). This discrepancy suggests that the metric used in the USEPA (2017) approach to assess population-level effects of perchlorate, i.e., a 1% or 5% increase the proportion of thyroxinemic mothers in early pregnancy assuming that all individuals have a low (75  $\mu$ g/day) iodine intake and an inadequate TSH response (pTSH = 0.398 vs. 1), may be overly conservative.

Table 4. Summary of Results for the Amount of Perchlorate Needed to Increase the Proportion of Hypothyroxinemic, Low Iodine Individuals by a Defined Percentage (with hypothyroxinemia defined as  $fT4 < 10^{th}$  Percentile) (USEPA 2017)

Gestational Week	fT4 (pmol/L) at the Hypothyroxinemic Cut Point (i.e. 10 <sup>th</sup> Percentile of 170  µg/day Iodine Intake Group)  (Column 1)	Corresponding Percentile in 75  µg/day Iodine Intake Group  (Column 2) <sup>a</sup>	Perchlorate Dose  (µg/kg/day) Associated  with a 1 Percent  Increase in Proportion  Hypothyroxinemic  (Column 3) <sup>a</sup>	Perchlorate Dose  (µg/kg/day)  Associated with a 5  Percent Increase in  Proportion  Hypothyroxinemic  (Column 4)a
12	8.80	48.4	0.4	2.2
13	8.78	47.9	0.4	2.2
16	8.63	52.6	0.3	2.1

<sup>&</sup>lt;sup>a</sup> Results based on central effect estimates, pTSH in BBDR model set to 0.398

### Evaluation of the effect of model assumptions on predicted PODs

In order to assess the potential quantitative impact of some of the uncertainties in the BBDR model, we compared model predictions of percent change in fT4 and TSH for a range of perchlorate concentrations using two alternative parameterizations: (1) the parameterization used by the USEPA (2017) to generate their Table 3, and (2) replacing the equation for HCGREG with the equation for VCHNG (in order to provide an appropriately coupled response to hCG stimulation of thyroidal iodine uptake and thyroid hormone production), and also setting pTSH = 1 (the nominal value, as opposed to the lower-bound value of 0.398 used by the USEPA). The simulations (Table 5) were performed with the model calibrated to either the median population thyroid hormone levels (using the script medset.r) or a low (thyroxinemic) population defined as fT4 < 10<sup>th</sup> percentile (using the script lowset.r). When predicting the effect of perchlorate exposure on fT4 for the median population there is not a significant difference between the USEPA results and the alternative parameterization; however, the USEPA model parameterization results in more than a factor of 2 greater sensitivity of TSH levels to perchlorate compared to the alternative parameterization. This difference is primarily

due to the change in pTSH. On the other hand, when predicting the effect of perchlorate exposure on hypothyroxemic individuals, both fT4 and TSH responses to perchlorate exposure are significantly lower using the alternative parameterization. Thus, the parameters that were altered in the recent revision of the model (VCHNG, HCGreg, pTSH, KmNiS\_p) increase the predicted effect on thyroid hormone levels compared to the expected response with the well-validated precursor event of iodide inhibition. The sensitivity of the prediction to changes in these parameters, and the disconnect between the prediction of iodide inhibition and thyroid hormone levels, calls for better justification – and evaluation – of the given parameter values.

	fT4 (pmol/L)				TSH (mIU/L) (% Change from 0 Dose)			
Perchiorate Dose	(% Change from 0 Dose)							
(µg/kg/day)		USEPA <sup>0</sup>	VCHNG + VCHNG	VCHNG +		USEPA <sup>0</sup>	VCHNG +	VCHNG +
		USEPA	pTSH <sup>1</sup>	pTSH <sup>1</sup>		USEPA	pTSH <sup>1</sup>	pTSH <sup>1</sup>
	Population	Median	Median	Low	Population	Median	Median	Low
0	Absolute	8.6	9.9	7.5	Absolute	2.2	1.5	3.0
1		-0.74	-0.8	-0.31		3.3	1.4	1.9
2		-1.5	-1.6	-0.61		6.6	2.7	3.8
3	Percent	-2.1	-2.3	-0.9	Percent	10	4.1	5.7
4	Change	-2.8	-2.9	-1.2	Change	14	5.5	5.7
5		-3.4	-3.5	-1.5		17	6.9	7.7
10		-6.2	-6.2	-2.8		36	14	19

 $<sup>^{0}</sup>$  Results using pTSH = 0.398

### Review of Literature Linking BBDR Results to Neurodevelopment Outcomes

Chapter 5 of USEPA (2017) focuses on the SAB's recommendation to "Identify literature and conduct analyses to support the model outputs for the downstream steps" from the BBDR's predicted changes in thyroid hormones following exposure to perchlorate. Specifically, Chapter 5 was developed to present the process USEPA (2017) used to identify literature to support the draft approach for derivation of the MCLG for perchlorate. USEPA (2017) states, "Based on the recommendations of

Results using HCGREG replaced with VCHNG, and pTSH=1

previous peer review panels, USEPA assumed that changes in thyroid hormone levels would be
expected to lead to neurodevelopmental outcomes", and because of this assumption, a complete
systematic review of the body of literature on this topic was not performed. Instead, a "focused
review of the published literature" was conducted.
The approach is inconsistent with recent recommendations from the National Research Council (NRC
2014) regarding systematic review and evidence integration. These recommendations are currently
being incorporated into the USEPA's Integrated Risk Information System (IRIS) process and USEPA
has recently released scoping and problem formulation materials for several new Integrated Risk
Information System (IRIS) assessments, including ethylbenzene (USEPA 2014a), and naphthalene
(USEPA 2014b). The approach applied in these assessments is intended to follow recommendations
provided by the National Research Council (NRC 2013). While development of MCLGs are not part of
the IRIS process, the application of systematic review principles in the identification of studies to
define the relationship between fT4 and neurodevelopmental effects, is needed. The application of
these principles would not only assist in defining the highest quality studies to address a specific
research question, they also provide a way to integrate all of the available evidence for the specific
research questions raised by the SAB. Systematic reviews include the formulation of a specific
question to be addressed and developing a protocol that specifies the methods that will be used to
address the question. While a broad research question can lead to a large systematic review, if the
research question is limited, such as in the case of perchlorate, then the systematic review becomes
more focused.
For the USEPA (2017) draft MCLG approach, a systematic review question could have been easily
developed based on the SAB recommendation (i.e. "Identify literature and conduct analyses to support
the model outputs for the downstream steps") and the protocol would simply be focused on the
methods for conducting the systematic review to address this very focused systematic review question
in a transparent manner. Transparency being defined by USEPA as "sufficient information will be
$available\ to\ understand\ the\ scientific\ rational\ behind\ decisions,\ as\ well\ as,\ reproduce\ methods\ used\ to$
identify and evaluate data". However, in the case of the literature identified for consideration in the
draft MCLG approach for perchlorate, a well-defined protocol for all steps of the process has not been
developed and therefore is inconsistent with the recommendations of the NRC (2013):

516 "A priori decisions and a predefined protocol are critical during the systematic review 517 process (Berlin and Colditz 1999; Dickersin 2002); the protocol should describe the 518 following steps: the research question, the search strategy and data sources, the 519 study inclusion and exclusion criteria, the data to be abstracted and derived from the 520 original studies (such as sample size, exposure and outcome assessment methods, 521 and confounders evaluated), the criteria and methods for pooling effect estimates and 522 measures of variability among studies. Systematic reviews and meta-analyses need 523 to be replicable; other investigators following the same steps should be able to identify 524 the same articles, abstract the same data, and reach similar conclusions." 525 At each step of the process for identifying studies for use in the development of the MCLG 526 approach for perchlorate, a detailed set of criteria is needed. For example, if decisions are 527 made to include or exclude any studies, there should be very detailed criteria indicating why 528 studies were included or excluded and it should be specified prior to the initiation of the 529 literature searching process. The criteria for each step should be described in such a way that 530 an independent reviewer could use it to replicate the results of the literature search and review; however, there are several areas in the USEPA (2017) draft MCLG approach for 531 532 perchlorate where this level of detail is lacking, making it difficult for an independent reviewer 533 to replicate the results. 534 Systematic Review Research Questions An overall hypothesis or systematic review research question should be developed that is based on the 535 536 SAB recommendation to clarify the focus of the review and the linkage between altered maternal fT4 537 (as predicted by the BBDR model) and the potential for adverse neurodevelopmental effects in 538 offspring. Some additional explanation as to how USEPA arrived at the specific neurodevelopmental 539 outcomes of concern should be provided. 540 Searching the Published Literature 541 While the literature search key words are presented in the USEPA (2017) report, there is a lack of 542 explanation as to the reasoning behind the focus on the outcome of concern. The research question

should be used to develop the literature search. The major points used or considered in developing

the literature search strategy should be presented. In addition, there should be a detailed explanation
of the criteria used to screen the literature search results. Furthermore, USEPA (2017) does not
report the details of the literature search results. For each search string reported in Table 9 of the
USEPA (2017) report, a total number of citations identified should be reported. In addition, the
criteria used to screen the original search results should be clearly reported in the document.
Essentially, each step of the literature search and review should be reported in such a way that any
independent party could easily reproduce the results reported in Chapter 5 of USEPA (2017). The lack
of this type of information does not allow the reader to determine if any key studies may have been
removed from consideration.
Literature Screening Approach and Selection of Key Studies
USEPA (2017) states that a 3 step approach was used to identify studies for consideration in the
development of the approach for derivation of the MCLG for perchlorate. The approaches utilized by
USEPA (2017) to identify the epidemiological studies for this evaluation were strictly focused on the
appropriateness of the quantitative data for consideration in combination with the output of the BBDR
model. Group 2 (studies with categorical analyses only) and Group 3 (studies with analyses not
directly compatible with BBDR output) studies were apparently eliminated from consideration in the
assessment. While not directly compatible with BBDR modeling output, it is possible that these
studies may provide information important in understanding the potential relationship between
changes in thyroid hormones and the potential for neurodevelopmental effects, as well as potential
key confounders.
While 15 studies were identified in Group 1, only 5 of these were determined by USEPA to include
analyses that could be used to connect the results of the BBDR model to incremental changes in
adverse neurodevelopmental effects. A clearly defined set of inclusion and exclusion criteria should be
provided to clearly convey to the reader why the other 40 studies in Groups 1, 2, and 3 were not
considered. In addition, studies that provide no evidence of an inverse relationship between
perchlorate exposure and serum thyroid function (e.g. Ghassabian et al. 2014; Modesto et al. 2015;
Moleti et al. 2016; Noten et al. 2015) should also be considered to not only understand why these
results are in contrast to the potential research question, but also that the overall weight of evidence
can be determined. It is possible that the majority of studies provide evidence that critical factors that

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573	are not reported in some of the available studies may explain the reported changes in serum thyroid
574	function.
575	Assessment of Study Quality and Risk of Bias
576	According to recent recommendations from the National Research Council (NRC 2014), the National
577	Toxicology Program's (NTP) Office of Health Assessment and Translation (OHAT) method for the
578	assessment of study quality and risk of bias of the literature (NTP 2015) is one method that should be
579	considered for qualitative and quantitative assessments. "An assessment of study quality evaluates
580	the extent to which the researchers conducted their research to the highest possible standards and
581	how a study is reported. Risk of bias is related to the internal validity of a study and reflects study-
582	design characteristics that can introduce a systematic error (or deviation from the true effect) that
583	might affect the magnitude and even the direction of the apparent effect" (NRC 2014). Each study
584	meeting inclusion criteria in Group 1, 2, and 3, should be evaluated against a predetermined set of
585	study quality and risk of bias criteria and the results of this evaluation should be presented in the
586	perchlorate MCLG approach report.
F07	Ungertainting Critical to Characterising Changes in The Sid Houses of Levels in December Women
587 588	Uncertainties Critical to Characterizing Changes in Thyroid Hormone Levels in Pregnant Women  Associated with Neurodevelopmental Changes in Offspring
589	The draft MCLG approach presented in USEPA (2017) to predict doses of perchlorate that would result
590	in per unit changes in neurodevelopmental measures, is, as noted by USEPA (2017), "dependent
591	upon predictions from the BBDR model, the derivation of the distribution of fT4, and the evaluations of
592	the relationship between fT4 and neurodevelopment. Each of these steps has inherent uncertainties
593	associated with it."
594	A major source of uncertainty is related to the five studies in Group 1 with data that could be used to
595	quantitatively describe the relationship between thyroid hormone levels in early pregnancy and
596	changes in neurodevelopment (Pop et al. 1999, 2003; Finken et al. 2013; Korevaar et al. 2016;
597	Vermiglio et al. 2004). None of these five studies relied upon data from US populations or have been
598	demonstrated to have iodine intake similar to US populations. Yet according to the American Thyroid
599	Association (Alexander et al. 2017), the reference range of both TSH and fT4 in pregnant women
600	varies depending upon ethnicity. While two studies in Group 1 focused on population groups within

the United States, neither were considered for the model because T4 and not fT4 was measured in the
pregnant females (Oken et al. 2009) and the relationship between fT4 and neurodevelopment was
evaluated in late pregnancy and did not reach statistical significance (Chevrier et al. 2011). USEPA
(2017) (Section 6.5.1) states "there is no reason to believe that the impact of fT4 on
neurodevelopment would differ by country, unless there is a substantial difference in iodine intake".
While USEPA (2017) does make an effort to evaluate changes in iodine intake in women from various
populations, including the US, there are not substantial data reported in the peer-reviewed literature
to validate the conclusions that the impact of fT4 on neurodevelopment would differ by population or
uncertainty in iodine intake levels would have an impact on the derivation of the MCLG. This is
inconsistent with data from the American Thyroid Association (Alexander et al. 2017) that suggest
variability in the distribution of thyroid hormone levels across populations and even within ethnicities
within a single population.
USEPA (2017) also notes that all five studies used for quantitative analysis relied on a one-time fT4
level during pregnancy (Section 6.5.5). The influence of changes in maternal fT4 on fetal brain
development is likely greatest during early pregnancy. The variability in maternal fT4 levels during
pregnancy and the lack of measurement of fT4 at time points throughout pregnancy in the studies
provides a substantial data gap and lack of information needed to validate some of the assumptions
relied upon in the development of the BBDR current model as well as the resulting predictions of the
model. As stated in USEPA (2017),
"Circulating T3 and T4 levels in an individual are maintained within a narrow range by
a negative feedback loop with TSH from the pituitary and TRH from the hypothalamus
that operates around a "set-point." This set-point is different from individual to
individual, which generates a population variance in blood levels of thyroid hormone
that is considerably broader than the individual variance (Andersen, Pedersen, Bruun,
& Laurberg 2002). Therefore, in euthyroid individuals, serum T4 and T3 fluctuate
within a fairly narrow range (about 10% of the population variance), maintained by
the negative feedback relationship with serum TSH from the pituitary gland. This
normal variation creates a situation where single measures of free or total T4 and TSH

are a somewhat imprecise measure of an individual's average T4 and TSH
concentrations (Andersen et al. 2002)."
Several other areas of uncertainty are also highlighted by USEPA (2017). Specifically, USEPA
(2017) noted that none of the five studies carried forward provided iodine intake levels
(Section 6.5.3), which adds significant uncertainty to the estimates. Three of the 5 studies
(Pop et al. 1999, 2003; Vermiglio et al. 2004) also have populations of less than 30
decreasing the statistical power of the studies (section 6.5.4) relied upon for establishing the
relationship between changes in fT4 and neurodevelopmental changes. USEPA (2017) also
noted uncertainties in regard to the analytical methods used to evaluate fT4 levels and while
approaches are being introduced to standardize analytical methods, results at different time
points and from different countries may vary considerably due to differences in analytical
procedures (USEPA 2017). USEPA (2017) also notes that "there is uncertainty regarding the
true fT4 levels at various percentiles in the distribution around the median output from the
BBDR model. This is exemplified by the fact that in this analysis larger unit changes are being
seen with increasing percentiles of fT4 in most analyses." Finally, other confounders such as
iron deficiency were not considered in the analysis. Iron deficiency in pregnant mothers,
which is noted in approximately 18% of pregnant women in the US (Cantor et al. 2015), may
also be associated with hypothyroxinemia (Yu et al. 2015) and failing to directly account for a
relationship between iron deficiency and hypothyroxinemia may introduce an uncertainty into
this analysis.
While all these uncertainties are noted by USEPA (2017), there is no attempt to adjust the
draft MCLG approach in any way to account for these uncertainties. Many of these, especially
confounders such as iron deficiency in the study population and a lack of information on iodide
intake, can have a significant effect in characterizing changes in thyroid hormone levels
associated with changes in neurodevelopmental outcomes. In the absence of adequately
accounting for these uncertainties, it is difficult to have confidence that BBDR model
predictions of small changes in a specific thyroid hormone (e.g. fT4) may accurately predict
the potential for neurodevelopmental effects.

The inadequacy of the USEPA (2017) literature review is substantiated by the comments of	of the
External Peer Review for USEPA's Proposed Approaches to Inform the Derivation of a	
Maximum Contaminant Level Goal for Perchlorate in Drinking Water (Versar 2018).	
Comments regarding the USEPA (2017) literature search included:	

- "The literature search produced ten studies (that assessed maternal serum FT4 concentrations as a continuous measure which did not observe an adverse effect on offspring neurocognition), as well as those in Group 2 that assessed serum FT4 as categorical measures. Although their inclusion may not necessarily be recommended in the final model, comparison of the estimated effects on the various neurocognitive outcomes with and without these may indeed inform the degree of uncertainty inherent in the present model. Several of the studies in Group 2 were able to demonstrate significant adverse outcomes (Berbel 2009 as one excellent example), and also their more global nature would help support the generalizability of the present model."
  - "Excluding these studies lessens the power of the total sample size and thus the ability to detect an association between maternal hypothyroxinemia and any of the offspring outcomes, but provides what may be a somewhat exaggerated estimate of the potential adverse effects of perchlorate exposure. This approach is more conservative, to which there are pros and cons of doing so, toward derivation of a perchlorate MCLG. With this approach, the goal is to minimize exposure to the lowest perchlorate concentration associated with any number of adverse outcomes. I would favor the more liberal public health approach, which is inclusion of all available studies, whether they are positive or negative. Although the perchlorate MCLG may be higher, this latter approach would be consistent with using all available evidence to improve the scientific rigor of the proposed study question."

The peer reviewers also suggested a number of additional peer-reviewed studies that they felt should have been considered to inform BBDR modeling of the quantitative relationship between thyroid hormone levels and neurodevelopmental outcomes:

685	$\infty$	Bárez-López S, Jesus-Obregon M, Bernal J, Guadaño-Ferraz A. 2017. Thyroid Hormone
686		Economy in the Perinatal Mouse Brain: Implications for Cerebral Cortex Development.
687		Cerebral Cortex, 28(5): 1783-1793.
688	$\infty$	Bath S, Steer C, Golding J, Emmett P, Raymen M. 2013. Effect of inadequate iodine
689		status in UK pregnant women on cognitive outcomes in their children: results from the
690		Avon Longitudinal Study of Parents and Children (ALSPAC). The Lancet, 382(9889):
691		331-337.
692	∞	Bernal J. 2017. Thyroid hormone regulated genes in cerebral cortex development.
693		Journal of Endocrinology, 232(2): R83-R97.
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	In the draft MCLG approach, USEPA (2017) focused on five studies that evaluated the relationship of	
	maternal fT4 and several neurodevelopmental endpoints (IQ, mental development index (MDI),	
	psychomotor development index (PDI), standard deviation of reaction time), based on measurements	
	of fT4 during early pregnancy. Results from previous studies have provided the basis for No Observed	
	Effect Levels (NOELs) for health effects of perchlorate in the development of Reference Doses and	
	currently recommended Health Reference Levels (HRLs), including Greer et al. (2002) in which adult	
	men and women were exposed to perchlorate in drinking water at doses of 0.007, 0.02, 0.1, or 0.5	
	mg/kg/day for 14 days demonstrated a NOEL for perchlorate inhibition of radioiodide uptake by the	
	thyroid NIS following exposure to 7 $\mu g/kg/day$ . The point of departure from the Greer et al. (2002)	
	study represents a perchlorate level that precedes the inhibition of iodine uptake by the thyroid. The	
	NAS RfD developed based on the point of departure (POD) from this study is a deviation from the	
	Agency's traditional approach of using a No Observed Adverse Effect Level (NOAEL) for regulatory	
	actions. The NAS's use of a No Observed Effect Level (NOEL) is based on "using a nonadverse effect	
	that is upstream of the adverse effect [which] is a more conservative and health protective approach".	
	While these studies have not been conducted in pregnant women (the population of interest for the	
	draft MCLG approach), as noted by in USEPA (2017):	
	"the BBDR model predicts very little difference in non-pregnant and first-trimester	
	response to perchlorate. This likely occurs because the half-life of (organified) iodine	
	in the adult thyroid is around six months, hence the availability of thyroidal iodine in	
	the first trimester pregnant woman is determined to a very large extent by her	
	nutrition and perchlorate exposure several years preceding pregnancy."	
	This suggests that a comparison of the current modeling results to those from studies conducted in	
adults should provide insight into the predictions of the model and the conclusions regarding the		
	changes in thyroid hormone levels that may result in neurodevelopmental effects.	

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742	The current draft approach for deriving the MCLG assumes any exposure to perchlorate reduces fT4 to
743	some extent (p. 3-17 of USEPA 2017). In addition, linear regression analyses conducted to evaluate
744	the relationship between changes in fT4 and neurodevelopmental effects further assumes any change
745	in fT4 results in some risk of neurodevelopmental effects. These assumptions are in contrast to the
746	results from Greer et al. (2002) in which exposures to perchlorate were as high as 500 $\mu g/kg/day$ and
747	no impact on thyroid hormone levels was observed. This was true for both men and women. In
748	addition, in a study conducted by Braverman et al. (2006), 6 months of exposure to perchlorate in
749	capsules at doses up to 3 mg/day (approximately 40 µg/kg/day) was reported to have no effect on
750	thyroid function, including inhibition of thyroid iodide uptake as well as serum levels of thyroid
751	hormones, TSH, and Tg in a small group of volunteers.
752	USEPA (2017) notes (p. 6-16) that from results of the literature review, it appears the relationship
753	between maternal fT4 and fetal brain development has a temporal relationship, with this influence
754	likely being greatest in early pregnancy (i.e. prior to mid-gestation). The focus of the evaluation is on
755	gestational weeks 12, 13, and 16, where the mother's fT4 levels will have the greatest impact on the
756	fetus. This should allow for comparison to the model results in pregnant women to results from
757	previous studies focused on identification of perchlorate concentrations that would impact fT4 levels in
758	adult women, such as the Greer et al. (2002) study.
759	Based on the BBDR model predictions, USEPA (2017) estimates that a perchlorate dose of 0.3-0.4
760	μg/kg/day would result in a 1% increase in the proportion of the population with hypothyroxinemia
761	and a perchlorate dose of 2.1-2.2 µg/kg/day would result in a 5% increase in proportion of the
762	population with hypothyroxinemia. These modeling results suggest a potential for a significant change
763	in thyroid hormones, as well as adverse effects on neurodevelopment at doses of perchlorate exposure
764	for which there is evidence that decreases in fT4 are not observed. Based on the mode of action
765	proposed by USEPA (2017), decreases in fT4 and increases in TSH would be prerequisite steps for the
766	potential for neurodevelopmental effects. These changes in hormone levels are not observed in the
767	Greer et al. (2002) study following exposure up to 500 $\mu g/kg/day$ . The draft MCLG approach suggests
768	population changes in fT4 would be observed that would shift the proportion of pregnant women that
769	would be hypothyroxinemic at doses of perchlorate below the previously defined NOEL (7 μg/kg/day).

Table 6 (Table 39 of USEPA 2017) provides the predicted dose of perchlorate per unit change in			
neurodevelopmental measure for low iodine intake individuals. Those for IQ are approximately at or			
above (6.5 – 45 $\mu g/kg/day$ ) the NOEL from Greer et al. (2002) and are associated with decreases in			
fT4 of 4.3 to 18.7%. The doses associated with other neurodevelopmental endpoints are 1.7 to 3.0			
$\mu g/kg/day$ and are associated with decreases in fT4 of 1.3 to 2.4%. These percent changes in fT4 are			
very small and considering the potential uncertainty and variability in measuring fT4 levels, there is a			
lack of evidence that such small changes in fT4 will result in clinical observations. Reference ranges			
for fT4 are 0.9 - 2.5 ng/dL in infants (0-5 days) and 0.9 - 1.7 ng/dL in adults (> 20 yrs)			
$(\underline{https://www.mayomedicallaboratories.com/test-catalog/Clinical+and+Interpretive/8725}). \ \ Thus, for a superior of the property of the pr$			
an adult, at the low end of the reference range, we would expect a change from 0.900 to 0.878 ng/dL			
a value that given the number of significant figures in the reference value would not be measurable.			
The dose of perchlorate estimated to result in a 1% or 5% increase in the proportion of			
hypothyroxinemic pregnant women is even lower, ranging from 0.3 to 2.2 $\mu g/kg/day$ . USEPA (2017)			
findings are contrary to multiple studies in adults and pregnant women (Greer et al. 2002; Bravermar			
et al. 2006; Téllez Téllez et al. 2005a, 2005b) provide robust evidence that no impact on iodine			
uptake or thyroid hormone levels would be expected at these dose levels. Based on the mode of			
action proposed by USEPA (2017), these precursor impacts are necessary to generate the			
neurodevelopmental effects derived from the BBDR model.			

Table 6. Predicted Dose of Perchlorate per Unit Change in Neurodevelopmental Measure for Low Iodine
Intake Individuals based on Central Effect Estimates at the Median fT4 level (USEPA 2017)

Study	Endpoint	ΔfT4 in pmol/L  (% ΔfT4 from 0 dose  perchlorate, iodine intake  = 75 μg/day)	Dose of perchlorate  per unit change in  endpoint  (µg/kg/day) <sup>a</sup>
Korevaar et al. (2016) Quadratic	IQ	-1.08 (12.2%)	23
Korevaar et al. (2016) USEPA Independent Analysis: Bivariate	IQ	-0.98 (11.1%)	20
Korevaar et al. (2016) USEPA Independent Analysis: Multivariate	IQ	-1.66 (18.7%)	45
Vermiglio et al. (2004)	IQ	-0.37 (4.3%)	6.5
Pop et al. (2003)	MDI	-0.15 (1.7%)	2.2
Pop et al. (2003)	PDI	-0.12 (1.3%)	1.7
Pop et al. (1999)	PDÍ	-0.12 (1.3%)	1.7
Finken et al. (2013)	SD of Reaction Time  1% or 5% increase in	-0.21 (2.4%)	3.0
BBDR model (USEPA 2017)	proportion of hypothyroxinemic pregnant women <sup>b</sup>	1% or 5%	0.3 - 0.4° [1%] 2.1 - 2.2° [5%]

<sup>&</sup>lt;sup>a</sup> Based on the regression analysis for the range of fT4 data within each study. Central beta estimates of the low iodide intake population (=  $75 \mu g/day$ ) are presented.

 $<sup>^{\</sup>rm b}$  Hypothyroxinemia defined as fT4 < 10 $^{\rm th}$  percentile

 $<sup>^{\</sup>rm c}$  Range based on gestational week used to perform the analysis (12 to 16 weeks).

### DISCUSSION

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A critical review of the (USEPA) 2017 report entitled "Draft Report: Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water", as well as the BBDR model that was proposed for use in derivation of the MCLG, was conducted. Overall, conducting this review and assessment of the BBDR model was beset by multiple challenges and the effort highlighted a number of uncertainties in the use of the model. The main challenges that the review presented were due to the complexity of the BBDR model itself. The co-authors of this review, who are widely considered to be experts in the area of PBPK and BBDR model development, found it difficult to evaluate the complex interactions of model parameters and their relationship to the predictions of the model. In our efforts to reproduce simulations provided in USEPA (2017), it was noted that instructions for running the model for different scenarios, and documentation of the rationale for the model parameter values associated with them, were sometimes inadequate; this deficiency, which is inevitable in a complex model, was exacerbated by the number of code scripts required to set the parameters used to generate the various results in the document. As a result, the ability to independently verify all aspects of the model were impeded by uncertainties associated with the steps necessary to reproduce figures and tables in the report, or to perform comparisons of model predictions to data for alternative exposure scenarios or studies. As suggested by C.A.R. Hoare in his 1980 ACM Turing Award Lecture: "There are two ways of constructing a software design: One way is to make it so simple that there are obviously no deficiencies and the other way is to make it so complicated that there are no obvious deficiencies." By their nature, BBDR models are seldom simple; to the extent that BBDR models attempt to describe complex biological systems they will inherently be difficult to comprehend. The criticisms of the perchlorate PBPK model in this case study are not meant to suggest that the model is incorrect or unuseful, and they should not be taken as criticisms of the utility of BBDR modeling in general. Used appropriately, BBDR models can provide important information for better risk assessment decisionmaking. The issue that needs to be addressed in each case is whether a BBDR model is fit for the intended purpose of using it in the risk assessment. The first use of PBPK modeling in risk assessments dates back to the 1980s (USEPA 1987) and yet the application of PBPK modeling to replace default dosimetry remains controversial, primarily due to

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818	concerns regarding model uncertainty. To address these concerns, the OMB (2007) memorandum on
819	risk analysis recommended the presentation of results from multiple dose-response approaches to
820	provide a more robust risk characterization. In this scenario, a fit-for-purpose BBDR model can
821	provide information on the most scientifically plausible risk estimate for comparison with the results of
822	default approaches (Clewell et al. 2008). Consistent with this OMB recommendation, one focus of our
823	evaluation was determining how the results of the BBDR modeling could inform the likelihood that the
824	current perchlorate guideline (USEPA 2005), which is based on inhibition of thyroidal iodine uptake in
825	adults, is also protective of concerns regarding neurodevelopmental effects of perchlorate. This
826	question is discussed in the Conclusion.
827	The current BBDR model that was relied upon for the USEPA (2017) draft approach is an extension of
828	previous models that have been validated and published in the peer-reviewed literature (Clewell et al.
829	2007; Merrill et al. 2003; Lumen et al. 2013). Similar values for key parameters have been
830	successfully used across the previous models, yet changes were made in the current model or new
831	parameters added (e.g. VCHNG, HCGreg, pTSH, KmNiS), often with little or no evidence or
832	justification provided to support these revisions in the USEPA (2017) documentation. Additional
833	support for these changes will be needed to provide validation of the current revisions to the BBDR
834	model and to provide confidence in the predictions of changes in fT4 made by the model.
835	Certainly, confidence in the BBDR model predictions is undermined by the model's inability to simulate
836	the results from the Steinmaus et al. (2016) study. In Appendix B of USEPA (2017), a comparison of
837	the predicted changes in both fT4 and TSH from the BBDR model were compared to the results
838	reported by Steinmaus et al. (2016) (reproduced in Figure 6). The Steinmaus et al. (2016) study was
839	conducted to evaluate the potential for perchlorate exposure to impact thyroid hormone levels in
840	pregnant women in San Diego. This comparison clearly highlights the differences between the model
841	predictions and those from a published study. The baseline BBDR simulations with normal iodine
842	intake (170 $\mu$ g/day) demonstrate no change in fT4, which is consistent with other studies in which no
843	impact on fT4 has been observed at doses up to 7 $\mu g/kg/day$ (Greer et al. 2002; Braverman et al.
844	2006). The BBDR model underpredicted changes in fT4, even in the scenario with low dietary iodine
845	intake (75 $\mu g/day$ ), when compared to the changes reported by Steinmaus et al. (2016). This
846	discrepancy calls into question the ability of the model to predict changes in fT4 associated with

perchlorate exposure. In particular, the proposed MCLG approach depends on model predictions of
small changes in fT4 as low as approximately 1% (Table 6) being associated with unit changes in
neurodevelopmental endpoints. Predictions of this precision would require a level of model precision
that has not been demonstrated by comparisons to existing data.
Many of the changes in fT4 that are predicted by the draft MCLG approach to estimate impact on the
population distribution of fT4 and therefore result in per unit changes in neurodevelopmental
outcomes are small percent changes (some as low as a 1.3-4.3% change). This would appear to
suggest that the extended version of the BBDR model has a capability to estimate small changes in
fT4 with a level of precision that is not demonstrated by any adequate validation. In fact, BBDR mode
predictions of fT4 underpredict observed data in human studies (Tables 1 and 3) by as much as 25-
35%. Moreover, considering the variability of fT4 in the populations of interest, there is uncertainty as
to whether these slight changes could be measured clinically, considering the greater impact of iodine
intake on hormone levels. Considering the lack of data to support critical parameters and
assumptions in the model, as well as the impact of the variability of iodine intake on model
predictions, it seems crucial that validation of the BBDR model by comparison with observed data be
used to provide confidence in the predictions of the BBDR model. However, the BBDR model clearly
fails the only comparison that has been conducted (Figure 6), with the BBDR model predictions falling
outside the bounds of the statistical confidence limits estimated for the Steinmaus et al. (2016)
relationship between perchlorate dose and fT4. Each of the components of the BBDR model combined
result in compounded uncertainty in the modeling results.
Until additional data are available to validate current extensions of the BBDR model to the pregnant
woman, the Greer et al. (2002) and Braverman et al. (2006) studies provide the critical information in
determining concentrations of perchlorate that do not result in significant inhibition of iodide uptake
and, therefore, impacts on fT4. Based on recommendations from the National Academy of Sciences
(2005), points of departure provided by these studies used in combination with uncertainty factors
were considered to be protective of sensitive subpopulations, this approach has previously been relied
upon to support guidelines for perchlorate in drinking water under the Safe Drinking Water Act (USEPA
2008), and has also been used more recently by JECFA (2011) and EFSA (2014) in their regulation of
perchlorate.

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We applaud the USEPA for the application of a BBDR model in their draft MCLG approaches, as these models integrate the available science for a compound of interest. However, while the hormone component of the model is a scientific improvement in terms of incorporating the available biology, there is a lack of data to provide critical validation in multiple steps of the proposed approach and to support several assumptions/parameters within the BBDR model. In particular, while no major structural defects in the USEPA (2017) BBDR model were identified, there are a number of uncertainties in the model parameterization that call into question its use for predicting very small changes in clinical hormone values, such as a 1% change in fT4 (Tables 4-6) While the model prediction for 1% change in fT4 (0.3-0.4 µg/kg/day) would yield a POD lower than the USEPA (2005) RfD, that level of precision is a not supported by the comparison of the model predictions with available data. Nonetheless, the consistency of the model-predicted PODs based on the epidemiological endpoints (Table 6), and the relationship of these results with previous risk assessments based on biologically sound precursors (iodide inhibition in thyroid), indicate that the interim health standard would be sufficiently protective against the developmental neurological endpoints of concern, as illustrated in Figure 7, which compares the point of departure from the USEPA (2005) IRIS assessment with the PoDs calculated by the BBDR model in the USEPA (2017) report (Table 6). The USEPA (2005) RfD (red bar) is protective for all of the endpoints from epidemiological studies and is consistent with a change in population fT4 levels of less than 5%.

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Beginning with the initial risk characterization for perchlorate (USEPA 2002), the fundamental underpinning of the agency's risk assessment approach has been the use of an obligatory precursor as a conservative basis for protecting against downstream health effects. As elaborated in the original documentation (USEPA 2003), the effects of perchlorate are mediated by the inhibition of thyroidal iodine uptake by perchlorate. Unless perchlorate concentrations in the blood are sufficient to disrupt iodine uptake, there is no plausible basis for suggesting an effect of perchlorate on thyroid hormone homeostasis or subsequent events leading to developmental or (in the rat) carcinogenic effects. The recent studies suggesting a relationship between perchlorate exposure and decreased fT4 do not impeach this causal relationship. Therefore, until the significant uncertainties in the current BBDR

model and draft MCLG approaches can be addressed, USEPA should continue to rely on the RfD
approach based on inhibition of thyroidal iodine uptake (USEPA (2005), as recommended by the
National Academy of Sciences (2005) for any further regulatory action. The USEPA (2005) RfD
includes an intraspecies uncertainty factor of 10 "to protect the most sensitive population, the fetuses
of pregnant women who might have hypothyroidism or iodide deficiency." None of the predictions of
the BBDR model suggest that this uncertainty factor is inadequate.

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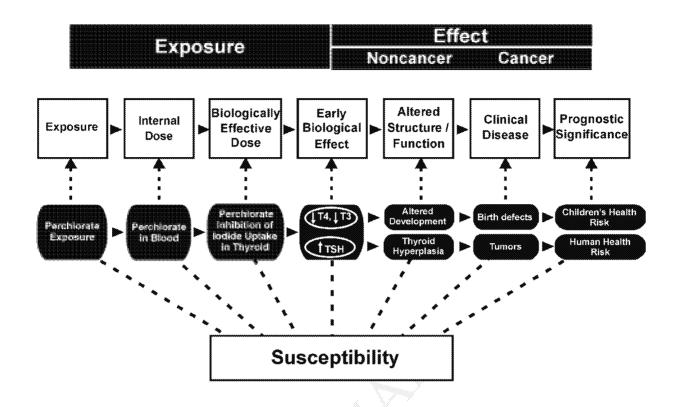
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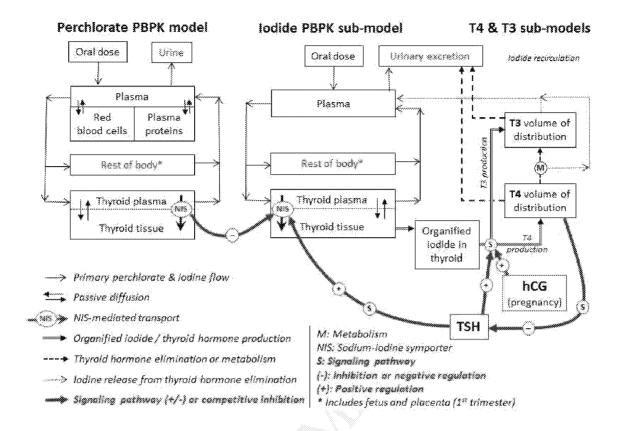
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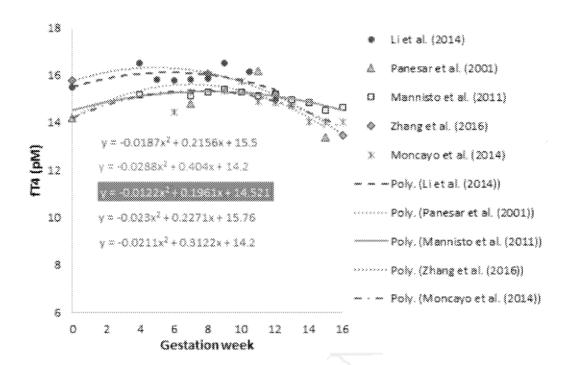
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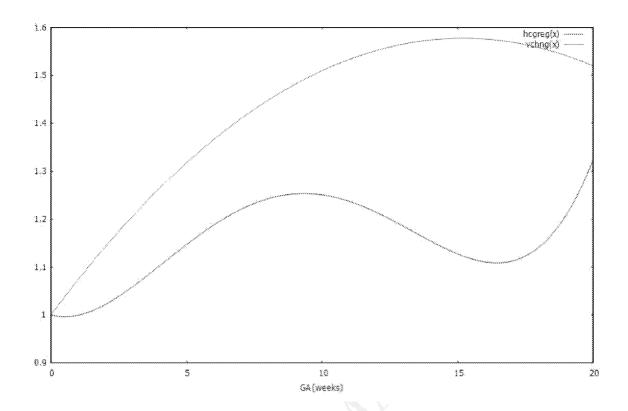
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1141	Figure 1. Mode-of-action model for perchlorate toxicity proposed by the USEPA (2002).
1142	Inhibition of iodide uptake in the thyroid by perchlorate is an obligatory precursor for all
1143	downstream cancer and noncancer endpoints, including neurodevelopment.
1144	Figure 2: Structure of the Early Pregnancy BBDR (USEPA 2017)
1145	Figure 3: Variation in free T4 (fT4 in early pregnancy (as reported in USEPA 2017))
1146	Figure 4, Comparison of parameters controlling hCG-dependent changes in thyroidal uptake
1147	(VCHNG, green) and thyroid hormone production rate (HCGREG, purple) in the BBDR model as
1148	a function of gestational age. Despite the fact that both parameters are dependent upon hCG
1149	levels, the predicted trends across gestation are not consistent.
1150	Figure 5. Panel a: Model predictions for free T4 (fT4) in non-pregnant women as a function of
1151	iodine intake compared to data from NHANES 2007-2012 (USEPA 2017). Panel b: Underlying
1152	NHANES data without model predictions. Note the lack of evidence for any correlation
1153	between iodine intake and fT4 in the NHANES data in the range from 20 to 90 $\mu g/d$ , in
1154	contrast to model predictions.
1155	Figure 6. Comparison of BBDR model predicted free T4 (fT4) changes as a function of
1156	perchlorate dose with data from Steinmaus et al. (2016). Reproduced from USEPA (2017).
1157	Blue boxes and diamonds represent the BBDR model predictions for median (170 $\mu g/d$ ) and
1158	low (90 $\mu$ g/d) iodine intake populations (GW 13-16); red +'s represent the central estimate
1159	from the analysis of the Steinmaus et al. (2016) study and the red x's represent the upper
1160	and lower confidence limits for that estimate.
1161	Figure 7. Comparison of PoDs calculated using the USEPA (2017) BBDR model-based PoDs
1162	(blue and green bars) with the USEPA (2005) RfD (red bar).
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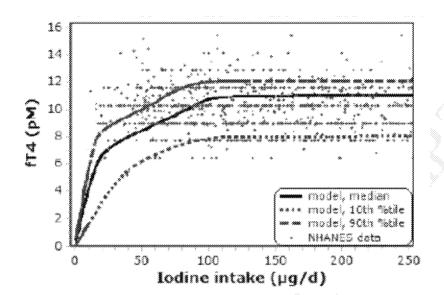




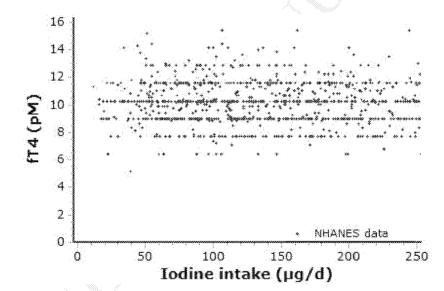


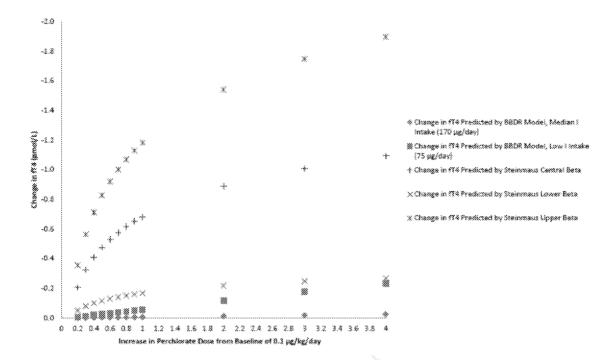


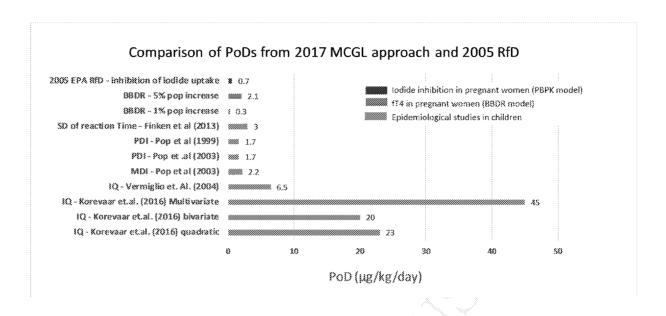
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### Highlights (maximum 125 characters, including spaces)

- $_{\infty}$  The USEPA (2017) BBDR model plausibly describes perchlorate effects on thyroid hormone regulation during early pregnancy.
- $\infty$  The model is a valuable tool for investigating the effects of perchlorate on thyroid function during early gestation.
- $_{\infty}$  BBDR modeling results indicate that the current USEPA RfD, based on adult effects, is also protective for fetal effects.
- $\infty$  However, current model uncertainties dictate against its use to replace the existing RfD for perchlorate.

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ENVIRONMENTAL PROTECTION AGENCY

40 CFR Parts 141 and 142

[EPA-HQ-OW-2018-0780, EPA-HQ-OW-2008-0692, EPA-HQ-OW-2009-0297; FRL-

XXXX-XX-OW]

**RIN 2040-AF28** 

**Drinking Water: Notice of Final Action on Perchlorate** 

**AGENCY:** Environmental Protection Agency (EPA).

**ACTION:** Final Action.

**SUMMARY:** The Environmental Protection Agency (EPA) is announcing its withdrawal of the 2011 determination to regulate perchlorate in accordance with the Safe Drinking Water Act (SDWA). On February 11, 2011, the EPA published a Federal Register notice in which the Agency determined that perchlorate met the SDWA's criteria for regulating a contaminant. On June 26, 2019, the EPA published a proposed national primary drinking water regulation (NPDWR) for perchlorate and requested public comments on multiple alternative regulatory actions, including the alternative of withdrawing the 2011 regulatory determination for perchlorate. The EPA received approximately 1,500 comments on the proposed rule. The EPA has considered these public comments and based on the best available information the Agency is

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* withdrawing the 2011 regulatory determination and is making a final determination to not regulate perchlorate. The EPA has determined that perchlorate does not occur with a frequency and at levels of public health concern, and that regulation of perchlorate does not present a meaningful opportunity for health risk reduction for persons served by public water systems.

**DATES**: For purposes of judicial review, the regulatory determination in this document is issued as of *[insert date of publication in the Federal Register]*.

**FOR FURTHER INFORMATION CONTACT**: Samuel Hernandez, Office of Ground Water and Drinking Water, Standards and Risk Management Division (Mail Code 4607M), Environmental Protection Agency, 1200 Pennsylvania Avenue, NW, Washington, DC 20460; telephone number: (202) 564-1735; email address: hernandez.samuel@epa.gov.

**SUPPLEMENTARY INFORMATION**: This notice is organized as follows:

### I. General Information

- A. Does this Action Apply to Me?
- B. How can I get Copies of this Document and other Related Information?

### II. Background

- A. What is Perchlorate?
- B. What is the Purpose of this Action?
- C. What is the EPA's statutory authority for this action?
- D. Statutory Framework and Perchlorate Regulatory History

### III. Final Regulatory Determination for Perchlorate

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- A. May perchlorate have an adverse effect on the health of persons?
- B. Is perchlorate known to occur or is there a substantial likelihood that perchlorate will occur in public water systems with a frequency and at levels of public health concern?
- C. Is there a meaningful opportunity for the reduction of health risks from perchlorate for persons served by public water systems?
- D. What is the EPA's final regulatory determination on perchlorate?

### IV. Summary of Key Public Comments on Perchlorate

- A. Health Effects Assessment
- B. Occurrence
- C. Regulatory Proposal and Alternatives
- D. SDWA Statutory Requirements
- E. Regulatory Determination Withdrawal

### V. Conclusion

### VI. References

### I. General Information

A. Does This Action Apply to Me?

This action will not impose any requirements on anyone. Instead, this action notifies interested parties of the EPA's withdrawal of the 2011 regulatory determination for perchlorate and the final regulatory determination to not regulate perchlorate. This notice also provides a

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summary of the major comments received on the June 26, 2019 (84 FR 30524) proposed
NPDWR for perchlorate.

B. How can I get Copies of this Document and other Related Information?

The EPA has established a docket for this action under Docket ID No. EPA-HQ-OW-2018-0780. Publicly available docket materials are available electronically at [ HYPERLINK "http://www.regulations.gov/docket?D=EPA-HQ-OW-2018-0780" ].

### II. Background

### A. What is Perchlorate?

Perchlorate is a negatively charged inorganic ion that is comprised of one chlorine atom bound to four oxygen atoms (ClO<sub>4</sub>-), which is highly stable and mobile in the aqueous environment. Perchlorate comes from both natural and manmade sources. It is formed naturally via atmospheric processes and can be found within mineral deposits in certain geographical areas. It is also produced in the United States, and the most common compounds include ammonium perchlorate and potassium perchlorate used primarily as oxidizers in solid fuels to power rockets, missiles, and fireworks. Perchlorate can also result from the degradation of hypochlorite solutions used for water disinfection. The degradation into perchlorate occurs when hypochlorite solutions are improperly stored and handled. For the general population, most perchlorate exposure is through the ingestion of contaminated food or drinking water. At certain levels, perchlorate can prevent the thyroid gland from getting enough iodine, which can affect thyroid hormone

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production. For pregnant women with low iodine levels, sufficient changes in thyroid
hormone levels may cause changes in the child's brain development. For infants, changes
to thyroid hormone function can also impact brain development.

### B. What is the purpose of this action?

The purpose of this action is to publish the EPA's notice to withdraw the 2011 regulatory determination and issue a final determination to not regulate perchlorate in drinking water. This notice presents the EPA's basis for this withdrawal and final regulatory determination, and the EPA's response to key issues raised by commenters in response to the June 26, 2019 (84 FR 30524) proposed rule (referred to hereinafter as "the 2019 proposal").

### C. What is the EPA's statutory authority for this action?

The SDWA sets forth three criteria that must be met for the EPA to issue a maximum contaminant level goal (MCLG) and promulgate a national primary drinking water regulation (NPDWR). Specifically, the Administrator must determine that (1) "the contaminant may have an adverse effect on the health of persons"; (2) "the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern"; and (3) "in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by public water systems" (SDWA 1412(b)(1)(A)).

The EPA has determined, based on data and analysis since the issuance of the 2011 regulatory determination, that perchlorate does not in fact meet the statutorily-prescribed criteria

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* for regulation. As described in Sections III & VI of the 2019 proposal, the data and analysis in the record indicate that perchlorate does not occur in public water systems with a frequency and at levels of public health concern. Specifically, the peer-reviewed health effects analysis indicates that the concentration of perchlorate representing the levels of public health concern (i.e., the proposed MCLG levels, 18-90 µg/L) is higher than the concentration considered in issuance of the 2011 regulatory determination (1-47 µg/L) (USEPA, 2019a). In addition, based on an evaluation of the nationally representative UCMR 1 systems, the updated occurrence analysis shows that the frequency of occurrence of perchlorate in public water systems at levels exceeding any of the alternative proposed MCLGs is significantly lower (0.38% - 0.02%) than the frequency considered in the analysis for the 2011 regulatory determination (4% - 0.39%) (USEPA, 2019b). The EPA estimates that, even at the most stringent regulatory level considered in the 2019 proposal (18 µg/L), not more than 15 systems (0.03% of all water systems in the U.S.) would need to take action to reduce levels of perchlorate. Based on this information, the EPA determines that perchlorate does not occur in public water systems "with a frequency...of public health concern" and thus does not meet the second criterion of the three required for regulation under the SDWA. In addition, while the third criterion is "in the sole judgement of the Administrator," the low occurrence provides ample support for the EPA's conclusion that the regulation of perchlorate does not present a "meaningful opportunity for health risk reduction for persons served by public water systems," within the meaning of 1412(b)(1)(A)(iii). Accordingly,

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because perchlorate no longer meets the statutory criteria for regulation, the EPA does not have
the authority to issue a MCLG or promulgate a NPDWR for perchlorate.

The EPA's decision to withdraw the regulatory determination is supported by the legislative history underlying the 1996 amendments to the SDWA, which repealed the statutory requirement for the EPA to regulate an additional 25 contaminants every 3 years and replaced it with the current requirement for the EPA to determine whether regulation is warranted for five contaminants every five years. In describing the need for such amendment, the legislative history points to the view expressed at the Committee Hearing that "the current law is a one-size-fits-all program. It forces our water quality experts to spend scarce resources searching for dangers that often do not exist rather than identifying and removing real health risks from our drinking water" (S. Rep. 104-169 (1995) at 12). This amendment reflected Congress' clear intent that the EPA prioritize actual health risks in determining whether to regulate any particular contaminant. See id at 12 (noting that the amendment "repeals the requirement that the EPA regulate an additional 25 contaminants every 3 years replacing it with a new selection process that gives the EPA the discretion to identify contaminants that warrant regulation in the future").

The EPA's decision to withdraw the regulatory determination is also consistent with Congress' direction to prioritize the SDWA decisions based on the best available public health information. *See* 1412(b)(1)(B)(ii)(II) (findings supporting a determination to regulate "shall be based on the best available public health information"); 1412(b)(2)(A) (requiring that the EPA use "the best available, peer-reviewed science and supporting studies..." in carrying out any

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* actions under this section). Although the EPA determined in 2011 that perchlorate met the criteria for regulation, new data and analysis developed by the Agency as part of the 2019 proposal demonstrate that the occurrence and health effects information used as the basis for the 2011 determination no longer constitute "best available information," are no longer accurate and no longer support the Agency's prioritization of perchlorate for regulation. Accordingly, not only is EPA not authorized to issue a MCLG or promulgate a NPDWR for perchlorate, but it would not be in the public interest to do so.

The EPA recognizes that the Act does not include a provision explicitly authorizing withdrawal of a regulatory determination. However, such authority is inherent in the authority to issue a regulatory determination under 1412(b)(1)(B)(ii)(II), particularly given the requirement that such determination be based on the "best available public health information," as discussed above. Accordingly, the EPA must have the inherent authority to withdraw a regulatory determination if the underlying information changes between regulatory determination and promulgation. In light of its concern that the EPA focus new contaminant regulations on priority health concerns, Congress could not have intended that the EPA's regulatory decision-making be hamstrung by older data when newer, more accurate scientific and public health data are available, especially when those data demonstrate that regulation of a new contaminant would not present a meaningful opportunity for health risk reduction.

Moreover, the EPA notes that the statute specifically provides that a decision to <u>not</u> regulate a contaminant is a final Agency action subject to judicial review. SDWA, section

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1412(b)(1)(B)(ii)(IV). Congress could have – but did not – specify the same with respect to determinations to regulate. Congress also did not explicitly prohibit the EPA from withdrawing or modifying a regulatory determination. Congress' silence with respect to determinations to regulate suggests that Congress intended that such a determination is not itself a final agency action, but rather a preliminary step in a decision-making process culminating in a NPDWR and thus subject to reconsideration based on new data and analysis considered during the 36 month promulgation process specified in the statute. Accordingly, reconsideration of this preliminary finding – and withdrawal of the determination based on subsequent analysis mandated for NPDWR development – is fully consistent with the statutory decision-making framework.

Section 1412(b)(1)(B)(i) of the SDWA requires the EPA to publish every five years a Contaminant Candidate List (CCL). The CCL is a list of drinking water contaminants that are known or anticipated to occur in public water systems and are not currently subject to federal drinking water regulations. The EPA uses the CCL to identify priority contaminants for regulatory decision-making and information collection. Contaminants listed on the CCL may require future regulation under the SDWA. The EPA included perchlorate on the first, second, and third CCLs published in 1998 (63 FR 10274), 2005 (70 FR 9071), and 2009 (74 FR 51850).

D. Statutory Framework and Perchlorate Regulatory History

The EPA collects data on the CCL contaminants to better understand their potential health effects and to determine the levels at which they occur in public water systems. SDWA, section 1412(b)(1)(B)(ii) requires that, every five years, the EPA, after consideration of public

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* comment, issue a determination of whether or not to regulate at least five contaminants on each CCL. For any contaminant that the EPA determines meets the SDWA criteria for regulation, under SDWA, section 1412(b)(1)(E), the EPA must propose a NPDWR within two years and promulgate a final regulation within 18 months of the proposal (which may be extended by 9 additional months).

As part of its responsibilities under the SDWA, the EPA implements section 1445(a)(2), "Monitoring Program for Unregulated Contaminants." This section requires that once every five years, the EPA issue a list of no more than 30 unregulated contaminants to be monitored by public water systems. This monitoring is implemented through the Unregulated Contaminant Monitoring Rule (UCMR), which collects data from community water systems and non-transient, non-community water systems. The first four UCMRs collected data from a census of large water systems (serving more than 10,000 people) and from a statistically representative sample of small water systems. On September 17, 1999, the EPA published its first UCMR (64 FR 50556), which required all large systems and a representative sample of small systems to monitor for perchlorate and 25 other contaminants (USEPA, 1999). Water system monitoring data for perchlorate was collected from 2001 to 2005.

The EPA and other federal agencies asked the National Research Council (NRC) to evaluate the health implications of perchlorate ingestion. In its 2005 report, the NRC concluded

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* that perchlorate exposure inhibits the transport of iodide<sup>1</sup> into the thyroid by a protein molecule known as the sodium/iodide symporter (NIS), which may lead to decreases in two thyroid hormones, thyroxine (T3) and triiodothyronine (T4), and increases in thyroid-stimulating hormone (TSH) [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"a1mn5hjprkt","properties":{"formattedCitation":"(National Research Council (NRC), 2005b)", "plainCitation": "(National Research Council (NRC), 2005b)", "noteIndex":0}, "citationItems":[{"id":350, "uris":["http://zotero.org/groups/945096/item s/TN6HMC9D"], "uri": ["http://zotero.org/groups/945096/items/TN6HMC9D"], "itemData": {"id": 350, "type": "book", "title": "Health Implications of Perchlorate Ingestion", "publisher": "National Academies Press", "publisher-place": "Washington, DC", "event-place": "Washington, DC", "author": [{"literal": "National Research Council (NRC)"}], "issued": {"dateparts":[["2005"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Additionally, the NRC concluded that the most sensitive population to perchlorate exposure are "the fetuses of pregnant women who might have hypothyroidism or iodide deficiency" (p. 178). The EPA established a reference dose (RfD) consistent with the NRC's recommended RfD of 0.7 µg/kg/day for perchlorate. The reference dose is an estimate of a human's daily exposure to perchlorate that is likely to be without an

<sup>&</sup>lt;sup>1</sup> For the purposes of this notice, "iodine" will be used to refer to dietary intake before entering the body. Once in the body, "iodide" will be used to refer to the ionic form.

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* appreciable risk of adverse effects. This RfD was based on a study [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "a3u94lt6me", "properties": {"formattedCitation": "(Greer, Goodman, Pleus, & Greer, 2002)", "plainCitation": "(Greer, Goodman, Pleus, & Greer, 2002)","noteIndex":0},"citationItems":[{"id":387,"uris":["http://zotero.org/groups/945096/items/ 6AKUNIX6"], "uri": ["http://zotero.org/groups/945096/items/6AKUNIX6"], "itemData": {"id":387 "type":"article-journal", "title":"Health effects assessment for environmental perchlorate contamination: the dose response for inhibition of thyroidal radioiodine uptake in humans", "container-title": "Environmental Health Perspectives","page":"927","volume":"110","issue":"9","author":[{"family":"Greer","given":"M onte A."}, {"family": "Goodman", "given": "Gay"}, {"family": "Pleus", "given": "Richard C."},{"family":"Greer","given":"Susan E."}],"issued":{"dateparts":[["2002"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] of perchlorate's inhibition of radioactive iodine uptake in healthy adults and the application of an uncertainty factor of 10 for intraspecies variability [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID": "00Hz805e", "properties": {"formattedCitation": "(USEPA, 2005b)", "plainCitation": "(USEPA, 2005b)", "noteIndex":0}, "citationItems":[{"id":980, "uris":["http://zotero.org/groups/945096/item s/LHANJBR6"],"uri":["http://zotero.org/groups/945096/items/LHANJBR6"],"itemData": {"id":9 80,"type":"article","title":"Integrated Risk Information System (IRIS) Chemical Assessment

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Summary: Perchlorate (ClO4-) and Perchlorate Salts","publisher":"USEPA National Center for Environmental Assessment","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2005"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

In October 2008, the EPA published a preliminary regulatory determination to not regulate perchlorate in drinking water and requested public comment (73 FR 60262). In that preliminary determination, the EPA found that perchlorate did not occur with a frequency and at levels of public health concern and that development of a regulation did not present a meaningful opportunity for health risk reduction for persons served by public water systems. The EPA derived and used a Health Reference Level (HRL) of 15 µg/L based on the RfD of 0.7 µg/kg/day and body weight and exposure information for pregnant women in making this conclusion [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"FZ6WMtAv","properties": {"formattedCitation":"(USEPA, 2008a)","plainCitation":"(USEPA,

2008a)","noteIndex":0},"citationItems":[{"id":934,"uris":["http://zotero.org/groups/945096/items/HBX88QM9"],"uri":["http://zotero.org/groups/945096/items/HBX88QM9"],"itemData":{"id":934,"type":"article-journal","title":"Drinking water: Preliminary regulatory determination on perchlorate","container-title":"Federal

Register", "volume": "73", "issue": "198", "abstract": "SUMMARY: This action presents EPA's preliminary regulatory determination for perchlorate in accordance with the Safe Drinking Water

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* Act (SDWA). The Agency has determined that a national primary drinking water regulation (NPDWR) for perchlorate would not present \"a meaningful opportunity for health risk reduction for persons served by public water systems." The SDWA requires EPA to make determinations every five years of whether to regulate at least five contaminants on the Contaminant Candidate List (CCL). EPA included perchlorate on the first and second CCLs that were published in the Federal Register on March 2, 1998 and February 24, 2005. Most recently, EPA presented final regulatory determinations regarding 11 contaminants on the second CCL in a notice published in the Federal Register on July 30, 2008. In today's action, EPA presents supporting rationale and requests public comment on its preliminary regulatory determination for perchlorate. EPA will make a final regulatory determination for perchlorate after considering comments and information provided in the 30-day comment period following this notice. EPA plans to publish a health advisory for perchlorate at the time the Agency publishes its final regulatory determination to provide State and local public health officials with technical information that they may use in addressing local contamination.", "ISSN": "ISSN 0097-6326 EISSN 2167-2520", "shortTitle": "Federal Register", "journalAbbreviation": "Fed. Reg.", "language": "English", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2008"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Using the UCMR 1 occurrence data, the EPA estimated that less than 1% of drinking water systems (serving approximately 1 million people)

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had perchlorate levels above the HRL of 15 µg/L. Based on this information the EPA found that

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perchlorate did not occur at a frequency and at levels of public health concern. The EPA also determined there was not a meaningful opportunity for a NPDWR for perchlorate to reduce health risks.

In August 2009, the EPA published a supplemental request for comment with new analysis that derived potential alternative Health Reference Levels (HRLs) for 14 life stages, including infants and children. The analysis used the RfD of 0.7 µg/kg/day and life stage-specific bodyweight and exposure information, resulting in comparable perchlorate concentrations in drinking water, based on life stage, of between 1 µg/l to 47 µg/l (74 FR 41883; USEPA, 2009).

In February 11, 2011, the EPA published its determination to regulate perchlorate (76 FR 7762; USEPA, 2011) after careful consideration of public comments on the October 2008 and August 2009 notices. The EPA found at that time that perchlorate may have an adverse effect on the health of persons, it is known to occur in public drinking water systems with a frequency and at levels that present a public health concern, and regulation of perchlorate presented a meaningful opportunity for health risk reduction for persons served by public water systems. The EPA stated then that: "Based on the data in Table 1 and the range of potential alternative HRLs, EPA has determined that perchlorate is known to occur or there is a substantial likelihood that it will occur with a frequency and at levels of public health concern." (USEPA, 2011, p. 7765). The EPA found that as many as 16 million people could potentially be exposed to perchlorate at levels of concern, up from 1 million people originally estimated in the 2008 notice.

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As a result of the determination, and as required by SDWA, section 1412(b)(1)(E), the EPA initiated the process to develop a MCLG and a NPDWR for perchlorate.

In September 2012, the U.S. Chamber of Commerce (the Chamber) submitted to the EPA a Request for Correction under the Information Quality Act regarding the EPA's regulatory determination. In the request, the Chamber claimed that the UCMR 1 data used in the EPA's occurrence analysis did not comply with data quality guidelines and were not representative of current conditions. In response to this request, the EPA reassessed the data and removed certain source water samples that could be paired with appropriate follow-up samples located at the entry point to the distribution system. The EPA also updated the UCMR 1 data in the analysis for systems in California and Massachusetts, using state compliance data to reflect current occurrence conditions after state regulatory limits for perchlorate were implemented.

As required by section 1412(d) of the SDWA, as part of the NPDWR development process, the EPA requested comments from the Science Advisory Board (SAB) in 2012, seeking guidance on how best to consider and interpret the life stage information, the epidemiologic and biomonitoring data since the NRC report, physiologically-based pharmacokinetic (PBPK) analyses, and the totality of perchlorate health information to derive an MCLG for perchlorate. In May 2013, the SAB recommended that the EPA:

 derive a perchlorate MCLG that addresses sensitive life stages through physiologically-based pharmacokinetic/pharmacodynamic modeling based upon its mode of action rather than the default MCLG approach using the RfD and specific chemical exposure parameters;

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- expand the modeling approach to account for thyroid hormone perturbations and potential adverse neurodevelopmental outcomes from perchlorate exposure;
- utilize a mode-of-action framework for developing the MCLG that links the steps in the
  proposed mechanism leading from perchlorate exposure through iodide uptake inhibition—to
  thyroid hormone changes—and finally to neurodevelopmental impacts; and
- "Extend the [BBDR] model expeditiously to . . . provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (SAB for the U.S. EPA, 2013, p. 19).

To address the SAB recommendations, the EPA revised an existing PBPK/PD model that describes the dynamics of perchlorate, iodide, and thyroid hormones in a woman during the third trimester of pregnancy (Lumen, Mattie, & Fisher, 2013; USEPA, 2009b). The EPA also created its own Biologically Based Dose Response (BBDR) models that included the additional sensitive life stages identified by the SAB, *i.e.*, breast- and bottle-fed neonates and infants (SAB for the U.S. EPA, 2013, p. 19).

To determine whether the Agency had implemented the SAB recommendations for modeling thyroid hormone changes, the EPA convened an independent peer review panel to evaluate the BBDR models in January 2017 (External Peer Reviewers for USEPA, 2017). The EPA considered the recommendations from the 2017 peer review and made necessary model revisions to increase the scientific rigor of the model and the modeling results.

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The EPA convened a second independent peer review panel in January 2018 to evaluate the revisions to the BBDR model. The EPA also presented several approaches to link the thyroid hormone changes in a pregnant mother predicted by the BBDR model to neurodevelopmental effects using evidence from the epidemiological literature (External Peer Review for U.S. EPA, 2018).

In response to a lawsuit brought to enforce the deadlines in SDWA, section 1412(b)(1)(E), on October 18, 2016, the U.S. District Court for the Southern District of New York entered a consent decree, requiring the EPA to sign for publication a proposal for a MCLG and NPDWR for perchlorate in drinking water no later than October 31, 2018, and to sign for publication a final MCLG and NPDWR for perchlorate in drinking water no later than December 19, 2019. The deadline for the EPA to propose a MCLG and NPDWR for perchlorate in drinking water was later extended to May 28, 2019, and the date for signature of a final MCLG and NPDWR was extended to be no later than June 19, 2020. The consent decree is available in the docket for this action.

In compliance with the deadline established in the consent decree, on May 23, 2019, the EPA Administrator signed a proposed rulemaking notice seeking public comment on a range of options regarding the regulation of perchlorate in public drinking water systems. The proposed rulemaking notice was published in the *Federal Register* on June 26, 2019. 84 Fed. Reg. 30524. The EPA proposed a NPDWR for perchlorate with an MCL and MCLG of 56 μg/L. The proposed MCLG of 56 μg/L was based on avoiding a 2 point IQ decrement associated with

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exposure to perchlorate in drinking water during the most sensitive life stage (the fetus) within a specific segment of the population (iodine deficient pregnant women).

The EPA also requested comment on two alternative MCL/MCLG values of 18 µg/L and 90 µg/L. These alternatives were based upon avoiding 1 point and 3 point IQ decrements respectively, associated with perchlorate exposure. Additionally, the EPA requested comment on whether the 2011 regulatory determination should be withdrawn, based on new information including updated occurrence data on perchlorate in drinking water and new analysis of the concentration of perchlorate in drinking water that represents a level of health concern.

# III. Withdrawal of the 2011 Regulatory Determination and Final Determination to Not Regulate Perchlorate

In determining whether to regulate a particular contaminant, the EPA must follow the criteria mandated by the 1996 SDWA Amendments. Specifically, in order to issue a MCLG and NPDWR for perchlorate, the EPA must determine that perchlorate "may have an adverse effect on the health of persons," that perchlorate occurs at "a frequency and at levels of public health concern" in public water systems, and that regulation of perchlorate in drinking water systems "presents a meaningful opportunity for health risk reduction for persons served by public water systems." SDWA, section 1412(b)(1)(A). In preparing the 2019 proposal for perchlorate, the EPA updated and improved information on the levels of public health concern and the frequency and levels of perchlorate in public water systems. The following is the EPA's reassessment of the

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regulatory determination criteria applied to the best available health science and occurrence data
for perchlorate.

A. May perchlorate have an adverse effect on the health of persons?

Yes, perchlorate may have adverse health effects. The perchlorate anion is biologically significant specifically with respect to the functioning of the thyroid gland. Perchlorate can interfere with the normal functioning of the thyroid gland by inhibiting the transport of iodide into the thyroid, resulting in a deficiency of iodide in the thyroid. Perchlorate inhibits (or blocks) iodide transport into the thyroid by chemically competing with iodide, which has a similar shape and electric charge. The transfer of iodide from the blood into the thyroid is an essential step in the synthesis of thyroid hormones. Thyroid hormones play an important role in the regulation of metabolic processes throughout the body and are also critical to developing fetuses and infants, especially for brain development. Because the developing fetus depends on an adequate supply of maternal thyroid hormones for its central nervous system development during the first and second trimester of pregnancy, iodide uptake inhibition from perchlorate exposure has been identified as a concern in connection with increasing risk of neurodevelopmental impairment in fetuses of pregnant women with low dietary iodine. Poor iodide uptake and subsequent impairment of the thyroid function in pregnant and lactating women have been linked to delayed development and decreased learning capability in their infants and children (NRC, 2005). Therefore, the EPA continues to find that perchlorate may have an adverse effect on the health of persons.

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B. Is perchlorate known to occur or is there a substantial likelihood that perchlorate will occur in public water systems with a frequency and at levels of public health concern?

The EPA has determined that perchlorate does not occur with a frequency and at levels of public health concern in public water systems. The EPA has made this determination by comparing the best available data on the occurrence of perchlorate in public water systems to potential MCLGs for perchlorate.

In past regulatory determinations, the EPA has identified HRLs as benchmarks against which the EPA compares the concentration of a contaminant found in public water systems to determine if it occurs at levels of public health concern. For the 2011 regulatory determination the EPA identified potential HRLs values ranging from 1 to 47  $\mu$ g/L for 14 different life stages. These HRLs were not final decisions about the level of perchlorate in drinking water that is without adverse effects. For the 2019 proposal, the EPA derived three potential MCLGs for perchlorate of 18, 56, and 90  $\mu$ g/L for the most sensitive life stage using the best available peer reviewed science in accordance with the SDWA. After considering public comment, the EPA used these potential MCLGs as the levels of public health concern in assessing the frequency of occurrence of perchlorate in this regulatory determination. These MCLGs were set at levels to avoid IQ decrements of 1, 2, and 3 points respectively in the most sensitive life stage, the children of hypothyroxinemic women with low iodine intake. The EPA proposed an MCLG of 56  $\mu$ g/L and alternative MCLG values of 18 and 90  $\mu$ g/L.

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The rationale used in deriving the numerical values is presented in greater detail in the EPA's technical support document titled "Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water" (USEPA, 2019b).

The EPA compared these potential MCLG values to the updated perchlorate UCMR 1 occurrence data set. A comprehensive description of the perchlorate occurrence data is presented in Section VI of the 2019 proposal. It is also available in the "Perchlorate Occurrence and Monitoring Report" (USEPA, 2019a).

The occurrence data for perchlorate were collected from 3,865 PWSs between 2001 and 2005 under the UCMR 1. In the 2019 proposal, the EPA modified the UCMR 1 data set in response to concerns raised by stakeholders regarding the data quality and to represent current conditions in California and Massachusetts, which have enacted perchlorate regulations since the UCMR 1 data were collected. Massachusetts promulgated a drinking water standard for perchlorate of 2 µg/L in 2006 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"8DPpSrv3","properties": {"formattedCitation":"(MassDEP, 2006)","plainCitation":"(MassDEP, 2006)","noteIndex":0},"citationItems":[{"id":151,"uris":["http://zotero.org/groups/945096/items/9893MBZH"],"itemData": {"id":151,"type":"personal\_communication","title":"Letter to Public Water Suppliers concerning new perchlorate regulations","URL":"https://www.mass.gov/lists/perchlorate-background-information-and-standards#perchlorate---final-standards-

\*\*\*E.O. 12866 Review - Draft - Do Not Cite, Quote, or Release During Review\*\*\* ","author":[{"literal":"MassDEP"}],"issued":{"dateparts":[["2006"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ], and California promulgated a drinking water standard of 6  $\mu$ g/L in 2007 [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"cfr6HNhg","properties":{"formattedCitation":"(California Department of Public Health, 2007)", "plainCitation": "(California Department of Public Health, 2007)","noteIndex":0},"citationItems":[{"id":150,"uris":["http://zotero.org/groups/945096/items/ RA45NKLQ"],"uri":["http://zotero.org/groups/945096/items/RA45NKLQ"],"itemData":{"id":15 0,"type":"personal communication","title":"State Adoption of a Perchlorate Standard", "URL": "https://www.waterboards.ca.gov/drinking water/certlic/drinkingwater/docum ents/perchlorate/AdoptionMemotoWaterSystems-10-2007.pdf", "author": [{"literal": "California Department of Public Health"}],"issued":{"dateparts":[["2007"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Systems in these states are now required to keep perchlorate levels in drinking water below their state limits. As discussed below, the EPA finds that perchlorate levels in drinking water and sources of drinking water have decreased since the UCMR 1 data collection. The main factors contributing to the decrease in perchlorate levels are the promulgation of drinking water regulations for perchlorate in California and Massachusetts and the ongoing remediation efforts in the state of Nevada to address perchlorate contamination in groundwater adjacent to the lower Colorado River upstream of Lake Mead.

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To update the occurrence data for systems sampled during UCMR 1 from California and Massachusetts, the EPA identified all systems and corresponding entry points which had reported perchlorate detections in UCMR 1. Once the systems and entry points with detections were appropriately identified, the EPA then used a combination of available data from Consumer Confidence Reports (CCRs) and perchlorate compliance monitoring data from California (https://sdwis.waterboards.ca.gov/PDWW/) and Massachusetts (https://www.mass.gov/service-details/public-water-supplier-document-search) to match current compliance monitoring data (where available) to the corresponding water systems and entry points sampled during UCMR 1.

The EPA has determined that the UCMR 1 data with these updates are the best available data collected in accordance with accepted methods regarding the frequency and level of perchlorate nationally. The UCMR 1 data are from a census of the large water systems (serving more than 10,000 people) and a statistically representative sample of small water systems that provides the best available, national assessment of perchlorate occurrence in drinking water.

The EPA used entry point maximum measurements to estimate potential baseline occurrence and exposure at levels that exceed the potential MCLG thresholds. The maximum measurements indicate highest perchlorate levels reported in at least one quarterly sample from surface water systems and at least one semi-annual sample from ground water systems.

Table 1: Perchlorate Occurrence and Exposure (Updated UCMR 1 Data Set)

Threshold Concentration	Entry Points with Detections above		Percent of U.S. Water Systems with Detections	Population Served
(ug/L)	Threshold	above Threshold	•	Served

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18 μg/L	17	15	0.03 %	620,560
56 μg/L	2	2	0.004 %	32,432
90 μg/L	1	1	0.002 %	25,972

Table 1 presents the number and percentage of water systems that reported perchlorate at levels exceeding the three proposed MCLG threshold concentrations. In summary, the updated perchlorate occurrence information suggests that at an MCLG of 18  $\mu$ g/L, there would be 15 systems (0.03% of all water systems in the U.S.) that would exceed the threshold, at an MCLG of 56  $\mu$ g/L, two systems (0.004% of all water systems in the U.S.) would exceed the threshold, and finally one system would exceed the MCLG threshold of 90  $\mu$ g/L. Based on the analysis of drinking water occurrence presented in the 2019 proposal and the data summarized in Table 1 and the range of potential MCLGs, the EPA concludes that perchlorate does not occur with a frequency and at levels of public health concern in public water systems.

While the EPA has made its conclusion that perchlorate does not occur at a frequency and at levels of public health concern in public water systems based on the updated UCMR 1 data, the EPA also sought to find additional information about the perchlorate levels at the 15 water systems that had at least one reported result greater than 18 µg/L in the updated UCMR 1 data. The EPA found that perchlorate levels have been reduced at many of these water systems. Although these water systems were not required to take actions to reduce perchlorate in drinking

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water, many had conducted additional monitoring for perchlorate and found decreased levels or
had taken mitigation efforts to address perchlorate, confirming the EPA's conclusion described
above. The status of each of these systems is described in Table 2 below.

Table 2: Update on Systems with Perchlorate levels above 18  $\mu g/L$  in the UCMR 1

State	System Name	Range of UCMR 1 Results (μg/L)**	Update on Mitigation and Levels of Perchlorate++
Florida	Sebring Water	ND-70	The EPA contacted the Sebring system in January 2020. Operations personnel indicated that no follow-up/updated monitoring data for perchlorate are available.
Florida	Manatee County Utilities Dept	ND-30	Researchers contacted the system to identify the source of perchlorate. System personnel attributed the sole perchlorate detection under UCMR 1 to analytical error. System personnel indicated that three other quarterly samples collected under UCMR 1 as well as other subsequent perchlorate sampling efforts were non-detect. Source: AWWA (2008)
Georgia	Oconee Co Watkinsville	38 (single sample)	Researchers contacted the system and found that a perchlorate contaminated well was removed from service in 2003. The system indicates that perchlorate is no longer detected. Source: Luis et al. (2019)
Louisiana	St. Charles Water District 1 East Bank	ND-24	The EPA was not able to identify updated data on perchlorate levels for this system.
Maryland	City of Aberdeen	ND-19	The system's 2018 Consumer Confidence Report (CCR) indicates that perchlorate was not detected. According to the Maryland Department of Environment, perchlorate was not detected in this system in 2019. In addition, researchers contacted the system and found that there has been no detection of

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State	System Name	Range of UCMR 1 Results (μg/L)**	Update on Mitigation and Levels of Perchlorate++
			perchlorate since treatment was installed in 2009.
			Source: Luis et al. (2019)
Maryland	Chapel Hill - Aberdeen Proving Grounds	ND-20	The EPA contacted the Chapel Hill System in January 2020. Water system personnel indicate that the Chapel Hill WTP was taken off-line and was replaced with a new treatment plant and five new production wells. The new treatment plant started operations on January 27, 2020. System personnel also indicate that monitoring was conducted in November 2019 and perchlorate was not detected in either the source well water or the finished water. In addition, according to the Maryland Department of Environment, perchlorate was not detected in this system in 2019.
Mississippi	Hilldale Water District	ND-20	The EPA contacted the Hilldale System in January 2020. Water system personnel indicated that no follow-up/updated monitoring data for perchlorate are available.
New Mexico	Deming Municipal Water System	15-20	Data from the EPA's SDWIS/FED database indicates that the entry point that reported detections in UCMR 1 (Well #3) is now inactive (i.e., the contaminated source is no longer in use).
Nevada	City of Henderson	6-23	Researchers report that the perchlorate levels described in the system's CCR ranged from non-detect to 9.7 µg/L. Source AWWA (2008).
Ohio	Fairfield City PWS	6-27	The EPA contacted the Fairfield City System in January 2020. Water system personnel indicated that follow-up monitoring was conducted after UCMR 1, between 2002 and 2004. The Ohio EPA provided copies of the follow-up monitoring results which

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State	System Name	Range of UCMR 1 Results (μg/L)**	Update on Mitigation and Levels of Perchlorate <sup>++</sup>
		V S	indicate that results at the entry point ranged from non-detect to 13 µg/L.
Ohio	Hecla Water Association-Plant PWS	ND-32	The EPA contacted the Hecla Water Association System in January 2020. Water system personnel indicated that that no follow-up/updated monitoring data for perchlorate are available.
Oklahoma	Enid	ND-30	The EPA reviewed Oklahoma's monitoring data and did not find any monitoring results reported for perchlorate.
Pennsylvania	Meadville Area Water Authority	ND-33	The EPA contacted the Meadville System in January 2020. Water system personnel indicated that no follow-up/updated monitoring data for perchlorate are available.
Puerto Rico	Utuado Urbano	ND-420	The EPA contacted the Puerto Rico Aqueduct and Sewer Authority (PRASA) in January 2019. PRASA personnel indicated that no updated monitoring data for perchlorate are available. NOTE: The PRASA personnel stated that the Utuado water system was significantly impacted by hurricane Maria and monitoring records from years prior to 2017 were lost.
Texas	City of Levelland	ND-32	Researchers found that a water storage tank was the source of perchlorate contamination, the wells feeding the tank were tested by the state and perchlorate was not detected. The water tank was shut off from service. Source: Luis et al. (2019).

<sup>\*\* -</sup> Values have been rounded. ND describes a sampling event where perchlorate was not detected at or above the UCMR 1 minimum reporting level of 4  $\mu$ g/L. UCMR 1 results collected between 2001 and 2005.

<sup>++ -</sup> To obtain updated data and/or information regarding perchlorate levels, the EPA reviewed Consumer Confidence Reports and other publicly available data, as well as published studies. In addition, the EPA contacted some water systems for information about current perchlorate levels. (USEPA, 2020b)

C. Is there a meaningful opportunity for the reduction of health risks from perchlorate for persons served by public water systems?

The EPA's analysis presented in the 2019 proposal demonstrates that a NPDWR for perchlorate does not present a meaningful opportunity for health risk reduction for persons served by public water systems. As discussed above, the EPA found that perchlorate occurs with very low frequency at levels of public health concern. Based on updated UCMR 1 occurrence information, there were 15 water systems (0.03% of all water systems in the U.S.) that detected perchlorate in drinking water above the lowest proposed alternative MCLG of 18 µg/L and only 1 system had a detection above the proposed alternative MCLG of 90 µg/L. Specifically, Table 1 presents the population served by PWSs that were monitored under UCMR 1 for which the highest reported perchlorate concentration was greater than the identified thresholds. The EPA estimates² that the number of people who may be potentially consuming water containing perchlorate at levels that could exceed the levels of concern for perchlorate could range between 26,000 and 620,000. The small number of water systems with perchlorate levels greater than identified thresholds and the corresponding small population served provides ample support for the EPA's conclusion that the regulation of perchlorate does not present a "meaningful"

 $<sup>^2</sup>$  The values shown in Table 1 are based on the revised UCMR 1 data. The EPA also applied statistical sampling weights to the small systems results to extrapolate to national results. There was one small system included in the statistical sample stratum which had a perchlorate measurement exceeding 18  $\mu g/L$ . Accordingly, the EPA estimates that approximately 41,000 small system customers may be exposed to perchlorate greater than 18  $\mu g/L$ . Page [ PAGE ] of [ NUMPAGES ]

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* opportunity for health risk reduction for persons served by public water systems," within the meaning of SDWA, section 1412(b)(1)(A)(iii).

The EPA also considered the findings of the Health Risk Reduction and Cost Analysis (HRRCA, USEPA 2019c) as additional information supporting withdrawal of the regulatory determination. The HRRCA for perchlorate (which was presented in the 2019 proposal) provides a unique set of economic data indicators that are not available for regulatory determinations because the HRRCA is required for a proposed NPDWR under SDWA Section 1412(b)(3)(C), but is not required to support a regulatory determination. Accordingly, because the EPA initially determined that perchlorate met the criteria for regulation and began the regulatory analysis process, the HRRCA was available with respect to perchlorate, and the Agency considered this comprehensive economic analysis in informing its decision to withdraw the regulatory determination.

Specifically, the HRRCA provides a description of the potential benefits and costs of a drinking water regulation for perchlorate. For all potential regulatory levels considered for perchlorate (18, 56, and 90  $\mu$ g/L) the total costs associated with establishing a regulation were substantially higher than the potential range of benefits. The infrequent occurrence of perchlorate at levels of health concern imposes high monitoring and administrative cost burdens on public water systems and the states, while having little impact on health risk reductions and the associated low estimates of benefits.

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Based on a comparison of costs and benefits estimated at the three potential regulatory

levels, the EPA determined in the 2019 proposal that the benefits of establishing a drinking water

regulation for perchlorate do not justify the potential costs.

A drinking water regulation for perchlorate would impose significant burden on states and water systems, mainly associated with requirements for monitoring but which would result in very few systems having to take action to reduce perchlorate levels. It is of paramount importance that water systems (particularly medium, small and economically distressed systems) focus their limited resources on actions that ensure compliance with existing NPDWRs and maintain their technical, managerial, and financial capacity to improve system operations and the quality of water being provided to their customers rather than spending resources monitoring for contaminants that are unlikely to occur.

D. What is the EPA's final regulatory determination on perchlorate?

Based on the EPA's analysis of the best available public health information, and after careful review and consideration of public comments on the June 2019 proposal, the Agency is withdrawing its 2011 determination and is making a final determination to not regulate perchlorate. Accordingly, the EPA will not issue a NPDWR for perchlorate at this time. While the EPA has found that perchlorate may have an adverse effect on human health, based on the analysis presented in this notice and supporting record, the EPA has determined that perchlorate does not occur in public water systems with a frequency and at levels of public health concern and that regulation of perchlorate does not present a meaningful opportunity to reduce health risks for persons served by public water systems. This conclusion is based on the best available Page [ PAGE ] of [ NUMPAGES ]

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peer reviewed science and data collected in accordance with accepted methods on perchlorate health effects and occurrence.

## IV. Summary of Key Public Comments on Perchlorate

The EPA received approximately 1,500 comments from individuals or organizations on the June 2019 proposal. This section briefly discusses the key technical issues raised by commenters and the EPA's response. Comments are also addressed in the "Comment Response Document for the Final Regulatory Action for Perchlorate" (USEPA, 2020a) available at http://www.regulations.gov (Docket ID No. EPA–HQ–OW–2018–0780).

## A. SDWA Statutory Requirements and the EPA's Authority

The EPA received comments stating the Agency should promulgate an MCLG and MCL for perchlorate and comments stating the Agency should not promulgate a regulation. After considering these comments the EPA has re-evaluated perchlorate in accordance with SDWA, section 1412(b)(1)(A), which requires that the Agency promulgate a NPDWR if (i) the contaminant may have an adverse effect on the health of persons; (ii) the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern; and (iii) in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by public water systems.

The EPA has determined, based upon the best available peer reviewed science and data collected in accordance with accepted methods, that perchlorate does not occur at a

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frequency and at levels of public health concern, and that regulation of perchlorate does not
present a meaningful opportunity for health risk reduction. Because perchlorate does not
meet the statutory criteria for regulation, the EPA lacks the authority to issue a MCLG or
NPDWR for perchlorate, and is therefore withdrawing its 2011 regulatory determination and
issuing this final determination to not regulate perchlorate. For more information regarding
EPA's statutory authority to withdraw its regulatory determination, see Section II.C above.

## B. Health Effects Assessment

## Health Effects/MCLG Derivation

The EPA received comments indicating that the Agency should utilize different approaches to derive the MCLG for perchlorate including approaches that some states used to develop their perchlorate advisory levels or drinking water standards. The EPA considered a number of alternative approaches to develop the MCLG for perchlorate and in accordance with SDWA, section 1412(e), the Agency sought recommendations from the Science Advisory Board. The EPA derived the proposed MCLG for perchlorate based on the approach recommended by the Science Advisory Board (SAB) (SAB for the U.S. EPA, 2013). The SAB recommended that "the EPA derive a perchlorate MCLG that addresses sensitive life stages through physiologically-based pharmacokinetic/pharmacodynamic modeling based upon its mode of action rather than the default MCLG approach using the RfD and specific chemical exposure parameters." The EPA has implemented these recommendations and has obtained two independent peer reviews of the analysis. These peer

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reviewers stated that: "Overall, the panel agreed that the EPA and its collaborators have
prepared a highly innovative state-of-the-science set of quantitative tools to evaluate
neurodevelopmental effects that could arise from drinking water exposure to perchlorate.
While there is always room for improvement of the models, with limited additional work to
address the committee's comments below, the current models are fit-for-purpose to
determine an MCLG" (External Peer Reviewers for USEPA, 2018, p. 2).

The EPA received comments indicating the most sensitive life stages were not selected and/or considered in the Agency's approach. The EPA disagrees. Gestational exposure to perchlorate during neurodevelopment is the most sensitive time period. The NRC concluded that the population most sensitive to perchlorate exposure are "the fetuses of pregnant women who might have hypothyroidism or iodide deficiency" [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"a1mn5hjprkt","properties": {"formattedCitation":"(National Research Council (NRC), 2005b)","plainCitation":"(National Research Council (NRC), 2005b)","noteIndex":0},"citationItems": [{"id":350,"uris": ["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData" : {"id":350,"type":"book","title":"Health Implications of Perchlorate
Ingestion","publisher":"National Academies Press","publisher-place":"Washington, DC","event-place":"Washington, DC","author": [{"literal":"National Research Council (NRC)"}],"issued": {"date-parts": [["2005"]]}}}],"schema":"https://github.com/citation-style-

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\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* language/schema/raw/master/csl-citation.json"} ]. In addition, there is clear evidence that disrupted maternal thyroid hormone levels during gestation can impact neurodevelopment later in life (Alexander et al., 2017; Costeira et al., 2011; Endendijk et al., 2017; Ghassabian, Bongers-Schokking, Henrichs, Jaddoe, & Visser, 2011; Glinoer & Delange, 2000; Glinoer & Rovet, 2009; Gyllenberg et al., 2016; Henrichs et al., 2010; Korevaar et al., 2016; Morreale de Escobar, Obregón, & Escobar del Rey, 2004; Noten et al., 2015; Pop et al., 2003, 1999; SAB for the U.S. EPA, 2013; Thompson et al., 2018; van Mil et al., 2012; Wang et al., 2016; Zoeller & Rovet, 2004; Zoeller et al., 2007). The available data demonstrate that the fetus of the first trimester pregnant mother, when compared to other life-stages, experiences the greatest impact from the same dose of perchlorate, which is described in detail in Section 6 of the document "Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water" (USEPA, 2019a). Some commenters suggested that the bottle-fed infant is a more sensitive life-stage. The EPA disagrees as described in the January 2017 Peer Review Report on the original Biologically Based Dose Response (BBDR) model, the bottle-fed infant's thyroid hormone levels were not impacted by doses of perchlorate up to 20 µg/day (External Peer Reviewers for USEPA, 2017). This lack of any impact is due primarily to the iodine in the formula, which offsets the impact of perchlorate on the thyroid.

The EPA received comments advocating for the use of the population-based approach evaluating the shift in the proportion of a population that would fall below a hypothyroxinemic cut point under a perchlorate exposure scenario. The EPA chose to

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develop the MCLG using dose-response functions from the epidemiological literature to estimate neurodevelopmental impacts in the offspring of pregnant women exposed to perchlorate. The EPA selected this proposed approach because it is consistent with the SDWA's definition of a MCLG to avoid adverse health effects and because it is most consistent with the SAB recommendations. In addition, the fact that thyroid hormone levels vary by reference population and that there is not a defined value representing hypothyroxinemia makes the population-based approach less desirable than the approach selected (USEPA, 2018).

## End Point Selection/Basis

The EPA received comments regarding the magnitude of an IQ change which should be used in deriving the MCLG. The EPA's proposed MCLG was based upon avoiding a 2% change in IQ in the most sensitive life stage and the EPA also requested comment on alternative options for the MCLG that would respectively avoid 1% or 3% change in IQ in the most sensitive life stage. Many comments stated that the EPA should at most consider a 1% IQ change. However, several commenters stated a 3% change is too small to have a meaningful impact and suggested the EPA consider a higher IQ percent change.

The EPA uses a variety of science policy approaches to select points of departure for developing regulatory values. For instance, in noncancer risk assessment the EPA often uses a percentage change in value. When assessing toxicological data, a 10 % extra risk (for discrete data), or a 1 standard deviation (i.e., 15 IQ points) change from the mean (for

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continuous data) is often used (USEPA, 2012). A smaller response to inform a POD has been applied when using epidemiological literature because there is an inherently more direct relationship between the study results and the exposure context and health endpoint.

Given the difficulty in identifying a response below which no adverse impact occurs when considering a continuous outcome in the human population, the EPA looked to its Benchmark Dose Guidance (2012) for insight regarding a starting point. Specifically, "[a] BMR of 1% has typically been used for quantal human data from epidemiology studies" (p. 21, USEPA, 2012). For the specific context of setting an MCLG for perchlorate, the EPA evaluated the level of perchlorate in water associated with a 1% decrease, a 2% decrease, and a 3 percent decrease in the mean population IQ (i.e., 1, 2 and 3 IQ points).

In evaluating the frequency and level of occurrence of perchlorate in drinking water the EPA has found that perchlorate does not occur with frequency even at the lowest alternative MCLG of 18  $\mu$ g/L which is based upon avoiding a 1% change in IQ in the most sensitive life stage.

The EPA received comments that the proposed MCLG did not incorporate an adequate margin of safety to comply with the SDWA. The EPA disagrees that it failed to use an adequate margin of safety. The EPA's assessment focused upon the most sensitive subset of the population, specifically offspring whose mothers had low (75 µg/day) iodine intake and were hypothyroxinemic (fT4 in the lowest 10th percentile of the population). In addition, to account for uncertainties and to ensure the most sensitive subset of the population is

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protected with an adequate margin of safety, a 3-fold uncertainty factor was applied to the
proposed MCLG calculation (USEPA, 2019a). More discussion on the uncertainty factor is
presented in the section "Consideration of Uncertainties."

The EPA received some comments stating that the selection of the study for informing the relationship between maternal hormone levels (fT4) and IQ was inadequately described. Other comments supported the EPA's study selection. The EPA concludes that selection of the Korevaar et al. (2016) study is appropriate because that study provides the most robust data available with a clear measure of neurodevelopment that can be expressed as a function of changing maternal fT4 exposure, which is necessary to the development of the model.

#### BBDR and PBPK Models

The EPA received comments indicating the BBDR model was not transparent, scientifically valid, or based on robust data. The EPA disagrees. The model represents the best available peer reviewed science and uses the best available data to inform a MCLG for perchlorate. The EPA disagrees that there is a significant lack of transparency with respect to the assumptions related to the BBDR model. Appendix A of the EPA's Proposed MCLG Approaches report outlines the justification for all assumptions used in the development of the BBDR model (USEPA, 2019a). The EPA also disagrees with the assertion the BBDR model is far too uncertain to be relied upon as the basis for the derivation of the RfD. The EPA has used the best available science to calibrate the pharmacokinetic aspects of the

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BBDR model. The development of the BBDR model was in response to SAB recommendations and a model was deemed to be a more refined approach to estimating a dose-response relationship between perchlorate exposure and maternal fT4 than anything that was available in the current scientific literature. The EPA disputes the claim that there are issues with the scientific validity of the BBDR model as the Agency conducted a peer review of the approach proposed and the reviewers stated the approach was "fit for purpose" to inform a MCLG for perchlorate (External Peer Reviewers for U.S. EPA, 2018, p. 2).

#### Consideration of Uncertainties

The EPA received comments on the Agency's use of Uncertainty Factors (UFs); with most commenters suggesting that the EPA should consider a higher UF. The EPA thoroughly considered the application of UFs when deriving the RfDs and followed guidance presented in "A review of the reference dose and reference concentration processes" (USEPA, 2002). The EPA concluded that the UFs are adequately justified and subsequently no changes have been made. Justification for each of the UFs can be found in Section 11 of the Agency's MCLG Derivation report (USEPA, 2019a).

The EPA selected a UF of 3 for inter-individual variability because the Agency specifically modeled groups within the population that are identified as likely to be at greater risk of the adverse effects from perchlorate in drinking water (i.e., the fetus of the iodide deficient pregnant mother). The EPA selected model parameters to account for the most sensitive individuals in that group (i.e., muted TSH feedback, low fT4 values, low-iodine

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* intake). As discussed in the MCLG Derivation report, the EPA has attempted to select the most appropriate inputs to protect the most sensitive population with an adequate margin of safety (USEPA, 2019a). The EPA has determined that the selection of a UF of 3 for interindividual variability is justified. As described in the MCLG Derivation report, because the output from the BBDR model is specific to the sensitive population the EPA concluded that the UF of 3 is appropriate. In regards to variation in sensitivity among the members of the human population (i.e., inter-individual variability), section 4.4.5.3 of the EPA guidance "A review of the reference dose and reference concentration process" (USEPA, 2002) document states, "In general, the Technical Panel reaffirms the importance of this UF, recommending that reduction of the intraspecies UF from a default of 10 be considered only if data are sufficiently representative of the exposure/dose-response data for the most susceptible subpopulation(s). Similar to the interspecies UF, the intraspecies UF can be considered to consist of both a toxicokinetic and toxicodynamic portion (i.e. 10\^0.5 each)" (USEPA, 2002). Given that the BBDR model significantly accounts for differences within the human population, the full UF of 10 is not warranted.

One commenter suggested using a UF greater than 1 to account for the extrapolation of the lowest-observed adverse effect level (LOAEL) to the no-observed-adverse-effect-level (NOAEL). LOAELs and NOAELs were not identified or used by the EPA in its assessment because the Agency employed a sophisticated BBDR modeling approach, which was coupled with extrapolation to changes in IQ using linear regression, to determine a POD that would

\*\*\*E.O. 12866 Review – Draft – Do Not Cite, Quote, or Release During Review\*\*\* not be expected to represent an adverse effect. Therefore, a UF of 1 is appropriate. Other commenters suggested incorporating UFs for database deficiencies. Based on the findings of the NRC report, the EPA has previously concluded that this UF was not needed for deficiencies in the perchlorate database (NRC, 2005; USEPA, 2005a). The EPA determined that a UF of 1 to account for database deficiencies is still appropriate given that the state of the perchlorate database has only increased since 2005.

## C. Occurrence Analysis

The EPA received comments suggesting that the revised UCMR 1 data did not provide an adequate estimate of the perchlorate occurrence in drinking water systems. Some commenters indicated that the age of the collected data rendered the occurrence analysis obsolete and overestimated, since it no longer captures current lower contamination conditions that have been achieved due to mitigation measures taken in the Colorado River Basin. Other commenters criticized the EPA for replacing UCMR 1 data with compliance data for the States of California and Massachusetts.

The EPA recognizes that changes in perchlorate levels (increasing or decreasing) may have occurred in water systems since the UCMR 1 samples were collected between 2001 to 2005. The EPA updated the UCMR 1 data set to improve its accuracy in representing the current conditions for states that have enacted perchlorate regulations since the UCMR 1 monitoring was conducted. As outlined in the June 26, 2019 proposal, the EPA updated occurrence data for California and Massachusetts with current compliance data as reported

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by the states. Systems from these two states that were sampled during the UCMR 1 and that
had reported perchlorate detections were updated with more recently measured values taken
from current compliance monitoring data from Consumer Confidence Reports and state-level
perchlorate compliance monitoring data to match corresponding water systems and entry
points.

The EPA has determined that the updated UCMR 1 data are the best available data collected in accordance with accepted methods on the frequency and level of perchlorate occurrence in drinking water on a national scale.

#### V. Conclusion

With this withdrawal of the 2011 perchlorate regulatory determination and final determination to not regulate perchlorate, the EPA announces that there will be no NPDWR for perchlorate at this time. The EPA could consider re-listing perchlorate on the CCL and could proceed to regulation in the future if the occurrence or health risk information changes. As with other unregulated contaminants, the EPA will consider addressing limited instances of elevated levels of perchlorate by working with the affected system and state, as appropriate.

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Zoeller, R. T., Tan, S. W., & Tyl, R. W. (2007). General background on the hypothalamic-pituitary-thyroid (HPT) axis. Critical reviews in toxicology, 37(1-2), 11-53.

## List of Subjects in 40 CFR Parts 141 and 142

Administrative practice and procedure, Chemicals, Indians-lands, Intergovernmental relations, Radiation protection, Reporting and recordkeeping requirements, Water supply.

Andrew Wheeler,

Administrator.

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***E.O. 12866 Review	w – Draft – Do Not Cite, Quo	ote, or Release During Re	eview***
	Page [ PAGE ] of [ NUM]	PAGES ]	

### **MEMORANDUM**

**SUBJECT:** Notice of Final Action on Perchlorate

(Tier 1 Action; SAN 5555; RIN 2040-AF28) - ACTION MEMORANDUM

**FROM:** David P. Ross

Assistant Administrator (4101M)

**THRU:** Office of Policy (1803A)

Office of Executive Secretariat (1105A)

**TO:** Andrew R. Wheeler

Administrator (1101A)

#### **PURPOSE**

Attached for your signature is the action titled "Notice of Final Action on Perchlorate."

On February 11, 2011, the U.S. Environmental Protection Agency (EPA or Agency) published a determination to regulate perchlorate in drinking water (76 FR 7762). On June 26, 2019 (84 FR 30524), the EPA published the proposed National Primary Drinking Water Regulations (NPDWR) for Perchlorate and requested public comments on multiple alternative actions, including withdrawing the Agency's 2011 determination to regulate perchlorate. The EPA received approximately 1,500 comments on the proposed rule.

In this notice, the EPA is withdrawing the 2011 Regulatory Determination and is making a final determination not to regulate perchlorate based on the Agency's consideration of public comments and the best available information.

#### **DEADLINE/TIMELINE**

Section 1412(b)(1)(A) of the Safe Drinking Water Act (SDWA) requires the EPA to issue a proposed NPDWR within 24 months of the final regulatory determination and a final NPDWR within 18 months after the proposal. However, when the EPA consulted with the Science Advisory Board (SAB) regarding a planned methodology for deriving the maximum contaminant level goal (MCLG) for perchlorate, the Agency received recommendations to develop a physiologically based pharmacokinetic model (i.e., a biologically based dose-response model (BBDR)) to predict the effects of perchlorate exposure on thyroid function in pregnant women and their children, instead. The EPA collaborated with Food and Drug Administration scientists to perform the modeling recommended by the SAB and completed the analysis and associated peer reviews in March 2018. This delayed the EPA in proposing a NPDWR within 24 months.

In February 2016, the Natural Resources Defense Council (NRDC) filed a lawsuit for failure of the EPA to perform its mandatory duties of proposing and finalizing a regulation for perchlorate in accordance with timelines provided in the SDWA. On October 18, 2016, the U.S. District Court for the Southern District of New York entered a Consent Decree, requiring the EPA to sign for publication a proposal for a MCLG and NPDWR for perchlorate in drinking water no later than October 31, 2018, and to sign for publication a final MCLG and NPDWR for perchlorate in drinking water no later than December 19, 2019. The Court later extended the deadline for the EPA to propose a MCLG and NPDWR for perchlorate in drinking water to May 28, 2019, and extended the date for signature of a final MCLG and NPDWR no later than June 19, 2020.

In compliance with the deadline established in the Consent Decree, on May 23, 2019, the Administrator signed a proposed rulemaking notice seeking public comment on a range of options regarding the regulation of perchlorate in public drinking water systems. The EPA published the proposed rule in the *Federal Register* on June 26, 2019. The public comment period for the proposal ended on August 26, 2019, and the EPA received approximately 1,500 comments.

#### DESCRIPTION OF THE ACTION

Perchlorate is an inorganic anion that occurs naturally. It is also manufactured as an oxidizer for rockets, missiles, and fireworks and can be an impurity in hypochlorite disinfectants. The public may be exposed to perchlorate through food and drinking water. At certain levels, perchlorate can prevent the thyroid gland from getting enough iodine, which can affect thyroid hormone production. For pregnant women with low iodine levels, sufficient changes in thyroid hormone levels may cause changes in the child's brain development. For infants, changes to thyroid hormone function can also impact brain development.

The SDWA sets forth three criteria that must be met for the EPA to issue a MCLG and promulgate a NPDWR. Specifically, the EPA must determine that (1) "the contaminant may have an adverse effect on the health of persons;" (2) "the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern;" and (3) "in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by public water systems" (SDWA 1412(b)(1)(A)).

In the attached notice, the EPA concludes that, based on new data and the Agency's analysis since the issuance of the 2011 Regulatory Determination, perchlorate does not in fact meet the statutorily prescribed criteria for regulation. The new data and analysis indicate that perchlorate does not occur in public water systems with a frequency and at levels of public health concern. Specifically, the new peer-reviewed health effects analysis resulted in the health based proposed MCLG and proposed alternative MCLGs for perchlorate that are higher concentrations in drinking water (18 – 90  $\mu$ g/L) than the concentrations that the EPA considered to be levels of public health concern in the Agency's analysis for the determination to regulate in 2011 (1 – 47  $\mu$ g/L). In addition, the updated occurrence analysis shows that the frequency of occurrence of perchlorate in public water systems at levels exceeding any of the alternative proposed MCLGs (0.38% – 0.02%) is significantly lower than the frequency considered in the

[ PAGE \\* MERGEFORMAT ]

EPA's analysis for the 2011 Regulatory Determination (4% - 0.39%). Based on this information, the EPA is announcing the Agency's conclusion that perchlorate does not occur in public water systems "with a frequency...of public health concern" and, therefore, regulation of perchlorate does not present a "meaningful opportunity for health risk reduction for persons served by public water systems" as required for regulation under the SDWA. Accordingly, perchlorate no longer meets the statutory criteria for regulation because the EPA does not have the authority to issue a MCLG or promulgate a NPDWR for perchlorate.

Therefore, the EPA is not issuing a final MCLG or NPDWR for perchlorate. The EPA will consider addressing limited instances of elevated levels of perchlorate by working with the affected system and state, as appropriate.

#### STAKEHOLDER INVOLVEMENT AND ANTICIPATED RESPONSE

The EPA considered the approximately 1,500 comments that were submitted on the proposed regulation. The EPA also consulted with the National Drinking Water Advisory Committee regarding the proposed regulation. The EPA expects a variety of reactions and responses from stakeholders. The NRDC will likely sue the EPA for failure to comply with the Consent Decree and will likely challenge the Agency's authority to withdraw a Regulatory Determination. Officials from the States of California and Massachusetts, public health groups and environmental groups will likely state that a low perchlorate maximum contaminant level is needed to protect children's health. Industry groups, including the American Water Works Association, the Perchlorate Study Group, the American Chemistry Council, and the U.S. Chamber of Commerce will support the decision not to regulate perchlorate in drinking water. These groups will agree with the EPA's determinations that perchlorate does not occur frequently at levels of public health concern and there is not a meaningful opportunity for health risk reduction for persons served by public water systems.

### INTERNAL DEVELOPMENT AND REVIEW PROCESS

The attached notice reflects the direction provided by the Administrator in the Options Selection and follow-up meetings held on January 9 and March 18, 2020. The Office of Water (OW) convened a Final Agency Review meeting for this action on May 7, 2020. The following offices concurred without comment: the Office of Research and Development, the Office of Land and Emergency Management, the Office of Air and Radiation, and the Office of Chemical Safety and Pollution Prevention. The following offices concurred with comment: the Office of General Counsel (OGC), the Office of Policy (OP), and the Office of Children's Health Protection (OCHP). OW has incorporated revisions identified in the comments from the OGC. OW has also incorporated most of the suggested revisions identified by OP, the key exception being that we are not incorporating OP's recommendation to not list the cost benefit analysis as a factor in the decision to withdraw the regulatory determination. OW has worked with OGC to incorporate language that clarifies that this does not set a precedent for future regulatory determinations. OW is not incorporating the majority of recommendations made by OCHP, which address the health effects and occurrence analysis and are issues we have evaluated previously, including in response to OCHP's input on the proposal and in response to public comments.

### **INTERAGENCY REVIEW**

[ PAGE \\* MERGEFORMAT ]

The Office of Management and Budget initiated review of the *Federal Register* notice: "Notice of Final Action on Perchlorate" on [date placeholder].

#### PEER REVIEW

For the proposed rulemaking, OW followed the EPA's Peer Review Handbook and Agency policy titled "Conflicts of Interest Review Process for Contractor-Managed Peer Reviews of EPA HISA and ISI Documents" when conducting the peer review of models used to derive the proposed MCLGs for perchlorate. The EPA convened an independent peer review panel to evaluate the BBDR models in 2017 and a second, expert peer review panel in 2018 to evaluate the update of the BBDR model and approaches to link the BBDR model output to neurodevelopment endpoints in epidemiology studies to derive an MCLG. The EPA also sought input from the SAB, as required by the SDWA, prior to developing the proposed MCLGs.

### RECOMMENDATION

I recommend that you sign the attached *Federal Register* notice titled "Notice of Final Action on Perchlorate."

Attachments (2)



## UNITED STATES ENVIRONMENTAL PROTECTION AGENCY WASHINGTON, D.C. 20460

OFFICE OF WATER

### **MEMORANDUM**

**SUBJECT:** Transmittal of the Notice of Final Action on Perchlorate (Tier 1 Action; SAN 5555; RIN

2040-AF28) to the Office of Management and Budget for Executive Order 12866 Review

**FROM:** Charlotte Bertrand,

Deputy Assistant Administrator

**TO:** Brittany Bolen, Associate Administrator

Office of Policy

Attached for the Office of Policy's review and transmittal to the Office of Management and Budget (OMB) for Executive Order 12866 interagency review is a *Federal Register* notice titled: "Notice of Final Action on Perchlorate."

On February 11, 2011, the U.S. Environmental Protection Agency (EPA or Agency) published a determination to regulate perchlorate in drinking water (76 FR 7762). On June 26, 2019 (84 FR 30524), the EPA published the proposed National Primary Drinking Water Regulations for Perchlorate and requested public comments on multiple alternative actions, including withdrawing the 2011 determination to regulate perchlorate. The EPA received approximately 1,500 comments on the proposed rule. In the attached notice, the EPA is withdrawing the 2011 Regulatory Determination and is making a final determination not to regulate perchlorate based on the Agency's consideration of public comments and the best available science.

I request that the Office of Policy seek an expedited interagency review in accordance with Executive Order 12866 to meet a consent decree deadline for signature of the final action by June 19, 2020.

If you have any questions, please contact me or have your staff contact Samuel Hernandez at 202-564-1735.

Attachments

#### Message

From: Mclain, Jennifer [Mclain.Jennifer@epa.gov]

**Sent**: 12/2/2019 10:40:17 PM

To: Ross, David P [ross.davidp@epa.gov]

CC: Forsgren, Lee [Forsgren.Lee@epa.gov]; Bertrand, Charlotte [Bertrand.Charlotte@epa.gov]; Wildeman, Anna

[wildeman.anna@epa.gov]; Best-Wong, Benita [Best-Wong.Benita@epa.gov]; Aguirre, Janita

[Aguirre.Janita@epa.gov]; Tiago, Joseph [Tiago.Joseph@epa.gov]; Guilaran, Yu-Ting [Guilaran.Yu-Ting@epa.gov];

Burneson, Eric [Burneson.Eric@epa.gov]; Nagle, Deborah [Nagle.Deborah@epa.gov]

**Subject**: Perchlorate - Administrator's Option Selection Briefing **Attachments**: Option Selection for Perchlorate 12-2-19 v2.docx

#### Dave

As we discussed at the 11/14 OGWDW Biweekly meeting, we developed the attached Option Selection briefing document for the Administrator's consideration. As you recommended, tomorrow we are briefing OLEM on the comments we received on the proposed regulation and the options for a final action. We plan to discuss the potential impacts to their program under different options (these impacts are included in the briefing document). The meeting invitation is on Peter Wright's calendar although we are not certain if he will attend. We will follow up with you regarding next steps with the Administrator after our meeting with OLEM.

Jennifer

#### Message

From: McLain, Jennifer L. [McLain.Jennifer@epa.gov]

**Sent**: 5/19/2020 1:17:18 PM

**To**: Bertrand, Charlotte [Bertrand.Charlotte@epa.gov]

CC: Guilaran, Yu-Ting [Guilaran.Yu-Ting@epa.gov]; Braschayko, Kelley [braschayko.kelley@epa.gov]; Tiago, Joseph

[Tiago.Joseph@epa.gov]; Aguirre, Janita [Aguirre.Janita@epa.gov]

**Subject**: FW: Notice of Final Action on Perchlorate

Attachments: Perchlorate Action Memo 5-18-20.cb.srmd.docx; Draft Perchlorate Final Action FRN 5-18-20 v1 Redline.cb.srmd.docx

Charlotte – the attached provide edits and responses to your questions. We will start the process to formally send to OW for submittal to OP.

Thanks Jennifer

From: Bertrand, Charlotte <Bertrand.Charlotte@epa.gov>

**Sent:** Monday, May 18, 2020 5:29 PM

To: McLain, Jennifer L. < McLain. Jennifer@epa.gov>

<Tiago.Joseph@epa.gov>; Aguirre, Janita <Aguirre.Janita@epa.gov>

Subject: RE: Notice of Final Action on Perchlorate

Thanks – couple of bubble box questions and then I had one redline edit I added to the Notice.

From: McLain, Jennifer L. < McLain. Jennifer@epa.gov>

Sent: Monday, May 18, 2020 3:01 PM

To: Bertrand, Charlotte <Bertrand.Charlotte@epa.gov>

Cc: Guilaran, Yu-Ting <a href="mailto:Guilaran.Yu-Ting@epa.gov">Guilaran.Yu-Ting@epa.gov</a>; Braschayko, Kelley <a href="mailto:braschayko.kelley@epa.gov">braschayko.kelley@epa.gov</a>; Tiago, Joseph

<Tiago.Joseph@epa.gov>; Aguirre, Janita <Aguirre.Janita@epa.gov>

Subject: FW: Notice of Final Action on Perchlorate

Charlotte – as agreed, I'm sending you the draft final perchlorate FRN for review. The redline includes the changes made since the FAR. I'm also including the draft Action Memo. Please let us know if your preference is to have these submitted to OW through CMS now or after you have reviewed. Let me know if you want to talk.

#### Jennifer

From: Burneson, Eric < Burneson. Eric@epa.gov>

Sent: Monday, May 18, 2020 2:48 PM

To: McLain, Jennifer L. < McLain, Jennifer@epa.gov>

Cc: Guilaran, Yu-Ting <a href="mailto:Suilaran.Yu-Ting@epa.gov">Guilaran, Yu-Ting@epa.gov</a>; Christ, Lisa

<<u>Christ.Lisa@epa.gov</u>>; Hernandez-Quinones, Samuel <<u>Hernandez.Samuel@epa.gov</u>>

Subject: FW: Notice of Final Action on Perchlorate

#### Jennifer:

Attached for transmission to OW are revised versions of the FRN for the Perchlorate Final Action. There is both a clean and track changes version that includes edits made since initiating FAR (including the edits you asked for on Saturday and adding 3 more SAB recommendations to page 14 that were in the proposal but were not included in the draft we provided you on Friday). Also please find clean version of the transmittal memo from you to Dave Ross and the Action memo incorporating your edits.

Please note that there is also a redline version of the Action Memo for you to see the responses to your comments on the document. I do not recommend transmitting that memo to OW.

Eric Burneson, P.E.
Director of Standards and Risk Management
Office of Ground Water and Drinking Water
U.S. Environmental Protection Agency
202 564 5250

From: Hernandez-Quinones, Samuel < Hernandez. Samuel@epa.gov>

Sent: Monday, May 18, 2020 2:18 PM

**To:** Burneson, Eric < <u>Burneson.Eric@epa.gov</u>> **Cc:** Christ, Lisa < Christ.Lisa@epa.gov>

Subject: RE: Notice of Final Action on Perchlorate

Hi Eric,

Attached are the revised Redline and Clean versions of the Perchlorate FR Notice. Once we are ready for OP's submittal to OMB let me know and I will provide a version that adheres to OP's file name formatting guidelines.

Thanks Sam

Samuel Hernández Quiñones, P.E. Environmental Engineer Office of Water Environmental Protection Agency 1200 Pennsylvania Ave. NW Washington, DC 20460 202-564-1735

"USEPA Protecting Human Health and the Environment"

From: Burneson, Eric < Burneson. Eric@epa.gov>

**Sent:** Monday, May 18, 2020 1:14 PM

To: Hernandez-Quinones, Samuel < Hernandez. Samuel@epa.gov>

Cc: Christ, Lisa < Christ.Lisa@epa.gov>

Subject: RE: Notice of Final Action on Perchlorate

#### Sam

- 1. Change the title please. This was requested by OGC at Sr. Leadership levels.
- 2. Provide the same level of detail on the SAB recommendations as was included in the proposal.
- 3. I don't think the HRRCA text is necessary and do not want to add it at this stage since there are OGC edits that already make this clear.

Eric

From: Hernandez-Quinones, Samuel < Hernandez. Samuel@epa.gov>

Sent: Monday, May 18, 2020 12:41 PM
To: Burneson, Eric <Burneson. Eric @epa.gov>

Cc: Christ, Lisa < Christ.Lisa@epa.gov>

Subject: RE: Notice of Final Action on Perchlorate

Hi Eric,

Here is a revised Redline of the document (from FAR). We had a few questions/issues for your consideration about the attached file. Specifically:,

- 1- Page #1, Notice Tittle: We did not accept the edits to the notice tittle. Because, the tittle of the notice was specifically crafted by OGC to capture the multiple actions EPA is taking. Suggest consulting with OGC before modifying this tittle.
- 2- Page #14, SAB Recommendations: SAB provided 4 main recommendations in 2013 but we only listed the first recommendation. Please advise if we should list all 4 recommendations here or not.
- 3- Page #26, Missing HRRCA Text: This language was offered by TAB in its 5-13-20 version of the draft FRN, but it did not show up in the version provided by OGWDW with Eric's & Jennifer's comments. We have inserted the language here for the reviewer's consideration. Please advise if we should keep it.

Once you provide your feedback, I will modify the redline version and also provide a Clean copy for transmittal.

Thanks Sam

Samuel Hernández Quiñones, P.E. Environmental Engineer Office of Water Environmental Protection Agency 1200 Pennsylvania Ave. NW Washington, DC 20460 202-564-1735

"USEPA Protecting Human Health and the Environment"

From: Burneson, Eric < Burneson. Eric@epa.gov>

Sent: Monday, May 18, 2020 8:42 AM

To: Christ, Lisa < <a href="mailto:Christ, Lisa@epa.gov">Christ, Lisa@epa.gov</a>; Hernandez-Quinones, Samuel <a href="mailto:Hernandez.Samuel@epa.gov">Hernandez.Samuel@epa.gov</a>>

Cc: McLain, Jennifer L. < <a href="McLain.Jennifer@epa.gov">McLain.Jennifer@epa.gov</a>; Tiago, Joseph < <a href="Tiago.Joseph@epa.gov">Tiago.Joseph@epa.gov</a>; Guilaran, Yu-Ting

<Guilaran.Yu-Ting@epa.gov>

Subject: FW: Notice of Final Action on Perchlorate

#### Lisa and Sam

Attached are Jennifer's comments and edits on the draft FRN. I have responded to her questions in the attached and made some additional edits. Can you please get a revised clean version and another redline version that compares this document and the version that was distributed to FAR?

Thanks for your work on this.

Eric

From: McLain, Jennifer L. < McLain, Jennifer@epa.gov>

**Sent:** Saturday, May 16, 2020 11:39 AM **To:** Burneson, Eric < <u>Burneson.Eric@epa.gov</u>> **Subject:** RE: Notice of Final Action on Perchlorate

Looks very good. See p. 6 for my only concern w/the revisions.

From: Burneson, Eric < Burneson, Eric@epa.gov>

**Sent:** Friday, May 15, 2020 5:03 PM

To: McLain, Jennifer L. < McLain. Jennifer@epa.gov>

Cc: Christ, Lisa < <a href="Christ.Lisa@epa.gov">Christ, Lisa@epa.gov</a>; Tiago, Joseph

<Tiago.Joseph@epa.gov>; Guilaran, Yu-Ting <Guilaran.Yu-Ting@epa.gov>

Subject: Notice of Final Action on Perchlorate

#### Jennifer

Attached for your approval and transmittal to the Office of Water for their approval and transmittal to the Office Policy for initiation of interagency review is a *Federal Register* notice titled: "Notice of Final Action on Perchlorate." Also attached for your review are a draft transmittal memo from you to the Assistant Administrator of Water, a draft Action Memorandum and a track changes version of the FR notice that denotes the changes made as a result of Final Agency Review.

On February 11, 2011, the EPA published a determination to regulate perchlorate in drinking water (76 FR 7762). On June 26, 2019 (84 FR 30524), the EPA published the proposed National Primary Drinking Water Regulation for Perchlorate and requested public comments on multiple alternative actions, including withdrawing the 2011 determination to regulate perchlorate. The EPA received approximately 1,500 comments on the proposed rule. In the attached notice, the EPA is withdrawing the 2011 Regulatory Determination and is making a final determination not to regulate perchlorate based on the Agency's consideration of public comments and the best available information.

I recommend that you approve and transmit the attached notice to the Office of Water for their review, approval and transmission to the Office of Policy to initiate interagency review in accordance with Executive Order 12866. If you need additional information or have questions pertaining to any aspect of this notice, please call me or have your staff contact Samuel Hernandez at 202-564-1735.

Eric Burneson, P.E.
Director of Standards and Risk Management
Office of Ground Water and Drinking Water
U.S. Environmental Protection Agency
202 564 5250

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6560-50-P

ENVIRONMENTAL PROTECTION AGENCY

40 CFR Parts 141 and 142

[EPA-HQ-OW-2018-0780, EPA-HQ-OW-2008-0692, EPA-HQ-OW-2009-0297; FRL-

XXXX-XX-OW]

RIN 2040-AF28

Drinking Water: Notice of Withdrawal of the 2011 Perchlorate Regulatory Determination and Publication of the Final <u>Action</u> Regulatory Determination on Perchlorate

AGENCY: Environmental Protection Agency (EPA).

ACTION: Withdrawal of Regulatory Determination and Final Across Regulatory

Determination.

SUMMARY: The Environmental Protection Agency (EPA) is announcing its withdrawal of the 2011 determination to regulate perchlorate in accordance with the Safe Drinking Water Act (SDWA). On February 11, 2011 (76 FR 7762), the Agency EPA published a Federal Register notice in which the EPA Agency determined that perchlorate met the SDWA's criteria for regulating a contaminant. On June 26, 2019 (84 FR 30524), the EPA published a proposed national primary drinking water regulation (NPDWR) for perchlorate and requested public comments on multiple alternative regulatory actions, including the alternative of withdrawing the 2011 regulatory determination for perchlorate. The Agency EPA received approximately 1,500 comments on the proposed rule. The EPA has considered these public comments and based on

Page [ PAGE ] of [ NUMPAGES ]

Commented [BC1]: Will this show up publicly in ROCIS?

Commented [BEZR1]: This information is used to populate the Stage field on the RegInfo gov site at least that is how it is used for Lead Free. I have edited this information to avoid disclosing the outcome but its not clear if this is going to work for the system the best available information the Agency is withdrawing the 2011 regulatory determination and is making a final determination to not to regulate perchlorate. The Agency The has determined that perchlorate does not occur with a frequency and at levels of public health concern, and that regulation of perchlorate does not present a meaningful opportunity for health risk reduction for

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persons served by public water systems.

**DATES**: For purposes of judicial review, the regulatory determination in this document is issued as of *[insert date of publication in the Federal Register]*.

**FOR FURTHER INFORMATION CONTACT**: Samuel Hernandez, Office of Ground Water and Drinking Water, Standards and Risk Management Division (Mail Code 4607M), Environmental Protection Agency, 1200 Pennsylvania Avenue, NW, Washington, DC 20460; telephone number: (202) 564-1735; email address: hernandez.samuel@epa.gov.

**SUPPLEMENTARY INFORMATION**: This notice is organized as follows:

#### I. General Information

- A. Does this Action Apply to Me?
- B. How can I get Copies of this Document and other Related Information?

#### II. Background

- A. What is Perchlorate?
- B. What is the Purpose of this Action?
- C. What is the EPA's statutory authority for this action?
- D. Statutory Framework and Perchlorate Regulatory History

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#### III. Final Regulatory Determination for Perchlorate

- A. May perchlorate have an adverse effect on the health of persons?
- B. Is perchlorate known to occur or is there a substantial likelihood that perchlorate will occur in public water systems with a frequency and at levels of public health concern?
- C. Is there a meaningful opportunity for the reduction of health risks from perchlorate for persons served by public water systems?
- D. What is the EPA's final regulatory determination on perchlorate?

#### IV. Summary of Key Public Comments on Perchlorate

- A. Health Effects Assessment
- B. Occurrence
- C. Regulatory Proposal and Alternatives
- D. SDWA Statutory Requirements
- E. Regulatory Determination Withdrawal

#### V. Conclusion

#### VI. References

## I. General Information

A. Does This Action Apply to Me?

This action will not impose any requirements on anyone. Instead, this action notifies interested parties of the EPA's withdrawal of the 2011 regulatory determination for perchlorate and the final regulatory determination to not regulate perchlorate based on new information.

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This notice also provides a summary of the major comments received on the June 26, 2019 (84

FR 30524) proposed NPDWR for perchlorate.

B. How can I get Copies of this Document and other Related Information?
The EPA has established a docket for this action under Docket ID No. EPA-HQ-OW-2018-0780. Publicly available docket materials are available electronically at [ HYPERLINK "http://www.regulations.gov/docket?D=EPA-HQ-OW-2018-0780" ].

#### II. Background

#### A. What is Perchlorate?

Perchlorate is a negatively charged inorganic ion that is comprised of one chlorine atom bound to four oxygen atoms (ClO<sub>4</sub>-), which is highly stable and mobile in the aqueous environment. Perchlorate comes from both natural and manmade sources. It is formed naturally via atmospheric processes and can be found within mineral deposits in certain geographical areas. It is also produced in the United States, and the most common compounds include ammonium perchlorate and potassium perchlorate used primarily as oxidizers in solid fuels to power rockets, missiles, and fireworks. Perchlorate can also result from the degradation of hypochlorite solutions used for water disinfection. The degradation into perchlorate occurs when hypochlorite solutions are improperly stored and handled. For the general population, most perchlorate exposure is through the ingestion of contaminated food or drinking water. At certain levels, perchlorate can prevent the thyroid gland from getting enough iodine, which can affect thyroid hormone

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production. For pregnant women with low iodine levels, sufficient changes in thyroid
hormone levels may cause changes in the child's brain development. For infants, changes
to thyroid hormone function can also impact brain development.

#### B. What is the purpose of this action?

The purpose of <u>today's this</u> action is to publish the EPA's notice to withdraw the 2011 regulatory determination and issue a final determination to not regulate perchlorate in drinking water. This notice presents the EPA's basis for this withdrawal and final regulatory determination, and the EPA's response to key issues raised by commenters in response to the June 26, 2019 (84 FR 30524) proposed rule (referred to hereinafter as "the 2019 proposal").

C. What is the EPA's statutory authority for this action?

The SDWA sets forth three criteria that must be met for the EPA to issue a maximum contaminant level goal (MCLG) and promulgate a national primary drinking water regulation (NPDWR). Specifically, the Administrator must determine that (1) "the contaminant may have an adverse effect on the health of persons"; (2) "the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern"; and (3) "in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by the public water systems" (SDWA 1412(b)(1)(A)).

The EPA has determined, based on www data and analysis since the issuance of the 2011 regulatory determination, that perchlorate does not in fact meet the statutorily-prescribed criteria

\*\*\* Deliberative Draft - Do Not Cite, Quote, or Release During Review\*\*\* for regulation. As described in Sections III & VI of the 2019 proposal, the wordata and analysis in the record indicate that perchlorate does not occur in public water systems with a frequency and at levels of public health concern. Specifically, the \*\*\* peer-reviewed health effects analysis yields a health based proposed MCLG and proposed alternative MCLGs for indicates that the concentration of perchlorate that are higher concentrations in drinking water (18 - 90 464) than the concentrations that the EPA considered to berepresenting the proposaled MCLG alternatives for levels of public health concern in (18-90 µg/L) is higher than the concentration considered in issuance of the analysis for the 2011 regulatory determination to regulate in 2011 (1-47 μg/L) (USEPA, 2019a). In addition, based on an evaluation of the nationally representative UCMR 1 systems, the updated occurrence analysis shows that the frequency of occurrence of perchlorate in public water systems at levels exceeding any of the alternative proposed MCLGs (4.38%-0.02%) is significantly lower (0.38% - 0.02%) than the frequency considered in the analysis for the 2011 regulatory determination (4% - 0.39%) (USEPA, 2019b). The EPA estimates that, even at the most stringent regulatory level considered in the 2019 proposal, (18 µg/L), not more than 15 systems (0.03% of all water systems in the U.S.) would need to take action to reduce levels of perchlorate. Based on this information, the EPA determines that perchlorate does not occur in public water systems "with a frequency...of public health concern" and thus does not meet the second criterion of the three required for regulation under the SDWA. In addition, while the third criterion is "in the sole judgement of the Administrator," the low occurrence provides ample support for the EPA's conclusion that the regulation of perchlorate does not present a

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"meaningful opportunity for health risk reduction for persons served by public water systems,"

within the meaning of 1412(b)(1)(A)(iii). Accordingly, because perchlorate no longer meets the statutory criteria for regulation, the EPA does not have the authority to issue a MCLG or promulgate a NPDWR for perchlorate.

The EPA's decision to withdraw the regulatory determination is supported by the legislative history underlying the 1996 amendments to the SDWA, which repealed the blanket rule requirement for the EPA to regulate an additional 25 contaminants every 3 years and replaced it with the current requirement for the EPA to determine whether regulation is warranted for stive contaminants every stive years. In describing the need for such amendment, the legislative history points to the view expressed at the Committee Hearing that "the current law is a one-size-fits-all program. It forces our water quality experts to spend scarce resources searching for dangers that often do not exist rather than identifying and removing real health risks from our drinking water" (S. Rep. 104-169 (1995) at 12). This amendment reflected Congress' clear intent that the EPA prioritize actual health risks in determining whether to regulate any particular contaminant. See id at 12 (noting that the amendment "repeals the requirement that the EPA regulate an additional 25 contaminants every 3 years replacing it with a new selection process that gives the EPA the discretion to identify contaminants that warrant regulation in the future").

The EPA's decision to withdraw the regulatory determination is also consistent with Congress' direction to prioritize the SDWA decisions based on the best available public health

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information. See 1412(b)(1)(B)(ii)(II) (findings supporting a determination to regulate "shall be based on the best available public health information"); 1412(b)(2)(A) (requiring that the EPA use "the best available, peer-reviewed science and supporting studies..." in carrying out any actions under this section). Although the EPA determined in 2011 that perchlorate met the criteria for regulation, new data and analysis developed by the PAAGENCY as part of the 2019 proposal demonstrate that the occurrence and health effects information used as the basis for the 2011 determination no longer constitute "best available information," are no longer accurate and no longer support the Agency's prioritization of perchlorate for regulation. Accordingly, not only is EPA not authorized to issue a MCLG or promulgate a NPDWR for perchlorate, but it would not be in the public interest to do so.

The EPA recognizes that the Act does not include a provision explicitly authorizing withdrawal of a regulatory determination. However, such authority is inherent in the authority to issue a regulatory determination under 1412(b)(1)(B)(ii)(II), particularly given the requirement that such determination be based on the "best available public health information," as discussed above. Accordingly, the EPA must have the inherent authority to withdraw a regulatory determination if the underlying information changes between regulatory determination and promulgation. Particularly In light of its concern that the EPA focus new contaminant regulators on priority health concerns, Congress could not have intended that the EPA's regulatory decision-making be hamstrung by older data when newer, more accurate scientific and public health data are available, especially when those data demonstrate that

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regulation of a new contaminant would not be present a meaningful opportunity for health risk reduction.

#### D. Statutory Framework and Perchlorate Regulatory History

Section 1412(b)(1)(B)(i) of the SDWA requires the EPA to publish every five years a Contaminant Candidate List (CCL). The CCL is a list of drinking water contaminants that are known or anticipated to occur in public water systems and are not currently subject to federal drinking water regulations. The EPA uses the CCL to identify priority contaminants for

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regulatory decision-making and information collection. Contaminants listed on the CCL may
require future regulation under the SDWA. The EPA included perchlorate on the first, second,
and third CCLs published in 1998 (63 FR 10274), 2005 (70 FR 9071), and 2009 (74 FR 51850).

The Agency EPA collects data on the CCL contaminants to better understand their potential health effects and to determine the levels at which they occur in public water systems.

The SDWA section 1412(b)(1)(B)(ii) requires that, every five years, the EPA, after consideration of public comment, issue a determination of whether or not to regulate at least five contaminants on each CCL. For any contaminant that the EPA determines meets the SDWA criteria for regulation, under the SDWA section 1412(b)(1)(E), the EPA must propose a NPDWR within two years and promulgate a final regulation within 18 months of the proposal (which may be extended by 9 additional months).

As part of its responsibilities under the SDWA, the EPA implements section 1445(a)(2), "Monitoring Program for Unregulated Contaminants." This section requires that once every five years, the EPA issue a list of no more than 30 unregulated contaminants to be monitored by public water systems. This monitoring is implemented through the Unregulated Contaminant Monitoring Rule (UCMR), which collects data from community water systems (CWS) and non-transient, non-community water systems (NTNCWS). The first four UCMRs collected data from a census of large water systems (serving more than 10,000 people) and from a statistically representative sample of small water systems. On September 17, 1999, the EPA published its first UCMR (64 FR 50556), which required all large systems and a representative sample of

\*\*\* Deliberative Draft – Do Not Cite, Quote, or Release During Review\*\*\* small systems to monitor for perchlorate and 25 other contaminants (USEPA, 1999). Water system monitoring data for perchlorate was collected from 2001 to 2005.

The EPA and other federal agencies asked the National Research Council (NRC) to evaluate the health implications of perchlorate ingestion. In its 2005 report, the NRC concluded that perchlorate exposure inhibits the transport of iodide<sup>1</sup> into the thyroid by a protein molecule known as the sodium/iodide symporter (NIS), which may lead to decreases in two thyroid hormones, thyroxine (T3) and triiodothyronine (T4), and increases in thyroid-stimulating hormone (TSH) [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"a1mn5hjprkt","properties": {"formattedCitation":"(National Research Council (NRC), 2005b)","plainCitation":"(National Research Council (NRC), 2005b)","noteIndex":0},"citationItems":[{"id":350,"uris":["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData": {"id":350,"type":"book","title":"Health Implications of Perchlorate Ingestion","publisher":"National Academies Press","publisher-place":"Washington, DC","event-place":"Washington, DC","author":[{"literal":"National Research Council (NRC)"}],"issued": {"date-parts":[["2005"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Additionally, the NRC concluded that the

<sup>1</sup> For the purposes of this notice, "iodine" will be used to refer to dietary intake before entering the body. Once in the body, "iodide" will be used to refer to the ionic form.

\*\*\* Deliberative Draft - Do Not Cite, Quote, or Release During Review\*\*\* most sensitive population to perchlorate exposure are "the fetuses of pregnant women who might have hypothyroidism or iodide deficiency" (p. 178). The EPA established a reference dose (RfD) consistent with the NRC's recommended RfD of 0.7 µg/kg/day for perchlorate. The reference dose is an estimate of a human's daily exposure to perchlorate that is likely to be without an appreciable risk of adverse effects. This RfD was based on a study [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"a3u94lt6me","properties":{"formattedCitation":"(Greer, Goodman, Pleus, & Greer, 2002)", "plainCitation": "(Greer, Goodman, Pleus, & Greer, 2002)","noteIndex":0},"citationItems":[{"id":387,"uris":["http://zotero.org/groups/945096/items/ 6AKUNIX6"],"uri":["http://zotero.org/groups/945096/items/6AKUNIX6"],"itemData":{"id":387 "type": "article-journal", "title": "Health effects assessment for environmental perchlorate," contamination: the dose response for inhibition of thyroidal radioiodine uptake in humans", "container-title": "Environmental Health Perspectives","page":"927","volume":"110","issue":"9","author":[{"family":"Greer","given":"M onte A."},{"family":"Goodman","given":"Gay"},{"family":"Pleus","given":"Richard C."},{"family":"Greer","given":"Susan E."}],"issued":{"dateparts":[["2002"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] of perchlorate's inhibition of radioactive iodine uptake in healthy adults and the application of an uncertainty factor of 10 for intraspecies variability [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"0oHz805e", "properties": {"formattedCitation":"(USEPA,

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2005b)", "plainCitation": "(USEPA,

2005b)", "noteIndex":0}, "citationItems": [{"id":980, "uris": ["http://zotero.org/groups/945096/item

s/LHANJBR6"], "uri": ["http://zotero.org/groups/945096/items/LHANJBR6"], "itemData": {"id":9

80, "type": "article", "title": "Integrated Risk Information System (IRIS) Chemical Assessment

Summary: Perchlorate (ClO4-) and Perchlorate Salts", "publisher": "USEPA National Center for

Environmental Assessment", "author": [{"literal": "USEPA"}], "issued": {"dateparts": [["2005"]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}].

In October 2008, the EPA published a preliminary regulatory determination not he regulate perchlorate in drinking water and requested public comment (73 FR 60262). In that preliminary determination, the EPA found that perchlorate did not occur with a frequency and at levels of public health concern and that development of a regulation did not present a meaningful opportunity for health risk reduction for persons served by public water systems. The EPA derived and used a Health Reference Level (HRL) of 15 μg/L based on the RfD of 0.7 μg/kg/day and body weight and exposure information for pregnant women in making this conclusion [ ADDIN ZOTERO ITEM CSL CITATION

 $\label{lem:control} \begin{tabular}{ll} $\{$"citationID":"FZ6WMtAv","properties": \{$$"formattedCitation":"(USEPA, 2008a)","plainCitation":"(USEPA, 2008a)","(USEPA, 2008a)","(USEPA$ 

\*\*\* Deliberative Draft - Do Not Cite, Quote, or Release During Review\*\*\* 934,"type": "article-journal", "title": "Drinking water: Preliminary regulatory determination on perchlorate", "container-title": "Federal Register", "volume": "73", "issue": "198", "abstract": "SUMMARY: This action presents EPA's preliminary regulatory determination for perchlorate in accordance with the Safe Drinking Water Act (SDWA). The Agency has determined that a national primary drinking water regulation (NPDWR) for perchlorate would not present \"a meaningful opportunity for health risk reduction for persons served by public water systems." The SDWA requires EPA to make determinations every five years of whether to regulate at least five contaminants on the Contaminant Candidate List (CCL). EPA included perchlorate on the first and second CCLs that were published in the Federal Register on March 2, 1998 and February 24, 2005. Most recently, EPA presented final regulatory determinations regarding 11 contaminants on the second CCL in a notice published in the Federal Register on July 30, 2008. In today's action, EPA presents supporting rationale and requests public comment on its preliminary regulatory determination for perchlorate. EPA will make a final regulatory determination for perchlorate after considering comments and information provided in the 30-day comment period following this notice. EPA plans to publish a health advisory for perchlorate at the time the Agency publishes its final regulatory determination to provide State and local public health officials with technical information that they may use in addressing local contamination.", "ISSN": "ISSN 0097-6326 EISSN 2167-2520", "shortTitle": "Federal Register", "journalAbbreviation": "Fed.

Reg.", "language": "English", "author": [{"literal": "USEPA"}], "issued": {"date-

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parts":[["2008"]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}]. Using the UCMR 1 occurrence data, the EPA estimated that less than 1% of drinking water systems (serving approximately 1 million people) had perchlorate levels above the HRL of 15 µg/L. Based on this information the AgencyEPA found that perchlorate did not occur frequently at a frequency and at levels of public health concern. The EPA also determined there was not a meaningful opportunity for a NPDWR for perchlorate to reduce health risks.

In August 2009, the EPA published a supplemental request for comment with new analysis that derived potential alternative Health Reference Levels (HRLs) for 14 life stages, including infants and children. The analysis used the RfD of 0.7 μg/kg/day and life stage-specific bodyweight and exposure information, resulting in comparable perchlorate concentrations in drinking water, based on life stage, of between 1 μg/l to 47 μg/l (74 FR 41883; USEPA, 2009).

FR 7762; USEPA, 2011) after careful consideration of public comments on the October 2008 and August 2009 notices. The Agency PA found at that time that perchlorate may have an adverse effect on the health of persons, it is known to occur in public drinking water systems with a frequency and at levels that present a public health concern, and regulation of perchlorate presented a meaningful opportunity for health risk reduction for persons served by public water systems. The Agency PA stated then that: "Based on the data in Table 1 and the range of potential alternative HRLs, EPA has determined that perchlorate is known to occur or there is a

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substantial likelihood that it will occur with a frequency and at levels of public health

concern."(USEPA, 2011, p. 7765). The EPA found that as many as 16 million people could

potentially be exposed to perchlorate at levels of concern, up from 1 million people originally
estimated in the 2008 notice.

As a result of the determination, and as required by the SDWA, section 1412(b)(1)(E), the EPA initiated the process to develop a MCLG and a NPDWR for perchlorate.

In September 2012, the U.S. Chamber of Commerce (the Chamber) submitted to the EPA a Request for Correction under the Information Quality Act regarding the EPA's regulatory determination. In the request, the Chamber claimed that the UCMR 1 data used in the EPA's occurrence analysis did not comply with data quality guidelines and were not representative of current conditions. In response to this request, the EPA reassessed the data and removed certain source water samples that could be paired with appropriate follow-up samples located at the entry point to the distribution system. The EPA also updated the UCMR 1 data in the analysis for systems in California and Massachusetts, using state compliance data to reflect current occurrence conditions after state regulatory limits for perchlorate were implemented.

As required by section 1412(d) of the SDWA, as part of the NPDWR development process, the EPA requested comments from the Science Advisory Board (SAB) in 2012, seeking guidance on how best to consider and interpret the life stage information, the epidemiologic and biomonitoring data since the NRC report, physiologically-based pharmacokinetic (PBPK)

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analyses, and the totality of perchlorate health information to derive an MCLG for perchlorate. In May 2013, the SAB recommended that the EPA:

- derive a perchlorate MCLG that addresses sensitive life stages through physiologically-based pharmacokinetic/pharmacodynamic modeling based upon its mode of action rather than the default MCLG approach using the RfD and specific chemical exposure parameters;
- expand the modeling approach to account for thyroid hormone perturbations and potential adverse neurodevelopmental outcomes from perchlorate exposure;
- utilize a mode-of-action framework for developing the MCLG that links the steps in the proposed mechanism leading from perchlorate exposure through iodide uptake inhibition—to thyroid hormone changes—and finally to neurodevelopmental impacts; and
- "Extend the [BBDR] model expeditiously to \_\_\_\_\_provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (SAB for the U.S. EPA, 2013, p. 19).

To address the SAB recommendations, the EPA revised an existing PBPK/PD model that describes the dynamics of perchlorate, iodide, and thyroid hormones in a woman during the third trimester of pregnancy (Lumen, Mattie, & Fisher, 2013; USEPA, 2009b). The EPA also created its own Biologically Based Dose Response (BBDR) models that included the additional sensitive

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life stages identified by the SAB, i.e., breast- and bottle-fed neonates and infants (SAB for the U.S. EPA, 2013, p. 19).

To determine whether the Agency had implemented the SAB recommendations for modeling thyroid hormone changes, the EPA convened an independent peer review panel to evaluate the BBDR models in January 2017 (External Peer Reviewers for USEPA, 2017). The EPA considered the recommendations from the 2017 peer review and made necessary model revisions to increase the scientific rigor of the model and the modeling results.

The EPA convened a second independent peer review panel in January 2018 to evaluate the revisions to the BBDR model. The EPA also presented several approaches to link the thyroid hormone changes in a pregnant mother predicted by the BBDR model to neurodevelopmental effects using evidence from the epidemiological literature (External Peer Review for U.S. EPA, 2018).

In response to a lawsuit brought to enforce the deadlines in the SDWA<sub>2</sub> section 1412(b)(1)(E), on October 18, 2016, the U.S. District Court for the Southern District of New York entered a consent decree, requiring the EPA to sign for publication a proposal for a MCLG and NPDWR for perchlorate in drinking water no later than October 31, 2018, and to sign for publication a final MCLG and NPDWR for perchlorate in drinking water no later than December 19, 2019. The deadline for the EPA to propose a MCLG and NPDWR for perchlorate in drinking water was later extended to May 28, 2019, and the date for signature of a final MCLG and

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NPDWR was extended to be no later than June 19, 2020. The consent decree is available in the

docket for today at this action.

In compliance with the deadline established in the consent decree, on May 23, 2019, the

EPA Administrator signed a proposed rulemaking notice seeking public comment on a range of

options regarding the regulation of perchlorate in public drinking water systems. The proposed

rulemaking notice was published in the Federal Register on June 26, 2019. 84 Fed. Reg. 30524.

The EPA proposed a NPDWR for perchlorate with an MCL and MCLG of 56  $\mu$ g/L. The

proposed MCLG of 56  $\mu\text{g/L}$  was based on avoiding a 2 point IQ decrement associated with

exposure to perchlorate in drinking water during the most sensitive life stage (the fetus) within a

specific segment of the population (iodine deficient pregnant women).

The AgencyEPA also requested comment on two alternative MCL/MCLG values of 18

 $\mu g/L$  and 90  $\mu g/L$ , respectively. These alternatives were based upon avoiding 1 point and 3 point

IQ decrements respectively, associated with perchlorate exposure. Additionally, the EPA

requested comment on whether the 2011 regulatory determination should be withdrawn, based

on new information including updated occurrence data on perchlorate in drinking water and new

analysis of the concentration of perchlorate in drinking water that represents a level of health

concern.

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# III. Withdrawal of the 2011 Regulatory Determination and Final Determination to Not Regulate Perchlorate

In determining whether to regulate a particular contaminant, the EPA must follow the criteria mandated by the 1996 SDWA Amendments. Specifically, in order to issue a MCLG and NPDWR for perchlorate, the EPA must determine that perchlorate "may have an adverse effect on the health of persons," that perchlorate occurs at "a frequency and at levels of public health concern" in public water systems, and that regulation of perchlorate in drinking water systems "presents a meaningful opportunity for health risk reductions for persons served by public water systems." The SDWA<sub>2</sub> section 1412(b)(1)(A). In preparing the 2019 proposal for perchlorate, the EPA updated and improved information on the levels of public health concern and the frequency and levels of perchlorate in public water systems. The following is the EPA's reassessment of the regulatory determination criteria applied to the improved basis available health science and occurrence data available for perchlorate.

#### A. May perchlorate have an adverse effect on the health of persons?

Yes, perchlorate may have adverse health effects. The perchlorate anion is biologically significant specifically with respect to the functioning of the thyroid gland. Perchlorate can interfere with the normal functioning of the thyroid gland by inhibiting the transport of iodide into the thyroid, resulting in a deficiency of iodide in the thyroid. Perchlorate inhibits (or blocks) iodide transport into the thyroid by chemically competing with iodide, which has a similar shape and electric charge. The transfer of iodide from the blood into the thyroid is an essential step in

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the synthesis of thyroid hormones. Thyroid hormones play an important role in the regulation of metabolic processes throughout the body and are also critical to developing fetuses and infants, especially for brain development. Because the developing fetus depends on an adequate supply of maternal thyroid hormones for its central nervous system development during the first and second trimester of pregnancy, iodide uptake inhibition from perchlorate exposure has been identified as a concern in connection with increasing risk of neurodevelopmental impairment in fetuses of pregnant women with low dietary iodine. Poor iodide uptake and subsequent impairment of the thyroid function in pregnant and lactating women have been linked to delayed development and decreased learning capability in their infants and children (NRC, 2005).

Therefore, the EPA continues to find that perchlorate may have an adverse effect on the health of persons.

B. Is perchlorate known to occur or is there a substantial likelihood that perchlorate will occur in public water systems with a frequency and at levels of public health concern?

The EPA has determined that perchlorate does not occur at a frequency and at levels of public health concern in public water systems. The EPA has made this determination by comparing the best available data on the occurrence of perchlorate in public water systems to potential MCLGs for perchlorate.

In past regulatory determinations, the EPA has identified HRLs as benchmarks against which the EPA compares the concentration of a contaminant found in public water systems to determine if it isoccurs at levels of public health concern. For the 2011 regulatory determination the EPA identified potential HRLs values ranging from 1 to 47 µg/L for 14 different life\_stages.

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These HRLs were not final decisions about the level of perchlorate in drinking water that is

\*\*\* necessary to protect any particular population without adverse effects. For the 2019 proposal, the

EPA derived three potential MCLGs for perchlorate of 18, 56, and 90 µg/L for the most sensitive

life-stage \*\* stage \*

The rationale used in deriving the numerical values is presented in greater detail in the EPA's technical support document titled "Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water" (USEPA, 2019b).

The EPA compared these potential MCLG values to the updated perchlorate UCMR 1 occurrence data set. A comprehensive description of the perchlorate occurrence data is presented in Section VI of the 2019 proposal. It is also available in the "Perchlorate Occurrence and Monitoring Report" (USEPA, 2019a).

The occurrence data for perchlorate \*\*\*\* collected from 3,865 PWSs between 2001 and 2005 under the UCMR 1. The Agency has in the 2019 proposal, the EPA modified its the UCMR 1 data set in response to concerns raised by stakeholders regarding the data quality and to

\*\*\* Deliberative Draft - Do Not Cite, Quote, or Release During Review\*\*\* represent current conditions in California and Massachusetts, which have enacted perchlorate regulations since the UCMR 1 data was were collected. Massachusetts promulgated a drinking water standard for perchlorate of 2 µg/L in 2006 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"8DPpSrv3","properties":{"formattedCitation":"(MassDEP, 2006)", "plainCitation": "(MassDEP, 2006)","noteIndex":0},"citationItems":[{"id":151,"uris":["http://zotero.org/groups/945096/items/ 9893MBZH"],"uri":["http://zotero.org/groups/945096/items/9893MBZH"],"itemData": {"id":151 "type":"personal\_communication", "title": "Letter to Public Water Suppliers concerning new, perchlorate regulations", "URL": "https://www.mass.gov/lists/perchlorate-backgroundinformation-and-standards#perchlorate---final-standards-","author":[{"literal":"MassDEP"}],"issued":{"dateparts":[["2006"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ], and California promulgated a drinking water standard of 6 µg/L in 2007 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID": "cfr6HNhg", "properties": {"formattedCitation": "(California Department of Public Health, 2007)", "plainCitation": "(California Department of Public Health, 2007)", "noteIndex":0}, "citationItems":[{"id":150, "uris":["http://zotero.org/groups/945096/items/ RA45NKLQ"],"uri":["http://zotero.org/groups/945096/items/RA45NKLQ"],"itemData": {"id":15 0,"type":"personal communication", "title": "State Adoption of a Perchlorate Standard", "URL": "https://www.waterboards.ca.gov/drinking water/certlic/drinkingwater/docum

Page [ PAGE ] of [ NUMPAGES ]

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ents/perchlorate/AdoptionMemotoWaterSystems-10-2007.pdf", "author": [{"literal": "California Department of Public Health"}], "issued": {"date-parts": [["2007"]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Systems in these states are now required to keep perchlorate levels in drinking water below their state limits. As discussed below, the EPA finds that perchlorate levels in drinking water and sources of drinking water have decreased since the UCMRAUCMR\_1 data collection. The main factors contributing to the decrease in perchlorate levels are the promulgation of drinking water regulations for perchlorate in California and Massachusetts and the ongoing remediation efforts in the state of Nevada to address perchlorate

To update the occurrence data for systems sampled during UCMR 1 from California and Massachusetts, the EPA identified all systems and corresponding entry points which had reported perchlorate detections in UCMR 1. Once the systems and entry points with detections were appropriately identified, the EPA then used a combination of available data from Consumer Confidence Reports (CCRs) and perchlorate compliance monitoring data from California (https://sdwis.waterboards.ca.gov/PDWW/) and Massachusetts (https://www.mass.gov/service-details/public-water-supplier-document-search) to match current compliance monitoring data (where available) to the corresponding water systems and entry points sampled during UCMR 1.

contamination in groundwater adjacent to the lower Colorado River upstream of Lake Mead.

With these updates. The EPA has determined that the UCMR 1 data with these updates are the best available data collected in accordance with accepted methods regarding the

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frequency and level of perchlorate nationally. The UCMR 1 data are from a census of the large water systems (serving more than 10,000 people) and a statistically representative sample of small water systems that provides the best available, national assessment of perchlorate occurrence in drinking water.

The EPA used entry point maximum measurements to estimate potential baseline occurrence and exposure at levels that exceed the potential MCLG thresholds. The maximum measurements indicate highest perchlorate levels reported in at least one quarterly sample from surface water systems and at least one semi-annual sample from ground water systems.

Table 1: Perchlorate Occurrence and Exposure (Updated UCMR 1 Data Set)

Threshold Concentration (µg/L)	Entry Points with Detections above Threshold	Water Systems with Detections above Threshold	Percent of U.S. Water Systems with Detections above Threshold	Population Served
18 μg/L	17	15	0.03 %	620,560
56 μg/L	2	2	0.004 %	32,432
90 μg/L	1	1	0.002 %	25,972

Table 1 presents the number and percentage of water systems that reported perchlorate at levels exceeding the three proposed MCLG threshold concentrations. In summary, the updated perchlorate occurrence information suggests that at an MCLG of 18  $\mu$ g/L, there would be 15 systems (0.03% of all water systems in the U.S.) that would exceed the threshold, at an MCLG of 56  $\mu$ g/L, two systems (0.004% of all water systems in the U.S.) would exceed the threshold,

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and finally one system would exceed the MCLG threshold of 90 µg/L. Based on the analysis of drinking water occurrence presented in the 2019 proposal and the data summarized in Table 1 and the range of potential MCLGs, the EPA concludes that perchlorate does not occur at a frequency and at levels of public health concern in public water systems.

While the EPA has made its conclusion that perchlorate occurs infrequently does not occur at a frequency and at levels of public health concern in public water systems based on the updated UCMR 1 data, the EPA also sought to find additional information about the perchlorate levels at the 15 water systems that had at least one reported result greater than 18 µg/L in the updated UCMR 1 data. The EPA found that perchlorate levels have been reduced at many of these water systems. Although these water systems were not required to take actions to reduce perchlorate in drinking water, many had conducted additional monitoring for perchlorate and found decreased levels or had taken mitigation efforts to address perchlorate, confirming the EPA's conclusion described above. The status of each of these systems is described in Table 2 below.

Table 2: Update on Systems with Perchlorate levels above 18  $\mu g/L$  in the UCMR 1

State	System Name	Range of UCMR 1 Results (µg/L)**	Update on Mitigation and Levels of Perchlorate++
Florida	Sebring Water	ND-70	The EPA contacted the Sebring system in January 2020. Operations personnel indicated that no follow-up/updated monitoring data for perchlorate are available.

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State	System Name	Range of UCMR 1 Results (µg/L)**	Update on Mitigation and Levels of Perchlorate <sup>++</sup>
Florida	Manatee County Utilities Dept	ND-30	Researchers contacted the system to identify the source of perchlorate. System personnel attributed the sole perchlorate detection under <u>UCMR+UCMR 1</u> to analytical error. System personnel indicated that three other quarterly samples collected under <u>UCMR+UCMR 1</u> as well as other subsequent perchlorate sampling efforts were non-detect. Source: AWWA (2008)
Georgia	Oconee Co Watkinsville	38 (single sample)	Researchers contacted the system and found that a perchlorate contaminated well was removed from service in 2003. The system indicates that perchlorate is no longer detected. Source: Luis et al. (2019)
Louisiana	St. Charles Water District 1 East Bank	ND-24	The EPA was not able to identify updated data on perchlorate levels for this system.
Maryland	City of Aberdeen	ND-19	The system's 2018 Consumer Confidence Report (CCR) indicates that perchlorate was not detected. According to the Maryland Department of Environment, perchlorate was not detected in this system in 2019. In addition, researchers contacted the system and found that there has been no detection of perchlorate since treatment was installed in 2009. Source: Luis et al. (2019)
Maryland	Chapel Hill - Aberdeen Proving Grounds	ND-20	The EPA contacted the Chapel Hill System in January 2020. Water system personnel indicate that the Chapel Hill WTP was taken off-line and was replaced with a new treatment plant and five new production wells. The new treatment plant started operations on January 27, 2020. System personnel also indicate that monitoring was conducted in November 2019 and perchlorate was not detected in either the source well water or the finished water. In

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State	System Name	Range of UCMR 1 Results (µg/L)**	Update on Mitigation and Levels of Perchlorate++	
			addition, according to the Maryland Department of Environment, perchlorate was not detected in this system in 2019.	
Mississippi	Hilldale Water District	ND-20	The EPA contacted the Hilldale System in January 2020. Water system personnel indicated that no follow-up/updated monitoring data for perchlorate are available.	
New Mexico	Deming Municipal Water System	15-20	Data from the EPA's SDWIS/FED database indicates that the entry point that reported detections in UCMR-1 (Well #3) is now inactive (i.e., the contaminated source is no longer in use). Source: SDWIS/FED (2016)	
Nevada	City of Henderson	6-23	Researchers report that the perchlorate levels described in the system's CCR ranged from non-detect to 9.7 µg/L. Source AWWA (2008).	
Ohio	Fairfield City PWS	6-27	The EPA contacted the Fairfield City System in January 2020. Water system personnel indicated that follow-up monitoring was conducted after UCMR+UCMR+1, between 2002 and 2004. The Ohio EPA provided copies of the follow-up monitoring results which indicate that results at the entry point ranged from non-detect to 13 µg/L.	
Ohio	Hecla Water Association-Plant PWS	ND-32	The EPA contacted the Hecla Water Association System in January 2020. Water system personnel indicated that that no follow-up/updated monitoring data for perchlorate are available.	
Oklahoma	Enid	ND-30	The EPA reviewed Oklahoma's monitoring data and did not find any monitoring results reported for perchlorate.	
Pennsylvania	Meadville Area Water Authority	ND-33	The EPA contacted the Meadville System in January 2020. Water system personnel indicated that no	

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State	System Name	Range of UCMR 1 Results (µg/L)**	Update on Mitigation and Levels of Perchlorate++
			follow-up/updated monitoring data for perchlorate are available.
Puerto Rico	Utuado Urbano	ND-420	The EPA contacted the Puerto Rico Aqueduct and Sewer Authority (PRASA) in January 2019. PRASA personnel indicated that no updated monitoring data for perchlorate are available. NOTE: The PRASA personnel stated that the Utuado water system was significantly impacted by hurricane Maria and monitoring records from years prior to 2017 were lost.
Texas	City of Levelland	ND-32	Researchers found that a water storage tank was the source of perchlorate contamination, the wells feeding the tank were tested by the state and perchlorate was not detected. The water tank was shut off from service. Source: Luis et al. (2019).

<sup>\*\* -</sup> Values have been rounded. ND describes a sampling event where perchlorate was not detected at or above the UCMR 1 minimum reporting level of 4  $\mu$ g/L. UCMR 1 results collected between 2001 and 2005.

C. Is there a meaningful opportunity for the reduction of health risks from perchlorate for persons served by public water systems?

The Agency's <u>EPA's</u> analysis presented in the 2019 proposal demonstrates that a NPDWR for perchlorate does not present a meaningful opportunity for health risk reduction for persons served by public water systems. As discussed above, the EPA found that perchlorate occurs with very low frequency at levels of public health concern. Based on updated UCMR 1

<sup>++ -</sup> To obtain updated data and/or information regarding perchlorate levels, the EPA reviewed Consumer Confidence Reports and other publicly available data, as well as published studies. In addition, the EPA contacted some water systems for information about current perchlorate levels. (USEPA, 2020b)

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occurrence information, there were 15 water systems (0.03% of all water systems in the U.S.)

that detected perchlorate in drinking water above the lowest proposed alternative MCLG of 18

µg/L and only 1 system had a detection above the proposed alternative MCLG of 90 µg/L.

Specifically, Table 1 presents the population served by PWSs that were monitored under UCMR

1 for which the highest reported perchlorate concentration was greater than the identified thresholds. The EPA estimates² that the number of people who may be potentially consuming water containing perchlorate at levels that could exceed the levels of concern for perchlorate could range between 26,000 total 620,000. The small number of water systems with perchlorate levels greater than identified thresholds and the corresponding small population served provides ample support for the EPA's conclusion that the regulation of perchlorate does not present a "meaningful opportunity for health risk reduction for persons served by public water systems," within the meaning of the SDWA, section 1412(b)(1)(A)(iii).

The EPA also considered the findings of the Health Risk Reduction and Cost Analysis (HRRCA, USEPA 2019c) as additional information supporting withdrawal of the regulatory determination. The HRRCA for perchlorate (which was presented in the 2019 proposal) provides a unique set of economic data indicators that are not available for regulatory determinations because the HRRCA is required for a proposed NPDWR under SDWA Section 1412(b)(3)(C),

 $<sup>^2</sup>$  The values shown in Table 1 are based on the revised UCMR 1 data. The EPA also applied statistical sampling weights to the small systems results to extrapolate to national results. There was one small system included in the statistical sample stratum which had a perchlorate measurement exceeding 18  $\mu g/L$ . Accordingly, the EPA estimates that approximately 41,000 small system customers may be exposed to perchlorate greater than 18  $\mu g/L$ . Page [ PAGE ] of [ NUMPAGES ]

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but is not prepared required to support a regulatory determination. Perchlorate is a unique
contaminant for which Accordingly, because the Agency has done significant new analysis not
undertaken for EPA untially determined that perchlorate met the criteria for regulation and began
the regulatory determinations. Accordingly analysis process, the HRRCA was available with
respect to perchlorate, and the Agency considered this comprehensive economic analysis in
informing its decision to withdraw the regulatory determination.

Specifically, the HRRCA provides a description of the potential quantifiable and non-quantifiable benefits and costs of a drinking water regulation for perchlorate. For all potential regulatory levels considered for perchlorate (18, 56, and 90 µg/L) the total costs associated with establishing a regulation were substantially higher than the potential range of quantifiable benefits. The infrequent occurrence of perchlorate at levels of health concern imposes high monitoring and administrative cost burdens on public water systems and the states, while having little impact on health risk reductions and the associated low estimates of benefits.

Based on a comparison of costs and benefits estimated at the three potential regulatory levels, the EPA determined in the 2019 proposal that the benefits of establishing a drinking water regulation for perchlorate do not justify the potential costs.

A drinking water regulation for perchlorate would impose significant burden on states and water systems, mainly associated with requirements for monitoring but which would result in very few systems having to take action to reduce perchlorate levels. It is of paramount importance that water systems (particularly medium, small and economically distressed systems)

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focus their limited resources on actions that ensure compliance with existing NPDWRNPDWRs

and sustainmantain their technical, managerial, and financial capacity mantain to improve

system operations and -the quality of water being provided to their customers rather than

spending resources monitoring for contaminants that are unlikely to occur.

D. What is the EPA's final regulatory determination on perchlorate?

Based on the EPA's new analysis of the best available public health information, and after careful review and consideration of public comments on the June 2019 proposal, the Agency is withdrawing its 2011 determination and is now making a final determination to not to regulate perchlorate. Accordingly, the EPA will not issue a NPDWR for perchlorate at this time. While the EPA has found that perchlorate may have an adverse effect on human health, based on the analysis presented in this notice and supporting record, the EPA has determined that perchlorate does not occur in public water systems atwith a frequency and at levels of public health concern and that regulation of perchlorate does not present a meaningful opportunity to reduce health risks for persons served by public water systems. This conclusion is based on the best available peer reviewed science and data collected in accordance with accepted methods on perchlorate health effects and occurrence.

## IV. Summary of Key Public Comments on Perchlorate

The EPA received approximately 1,500 comments from individuals or organizations on the June 2019 proposal. This section briefly discusses the key technical issues raised by commenters and the EPA's response. Comments are also addressed in the "Comment Response

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Document for the Final Regulatory Action for Perchlorate' (USEPA, 2020a) available at http://www.regulations.gov (Docket ID No. EPA–HQ–OW–2018–0780).

A. SDWA Statutory Requirements and the EPA's Authority

The EPA received comments stating the Agency should promulgate an MCLG and MCL for perchlorate and comments stating the Agency should not promulgate a regulation. After considering these comments the Agency EPA has re-evaluated perchlorate in accordance with the SDWA, section 1412-(b-(1)(A)), which requires that the EPAAgency promulgate a NPDWR if (i) the contaminant may have an adverse effect on the health of persons; (ii) the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern; and (iii) in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by public water systems.

The AgencyEPA has determined, based upon the best available peer reviewed science and data collected in accordance with accepted methods, that perchlorate does not occur withat a frequency and at levels of public health concern, and there is not that regulation of perchlorate does not present a meaningful opportunity for health risk reduction.

Therefore, Because perchlorate does not meet the Agency has determined not statutory criteria for regulation, the EPA lacks the authority to promulgate issue a MCLG or NPDWR for perchlorate—, and is therefore withdrawing its 2011 regulatory determination and issuing

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this final determination to not regulate perchlorate. For more information regarding EPA's
statutory authority to withdraw its regulatory determination, see Section II C above.

## B. Health Effects Assessment

## Health Effects/MCLG Derivation

The Agency EPA received comments indicating that the EPAAgency should utilize different approaches to derive the MCLG for perchlorate including approaches that some states used to develop their perchlorate advisory levels or drinking water standards. The AgencyEPA considered a number of alternative approaches to develop the MCLG for perchlorate and in accordance with as SDWA, section 1412(e), the Agency sought recommendations from the Science Advisory Board. The EPA derived the proposed MCLG for perchlorate based on the approach recommended by the Science Advisory Board (SAB) (SAB for the U.S. EPA, 2013). The SAB recommended that "the EPA derive a perchlorate MCLG that addresses sensitive life stages through physiologically-based pharmacokinetic/pharmacodynamic modeling based upon its mode of action rather than the default MCLG approach using the RfD and specific chemical exposure parameters." The EPA has implemented these recommendations and has obtained two independent peer reviews of the analysis. These peer reviewers stated that: "Overall, the panel agreed that the EPA and its collaborators have prepared a highly innovative state-of-the-science set of quantitative tools to evaluate neurodevelopmental effects that could arise from drinking water exposure to perchlorate. While there is always room for improvement of the models,

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with limited additional work to address the committee's comments below, the current models
are fit-for-purpose to determine an MCLG' (External Peer Reviewers for USEPA,
2018, p. 2).

The EPA received comments indicating the most sensitive life stages were not selected and/or considered in the Agency's approach. The EPA disagrees. Gestational exposure to perchlorate during neurodevelopment is the most sensitive time period. The NRC concluded that the population most sensitive population to perchlorate exposure are "the fetuses of pregnant women who might have hypothyroidism or iodide deficiency" [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"a1mn5hjprkt", "properties": {"formattedCitation":"(National Research Council (NRC), 2005b)", "plainCitation": "(National Research Council (NRC), 2005b)", "noteIndex":0}, "citationItems":[{"id":350, "uris":["http://zotero.org/groups/945096/it ems/TN6HMC9D"],"uri":["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData" :{"id":350,"type":"book","title":"Health Implications of Perchlorate Ingestion", "publisher": "National Academies Press", "publisher-place": "Washington, DC", "event-place": "Washington, DC", "author": [{"literal": "National Research Council  $(NRC)"\}], "issued": \{"date-parts": [["2005"]]\}\}\}], "schema": "https://github.com/citation-style-parts": [["2005"]]]\}$ language/schema/raw/master/csl-citation.json"} ]. In addition, there is clear evidence that disrupted maternal thyroid hormone levels during gestation can impact neurodevelopment later in life (Alexander et al., 2017; Costeira et al., 2011; Endendijk et al., 2017; Ghassabian,

\*\*\* Deliberative Draft - Do Not Cite, Quote, or Release During Review\*\*\* Bongers-Schokking, Henrichs, Jaddoe, & Visser, 2011; Glinoer & Delange, 2000; Glinoer & Rovet, 2009; Gyllenberg et al., 2016; Henrichs et al., 2010; Korevaar et al., 2016; Morreale de Escobar, Obregón, & Escobar del Rey, 2004; Noten et al., 2015; Pop et al., 2003, 1999; SAB for the U.S. EPA, 2013; Thompson et al., 2018; van Mil et al., 2012; Wang et al., 2016; Zoeller & Rovet, 2004; Zoeller et al., 2007). The EPA's analysis concludes that The available data demonstrate that the fetus of the first trimester pregnant mother, when compared to other life-stages, experiences the greatest impact from equivalent-desesthe same dose of perchlorate-exposure, which is described in detail in Section 6 of the document "Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water" (USEPA, 2019a). In addition, the EPA disagrees Some commenters suggested that the bottle-fed infants are the most infant is a more sensitive population life-stage. The EPA disagrees as described in the January 2017 Peer Review Report on the original Biologically Based Dose Response (BBDR) model, the bottle-fed infant's thyroid hormone levels were not impacted by doses of perchlorate up to 20 µg/day (External Peer Reviewers for USEPA, 2017). This lack of any impact is due primarily to the iodine in the formula, which offsets the impact of perchlorate on the thyroid.

The AgencyEPA received comments advocating for the use of athe population-based approach- evaluating the shift in the proportion of a population that would fall below a hypothyroxinemic cut point under a perchlorate exposure scenario. The AgencyEPA chose to develop the MCLG using dose-response functions from the epidemiological literature to

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estimate neurodevelopmental impacts in the offspring of pregnant women exposed to
perchlorate. The EPA selected this proposed approach because it is consistent with the
SDWA's definition of a MCLG to avoid adverse health effects and because it is most
consistent with the SAB recommendations. In addition, the fact that thyroid hormone levels
vary by reference population and that there is not a defined value representing
hypothyroxinemia makes the population-based approach less desirable than the approach
selected (USEPA, 2018b2018).

## End Point Selection/Basis

The Agency EPA received comments regarding the magnitude of an IQ change which should be used in considering the MCLG. Many comments stated that the Agency should at most consider a 1% IQ change. However, several commenters stated a 3% change is too small to have a meaningful impact and suggested the Agency consider a higher IQ percent change. The Agency's EPA's proposed MCLG was based upon avoiding a 2% change in IQ in the most sensitive life stage and the EPA also requested comment on alternative options for the MCLG that would respectively avoid 1% or 3% change in IQ in the most sensitive lifestage Many comments stated that the EPA should at most consider a 1% IQ change. However, several commenters stated a 3% change is too small to have a meaningful impact and suggested the EPA consider a higher IQ percent change.

The EPA uses a variety of science policy approaches to select points of departure for developing regulatory values. For instance, in noncancer risk assessment the EPA often uses

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a percentage change in value. When assessing toxicological data, a 10 percent extra risk

(for discrete data), or a 1 standard deviation (i.e., 15 IQ points) change from the mean (for continuous data) is often used (USEPA, 2012). A smaller response to inform a POD has been applied when using epidemiological literature because there is an inherently more direct relationship between the study results and the exposure context and health endpoint.

Given the difficulty in identifying a response below which no adverse impact occurs when considering a continuous outcome in the human population, the EPA looked to its Benchmark Dose Guidance (2012) for insight regarding a starting point. Specifically, "[a] BMR of 1% has typically been used for quantal human data from epidemiology studies" (p. 21, USEPA, 2012). For the specific context of setting an MCLG for perchlorate, the EPA made a policy decision to evaluate valuated the level of perchlorate in water associated with a 1-percent? decrease, a 2-percent? decrease, and a 3 percent decrease in the mean population IQ (i.e., 1, 2 and 3 IQ points).

In evaluating the frequency and level of occurrence of perchlorate in drinking water the Agency EPA has found that perchlorate does not occur with frequency even at the lowest alternative MCLG of 18  $\mu$ g/L which is based upon avoiding a 1% change in IQ in the most sensitive breatagelife stage.

The Agency EPA received comments that the proposed MCLG did not incorporate an adequate margin of safety to comply with the SDWA. The Agency EPA disagrees that there was a failure it failed to use an adequate margin of safety. The Agency's EPA's assessment

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Commented [BC5]: Can you send me the source for this and the page number? Thanks.

Commented [BE6R5]: See page 21 [HYPERLINK "https://www.epa.gov/sites/production/files/2015-01/documents/benchmark\_dose\_guidance.pdf" ]
Also note this text was in the perchlorate proposal near the bottom of page 30356 [HYPERLINK "https://www.federalregister.gov/documents/2019/06/26/2019-12773/national-primary-drinking-water-regulations-perchlorate" ]

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focused upon the most sensitive subset of the population, specifically offspring whose
mothers had low (75 µg/day) iodine intake and were hypothyroximemic hypothyroximemic
(fT4 in the lowest 10th percentile of the population). In addition, to account for uncertainties
and to ensure the most sensitive subset of the population is protected within an adequate
margin of safety, a 3-fold uncertainty factor was applied to the proposed MCLG calculation
(USEPA, 2019a). More discussion on the uncertainty factor is presented in the section
"Consideration of Uncertainties."

The EPA received some comments stating the selection of the study for informing the relationship between maternal hormone levels (fT4) and IQ was inadequately described. Other comments supportsupported the Agency's EPA's study selection. The EPA concludes that selection of the Korevaar et al. (2016) study is appropriate because that study provides the most robust data available with a clear measure of neurodevelopment that can be expressed as a function of changing maternal fT4 exposure, which is necessary to the development of the model.

## BBDR and PBPK Models

The Agency PA received comments indicating the BBDR model was not transparent, scientifically valid, or based on robust data. The Agency PA disagrees. The model represents the best available peer reviewed science and will be best available data to inform a MCLG for perchlorate. The EPA does not believe there is a significant lack of transparency with respect to the assumptions related to the BBDR model. Appendix A of

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the EPA's Proposed MCLG Approaches report outlines the justification for all assumptions used in the development of the BBDR model (USEPA, 2019a). The EPA also disagrees with the assertion the BBDR model is far too uncertain to be relied upon as the basis for the derivation of the RfD. The EPA has used the best available science to calibrate the pharmacokinetic aspects of the BBDR model. The development of the BBDR model was performed in response to SAB recommendations and a model was deemed to be a more superiorefined approach to estimating a dose-response relationship between perchlorate exposure and maternal fT4 than anything that was available in the current scientific literature. The EPA disputes the claim that there are issues with the scientific validity of the BBDR model as the Agency conducted a peer review of the approach proposed and the reviewers stated the approach was "fit for purpose" to inform a MCLG for perchlorate (External Peer Reviewers for U.S. EPA, 2018, p. 2).

## Consideration of Uncertainties

The AgencyEPA received comments on the EPA's Agency's use of Uncertainty

Factors (UFs): with most commenters suggested suggesting that the EPA should consider a higher UF. The AgencyEPA thoroughly considered the application of UFs when deriving the RfDs and followed guidance presented in "A review of the reference dose and reference concentration processes" (USEPA, 2002). The Agency believes EPA concluded that the UFs are adequately justified and subsequently no changes have been made. Justification for each

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of the UFs described in this comment can be found in Section 11 of the Agency's MCLG

Derivation report (USEPA, 2019a).

The Asserted a UF of 3 for inter-individual variability because the Agency specifically modeled groups within the population that are identified as likely to be at greater risk and the adverse effects from perchlorate in drinking water (i.e., the fetus of the iodide deficient pregnant mother). The Assessy EPA selected model parameters to account for the most sensitive individuals in that group (i.e., muted TSH feedback, low fT4 values, lowiodine intake). As discussed in the MCLG Derivation report, the EPA has attempted to select the most appropriate inputs to protect the most sensitive population with an adequate margin of safety (USEPA, 2019a). The Agency EPA has determined that the selection of a UF of 3 for inter-individual variability is justified. As described in the MCLG Derivation report, because the output from the BBDR model is specific to the sensitive population and therefore the Agency has made no change in the EPA concluded that the UF of 3 is appropriate. In regards to variation in sensitivity among the members of the human population (i.e., interindividual variability), section 4.4.5.3 of the EPA guidance "A review of the reference dose and reference concentration process" (USEPA, 2002) document states, "In general, the Technical Panel reaffirms the importance of this UF, recommending that reduction of the intraspecies UF from a default of 10 be considered only if data are sufficiently representative of the exposure/dose-response data for the most susceptible subpopulation(s). Similar to the interspecies UF, the intraspecies UF can be considered to consist of both a toxicokinetic and

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toxicodynamic portion (i.e. 10^0.5 each)" (USEPA, 2002). Given that the BBDR model significantly accounts for differences within the human population, the full UF of 10 is not warranted.

NOAEL the extrapolation of the lowest-observed adverse effect level (LOAEL) to the noobserved-adverse-effect-level (NOAEL). LOAELs and NOAELs were not identified or used
by the EPA in its assessment because the Agency bas determined that the IQ employed a
sophisticated BBDR modeling approach, which was coupled with extrapolation to changes
presented as options in the 2019 proposalin IQ using linear regression, to determine a POD
that would not be expected to represent NOAELs an adverse effect. Therefore, including a
UF to account for extrapolating from a LOAEL to a NOAEL of 1 is not needed.
Additional appropriate. Other commenters suggested incorporating UFs for database
deficiencies. Based on the findings of the NRC report, the EPA has previously concluded that
this UF was not needed for deficiencies in the perchlorate database (NRC, 2005; USEPA,
2005a). The EPA believes that a UF of 1 to account for database deficiencies is still
appropriate given that the state of the perchlorate database has only increased since 2005.

## C. Occurrence Analysis

The EPA received comments suggesting that the revised UCMR 1 data did not provide an adequate estimate where the perchlorate occurrence in drinking water systems. Some commenters indicated that the age of the collected data rendered the occurrence

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analysis obsolete and overestimated, since it no longer captures current lower contamination
conditions which have been achieved due to mitigation measures taken in the Colorado
River Basin. Other commenters criticized the EPA for replacing UCMR 1 data with
compliance data for the States of California and Massachusetts.

The EPA recognizes that changes in perchlorate levels (increasing or decreasing) may have occurred in water systems since the UCMR 1 samples were collected between 2001 to 2005. The AgencyEPA updated the UCMR 1 data set to improve its accuracy in representing the current conditions for states that have enacted perchlorate regulations since the UCMR 1 monitoring was conducted. As outlined in the June 26, 2019 proposal, the EPA updated occurrence data for California and Massachusetts with current compliance data as reported by the states. Systems from these two states that were sampled during the UCMR 1 and that had reported perchlorate detections were updated with more recently measured values taken from current compliance monitoring data from Consumer Confidence Reports and state-level perchlorate compliance monitoring data to match corresponding water systems and entry points between the two sources.

The EPA has determined that the updated UCMR 1 data are the best available data collected in accordance with accepted methods on the frequency and level of perchlorate occurrence in drinking water on a national scale.

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## V. Conclusion

With this withdrawal of the 2011 perchlorate regulatory determination and final determination to not to regulate perchlorate, the EPA announces that there will be no NPDWR for perchlorate at this time. The EPA could consider re-listing perchlorate on the CCL and could proceed to regulation in the future if the occurrence or health risk information changes. As with other unregulated contaminants, the EPA sould address the will consider addressing limited instances of elevated levels of perchlorate by working with the affected system and state, as appropriate.

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#### MEMORANDUM

**SUBJECT:** Notice of Final Action on Perchlorate

(Tier 1 Action; SAN 5555; RIN 2040-AF28) - ACTION MEMORANDUM

FROM: David P. Ross

Assistant Administrator (4101M)

**THRU:** Office of Policy (1803A)

Office of Executive Secretariat (1105A)

**TO:** Andrew J. Wheeler

Administrator (1101A)

#### **PURPOSE**

Attached for your signature is the action titled "Notice of Final Action on Perchlorate."

On February 11, 2011, the EPA published a determination to regulate perchlorate in drinking water (76 FR 7762). On June 26, 2019 (84 FR 30524), the EPA published the proposed National Primary Drinking Water Regulation (NPDWR) for Perchlorate and requested public comments on multiple alternative actions, including withdrawing the Agency's 2011 determination to regulate perchlorate. The EPA received approximately 1,500 comments on the proposed rule.

In this notice, the EPA is withdrawing the 2011 Regulatory Determination and is making a final determination not to regulate perchlorate based on the Agency's consideration of public comments and the best available information.

### DEADLINE/TIMELINE

Section 1412(b)(1)(A) of the Safe Drinking Water Act (SDWA) requires the EPA to issue a proposed NPDWR within 24 months of the final regulatory determination and a final NPDWR within 18 months after the proposal. However, when the EPA consulted with the Science Advisory Board (SAB) regarding a planned methodology for deriving the maximum contaminant level goal (MCLG) for perchlorate, the Agency received recommendations to develop a physiologically based pharmacokinetic model (i.e., a biologically based dose-response model (BBDR)) to predict the effects of perchlorate exposure on thyroid function in pregnant women and their children, instead. The EPA collaborated with Food and Drug Administration scientists to perform the modeling recommended by the SAB and completed the analysis and associated peer reviews in March 2018. This delayed the EPA in proposing a NPDWR within 24 months.

In February 2016, the Natural Resources Defense Council (NRDC) filed a lawsuit for failure of

the EPA to perform its mandatory duties of proposing and finalizing a regulation for perchlorate in accordance with timelines provided in the SDWA. On October 18, 2016, the U.S. District Court for the Southern District of New York entered a Consent Decree, requiring the EPA to sign for publication a proposal for a MCLG and NPDWR for perchlorate in drinking water no later than October 31, 2018, and to sign for publication a final MCLG and NPDWR for perchlorate in drinking water no later than December 19, 2019. The Court later extended the deadline for the EPA to propose a MCLG and NPDWR for perchlorate in drinking water to May 28, 2019, and extended the date for signature of a final MCLG and NPDWR no later than June 19, 2020.

In compliance with the deadline established in the Consent Decree, on May 23, 2019, the Administrator signed a proposed rulemaking notice seeking public comment on a range of options regarding the regulation of perchlorate in public drinking water systems. The EPA published the proposed rule in the *Federal Register* on June 26, 2019. The public comment period for the proposal ended on August 26, 2019, and the EPA received approximately 1,500 comments.

#### DESCRIPTION of the ATION

Perchlorate is an inorganic anion that occurs naturally. It is also manufactured as an oxidizer for rockets, missiles, and fireworks and can be an impurity in hypochlorite disinfectants. The public may be exposed to perchlorate through food and drinking water. At certain levels, perchlorate can prevent the thyroid gland from getting enough iodine, which can affect thyroid hormone production. For pregnant women with low iodine levels, sufficient changes in thyroid hormone levels may cause changes in the child's brain development. For infants, changes to thyroid hormone function can also impact brain development.

The SDWA sets forth three criteria that must be met for the EPA to issue a MCLG and promulgate a NPDWR. Specifically, the EPA must determine that (1) "the contaminant may have an adverse effect on the health of persons;" (2) "the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems with a frequency and at levels of public health concern;" and (3) "in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by public water systems" (SDWA 1412(b)(1)(A)).

In the attached notice, the EPA concludes that, based on new data and the Agency's analysis since the issuance of the 2011 Regulatory Determination, perchlorate does not in fact meet the statutorily prescribed criteria for regulation. The new data and analysis indicate that perchlorate does not occur in public water systems with a frequency and at levels of public health concern. Specifically, the new peer-reviewed health effects analysis resulted in the health based proposed MCLG and proposed alternative MCLGs for perchlorate that are higher concentrations in drinking water (18 – 90  $\mu$ g/L) than the concentrations that the EPA considered to be levels of public health concern in the Agency's analysis for the determination to regulate in 2011 (1-47  $\mu$ g/L). In addition, the updated occurrence analysis shows that the frequency of occurrence of perchlorate in public water systems at levels exceeding any of the alternative proposed MCLGs (0.38% - 0.02%) is significantly lower than the frequency considered in the EPA's analysis for the 2011 Regulatory Determination (4% - 0.39%). Based on this information, the

EPA is announcing the Agency's conclusion that perchlorate does not occur in public water systems "with a frequency...of public health concern" and, therefore, regulation of perchlorate does not present a "meaningful opportunity for health risk reduction for persons served by public water systems" as required for regulation under the SDWA. Accordingly, perchlorate no longer meets the statutory criteria for regulation because the EPA does not have the authority to issue a MCLG or promulgate a NPDWR for perchlorate.

Therefore, the EPA is not issuing a final MCLG or NPDWR for perchlorate. However, the LiPA maintains the authority to re-list perchlorate on future Contaminant Candidate Lists and proceed with regulating perchlorate in the future if occurrence or risk information changes. The EPA will consider addressing limited instances of elevated levels of perchlorate by working with the affected system and state, as appropriate.

## STAKEHOLDER INVOLVEMENT and ANTICIPATED RESPONSE

The EPA considered the approximately 1,500 comments that were submitted on the proposed regulation. The EPA also consulted with the National Drinking Water Advisory regarding the proposed regulation. The EPA expects a variety of reactions and responses from stakeholders. The NRDC will likely sue the EPA for failure to comply with the Consent Decree and will likely challenge the Agency's authority to withdraw a Regulatory Determination. Officials from the States of California and Massachusetts, public health groups and environmental groups will likely state that a low perchlorate maximum contaminant level is needed to protect children's health. Industry groups, including the American Water Works Association, the Perchlorate Study Group, the American Chemistry Council, and the U.S. Chamber of Commerce will support the decision not to regulate perchlorate in drinking water. These groups will agree with the EPA's determinations that perchlorate does not occur frequently at levels of public health concern and there is not a meaningful opportunity for health risk reduction for persons served by public water systems.

## INTERNAL DEVELOPMENT and REVIEW PROCESS

The attached notice reflects the direction provided by the Administrator in the Options Selection meetings held on January 9 and March 18, 2020. The Office of Water (OW) convened a Final Agency Review meeting for this action on May 7, 2020. The following offices concurred without comment: the Office of Research and Development, the Office of Land and Emergency Management, the Office of Air and Radiation, and the Office of Chemical Safety and Pollution Prevention. The following offices concurred with comment: the Office of General Counsel (OGC), the Office of Policy (OP), and the Office of Children's Health Protection (OCHP). The OW has incorporated revisions identified in the comments from the OGC. The OW has also incorporated most of the suggested revisions identified by the OP, the key exception being that we are not incorporating OP's recommendation to not list the cost benefit analysis as a factor in the decision to withdraw the regulatory determination. The OW has worked with OGC to incorporate language that clarifies that this does not set a precedent for future regulatory determinations. The OW is not incorporating the majority of recommendations made by the OCHP, which address the health effects and occurrence analysis and are issues we have evaluated previously, including in response to OCHP's input on the proposal and in response to public comments.

Commented [BC1]: Is this statement in the final action? Why are we including it here?

Commented [BE2R1]: There is similar language in conclusion section. However it is not necessary to include in the Action memo

## INTERAGENCY REVIEW

The Office of Management and Budget initiated review of the *Federal Register* notice: "Notice of Final Action on Perchlorate" on [date placeholder].

## PEER REVIEW

For the proposed rulemaking, the OW followed the EPA's Peer Review Handbook and Agency policy titled "Conflicts of Interest Review Process for Contractor-Managed Peer Reviews of EPA HISA and ISI Documents" when conducting the peer review of models used to derive the proposed MCLGs for perchlorate. The EPA convened an independent peer review panel to evaluate the BBDR models in 2017 and a second, expert peer review panel in 2018 to evaluate the update of the BBDR model and approaches to link the BBDR model output to neurodevelopment endpoints in epidemiology studies to derive an MCLG. The EPA also sought input from the SAB, as required by the SDWA, prior to developing the proposed MCLGs.

## RECOMMENDATION

I recommend that you sign the attached *Federal Register* notice titled "Notice of Final Action on Perchlorate."

Attachments (2)

#### Message

From: Miller, Wynne [Miller.Wynne@epa.gov]

**Sent**: 2/6/2019 2:06:37 PM

To: Bertrand, Charlotte [Bertrand.Charlotte@epa.gov]; Beck, Nancy [Beck.Nancy@epa.gov]; Morris, Jeff

[Morris.Jeff@epa.gov]

CC: Keigwin, Richard [Keigwin.Richard@epa.gov]; Messina, Edward [Messina.Edward@epa.gov]; Dunn, Alexandra

[dunn.alexandra@epa.gov]; Baptist, Erik [Baptist.Erik@epa.gov]; Dinkins, Darlene [Dinkins.Darlene@epa.gov]

**Subject**: FW: Technical Support Documents - Draft Perchlorate Rule Proposal

Attachments: 181129\_DRAFT MCLG TSD.DOCX; Draft Perchlorate HRRCA 2019-02-03.docx;

PerchlorateOccMonitoringReport Updated 2-1-19.docx; Perchlorate T&C November 2018.docx; Perchlorate BAT

and SSCT November 2018.docx; WA459\_T4\_Vol1 Main Report\_181113.docx

Per my previous note ... here's the second email with the support documents for perchlorate.

From: Hernandez-Quinones, Samuel

Sent: Monday, February 04, 2019 3:55 PM

To: Johnson, Ann <Johnson.Ann@epa.gov>; Dockins, Chris <Dockins.Chris@epa.gov>; Shao, Nicole

<Shao.Nicole@epa.gov>; Miller, Gregory <Miller.Gregory@epa.gov>; Flowers, Lynn <Flowers.Lynn@epa.gov>; Foster,

Stiven <Foster.Stiven@epa.gov>; Kyprianou, Rose <Kyprianou.Rose@epa.gov>; Miller, Wynne

<Miller.Wynne@epa.gov>; Raffaele, Kathleen <raffaele.kathleen@epa.gov>
Cc: Christ, Lisa <Christ.Lisa@epa.gov>; Khera, Rajiv <Khera.Rajiv@epa.gov>
Subject: Technical Support Documents - Draft Perchlorate Rule Proposal

## Preliminary draft for internal EPA review

Hi,

Attached are copies of the draft technical support documents that were used as basis for the Perchlorate Rule Proposal. As mentioned earlier today we are not requesting comments on these documents within the two-week review time frame.

Economic Analysis
MCLG Derivation Document
Treatment Technologies & Costs Document
Occurrence & Monitoring Report
MCLG Approaches Document
Best Available Technologies Document

Thanks Sam

Samuel Hernández Quiñones, P.E.

Environmental Engineer

Office of Water

Environmental Protection Agency 1200 Pennsylvania Ave. NW

Washington, DC 20460

202-564-1735

"USEPA Protecting Human Health and the Environment"

From: Hernandez-Quinones, Samuel

Sent: Monday, February 04, 2019 10:32 AM

To: Cc: Christ, Lisa <a href="mailto:christ.lisa@epa.gov">christ.lisa@epa.gov">christ.lisa@epa.gov</a>; Burneson, Eric <a href="mailto:burneson.eric@epa.gov">christ.lisa@epa.gov</a>; Burneson, Eric <a href="mailto:burneson.eric@epa.gov">christ.lisa@epa.gov</a>; Messier, Dawn@epa.gov</a>; Wehling, Carrie@epa.gov</a>; Huff, Lisa <a href="mailto:huff.Lisa@epa.gov">huff.Lisa@epa.gov</a>; Khera, Rajiv <a href="mailto:khera.Rajiv@epa.gov">khera.Rajiv@epa.gov</a>>

Subject: Request for review - Draft Perchlorate Rule Proposal

## Preliminary draft for internal EPA review

Hi,

Attached for your review is the Draft Perchlorate Rule Proposal (Preamble and Regulatory Text). As discussed previously we are asking that you provide your questions and or comments on the draft document to me by COB 2/15/2019. In order to keep with the consent decree deadline, we need to complete this step of the process within two weeks, if there are any issues that require technical discussions please let me know so that we can arrange any necessary working meeting.

I will follow-up with another email later today that contains the draft version of the technical support documents that were used as the basis for the Perchlorate proposal. Since those are very lengthy documents we are not requesting that you submit input and/or comments on those within the two- week time frame.

Please let me know if you have any questions.

Thanks Sam

Samuel Hernández Quiñones, P.E. Environmental Engineer Office of Water Environmental Protection Agency 1200 Pennsylvania Ave. NW Washington, DC 20460 202-564-1735

"USEPA Protecting Human Health and the Environment"



# DRAFT Technical Support Document: Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water

NOTICE	
This document is a <b>review draft.</b> It has not been formally disseminated by the and should not be construed to represent any Agency determination or policy public comment only.	ne EPA. It does not represent v. It is being circulated for
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## **Table of Contents**

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## **List of Tables**

[ TOC \h \z \c "Table" ]

## **List of Figures**

[ TOC \h \z \c "Figure" ]

# **Abbreviations and Acronyms**

Agency U.S. Environmental Protection Agency

Approaches Report Proposed Approaches to Inform the Derivation of a Maximum Contaminant

Level Goal for Perchlorate in Drinking Water

ATA American Thyroid Association

BBDR biologically based dose-response

BW bodyweight

CalEPA California Environmental Protection Agency

CBCL Child Behavior Checklist CCL Contaminant Candidate List

CI confidence interval

CSFII Continuing Survey of Food Intakes by Individuals

DWI drinking water ingestion

EFH Exposure Factors Handbook

EPA U.S. Environmental Protection Agency

FDA Food and Drug Administration

fT4 free thyroxine

GW gestational week

HPT hypothalamic-pituitary-thyroid

HRL Health Reference Level

IQ intelligence quotient

IRIS Integrated Risk Information System

LOAEL lowest-observed adverse effect level

LOD level of detection

MCL Maximum Contaminant Level
MCLG Maximum Contaminant Level Goal
MDI Mental Developmental Index

NAAQS National Ambient Air Quality Standards

NHANES National Health and Nutrition Examination Survey

NIS sodium-iodide symporter
NOAEL no observed adverse effect level
NOAL no observed effect level

NPDWR National Primary Drinking Water Regulation

NRC National Research Council

OHAT Office of Health Assessment and Translation

PBPK physiologically based pharmacokinetic

PD pharmacodynamic

PDI Psychomotor Development Index

POD point of departure

RfD reference dose ROB Risk of Bias

RSC relative source contribution

SAB Science Advisory Board SD standard deviation SDWA Safe Drinking Water Act

T3 triiodothyronine
T4 thyroxine
TDS Total Diet Study

THOP transient hypothyroxinemia of prematurity

TSH thyroid stimulating hormone

UF uncertainty factor

 $\begin{array}{lll} UF_A & Uncertainty factor, animal-to-human \\ UF_D & Uncertainty factor, database deficiency \\ UF_H & Uncertainty factor, within-human variability \\ UF_L & Uncertainty factor, LOAEL-to-NOAEL \\ UF_S & Uncertainty factor, subchronic-to-chronic \\ \end{array}$ 

WHO World Health Organization
WWEIA What We Eat in America

# Acknowledgements

This document was prepared by the Environmental Protection Agency, Office of Ground Water and Drinking Water. Abt Associates Inc. provided assistance in performing the analyses detailed in this document.

#### 1. Introduction

The purpose of this document is to provide the U.S. Environmental Protection Agency's (EPA or the Agency) justification for the selected approach to inform the Maximum Contaminant Level Goal (MCLG) for perchlorate in drinking water. MCLGs are non-enforceable, health-based goals that the EPA sets for each regulated drinking water contaminant. In accordance with the Safe Drinking Water Act (SDWA), the EPA sets MCLGs at a level at which no known or anticipated adverse effects on the health of persons occur, and which allows an adequate margin of safety. SDWA requires that the EPA use the best available, peer reviewed science and supporting studies conducted in accordance with accepted practices; and data collected by accepted methods or best available methods. The process of setting an MCLG considers only public health, and the limits of analytical measurement and treatment technology effectiveness are not taken into account. The SDWA requires that the EPA establish the enforceable Maximum Contaminant Level (MCL) as close as feasible to the MCLG, taking costs into consideration. SDWA also requires that the EPA prepare a Health Risk Reduction Cost Analysis that includes an estimate of the costs, benefits, and an assessment of effects of the contaminant on the general population and on groups within the population such as infants, pregnant women, the elderly or other subpopulations that are likely to be at greater risk of adverse health effects.

The perchlorate MCLG will be based on analyses presented in the peer-reviewed report, *Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water* (hereafter referred to as the "Approaches Report") [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</author><Year>2018</year><RecNum>1995</RecNum><DisplayText>(U.S. EPA, 2018)</DisplayText><record><rec-number>1995</rec-number><forcign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1527619441">1995</key></forcign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author

The Approaches Report presents a Biologically Based Dose Response (BBDR) model that predicts thyroid hormone changes that may result from perchlorate exposure in pregnant women. The Approaches Report also describes the EPA's analysis of epidemiologic studies that allow for a connection between the thyroid hormone changes in pregnant women predicted by the BBDR model and neurodevelopmental effects in their offspring.

## 1.1 Evaluation of Perchlorate under the Safe Drinking Water Act

The EPA included perchlorate on the first three iterations of the Contaminant Candidate List (CCL), starting in 1998, after data collection and requests for comments. The CCL is a list of drinking water contaminants that are known or anticipated to occur in public water systems and are not currently subject to the EPA drinking water regulations. Contaminants listed on the CCL may require future regulation under the SDWA. The EPA uses the CCL to identify priority contaminants for regulatory decision-making and information collection.

In 2005, at the request of the EPA and other federal agencies, the National Research Council (NRC) evaluated the health implications of perchlorate ingestion. The NRC concluded that perchlorate exposure

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Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water
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could inhibit the transport of iodide into the thyroid, leading to thyroid hormone deficiency [ ADDIN
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uptake results in intra-thyroid iodide deficiency, decreased synthesis of key thyroid hormones
(triiodothyronine (T3) and thyroxine (T4)), and increased thyroid stimulating hormone (TSH). The NRC
also concluded that a prolonged decrease of thyroid hormones is potentially more likely to have adverse
effects in sensitive populations (e.g., people with thyroid disorders, pregnant women, fetuses, and
infants). Based on NRC's recommendations, the EPA adopted a reference dose (RfD) of 0.7 μg/kg/day in
2005 [ ADDIN EN.CITE < EndNote > < Cite > < Author > U.S.
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7 μg/kg/day) identified by a study [ ADDIN EN.CITE
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In healthy adults for the inhibition of radioactive iodide uptake and the application of an uncertainty factor of 10 for intraspecies variability.

In October 2008, the EPA published a preliminary determination not to regulate perchlorate in drinking water using a Health Reference Level (HRL) of 15  $\mu$ g/L, which was derived from the RfD of 0.7  $\mu$ g/kg/day, using a default bodyweight of 70 kg, a default drinking water consumption rate of 2 L/day, and a perchlorate-specific relative source contribution (RSC) of 62 percent for a pregnant woman [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

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size="100%">https://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=P1004X7Q.TXT</style></url></related-urls></url></record></EndNote>]. The RSC is the percentage of the RfD remaining for drinking water after other sources of exposure to perchlorate (e.g., food) have been considered. In December 2008, the EPA issued an interim health advisory (15 μg/L perchlorate in drinking water) to provide guidance to state and local officials in their efforts to address this issue. In August 2009, the EPA published a supplemental request for comment with a new analysis that derived potential alternative HRLs for 14 life stages, including infants and children. The analysis used the RfD of 0.7 μg/kg/day and the life stage-specific bodyweight and exposure information (i.e., drinking water intake, RSC) [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

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After considering comments on the October 2008 and August 2009 notices, the EPA made a final determination to regulate perchlorate in drinking water in February 2011 [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2011</Year><RecNum>69</RecNum><DisplayText>(U.S. EPA, 2011a)</DisplayText><record><rec-number>69</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1438181138">69</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>

In accordance with the SDWA, the Agency requested the EPA Science Advisory Board (SAB) to provide recommendations for how to consider available data in deriving an MCLG for use in developing a NPDWR for perchlorate. The EPA presented the SAB with a description of the general approach used by the Agency to derive an MCLG for non-carcinogenic chemicals. This approach relies on the following equation:

$$MCLG\left(\frac{\mu g}{L}\right) = \frac{RfD \times BW}{DWL} \times RSC$$

where:

RfD = RfD of the contaminant ( $\mu g/kg/day$ ) (in the initial assessment for perchlorate, the EPA used NRC's recommended RfD of 0.7  $\mu g/kg/day$ );

BW = bodyweight (kg); and

DWI = drinking water ingestion rate (L/day).

The EPA also presented information to the SAB about life-stage considerations, physiologically based pharmacokinetic modeling, and epidemiologic and biomonitoring studies. In 2013, the SAB recommended the following:

- Derive a perchlorate MCLG that addresses sensitive life stages through physiologically based pharmacokinetic/pharmacodynamic (PBPK/PD) modeling;
- Expand the modeling approach to account for thyroid hormone perturbations and potential adverse neurodevelopmental outcomes from perchlorate exposure;
- Utilize a mode-of-action framework for developing the MCLG that links the steps in the proposed mechanism leading from perchlorate exposure through iodide uptake inhibition to thyroid hormone changes and finally to neurodevelopmental impacts; and
- "Extend the [BBDR] model expeditiously to...provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (SAB, 2013, p. 19).

The SAB's framework incorporates the endpoint of iodide uptake inhibition that was the basis for the RfD as part of this broader and more comprehensive framework that links perchlorate exposure to adverse neurodevelopmental outcomes.

In addition, the SAB stated that the EPA should more directly consider thyroid hormone changes as relevant to sensitive life stages; specifically, fetuses of hypothyroxinemic pregnant women and infants and neonates exposed to perchlorate through either water-based formula preparations or the breast milk of lactating women [ ADDIN EN.CITE

<EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><DisplayText >(SAB, 2013)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreignkeys><ref-type name="Government Document">46</ref-

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<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayTe xt>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-date>National Research Council</modified-date></record></Cite></EndNote>]. The SAB approach focuses on the subtler changes in thyroid hormones (specifically fT4) that are associated with maternal hypothyroxinemia rather than the broader changes in thyroid hormones (both fT4 and TSH) that are associated with hypothyroidism[]. Furthermore, the SAB recommended that the EPA consider available data on potential adverse health effects (i.e., neurodevelopmental outcomes) due to thyroid hormone level perturbations, regardless of the cause of those perturbations [ ADDIN EN.CITE

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<sup>&</sup>lt;sup>1</sup> The serum-free thyroxine (fT4) value is at the lower end of the population reference range with levels of serum TSH within the population reference range, as opposed to hypothyroidism, which consists of serum fT4 levels below and levels of serum TSH above the respective reference ranges. For the purposes of this report, fT4 and TSH values will be assumed to be measured via the serum unless otherwise stated.

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type><contributors><author>SAB,</author></author></author>><author>SAB,</author></author>><author>><author>>U.S. Environmental Protection Agency,</author></secondary-authors></contributors><titles><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><quel><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author

To address these recommendations, the EPA revised existing biologically based dose-response (BBDR) and PBPK/PD models [ ADDIN EN.CITE

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J.W.</author></authors></contributors><titles><title>Evaluation of perturbations in serum thyroid hormones during human pregnancy due to dietary iodide and perchlorate exposure using a biologically based dose-response model</title><secondary-title>Toxicological Sciences</secondary-title></title></periodical><full-title>Toxicological Sciences</full-title></periodical><pages>320-341</pages><volume>133
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EPA,</author></authors></contributors><titles><title>Inhibition of the sodium-iodide symporter by perchlorate: an evaluation of lifestage sensitivity using physiologically based pharmacokinetic (PBPK) modeling (Final Report)</title></title></dates><year>2009</year></dates><pub-location>Washington, D.C. </pub-location><isbn>EPA/600/R-08/106A</isbn><urls></urls></record></EndNote>] to create a BBDR model that predicts changes in thyroid hormone (i.e., T4, fT4, and T3) levels as a result of nutritional iodine intake and perchlorate exposure in women prior to pregnancy and early gestation. The EPA originally developed a set of BBDR models that included all sensitive life stages identified by the SAB (i.e., the fetus (by modeling a pregnant mother at 40 gestational weeks (GWs)), neonates, and infants (SAB, 2013)), with the pregnancy model representing the third trimester. These models were peer reviewed in January 2017.² Reviewers stressed the importance of developing an early pregnancy model when considering adverse neurodevelopmental impacts.

<sup>&</sup>lt;sup>2</sup> For a more detailed discussion of the development of the draft model, its calibration, and dose-response evaluations, please see the draft BBDR model report entitled, *Biologically Based Dose Response Models for the Effect of Perchlorate on Thyroid Hormones in the Infant, Breast Feeding Mother, Pregnant Mother, and Fetus: Model Development, Revision, and Preliminary Dose-Response Analysis.* The report is available through the docket at [ HYPERLINK "http://www.regulations.gov" ] (Docket ID No. EPA-HQ-OW-2016-0439).

The EPA considered the recommendations from the January 2017 peer review and focused on those that were anticipated to be most important for increasing the scientific rigor of the model and modeling results.<sup>3</sup> Model revisions focused on the following key recommendations:

- Extending the model to early pregnancy;
- Incorporating biological feedback control of hormone production via TSH signaling, such that the model can describe lower levels of iodide nutrition;
- Calibrating the model and evaluating its behavior for upper and lower percentiles of the population, as well as the population median; and
- Conducting an uncertainty analysis for key parameters.

Consequently, the MCLG is developed to protect the fetuses of hypothyroxinemic women early in pregnancy. The EPA carried out a subsequent peer review in January 2018 to evaluate updates to the BBDR model and presented several approaches to link the revised perchlorate BBDR model predictions to neurodevelopmental effects. The January 2018 peer review was largely supportive of the efforts described in the Draft Approaches Report, as evidenced by the following from the peer-reviewed final report:

The panel commends EPA for the substantial amount of work done in creating the new modeling and preparing the report under review. It was highly responsive to the review comments from a year ago. Overall, the panel agreed that the EPA and its collaborators have prepared a highly innovative state-of-the-science set of quantitative tools to evaluate neurodevelopmental effects that could arise from drinking water exposure to perchlorate. While there is always room for improvement of the models, with limited additional work to address the committee's comments [in the peer-reviewed report], the current models are fit-for-purpose to determine an MCLG.

The purpose of the Draft Approaches Report was to "present the revisions to the BBDR model and alternative approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects in response to the prior peer reviews" (U.S. EPA, 2018, p. xi). Additionally, the Draft Approaches Report connected perchlorate doses (from any source) to neurodevelopmental outcomes. The document did not derive an MCLG; rather, it presented alternative approaches that might be used to inform future decisions, including the derivation of an MCLG in accordance with the SDWA.

### 1.2 Overview of this Report

The remainder of this document describes the approach and inputs the EPA deemed most appropriate to inform an MCLG for perchlorate. The approach to inform the MCLG is a multistep process with many inputs and decision points that will be laid out in this document. In addition, this document provides background information on perchlorate including it chemical and physical properties (Section 2), common uses and environmental fate (Section [ REF \_Ref529977182 \r \h ]), occurrence of perchlorate in drinking water (Section 4), and occurrence of perchlorate in food or other sources (Section [ REF

<sup>&</sup>lt;sup>3</sup> Model revisions are summarized in Section 3 of the Approaches Report, with additional detail provided in Appendix A.

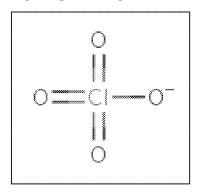
\_Ref529977466 \r \h ]). To put the MCLG analysis into context, an overview of thyroid hormone physiology is provided along with a description between thyroid hormones and neurodevelopment (Section 6). Section 7 provides a brief overview of perchlorate's mode of action and the evidence evaluating perchlorate's impact on the thyroid. Section [ REF \_Ref530556732 \r \h ] of this report describes the sensitive life stages identified by the SAB and evaluates if protecting the fetuses of hypothyroxinemic pregnant women would also be protective of breast- and formula-fed infants.

Section [ REF \_Ref530556719 \r \h ] is where EPA begins to describe the approach of deriving an MCLG for perchlorate, starting with the selection of an approach to inform the point of departure (POD) for perchlorate (i.e., the selected key parameters for the BBDR model runs and the key epidemiological data selected to inform the MCLG are described). Section [ REF \_Ref515367288 \r \h \\* MERGEFORMAT ] then describes the EPA's selection of the benchmark response to identify the POD from which an RfD can be derived and Section [ REF \_Ref515367295 \r \h \\* MERGEFORMAT ] goes on to describe how to translate the identified POD to an RfD by considering and applying appropriate uncertainty factors. Section [ REF \_Ref515367301 \r \h \\* MERGEFORMAT ] describes the process to convert the RfD to a drinking water concentration by utilizing population-specific exposure factors and the relative source contribution of perchlorate in drinking water, which, ultimately will be deemed the MCLG.

## 2. Identity: Chemical and Physical Properties

Perchlorate is an inorganic compound containing one chlorine atom bound to four oxygen atoms in a tetrahedral configuration (See [ REF \_Ref528589635 \h ]). As such, perchlorate (ClO<sub>4</sub>-) is an anion that forms salts with most cations. Commonly used perchlorate salts include ammonium perchlorate and potassium perchlorate. Other perchlorate compounds include: sodium perchlorate, aluminum perchlorate, hydrozylammonium perchlorate, lithium perchlorate, magnesium perchlorate, nitronium perchlorate, and perchloric acid.

Figure [ SEQ Figure \\* ARABIC ]. Chemical Structure of Perchlorate



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The chemical structure of perchlorate is shown above ([ REF \_Ref528589635 \h ]). Its physical and chemical properties and Chemical Abstracts Service (CAS) registry numbers for the most common forms of perchlorate are listed in [ REF \_Ref528589694 \h ].

Table [ SEQ Table \\* ARABIC ]. Chemical and Physical Properties of Perchlorate

Perchlorate and Its Common Salts					
	Perchlorate	Ammonium perchlorate	Potassium perchlorate	Magnesium perchlorate	Sodium perchlorate
CAS number	14797-73-0	7790-98-9	7778-74-7	10034-81-8	7601-89-0
Molecular Formula	ClO <sub>4</sub> -	NH <sub>4</sub> ClO <sub>4</sub>	KCIO <sub>4</sub>	Mg(ClO <sub>4</sub> ) <sub>2</sub>	NaClO <sub>4</sub>

Physical and Chemic	al Properties				
Boiling Point	-	-	400 °C ¹	-	-
Melting Point	-	439 °C ²	525 °C ⁴	250 °C 4	480 °C 4
Molecular Weight	99.45 g/mol <sup>a</sup>	117.49 g/mol <sup>b</sup>	138.55 g/mol <sup>a</sup>	223.20 g/mol <sup>d</sup>	122.4 g/mol <sup>a</sup>
Water Solubility	-	200 g/L @ 25°C°	15 g/L @ 25°C <sup></sup>	99 g/1000g @ 25°C°	209 g/100 g @ 25 °C °

#### Budavari [ ADDIN EN.CITE < EndNote > < Cite

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HSDB</Author><Year>2010</Year><RecNum>2232</RecNum><DisplayText>(2010)</DisplayText><record><recnumber>2232</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543327203">2232</key></foreign-keys><ref-type name="Web Page">12</ref-

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#### 3. Uses and Environmental Fate

This section summarizes information pertaining to the uses, manufacture, and environmental fate of perchlorate.

#### 3.1 Production, Use and Occurrence

Ammonium perchlorate is the form of perchlorate most widely used in the United States [ ADDIN EN.CITE

<EndNote><Cite><Author>Mendiratta</Author><Year>1996</Year><RecNum>2033</RecNum>< DisplayText>(Mendiratta, Dotson, & Brooker, 1996)</DisplayText><record><recnumber>2033</recnumber><foreign-keys><key app="EN" db-

authors></contributors><title>Perchloric acid and perchlorates</title><secondary-title>Kirk-Othmer encyclopedia of chemical technology. </secondary-title></title><pages>157-

170</pages><volume>18</volume><dates><year>1996</year></dates><pub-location>New York, NY</pub-location><publisher>John Wiley & Dons, Sons, Sons,

Inc.</publisher><urls></urls></record></Cite></EndNote>]. It is primarily used as an oxidizer in solid fuels to power rockets, missiles, and fireworks. According to Wang et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Wang</Author><Year>2002</Year><RecNum>2038</RecNum><Disp layText>(2002)</DisplayText><record><rec-number>2038</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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Occurrence</title></title></pages>Abstract</pages><dates><year>2002</year></dates><publocation>Denver, CO</pub-location><publisher>Awwa Research Foundation and American Water Works Association</publisher><urls></urls></record></Cite></EndNote>] approximately 90% of perchlorate is manufactured for this application. Data on the amount of ammonium perchlorate used in solid rocket boosters are not available, but the solid propellants used in the booster rockets in the space shuttle were approximately 70% ammonium perchlorate by weight [ADDIN EN.CITE <EndNote><Cite><Author>Conkling</Author><Year>1996</Year><RecNum>2039</RecNum><D isplayText>(Conkling, 1996)</DisplayText><record><rec-number>2039</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541693880">2039</key></foreign-keys><ref-type name="Encyclopedia">53</reftype><contributors><authors><author>Conkling, J.A.</author></author><secondaryauthors><author>Kroschwitz, J.I. and M. Howe-Grant.</author></secondaryauthors></contributors><title>Pyrotechnics</title><secondary-title>Kirk-Othmer encyclopedia of chemical technology</secondary-title></title></pages>680-697</pages><volume>20</volume><dates><year>1996</year></dates><pub-location>New York, NY</pub-location><publisher>John Wiley & Dons, Inc.</publisher><urls></urls></record></Cite></EndNote>]. Other perchlorates, such as potassium

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perchlorate, are used as oxidizers in solid rocket boosters [ ADDIN EN.CITE
<EndNote><Cite><Author>Lindner</Author><Year>1993</Year><RecNum>2040</RecNum><Dis
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125</pages><volume>10</volume><dates><year>1993</year></dates><pub-location>New York,
NY</pub-location><publisher>John Wiley & amp;
Sons</publisher><urls></urls></record></Cite></EndNote>].
Perchlorate is also generated during the manufacture and use of pyrotechnic devices such as fireworks
[ ADDIN EN.CITE
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location>Atlanta, GA</pub-location><publisher>U.S. Department of Health and Human
Services</publisher><urls></urls></record></Cite></EndNote>]. Potassium perchlorate and
ammonium perchlorate are the most commonly used salts in fireworks. Some fireworks components
may contain anywhere from 0-70% perchlorate [ ADDIN EN.CITE
<EndNote><Cite><Author>Sijimol</Author><Year>2014</Year><RecNum>2041</RecNum><Dis
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7210</pages><volume>186</volume><dates><year>2014</year></dates><urls></urls></record></
Cite></EndNote>]. Ammonium perchlorate is used in small amounts in gun powder [ ADDIN
EN.CITE
<EndNote><Cite><Author>Lindner</Author><Year>1993</Year><RecNum>2040</RecNum><Dis
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Othmer encyclopedia of chemical technology</secondary-title></title></secondary-title></title>

NY</pub-location><publisher>John Wiley & amp;

Sons</publisher><urls></urls></record></Cite></EndNote>]. Ammonium and potassium perchlorate are used in signal flares, such as road flares and signaling devices on ships [ ADDIN **EN.CITE** 

<EndNote><Cite><Author>Conkling</Author><Year>1996</Year><RecNum>2039</RecNum><D isplayText>(Conkling, 1996)</DisplayText><record><rec-number>2039</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541693880">2039</key></foreign-keys><ref-type name="Encyclopedia">53</reftype><contributors><authors><author>Conkling, J.A.</author></author></author> authors><author>Kroschwitz, J.I. and M. Howe-Grant.</author></secondaryauthors></contributors><title>Pyrotechnics</title><secondary-title>Kirk-Othmer encyclopedia of chemical technology</secondary-title></title></pages>680-697</pages><volume>20</volume><dates><year>1996</year></dates><pub-location>New York, NY</pub-location><publisher>John Wiley & Dons,

Inc.</publisher><urls></urls></record></Cite></EndNote>].

Perchlorate can also be present in lubricating oils, matches, aluminum refining, rubber manufacturing, paint and enamel manufacturing, leather tanning, paper and pulp processing (as an ingredient in bleaching powder), and as a dye mordant [ ADDIN EN.CITE <EndNote><Cite><Author>U.S. EPA</Author><Year>2006</Year><RecNum>2234</RecNum><DisplayText>(U.S. EPA, 2006)</DisplayText><record><rec-number>2234</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543332964">2234</key></foreignkeys><ref-type name="Report">27</ref-type><contributors><authors><author>U.S. EPA,</author></authors></contributors></title>Regulatory Determinations Support Document for Selected Contaminants from the Second Drinking Water Contaminant Candidate List (CCL2).</title></title><number>EPA Report 815-D-06-007</number><dates><year>2006</year></dates><pub-location>Office of Ground Water and Drinking Water, U.S. Environmental Protection Agency, Washington, DC</publocation><urls></urls></record></Cite></EndNote>]. Potassium perchlorate, mixed with metals such as iron or zirconium, has been used in heat pellets for the activation of reserve battery cells, and lithium and magnesium perchlorate have been used in batteries [ ADDIN EN.CITE <EndNote><Cite><Author>Cohen</Author><Year>1993</Year><RccNum>2042</RecNum><Disp layText>(Cohen, 1993)</br>

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York, NY</pub-location><publisher>VCH
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Publishers</publisher><urls></urls></record></Cite></EndNote>]. Perchlorate was also used to adjust the ionic strength of electroplating baths in plating razor blades [ ADDIN EN.CITE <EndNote><Cite><Author>Schilt</Author><Year>1979</Year><RecNum>2044</RecNum><Displ ayText>(Schilt, 1979)</DisplayText><record><rec-number>2044</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmedxe5vxfpkax2vzp0ftv29"

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A.A.</author></authors></contributors></title>Preparation and properties of perchloric acid. Perchloric acid and perchlorates. </title></title></e>

63 < / extion > < dates > < year > 1979 < / year > < / dates > < pub-location > Columbus, OH < / pub-location > Columbus, OH

location><publisher>The G. Frederick Smith Chemical

Company</publisher><urls></urls></cite></EndNote>]. It is not clear to what extent these other uses of perchlorate are still in practice.

During the 1950s and early 1960s, perchlorate was widely used in the treatment of hyperthyroidism, especially for patients with Graves' disease [ ADDIN EN.CITE <EndNote><Cite><Author>Von Burg</Author><Year>1995</Year><RecNum>2045</RecNum><DisplayText>(Von Burg, 1995)</DisplayText><record><rec-number>2045</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541695707">2045</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Von Burg, R.</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></author></a></author></a></author></a></author></a></author></author></a></author></a></author></a></al></arthor></a></al></arthor></a></al></arthor></a></al></arthor></a></al></arthor></a></al>

241</pages><volume>15</volume><dates><year>1995</year></dates><urls></record></Cit e></EndNote>]. Perchlorate is currently used in the U.S. to block radioactive technetium in the thyroid and other glands during medical imaging of the brain, blood, and placenta [ ADDIN EN.CITE <EndNote><Cite><Author>Gibbs</Author><Year>1998</Year><RecNum>2046</RecNum><Displ ayText>(Gibbs et al., 1998)</DisplayText><record><rec-number>2046</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamn="1541695841">2046</key></foreign-keys><ref-type name="Journal Article">17</ref-

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Leveille</author><author>J.E. Findley</author><author>M.

Francis</author></authors></contributors><titles><title>Evaluation of a population with occupational exposure to airborne ammonium perchlorate for possible acute or chronic effects on thyroid function</title><secondary-title>J. Occup. Environ. Med. </secondary-

title></titles><periodical><full-title>J. Occup. Environ. Med.</full-title></periodical><pages>1072-1082</pages><volume>40</volume><number>12</number><dates><year>1998</year></dates><ur

ls></urls></record></Cite></EndNote>] and as part of a treatment to counter the effects of the drug amiodarone on the thyroid [ ADDIN EN.CITE

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Braverman</author></authors></contributors><title>The effects of amiodarone on the thyroid</title><secondary-title>Endocr. Rev.</secondary-title></title><periodical><full-title>Endocr. Rev.</full-title></periodical><pages>240-

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Rajagopalan</author></authors></contributors></title>The origin of naturally occurring perchlorate: The role of atmospheric processes</title><secondary-title>Environ. Sci.

Technol.</secondary-title></title><periodical><full-title>Environ. Sci. Technol.</full-title></periodical><pages>1569-

1575</pages><volume>39</volume><number>6</number><dates><year>2005</year></dates><url s></urls></record></Cite></EndNote>] detected perchlorate in rain and snow samples, and showed that perchlorate is formed by a variety of simulated atmospheric processes, suggesting that natural, atmospherically-derived perchlorate is present in the environment. This study suggested that perchlorate can be produced in the natural environment by heterogeneous reactions of chlorine compounds with ozone or UV light and by electrical discharge in the air. In a series of experiments, Kang *et al.* [ ADDIN EN.CITE <EndNote><Cite

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Anderson</author><author>W.A.

Jackson</author></authors></contributors><titles><title>Photochemical formation of perchlorate from aqueous oxychlorine anions</title><secondary-title>Anal. Chim. Acta </secondary-title></periodical><pages>48-56</pages><volume>567</volume><number>1</number><dates><year>2006</year></dates><urls></urls></record></Cite></EndNote>] exposed various chlorine intermediates to UV radiation and produced perchlorate in aqueous salt solutions with initial concentrations of hypochlorite, chlorite, or chlorate between 100 and 10,000 mg/L. Rao *et al.* [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1">EndNote><RecNum>1463</RecNum><Displa</pre>

yText>(2007)</DisplayText><record><rec-number>1463</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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A.</author><author>Rajagopalan,

S.</author><author>Sandvig, R. M.</author><author>Scanlon, B. R.</author><author>Stonestrom,

D. A.</author><author>Walvoord, M. A.</author><author>Jackson, W.

A.</author></authors></contributors><titles><title>Widespread natural perchlorate in unsaturated zones of the southwest United States</title><secondary-title>Environmental Science and Technology</secondary-title>Environ Sci Technol</alt-title>Environmental Science and Technology</short-title></title><periodical><full-title>Environmental Science and Technology</full-title><abbr-1>Environ Sci Technol</abbr-1></periodical><full-title>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1>Environ Sci Technol</abbr-1></alt-periodical><pages>4522-

5851</isbn><label>2140862</label><urls></urls><language>English</language></record></Cite></EndNote>] showed that a substantial reservoir of natural perchlorate is present in the vadose (unsaturated) zone of arid and semiarid areas of the U.S., which is capable of producing high levels of perchlorate in groundwater.

Chile possesses caliche ores rich in sodium nitrate (NaNO<sub>3</sub>), which are also a natural source of perchlorate. These Chilean nitrate salts (saltpeter) have been mined and refined to produce commercial fertilizers, which (before 2001), accounted for about 0.14 percent of U.S. fertilizer application [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2001</Year><RecNum>2235</RecNum><DisplayText>(U.S. EPA, 2001)</DisplayText><record><rec-number>2235</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543335304">2235</key></foreign-keys><ref-type name="Report">27</ref-type><contributors><author>U.S.

EPA,</author></authors></contributors><title>Survey of Fertilizers and Related Materials for Perchlorate (ClO4): Final Report</title></title><volume>EPA/600/R-

01/049</volume><dates><year>2001</year></dates><pub-location>Office of Research and Development, U.S. Environmental Protection Agency, Cincinnati, OH</pub-

location><urls></urls></record></Cite></EndNote>]. The U.S. EPA [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>U.S.

EPA</Author><Year>2001</Year><RecNum>2235</RecNum><DisplayText>(2001)</DisplayText><record><rec-number>2235</rec-number><foreign-keys><key app="EN" db-

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EPA,</author></authors></contributors><title>Survey of Fertilizers and Related Materials for Perchlorate (ClO4): Final Report</title></title></tole>

01/049</volume><dates><year>2001</year></dates><pub-location>Office of Research and Development, U.S. Environmental Protection Agency, Cincinnati, OH</pub-

location><urls></urls></record></EndNote>] conducted a broad survey of fertilizers and other raw materials and found that all products surveyed were devoid of perchlorate except for those known to contain or to be derived from mined Chilean saltpeter.

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Perchlorate production began in the U.S. with the production of magnesium perchlorate in 1928 for
use as a desiccant [ ADDIN EN.CITE
<EndNote><Cite><Author>GFS</Author><Year>1997</Year><RecNum>2055</RecNum><Displa
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Chemical Company Catalog</title></title></dates><year>1997</year></dates><pub-
location>Powell, OH</pub-location><publisher>GFS Chemicals,
Inc.</publisher><urls></urls></record></Cite></EndNote>]. The total worldwide production of
perchlorate was less than 3.6 million pounds until World War II; after World War II, annual
perchlorate production increased to 36 million pounds [ ADDIN EN.CITE
<EndNote><Cite><Author>Mendiratta</Author><Year>1996</Year><RecNum>2033</RecNum><
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Othmer encyclopedia of chemical technology. </secondary-title></title></secondary-title></title>
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Inc.</publisher><urls></urls></record></Cite></EndNote>] and by 1974, reached 50 million pounds
[ ADDIN EN.CITE
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Campbell, R. Pfefferkorn, and J.F. Rounsaville</author></secondary-
authors></contributors><title>Chlorine oxides and chlorine oxygen acids</title><secondary-
title>Ullmann's Encyclopedia of Industrial Chemistry</secondary-title></title></graphs="2">title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>
525</pages><volume>A6</volume><dates><year>1986</year></dates><pub-location>New York,
NY</pub-location><publisher>VCH
Publishers</publisher></urls></record></Cite></EndNote>]. In 1994, production data for
ammonium perchlorate was estimated at 22 million pounds [ ADDIN EN.CITE
<EndNote><Cite><Author>Mendiratta</Author><Year>1996</Year><RecNum>2033</RecNum><
DisplayText>(Mendiratta et al., 1996)</DisplayText><record><rec-number>2033</rec-
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type><contributors><author>Mendiratta, S.K.</author><author>R.L.
Dotson</author><author>R.T. Brooker</author></authors><secondary-
authors><author>Kroschwitz, J.I. and M. Howe-Grant.</author></secondary-
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authors></contributors><titles><title>Perchloric acid and perchlorates</title><secondary-title>Kirk-Othmer encyclopedia of chemical technology. </secondary-title></title><epages>157-170</pages><volume>18</volume><dates><year>1996</year></dates><pub-location>New York, NY</pub-location><publisher>John Wiley & Dons, Inc.</publisher><urls></record></Cite></EndNote>]. Recent production data for perchlorate are not available. Also, no information on production of perchlorate is available from the Toxic Release Inventory (TRI) database for facilities that produce or process perchlorate because

#### 3.2 Environmental Release

perchlorate is not on the list of chemicals that must be reported.

Perchlorate may be released to water from manufacturing and processing facilities. Since wastewater treatment processes (including stripping, precipitation, filtration, oxidation, or aerobic biodegradation) do not effectively remove perchlorate from waste streams, perchlorate in wastewater may eventually be released to surface water [ ADDIN EN.CITE <EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><Dis playText>(ATSDR, 2008)</br>
//DisplayText><record><rec-number>115</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</reftype><contributors><authors><author>ATSDR</author></author>></contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year></dates><publocation>Atlanta, GA</pub-location><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>]. There is no quantitative information available on the release of ammonium perchlorate from the manufacture, maintenance, or testing of solid rocket propellants, however it has been reported that perchlorate concentrations from these activities may reach the g/L level [ ADDIN EN.CITE <EndNote><Cite><Author>Herman</Author><Year>1998</Year><RecNum>2063</RecNum><Di splayText>(Herman & DisplayText) number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541731415">2063</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Herman, D.C.</author><author>W.T. Frankenberger</author></authors></contributors></title>Microbial-mediated reduction of perchlorate in groundwater</title><secondary-title>J. Environ. Qual.</secondarytitle></title></periodical><full-title>J. Environ. Qual.</full-title></periodical><pages>750-754</pages><volume>27</volume><dates><year>1998</year></dates><urls></record></Cit e></EndNote>]. According to one report, the amount of ammonium perchlorate released to wastewater from decommissioning rockets during that decade may reach 8.5 million pounds [ ADDIN EN.CITE <EndNote><Cite><Author>Buckley</Author><Year>1999</Year><RecNum>2064</RecNum><Di splayText>(Buckley, Sclippa, & Sc number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541731604">2064</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author>Buckley, S.G</author><author>G.C. Sclippa</author><author>L.L. Baxter</author></authors></contributors><titles><title>Combustion tests of rocket motor washout material: Focus on air toxics formation potential of asbestos

remediation. DE00005953</title></title></dates><year>1999</year></dates><pub-

location>Albuquerque, NM</pub-location><publisher>Sandia National Laboratories</publisher><urls></urls></record></Cite></EndNote>]. Likewise, there is no quantitative information available on the release of perchlorate from the manufacture or detonation of fireworks. However, elevated levels of perchlorate in water have been measured near fireworks manufacturing facilities in China and India [ ADDIN EN.CITE <EndNote><Cite><Author>Sijimol</Author><Year>2014</Year><RecNum>2041</RecNum><Dis playText>(Sijimol & DisplayText><record><rec-number>2041</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541694150">2041</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Sijimol, M.R.</author><author>M. Mohan</author></authors></contributors></title>Environmental impacts of perchlorate with special reference to fireworks - a review. </title><secondary-title>Environ. Monit. Assess.</secondary-title></title></erodical><full-title>Environ. Monit. Assess.</fulltitle></periodical><pages>7203-7210</pages><volume>186</volume><dates><year>2014</year></dates><urls></urls></record></ Cite></EndNote>]. Isobe et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Isobe</Author><Year>2013</Year><RecNum>2236</RecNum><Displ ayText>(2013)</br>

ayText>
<rec-number>
236</rec-number>
foreign-keys><key</td> app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543341711">2236</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Isobe, T.</author><author>Ogawa, S.P.</author><author>Sugimoto, R.</author><author>Ramu, K.</author><author>Sudaryanto, A.</author><author>Malarvannan, G.</author><author>Devanathan, G.</author><author>Ramaswamy, B.R.</author><author>Munuswamy, N.</author><author>Ganesh, D.S.</author><author>Sivakumar, J.</author><author>Sethuraman, A.</author><author>Parthasarathy, V.</author><author>Subramanian, A.</author><author>Field, J.</author><author>Tanabe, S.</author></authors></contributors></title>Perchlorate contamination of groundwater from fireworks manufacturing area in South India</title><secondarytitle>Environmental Monitoring and Assessment</secondary-title></title></fitles><periodical><fulltitle>Environmental Monitoring and Assessment</full-title></periodical><pages>5627-37</pages><volume>185</volume><number>7</number><dates><year>2013</year></dates><urls ></urls></record></Cite></EndNote>], studied the contamination of perchlorate in groundwater from areas in South India that manufactured fireworks. Concentrations of perchlorate were found to be  $<0.005-7,690 \mu g/L$  in groundwater,  $<0.005-30.2 \mu g/L$  in surface water, and  $0.063-0.393 \mu g/L$  in

Another source of perchlorate in water may be from the release of the compound to the atmosphere and deposition onto the surface of oceans, rivers, lakes, and ponds [ ADDIN EN.CITE <EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><Dis playText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>ATSDR</author></author></author></author></author></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor></actor>

location>Atlanta, GA</pub-location><publisher>U.S. Department of Health and Human

tap water. The concentration of perchlorate was higher in the area of the firework factory than in any

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other location.

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Services</publisher><urls></record></Cite></EndNote>]. Perchlorate may also be released to surface water from runoff from sand or soil that contains the compound [ ADDIN EN.CITE <EndNote><Cite><Author>Herman</author><Year>1998</Year><RecNum>2063</recNum><Di splayText>(Herman & Di splayText>(Herman & Di splayText)</a> (Herman & Di splayText></record><rec-number>2063</rec-number></record><rec-number>2063</rec-number></foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541731415">2063</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Herman, D.C.</author><author>W.T.</a>
Frankenberger</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a>
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type><contributors><author>ATSDR</author></authors></contributors><title>To xicological profile for perchlorates</title></title></title></dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></Cite></EndNote>].

All forms of perchlorate have very low vapor pressures, therefore perchlorate compounds are not expected to volatilize into air. However, releases of unspent perchlorate to air may occur during its use as a component of solid rocket boosters, as gases are released to provide propulsion of rockets through the atmosphere [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></Cite></EndNote>]. Perchlorates may also be released to the atmosphere during catastrophic failure of rockets [ ADDIN EN.CITE

<EndNote><Cite><Author>Merrill</Author><Year>1998</Year><RecNum>2069</RecNum><DisplayText>(Merrill & DisplayText><record><recnumber>2069</recnumber><foreign-keys><key app="EN" db-"

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Characterization Subcommittee and Safety & Environmental Protection Subcommittee Joint Meeting. Vol. 1. CPIA Publ. </title></title></title></title></title>

567</pages><volume>674</volume><dates><year>1998</year></dates><urls></urls></record></C

ite></EndNote>] or during open-burn decommissioning of rocket booster propellants or munitions [ ADDIN EN.CITE

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206</pages><volume>185</volume><dates><year>2000</year></dates><pub-location>Reston, VA</pub-location><publisher>American Institute of Aeronautics and Astronautics,

Inc.</publisher><urls></urls></record></Cite></EndNote>]. Other sources of perchlorate in air include release of the compound during the detonation of fireworks and other pyrotechnic devices, the wind-borne erosion of contaminated soil, sand, or particulate matter, or the production of perchlorate in the atmosphere after volcanic eruptions [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><title>><title>To xicological profile for perchlorates</title></title></dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>].

#### 3.3 Environmental Fate

In water, perchlorate compounds are expected to rapidly dissolve and dissociate into their component ions. The water solubility of perchlorate indicates that it is not expected to adsorb to sediment and organic matter and will not undergo removal through the formation of insoluble metal complexes [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><titles><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></Cite></EndNote>]. Perchlorate is persistent in the environment [ ADDIN EN.CITE

<EndNote><Cite><Author>Logan</Author><Year>1998</Year><RecNum>2066</RecNum><Disp layText>(Logan, 1998)</DisplayText><rec-number>2066</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541732141">2066</key></foreign-keys><ref-type name="Journal Article">17</ref-type name="

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type><contributors><author>Logan, B.E.
</author></author></authors></title>A review of chlorate- and perchlorate-respiring
```

microorganisms</title><secondary-title>Bioremediat. J. </secondary-title></title></periodical><pages>69-

79</pages><volume>2</volume><number>2</number><dates><year>1998</year></dates><urls></urls></record></Cite></EndNote>]. *In situ* removal of perchlorate in water has not been demonstrated [ ADDIN EN.CITE

<EndNote><Cite><Author>Coates/Author><Year>2000</Year><RecNum>2071/RecNum><Dis playText>(Coates & DisplayText)/Coates & DisplayText><record><rec-number>2071/rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541769502">2071/key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Coates, J.D./author><author>R.T.

Anderson</author></authors></contributors></title>Emerging techniques for anaerobic bioremediation of contaminated environments</title><secondary-title>Trends
Biotechnol.</secondary-title></title>Trends Biotechnol.</full-title></periodical><pages>408-

412</pages><volume>18</volume><number>10</number><dates><year>2000</year></dates><url s></urls></record></Cite></EndNote>], and removal processes such as photooxidation, biodegradation, hydrolysis, or photochemical degradation are not expected to occur [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>].

In the laboratory, soil microorganisms have been shown to reduce perchlorate under anaerobic conditions [ ADDIN EN.CITE

<EndNote><Cite><Author>Herman</Author><Year>1998</Year><RecNum>>2063</RecNum><Di splayText>(Herman & Di Sampi Frankenberger, 1998; Logan, 1998)</DisplayText><record><recnumber>>2063</recnumber><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541731415">2063</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Herman, D.C.</author><author>W.T.

Frankenberger</author></authors></contributors></title>Microbial-mediated reduction of perchlorate in groundwater</title><secondary-title>J. Environ. Qual.</secondary-title></title></periodical><full-title>J. Environ. Qual.</full-title></periodical><pages>750-754</pages><volume>27</volume><dates><year>1998</year></dates><urls></urls></record></cite><Cite><Author>Logan</Author><Year>1998</year></record></cr>

number>2066</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541732141">2066</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><au

</author></authors></contributors><titles>< title>A review of chlorate- and perchlorate-respiring

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Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water
microorganisms</title><secondary-title>Bioremediat. J. </secondary-
title></titles><periodical><full-title>Bioremediat. J.</full-title></periodical><pages>69-
79</pages><volume>2</volume><number>2</number><dates><year>1998</year></dates><urls><
/urls></record></Cite></EndNote>]. However perchlorate has been shown to be persistent in soil,
and thus the significance of this removal process is questionable. One study reported that the
necessary criteria for degrading perchlorate in soil are anaerobic conditions, an adequate carbon
source, and an active perchlorate-degrading microbial population [ ADDIN EN.CITE
<EndNote><Cite><Author>Tipton</Author><Year>2003</Year><RccNum>2067</RecNum><Disp
layText>(Tipton, Rolston, & DisplayText>(Tipton, Rolston, & DisplayText) (Tipton, Rolston, & Displa
number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
timestamp="1541732227">2067</key></foreign-keys><ref-type name="Journal Article">17</ref-
type><contributors><author>Tipton, D.K.</author><author>D.E.
Rolston</author><author>K.M.
Scow</author></authors></contributors><titles><title>Bioremediation and biodegradation:
Transport and biodegradation of perchlorate in soils</title><secondary-title>J. Environ. Qual.
</secondary-title></title>><periodical><full-title>J. Environ. Qual.</full-
title></periodical><pages>40-
46</pages><volume>32</volume><number>1</number><dates><year>2003</year></dates><urls>
</urls></record></Cite></EndNote>].
Monitoring data suggest that perchlorate may accumulate in living organisms, as the compound was
detected in fish, vegetation, insect, amphibian, and rodent samples near a site of known contamination
[ ADDIN EN.CITE
<EndNote><Cite><Author>Smith</Author><Year>2001</Year><RecNum>2072</RecNum><Displ
ayText>(Smith, Theodorakis, Anderson, & DisplayText><record><rec-
number>2072</rec-number><foreign-keys><key app="EN" db-
id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541769635">2072</key></foreign-
keys><ref-type name="Journal Article">17</ref-type><contributors><author>Smith,
P.N.</author><author>C.W. Theodorakis</author><author>T.A. Anderson</author><author>R.J.
Kendall</author></authors></contributors><title>Preliminary assessment of perchlorate in
ecological receptors at the Longhorn Army ammunition plant (LHAAP), Karnack,
Texas</title><secondary-title>Ecotoxicology</secondary-title></title><periodical><full-
title>Ecotoxicology</full-title></periodical><pages>305-
```

313</pages><volume>10</volume><dates><year>2001</year></dates><urls></urls></record></Cit e></EndNote>]). However, experimental studies indicate that the bioconcentration of perchlorate in aquatic organisms is low, with bioconcentration factors of 1.854 and 0.70 reported for the Asiatic clam and bluegill, respectively [ ADDIN EN.CITE

<EndNote><Cite><Author>Dean</Author><Year>2004</Year><RecNum>2073</RecNum><Displ ayText>(Dean et al., 2004)</DisplayText><record><rec-number>2073</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541769720">2073</key></foreign-keys></ref-type name="Journal Article">17</ref-

type><contributors><author>Dean, K.E.</author><author>R.M.

Palachek</author><author>I.M. Noel</author><author>R. Warbritton</author><author>J. Aufderheide</author><author> J.

Wireman</author></authors></contributors><title>Development of freshwater water-quality criteria for perchlorate</title><secondary-title>Environ. Toxicol. Chem.</secondarytitle></title></periodical><full-title>Environ. Toxicol. Chem.</full-title></periodical><pages>14411451 < pages > < volume > 23 < / volume > 6 < / number > 6 < / volume > < dates > < year > 2004 < / year > < / dates > < url s > < / cord > < / Cite > < / EndNote > 1.

### 3.4 Summary

Perchlorate is primarily used as an oxidizer in solid fuels to power rockets, missiles, and fireworks. Ammonium perchlorate is the form most widely used in the U.S. A source of release of perchlorate to the environment is thought to be from the manufacture of perchlorate for rocket booster engines, and their testing and decommissioning. Perchlorate may also be released to the environment from the manufacture of munitions, the catastrophic failure of booster rockets, and during its use in pyrotechnic devices. It may also be present in the environment from its use as a fertilizer and from natural sources. Perchlorate has been found to be persistent in water and soil and is not removed by most common degradation processes. It is not expected to volatilize to the atmosphere. Perchlorate may accumulate in living organisms; however, bioconcentration factors have been reported to be low.

## 4. Drinking Water Occurrence

EPA's Perchlorate Occurrence and Monitoring Document provides estimates of the baseline perchlorate occurrence in PWSs [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"qUTie4VE","properties":{"formattedCitation":"(USEPA, 2018a)", "plainCitation": "(USEPA, 2018a)", "noteIndex":0}, "citationItems":[{"id":1675, "uris":["http://zotero.org/groups/945096/items/Y ERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":1675,"t ype":"article","title":"Perchlorate Occurrence and Monitoring Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json" \ ]. After reviewing the available data on perchlorate in drinking water, the EPA determined that the best nationally representative source is data from the Unregulated Contaminant Monitoring Rule 1 (UCMR1). This section summarizes the EPA's perchlorate occurrence analysis [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"UCKVzf9d","properties":{"formattedCitation":"(USEPA, 2018a)", "plainCitation": "(USEPA, 2018a)", "noteIndex":0}, "citationItems":[{"id":1675, "uris":["http://zotero.org/groups/945096/items/Y ERQWPRZ"], "uri": ["http://zotero.org/groups/945096/items/YERQWPRZ"], "itemData": {"id":1675, "t ype":"article","title":"Perchlorate Occurrence and Monitoring Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json"} ].

The UCMR is a national drinking water monitoring program administered by the EPA. The UCMR1 monitoring cycle included a census of all large CWSs and NTNCWSs (i.e., those serving more than 10,000 people) and a statistical sample of 800 small CWSs and NTNCWSs (i.e., those serving 10,000 people or fewer) [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"sOBUvFhD","properties":{"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA, 2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","author":[{"family":"USEPA","given":""}],"issued":{"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The UMCR1 cycle was from 2001-2005; most data collection occurred between 2001 and 2003.

During UMCR1, the EPA obtained perchlorate sample data from systems in all 50 states, the District of Columbia, Tribal Nations, and four U.S. territories (Puerto Rico, Virgin Islands, Guam, and the Northern Mariana Islands). Nearly all systems in either the statistical sample or the census sample responded: 99.6% of small systems and 99.0% of large systems provided data [ ADDIN ZOTERO\_ITEM CSL\_CITATION

```
{"citationID":"45nk8Bn9","properties":{"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA,
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2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"t

Deriving a Maximum Contaminant Level Goal for Perchlorate in Drinking Water

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ype":"article","title":"Perchlorate Occurrence and Monitoring
Report","author":[{"family":"USEPA","given":""}],"issued":{"date-
parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-
citation.json"}].
```

Systems collected samples at each entry point to the customer distribution system.<sup>4</sup> Entry points are the point of compliance with drinking water standards. Systems can have multiple entry points. Sampling frequency varied by source water: four quarterly samples in a one-year period for surface water systems, and two samples at least 6 months apart for ground water systems [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"8NSuLNK0","properties": {"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA,

2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/Y ERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"type":"article","title":"Perchlorate Occurrence and Monitoring

Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The minimum reporting level (MRL) was 4 μg/L [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"k4YEKduX","properties":{"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA,

2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/Y ERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"t ype":"article","title":"Perchlorate Occurrence and Monitoring

Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

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parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-

citation.json"}].

<sup>&</sup>lt;sup>4</sup> In response to comments on UCMR1 data quality [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"G1AISvS5", "properties": {"formattedCitation":"(U.S. Chamber of Commerce. 2012)", "plainCitation": "(U.S. Chamber of Commerce, 2012)", "noteIndex":3}, "citationItems":[{"id":152, "uris":["http://zotero.org/groups/945096/items/CMDYPKE9 "],"uri":["http://zotero.org/groups/945096/items/CMDYPKE9"],"itemData": {"id":152,"type":"article","title":"I nformation Quality Guidelines (IQG) Request for Correction", "publisher": "Letter to USEPA. September.", "author": [{"family": "U.S. Chamber of Commerce", "given": ""}], "issued": {"dateparts":[["2012"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json" | 1, the EPA reviewed the UCMR1 data to identify instances where source water monitoring samples were accompanied by corresponding 'downstream' entry point monitoring samples. In these instances, only the entry point samples provide the perchlorate concentration in water delivered to customers. Therefore, the 2013 version of the UCMR1 dataset excludes these types of source water samples [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"DTSSVkcz", "properties": {"formattedCitation":"(USEPA, 2018a)", "plainCitation": "(USEPA, 2018a)", "noteIndex":3}, "citationItems":[{"id":1675, "uris":["http://zotero.org/groups/945096/items/YERQWP RZ"], "uri": ["http://zotero.org/groups/945096/items/YERQWPRZ"], "itemData": {"id":1675, "type": "article", "tit le":"Perchlorate Occurrence and Monitoring Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

The maximum reported concentration of perchlorate was 420  $\mu$ g/L, which was found in a single surface water sample from a PWS in Puerto Rico. The average concentration of perchlorate for those samples with positive detections for perchlorate was 9.9  $\mu$ g/L and the median concentration was 6.4  $\mu$ g/L.

The summary statistics in [ REF\_Ref528590138 \h ] show total samples, entry points, and systems in the UCMR1 perchlorate dataset. It also shows the number of reported perchlorate detections ( $\geq 4 \mu g/L$ ) and the number of detections strictly greater than the MRL of 4  $\mu g/L$ .

Table [ SEQ Table \\* ARABIC ]. UCMR1 Data Summary Statistics

ltem	Small System Sample	Large System Census	Sum
Total samples	3,295	30,837	34,132
Detections ≥ 4 μg/L	15	525	540
• Detections > 4 μg/L	15	510	525
Total entry points	1,454	13,482	14,936
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	8	328	336
• Detections > 4 μg/L	8	314	322
Total systems	797	3,068	3,865
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	8	141	149
• Detections > 4 μg/L	8	136	144

Source: [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"UAoGFPZv","properties":{"formattedCitation":"(USEPA,

2018a)", "plainCitation": "(USEPA,

2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/Y ERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"type":"article","title":"Perchlorate Occurrence and Monitoring

Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]

[ REF \_Ref528590178 \h ] shows the populations that correspond with the occurrence summary in [ REF \_Ref528590138 ]. The entry point population estimates reflect the assumption that system population is uniformly distributed across entry points; e.g., the entry point population for a system with two entry points is one-half the total system population.

#### Table [ SEQ Table \\* ARABIC ]. UCMR1 Data Service Population Summary Statistics

Item	Small System Sample	Large System Census	Sum
Total entry point population	2,760,570	222,853,101	225,613,671
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	9,484	4,063,241	4,072,725
Detections > 4 μg/L	9,484	3,965,473	3,974,957
Total system population	2,760,570	222,853,101	225,613,671
<ul> <li>Detections ≥4 μg/L</li> </ul>	13,483	16,159,082	16,172,565
Detections > 4 μg/L	13,483	15,922,821	15,936,304

Source: [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID": "ChxDKgDr", "properties": {"formattedCitation": "(USEPA,

2018a)", "plainCitation": "(USEPA,

2018a)","noteIndex":0},"citationItems":[{"id":1675,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":1675,"type":"article","title":"Perchlorate Occurrence and Monitoring

Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]

Because the UCMR1 data are well over a decade old, the EPA considered potential sources of uncertainty because of changes between current conditions and conditions at the time of data collection. One important change is the adoption of regulatory limits in two states: Massachusetts adopted a drinking water standard for perchlorate of 2 µg/L in 2006 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"8DPpSrv3","properties":{"formattedCitation":"(MassDEP, 2006)","plainCitation":"(MassDEP,

2006)","noteIndex":0},"citationItems":[{"id":151,"uris":["http://zotero.org/groups/945096/items/9893 MBZH"],"uri":["http://zotero.org/groups/945096/items/9893MBZH"],"itemData":{"id":151,"type":"p ersonal\_communication","title":"Letter to Public Water Suppliers concerning new perchlorate regulations","URL":"https://www.mass.gov/lists/perchlorate-background-information-and-standards#perchlorate---final-standards-

","author":[{"family":"MassDEP","given":""}],"issued":{"date-

parts":[["2006"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], and California promulgated a drinking water standard of 6  $\mu$ g/L in 2007 [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"cfr6HNhg","properties": {"formattedCitation":"(California Department of Public Health, 2007)", "plainCitation":"(California Department of Public Health,

2007)","noteIndex":0},"citationItems":[{"id":150,"uris":["http://zotero.org/groups/945096/items/RA45NKLQ"],"uri":["http://zotero.org/groups/945096/items/RA45NKLQ"],"itemData":{"id":150,"type": "personal\_communication","title":"State Adoption of a Perchlorate

Department of Public Health", "given": ""}], "issued": {"date-

parts":[["2007"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Systems in these states cannot exceed these limits. Subsequently EPA re-evaluated and excluded the California and Massachusetts from the remainder of this analysis. [ REF Ref528590235 \h \\* MERGEFORMAT] summarizes the UCMR1 data pursuant to this assumption.

Table [ SEQ Table \\* ARABIC ]. UCMR1 Data Summary Statistics, Excluding California and Massachusetts

ltem	Small System Sample	Large System Census	Sum
Total samples	2,984	21,128	24,112
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	13	206	219
<ul> <li>Detections &gt; 4 μg/L</li> </ul>	13	195	208
Total entry points	1,327	9,118	10,445
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	7	159	166
<ul> <li>Detections &gt; 4 μg/L</li> </ul>	7	149	156
Total systems	737	2591	3,328
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	7	91	98
<ul> <li>Detections &gt; 4 μg/L</li> </ul>	7	91	98
Total entry point population	2,537,888	183,525,431	186,063,319
<ul> <li>Detections ≥ 4 μg/L</li> </ul>	5,430	1,760,855	1,766,285
<ul> <li>Detections &gt; 4 μg/L</li> </ul>	9,484	3,965,473	3,974,957
Total system population	2,537,888	183,525,431	186,063,319
• Detections ≥4 μg/L	9,429	7,762,593	7,772,022
• Detections > 4 μg/L	13,483	15,922,821	15,936,304

Another change involves remediation efforts to reduce the sources of perchlorate in drinking water.

USEPA [ ADDIN ZOTERO ITEM CSL CITATION

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2018a)", "plainCitation": "(USEPA,

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Report", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] describes remediation efforts that have effectively reduced perchlorate levels in the Colorado River water from a range of 4  $\mu$ g/L to 9  $\mu$ g/L during the UCMR1 data collection period to 1  $\mu$ g/L to 2  $\mu$ g/L after 2009.

Both of these changes tend to reduce current perchlorate concentrations compared to the levels observed in the UCMR1 data. A third type of change that has an uncertain impact on occurrence is the change in the universe of systems over time. Some systems operating during the UCMR1 data

collection period are now inactive. There are also new systems that were not operating during the UCMR1 period.

## 5. Exposure or Occurrence from Food and Other Sources

This section summarizes studies of perchlorate occurrence in food, as well as the most recent perchlorate exposure estimates from the consumption of food (Section [ REF \_Ref531260962 \r \h ]). This section also summarizes studies of the occurrence of perchlorate in air and soil (Section [ REF \_Ref531261023 \r \h ] and [ REF \_Ref531261025 \r \h ]), as well as studies focused on other environmental and occupational sources of perchlorate exposure (Sections [ REF \_Ref531261047 \r \h ] and [ REF \_Ref531261049 \r \h ]).

### 5.1 Exposure/Occurrence from Food

Although perchlorate has been detected in many different kinds of food, studies of perchlorate in food generally show the highest levels of perchlorate in produce and milk products. A study by the European Food Safety Authority found relatively high mean occurrence values in dried products such as "Tea and herbs for infusion" (324 μg /kg) and "Herbs, spices and condiments" (63 μg /kg), as well as in fresh vegetables such as "Radishes" (117 μg/kg), "Rocket salad, rucola" (75 μg /kg) and "Spinach (fresh)" (132 μg/kg) [ ADDIN EN.CITE <EndNote><Cite><Author>Arcella</Author><Year>2017</Year><RecNum>2217</RecNum><Dis playText>(Arcella, Binaglia, &amp; Vernazza, 2017)</DisplayText><record><recnumber>2217</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1542749308">2217</key><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1542749308">2217</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><<contributors><author>Arcella, Davide</author><author>Binaglia,

Marco</author><author>Vernazza,
Francesco</author></authors></contributors><title>Dietary exposure assessment to
perchlorate in the European population</title><secondary-title>EFSA Journal</secondarytitle></title><periodical><full-title>EFSA Journal

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num>10.2903/j.efsa.2017.5043</electronic-resource-num></record></Cite></EndNote>]. [
HYPERLINK \l "\_ENREF\_71" \o "Lau, 2013 #2224" ] found that consumption mass of "Milk
products" predicted higher urinary perchlorate concentrations in children age 6 to 11 years using
National Health and Nutrition Examination Survey (NHANES)<sup>5</sup> data. Lau et al. (2013) [
HYPERLINK \l "\_ENREF\_71" \o "Lau, 2013 #2224" ] also found a statistically significant
relationship between consumption mass of "Milk products" and higher urinary perchlorate
concentrations in adults, along with consumption mass of "Vegetables" and "Dark-green leafy
vegetables". Additionally, the FDA Exploratory Survey on Perchlorate in Food [ ADDIN EN.CITE
<EndNote><Cite ExcludeAuth="1"><Author>Food and Drug Administration
(FDA)</Author><Year>2015</Year><RecNum>1916</RecNum><DisplayText>(2015)</DisplayTe
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<sup>&</sup>lt;sup>5</sup> The NHANES is run by the Centers for Disease Control and Prevention's National Center for Health Statistics and was designed to collect information on the health and nutritional status of the U.S. civilian, non-institutionalized population through in-home interviews and physical examinations

keys><ref-type name="Web Page">12</ref-type><contributors><authors><author>Food and Drug Administration (FDA), </author></contributors></title>Preliminary estimation of perchlorate dietary exposure based on FDA 2004/2005 exploratory data</title></title></dates><year>2015</year></dates><urls><related-urls><url> face="underline" font="default" size="100%">https://www.fda.gov/Food/FoodborneIllnessContaminants/ChemicalContaminants/ucm 077653.htm</style></url></related-urls></urls></record></Cite></EndNote>] found that milk contributed the highest perchlorate exposure (47% of total exposure) for the mean population aged 2 years and older. [ HYPERLINK \1" ENREF 74" \0 "Lee, 2012 #2225" ] found that the highest average concentration of perchlorate for Korean food was found in dairy products (6.34 µg/kg), with over 85% of samples having a detection. [HYPERLINK \1" ENREF 74" \0 "Lee, 2012 #2225"] also found that spinach was the food with the highest average perchlorate concentration (39.9 μg/kg) followed by tomatoes (19.8 μg/kg) and green pumpkin (11.6 μg/kg). Alomirah et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Alomiraha</Author><Year>2016</Year><RecNum>2239</RecNum>< DisplayText>(2016)</DisplayText><rec-number>2239</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543344227">2239</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><authors><author>Alomiraha, H.F.</author><author>Al-Zenkia, S.F.</author><author>Alaswada, M.C.</author><author>Alruwaiha, N.A.</author><author>Wub, O.</author><author>Kannanb, K.</author></authors></contributors><title>Widespread occurrence of perchlorate in water, foodstuffs and human urine collected from Kuwait and its contribution to human exposure</title><secondary-title>Food Additives and Contaminants</secondary-title></titles><periodical><full-title>Food Additives and Contaminants</full-title><abbr-1>Food Addit Contam</abbr-1></periodical><pages>1016-1025</pages><volume>33</volume><number>6</number><dates><vear>2016</vear></dates><url s></urls></record></Cite></EndNote>] found that leafy vegetables accounted for 36.2% of perchlorate exposure among the Kuwait population, with fruits contributing 15.3% and non-leafy vegetables contributing 10.1%. In addition to the perchlorate detected in produce and milk products, perchlorate has also been detected in beer, wine, dietary supplements, and vitamins [ ADDIN **EN.CITE** <EndNote><Cite><Author>Aribi</Author><Year>2006</Year><RecNum>2081</RecNum><Displ ayText>(Aribi, Blanc, Antonsen, & Damp; Sakuma, 2006; Snyder, Pleus, Vanderford, & Damp; Holady., 2006)</DisplayText><record><rec-number>2081</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541770654">2081</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Aribi, H.</author><author>Y.J.C. Le Blanc</author><author>S. Antonsen</author><author>T. Sakuma</author></authors></contributors></title>>Analysis of Perchlorate in Foods and Beverages by Ion Chromatography Couple with Tandem Mass Spectrometry (IC-ESI-MS/MS)</title><secondary-title>Anal. Chim. Acta</secondary-title></title><periodical><fulltitle>Anal. Chim. Acta</full-title></periodical><pages>39-47</pages><volume>567</volume><number>1</number><dates><vear>2006</vear></dates><urls ></urls></record></Cite><Cite><Author>Snyder</Author><Year>2006</Year><RecNum>2082</ RecNum><record><rec-number>2082</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541770749">2082</key></foreign-

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S.A.</author><author>R.C. Pleus</author><author>B.J. Vanderford</author><author>J.C. Holady. </author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></al></al></al></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></author></a></author></a></author></a></author></author></a></author></a></author></a></al></al></al>

The most recent and most comprehensive information available on total daily exposure to perchlorate through the diet for the U.S. population comes from an analysis carried out by scientists at the FDA in the Abt et al. [ ADDIN EN.CITE < EndNote > < Cite

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title></titles><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016</year></dates><urls></record></Cite></EndNote>] study, which combines perchlorate food contamination data from the FDA's Total Diet Study (TDS) with food consumption data from CDC's What We Eat in America (WWEIA). The Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-

title></titles><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016<//year></dates><urls></urls></record></Cite></EndNote>] analysis was modeled after a previous FDA study using TDS perchlorate data to estimate mean intake of perchlorate for 14 age/sex groups [ ADDIN EN.CITE | ADDIN EN.CITE.DATA | ]. The EPA also performed a follow-on study to the Abt et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><ReeNum>1917</ReeNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</td>

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J.</author><author>Fouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></author></author></title>Update on dietary intake of perchlorate and

iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>] analysis to estimate perchlorate intake using the same methods but for additional populations of interest. This remainder of this section describes these EPA and FDA perchlorate in food analyses.

#### 5.1.1 FDA Total Diet Study

The Food and Drug Administration's (FDA's) Total Diet Study (TDS) is an ongoing FDA program that collects information on levels of various contaminants, including perchlorate, which occur in food and beverages commonly consumed by the U.S. population [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

FDA</Author><Year>2015</Year><RecNum>2006</RecNum><DisplayText>(U.S. FDA, 2015)</DisplayText><record><rec-number>2006</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1530038884">2006</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><author>><author>>U.S. FDA,</author></author></author>></contributors><tittles><tittle>Total diet study — study design</title></title></title></title></ref-type><contributors><ref-type><contributors><ref-type><contributors><ref-type><contributors></archive=</a>

urls><url>http://www.fda.gov/food/foodscienceresearch/totaldietstudy/ucm184232.htm#</url></relat ed-urls></urls></record></EndNote>]. To estimate the levels of contaminants in these food products, the FDA buys these foods as a consumer would, prepares them as directed,<sup>6</sup> and then analyzes the prepared foods for levels of the contaminants of interest. This process yields nationally representative estimates of contaminant levels in more than 280 kinds of food and beverages. For the purposes of estimating perchlorate concentrations in food, TDS data on perchlorate were collected from 2008 to 2012. These years of TDS perchlorate data were used for both the Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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title></periodical><dates><year>2016</year></dates><urls></record></Cite></EndNote>]st udy, as well as the EPA perchlorate in food analysis. FDA also included perchlorate as an analyte in TDS baby foods in 2005 and in all other TDS foods in 2006, which were used as the basis for the earlier Murray et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] study that served as the

<sup>&</sup>lt;sup>6</sup> The FDA prepares their samples using deionized water.

model for carrying out the Abt et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1503519671">1917</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Abt, E.</author><author>Spungen, J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-title></periodical><dates><year>2016

#### 5.1.2 Abt et al. (2016)

The aim of the Abt et al. [ ADDIN EN.CITE < EndNote > < Cite

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title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>] study was to use the available TDS data to estimate daily perchlorate and iodine intake rates for 14 age/sex groups. The study was intended to provide an update to the Murray et al. [ ADDIN EN.CITE

ADDIN EN.CITE.DATA ] study, which also estimated perchlorate exposure for the same 14 sex/age groups using 2005 to 2006 TDS data combined with food consumption data from USDA's 1994 to 1996 and 1998 Continuing Survey of Food Intakes by Individuals.

The Abt et al. [ ADDIN EN.CITE < EndNote > < Cite

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J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title>

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study instead combined 2008 to 2012 TDS food contaminant data with food consumption and
demographic data from the 2007 to 2012 NHANES [ ADDIN EN.CITE
<EndNote><Cite><Author>CDC</Author><Year>2007-2008, 2009-2010, 2011-
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urls></urls></record></Cite></EndNote>]. The dietary intake measurement component of the survey
is called What We Eat in America (WWEIA) and provides 24-hour food diary data. Dietary data are
collected for up to two days for each respondent [ ADDIN EN.CITE
<EndNote><Cite><Author>CDC</Author><Year>2007-2008, 2009-2010, 2011-
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With four samples taken for each food for each year of data in the TDS, the study had a total of 20 perchlorate samples for each of the 282 foods in the TDS. Many of the measurements of perchlorate in the TDS fall below the limit of detection (LOD). In order to account for values below the LOD, Abt et al. [ ADDIN EN.CITE <EndNote><Cite

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title></periodical><dates><year>2016</year></dates><urls></record></Cite></EndNote>] used a method suggested by WHO to calculate both the lower and upper bound concentrations in the foods by substituting all values below the LOD with zero to estimate the lower bound, and substituting all values below the LOD with the LOD to estimate the upper bound [ ADDIN EN.CITE <EndNote><Cite><Author>European Food Safety

Authority</Author><Year>2010</Year><RecNum>2007</RecNum><DisplayText>(Authority,

Epidemiology</full-

2010)</br>
2010)
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96</pages><number>8</number><dates><year>2010</year></dates><urls></urls></record></Cite></EndNote>]. Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title>

title></periodical><dates><year>2016</year></dates><urls></record></Cite></EndNote>] also implemented a Bayesian method for substituting values below the LOD known as the clustered zero-inflated lognormal (CZILN) method, which groups foods with similar exposure profiles together and models the values below the LOD based on the trends of the known values of the grouped foods above the LOD. Once these methods were applied to each of the values below the LOD, average perchlorate concentrations for each TDS food were calculated and carried forward for use in the perchlorate intake analysis.

The WWEIA food diary data include each individual food that was consumed during each 24-hour study period, as well as the amount of each food consumed measured in grams. For a large subset of respondents, food diary data are available for two days. For this analysis, the study authors included only participants with two days of dietary survey data, and used the appropriate NHANES survey weights for this segment of the dataset. Foods that are included in the TDS are not characterized and coded exactly the same way as the foods in the WWEIA food diaries. Therefore, it is necessary to use expert judgement to match the TDS foods with the WWEIA foods. The study authors developed a "matching" file that enabled them to link each food consumed in the 2007 to 2012 WWEIA data with one of the 282 foods in the 2008 to 2012 TDS data. Once each consumed food in the WWEIA was linked to a perchlorate concentration from TDS, the authors were able to develop estimates of percapita mean daily intake for each participant based on both the WHO method for values below the LOD, as well as the CZILN method, by adding up the total perchlorate intake for each of the two days of food intake in the WWEIA, and calculating the average total perchlorate intake between the two days. CZILN estimated and WHO method lower and upper bound estimated mean daily intake results for the 14 age/sex groups from Abt et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa vText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1503519671">1917</key></foreign-keys></ref-type name="Journal Article">17</ref-

[PAGE \\* MERGEFORMAT]

type><contributors><author>Abt, E.</author><author>Spungen, J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title>>cperiodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-title></periodical><dates><year>2016</per></dates><urls></urls></record></crite></EndNote>] are reported in [ REF Ref528663468 \h ].

Table [ SEQ Table \\* ARABIC ]. CZILN and WHO Method Lower- and Upper-bound Perchlorate Intakes from Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><DisplayText>(2016)</DisplayText><record><rec-number>1917</rec-

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E.</author><author>Spungen, J.</author><author>Pouillot,

R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz,

M.</author></contributors><title>><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>]

		Total Average Intake (μg/person/day)			Total Average Intake (μg/kg/day)			
Population Group		Lower-bound	Upper-bound	CZILN Method	Lower-bound	Upper-bound	CZILN Method	
Infants	6-11 mo	3.3	3.6	3.2	0.36	0.39	0.36	
Children	2 yr	6.1	6.6	6.0	0.44	0.48	0.43	
Children	6 yr	6.5	7.3	6.4	0.28	0.31	0.28	
Children	10 yr	6.1	7.1	6.2	0.16	0.18	0.16	
Teenage Girls	14-16 yr	5.7	6.8	6.0	0.10	0.12	0.10	
Teenage Boys	14-16 yr	7.4	8.7	7.4	0.11	0.13	0.11	
Women	25-30 yr	6.3	7.8	6.5	0.09	0.11	0.09	
Men	25-30 yr	7.8	10.0	8.3	0.09	0.12	0.10	
Women	40-45 yr	6.6	8.3	6.9	0.09	0.12	0.10	
Men	40-45 yr	9.0	11.2	9.3	0.10	0.13	0.10	
Women	60-65 yr	7.5	9.1	7.4	0.10	0.12	0.10	
Men	60-65 yr	7.9	9.9	8.3	0.09	0.11	0.09	
Women	70+ yr	6.5	7.7	6.6	0.10	0.11	0.10	
Men	70+ yr	7.6	9.0	7.7	0.09	0.11	0.09	

Source: Abt et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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The Abt et al [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><DisplayText>(2016)</DisplayText><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1503519671">1917</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Abt, E.</author><author>Spungen,

J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title>

title></periodical><dates><year>2016</year></dates><urls></record></cite></EndNote>] study found that 97% of the TDS foods had detectable levels of perchlorate in at least one sample, which is higher than was previously reported in the Murray et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA] study that used 2005 to 2006 TDS data. The authors concluded that the majority of perchlorate intake for infants comes from baby food (56%), while the other child and teenage groups had the largest proportion of perchlorate intake from consumption of dairy (from 53% of perchlorate intake in two year old children, to 28% of perchlorate intake in girls age 14 to 16 years). Adult groups had the largest proportion of their perchlorate intake from either "meat, poultry, fish" or

#### 5.2 Occurrence in Air

Epidemiology</full-

"vegetables".

Perchlorate in wet deposition was measured at 26 sites across the U.S. in a 3-year period (2004-2007) ADDIN EN.CITE

<EndNote><Cite><Author>Rajagopalan</Author><Year>2009</Year><RecNum>2084</RecNum><br/>
<DisplayText>(Rajagopalan et al., 2009)</DisplayText><record><rec-number>2084</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541770962">2084</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Rajagopalan, S.</author><author>T.</a>

Anderson</author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><auth

Cheng</author><author>W.A. Jackson</author></authors></contributors><titles><title>Perchlorate in wet deposition across North America</title><secondary-title>Environ. Sci. Technol. </secondary-title></title></periodical><full-title>Environ. Sci. Technol.</full-title></periodical><pages>616-622</pages><volume>43</volume><dates><year>2009</year></dates><urls></urls></record></Cit e></EndNote>]. Perchlorate concentrations varied from < 5 ng/L to a high of 102 ng/L, with a mean

of  $14.1 \pm 13.5$  ng/L (standard deviation). Perchlorate concentrations in wet deposition were highest in May-August and lowest in December-February and generally increased with distance from the coast. Perchlorate concentrations were also significantly associated with other monitored ions (calcium, sodium, potassium, ammonia, nitrate, chlorine, and sulfate).

Several studies [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] have documented the increased concentration in air due to July 4<sup>th</sup> fireworks, including deposition on soil. Wu et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Wu</Author><Year>2011</Year><RecNum>2087</RecNum><Displa yText>(2011)</DisplayText><rec-number>2087</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1541771183">2087</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Wu, Q.</author><author>J.F. Oldi</author><author>K. Kanna</author></author></contributors></title> Fate of Perchlorate in Man-made Reflecting Pond Following a Fireworks Display in Albany, New York, USA</title><secondary-title>Environmental Toxicology and Chemistry</secondary-title></title>/title>

title>Environmental Toxicology and Chemistry</full-title><abbr-1>Environ Toxicol Chem</abbr-1></periodical><pages>2449-

2455 < pages > volume > 30 < volume > number > 11 < number > < dates > < year > 2011 < year > < / dates > < urls > < / urls > < / cite > < / EndNote > ] noted that aerial deposition from fireworks displays ranged from 670 to 2.620 g/ha.

#### 5.3 Occurrence in Soil

The maximum reported perchlorate concentration in soil at 27 DoD facilities in 12 states ranged from approximately 32 to 2,000,000 ppb. Maximum reported perchlorate concentrations in sediment ranged from 17 to 230 ppb at four Federal defense facilities [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2005</Year><RecNum>2237</RecNum><DisplayText>(U.S. EPA, 2005b)</DisplayText><record><rec-number>2237</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543342576">2237</key></foreign-keys><ref-type name="Report">27</ref-type><contributors><author>U.S.

EPA,</author></authors></contributors><titles><title>Known perchlorate releases in the U.S. - March 25, 2005</title></title></disp><tycar></dates><pub-location>Washington, D.C.</pub-location><publisher>U.S. Environmental Protection

Agency</publisher><urls></record></Cite></EndNote>] and perchlorate levels in four of 12 sediment samples at an ammunition plant ranged from 12 to 704 μg/L [ ADDIN EN.CITE <EndNote><Cite><Author>Smith</Author><Year>2001</Year><RecNum>2072</RecNum><Displ

ayText>(Smith et al., 2001)</DisplayText><record><rec-number>2072</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1541769635">2072</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Smith, P.N.</author><author>C.W.

Theodorakis</author><author>T.A. Anderson</author><author>R.J.

Kendall</author></authors></contributors><titles><title>Preliminary assessment of perchlorate in ecological receptors at the Longhorn Army ammunition plant (LHAAP), Karnack,

Texas</title><secondary-title>Ecotoxicology</secondary-title></title><periodical><full-title>Ecotoxicology</full-title></periodical><pages>305-

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313</pages><volume>10</volume><dates><year>2001</year></dates><urls></urls></record></Cit
e></EndNote>]. Perchlorate was detected in 38% of 113 soil samples and 93% of sediment samples
collected from the Lake Mead area of Nevada with average/maximum concentrations of 57.7/1,470
mg/kg and 12.8/56.0 mg/kg, respectively [ ADDIN EN.CITE
<EndNote><Cite><Author>Dcan</Author><Year>2004</Year><RecNum>2073</RecNum><Displ
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timestamp="1541769720">2073</key></foreign-keys></ref-type name="Journal Article">17</ref-
type><contributors><author>Dean, K.E.</author><author>R.M.
Palachek</author><author>J.M. Noel</author><author>R. Warbritton</author><author>J.
Aufderheide</author><author> J.
Wireman</author></authors></contributors><title>Development of freshwater water-quality
criteria for perchlorate</title><secondary-title>Environ. Toxicol. Chem.</secondary-
title></title><periodical><full-title>Environ. Toxicol. Chem.</full-title></periodical><pages>1441-
1451</pages><volume>23</volume><number>6</number><dates><year>2004</year></dates><url
s></urls></record></Cite></EndNote>]. A mean perchlorate concentration of 24.7 µg/g in 51 soil
samples was reported from three sites in Nevada [ ADDIN EN.CITE
<EndNote><Cite><Author>Smith</Author><Year><2004</Year><RecNum><2088</RecNum><Displ
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P.N.</author><author>Y. Lu</author><author>S.T. McMurry</author><author>T.A.
Anderson</author></authors></contributors><title>Perchlorate in water, soil, vegetation, and
rodents collected from the Las Vegas Wash, Nevada, U.S.A.</title><secondary-title>Environ. Pollut.
</secondary-title></title>><periodical><full-title>Environ. Pollut.</full-
title></periodical><pages>121-
127</pages><volume>132</volume><number>1</number><dates><year>2004</year></dates><url
s></urls></record></Cite></EndNote>|while soil samples from a tobacco field that had been
fertilized with products derived from Chilean caliches reported perchlorate at a concentration of 340
μg/g [ ADDIN EN.CITE
<EndNote><Cite><Author>Ellington</Author><Year>2001</Year><RecNum>2056</RecNum><D
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type><contributors><author>Ellington, J.J.</author><author>N.L.
Wolfe</author><author>A.W. Garrison</author><author>J.J. Evans</author><author>J.K
Avants</author><author>Q. Teng</author></authors></contributors><title>Determination
of perchlorate in tobacco plants and tobacco products</title><secondary-title>Environ. Sci.
Technol.</secondary-title></titles><periodical><full-title>Environ. Sci. Technol.</full-
title></periodical><pages>3213-
3218</pages><volume>35</volume><number>15</number><dates><year>2001</year></dates><ur
ls></urls></record></Cite></EndNote>]. Natural perchlorate, unrelated to the Chilean source, has
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been widely detected in the arid U.S. Southwest. Concentrations in unconsolidated surficial material

from Death Valley, CA range from 0.25-1.7 mg/kg (Jackson et al. 2010).

#### 5.4 Other Residential Exposures

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Perchlorate has been detected in tobacco products [ ADDIN EN.CITE
<EndNote><Cite><Author>Ellington</Author><Year>2001</Year><RecNum>2056</RecNum><D
isplayText>(Ellington et al., 2001)</DisplayText><record><rec-number>2056</rec-
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timestamp="1541730269">2056</key></foreign-keys></ref-type name="Journal Article">17</ref-
type><contributors><author>Ellington, J.J.</author><author>N.L.
Wolfe</author><author>A.W. Garrison</author><author>J.J. Evans</author><author>J.K
Avants</author><author>Q. Teng</author></authors></title>><title>Determination
of perchlorate in tobacco plants and tobacco products</title><secondary-title>Environ. Sci.
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Technol.</secondary-title></title><periodical><full-title>Environ. Sci. Technol.</fulltitle></periodical><pages>3213-

3218</pages><volume>35</volume><number>15</number><dates><year>2001</year></dates><ur ls></urls></record></Cite></EndNote>], resulting in exposure to individuals who smoke, and is a component of gunpowder, exposing individuals who reload their own ammunition [ ADDIN **EN.CITE** 

<EndNote><Cite><Author>Lindner</Author><Year>1993</Year><RecNum>2040</RecNum><Dis playText>(Lindner, 1993)</br>

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Othmer encyclopedia of chemical technology</secondary-title></title></title> 125</pages><volume>10</volume><dates><year>1993</year></dates><pub-location>New York, NY</pub-location><publisher>John Wiley & amp;

Sons</publisher><urls></urls></record></Cite></EndNote>]. Small amounts of perchlorate are used in different types of medical imaging, resulting in exposure to members of the population undergoing health tests [ ADDIN EN.CITE

<EndNote><Cite><Author>Gibbs</Author><Year>1998</Year><RecNum>2046</RecNum><Displ ayText>(Gibbs et al., 1998)</DisplayText><record><rec-number>2046</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1541695841">2046</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Gibbs, J.P.</author><author>R.

Ahmad</author><author>K.S. Crump</author><author>D.P. Houck</author><author>T.S. Leveille</author><author>J.E. Findley</author><author>M.

Francis</author></authors></contributors></title>Evaluation of a population with occupational exposure to airborne ammonium perchlorate for possible acute or chronic effects on thyroid function</title><secondary-title>J. Occup. Environ. Med. </secondary-

title></title></periodical><full-title>J. Occup. Environ. Med.</full-title></periodical><pages>1072-1082</pages><volume>40</volume><number>12</number><dates><year>1998</year></dates><ur ls></urls></record></Cite></EndNote>]. Wan et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Wan</Author><Year>2015</Year><RecNum>2238</RecNum><Displ ayText>(2015)</DisplayText><record><rec-number>2238</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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B.</author><author>Kumosani, T.A.</author><author>Malarvannan, G.</author><author>Moon,
H. </author><author>Nakata, H.</author><author>Sinha, R.K. </author><author>Minh, T.B.
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perchlorate in indoor dust from the United States and eleven other countries: Implications for human
exposure</title><secondary-title>Environmental International </secondary-
title></titles><periodical><full-title>Environmental International</full-
title></periodical><pages>166-
171</pages><volume>75</volume>dates><year>2015</year></dates><urls></record></Cit
e></EndNote>] collected dust samples from various microenvironments. Concentrations of indoor
dust ranged from 0.03-1.18 \mu g/g. They estimated a median daily intake of perchlorate for toddlers
through ingestion of dust to be 1.89 ng/kg bodyweight/day in the United States [ ADDIN EN.CITE
<EndNote><Cite><Author>Wan</Author><Year>2015</Year><RecNum>2238</RecNum><Displa
yText>(Wan et al., 2015)</DisplayText><record><rec-number>2238</rec-number><foreign-
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type><contributors><author>Wan, Y.</author><author>Wu,
Q.</author><author>Abualnaja, K.O.</author><author>Asimakopoulos,
A.G. </author><author>Covaci, A. </author><author>Gevao, B. </author><author>Johnson-Restrepo.
B.</author><author>Kumosani, T.A. </author><author>Malarvannan, G.</author><author>Moon,
H. </author><author>Nakata, H.</author><author>Sinha, R.K. </author><author>Minh, T.B.
</author><author>Kannan, K.</author></authors></contributors><title>><title>Occurrence of
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title></periodical><pages>166-
171</pages><volume>75</volume><dates><year>2015</year></dates><urls></urls></record></Cit
e></EndNote>].
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#### 5.5 Occupational Exposures

Exposure to perchlorate may occur at facilities where perchlorate is manufactured or used [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><authors><authors><authors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author>

ls></urls></record></Cite></EndNote>].

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Ahmad</author><author>K.S. Crump</author><author>D.P. Houck</author><author>T.S.
Leveille</author><author>J.E. Findley</author><author>M.
Francis</author></author></authors></contributors><titles><title>Evaluation of a population with occupational exposure to airborne ammonium perchlorate for possible acute or chronic effects on thyroid function</title><secondary-title>J. Occup. Environ. Med. </secondary-title>
/title></periodical><pages>1072-1082</pages>
/pages></pages></pages></pages></pages>
/dates></page></page>
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Workers at an ammonium perchlorate facility received a 35 μg/kg average absorbed dose of perchlorate over a single-shift, with a range of 0.2 to 436 μg/kg. The lifetime cumulative dose, over an average of 8.3 years, ranged from 8,000 to 88,000 µg/kg [ ADDIN EN.CITE <EndNote><Cite><Author>Gibbs</Author><Year>1998</Year><RecNum>2046</RecNum><Displ ayText>(Gibbs et al., 1998)</DisplayText><record><rec-number>2046</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541695841">2046</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Gibbs, J.P.</author><author>R. Ahmad</author><author>K.S. Crump</author><author>D.P. Houck</author><author>T.S. Leveille</author><author>J.E. Findley</author><author>M. Francis</author></authors></contributors></title>Evaluation of a population with occupational exposure to airborne ammonium perchlorate for possible acute or chronic effects on thyroid function</title><secondary-title>J. Occup. Environ. Med. </secondarytitle></titles><periodical><full-title>J. Occup. Environ. Med.</full-title></periodical><pages>1072-1082</pages><volume>40</volume><number>12</number><dates><year>1998</year></dates><ur ls></urls></record></Cite></EndNote>]. Respirable air samples at an ammonium perchlorate manufacturing facility had an average perchlorate concentration of 0.091 mg/day for workers at low dust-forming operations and an average of 0.601 and 8.591 mg/day at moderate and high dustforming operations, respectively [ ADDIN EN.CITE <EndNote><Cite><Author>Lamm</Author><Year>1999</Year><RecNum>2090</RecNum><Disp layText>(Lamm et al., 1999)</DisplayText><record><rec-number>2090</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541771491">2090</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Lamm, S.H.</author><author>L.E. Braverman</author><author>F.X. Li</author><author>K. Richman</author><author>S. Pino</author><author>G. Howearth</author></authors></contributors><title>Thyroid health status of ammonium perchlorate workers: A cross-sectional occupational health study</title><secondary-title>J. Occup. Environ. Med. </secondary-title></title>><periodical><fulltitle>J. Occup. Environ. Med.</full-title></periodical><pages>248-260</pages><volume>41</volume><number>4</number><dates><year>1999</year></dates><urls ></urls></record></Cite></EndNote>].

#### 5.6 Summary

Perchlorate exposure for the general population is mostly through the ingestion of food. Perchlorate has been detected in a wide range of food items, including fruits, vegetables, and milk and has also been detected in beer, wine, dietary supplements, and vitamins. The Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1503519671">1917</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Abt, E.</author><author>Spungen,

J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title><secondary-title></title><secondary-title></title><secondary-title></title><secondary-title></title><secondary-title></title><secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></sec

title></titles><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>] study estimated average perchlorate exposure from food as follows: infants (6-11 months): 0.36 to 0.39  $\mu$ g/kg/day; children (2 – 10 years): 0.16 to 0.48  $\mu$ g/kg/day; women (25 – 70+ years): 0.09 to 0.12  $\mu$ g/kg/day; men (25 – 70+ years): 0.09 to 0.13  $\mu$ g/kg/day. EPA also analyzed the same data as Abt et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa yText>(2016)</DisplayText><record><rec-number>1917</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers, M.</author><author>Wirtz, M.</author></authors></contributors><titles><title>Update on dietary intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-

title></titles><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-

title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>] and additional upper bound exposure estimates with values ranging from 0.23 µg/kg/day to 1.17 µg/kg/day when looking at the mean to 99<sup>th</sup> percentile of intake rates and considering the 95<sup>th</sup> percentile of perchlorate in food for women of child bearing age. Perchlorate has also been detected in air, with higher concentrations detected after firework displays.

## 6. Background on Thyroid Hormone Physiology and Connection to Neurodevelopment

In order to understand the approach the EPA has selected to derive an MCLG in accordance with SDWA, it is first necessary to have a basic understanding of thyroid physiology because perchlorate's main target is the thyroid gland. Therefore, an overview of thyroid physiology (Section [ REF \_Ref427059872 \r \h ]), including during pregnancy (Section [ REF \_Ref530559488 \r \h ]) is provided. This is followed by a description of thyroid pathophysiology (Section [ REF \_Ref486250699 \r \h ]) and how thyroid hormones impact neurodevelopment (Section [ REF \_Ref427059920 \r \h ]). These sections will provide a basis for understanding the approach to derive an MCLG presented in this document.

#### 6.1 Overview of Thyroid Physiology

The thyroid gland is part of a "self-regulating" loop referred to as the hypothalamic-pituitary-thyroid (HPT) axis ([ REF \_Ref454999548  $\h$  ]), which controls metabolism, growth, and brain development and function [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller<Author><Year>2007

2007)</DisplayText><record><rec-number>186</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202457">186</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Zoeller, R Thomas</author><author>Tan, Shirlee W</author><author>Tyl, Rochelle

W</author></authors></contributors><titles><title>General background on the hypothalamic-pituitary-thyroid (HPT) axis</title><secondary-title>Critical Reviews in Toxicology</secondary-title></title>

title></periodical><pages>11-53</pages><volume>37</volume><number>1-

2</number><dates><year>2007</year></dates><isbn>1040-

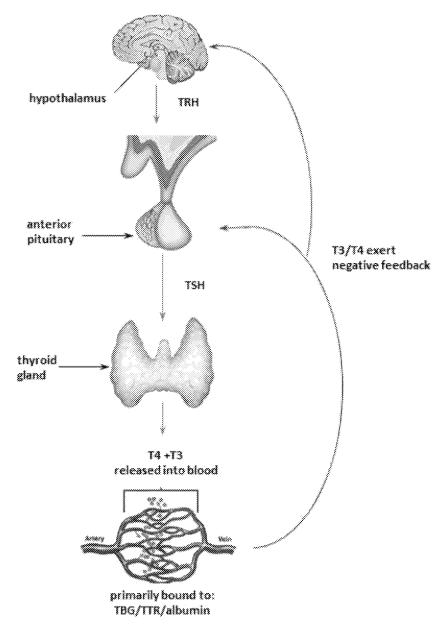
8444</isbn><urls></record></EndNote>]. Low circulating levels of thyroid hormones stimulate hypothalamic secretion of thyrotropin-releasing hormone (TRH), which diffuses into the anterior pituitary gland and stimulates the endocrine cells within the anterior pituitary (thyrotropes) to release TSH into the circulatory system. TSH acts on the thyroid gland to stimulate iodine uptake to result in thyroid hormone production. Iodide is taken up from circulation by the NIS; serum iodide is transported into thyroid gland follicles for synthesis into the thyroid hormones T4 and to a lesser extent T3. Endocrine disruptors, such as perchlorate, could interfere with this process by directly inhibiting the uptake of iodide, thereby reducing the production of thyroid hormone.

T3 and T4 are formed when thyroglobulin (Tg; the precursor protein of T3 and T4) produced by the thyroid follicular cells is iodinated by the enzyme thyroperoxidase, and the iodinated protein is cleaved. T4 is generally thought to act as the prohormone for the more biologically active T3, which is required for normal development of the central nervous system in fetuses and in infants, as well as for ultimate skeletal development and growth. However, it should be noted that the regulatory factors that modulate the delivery of T4 to the fetus are not fully understood. Both T3 and T4 are critical determinants of metabolic function in humans of all ages. T4 and T3 are released into circulation, where they are primarily bound to the carrier proteins thyroxine-binding globulin (TBG), transthyretin (TTR), albumin, and lipoproteins [ ADDIN EN.CITE ADDIN EN.CITE.DATA ].

T3 and T4 in the blood that are bound to these proteins act largely as reserves because they are less readily metabolized in the liver and cannot enter cells unless in free form [ ADDIN EN.CITE <EndNote><Cite><Author>Zoeller</Author><Year>2007</Year><RecNum>186</RecNum><Displ ayText>(Zoeller et al., 2007)</DisplayText><record><rec-number>186</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202457">186</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><authors><author>Zoeller, R Thomas</author><author>Tan, Shirlee W</author><author>Tyl, Rochelle W</author></authors></contributors><title>Sceneral background on the hypothalamic-pituitary-thyroid (HPT) axis</title><secondary-title>Critical Reviews in Toxicology</secondary-title></title>><periodical><full-title>Critical reviews in toxicology</full-title></periodical><pages>11-53</pages><volume>37</volume><number>1-2</number><dates><year>2007</year></dates><isbn>1040-8444</isbn><urls></urls></record></Cite></EndNote>]. However, in their free (or unbound) states (the free or unbound states of T4 and T3 are denoted as fT4 and fT3), both are available to be transported actively into specific cells. It is important to recognize that fT4 and fT3 are in a dynamic equilibrium with protein bound T<sub>4</sub> and T<sub>3</sub>, much as dissolved oxygen is in a dynamic equilibrium with that of oxygen bound to hemoglobin. T4 is transported through the cellular membrane by specific transporters and transformed by deiodinases into T3, which is then transferred to the nucleus to cause transcriptional changes. Circulating T3 is mostly derived from peripheral monodeionization of T4 [ ADDIN EN.CITE <EndNote><Cite><Author>Chanoine</Author><Year>1993</Year><RecNum>343</RecNum><Di splayText>(Chanoine, Braverman, & Earwell, 1993; Peeters & Earmer, Visser, 2017)</DisplayText><record><rec-number>343</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495028650">343</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>>cauthor>>cauthor><author>>ca J.P. </author><author>Braverman, L.E. </author><author>Farwell, A.P. </author></authors></contributors><titles><title>The thyroid gland is a major source of circulating T3 in the rat</title><secondary-title>Journal of Clinical Investigation</secondarytitle></title></title><periodical><full-title>Journal of Clinical Investigation</fulltitle></periodical><pages>2709-2713</pages><volume>91</volume><number>6</number><dates><year>1993</year></dates><url s></urls></record></Cite><Cite><Author>Peeters</Author><Year>2017</Year><RecNum>344</R ecNum><record><rec-number>344</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495029973">344</key></foreignkeys><ref-type name="Electronic Book">44</ref-type><contributors><author>Peeters, R.</author><author>Visser, T.</author></authors><secondary-authors><author>De Groot, LJ</author><author>Chrousos, G</author><author>Dungan, K</author><author>et al.</author></secondary-authors></contributors><title>Metabolism of thyroid hormone</title></title></dates><year>2017</year></dates><publisher>Endotext (Internet)</publisher><urls><related-urls><url><style face="underline" font="default" size="100%">https://www.ncbi.nlm.nih.gov/books/NBK285545/</style></url></relatedurls></urls></record></Cite></EndNote>]. Circulating T3 and T4 levels in an individual are maintained within a narrow range by a negative feedback loop with TSH from the pituitary and TRH from the hypothalamus (see [ REF Ref454999548 \h ]) that operates around a "set point." This set point is different from individual to individual, which generates a population variance in blood levels of thyroid hormone that is considerably broader than the individual variance [ ADDIN EN.CITE

<EndNote><Cite><Author>Andersen</Author><Year>2002</Year><RecNum>333</RecNum><Di splayText>(Andersen, Pedersen, Bruun, & DisplayText><record><recnumber>333</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1494274249">333</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Andersen, S.</author><author>Pedersen, K.M.</author><author>Bruun, N.H.</author><author>Laurberg, P.</author></authors></contributors><title>Narrow individual variations in serum T(4) and T(3) in normal subjects: a clue to the understanding of subclinical thyroid disease</title><secondarytitle>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><fulltitle>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1068-1072</pages><volume>87</volume><number>3</number><dates><year>2002</year></dates><url s></urls></record></Cite></EndNote>]. Therefore, in euthyroid individuals, serum T4 and T3 fluctuate within a fairly narrow range (about 10% of the population variance), maintained by the negative feedback relationship with serum TSH from the pituitary gland. This normal variation creates a situation where single measures of free or total T4 and TSH are a somewhat imprecise measure of an individual's average T4 and TSH concentrations [ ADDIN EN.CITE <EndNote><Cite><Author>Andersen</Author><Year>2002</Year><RecNum>333</RecNum><Di splayText>(Andersen et al., 2002)</DisplayText><record><rec-number>333</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1494274249">333</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Andersen, S.</author><author>Pedersen, K.M.</author><author>Bruun, N.H.</author><author>Laurberg, P.</author></authors></contributors></title>Narrow individual variations in serum T(4) and T(3) in normal subjects: a clue to the understanding of subclinical thyroid disease</title><secondarytitle>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><fulltitle>Journal of Clinical Endocrinology and Mctabolism</full-title></periodical><pages>1068-1072</pages><volume>87</volume><number>3</number><dates><year>2002</year></dates><url s></urls></record></Cite></EndNote>].

Figure [ SEQ Figure \\* ARABIC ]. Summary of HPT Axis Showing Negative Feedback of Increased Circulating T4 and T3 on Hypothalamus and Anterior Pituitary



It is generally thought that changes in thyroid hormone levels that occur during pregnancy operate independently of changes in the set point. For example, as the first trimester progresses, increases in human chorionic gonadotropin (hCG<sup>7</sup>; which operates outside the HPT axis), estrogen, and thyroxine-

binding protein result in a higher blood concentration of total T4, making more fT4 available to the developing fetus [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de Escobar</Author><Year>2007</Year><RecNum>201</RecNum><DisplayText>(G. Morreale de Escobar, Obregón, &amp; Escobar del Rey, 2007)</DisplayText><record><rec-number>201</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466438813">201</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Morreale de Escobar, G</author><author>Obregón, María Jesús</author><author>Escobar del Rey, F. </author></authors></contributors><titles><title>Iodine deficiency and brain development in the first half of pregnancy</title><secondary-title>Public Health Nutrition</secondary-title></title></periodical><pages>1554-

1570</pages><volume>10</volume><number>12A</number><dates><year>2007</year></dates><isbn>1475-2727</isbn><urls></record></Cite></EndNote>]. Serum TSH concentrations are also mildly suppressed due to the negative feedback of this elevated serum T4 on TSH.

#### 6.2 Thyroid Physiology in Pregnancy

In general, pregnancy requires the maternal thyroid gland to increase thyroid hormone production by around 50% [ ADDIN EN.CITE

<EndNote><Cite><Author>Skeaff</Author><Year>2011</Year><RecNum>180</RecNum><Prefix >for review see </Prefix><DisplayText>(for review see Skeaff, 2011)</DisplayText><record><recnumber>180</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466200929">180</key></foreign-

keys><ref-type name="Journal Article">17</ref-type><contributors><author>Skeaff, Sheila A</author></author>></contributors><title>Iodine deficiency in pregnancy: the effect on neurodevelopment in the child</title><secondary-title>Nutrients</secondary-

title></title></periodical><full-title>Nutrients</full-title></periodical><pages>265-

273</pages><volume>3</volume><number>2</number><dates><year>2011</year></dates><urls></urls></record></Cite></EndNote>] in order to provide sufficient fT4 for fetal brain development. In the first trimester of pregnancy, this increased output of thyroid hormones is, in part, due to estrogen increases during the first trimester, which increases the serum concentration of TBG, one of the main proteins to which T4 binds. The combination of increased TBG and hCG results in an increase in total T4 of about 50% over the pre-pregnancy level [ ADDIN EN.CITE

<EndNote><Cite><Author>Alexander
/Author><Year><2017</p>
/Year><RecNum>1895
/RecNum>
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(Alexander et al., 2017)
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N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan,
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S.</author></authors></contributors><titles><title>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondary-title>Thyroid</secondary-title></title></periodical><full-title>Thyroid</full-title></periodical><pages>315-

389</pages><volume>27</volume><number>3</number><dates><year>2017</year></dates><urls></urls></record></Cite></EndNote>]. This increase in serum total T4 also increases serum fT4 during the first trimester, although fT4 levels fall over the next two trimesters [ ADDIN EN.CITE <EndNote><Cite><Author>Alexander</Author><Year>2017</Year><RecNum>1895</RecNum>< DisplayText>(Alexander et al., 2017)</DisplayText><record><rec-number>1895</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497970921">1895</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Alexander, E. K.</author><author>Pearce, E. N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan,

S.</author></authors></contributors><titles><title>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondary-title>Thyroid</secondary-title></title></periodical><full-title>Thyroid</full-title></periodical><pages>315-

389</pages><volume>27</volume><number>3</number><dates><year>2017</year></dates><urls></urls></record></Cite></EndNote>]. The first-trimester increase in fT4 can lead in turn to a lower TSH concentration compared to the pre-pregnancy state.

Given the increased demand on the maternal thyroid in pregnancy, there is an increased demand for

iodine to support additional hormone synthesis. If the maternal thyroid is fully functional, then sufficient dietary iodine intake will enable this to occur. According to the SAB, maternal hypothyroxinemia has been defined by a "variety of cutoffs...ranging from fT4 below the 10th or 5th percentiles to below the 2.5th percentile" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Pages>1 0</Pages><DisplayText>(SAB, 2013, p. 10)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB,</author></author></author> authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></title> location>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>] in the population. Although a fully functional system (i.e., in euthyroid individuals) can compensate with additional iodine consumption, hypothyroxinemic individuals are already compromised, and iodine insufficiency can exacerbate this situation, putting the fetus at risk of neurological deficits [ ADDIN ADDIN EN.CITE.DATA ]. Maternal fT4 is particularly important in the first trimester of pregnancy because the fetal brain is solely dependent upon T4 of maternal origin during this period [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller</Author><Year>2004</Year><RecNum>194</RecNum><DisplayText>(Zoeller & DisplayText><record><rec-number>194</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Zoeller, R Thomas</author><author>Rovet, J</author></author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author

#### 6.3 Thyroid Pathophysiology

Historically, the most commonly observed thyroid conditions have been goiter (i.e., enlarged thyroid gland) and congenital hypothyroidism. Hypothyroidism is defined as serum fT4 below and serum TSH above the reference range. Hypothyroidism may result from iodine insufficiency, from genetic disorders, autoimmune disease, cancer, pituitary insufficiency, or environmental factors. The most common form of adult hypothyroidism results from an autoimmune thyroiditis (Hashimoto's disease) whereby a variety of cellular and antibody-related immune processes reduce thyroid hormone production within the gland [ ADDIN EN.CITE

<EndNote><Cite><Author>Pyzik</Author><Year>2015</Year><RccNum>198</RecNum><Prefix >for review see </Prefix><DisplayText>(for review see Pyzik, Grywalska, Matyjaszek-Matuszek, &amp; Roliński, 2015)
/DisplayText><record><rec-number>198</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</p>

timestamp="1466427626">198</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pyzik, Aleksandra</author><author>Grywalska, Ewelina</author><author>Matyjaszek-Matuszek, Beata</author><author>Roliński, Jacek</author></authors></contributors><titles><title>Immune disorders in Hashimoto's Thyroiditis: What do we know so far?</title><secondary-title>Journal of Immunology Research</secondary-title></title><periodical><full-title>Journal of immunology research</full-title></periodical><volume>2015</volume><dates><year>2015</year></dates><isbn>2314-8861</isbn><urls></record></Cite></EndNote>]. Both post-partum thyroiditis and subacute lymphocytic (or silent) thyroiditis are considered subtypes of this disorder. Post-partum thyroiditis can occur in up to 5% of women and can involve periods of hyper- and hypo-thyroid activity [

<EndNote><Cite><Author>Akamizu
/Author><Year><2015</p>
/Year><RecNum>169
/RecNum>
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/Prefix><DisplayText>
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J.</author><author>Rebar, R. </author><author>Singer, F. </author><author>Vinik, A. </author></secondary-authors></contributors><titles><title>Hashimoto's
Thyroiditis</title><secondary-title>Endotext </secondary-title></title></title></dates><quer><2015</year></dates><pub-location>South Dartmouth, MA</pub-location><publisher>MDText.com,

Inc.</publisher><urls></urls></record></Cite><Cite><Author>Pearce</Author><Year>2015</Year></RecNum>179</RecNum><record><rec-number>179</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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E.N.</author></authors></contributors><title>>Thyroid disorders during pregnancy and postpartum</title><secondary-title>Best Practice & Research Clinical Obstetrics & Research Clinical O

resolve itself; however, as many as 20% of cases may lead to permanent hypothyroidism.

Hashimoto's disease increases the risk of hypothyroidism, including both subclinical and overt hypothyroidism. Subclinical hypothyroidism is defined by only elevated (or outside the reference range) TSH concentrations, with serum T3 and T4 concentrations that remain within the population reference range, while overt hypothyroidism is defined by elevated serum TSH concentrations concurrent with decreased (or outside the reference range) serum T3 and T4 concentrations. Subclinical hypothyroidism has the potential to progress to overt hypothyroidism over time; thus, it requires serum thyroid function test monitoring. Other related autoimmune hypothyroid conditions include primary thyroid atrophy (Ord's disease), which is characterized by thyroid atrophy through invasion by T-lymphocytes, and is likely part of the continuum of hypothyroid conditions, rather than a separate disorder from Hashimoto's disease [ ADDIN EN.CITE

<EndNote><Cite><Author>Carlé</Author><Year>2009</Year><RecNum>172</RecNum><Displa yText>(Carlé et al., 2009)</DisplayText><record><rec-number>172</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1466200413">172</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Carlé, Allan</author><author>Pedersen, Inge Bulow</author><author>Knudsen, Nils</author><author>Perrild, Hans</author><author>Ovesen, Lars</author><author>Jørgensen, Torben</author><author>Laurberg,

Peter</author></authors></contributors></title>Thyroid volume in hypothyroidism due to autoimmune disease follows a unimodal distribution: evidence against primary thyroid atrophy and autoimmune thyroiditis being distinct diseases</title><secondary-title>The Journal of Clinical Endocrinology & Description (Secondary-title) (Secondary-title) (Periodical) (Secondary-title) (Periodical) (Periodi

839</pages><volume>94</volume><number>3</number><dates><year>2009</year></dates><isbn>0021-972X</isbn><urls></record></Cite></EndNote>]. Viral or other infections can result in the painful condition of subacute granulomatous thyroiditis. This condition, like post-partum thyroiditis, is generally transient and requires only beta-blocker and non-steroidal anti-inflammatory drugs (NSAIDs) or steroid treatment to resolve. Although not common, pituitary disorders or tumors can also result in reduced thyroid function through reduction in TSH output.

Thyroid peroxidase antibodies (also referred to as anti-thyroid peroxidase antibodies or TPO Abs) are a marker for thyroid autoimmunity, a condition in which a person's immune system antibodies mistakenly target parts of the thyroid gland or thyroid proteins. Thyroid autoimmunity leads to chronic inflammation of the thyroid, can cause tissue damage, and may disrupt thyroid function. The TPO Abs work against thyroid peroxidase, which is an enzyme involved in the synthesis of thyroid hormones [ ADDIN EN.CITE

<EndNote><Cite><Author>Braverman
/Author><Year>
2004</Year>
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key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
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/key></foreign-keys>
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### 6.4 Physiologic Connection between Thyroid Hormone Levels and Neurodevelopment

The profound effects of thyroid deficiencies during development were first documented in children with congenital hypothyroidism by Paracelsus [ ADDIN EN.CITE <EndNotc><Cite><Author>Cranefield</Author><Year>1963</Year><RecNum>1892</RecNum>< DisplayText>(Cranefield & DisplayText><record><rec-number>1892</rec-number>1892</rec-number> number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497969366">1892</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Cranefield, P.</author><author>Federn, W.</author></authors></contributors></title>Paracelsus on goiter and cretinism: A translation and discussion of " De Struma, Vulgo Der Kropf" </title><secondarytitle>Bulletin of the History of Medicine</secondary-title></titles><periodical><full-title>Bulletin of the History of Medicine</full-title></periodical><pages>463-471</pages><volume>37</volume><dates><year>1963</year></dates><urls></urls></record></Cit e></EndNote>]. Modern studies demonstrate that unless treated, children with congenital hypothyroidism will suffer from problems associated with somatic growth and an array of visuomotor and neurocognitive deficits. Aberrant patterns of growth, development, and organization of the cortex, cerebellum, hippocampus, basal ganglia, and brain stem can all result from reduced thyroid hormone exposure during development [ ADDIN EN.CITE <EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(Williams, 2008; Zoeller & DisplayText><record><recnumber>193</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202836">193</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Williams, GR</author></authors></contributors></title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondarytitle></titles><periodical><full-title>Journal of neuroendocrinology</fulltitle></periodical><pages>784794</pages><volume>20</volume><number>6</number><dates><year>2008</year></dates><isbn >1365-

2826</isbn><urls></record></cite><Cite><Author>Zoeller</Author><Year>2004</Year></record></cite><Cite><Author>Zoeller</Author><Year>2004</Year></rec>
RecNum>194</RecNum><record><rec-number>194</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Zoeller, R Thomas</author><author>Rovet, J</author></authors></contributors><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings</title><secondary-title>Journal of Neuroendocrinology</secondary-title>

\*\*Vittle></periodical><full-title>Journal of neuroendocrinology

\*\*Journal of ne

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>1365-2826</isbn><urls></record></Cite></EndNote>].

Thyroid hormones are essential for the development and differentiation of the developing brain. The brain and spinal cord begin development in the first half of the first trimester. Free T4 can reach the brain, though the exact mechanism by which is does so is actively being researched [ ADDIN EN.CITE

<EndNote><Cite><Author>Wirth</Author><Year>2014</Year><RecNum>2233</RecNum><Prefix>e.g.`, </Prefix><DisplayText>(e.g., Wirth, Schweizer, & DisplayText)

2014)</br>
2014)
DisplayText><record><rec-number>2233</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543328103">2233</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>>Wirth, E.K.</author><author>Schweizer, U.</author><author>Kohrle,

J.</author></authors></contributors><title>>Transport of thyroid hormone in brain</title><secondary-title>Frontiers in Endocrinology</secondary-

title></titles><periodical><full-title>Frontiers in endocrinology</full-title></periodical><pages>1-7</pages><volume>5</volume><dates><year>2014</year></dates><urls></urls></record></Cite><EndNote>]. However, once present in the brain, T4 is converted to T3 by the developing glial cells and then transported to neurons. T3 then interacts with nuclear receptors to tightly regulate gene expression so that neurogenesis, synaptogenesis, neuronal migration, cell differentiation, and myelination are developmentally appropriate. Deficiencies in thyroid hormones through iodine deficiency, congenital hypothyroidism, or maternal hypothyroidism/hypothyroxinemia can result in neurological impairments and intellectual deficits [ ADDIN EN.CITE

<EndNote><Cite><Author>Morreale de

Escobar</br>
Escobar
Author><Year>2000
/Year><RecNum>1885
/RecNum><DisplayText><(G. Morreale de Escobar, Obregón, & DisplayText>
rec-number>1885
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timestamp="1496433675">1885
/key>
/foreign-keys><ref-type name="Journal Article">17
/ref-type><contributors><author>>Morreale de Escobar, G</author><author>Obregón, M
J
/author><author>Escobar del Rey, F.</author></authors></contributors><title>><title>Is
neuropsychological development related to maternal hypothyroidism or to maternal hypothyroxinemia?
//title><secondary-title>Journal of Clinical Endocrinology and
Metabolism
/secondary-title>
/title>
/pages>3975-

3987</pages><volume>85</volume><number>11</number><dates><year>2000</year></dates><urls></urls></record></Cite></EndNote>].

The fetal thyroid gland first begins to concentrate iodide at about 11–12 weeks and begins significant contribution to fetal development around 16 weeks. However, the developing fetus still relies on some supply of maternal thyroid hormones until birth [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de

Escobar
Escobar
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(G. Morreale de Escobar, Obregón, & DisplayText
Escobar del Rey, 2004
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"Journal Article"
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Morreale de Escobar, G
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Obregón, M
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Escobar del Rey, F
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Escobar
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U37</pages><volume>151</volume><number>Suppl 3</number><dates><year>2004</year><published
dates><date>November 1, 2004</date></published
dates><dates><urls><relatedurls><url>http://www.eje-online.org/content/151/Suppl\_3/U25.abstract</url></relatedurls></urls><electronic-resource-num>10.1530/eje.0.151U025</electronic-resourcenum></record></Cite></EndNote>]. The fetal thyroid is not fully mature until birth, so premature
birth can also be associated with low thyroid hormone concentrations. Thus, alterations in thyroid
hormone levels can impact brain development in a variety of ways depending on when the alteration
occurs and for how long.

For example, early in brain development, fT4 deficiency can affect neuronal cell proliferation and cell migration in the hippocampus (essential for learning and memory, cognitive function), the cerebral cortex (essential for executive function, cognitive function), and the medial ganglionic eminence (transitory development structure responsible for guiding axonal migration) [ ADDIN EN.CITE <EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(Williams, 2008)</DisplayText><record><rec-number>193</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Williams,

GR</author></authors></contributors><titles><title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title></periodical><full-title>Journal of neuroendocrinology</full-title></periodical><pages>784-

794</pages><volume>20</volume>6</number>6</number><dates><year>2008</year></dates><isbn>1365-2826</isbn><urls></record></Cite></EndNote>]. Maternal hypothyroidism or hypothyroxinemia at these early stages may result in problems with gross motor skills and visual attention and processing [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller
Author><Year><2004</p>
Year><RecNum>
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foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</p>
timestamp="1466202917">194
/key>
foreign-keys><ref-type name="Journal Article">17
ref-type>
contributors><author>Zoeller, R Thomas</author><author>Rovet,
J
/author></author>
/author>
/contributors><title></title>Timing of thyroid hormone action in the
developing brain: clinical observations and experimental findings
/title><secondary-title>
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ref-type name="Journal Article">17
ref-type
contributors
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Neuroendocrinology
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neuroendocrinology</full-title></periodical><pages>809-

818 < pages > < volume > 16 < / volume > < number > 10 < / number > < dates > < year > 2004 < / year > < / dates > < isb n > 1365 - 2826 < / isbn > < / volume > < / Cite > < / EndNote > ]. As the first trimester progresses, increases in hCG result in a higher production of T4 [ ADDIN EN.CITE

<EndNote><Cite><Author>Morreale de

Escobar</Author><Year>2007</Year><RecNum>201</RecNum><DisplayText>(G. Morreale de Escobar et al., 2007)</DisplayText><record><rec-number>201</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1466438813">201</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Morreale de Escobar, G</author><author>Obregón, María Jesús</author><author>Escobar del Rey, F. </author></author></contributors><title>><title>Iodine deficiency and brain development in the first half of pregnancy</title><secondary-title>Public Health Nutrition</secondary-title></title>>cperiodical><full-title>Public health nutrition</full-title></periodical><pages>1554-

1570</pages><volume>10</volume><number>12A</number><dates><year>2007</year></dates><isbn>1475-2727</isbn><urls></record></Cite></EndNote>]. In addition, serum TBG increases during the first trimester, remaining elevated throughout pregnancy because estrogen increases the glycosylation of TBG, increasing its serum half-life. In the fetus this period also corresponds to neurogenesis in the midbrain and cortex, neuronal migration into the cortex, deiodinase activity for conversion of T4 to T3, and presence of nuclear thyroid receptors in the brain [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de

Escobar </br>
Escobar 
Author > < Year > 2004 < / Year > < RecNum > 49 < / RecNum > < Display Text > (G. Morreale de Escobar et al., 2004) < / Display Text > < record > < rec-number > 49 < / rec-number > < foreign-keys > < key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</td>

timestamp="1437077734">49</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Morreale de Escobar, G</author><author>Obregón, M J</author><author>Escobar del Rey, F</author></author></contributors><title>Role of thyroid hormone during early brain development</title><secondary-title>European Journal of Endocrinology</secondary-title></title><periodical><full-title>European Journal of Endocrinology</full-title></periodical><pages>U25-

U37</pages><volume>151</volume><number>Suppl 3</number><dates><year>2004</year><pubdates><date>November 1, 2004</date></pub-dates></dates><urls><related-urls><url>http://www.eje-online.org/content/151/Suppl\_3/U25.abstract</url></related-urls></urls><electronic-resource-num>10.1530/eje.0.151U025</electronic-resource-num></record></Cite></EndNote>].

Higher levels of maternal fT4 throughout pregnancy allow for normal neuron and glial cell migration and differentiation, axonal growth, dendritic branching, neurogenesis, and synaptogenesis to occur [ ADDIN EN.CITE

<EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(Williams, 2008)</DisplayText><record><rec-number>193</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</ref-

timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Williams,

GR</author></contributors></title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title></full-title>Journal of neuroendocrinology</full-

2016).

title></periodical><pages>784-

794</pages><volume>20</volume><number>6</number><dates><year>2008</year></dates><isbn >1365-2826</isbn><urls></record></Cite></EndNote>]. The complex interconnectedness of brain regions required for normal function could be impaired by effects on all of these processes simultaneously or in sequence, and across multiple brain regions. Maternal thyroid deficiencies during these processes further affect visual processing, development of fine motor skills, IQ, and selective learning problems [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller</Author><Year>2004</Year><RecNum>194</RecNum><DisplayText>(Zoeller & DisplayText><record><recnumber>194</rec-number><foreign-keys><key app="EN" db-"

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>>Zoeller, R Thomas</author><author>Rovet, J</author></author>></contributors><title>><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings</title><secondary-title>Journal of Neuroendocrinology</secondary-title>
/title></periodical><pages>809-

818 < pages > < volume > 16 < / volume > < number > 10 < / number > < dates > < year > 2004 < / year > < / dates > < isb n > 1365 -

2826</isbn><urls></record></Cite><Cite><Author>Zoeller</Author><Year>2007</Year></record></record></cite><Cite><Author>Zoeller</Author><Year>2007</Year></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></record></re

Additionally, humans continue to undergo considerable post-natal brain development well through puberty and into early adulthood. A properly functioning HPT is necessary for proper neurologic development at this stage [ ADDIN EN.CITE

there may be adverse effects associated with fT4 levels above the normal range (Korevaar et al.,

<EndNote><Cite><Author>Zoeller
Author><Year><2004</p>
Year><2004</p>
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Year

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>1365-2826</isbn><urls></record></Cite></EndNote>]. This is particularly true in the

hippocampus, which is considered to have a relatively high degree of neuroplasticity in humans and undergoes considerable adult neurogenesis. How the behavioral and cognitive deficits associated with subclinical hypothyroidism and hypothyroxinemia are manifested within the brain is a research area where further evaluation is needed.

#### 7. Overview of Perchlorate's Impact on the Thyroid

As stated previously, the main target for perchlorate's toxicity is the thyroid gland. In this section the mode of action for perchlorate is discussed along with an overview of studies that have evaluated perchlorate's impact on the human thyroid.

#### 7.1 Mode of Action

In 2005, the NRC proposed a mode of action for perchlorate. [ REF \_Ref455000569 \h ] is a reproduction of this mode of action modified to include hypothyroxinemia. Evaluating hypothyroxinemia as a result of perchlorate exposure is the mode of action that the SAB recommended the EPA utilize in evaluating the MCLG for perchlorate. [ REF \_Ref455000569 \h ] begins with exposure to perchlorate, which then leads to perchlorate in the bloodstream, followed by the inhibition of iodide uptake in the thyroid. The NRC concluded that these steps have all been observed in humans, and any subsequent steps or results are biologically possible, but have not been clearly demonstrated in humans. However, since the 2005 NRC report was published, several epidemiological studies have shown perchlorate exposure to be associated with changes in serum thyroid hormone levels [ ADDIN EN.CITE \_ ADDIN EN.CITE.DATA \_ ]. Furthermore, Taylor et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Taylor</Author><Year>2014</Year><RecNum>2032</RecNum><DisplayText>(2014)</DisplayText><record><rec-number>2032</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1541511371">2032</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Taylor, P.</author><author>Okosieme, O.

E.</author><author>Murphy, R.</author><author>Hales, C.</author><author>Chiusano,

E.</author><author>Maina, A., . . . </author><author>Paradice,

R.</author></authors></contributors><titles><title>Maternal perchlorate levels in women with borderline thyroid function during pregnancy and the cognitive development of their offspring: data from the controlled antenatal thyroid study</title><secondary-title>The Journal of Clinical Endocrinology & Metabolism</secondary-title></title><periodical><full-title>The Journal of Clinical Endocrinology & Metabolism</secondary-title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

title></periodical><pages>8</pages><volume>99</volume><section>4291</section><dates><year> 2014</year></dates><urls></record></Cite></EndNote>] demonstrated an association between high maternal perchlorate exposure and risk of low IQ in offspring.

In their review, NRC deemed the inhibition of iodide uptake in the thyroid to be a non-adverse effect, and concluded that the body's compensatory mechanisms would keep thyroid hormone levels within the population reference range [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-type><contributors><author>NRC</author></author></contributors><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><pub-

location>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-

date>National Research Council</modified-date></record></Cite></EndNote>]. However, concerns have been raised regarding this conclusion given the lack of empirical evidence provided to support it [ ADDIN EN.CITE

<EndNote><Cite><Author>Ginsberg</Author><Year>2007</Year><RecNum>345</RecNum><Dis playText>(Ginsberg, Hattis, Zoeller, & DisplayText><record><recnumber>345</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495122121">345</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Ginsberg, G.</author><author>Hattis, R.</author><author>Zoeller, R Thomas</author><author>Rice, D.C.</author></authors></contributors><title>Evaluation of the U.S. EPA/OSWER Preliminary Remediation Goal for Perchlorate in Groundwater: Focus on Exposure to Nursing Infants</title><secondary-title>Environmental Health Perspectives</secondarytitle></title></title><periodical><full-title>Environmental Health Perspectives</fulltitle></periodical><volume>115</volume><number>3</number><section>361</section><dates><y ear>2007</year></dates><urls></record></Cite></EndNote>]. The NRC did list subsequent biologically plausible steps, downstream from iodine uptake inhibition, that include the following: altered thyroid hormone levels (e.g., decreased T3 and T4, increased TSH; which has now been demonstrated in the epidemiologic literature), growth of the thyroid gland (e.g., thyroid hypertrophy and hyperplasia), and hypothyroidism.

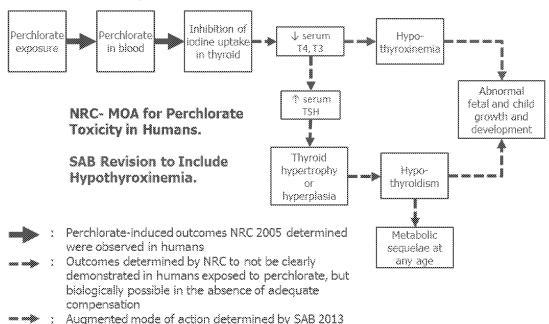
Hypothyroidism, deemed by the NRC to be the first adverse effect, could lead to abnormal fetal and child growth development, as well as metabolic disorders at all ages. Given the importance of a properly functioning thyroid for fetal and child growth [ ADDIN EN.CITE <EndNote><Cite><Author>Forhead</Author><Year>2014</Year><RecNum>1890</RecNum><Di splayText>(Forhead & DisplayText) (Forhead & Provided Amp; Fowden, 2014) (DisplayText) (Provided Amp; Fowden, 2014) (DisplayText) (Provided Amp; Fowden, 2014) (P number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497623515">1890</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Forhead, A J</author><author>Fowden, A L</author></authors></contributors></title>Thyroid hormones in fetal growth and prepartum maturation</title><secondary-title>Journal of Endocrinology</secondarytitle></titles><periodical><full-title>Journal of Endocrinology</full-title></periodical><pages>R87-R103</pages><volume>221</volume><number>3</number><dates><year>2014</year><pubdates><date>June 1, 2014</date></pub-dates></dates><urls><relatedurls><url>http://joe.endocrinology-journals.org/content/221/3/R87.abstract</url></relatedurls></urls><electronic-resource-num>10.1530/joc-14-0025</electronic-resourcenum></record></Cite></EndNote>], both hypothyroidism and hypothyroxinemia (low fT4 with TSH in the reference range) may also be associated with abnormal fetal and child growth.

In the SAB report on approaches to derive an MCLG for perchlorate, the SAB indicates that "the mode of action of perchlorate toxicity is well understood" [ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Pages>2 </Pages><DisplayText>(SAB, 2013, p. 2)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>SAB,</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></author></a></author></a></author></a></author></a></al>

goal for perchlorate. EPA-SAB-13-004</title></title></date>><year>2013</year></dates><publocation>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>]. The SAB finds that "hypothyroxinemia (i.e., low levels of thyroid hormone) is a more appropriate indicator of the potential adverse health effects than the more pronounced decreases in thyroid hormone associated with hypothyroidism" ([ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(SAB, 2013)</br>

Text>(sAB, 2013)
PisplayText><rec-number>50</rec-number><foreign-keys><key</td> app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB,</author></author></author> authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></title> location>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>], p. 2). Moreover, the "sensitive populations EPA should consider for exposure to perchlorate are the fetuses of hypothyroxinemic pregnant women" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Pages>2 </Pages><DisplayText>(SAB, 2013, p. 2)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB.</author></author></author> authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><year>2013</year></dates><publocation>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>].

# Figure [ SEQ Figure \\* ARABIC ]. Modified Representation of NRC's Suggested Mode of Action for Perchlorate Toxicity in Humans Indicating First Adverse Effect in the Continuum of Perchlorate Exposure to Effect, Revised to Include Hypothyroxinemia as recommended by the SAB



Source: Adapted from NRC [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>National Research Council (NRC)</Author><Year>2005</Year><RecNum>116</RecNum><Suffix>`, Figure 5-2 (modified)`, Figure 5-2 (modified)`, Figure 5-2 (modified), Figure 5-2 (m

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914312">116</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><authors><authors><authors>contributors><authors>contributors><authors>contributors>co

(NRC), </author></authors></contributors><titles><title>Health implications of perchlorate

exposure</title></title></title></title></title></title></title>></title></title></title></title></title></title></title></title>></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

Notes: Solid arrows represent outcomes that the NRC [ ADDIN EN.CITE < EndNote> < Cite Exclude Auth="1" > < Author > National Research Council (NRC) < / Author > < Year > 2005 < / Year > < RecNum > 116 < / RecNum > Council (NRC) < / Year > < Year > 116 < Year > (2005) <

number>116</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1465914312">116</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>National Research Council (NRC),</author></authors></contributors><title>Health implications of perchlorate exposure</title></title><dates><year>2005</year></dates><publication>Washington, D.C.</publication>Washington, D.C.

location><publisher>The National Academies Press</publisher><urls></urls></record></Cite></EndNote>] determined were observed in humans during perchlorate exposure. Black dashed arrows represent outcomes determined by the NRC [ ADDIN EN.CITE <EndNote><Cite ExcludeYear="1"><Author>National Research Council

(NRC)</Author><Year>2005</Year><RecNum>116</RecNum><DisplayText>(National Research Council

(NRC))</DisplayText><record><rec-number>116</rec-number><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914312">116</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><authors><author>National Research Council

(NRC),</author></contributors></title>Health implications of perchlorate

exposure</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>Active<th

## 7.2 Overview of Evidence Evaluating Perchlorate's Impact on the Human Thyroid

The proposed mode of action is informed by the knowledge developed from decades of research on perchlorate's impact on the thyroid gland conducted both through clinical trials and, more recently, through environmental epidemiological studies. Given the abundance of human data and the differences in serum binding and clearance rates and differences in thyroid stimulation by a placental hormone in pregnant women between rats and humans (NRC, 2005) the Agency does not summarize the toxicological literature in this section. Instead an overview of clinical studies and environmental epidemiological studies, evaluating the association between perchlorate exposure (measured as either a given dose or through biomonitoring) and impacts on measures of thyroid homeostasis are summarized.

#### 7.2.1 Clinical Studies

Understanding perchlorate's impact on the thyroid in a clinical setting has occurred since at least the 1950s. For example, Stanbury and Wyngaarden [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Stanbury</Author><Year>1952</Year><RecNum>2192</RecNum><D isplayText>(1952)</DisplayText><record><rec-number>2192</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541782564">2192</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Stanbury, J.B. </author><author>J.B. Wyngaarden</author></authors></contributors></title>Effect of perchlorate on the human thyroid gland</title><secondary-title>Metabolism</secondary-title></title><periodical><fulltitle>Metabolism</full-title><abbr-1>Metabolism</abbr-1></periodical><pages>533-539</pages><volume>1</volume><number>6</number><dates><year>1952</year></dates><urls> </urls></record></Cite></EndNote>], Godley and Stanbury [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>A.F.</Author><Year>1954</Year><RecNum>2113</RecNum><Displ ayText>(1954)</br>

ayText>
rec-number>
foreign-keys><key</td> app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541774730">2113</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Godley, A.F. </author><author>Stanbury, J.B.</author></authors></contributors></title>Preliminary experience in the treatment of hyperthyroidism with potassium perchlorate. </title><secondary-title>J. Clin. Endocrinol. </secondary-title></title></full-title>J. Clin. Endocrinol.</fulltitle></periodical><pages>70-78</pages><volume>14</volume><dates><vear>1954</vear></dates><urls></urls></record></Cite ></EndNote>], and Burgi et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Bürgi</Author><Year>1974</Year><RecNum>2115</RecNum><Disp layText>(1974)</DisplayText><record><rec-number>2115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541774836">2115</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Bürgi, H.</author><author>M. Benguerel</author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><aut Studer</author></authors></contributors><title>Influence of perchlorate on the secretion of non-thyroxine iodine by the normal human thyroid gland</title><secondary-title>Eur. J. Clin. Invest. </secondary-title></title>><periodical><full-title>Eur. J. Clin. Invest.</fulltitle></periodical><pages>6569</pages><volume>4</volume><dates><year>1974</year></dates><urls></urls></record></Cite></EndNote>] all noted perchlorate's impact on reducing iodine uptake by the thyroid gland.

In 1992, Brabant et al. used perchlorate to study changes in temporal patterns of TSH response to a decrease in iodine supply. Specifically, in this study the authors treated 5 healthy male volunteers (aged 25-28) with 200  $\mu$ g/day of iodine for 4 weeks. This was then followed by treatment of 300 mg of perchlorate three times per day over a four week period to induce an iodine depleted state in the study participants. As anticipated, the authors did see a significant decrease in iodine uptake with the perchlorate treatment. Further, the authors also noted that 24-hour mean serum TSH, TSH per pulse and serum fT4 levels were significantly decreased during treatment period. The authors did not observe any change in thyroid volume and total serum T3, T4 or fT3.

In another study, Lawrence et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Lawrence</Author><Year><2000</Year><RecNum><28</RecNum><D isplayText>(2000)</DisplayText><record><rec-number>228</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468246590">228</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Lawrence, J.E.</author><author>S.H. Lamm</author><author>S. Pino</author><author>K. Richman</author><author>Braverman, L</author></authors></contributors></title>The effect of short-term low-dose perchlorate on various aspects of thyroid function</title><secondary-title>Thyroid</secondarytitle></title></periodical><full-title>Thyroid</full-title></periodical><pages>659-663</pages><volume>10</volume><number>8</number><dates><year>2000</year></dates><url> ></urls></record></Cite></EndNote>] gave nine healthy men (ages 22-30 years old) 10 mg of potassium perchlorate (about 0.10 mg/kg/day perchlorate for a 70 kg person) in one liter of water daily for 14 days. Their serum perchlorate concentrations averaged 0.6 µg/mL, and the average urinary perchlorate excretion was 7.7 mg per day. There were no changes in serum T4, T3, or TSH concentrations during the 14-day period of perchlorate ingestion. A highly significant decrease in the thyroid <sup>123</sup>I uptake was observed at 14 days (38 percent below baseline) and a significant increase was observed 14 days after the treatment was stopped. Dietary iodine intakes were relatively high, as indicated by baseline urinary iodine levels of 254 µg/24 hr; 14 days of perchlorate exposure increased urinary iodide to 385 μg/24 hr, whereas after 14 days of recovery, urine iodine excretion was 208 μg/24 hr. The high dietary iodine intake in this study may have reduced the impact of perchlorate on the thyroid. However, the decreased iodine uptake during perchlorate administration, followed by an increase, indicates that total thyroidal colloid stores were being depleted, and replenished after treatment ceased.

In a follow-up study, the same investigators administered 3 mg of perchlorate as the potassium salt daily (perchlorate at about 0.04 mg/kg/day for a 70-kg person) to eight healthy men (ages not given) for 14 days [ ADDIN EN.CITE

<EndNote><Cite><Author>Lawrence</Author><Year>2001</Year><RecNum>292</RecNum><Di splayText>(Lawrence, Lamm, & Di splayText><record><recnumber>292</recnumber><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470160607">292</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Lawrence, J.E.</author><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor><author>Cauthor>Cauthor><author>Ca

L</author></authors></contributors></title>Low dose perchlorate (3 mg daily) and thyroid

function</title><secondary-title>Thyroid</secondary-title></title><periodical><full-title>Thyroid</full-title></periodical><pages>295-

295</pages><volume>11</volume><number>3</number><dates><year>2001</year></dates><isbn>1050-7256</isbn><urls></record></EndNote>]. The mean 24-hr thyroid uptake of radioactive iodine was 16.1% at baseline and 14.5% during perchlorate ingestion, and the mean 8-hr uptake values were 13.8% and 11.8%, respectively; neither change was statistically significant. There were no changes in serum T4, T3, or TSH concentrations. However, in evaluating the results of the shorter term (e.g., 2-week) studies one should consider that the adult euthyroid human thyroid contains several months of T4 stored in the colloid, it is not expected that a 2-week study would result in a change in thyroid status (Dunn & Dunn, 2000, and Brabant et al., 1992, both as cited in Greer et al., 2002).

Further supporting evidence that perchlorate inhibits iodine uptake comes from Greer et al. (2002). As previously described the NRC selected this as the critical study in their derivation of the RfD. In Greer et al. (2002) 37 subjects were recruited in an attempt to analyze the effects associated with perchlorate exposure from drinking water, including altered thyroid function that was measured by the inhibition of RAIU. The test subjects were assigned to one of two groups, which followed similar study protocols. The first group, Group A, consisted of 12 male and 12 female subjects. These 24 subjects were assigned to one of three dose groups of 0.02, 0.1 or 0.5 mg/kg-day of perchlorate. 4 individuals of each sex were in each dose group.

The other study group, Group B, contained the remaining 13 subjects of which 9 were female and 4 were male. In the uptake study, 6 females and 1 male were dosed with 0.007 mg/kg-day of perchlorate and the other 6 subjects were dosed with 0.02, 0.1 or 0.5 mg/kg-day of perchlorate (one male and one female were in each dose group). An overview of the number of subjects in Groups A and B can be seen in [ REF \_Ref293306990 \h \\* MERGEFORMAT ].

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Summary of Number of Subjects in Each of the Dose Groups from Greer et al. (2002) by Study Protocol

		Group A		Group B N					
Dose Group		N							
Dose Group (mg/kg-day)	Male	Female	Total	Male	Female	Total			
0.007	0	0	0	1	6	7			
0.02	4	4	8	1	1	2			
0.1	4	4	8	1	1	2			
0.5	4	4	8	1	1	2			
Tatal	12	12	24	4	9	13			
Total	37								

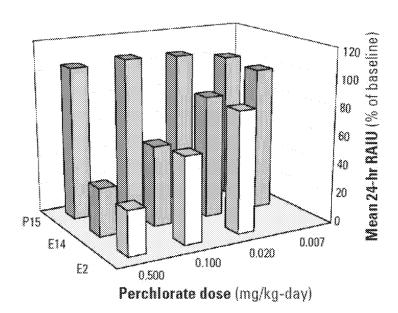
Each subject drank 400 mL of water containing their specified dose of perchlorate in 100 mL increments. This was done for 14 consecutive days. For the Group A subjects, <sup>123</sup>I was ingested four times: (1) the baseline visit (one day before exposure began); (2) the second day of exposure (E2); (3)

the 14th day of exposure (E14) and; (4) the post-exposure day 15 (P15). RAIU measurements were taken eight hours after ingestion on the day of <sup>123</sup>I ingestion, and then again 24 hours after ingestion.

For the Group B subjects <sup>123</sup>I was only ingested three times: (1) the baseline visit; (2) E14 and; (3) P15. After the baseline visit and E14, RAIU measurements were taken eight hours after ingestion on the day of <sup>123</sup>I ingestion, and then again 24 hours later. P15 RAIU measurements were completed only at 09:00 on the morning following <sup>123</sup>I ingestion for the uptake study subjects.

A strong correlation was observed between the eight- and 24-hour RAIU measurements over all of the dose groups and measurement days. The authors found no differences between the E2 and E14 RAIU measurements (the dose-response was a negative linear function of the logarithm of dose for the E2 and E24 measurements). Additionally, no difference was found between sexes. [ REF \_Ref293307032 \h \\* MERGEFORMAT ] presents the percent of RAIU change 24 hours after <sup>123</sup>I ingestion for each exposure day, compared to the baseline RAIU measure. No impact was seen on thyroid hormone levels.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. The Mean 24-hour RAIU (09:00, Day After $^{123}$ I Ingestion) Relative to Baseline for Each Dose Group on E2, E14, and P15



Source: Greer et al. (2002) Figure 3.

Greer et al. (2002) found the "lowest dose producing no statistically significant inhibition of RAIU uptake was 0.007 mg/kg-day." In this lowest dose group on E14, RAIU was 98.2% of the baseline with a standard deviation of 8.3%, indicating the variability seen within Group B. The dose group of 0.007 mg/kg-day, was deemed to be the NOEL for inhibition of RAIU. Given this was the lowest dose administered to study subjects, Greer derived an estimate for a "true no-effect level" and came up with 5.2 and 6.4  $\mu$ g/kg-day for the eight- and 24-hour uptake measurements on E14, respectively. The values Greer et al. (2002) calculated to be the "true no-effect level" doses would be ingested by

an adult (assuming default BW and drinking water ingestions assumptions<sup>8</sup>) if the drinking water supply contained perchlorate at concentrations of approximately 180  $\mu$ g/L and 220  $\mu$ g/L. Regarding the P15 measurements, the RAIU was not significantly different from the baseline, indicating that 15 days after perchlorate exposure ended, there were no lasting effects of iodide uptake inhibition (IUI). No data on subject iodine status was obtained in Greer et al. (2002), which is important given perchlorate's impact is greatest on those with lower iodine intake.

The lower dose studies summarized above are all short term studies that observed decreases in iodine uptake by the thyroid with perchlorate treatment. In these studies no impacts on thyroid hormone levels were noted. Given the several months of T4 stored in the thyroid impacts on hormone levels are not expected. There is one small (n = 13), longer-term clinical trial with six-months of perchlorate ingestion at rates of 0.5 or 3 mg of perchlorate ingestion per day (Braverman et al., 2006). No changes in thyroid function, including in iodine uptake, were observed in this study. However, urinary iodide levels among participants in this study were considerably higher than the national average (mean 257.8  $\mu$ g/total volume before 0.5 mg dose of perchlorate; mean 311.5  $\mu$ g/total volume before 3 mg dose of perchlorate), which could be protective. Subsequently, the results of this study are consistent with evidence that healthy adults may be protected against the thyroid hormone changes associated with perchlorate exposures given adequate dietary iodine intake [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></author>>/contributors><titles><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></Cite></EndNote>].

Overall, the clinical studies on perchlorate on the thyroid have shown that in healthy volunteers perchlorate has impacted iodine uptake by the thyroid (Lawrence et al., 2000; Lawrence et al., 2001; Greer et al., 2005). Additionally, in study participants with high iodine intake levels there may be a protective effect against the iodine uptake inhibition caused by perchlorate ingestion (Braverman et al. 2006). None of the clinical trials, with the exception of the higher dose study of Brabant et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Brabant</Author><Year>1992</Year><RecNum>2104</RecNum><Di splayText>(1992)</DisplayText><record><rec-number>2104</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1541773981">2104</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Brabant, G.P.</author><author>P.

Bergmann</author><author>C.M. Kirsch</author><author>J. Kohrle</author><author>R.D. Hesch</author><author>A. von zur Muhlen</author></author></contributors><titles><title>Early adaptation of thyrotropin and thyroglobulin secretion to experimentally decreased iodide supply in

[PAGE \\* MERGEFORMAT]

<sup>&</sup>lt;sup>8</sup> Default BW is 70 kg, and default ingestion rate is 2 L/day.

man</title><secondary-title>Metabolism</secondary-title></title><periodical><full-title>Metabolism</full-title><abbr-1>Metabolism</abbr-1></periodical><pages>1093-1096</pages><volume>41</volume><number>10</number><dates><year>1992</year></dates><urls></record></Cite></EndNote>] demonstrated an impact of perchlorate on thyroid hormone levels. However, it is unlikely healthy, euthyroid individuals with adequate iodine intake will experience changes in thyroid hormone levels in the short-term studies. In the longer term study (Braverman et al., 2006) in individuals with high iodine impact, no impact on the thyroid was seen.

# 7.2.2 Environmental Epidemiological Studies – General Population

In addition to what is seen in clinical exposure studies, there are several more recent epidemiological studies. Specifically, several studies have been published in the last twelve years which use data from NHANES and have found an association between low level perchlorate exposure (measured as urinary perchlorate) in the U.S. and thyroid hormone levels. This association is most noticeable in individuals with low urinary iodine levels and co-exposures to other goitrogens<sup>9</sup>.

The first such study was Blount et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Blount</Author><Year>2006</Year><RecNum>224</RecNum><Disp layText>(2006)</DisplayText><record><rec-number>224</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468246110">224</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>>Blount, B. C.</author><author>Pirkle, J. L.</author><author>Osterloh, J. D.</author><author>Valentin-Blasini, L.</author><author>Caldwell, K. L.</author></author></contributors><title>Urinary perchlorate and thyroid hormone levels in adolescent and adult men and women living in the United States</title><secondary-title>Environmental Health Perspectives</secondary-title></title>/title>/periodical><pages>1865-

71</pages><volume>114</volume><number>12</number><dates><year>2006</year></dates><url s></urls></record></Cite></EndNote>] who examined the relationship between urinary levels of perchlorate and serum levels of TSH and total T4 in 2,299 men and women (ages 12 years and older), who participated in CDC's 2001-2002 NHANES.<sup>10</sup> Blount et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Blount</Author><Year>2006</Year><RecNum>224</RecNum><Disp layText>(2006)</DisplayText><record><rec-number>224</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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L.</author><author>Osterloh, J. D.</author><author>Valentin-Blasini,

L.</author><author>Caldwell, K. L.</author></authors></contributors><title>Urinary

<sup>&</sup>lt;sup>9</sup> Substances that disrupt the production of thyroid hormones by interfering with iodine uptake in the thyroid gland.

<sup>&</sup>lt;sup>10</sup> While CDC researchers measured urinary perchlorate concentration for 2,820 NHANES participants, TSH and total T4 serum levels were only available for 2,299 of these participants.

perchlorate and thyroid hormone levels in adolescent and adult men and women living in the United States</title><secondary-title>Environmental Health Perspectives</secondary-title></title>Secondary-title>Environmental Health Perspectives</full-title>

 $71 < pages < volume > 114 < volume > number > 12 < number > dates > cyear > 2006 < year > c/dates > curl s > c/record > c/Cite > c/EndNote > 1 evaluated perchlorate along with covariates known or likely to be associated with T4 or TSH levels to assess the relationship between perchlorate and these hormones, and the influence of other factors on this relationship. The study found that perchlorate was a significant predictor of thyroid hormones in women, but not men. The 1,111 women were further analyzed after being divided into two categories, higher-iodine and lower-iodine, using a cut point of <math>100 \mu g/L$  of urinary iodine, based on the World Health Organization (WHO) definition of sufficient iodine intake. Hypothyroid women were excluded from the analysis. For women with urinary iodine levels less than  $100 \mu g/L$ , the author found urinary perchlorate to be inversely associated with serum T4 levels and directly associated with TSH levels. For women with urinary iodine levels greater than or equal to  $100 \mu g/L$ , the researchers found that perchlorate to be associated with increases in TSH but not a predictor of T4.

Steinmaus et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ], using the same data set, supported the findings of Blount et al. (2006) by demonstrating an interaction between thiocyanate/cotinine (byproducts of eigarette smoke), smoking, perchlorate, and impaired thyroid function in women with urinary iodine less than  $100 \, \mu g/L$ . From the base population of women from

 $<sup>^{11}</sup>$  WHO notes that the prevalence of goiter begins to increase in populations with a median iodide intake level below  $100~\mu g/L$  [ ADDIN EN.CITE

<sup>&</sup>lt;EndNote><Cite><Author>WHO</Author><Year>1999</Year><RecNum>2119</RecNum><DisplayTe
xt>(Organization), 1999)/DisplayText><record><rec-number>2119</rec-number><foreign-keys><key
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deficiency disorders (IDD)</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></ti>

<sup>&</sup>lt;sup>12</sup> Caldwell et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Caldwell</Author><Year>2005</Year><RecNum>2118</RecNum><Displa yText>(2005)</DisplayText><record><rec-number>2118</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1541775268">2118</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Caldwell, K.L.</author><author>R. Jones</author>J.G.

Howell</author></authors></contributors><titles><title>Urinary iodine concentration: United States National Health and Nutrition Examination Survey 2001-2002</title><secondary-title>Thyroid</secondary-title></title>Thyroid</full-title></periodical><pages>692-

<sup>699&</sup>lt;/pages><volume>15</volume><number>7</number><dates><year>2005</year></dates></urls></record></Cite></EndNote>] had found that about 36 percent of women living in the United States had iodine levels less than 100 µg/L in spot urine samples.

the NHANES (2001-2002). Steinmaus et al identified 385 women who had urinary iodine levels less than 100  $\mu$ g/L and also had NHANES data on urinary perchlorate, creatinine, and thyroid hormone levels. Steinmaus et al. collected additional information for covariates that may impact the perchlorate-thyroid interaction such as race, age and hours fasted. Similar to Blount et al. (2006), Steinmaus et al. (2007) found an association between the log of urinary perchlorate in women with urinary iodide less than 100  $\mu$ g/L and T4. The authors furthermore found when examining smokers (both as self-reported in NHANES or classified using cotinine levels), the association with T4 more than doubled in magnitude when compared to the entire study population. This same association was not seen with TSH. Additionally, when only looking at individuals with high thiocyanate exposure (> 1,800  $\mu$ g/L), there was more than a doubling of the impact of perchlorate on urinary T4; again, this same affect was not seen for TSH.

Steinmaus et al. (2007) also observed a clear stepwise increase in the relationship between perchlorate and T4 with increasing levels of serum cotinine. While clear associations between perchlorate, smoking, and decreased T4 were present in women with lower iodine intake levels, this relationship was not seen with TSH or with T4 in women with urinary iodide levels greater than or equal to  $100 \, \mu \text{g/L}$  or in men. Additionally, the strongest perchlorate thyroid hormone associations were found with thiocyanate levels in the upper tertile (> 1,800  $\,\mu \text{g/L}$ ) and clear associations were not seen at levels less than 751  $\,\mu \text{g/L}$ . These findings suggest that perchlorate may interact with tobacco smoke or thiocyanate to reduce iodide uptake by the thyroid and decrease thyroid function.

Building on the Blount et al. (2006) and Steinmaus et al. (2007) findings, Mendez and Eftim [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Mendez</Author><Year>2012</Year><RecNum>2244</RecNum><Di splayText>(2012)</DisplayText><record><rec-number>2244</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1543350257">2244</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Mendez, W.</author><author>Eftim,

S.E.</author></authors></contributors><title>Biomarkers of perchlorate exposure are correlated with circulating thyroid hormone levels in the 2007-2008 NHANES</title><secondary-title>Environmental Research</secondary-title></title><periodical><full-title>Environmental Research</full-title></periodical><volume>118

44</section><dates><year>2012</year></dates><urls></record></Cite></EndNote>] evaluated NHANES data from the 2007-2008 survey cycle. Specifically, Mendez and Eftim (2012) used generalized additive mixed models to estimate the relationship between thyroid hormones concentrations in 1,887 subjects (970 male, 907 female) and creatinine corrected urinary perchlorate. In this analysis Mendez and Eftim (2012) controlled for phthalate esters and BPA based on previous research demonstrating the association of these biomarkers with thyroid hormones. Mendez and Eftim (2012) found perchlorate to be significantly associated with free and total T4 and free T3 in the entire cohort. When looking at males and females separately the authors found an inverse association between creatinine corrected urinary perchlorate and fT3, fT4 and total T4, though for females the associations were not statistically significant at the p < 0.05 level.

<sup>&</sup>lt;sup>13</sup> Cotinine is an alkaloid found in tobacco and is also a metabolite of nicotine. Cotinine may be a more accurate indicator of current smoking status than self-reported smoking.

Similarly, Steinmaus et al (2013) also utilized the 2007-2008 NHANES data to test the hypothesis that individuals with the highest perchlorate and thiocyanate exposure, along with low iodine intake, would have lower thyroid hormone levels than those with lower exposure and adequate iodine intake. To study this research question Steinmaus et al. (2013) created three groups from NHANES data: (1) a "low exposure" reference group that had urinary perchlorate and urinary thiocyanate concentration in the lowest tertile of concentrations and urinary iodine intake greater than 1,000  $\mu$ g/L; (2) a "moderate exposure" group which included subjects who had a combination of urinary perchlorate and thiocyante in the middle tertile and urinary iodine greater than or equal to  $100~\mu$ g/L; and (3) a "high exposure" group which has perchlorate and thiocyanate in the highest tertile and urinary iodine less than  $100~\mu$ g/L. They then compared thyroid hormone concentrations in each group and determined the difference between them. They also evaluated the impact of each variable on thyroid hormone levels on their own. When comparing the highest tertile of perchlorate exposure participants to those with the lowest exposure there was a 5.0% difference in total T4. However, when comparing the "high exposure" group to the low exposure reference group the difference in total T4 was 12.9% indicating a substantial impact of the combined goitrogenic effect from the three variables assessed.

Suh, Abraham, Hixon & Proctor [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Suh</Author><Year>2014</Year><RecNum>2241</RecNum><Displa yText>(2014)</DisplayText><record><rec-number>2241</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543349148">2241</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Suh, M.</author><author>Abraham, L.</author><author>Hixon, J.G.</author><author>Proctor, D.M.</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author><

D.M.</author></authors></contributors><titles><title>The effects of perchlorate, nitrate, and thiocyanate on free thyroxine for potentially sensitive subpopulations of the 2001–2002 and 2007–2008 National Health and Nutrition Examination Surveys</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title><periodical><full-title>Journal of Exposure Science and Environmental Epidemiology</full-title></periodical><pages>579-

587</pages><volume>24</volume><dates><year>2014</year></dates><urls></urls></record></Cit e></EndNote>] also studied the associations between urinary perchlorate, nitrate and thiocyanate and serum fT4 in people with reduced urinary iodine levels using both 2001-2002 and 2007-2008 NHANES data (additional analysis was also done on pregnant women, which is summarized in the next section). Multivariate regression models, which included perchlorate, nitrate, thiocyanate and other covariates known to have an impact on the thyroid (e.g., TPO antibodies, race/ethnicity, BMI) were developed to test this association in both survey cycles. Additionally, a meta-analysis of the 2000-2001 data and the 2007-2008 data was also performed on the women only data. In non-pregnant women with creatinine-adjusted urinary iodine concentrations at or above 100 μg/L, perchlorate was found to be a significant predictor of serum fT4 in the 2001-2002 data cycle for one of the two presented models that included urinary perchlorate for this group. There was no statistically significant association between urinary perchlorate and serum fT4 for non-pregnant women with creatinine-adjusted urinary iodine concentrations below 100 μg/L.

McMullen, Ghassabian, Kohn & Trasande [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>McMullen</Author><Year>2017</Year><RecNum>2246</RecNum>< DisplayText>(2017)</DisplayText><record><rec-number>2246</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1543354250">2246</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>McMullen, J.</author><author>Ghassabian, A.</author><author>Kohn, B.</author><author>Trasande,

L.</author></authors></contributors><title>Identifying subpopulations vulnerable to the thyroid-blocking effects of perchlorate and thiocyanate</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>2637-

2645</pages><volume>102</volume><number>7</number><dates><year>2017</year></dates><urls></urls></record></EndNote>] evaluated exposure to perchlorate, thiocyanate, and nitrate (measured in urine) and associated impacts on serum fT4 and TSH in 3,151 individuals using NHANES data from 2009 to 2012. Multivariate linear regression models were developed to test these associations as stratified by age and sex. The authors evaluated potential confounders likely associated with the thyroid hormone/goitrogren relationship controling for serum cotinine, body mass index, total daily energy consumption, race/ethnicity, and income-to-poverty ratio. Urinary iodide levels were not found to be correlated with any goitrogen and was therefore not included included as a covariate in the model. For each log unit increase in perchlorate, fT4 decreased with statistical significance by 0.026 ng/dL in the general population (4% decrease relative to median fT4, p = 0.005), 0.029 ng/dL in females (4% decrease relative to median fT4, p = 0.035), and by 0.055 ng/dL in adolescent females (aged 12-21, 8% decrease relative to median fT4, p = 0.029). There were no statistically significant associations between urinary perchlorate and serum fT4 for males alone or the other, older female age groups.

Overall studies of the general population have found an association between low levels of perchlorate exposure (measured in urine) and altered thyroid hormone status. The Mendez and Eftim (2012) and McMullen et al. (2017) analyses demonstrated this association across the entire cohort of adults. Studies by Blount et al. (2006) and Steinmaus (2007; 2013) point to the impact low iodine and exposures to other goitrogens may have on the perchlorate/thyroid hormone relationship.

# 7.2.3 Environmental Epidemiology Studies – Pregnant Women

In contrast to the general population studies which find fairly consistent associations between urinary perchlorate and changes in thyroid hormone, particularly in sensitive populations, studies specifically in pregnant women are more mixed. For example, Pearce et al. (2010, 2011 and 2012) did not find an association between perchlorate biomarkers and altered thyroid function in pregnant women. Pearce et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] conducted a cross-sectional study from 2002 to 2006 on approximately 1,000 women in Wales and Italy, who were less than 16 weeks pregnant at the time of enrollment. Urinary perchlorate levels were not found to be associated with serum TSH or fT4 in women with normal thyroid function or in the hypothyroid/hyperthyroxinemic women (Pearce et al., 2010). Additionally, when Pearce et al. (2010) excluded women with iodine levels less than 100  $\mu$ g/L (n = 604), no association with TSH or fT4 was found. As noted by other researchers [ ADDIN EN.CITE <EndNote><Cite><Author>California Environmental Protection Agency

(CalEPA)</author><Year>2015</year><RecNum>62</RecNum><DisplayText>(California Environmental Protection Agency (CalEPA), 2015)</br>
<br/>
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Hazard Assessment</author></secondary-authors></contributors><title>Public health goal for perchlorate in drinking

water</title></title></title></title></title></fr> the median urine thiocyanate levels in Wales and Italy were much lower than those commonly found in the U.S., yet the reason for this is unknown given similar smoking prevalence in women in Wales, Italy, and the U.S. ADDIN EN.CITE <EndNote <Cite < Author > World Health Organization</Author><Year>2008</Year><RecNum>488</RecNum><DisplayText>(United Kindom's Office of National Statistics, 2008; World Health Organization, 2008)</DisplayText><record><rec-number>488</rec-number><foreign-keys><key app="EN" dbid="zr9taef095wwxfe2zaqxw224vdvttr052wsx">488</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>World Health Organization, </author></authors></contributors></title>Global Tobacco Epidemic 2008, Appendix III Table

3a</title></title></dates><year>2008</year></dates><urls></urls></record></Cite><Cite><Author >United Kindom's Office of National

Statistics</Author><Year>2008</Year><RecNum>489</RecNum><record><rec-number>489</rec number><foreign-keys><key app="EN" db-

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Statistics.</author></authors></contributors><title>General Lifestyle Survey: Smoking and Drinking among Adults, Table

1.10</title></title></dates><year>2008</year></dates><urls></urls></record></Cite></EndNote>]. Regardless of why thiocyanate levels differ this may explain why associations were not found. In the Steinmaus et al. (2007) study conducted in the U.S., the strongest perchlorate thyroid hormone associations were found with thiocyanate levels in the upper tertile (> 1800 µg/L) and clear associations were not seen at levels less than 751 µg/L. The mean thiocyanate level was 407.5 µg/L in Wales and 372.5 µg/L in Italy. It has been argued that "the Pearce et al. (2010) findings [were] consistent with those of Steinmaus et al. (2007)" (CalEPA, 2015, p. 108).

Similarly, Pearce et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pearce</Author><Year>2011</Year><RecNum>450</RecNum><Displ ayText>(2011)</DisplayText><record><rec-number>450</rec-number><foreign-keys><key app="EN" db-id="zr9taef095wwxfe2zaqxw224vdvttr052wsx">450</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Pearce, E.

N.</author><author>Spencer, C. A.</author><author>Mestman, J. H.</author><author>Lee, R.

H.</author><author>Bergoglio, L. M.</author><author>Mereshian, P.</author><author>He,

X.</author><author>Leung, A. M.</author><author>Braverman, L.

E.</author></authors></contributors><auth-address>Section of Endocrinology, Diabetes and Nutrition, Boston University School of Medicine, Boston, MA. <a href="mailto:duth-address">duth-address<<title><title>The</a> Effect of Environmental Perchlorate on Thyroid Function in Pregnant Women from Cordoba, Argentina, and Los Angeles, California </title> < secondary-title > Endocr Pract </secondarytitle></title></periodical><full-title>Endocr Pract</full-title></periodical><pages>1-17</pages><dates><year>2011</year><pub-dates><date>Feb 16</date></pubdates></dates><isbn>1934-2403 (Electronic)&#xD;1530-891X (Linking)</isbn><accession-

num>21324827</accession-num><urls><related-

urls><url>http://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=Retrieve&amp;db=PubMcd&amp;do

pt=Citation&list\_uids=21324827 </url></related-urls></urls><research-notes>Abt to Include</research-notes><language>Eng</language></record></EndNote>] found no association between urinary perchlorate concentrations and serum TSH, fT4 index, or total T3 values in pregnant women in Los Angeles, California, even though perchlorate was detected in the urine of all of the 134 women from California (median urinary perchlorate = 7.8 μg/L). The correlation between fT4 and urinary perchlorate was borderline significant, although it was in the opposite direction of what would have been expected from a biological standpoint (r = 0.12, p = 0.06). Their study population included women with low urinary iodine values (< 100 μg/L). The sample size in this study was much smaller (n = 241) which likely resulted in fewer participants with low iodide intake and less statistical power. Additionally, other goitrogenic anions in the blood and smoking status were not controlled in the study by Pearce et al. (2011). Pearce et al. (2011), however, did control for thyroperoxidase<sup>14</sup> antibodies in their analysis, which is an indicator of existing autoimmune thyroid disease, one of the factors that may affect the perchlorate-thyroid hormone relationship.

In a similar study based in Athens, Greece, Pearce et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pearce</Author><Year>2012</Year><RecNum>1396</RecNum><Dis playText>(2012)</DisplayText><record><rec-number>1396</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206441">1396</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Pearce, E. N.</author><author>Alexiou, M.</author><author>Koukkou, E.</author><author>Braverman, L. E.</author><author>He, X.</author><author>Ilias, I.</author><author>Alevizaki, M.</author><author>Markou, K. B.</author></authors></contributors><title>Perchlorate and thiocyanate exposure and thyroid function in first trimester pregnant women from Greece</title><secondary-title>Clinical Endocrinology</secondary-title><alt-title>Clin Endocrinol</alt-title><short-title>Clinical Endocrinology</short-title></title></full-title>Clinical Endocrinology</fulltitle></periodical><pages>471-474</pages><volume>77</volume><number>3</number><dates><year>2012</year></dates><isbn >ISSN 0300-0664
EISSN 1365-2265</isbn><label>1065842</label><urls><related-</p> urls><url>http://dx.doi.org/10.1111/j.1365-2265.2012.04407.x</url></relatedurls></urls><electronic-resource-num>10.1111/j.1365-2265.2012.04407.x</electronic-resourcenum><language>English</language></record></Cite></EndNote>] evaluated the association between urinary perchlorate, iodine, and thiocyanate, and serum concentrations of TSH, fT4, fT3, and TPO antibodies in 134 pregnant women. The mean gestational age of the cohort was 10.9 weeks (SD  $\pm$  2.3 weeks) and mean concentrations were 135  $\mu$ g/L (SD  $\pm$  77) for urinary iodine and 6.5  $\mu$ g/L (SD ± 11.2) for perchlorate. In a Spearman's rank univariate correlation analysis, an inverse correlation was found between urinary perchlorate and serum fT4 in all women (R = -0.19; p = 0.03) and in the subgroup of 50 mildly iodine-deficient women with iodine levels below 100  $\mu$ g/L (R = -0.23; p = 0.09). In multivariable regression modeling, with adjustments for urinary thiocyanate, urinary iodide, TPO antibody titers, gestational age, and maternal age, a relationship between the logarithm of urinary perchlorate (µg/L) and the logarithm of serum fT4 (pmol/L) was seen (regression coefficient

<sup>&</sup>lt;sup>14</sup> Thyroperoxidase is an enzyme expressed mainly in the thyroid that frees iodine for addition onto Tg for the production of T4 or T3.

 $(\beta)$  = -0.007; p = 0.8) but it did not reach statistically significance. Gestational age was found to be inversely associated with fT4 in multivariable regression analyses ( $\beta$  = -0.003; p = 0.002 for the association between gestational age in days and the logarithm of serum fT4), and positively correlated with urinary perchlorate concentration (R = 0.18; p=0.04). Based on these results, the authors hypothesized that gestational age appeared to be a confounder in the apparent correlation between urine perchlorate and serum fT4 (and fT3; fT3 results are presented in the article). Associations between urinary perchlorate and TSH were not seen in correlation or regression analyses.

In Suh et al. (2014; methods summarized in the preceding section), pregnant women in the 2001-2002 survey cycle, did not have a significant association between urinary iodine less than 150 µg/L, nitrate, perchlorate or thiocyanate and fT4. However there were less than 50 pregnant women in the NHANES cohort and this group had the lowest urinary perchlorate out of any evaluated by the study authors. Further, NHANES does not define the gestational week of pregnancy and thyroid hormones vary significantly over the course of pregnancy which may explain some of this null finding.

Mortensen et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Mortensen</Author><Year>2016</Year><RecNum>2242</RecNum>< DisplayText>(2016)</DisplayText><rec-number>2242</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1543349397">2242</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Mortensen, M.E.</author><author>Birch, B.</author><author>Wong, L.</author><author>Valentin-Blasini, L.</author><author>Boyle, E.B.</author><author>Caldwell, K.L.</author><author>Merrill, L.S.</author><author>Moye, J.</author><author>Blount, B.C.</author></author></contributors><title>Thyroid antagonists and thyroid indicators in U.S. pregnant women in the Vanguard Study of the National Children's Study</title><secondary-title>Environmental Research</secondarytitle></title></periodical><full-title>Environmental Research</full-title></periodical><pages>179-188</pages><volume>149</volume><dates><year>2016</year></dates><urls></urls></record></C ite></EndNote>] also did not find an association between urinary perchlorate and thyroid hormone levels. In a study of 330 pregnant women from the Vanguard Study of the National Children's Study (women in this cohort are from seven locations: Queens County, New York; Duplin County, North Carolina; Salt Lake County, Utah; Orange County, California; Montgomery County, Pennsylvania; Waukesha County, Wisconsin) the authors evaluated urinary perchlorate, thiocyanate, nitrate iodine and serum fT4 and TSH and cotinine. Using multiple regression models the authors assessed of the potential inhibitors on thyroid hormone levels. They also assessed the impacts of perchlorate equivalent concentrations (which estimates the combined inhibitory effect of the anions on the NIS). They authors did not find an association between perchlorate or perchlorate equivalent concentrations and TSH or fT4, even in women with low iodine.

Contrary to the null findings from the Pearce, Suh and Mortensen studies, Charatcharoenwitthaya et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Charatcharoenwitthaya</Author><Year>2014</Year><RecNum>2243</RecNum><DisplayText>(2014)</DisplayText><record><rec-number>2243</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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N.</author><author>Ongphiphadhanakul, B.</author><author>Pearce,

E.N.</author><author>Chanthasenanont, A.</author><author>He, X.</author><author>Chailurkit, L.</author><author>Braverman,

L.E.</author></authors></contributors><titles><title>The association between perchlorate and thiocyanate exposure and thyroid function in first-trimester pregnant Thai women</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title>><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>2365-2371</pages><volume>99</volume><number>7</number><dates><year>2014</year></dates><url>s></urls></record></Cite></EndNote>] found a direct association between urinary perchlorate and TSH and an inverse association between perchlorate and fT4. This study was conducted in 200 pregnant Thai women at 14 gestational weeks or less. In their multivariate analysis, adjusting for log thiocyanate to creatinine ratio, log iodide to creatinine ratio, and gestational age, log perchlorate to creatinine ratio was positively associated with log TSH and inversely associated with log free T4. Log thiocyanate to creatinine ratio was also a significant positive predictor of log TSH in women with a urine iodide level of less than 100  $\mu g/L$ .

# Horton et al., [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Horton</Author><Year>2015</Year><RecNum>2245</RecNum><DisplayText>(2015)</DisplayText><record><rec-number>2245</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1543350588">2245</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Horton, M.</author><author>Blount,

B.</author><author>Valentin-Blasini, L.</author><author>Wapner, R.</author><author>Whyatt, R.</author><author>Gennings, C.</author><author>Factor-Litvak,

P.</author></authors></contributors><titles><title>CO-occurring exposure to perchlorate, nitrate and thiocyanate alters thyroid function in healthy pregnant women</title><secondary-title>Environmental Research</secondary-title></title><periodical><full-title>Environmental Research</full-title></periodical><pages>1-

9</pages><volume>143</volume><dates><year>2015</year></dates><urls></record></Cite></EndNote>] also found an association between perchlorate, when exposure co-occurs with nitrate and thiocyanate, and thyroid hormone status (as indicated by TSH) in a cohort from New York city. In this study the cross-sectional relationship between urinary perchlorate, thiocyanate and nitrate concentrations and thyroid function in 284 pregnant women at 12 (±2.8) weeks was evaluated. A weighted quantile sum (WQS) regression was used in the analysis. Serum fT4 and TSH were measured along with urine samples measuring perchlorate, nitrate and iodide. Results found urinary perchlorate, thiocyanate and nitrate concentrations were significantly correlated (Spearman's r 04.-0.5, p<0.001). Linear regression analysis did not suggest an association between the individual goitrogens and thyroid hormone functioning. However, the WQS exposed a significant positive relationship between the weighted sum of urinary concentrations of perchlorate, thiocyanate and nitrate and increased TSH, a significant relationship was not seen when looking at fT4. The WQS suggests that perchlorate was the main contributor to this relationship (weighted at 75%). These findings suggests that exposure to perchlorate, nitrate and thiocyanate may alter maternal thyroid function, specifically TSH, during pregnancy.

Additionally, Steinmaus et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Steinmaus</Author><Year>2016</Year><RecNum>258</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>258</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1468518705">258</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Steinmaus, C</author><author>Pearl, M</author><author>Kharrazi, M</author><author>Blount, BC</author><author>Miller, MD</author><author>Pearce, EN</author><author>Valentin-Blasini, L</author><author>DeLorenze G</author><author>Hoofnagle, AN</author><author>Liaw J</author></authors></contributors></title>Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California</title><secondarytitle>Environmental Health Perspectives</secondary-title></titles><periodical><fulltitle>Environmental Health Perspectives</full-title></periodical><pages>861-867</pages><volume>124</volume><dates><year>2016</year></dates><urls></urls><electronicresource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resourcenum></record></Cite></EndNote>] found an association between perchlorate exposure (as measured by urinary perchlorate) and thyroid hormone levels (T4, fT4 and TSH) in pregnant women in San Diego. Data were collected as part of Project Baby's Breath during 2000-2003. During this time period the Colorado River, a large source of drinking water in the United States, was contaminated with perchlorate from a perchlorate manufacturing plant. Perchlorate concentrations during this time period in the river were 4 to 8 μg/L. Urine samples were collected from pregnant women at 7 gestational weeks (GW), and blood samples were collected from the same women around 15 to 20 GW. The median urinary perchlorate concentration in this population was 6.50 µg/L. According to Steinmaus et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Steinmaus</Author><Year>2016</Year><RecNum>258</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>258</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468518705">258</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Steinmaus, C</author><author>Pearl, M</author><author>Kharrazi, M</author><author>Blount, BC</author><author>Miller, MD</author><author>Pearce, EN</author><author>Valentin-Blasini, L</author><author>DeLorenze G</author><author>Hoofnagle, AN</author><author>Liaw J</author></authors></contributors></title>Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California </title><secondarytitle>Environmental Health Perspectives</secondary-title></titles><periodical><fulltitle>Environmental Health Perspectives</full-title></periodical><pages>861-867</pages><volume>124</volume><dates><year>2016</year></dates><urls></urls><electronicresource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resourcenum></record></Cite></EndNote>], this is more than double the reported concentration of perchlorate in NHANES from 2001 to 2002.

Using multivariate regression analysis, Steinmaus et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Steinmaus</Author><Year>2016</Year><RecNum>258</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>258</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468518705">258</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Steinmaus, C</author><author>Pearl, M</author><author>Kharrazi, M</author><author>Blount, BC</author><author>Miller, MD</author><author>Pearce, EN</author><author>Valentin-Blasini,

L</author><author>DeLorenze G</author><author>Hoofnagle, AN</author><author>Liaw J</author></authors></contributors></title>Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California</title><secondarytitle>Environmental Health Perspectives</secondary-title></titles><periodical><fulltitle>Environmental Health Perspectives</full-title></periodical><pages>861-867</pages><volume>124</volume><dates><vear>2016</vear></dates><urls></urls><electronicresource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resourcenum></record></Cite></EndNote>] evaluated the relationship between urinary perchlorate and thyroid hormone levels, specifically fT4, T4, and TSH. In their analysis perchlorate, thiocyanate, creatinine, and TSH were log<sub>10</sub>-transformed to create normal distributions. Variables included in the final models due to a priori decision making included urinary creatinine concentrations, maternal age and education, ethnicity, and gestational age at serum collection. Additionally, the final models were adjusted for urinary thiocyanate, as it changed regression coefficients by more than 10%. Perchloratethyroid hormone analyses were also estimated after stratifying by urinary iodide categories of < 100, 100-300, and  $> 300 \mu g/day$ .

When evaluating all iodine groups, Steinmaus et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Steinmaus</Author><Year>2016</Year><RecNum>258</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>258</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468518705">258</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Steinmaus, C</author><author>Pearl, M</author><author>Kharrazi, M</author><author>Blount, BC</author><author>Miller, MD</author><author>Pearce, EN</author><author>Valentin-Blasini, L</author><author>DeLorenze G</author><author>Hoofnagle, AN</author><author>Liaw J</author></authors></contributors></title>Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California</title><secondarytitle>Environmental Health Perspectives</secondary-title></titles><periodical><fulltitle>Environmental Health Perspectives</full-title></periodical><pages>861-867</pages><volume>124</volume><dates><year>2016</year></dates><urls></urls><electronicresource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resourcenum></record></Cite></EndNote>| found a statistically significant, inverse relationship between perchlorate and thyroid hormone (both fT4 and total T4). Further, there was a statistically significant, directly proportional relationship between perchlorate and log<sub>10</sub>TSH. Evaluating these results as a whole, it can be seen that perchlorate exposures in the San Diego cohort are associated with depressed fT4, which may in turn result in an increase in TSH. However, this increase in TSH does not compensate for the decreased fT4 as a result of perchlorate exposure. This finding is supported by Fitzgerald and Bean [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Fitzgerald</Author><Year>2016</Year><RecNum>1876</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>1876</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495213070">1876</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Fitzgerald, S. P.</author><author>Bean, N. G.</author></authors></contributors><auth-address>Department of Internal Medicine and Department of Endocrinology, The Royal Adelaide Hospital, Adelaide, SA 5000, Australia; School of Medicine, The University of Adelaide, Adelaide, SA 5005, Australia. & xD; School of Mathematical Sciences, The University of Adelaide, Adelaide, SA 5005, Australia; ARC Centre of Excellence for Mathematical and Statistical Frontiers, The University of Adelaide, Adelaide, SA 5005, Australia.</auth-address><titles><title>The relationship between population T4/TSH set point data and T4/TSH physiology</title><secondary-title>Journal of Thyroid Research</secondarytitle></title></title><periodical><full-title>Journal of Thyroid Research</fulltitle></periodical><pages>6351473</pages><volume>2016</volume><edition>2016/04/29</edition ><dates><year>2016</year></dates><isbn>2090-8067 (Print)</isbn><accessionnum>27123359</accession-num><urls></urls><custom2>4830732</custom2><electronic-resourcenum>10.1155/2016/6351473</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>], who evaluated thyroid hormone status in healthy and diseased populations and noted that although changes in T4 (as a result of levothyroxine dosing) have a predictable impact on TSH levels, increasing TSH levels do not have a consistently predictable impact on T4 levels [ ADDIN EN.CITE <EndNote><Cite><Author>Fitzgerald</Author><Year>2016</Year><RecNum>1876</RecNum>< DisplayText>(Fitzgerald & DisplayText><record><rec-number>1876</rec-number>1876</rec-number> number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495213070">1876</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Fitzgerald, S. P.</author><author>Bean, N. G.</author></authors></contributors><auth-address>Department of Internal Medicine and Department of Endocrinology, The Royal Adelaide Hospital, Adelaide, SA 5000, Australia; School of Medicine, The University of Adelaide, Adelaide, SA 5005, Australia. & xD; School of Mathematical Sciences, The University of Adelaide, Adelaide, SA 5005, Australia; ARC Centre of Excellence for Mathematical and Statistical Frontiers, The University of Adelaide, Adelaide, SA 5005, Australia.</auth-address><titles><title>The relationship between population T4/TSH set point data and T4/TSH physiology</title><secondary-title>Journal of Thyroid Research</secondarytitle></title></title><periodical><full-title>Journal of Thyroid Research</fulltitle></periodical><pages>6351473</pages><volume>2016</volume><edition>2016/04/29</edition ><dates><year>2016</year></dates><isbn>2090-8067 (Print)</isbn><accessionnum>27123359</accession-num><urls></urls><custom2>4830732</custom2><electronic-resourcenum>10.1155/2016/6351473</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>].

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resource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resource-num></record></Cite></EndNote>] found the effect of perchlorate on fT4 levels to be similar among women with low (<100  $\mu g/day$ ) and normal (100-300  $\mu g/day$ ) iodine intakes. They found a greater effect of perchlorate on fT4 in the high-iodine intake group (>300  $\mu g/day$ ). Their hypothesis for this finding is as follows:

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In most individuals, very high iodine intakes transiently and paradoxically inhibit
thyroid hormone production, termed the acute Wolff-Chaikoff effect [ ADDIN
EN.CITE
<EndNote><Cite><Author>Wolff</Author><Year>1948</Year><RecNum>1
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type><contributors><author><br/>Chaikoff, J.</author><author>Chaikoff,
I. L.</author></authors></contributors><titles><title>Plasma inorganic
iodide as a homeostatic regulator of thyroid function</title><secondary-
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351X</isbn><label>2140877</label><urls></urls><language>English</lang
uage></record></Cite></EndNote>]. Normally, this is only temporary, and
after a short disruption there is an "escape" and thyroid function returns to
normal after a few days, even if high iodine exposure continues [ ADDIN
EN.CITE
<EndNote><Cite><Author>Eng</Author><Year>1999</Year><RecNum>190
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S.</author><author>Carrasco,
N.</author></authors></contributors><titles><title>Escape from the acute
Wolff-Chaikoff effect is associated with a decrease in thyroid sodium/iodide
symporter messenger ribonucleic acid and protein</title><secondary-
title>Endocrine Journal</secondary-title></titles><periodical><full-
title>Endocrine Journal</full-title></periodical><pages>3404-
3410</pages><volume>140</volume><dates><year>1999</year></dates><
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urls></urls></record></Cite></EndNote>]. However, several studies have
reported increased rates of thyroid autoimmunity and hypothyroidism in areas
where people have chronically high iodine intakes, from diet (e.g., seaweed
consumption) or drinking water with naturally high iodine concentrations. This
would suggest that chronic excessive high iodine intakes can lead to long-term
hypothyroidism in susceptible individuals [ ADDIN EN.CITE
EN.CITE.DATA ]. Associations between very high urinary iodide
concentrations and decreased thyroid hormone production have also been seen in
NHANES [ ADDIN EN.CITE
<EndNote><Cite><Author>Cushing</Author><Year>2011</Year><RecNum
>1908</RecNum><DisplayText>(Cushing, Steinmaus, Miller, & Smith,
2011; Vanderver, Engel, & DisplayText><record><rec-
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name="Journal Article">17</ref-
type><contributors><author>Cushing,
L.</author><author>Steinmaus, C.</author><author>Miller,
M.D. < /author > < author > Smith,
A.H.</author></authors></contributors><titles><title>The effects of high
iodine intake on perchloraterelated decreases in thyroid hormone, NHANES
2001–2 [Abstract 0-0128]. In: Abstracts of the 2011 Conference of the
International Society for Environmental Epidemiology
(ISEE)</title><secondary-title>Environmental Health Perspectives</secondary-
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timestamp="1500666355">1907</key></foreign-keys><ref-type
name="Journal Article">17</ref-
type><contributors><authors><author>Vanderver,
G.B. </author><author>Engel, A. </author><author>Lamm,
S.</author></authors></contributors><titles><title>Cigarette smoking and
iodine as hypothyroxinemic stressors in U.S. women of childbearing age: a
NHANES III analysis</title><secondary-title>Thyroid</secondary-
title></titles><periodical><full-title>Thyroid</full-
title></periodical><volume>17</volume><section>741-
746</section><dates><vear>2007</vear></dates><urls></urls></record></
Cite></EndNote>]. Overall, these studies, combined with our findings, suggest
that excessive iodine intakes could lead to altered thyroid function and an
enhanced susceptibility to perchlorate. (p. 865)
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Steinmaus et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Steinmaus</Author><Year>2016</Year><RecNum>258</RecNum>< DisplayText>(2016)</DisplayText><record><rec-number>258</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1468518705">258</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Steinmaus, C</author><author>Pearl, M</author><author>Kharrazi, M</author><author>Blount, BC</author><author>Miller, MD</author><author>Pearce, EN</author><author>Valentin-Blasini, L</author><author>DeLorenze G</author><author>Hoofnagle, AN</author><author>Liaw J</author></authors></contributors></title>Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California</title><secondarytitle>Environmental Health Perspectives</secondary-title></titles><periodical><fulltitle>Environmental Health Perspectives</full-title></periodical><pages>861-867</pages><volume>124</volume><dates><year>2016</year></dates><urls></urls><electronicresource-num><style face="underline" font="default" size="100%">http://dx.doi.org/10.1289/ehp.1409614</style></electronic-resourcenum></record></Cite></EndNote>] also noted a correlation between perchlorate and iodine. It is difficult to hypothesize why exposure to perchlorate would be correlated with increased iodine levels. However, given this correlation, if individuals with higher perchlorate exposure also had higher iodine intake, there may have been uncontrolled confounding that could have diminished the beta estimate.

Additionally, Knight et al. (2018) evaluated 308 third trimester pregnant women at 36-38 weeks gestation and found urinary perchlorate to be inversely related to fT4. Specifically, in a cohort of women from South-West England, urinary iodine, perchlorate and thiocyante was measured along with fT4, TSH and TPO-Ab. Using multiregression analysis the authors found that for every 10 percent increase in urinary perchlorate there was a 0.03 pmol/L decrease in maternal serum fT4.

Overall, the studies specifically evaluating pregnant women's exposure to perchlorate and its potential association with thyroid hormone levels is mixed with the Pearce et al. (2010, 2011, 2012), Suh et al. (2014) and Mortensen et al. (2016) not finding an association while Steinmaus et al. (2016) and Charatcharoenwitthaya et al. (2014) see an association. Further, Knight et al. (2018) also saw an association when perchlorate co-occurs with other goitrogens. These possible explanations for the differences in findings include that the study populations are from different countries and different populations of women who may have different co-exposures, different iodine intake status and variable gestational age, which has a profound impact on circulating thyroid hormone levels.

#### 7.2.4 Environmental Epidemiology Studies – Infants

Cao et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Cao</Author><Year>2010</Year><RecNum>125</RecNum><Display Text>(2010)</DisplayText><record><rec-number>125</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1466187678">125</key></foreign-keys>< ref-type name="Journal Article">17</ ref-type>< contributors>< author>Cao, Y.</ author>Cauthor>Blount, B.

C.</author><author>Valentin-Blasini, L.</author><author>Bernbaum, J.

C.</author><author>Phillips, T. M.</author><author>Rogan, W.

J.</author></authors></contributors><auth-address>Epidemiology Branch, National Institute of Environmental Health Sciences, National Institutes of Health, Department of Health and Human Scrvices, Research Triangle Park, North Carolina, USA.</auth-address><titles><title>Goitrogenic anions, thyroid-stimulating hormone, and thyroid hormone in infants</title><secondary-title>Environ Health Perspect</secondary-title></title><periodical><full-title>Environ Health Perspect</full-

title></periodical><pages>1332-

7</pages><volume>118</volume>9</number>9</number><cdition>2010/05/05</edition><keywords ><keyword>Antithyroid Agents/urine</keyword><keyword>Environmental Exposure/adverse effects</keyword><keyword>Keyword>Humans</keyword><keyword>Infant</keyword><keyword>Infant</keyword><keyword>Infant</keyword><keyword>Infant</keyword><keyword>Infant</keyword><keyword>Infant</keyword><keyword><keyword>Infant</keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><keyword><k

Newborn</keyword>Iodides/urine</keyword>Keyword>Male</keyword>Keyword>Ni trates/toxicity/urine</keyword>Keyword>Perchlorates/toxicity/urine</keyword>Thiocya nates/toxicity/urine</keyword>Keyword>Thyroid Hormones/

urine</keyword><keyword>Thyrotropin/

toxicity</keyword></keywords><dates><year>2010</year><pub-dates><date>Sep</date></pubdates></dates><isbn>1552-9924 (Electronic)&#xD;0091-6765 (Linking)</isbn><accessionnum>20439182</accession-num><urls></urls><custom2>2944098</custom2><electronic-resourcenum>10.1289/ehp.0901736</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>] examined the association of urinary perchlorate, nitrate, iodide, and thiocyanate with urinary T4 and TSH in infants and examined if these associations differed by sex or iodide status, using data and samples from the Study of Estrogen Activity and Development. (Note: the other studies described evaluated serum, not urinary, T4 and TSH.) Cao et al.'s study included 92 full-term infants between birth and 1 year of age. Perchlorate thiocyanate, nitrate, and iodide were measured in 206 urine samples; TSH and T4 were measured in 50 blood and 206 urine samples. Using mixed models, Cao et al. determined infants with higher urinary perchlorate, nitrate, or thiocyanate had higher urinary TSH. When all three covariates were modeled together, children with higher nitrate and thiocyanate had higher urinary TSH, but higher perchlorate was associated with TSH only in children with low iodide. The interpretation of this study is made more difficult by the contradictory finding that with more perchlorate, the infants' urinary T4 levels tended to be higher; this was the opposite of what is expected. Further, how urinary TSH (or T4) relates to serum thyroid hormone levels is a question that complicates the interpretation of these study results.

Using a cohort of 64 mother infant pairs from Boston, Massachusetts Leung et al. [ ADDIN EN.CITE ADDIN EN.CITE ADDIN EN.CITE.DATA ] studied the potential association between infant (1-3 months old) exposures to perchlorate, thiocyanate and iodine (measured in maternal breast milk and also infant urine) and thyroid function. Maternal urinary perchlorate was also measured. Median (range) perchlorate concentrations were 4.4 μg/L (0.5-29.5 μg/L) in breast milk, 3.1 μg/L (0.2-22.4 μg/L) in maternal urine, and 4.7 μg/L (0.3-25.3 μg/L) in infant urine. Urine iodine concentrations were 101.9 μg/L (27-570 μg/L) in mothers and 197.5 μg/L (40-785 μg/L) in infants. In multivariable regression analyses which included breast milk, maternal urine, and infant urine thiocyanate concentrations, no evidence of an association was seen between breast milk, maternal urine, and infant urine perchlorate concentration and infant serum fT4 (logarithm transformed) or TSH (logarithm transformed). The authors noted that they were not able to achieve their targeted enrollment of 275 mother-infant pairs, and state that, "The recruited study population was underpowered to determine the statistical significance of perchlorate and thiocyanate exposures on serum infant thyroid function."

A study by Ucal et al. [ ADDIN EN.CITE <EndNote><Cite
ExcludeAuth="1"><Author>Ucal</Author><Year>2018</Year><RecNum>2240</RecNum><Displ
ayText>(2018)</DisplayText><record><rec-number>2240</rec-number><foreign-keys><key
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timestamp="1543347914">2240</key></foreign-keys><ref-type name="Journal Article">17</ref-

type><contributors><author>Ucal, Y.</author><author>Sahin, O.N. </author><author>Serdar, M. </author><author>Blount, B. </author><author>Kumru, P.</author><author>Muhcu, M.</author><author>Eroglu, M.</author><author>Akin-Levi, C.</author><author>Keles, Z.</author><author>Turam, C.</author><author>Valentin-Blasini, L.</author><author>Morel-Espinosa, M.</author><author>Serteser, M.</author><author>Unsal, I.</author><author>Ozpinar, A.</author></authors></contributors><title>Exposure to perchlorate in lactating women and its associations with newborn thyroid stimulating hormone</title><secondary-title>Frontiers in Endocrinology</secondarytitle></titles><periodical><full-title>Frontiers in endocrinology</fulltitle></periodical><volume>9</volume><section>348</section><dates><year>2018</year></dates> <url>></urls></record></Cite></EndNote>] studied 185 mother child pairs from Istanbul, Turkey and the association between maternal urinary perchlorate, thiocyanate and nitrate concentrations and thyroid hormone concentrations in infants. Thyroid hormones were measured in both the mothers (blood collected with 48 hours of delivery) and the infants (blood collected between 48 and 72 hour after delivery). Perchlorate was also analyzed in the colostrum. The medians of maternal urinary perchlorate (4.00μg/g creatinine), maternal urinary thiocyanate (403μg/g creatinine), and maternal urinary nitrate (49,117µg/g creatinine) were determined. Having concentrations of all three urinary NIS inhibitors (µg/g creatinine) above their 75th percentile levels was significantly correlated with higher newborn TSH(r = 0.21, p < 0.001), however having two or one of the NIS inhibitor concentrations above the 75th percentile had no statistically significant association with newborn TSH. Median colostrum perchlorate level concentration of all 185 participants was 2.30 µg/L. Colostrum perchlorate was not significantly correlated with newborn TSH (p > 0.05); however, there was a significant correlation between colostrum perchlorate level and maternal TSH (r = 0.21, p < 0.01). Similarly, there was a significant positive association between colostrum perchlorate and maternal urinary creatinine adjusted perchlorate (r = 0.32, p < 0.001).

The limited evidence on perchlorate exposures in infants (measured as a proxy through breast milk or maternal or infant urine) demonstrated mixed results on the association between perchlorate and thyroid function. The power in the cited studies to detect an effect is small, with the greatest n equal to 185.

# 7.3 Summary

In summary, most low dose clinical studies have not seen an association between perchlorate dosing and thyroid hormone levels, but in most instances have observed reduced iodine uptake by the thyroid gland. On the contrary, many U.S. based general population studies have observed an association between low level perchlorate exposure and changes in thyroid hormone levels, with the impact being greatest in individuals with low iodine or co-exposures to other goitrogens. Additionally, the evidence for potentially sensitive populations of pregnant women is more mixed. The evidence in infants is also much more limited.

Exposures to perchlorate and resulting changes in thyroid hormone levels may lead to adverse health effects, particularly in sensitive populations such as pregnant women, fetuses, and infants (Taylor et al., 2014). Thyroid hormones are critical to neurodevelopment of the fetus, and requirements for these hormones increase during pregnancy [ ADDIN EN.CITE

<EndNote><Cite><Author>Leung</Author><Year>2010</Year><RecNum>117</RecNum><DisplayText>(Leung, Pearce, & DisplayText><record><rec-number>117</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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N</author><author>Braverman, L</author></authors></contributors><titles><title>Perchlorate,
iodine and the thyroid</title><secondary-title>Best Practice & periodical Endocrinology
& periodical Endocrinology & periodical><full-title>Best Practice & periodical>Fractice & periodical>
Clinical Endocrinology & periodical>Fractice & periodical>
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# 8. Potentially Sensitive Life Stages

Because SDWA requires that the MCLG be set at the level at which there are no adverse effects to the health of persons and because SDWA requires that EPA assess the effects of the contaminant on subpopulations likely to be at greater risk from exposure to the contaminant, EPA identifies the most sensitive populations to derive the MCLG. The SAB states, "the sensitive populations that EPA should consider for exposure to perchlorate are the fetuses of hypothyroxinemic pregnant women, and infants exposed to perchlorate through either water-based formula preparations or the breast milk of lactating women" [ ADDIN EN.CITE

<EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Suffix>` ; p. 11</Suffix><DisplayText>(SAB, 2013; p. 11)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB,</author></author></author></arthrap authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></date>><vear>2013</vear></dates><publocation>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>]. The current analysis presented in this document aims to protect the fetus of a hypothyroxinemic pregnant woman. More specifically, the ultimate MCLG is based on protecting the fetus of a first trimester pregnant mother with low-iodine intake levels (i.e., 75 µg/kg/day), low fT4 levels (i.e., 10th percentile of a fT4 distribution for individuals with 75 µg/day iodine intake), and weak TSH feedback strength (defined as TSH feedback reduced to approximately 60 percent less effective than for the median individual). This section evaluates if protecting this population will also be protective of breast- and bottle-fed infants by evaluating two sensitivities: (1) the sensitivity of the thyroid to perchlorate exposure for each population, and (2) the sensitivity of the brain to thyroid hormone perturbations.

# 8.1 Sensitivity of the Thyroid to Perchlorate Exposure

To directly compare the relative sensitivities of the predicted impacts of perchlorate exposure on thyroid hormones, EPA can evaluate the BBDR model predictions for changes in fT4 for the breast-and formula-fed infant and compare these changes to the predictions for the first trimester pregnant mother. Given that the fetus does not yet have an operational thyroid in the first trimester, it is fully dependent on the mother's thyroid [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller
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Year><RecNum>194
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(Zoeller & amp; Rovet, 2004)
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ref-type>
contributors><author>Zoeller
R Thomas
author><author>Rovet
Rovet

<sup>&</sup>lt;sup>15</sup> The breast- and formula-fed infant BBDR model was peer reviewed in January 2017. No updates were made to the models for these life stages as a result of the peer review, given the strong recommendation from the review panel to develop a first trimester model. However, to directly compare the changes in fT4 for specific doses of perchlorate, the 2017 peer-reviewed model must be used. (For more information on these breast- and bottle-fed infant models, see the *Draft Biologically Based Dose-Response (BBDR) Model and Draft BBDR Model Report for Perchlorate in Drinking Water* (March 29, 2017).)

J</author></authors></contributors><titles><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title><periodical><full-title>Journal of neuroendocrinology</full-title></periodical><pages>809-

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>1365-2826</isbn><urls></record></Cite></EndNote>]. Subsequently, the thyroid hormone levels predicted for the first trimester pregnant mother represent the hormone levels available to the fetus.

Although formula-fed infants are potentially vulnerable to increased impacts of contaminants in drinking water given their high intake rates per unit of bodyweight, the BBDR model demonstrated that perchlorate barely impacts the thyroid hormone levels for 30-, 60-, and 90-day formula-fed infants, even up to doses as high as 20  $\mu$ g/kg/day. Specifically, "the range of iodine levels in formula is sufficient to almost entirely offset the effects of perchlorate exposure at 30, 60 and 90 days" [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2016</Year><RecNum>246</RecNum><Suffix>`; p. 73</Suffix><DisplayText>(U.S. EPA, 2016; p. 73)</DisplayText><record><rec-number>246</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468339271">246</key></foreign-keys><ref-type name="Generic">13</ref-type><contributors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><a

title></titles><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>]. \(^{16}\) The day-7 infant showed a slightly steeper dose-response curve than the very shallow dose-responses of the 30, 60 and 90 day infants. For the day-7 infant, it was assumed that perchlorate exposure occurred as a fetus through day 7 and that the mother had an iodine intake rate of 200  $\mu$ g/day. It is therefore concluded that the steeper dose-response for the day-7 infant was a result of "in utero exposure with

<sup>&</sup>lt;sup>16</sup> The ranges of iodine intake are based on data presented in [ HYPERLINK \l "\_ENREF\_108" \o "Pearce, 2004 #1400" ]. Note that two samples in [ HYPERLINK \l "\_ENREF\_108" \o "Pearce, 2004 #1400" ] were soy-based formulas. Soy-based formulas have isoflavones, which disrupt the normal thyroid function, although there is debate about the impact of isoflavones from soy formulas [ ADDIN EN.CITE <EndNote><Cite><Author>Canadian Paediatric

Society</author><Year>2009</Year><RecNum>2021</RecNum><DisplayText>(Canadian Paediatric Society, 2009)</br>
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Society, 2009)</br>
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<sup>113&</sup>lt;/pages><volume>14</volume><number>2</number><dates><year>2009</year></dates><urls></record></Cite></EndNote>]. The BBDR model did not consider other ingredients in the formula that may impact the perchlorate/thyroid hormone relationship.

border-line inadequate maternal iodine ingestion" [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2016</Year><RecNum>246</RecNum><Suffix>`; p.

73</Suffix><DisplayText>(U.S. EPA, 2016; p. 73)</DisplayText><record><rec-number>246</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468339271">246</key></foreign-keys><ref-type name="Generic">13</ref-type><contributors><author>><author>><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author

title></titles><dates><year>2016</year></dates><urls></record></Cite></EndNote>]. By day 30, it is clear the iodine in the formula has offset the inadequate maternal iodine intake and the fetal perchlorate. The EPA has concluded that based on the BBDR model results for the formula-fed infant, any value set for the MCLG based on the fetus of the first trimester hypothyroxinemic pregnant mother would also protect the formula-fed infant.

To determine if the same would be true for the breast-fed infant, the EPA compared the predicted percent change in fT4 experienced for given doses of perchlorate for both the breast-fed infant and the fetus of a first trimester, low-iodine intake (75  $\mu$ g/day) pregnant mother. This comparison is presented in [ REF \_Ref519676368 \h ]. Given that the current version of the BBDR model contains a TSH feedback loop and the infant models previously developed did not contain this loop, this comparison is done with the feedback loop turned off.<sup>17</sup>

[ REF \_Ref519676368 \h \\* MERGEFORMAT ] shows that when exposed to the same doses of perchlorate at iodine intakes of 75 and 100 µg/day, the first trimester pregnant mother has the largest resulting percent change in fT4 when compared to the 30- and 60-day infant. However, in the 50 µg/day iodine intake group, the 30-day infant is the most sensitive to changes in perchlorate, while approximately the same percent reduction in fT4 is seen when comparing the 60-day infant to the first trimester mother. In order to understand the relative sensitivity of the first trimester fetus compared to the breast-fed infant, it is necessary to understand the neurodevelopmental implications of the fT4 changes seen in [ REF \_Ref519676368 \h ], as similar perturbations in thyroid hormones could have different neurodevelopmental effects based on life-stage. The relative sensitivity of the brain to perturbations in thyroid hormones between the first trimester fetus and the breast-fed infant is considered in the subsequent section.

<sup>&</sup>lt;sup>17</sup> Ideally this comparison would be done with a TSH feedback loop in the infant models as well. However, given the need to move forward with the analysis in the given time frame this was not feasible.

Table [ SEQ Table \\* ARABIC ]. Predicted Changes in fT4 for Given Maternal Iodine Intakes and Perchlorate Doses for the Pregnant Mother, 30-Day and 60-Day Breast-Fed Infant

Maternal	Maternal Perchlorate Dose (µg/kg/day)	Predicted fT4 Concentration (pmol/L) (% change from 0 dose of perchlorate)				
lodine Intake		1st Trimester	Breast-Fed Infant			
(µg/d)		GW12*	30 days	60 days		
	0	10.32	25.31	25.23		
	2	10.13	25.07	25.00		
100	2	(-1.84%)	(-0.95%)	(-0.91%)		
	4	9.91	24.82	24.75		
	4	(-3.97%)	(-1.94%)	(-1.90%)		
	0	8.18	24.86	24.76		
	2	7.92	24.47	24.42		
75	2	(-3.18%)	(-1.65%)	(-1.37%)		
	4	7.67	24.04	24.05		
		(-6.23%)	(-3.30%)	(-2.87%)		
	0	5.50	23.02	23.54		
50	2	5.32	21.84	22.83		
		(-3.27%)	(-5.13%)	(-3.02%)		
	4	5.14	20.64	22.03		
		(-6.55%)	(-10.34%)	(-6.41%)		

<sup>&</sup>lt;sup>a</sup> The TSH feedback loop is turned off for these analyses in order to compare the infant model that did not have a TSH feedback loop component.

# 8.2 Sensitivity of the Brain to Perturbations in Thyroid Hormones

The literature clearly indicates that minor perturbations in thyroid hormone levels in the first trimester mother can adversely impact the offspring's neurodevelopment. As stated by the SAB (and supported by the literature review in the Approaches Report), "Children exposed gestationally to maternal hypothyroxinemia (without hypothyroidism) show reduced levels of global and specific cognitive abilities, as well as increased rates of behavior problems including greater dysregulation in early infancy and attentional disorders in childhood [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Notably these effects are correlated with both degree [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] and duration [ ADDIN EN.CITE

<EndNote><Cite><Author>Pop</Author><Year>2003</Year><RecNum><DisplayText>(Pop et al., 2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</td>

timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E

P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url

s></urls></record></Cite></EndNote>] of maternal hypothyroxinemia" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Suffix>`; p. 10</Suffix><DisplayText>(SAB, 2013; p. 10)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author><Author><Author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></author></a></al></al>

The literature does not provide analogous evidence linking minor perturbations in thyroid hormones in infants to adverse neurodevelopmental outcomes. As stated by the California Environmental Protection Agency (CalEPA) in their assessment of a public health goal for perchlorate, "the fetus is highly sensitive to any changes in thyroid hormone levels during pregnancy. It is unknown whether the neonate is similarly sensitive" [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1">Exc

water</title></title></title></dates><year>2015</year></dates><urls></record></Cite></EndNote> ]. One study was located that evaluated both the impact of maternal hypothyroxinemia and infant fT4 levels on subsequent neurodevelopmental outcomes. [HYPERLINK \l "\_ENREF\_33" \o "Costeira, 2011 #7" ] found that children born to mothers with low fT4 in the first trimester had increased odds of mild-to-severe delays in psychomotor development compared to children born to mothers with normal fT4 levels but that neonatal thyroid status (measured on day 3 after birth) did not influence development. [HYPERLINK \l "\_ENREF\_58" \o "Henrichs, 2010 #928" ] found in their evaluation that the neonatal thyroid status (measured in cord blood) did not explain the relationship between maternal hypothyroxinemia, early pregnancy, and children's cognitive impairment.

However, the SAB pointed out two lines of evidence that suggested that "the infant may be vulnerable to perchlorate exposure" preterm infants that experience transient hypothyroxinemia of prematurity (THOP) and infants that experience congenital hypothyroidism (SAB, 2013, p. 11). According to the SAB, "THOP arises because the fetal thyroid system is immature if a child is born preterm and the late-gestational maternal iodine and thyroid hormone supplies are no longer available [ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Follow-up studies of THOP report reduced [intelligence quotient] (IQ) [ADDIN EN.CITE

<EndNote><Cite><Author>Lucas</Author><Year>1996</Year><RecNum>2015</RecNum><Displ ayText>(Lucas, Morley, & DisplayText><record><rec-number>2015</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532362612">2015</key></foreign-keys><ref-type name="Journal Article">17</ref-type name="Journal Article">17</ref-

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type><contributors><author>A. Lucas</author>R. Morley </author>M. Fewtrell</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></a></author></a></author></author></a></ale author></a></ale author></ale author></a></ale author></a></ale author></ale author></ale author></ale author></ale author></ale author></ale author></ale author></a></ale author></ale auth
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3</pages><volume>312</volume><dates><year>1996</year></dates><urls></urls></record></Cite></EndNote>], impaired visual skills [ ADDIN EN.CITE

<EndNote><Cite><Author>Rovet</Author><Year>2008</Year><RecNum>2016</RecNum><Displ ayText>(Rovet & DisplayText><record><rec-number>2016</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532362698">2016</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Rovet, J</author><author>N.

Simic</author></authors></contributors><titles><title>The role of transient hypothyroxinemia of prematurity in development of visual abilities</title><secondary-title>Sem Perinatol</secondary-title></periodical><full-title>Sem Perinatol</full-title></periodical><pages>431-

7</pages><volume>32</volume><dates><year>2008</year></dates><urls></urls></record></Cite></EndNote>] and an increased incidence of neurological dysfunction and school failure [ ADDIN EN.CITE

<EndNote><Cite><Author>Ouden</Author><Year>1996</Year><RecNum>2017</RecNum><Disp layText>(Ouden, Kok, Verkerk, Brand, & Disp layText)

1996)</DisplayText><record><rec-number>2017</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532362793">2017</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>>A.L. Den Ouden</author><author><Author><author><R. Brand</author><author>S.P. Verloove-

Vanhorick</author></authors></contributors><titles><title>The relation between neonatal thyroxine levels and neurodevelopmental outcome at Age 5 and 9 years in a national cohort of very preterm and/or very low birth weight infants</title><secondary-title>Pediatr Res</secondary-title>Color of the state o

title></title></periodical><full-title>Pediatr Res</full-title></periodical><pages>142-

5</pages><volume>39</volume><dates><year>1996</year></dates><urls></urls></record></Cite></EndNote>], cognitive disabilities [ ADDIN EN.CITE

<EndNote><Cite><Author>Simic</Author><Year>2010</Year><RecNum>2014</RecNum><DisplayText>(Simic & DisplayText><record><rec-number>2014</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532362522">2014</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>N. Simic</author><author>J.

Rovet</author></contributors></title>Transient hypothyroxinemia of prematurity: current state of knowledge</title><secondary-title>Thyroid Intern</secondary-

title></title></periodical><full-title>Thyroid Intern</full-title></periodical><pages>1-

13 < pages > < dates > < year > 2010 < / year > < / dates > < / urls > < / record > < / Cite > < / EndNote > ], cerebral palsy [ ADDIN EN.CITE

<EndNote><Cite><Author>Reuss</Author><Year>1996</Year><RecNum>2018</RecNum><DisplayText>(Reuss, Paneth, Pinto-Martin, Lorena, & Susser, 1996)</DisplayText><record><recnumber>2018</rec-number><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532363214">2018</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>M. L.

Reuss</author><author>N. Paneth</author><author>J. A. Pinto-Martin</author><author>J. M. Lorena</author><author>M. Susser</author></contributors><title>><title>The relation of transient hypothyroxinemia in preterm infants to neurologic development at two years of age</title><secondary-title>New Engl J Med</secondary-title></title><periodical><full-title>New Engl J Med</full-title></periodical><pages>821-7</pages><volume>334</volume><dates></ear>1996</vear></dates><urls></urls></record></Cite ></EndNote>] with the most severe effects being seen in the lowest levels of thyroid hormones. Further, congenital hypothyroidism, which arises from a defect in thyroid gland formation or function or its central regulation by the hypothalamus and pituitary [ ADDIN EN.CITE <EndNote><Cite><Author>Rovet</Author><Year>2003</Year><RecNum>2019</RecNum><Displ ayText>(Rovet & Daneman, 2003)</br>

ayText>
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2019 number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532363287">2019</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Rovet, J</author><author>D. Daneman</author></authors></contributors><title>Congenital hypothyroidism: a review of current diagnostic procedures and treatment</title><secondary-title>Pediatric Drugs</secondarytitle></title><periodical><full-title>Pediatric Drugs</full-title></periodical><pages>141-149</pages><volume>5</volume><dates></ear>>2003</vear></dates><urls></urls></record></Cite ></EndNote>], is associated with mental retardation and severe behavior problems if untreated in the newborn period [ ADDIN EN.CITE <EndNote><Cite><Author>Rovet</Author><Year>1992</Year><RecNum>2020</RecNum><Displ ayText>(Rovet, Ehrlich, & Sorbara, 1992)</DisplayText><record><rec-number>2020</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1532363431">2020</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Rovet, J</author><author>R. M. Ehrlich</author><author>D. L. Sorbara</author></authors></contributors></title>Neurodevelopment in infants and preschool children with congenital hypothyroidism: etiological and treatment factors affecting outcome</title><secondary-title>Journal of Pediatric Psychology</secondarytitle></title></title>>cperiodical><full-title>Journal of Pediatric Psychology</fulltitle></periodical><pages>187-213</pages><volume>17</volume><number>2</number><dates><vear>1992</vear></dates><urls ></urls></record></Cite></EndNote>]" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Suffix>` ; p. 11</Suffix><DisplayText>(SAB, 2013; p. 11)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government

Clearly sufficient thyroid hormone levels in infancy are necessary for the infant brain to develop properly. However, the evidence linking perturbations in thyroid hormone levels to disrupted neurodevelopment for infants is in individuals with clinical conditions (i.e., THOP and congenital hypothyroidism). It is unknown if minor perturbations in thyroid hormones in infants would result in

Document">46</ref-type><contributors><author>SAB,</author></author></author></author>

authors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

authors><author>U.S. Environmental Protection Agency,</author></secondary-

location>Washington, DC</pub-location><urls></record></Cite></EndNote>].

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adverse neurodevelopmental outcomes similar to that seen in the literature for the offspring of first trimester pregnant mothers with hypothyroxinemia. Therefore, maternal thyroid hormone perturbations in the first trimester may be more significantly associated with the offspring neurodevelopment [ ADDIN EN.CITE | ADDIN EN.CITE.DATA |] than infant thyroid hormones. The EPA concludes that deriving and MCLG to protect a fetus in the first trimester will be protective of other potentially sensitive life stages as well.

# 9. Select the Approach to Inform the POD

Consistent with SAB recommendations, the EPA connected modeled thyroid hormone effects from perchlorate exposure in pregnant women to neurodevelopmental outcomes in their offspring. This was done using a combination of the BBDR model and the epidemiology literature that relates changes in maternal thyroid hormones to neurodevelopmental outcomes in children. This process is described in [ REF Ref516151159 \h ].

For each step of this process, decisions were made about the most appropriate inputs to the analysis. In the subsections that follow, EPA summarizes each step of the approach to inform an MCLG along with each decision point.

Figure [ SEQ Figure \\* ARABIC ]. Summary of Modeling Approach for Estimating Measurable Adverse Neurodevelopmental Impacts in Offspring from Perchlorate Exposure in Pregnant Woman

[EMBED Visio.Drawing.15]

# 9.1 Summary of BBDR Model and Decision Points (Step 1)

The BBDR model has two main components: (1) a pharmacokinetic model for perchlorate and iodide, which describes chemical absorption, distribution, metabolism, and excretion of these two anions; and (2) a pharmacodynamic model, which describes the joint effect of varying perchlorate and iodide blood concentrations on thyroidal uptake of iodide and subsequent production of thyroid hormones, most significantly T4. The pharmacokinetic portion contains a physiological description (e.g., organ volumes, blood flows) and chemical-specific information (e.g., partition coefficients, volume of distribution, rate constants for transport, metabolism, and elimination) that enable a prediction of perchlorate and iodide internal concentration at the critical target (i.e., thyroidal sodium-iodide symporter (NIS) in association with a particular exposure scenario (route of exposure, age, dose level)). This portion of the model is similar to other PBPK models and for perchlorate is simplified by the absence of metabolism. The pharmacodynamic portion of the model uses this internal concentration to simulate how the chemical will act within a known mechanism of action to perturb host systems and lead to a toxic effect. Thus, BBDR modeling attempts to predict the internal dose of a chemical associated with a particular exposure scenario and the perturbation this internal dose can have on host systems.<sup>18</sup>

The BBDR model predicts serum thyroid hormone levels dependent on variables for gestational week (GW), iodine intake, TSH feedback loop strength, and perchlorate dose. The EPA aimed to model a sensitive individual in their analysis to inform the MCLG in an effort to protect the sensitive population with an adequate margin of safety, using individuals with weaker than median TSH feedback and lower than median fT4 levels. To demonstrate the output from the BBDR model for GW 13, data from which will ultimately inform the MCLG, and to demonstrate the impact of iodine intake and the strength of the TSH feedback [ REF \_Ref517525852 \h \\* MERGEFORMAT ] is provided with the muted TSH feedback loop, and [ REF \_Ref512860812 \h \\* MERGEFORMAT ]

<sup>&</sup>lt;sup>18</sup> For additional information on the BBDR model, refer to Chapter 3 and Appendix A in the Approaches Report.

with the full TSH feedback loop. pTSH refers to the strength of the TSH feedback loop. pTSH=0.398, which is the coefficient for the muted TSH feedback loop used to produce the predictions in [ REF \_Ref512860801 \h ], is equal to the ratio of a median value for TSH from NHANES (non-pregnant women) to the 97.5 percentile value from NHANES (non-pregnant women). The full-strength TSH feedback loop used to produce the results in [ REF \_Ref529546636 \h ] has a pTSH equal to 1.

Table [ SEQ Table \\* ARABIC ]. Summary of BBDR Model Results for fT4 for GW 13 at 75 and 170  $\mu$ g/day lodine Intake at pTSH<sup>a</sup> = 0.398 []

Perchlorate Dose	Percentile fT4 (pmol/L) (% change from 0 dose)					
(µg/kg/day)	2.5th	5th	10th	50th	95th	97.5th
		la di s				
0	6.70	7.33	8.07	10.64	13.95	14.59
1	6.70	7.33	8.06	10.64	13.95	14.58
I	( -0.08%)	(-0.07%)	(-0.06%)	(-0.05%)	(-0.04%)	(-0.03%)
2	6.69	7.32	8.06	10.63	13.94	14.58
2	(-0.15%)	(-0.14%)	(-0.13%)	(-0.10%)	(-0.07%)	(-0.07%)
3	6.69	7.32	8.05	10.63	13.94	14.57
3	(-0.23%)	(-0.21%)	(-0.19%)	(-0.14%)	(-0.11%)	(-0.11%)
4	6.68	7.31	8.04	10.62	13.93	14.57
4	(-0.31%)	(-0.28%)	(-0.26%)	(-0.19%)	(-0.15%)	(-0.14%)
				6.29		
0	5.57	6.09	6.70	8.84	11.59	12.12
1	5.50	6.02	6.63	8.77	11.52	12.05
	(-1.26%)	(-1.15%)	(-1.04%)	(-0.79%)	(-0.60%)	(-0.58%)
2	5.43	5.96	6.56	8.71	11.45	11.98
	(-2.45%)	(-2.24%)	(-2.04%)	(-1.54%)	(-1.18%)	(-1.13%)
3	5.37	5.96	6.50	8.64	11.39	11.92
	(-3.59%)	(-3.28%)	(-2.98%)	(-2.26%)	(-1.72%)	(-1.65%)
4	5.31	5.83	6.44	8.58	11.33	11.86
4	(-4.68%)	(-4.28%)	(-3.89%)	(-2.95%)	(-2.25%)	(-2.15%)

<sup>&</sup>lt;sup>a</sup> See Section 3.1.2 for additional information on pTSH.

<sup>&</sup>lt;sup>b</sup> The 50th percentile is direct output from the BBDR model, and additional percentiles are estimated by assuming a normal distribution with a standard deviation (SD) of 2.01.

<sup>&</sup>lt;sup>c</sup> The 50th percentile is direct output from the BBDR model, and additional percentiles are estimated by assuming a normal distribution with a SD of 1.67.

Perchlorate Dose	Percentile fT4 (pmol/L) (% Change from 0 Dose)					
(μg/kg/day)	2.5th	5th	10th	50th	95th	97.5th
		collec		(Carlo		
0	6.70	7.33	8.07	10.64	13.95	14.59
1	6.70	7.33	8.06	10.64	13.95	14.58
1	(-0.04%)	(-0.04%)	(-0.04%)	(-0.03%)	(-0.02%)	(-0.02%)
2	6.69	7.33	8.06	10.64	13.95	14.58
2	(-0.09%)	(-0.08%)	(-0.07%)	(-0.05%)	(-0.04%)	(-0.04%)
•	6.69	7.33	8.06	10.63	13.94	14.58
3	(-0.13%)	(-0.12%)	(-0.11%)	(-0.08%)	(-0.06%)	(-0.06%)
4	6.69	7.32	8.05	10.63	13.94	14.58
4	(-0.17%)	(-0.16%)	(-0.14%)	(-0.11%)	(-0.08%)	(-0.08%)
		(4741)				
0	5.57	6.09	6.70	8.84	11.59	12.12
1	5.53	6.05	6.66	8.80	11.55	12.08
	(-0.73%)	(-0.67%)	(-0.61%)	(-0.46%)	(-0.35%)	(-0.34%)
2	5.49	6.01	6.62	8.76	11.51	12.04
	(-1.44%)	(-1.31%)	(-1.20%)	(-0.91%)	(-0.69%)	(-0.66%)
3	5.45	5.97	6.58	8.72	11.47	12.00
	(-2.13%)	(-1.94%)	(-1.77%)	(-1.34%)	(-1.02%)	(-0.98%)
4	5.41	5.94	6.54	8.69	11.44	11.96
4	(-2.80%)	(-2.56%)	(-2.32%)	(-1.76%)	(-1.34%)	(-1.29%)

<sup>&</sup>lt;sup>a</sup> See Section 3.1.2 for additional information on pTSH,

#### 9.1.1 Iodine Intake Level and Gestational Week

The BBDR model simulates the impact of perchlorate on thyroid hormones at any level of iodine intake and at each GW from conception to week 16. Thus, EPA was required to make informed decisions as to the appropriate level of iodine intake, as well as the week of gestation for which to model perchlorate exposure for the purposes of informing the MCLG.

Individuals with low-iodine intake have increased sensitivity to the impact of perchlorate on thyroid hormone levels as the functional reserve of the hypothalamic-pituitary-thyroid (HPT) system is limited with less iodine (Leung et al., 2010). The EPA selected an iodine intake level of 75  $\mu$ g/day to inform the MCLG to simulate an individual with low-iodine intake. This iodine intake level represents somewhere between the 15th and 20th percentile of the population distribution of estimated iodine intake from the NHANES. The EPA also considered conducting the MCLG analysis at the lower-iodine intake level of 50  $\mu$ g/day, which represents approximately the 5th percentile of the NHANES distribution. However, at 50  $\mu$ g/day of iodine intake, the BBDR model predicts TSH levels

<sup>&</sup>lt;sup>b</sup> The 50th percentile is direct output from the BBDR model, and additional percentiles are estimated by assuming a normal distribution with a SD of 2.01.

<sup>&</sup>lt;sup>c</sup>The 50th percentile is direct output from the BBDR model, and additional percentiles are estimated by assuming a normal distribution with a SD of 1.67.

would be elevated to within the clinically hypothyroid range before exposure to any perchlorate <sup>19</sup> (TSH ranges between 4.51 and 5.41 mIU/L at zero dose of perchlorate when evaluating GWs 12 or

<sup>19</sup> For the purposes of this analysis, the EPA evaluated the American Thyroid Association's (ATA's) 2017 recommendations for defining hypothyroidism [ ADDIN EN.CITE <EndNote><Cite><Author>Alexander</Author><Year>2017</Year><RecNum>1895</RecNum><DisplayTe xt>(Alexander et al., 2017)</br/>/DisplayText><record><rec-number>1895</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497970921">1895</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>Alexander, E. K.</author><author>Pearce, E. N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan, S.</author></authors></contributors></title>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondarytitle>Thyroid</secondary-title></title></periodical><full-title>Thyroid</full-title></periodical><pages>315-389</pages><volume>27</volume><number>3</number><dates></year>2017</year></dates><urls></rr record></Cite></EndNote>]. Specifically the ATA recommends "in the pregnancy setting, maternal hypothyroidism is defined as a TSH concentration elevated beyond the upper limit of the pregnancy-specific reference range" [ ADDIN EN.CITE <EndNote><Cite><Author>Alexander</Author><Year>2017</Year><RecNum>1895</RecNum><Pages>332 </Pages><DisplayText>(Alexander et al., 2017, p. 332)/DisplayText><record><rec-number>1895</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497970921">1895</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Alexander, E. K.</author><author>Pearce, E. N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan. S.</author></authors></contributors></title>>citle>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondarytitle>Thyroid</secondary-title></title></periodical><full-title>Thyroid</full-title></periodical><pages>315-389</pages><volume>27</volume><number>3</number><dates><year>2017</year></dates><urls></ record></Cite></EndNote>]. ATA goes on to state, in the absence of population- and trimester-specific reference ranges defined by a provider's institute or laboratory, that the TSH reference ranges should be obtained from similar patient populations. From their recommended studies with trimester-specific data on a U.S. population, Lambert-Meserlian et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Lambert-Messerlian</Author><Year>2008</Year><RecNum>100</RecNum><DisplayText>(2008)</DisplayText><re cord><rec-number>100</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1443808320">100</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Lambert-Messerlian, Geralyn</author><author>McClain, Monica</author><author>Haddow, James E</author><author>Palomaki, Glenn E</author><author>Canick, Jacob A</author><author>Cleary-Goldman, Jane</author><author>Malone, Fergal D</author><author>Porter, T Flint</author><author>Nyberg, David A</author><author>Bernstein. Peter</author></authors></contributors><title>First-and secondtrimester thyroid hormone reference data in pregnant women: a FaSTER (First-and Second-Trimester Evaluation of Risk for an euploidy) Research Consortium study</title><secondary-title>American Journal of Obstetrics and Gynecology</secondary-title></title> gynecology</full-title></periodical><pages>62e1</pages><volume>199</volume><number>1</number><dates><year>2008</year></dates><publisher>Else vier</publisher><isbn>0002-9378</isbn><urls></urls></record></Cite></EndNote>] is the largest U.S.-based

population with a reference range upper bound of 3.37 mIU/L for the first trimester (and 3.35 mIU/L for the

13). In contrast, at 75  $\mu$ g/day iodine, the BBDR-modeled concentrations of serum fT4 and TSH are significantly reduced from the population median but are still within the euthyroid range. This comports with the goal of evaluating perchlorate effects on the offspring of pregnant women who are borderline hypothyroxinemic on child neurodevelopment.

As previously stated, the EPA determined that modeling the first trimester would be protective of other potentially sensitive life stages. As will be described in Section [ REF \_Ref520707359 \r \h \\* MERGEFORMAT ], the EPA used the Korevaar et al. (2016) study to inform the relationship between maternal fT4 and neurodevelopmental outcomes in children.<sup>20</sup> This study had a median gestational age of 13.2 weeks at the time of thyroid hormone collection. Therefore, the EPA selected GW 13 BBDR output for Step 1 of the analysis.

### 9.1.2 Strength of the TSH Feedback Loop

TSH increases in response to decreases in T4 have been captured in numerous studies that document the relationship between these hormones [ ADDIN EN.CITE | ADDIN EN.CITE.DATA | ]. The BBDR model achieves this feedback regulation by adjusting a set of three parameters: the number of sodium-iodide symporter sites, the T4 synthesis rate, and the T3 synthesis rate. These three parameters are assumed to vary directly with the variation of the coefficient pTSH, which determines the strength of TSH stimulation. A lower bound of pTSH = 0.398 was used for the dose-response analysis to link changes in fT4 to neurodevelopmental outcomes. This assumes that sensitive individuals with high TSH and average fT4 levels exist because the stimulus strength of TSH is proportionally weaker. TSH increases can drive corresponding increases in iodide uptake and T4 and T3 production if the iodide supply is sufficient. With increasingly low-iodide intake, colloidal iodide stores are modeled to decline, which ultimately impairs the thyroid's response to elevations in TSH. The EPA chose to use a low TSH feedback coefficient to ensure protection for the sensitive population from adverse health effects.

#### 9.1.3 Percentile of fT4

When modeling changes in fT4 due to changes in perchlorate from the BBDR model, the baseline level of fT4 affects the magnitude of changes seen as a result of perchlorate exposure. While the BBDR model can be calibrated to match specific percentiles of the population, the distributions in the underlying parameters required to predict the distribution of thyroid hormone levels have not been determined. Therefore, in order to predict the impact of perchlorate exposure on the population distribution of fT4, the EPA estimated a distribution for fT4 plasma concentrations around the median modeled values, while accounting for the effects of perchlorate and iodine. To develop population distributions for fT4 concentrations around the BBDR-estimated median fT4 values, the EPA evaluated the fT4 data from studies that were used to calibrate the BBDR model [ADDIN EN.CITE

ADDIN EN.CITE.DATA ].<sup>21</sup> The EPA assumed the variation around predicted fT4 concentrations would likely be close to normal after accounting for perchlorate and iodine intake, and

second trimester). Therefore, these values were used to compare to BBDR output TSH values in the first trimester (or second trimester in cases of GWs 15 and 16) to determine the presence of hypothyroidism.

<sup>&</sup>lt;sup>20</sup> See Section 6.3.2 of the EPA's Approaches Report for details.

<sup>&</sup>lt;sup>21</sup> See Figure A-33 in Appendix A of the Approaches Report.

thus estimated a combined standard deviation (SD) using the distributional information from each of the studies [ADDIN EN.CITE ADDIN EN.CITE.DATA]. The EPA then used the estimated combined SD to predict a distribution of fT4 around the median fT4 estimated by the BBDR model, based on particular levels of iodine sufficiency and perchlorate doses. From this distribution around the median, the EPA chose to use the 10th percentile of baseline fT4 to conduct analyses in order to model an individual with low baseline fT4 without the uncertainty of modeling effects at more extreme percentiles (e.g., 2.5th percentile) of fT4. By using the 10th percentile of fT4 EPA is accounting for variability in thyroid hormones in the sensitive population.

# 9.2 Summary of Epidemiological Literature Connecting Changes in Maternal Thyroid Hormone Levels to Offspring Neurodevelopmental Outcomes and Decision Points (Step 2)

Data for the second step of the analysis come from epidemiological studies that evaluated maternal thyroid hormone levels in pregnancy and neurodevelopmental outcomes and not from studies evaluating perchlorate exposure. Based on recommendations of previous peer-review panels, the EPA evaluated epidemiological literature assessing the relationship between thyroid hormone levels and neurodevelopmental outcomes, regardless of the cause of hormone perturbations. As this evaluation followed the SAB recommendation, the EPA did not conduct a complete, systematic review of the body of literature and a weight-of-evidence evaluation on this topic. Instead, the EPA conducted a focused review of the published literature and identified epidemiological studies that examined thyroid hormone levels and neurodevelopmental outcomes.

#### 9.2.1 Selection of the Approach for Step 2

Two potential methods for using the epidemiological literature to inform an MCLG were outlined in the Approaches Report. In the first proposed approach, dose-response functions were culled from the epidemiological literature (or their underlying dataset) and incremental changes in neurodevelopmental endpoints were calculated based on a given change in thyroid hormone concentration as a result of perchlorate exposure.

In the peer-reviewed report, the peer reviewers summarized the strengths and limitations of this first approach (referred to by the panel as the "two-stage" approach) as such:

The panel agreed that strengths of the two-stage approach include explicit linkage between perchlorate exposure and a recognized adverse effect. Although the manner in which they are linked is highly technical, it is a concept that can be readily understood, facilitating risk communication related to the MCLG. The principal limitation of this approach is the small number of studies able to contribute data to the modeling and uncertainties associated with various modeling steps (External Peer Reviewers for U.S. EPA, 2018, p. 8).

The second proposed approach did not directly connect fT4 to neurodevelopment but instead used a distribution of fT4 to define a cut point at which hypothyroxinema occurs, and subsequently estimated the dose of perchlorate associated with a critical increase in the proportion of women predicted to be hypothyroxinemic due to perchlorate exposure. This approach was proposed based on the epidemiological evidence that early pregnancy hypothyroxinemia is a risk factor for a variety of adverse neurodevelopmental outcomes. In reviewing this approach, the peer reviewers noted the following strengths in their report:

1) The central premise, that hypothyroxinemia is associated with adverse neurodevelopmental effects is supported by a large number of studies, including categorical studies; 2) This approach encompasses a variety of adverse neurodevelopmental outcomes, as indicated by these studies, rather than focusing on one or a limited number of adverse outcomes, as with the two-stage approach; and 3) This approach avoids all of the uncertainties associated with determining a quantitative relationship between a specific maternal fT4 level and the magnitude an adverse neurodevelopmental effect (External Peer Reviewers for U.S. EPA, 2018, p. 7).

The peer reviewers expressed concern about hypothyroxinemia being a precursor effect, rather than an adverse health outcome, which they argued may create difficulties in explaining the basis for the MCLG to some audiences. However, the EPA has used precursor effects as the basis for setting regulatory and non-regulatory limits previously. The peer review panel also expressed concern that a standard definition of hypothyroxinemia has not yet been established, as clinicians use varying fT4 thresholds to define their own working definition of the condition. This also could lead to difficulties communicating the population at risk for developing this precursor effect as a result of perchlorate exposure.

Ultimately, the EPA chose to develop the MCLG using dose-response functions from the epidemiological literature to estimate neurodevelopmental impacts in the offspring of pregnant women exposed to perchlorate. EPA selected this proposed approach because it is consistent with SDWA's definition of an MCLG to avoid adverse health effects and because it is most consistent with the SAB recommendations.

# 9.2.2 Selection of the Dose-Response Function

As summarized in the Approaches Report, to identify studies that connected incremental changes in maternal T4 or fT4 to incremental changes in offspring neurodevelopment, the EPA assessed 71 epidemiological studies using a 4-step approach. First, studies were identified that were not compatible with BBDR model output, and these studies were not further evaluated. Next, studies were categorized depending on whether the analysis was based on a categorical or a continuous measure of thyroid hormones as they related to neurodevelopment. Because studies with continuous measures are the most useful for developing dose-response functions, they were selected over categorical studies. The studies with continuous measures of thyroid hormones that were deemed to have the potential to inform a dose-response analysis were then evaluated using the National Toxicology Program's Office of Health Assessment and Translation (OHAT) Risk of Bias (ROB) tool and scored. Studies that were identified as having a "definitely high risk of bias" by the ROB tool were removed from consideration. The remaining papers were evaluated to determine the feasibility of using data from the paper in a dose-response analysis.

Ultimately, the EPA focused on five studies, four related to cognition [ ADDIN EN.CITE ADDIN EN.CITE ADDIN EN.CITE.DATA ] and one related to behavior [ ADDIN EN.CITE <a href="mailto:kendones-endendijk-and-noise-endendijk-and-n

J.J.</author><author>Wijnen, H.A.</author><author>Pop, V.J.</author><author>van Baar, A.L.</author></author></contributors><titles><title>Maternal thyroid hormone trajectories during pregnancy and child behavioral problems</title><secondary-title>Hormones and Behavior</secondary-title></full-title>Hormones and Behavior</full-title></periodical><pages>84-

92</pages><volume>94</volume><dates><year>2017</year></dates><urls></record></Cite ></EndNote>], which could be used to quantitatively describe how incremental changes in thyroid hormone levels in early pregnancy result in subsequent changes in neurodevelopment.<sup>22</sup> For each study, an analysis was conducted to evaluate the dose-response relationship between maternal thyroid hormone levels (specifically fT4) and offspring neurodevelopment. Effect estimates were either presented in the paper, or derived through the digitization of figures<sup>23</sup> or through novel analysis of data provided by the study authors. Each effect estimate was presented or estimated along with a 95 percent confidence interval (CI). Consistent with the EPA risk assessment policies,<sup>24</sup> the EPA utilized the upper effect estimate from each study to determine the dose of perchlorate associated with a given change in fT4.

Results showing the perchlorate doses associated with a 1 percent and a 2 percent change in the endpoint specific to each study of interest from its mean using the upper beta estimate, assuming the 10th percentile of baseline fT4, are presented in [ REF \_Ref516151019 \h \\* MERGEFORMAT ]. These results provide a perspective on the potential impacts of perchlorate on maternal fT4 (as predicted by the BBDR model) and subsequent neurodevelopmental impacts (as predicted by the epidemiologic literature).

<sup>&</sup>lt;sup>22</sup> Detailed summaries of each study, as well as information about the EPA's independent reanalysis of the Korevaar et al. (2016) data, are presented in Chapters 5 and 6 of the Approaches Report.

<sup>&</sup>lt;sup>23</sup> For any publication for which this was done EPA attempted to contact the study authors for underlying data. When these attempts were unsuccessful, the EPA opted to use the WebPlotDigitizer Extension for Google Chrome to extract the data points from the plot, along with the linear regression function in Excel to estimate a regression on the extracted data. The EPA aimed to replicate the reported R<sup>2</sup> values reported in the studies in order to validate the use of the WebPlotDigitizer method as a means of accurately extracting the data from the study, and were successful in doing so.

<sup>&</sup>lt;sup>24</sup> The Benchmark Dose Technical Guidance document recommends the use of the 95 percent lower bound on a benchmark dose as the POD for noncancer effects, as using the lower bound accounts for the experimental variability inherent in a given study and assures (with 95 percent confidence for the experimental context) that the selected benchmark response is not exceeded (U.S. EPA, 2012).

# Table [ SEQ Table \\* ARABIC ]. Predicted Dose of Perchlorate per 1 Percent to 2 Percent Change<sup>a</sup> in Neurodevelopmental Measure for Low-lodine Intake Individuals Based on Upper Effect Estimates at the 10th Percentile fT4 Level<sup>b, c</sup>

Study	Endpoint	Dose-Response Function	β (95% CI)	ΔfT4 in pmol/L Associated with a 1% to 2% Change in Endpoint (% ΔfT4 from 0 dose perchlorate, iodine intake = 75 μg/day) <sup>a, d</sup>	
Korevaar et al. (2016) Quadratic	IQ	$\Delta IQ = (\beta_1 \times \ln f T 4_2 + \beta_2 \times \ln (f T 4_2)^2) - (\beta_1 \times \ln f T 4_1 + \beta_2 \times \ln (f T 4_1)^2)$	$ \beta_1 = 33.81 $ $ (9.8, 57.82) $ $ \beta_2 = -6.235 $ $ (-10.567, -1.903) $	-0.13 to -0.25 (1.9% to 3.8%)	1.9 to 3.9
Korevaar et al. (2016) EPA independent analysis	IQ	$\Delta IQ = (\beta_1 \times \ln(fT4_2)) - (\beta_1 \times \ln(fT4_1))$	17.26 (3.77, 30.75)	-0.21 to -0.41 (3.1% to 6.2%)	3.1 to 6.7
Pop et al. (2003)	MDI	$\Delta MDI = \beta \times \Delta fT4$	6.3 (1.92, 10.6)	-0.09 to -0.19 (1.0% to 2.8%)	1.3 to 2.8
Pop et al. (2003)	PDI	$\Delta PDI = \beta \times \Delta fT4$	8.4 (4.0, 12.8)	-0.08 to -0.16 (0.9% to 2.4%)	1.1 to 2.3
Pop et al. (1999)	PDI	$\Delta PDI = \beta \times \Delta fT4$	8.5 (0.01, 17.0)	-0.06 to -0.12 (0.6% to 1.8%)	0.8 to 1.7
Endendijk et al. (2017)	Anxiety/ depression score	$\Delta AD = \left(\frac{1}{\beta * fT4_2}\right) - \left(\frac{1}{\beta * fT4_1}\right)$	0.12 (0.11, 0.13)	-0.03 to -0.08 (0.45% to 1.2%)	0.4 to 1.1
Finken et al. (2013)	SD of reaction time	ΔSD Reaction Time (ms) = β × Δ fT4	-4.9 (-9.5, -0.2)	-0.28 to -0.57 (4.2% to 8.5%)	4.4 to 9.8

<sup>&</sup>lt;sup>a</sup> The analyses for IQ, Mental Development Index (MDI), and Psychomotor Development Index (PDI) are based on a 1% or 2% change from the standardized mean for each test, which in all instances is 100 points. The analysis for the anxiety/depression score is based on the mean score in the population (reported by sex) assessed in Endendijk et al. (2017). For a population of 51% boys, the mean anxiety/depression score to be equal to 0.51\*1.18 to 0.49\*1.11 = 1.1 and calculated a 1% change (i.e., 0.01 points) and a 2% change (i.e., 0.02 points) from this value. According to communication with Endendijk et al. (2017), there is not a standardized mean for the raw scores in the Child Behavior Checklist (CBCL). The analysis for the SD of reaction time is based on the mean SD of reaction time presented in the Finken et al. (2013) paper, which was 270 ms, so a 1% change would be 2.7 ms and a 2% change would be 5.4 ms. The EPA reached out to the Finken et al. authors to confirm the standardized mean for SD of reaction time but have not received a response.

<sup>&</sup>lt;sup>b</sup> This is based on the regression analysis for the range of fT4 data within each study using the upper beta estimates from the 95% CI. These results are for the low-iodide intake population of 75 μg/day. In all functions, fT4 is in units of pmol/L.

c The BBDR model with a pTSH of 0.398 was used for these analyses.

d Ranges show results from a 1% change in the endpoint from its mean value to a 2% change in the endpoint from its mean

From the seven analyses presented, the EPA chose the independent analysis of the Korevaar et al. (2016) data as the basis for calculating the POD. The reason for this selection is that a health impact function can be derived for the sensitive population of interest, adjusting for the appropriate set of confounders, with a readily interpretable endpoint.

The five identified papers evaluated a variety of endpoints with Korevaar et al. (2016) evaluating IQ, Pop et al. (1999, 2003) using the Bayley Scale to evaluate PDI and MDI, Finken et al. (2013) evaluating the SD of reaction time, and Endendijk et al. (2017) evaluating anxiety/depression scores using the CBCL. The SD of reaction time from Finken et al. (2013) was not well-received by the peer reviewers because it is difficult to ascertain the true implications of a change in the SD of reaction time. The Endendijk et al. (2017) study was identified after the peer review so no feedback was given on the appropriateness of the endpoint; however, the anxiety/depression raw score is not an intuitively interpretable endpoint. Further, neither the Endendijk et al. (2017) nor the Finken et al. (2013) analyses had functions for the sensitive life stage (i.e., their analyses were based on the full range of fT4 levels and did not concentrate on the impacts of low-end fT4 levels). For these reasons, the Endendijk et al. (2017) and Finken et al. (2013) papers were not selected for further evaluation.

The Korevaar et al. (2016) original and independent analyses are preferable compared to the Pop et al. (1999, 2003) studies because neither function derived from the Pop et al. studies was adjusted for confounders. Additionally, both Pop et al. papers have an  $N \le 50$  compared to the Korevaar et al. analyses, which have an N of greater than  $3,600.^{25}$ 

Although the original Korevaar et al. (2016) analysis was the most rigorous analysis available in the literature to date, the Korevaar et al. (2016) EPA reanalysis was chosen over the original analysis because it included changes made to the analysis at the suggestion of the peer-review panel. The revised analysis controls for a more appropriate set of confounders, leading to more precise estimates of the association between maternal fT4 and child IQ. The EPA was prompted to revisit the original Korevaar et al. (2016) model because of the feedback received during the peer review of the Approaches Report. Specifically, a member of the peer-review panel expressed the following suggestion:

Korevaar et al. controlled for instrumental variables (e.g. gestational week at fT4 measurement) as well as variables that are consequences of altered fT4 (e.g. maternal BMI), which may have biased estimates. This study also assumed a loglinear relation between fT4 and the outcome but it is unclear whether the data fit this functional form better than a linear form. Reanalysis of the data performed by EPA should not include the variables noted above, which may have driven measures of association towards the null, and should investigate the most appropriate functional form to inform decisions about transformation of fT4 values (External Peer Reviewers for U.S. EPA, 2018, pp. 61–62).

<sup>&</sup>lt;sup>25</sup> The original Korevaar et al. (2016) analysis included 3,839 mother/child pairs. The EPA reanalysis of the Korevaar et al. (2016) data had a slightly lower N of 3,609, due to the exclusion of subjects with imputed values for maternal fT4.

The EPA responded to this suggestion by developing a causal model for the effect of maternal fT4 on child IQ to identify the minimum set of confounding variables, testing the proper functional form of the relationship between maternal fT4 and child IQ in the Korevaar et al. (2016) data, and making decisions about data quality and influential data points in the analysis. <sup>26</sup> That is, the EPA determined that there were values of the independent variable of interest, fT4, in the original analysis that were imputed using multiple imputations. This could have impacted the effect estimate of the independent variable of interest with data that were not directly measured. The EPA reanalysis excludes these nonmeasured values as a precaution. Subsequently, the EPA selected the Korevaar et al. (2016) reanalysis as the most appropriate function from which to assess the relationship between fT4 and IQ.

<sup>&</sup>lt;sup>26</sup> The details of which are described in Section 6.3.2 of the Approaches Report.

## 10. Select Benchmark Response and Identify the POD

In conducting this analysis, the EPA seeks to identify an MCLG that is at a level at which no adverse neurodevelopmental impacts occur in the offspring of pregnant women who are exposed and that provides an adequate margin of safety that protect other potentially sensitive life stages from adverse health effects of perchlorate exposure. This is difficult when evaluating a continuous outcome such as IQ. Therefore, to select a dose to inform a POD, the percent change in IQ that results with no known or anticipated adverse effects needs to be defined. The EPA often uses percentage changes in noncancer risk assessment, including 5 percent or 10 percent, or a 0.5 to 1 SD change from the mean [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2012</Year><RecNum>2000</RecNum><DisplayText>(U.S. EPA, 2012)</DisplayText><record><rec-number>2000</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1529502864">2000</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>U.S. EPA,</author></authors></contributors><title>Benchmark dose technical guidance</title></title></dates><year>2012</year></dates><isbn>EPA/100/R-12/001</isbn><urls></urls></record></Cite></EndNote>]. However, these are often used when assessing toxicological data. A smaller response to inform a POD (e.g. 1%) is justified when using epidemiological literature [ ADDIN EN.CITE < EndNote > < Cite > < Author > U.S. EPA</Author><Year>2012</Year><RecNum>2000</RecNum><DisplayText>(U.S. EPA, 2012)</DisplayText><record><rec-number>2000</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1529502864">2000</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>U.S. EPA,</author></authors></contributors><title>Benchmark dose technical guidance</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title> 12/001</isbn><urls></urls></record></Cite></EndNote>]. Given that IQ is standardized to have a mean of 100, a 1 point change would be equivalent to a 1 percent change from the standardized mean IQ.

CalEPA concluded in Carlisle et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Carlisle</Author><Year>2009</Year><RecNum>59</RecNum><Displ ayText>(2009)</DisplayText><record><rec-number>59</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437145077">59</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Carlisle, J C</author><author>Dowling, K C</author><author>Siegel, D M</author><author>Alexeff, G V</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></a></author></a></al>
Author
</ap>
Author

1208</pages><volume>44</volume><number>12</number><section>1200</section><dates><year>>2009</year></dates><urls></record></Cite></EndNote>] that one IQ point should be considered the *de minimis* change that is appropriate for regulatory action. Specifically, Carlisle et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Carlisle</Author><Year>2009</Year><RecNum>59</RecNum><DisplayText>(2009)</DisplayText><record><rec-number>59</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1437145077">59</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Carlisle, J C</author><author>Dowling, K C</author>Siegel, D M</author><author>Alexeff, G

V</author></contributors></title>A blood lead benchmark for assessing risks from childhood lead exposure</title><secondary-title>Journal of Environmental Science and Health</secondary-title></title></periodical><full-title>Journal of Environmental Science and Health</full-title></periodical><pages>1200-

1208</pages><volume>44</volume><number>12</number><section>1200</section><dates><year>>2009</year></dates><urls></record></Cite></EndNote>] argued that "while a one-point change in WISC full-scale IQ is within the standard error of an individual measurement and would not be regarded as clinical disease or cause affected individuals to seek medical care, it is still highly significant on a population basis, since a small difference...is associated with large differences in the number of children in the two tails of the IQ distribution."

Additionally, the EPA used a small change in child IQ based on epidemiological literature when considering the epidemiological evidence that linked lead exposure to IQ. Specifically, when developing the National Ambient Air Quality Standards (NAAQS) for lead, the EPA stated that "the Administrator proposed to conclude that an air-related population mean IQ loss within the range of 1 to 2 points could be significant from a public health perspective, and that a standard level should be selected to provide protection from air-related population mean IQ loss in excess of this range" [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2008</Year><RecNum>105</RecNum><Pages>6698</Pages><DisplayText>(U.S. EPA, 2008b, p. 6698)</DisplayText><record><rec-number>105</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1444079856">105</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><author>U.S.

EPA,</author></contributors></title>National ambient air quality standards for lead. 73 FR 66964. Pages 66964-67062. (November 12, 2008) (to be codified at 40. C.F.R. 51, 51, 53, and 58).</title></title></title></tole>
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For the specific context of perchlorate, the EPA is exploring avoiding both a 1 percent change and a 2 percent change in IQ as the basis for an MCLG. The corresponding Benchmark Response percent decrease in fT4 from the zero perchlorate dose baseline are 3.1% for a 1 IQ point decrease and 6.2% for a 2 IQ point decrease ([ REF \_Ref516151019 \h ]). Combining these response rates with the results from the reanalysis of Korevaar et al. (2016) result in a POD dose of 3.1  $\mu$ g/kg/day for a 1 percent change in IQ, and a POD dose of 6.7  $\mu$ g/kg/day for a 2 percent change in IQ. The PODs associated with a 1 percent and a 2 percent change in IQ are likely protective of a large portion of the population not experiencing this IQ loss, given that it is based on the low end of the distribution of fT4 for a low-iodine intake population with a muted TSH feedback response. However, uncertainties in the analysis remain, as discussed in the next section.

#### 11. Translate POD to a RfD

To use the estimated POD to estimate an RfD, the application of uncertainty factors (UFs) needs to be evaluated. This section will explore each of the EPA's UFs and provide a brief overview of the process to derive a RfD.

#### 11.1 Consideration of Uncertainty Factors

Section 4.4.5 of *A Review of the RfD & RfC Processes* [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2002</Year><RecNum>1996</RecNum><DisplayText>(U.S. EPA, 2002)</DisplayText><record><rec-number>1996</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1527620504">1996</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><author>U.S. EPA,</author></authors></contributors><titles><title>A review of the reference dose and reference concentration process</title></title></dates><qaver>2002
year>2002
year></dates><isbn>EPA/630/P-02/002F</isbn><urls></record></Cite></EndNote>], presents the following uncertainty factors for deriving RfDs: inter-individual variability, interspecies uncertainty, extrapolating from subchronic to chronic exposure, extrapolating from a lowest-observed adverse effect level (LOAEL) rather than from a NOAEL, and an incomplete database. The EPA has considered each of these factors in deriving an RfD to inform an MCLG for perchlorate.

(1) Variation in sensitivity among the members of the human population (i.e., inter-individual variability) (uncertainty factor, within-human variability, UF<sub>H</sub>). Section 4.4.5.3 (p 4-42) of the 2002 document states, "In general, the Technical Panel reaffirms the importance of this UF, recommending that reduction of the intraspecies UF from a default of 10 be considered only if data are sufficiently representative of the exposure/dose-response data for the most susceptible subpopulation(s). ... Similar to the interspecies UF, the intraspecies UF can be considered to consist of a both a toxicokinetic and toxicodynamic portion (i.e. 10<sup>0.5</sup> each)" [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2002</Year><RecNum>1996</RecNum><DisplayText>(U.S. EPA, 2002)</DisplayText><record><rec-number>1996</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1527620504">1996</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author> U.S.

EPA,</author></authors></contributors></title>A review of the reference dose and reference concentration

process</title></title></title></dates></grave>2002</year></dates></sbn>EPA/630/P-02/002F</isbn><urls></urls></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes/likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></fraction-likes></

a. *Applicable*. The inter-individual variation in the NIS structure and/or regulation contributing to variance across the population in the impact of perchlorate exposure on thyroid function is currently unknown. Although the BBDR model does its best to model the relationship between perchlorate and fT4 in the identified sensitive population of first trimester fetuses, there is still uncertainty due to the lack of data, as explained in the Approaches Report [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2018</Year><RecNum>1995</RecNum><DisplayText>(U.S. EPA, 2018)</DisplayText><record><rec-number>1995</rec-number><foreign-keys><key

app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1527619441">1995</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>U.S. EPA,</author></authors></contributors><titles><title>Proposed approaches to inform the derivation of a maximum contaminant level goal for perchlorate in drinking water</title></title></title></dite></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

"On the pharmacokinetic side of the model.... there are very few calibration data for perchlorate kinetics in humans, particularly at the life stage of interest. This may be of particular importance given that available biomarker data to relate perchlorate external dose to internal serum concentration are highly variable and uncertain. For example, the Lumen et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Fisher</Author><Year>2013</Year><RecNum>249</ RecNum><DisplayText>(2013)</DisplayText><record><rec-number>249</recnumber><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468416440">249</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Fisher, J</author><author>Li, S.</author><author>Crofton, K.</author><author>Zoeller, R. T.</author><author>McLanahan, E. D.</author><author>Lumen, A.</author><author>Gilbert, M. E.</author></authors></contributors><authaddress>U.S. Food and Drug Administration, National Center for Toxicological Research, Jefferson, AR, USA. jeffrey.fisher@fda.hhs.gov</authaddress><title>Evaluation of iodide deficiency in the lactating rat and pup using a biologically based dose-response model</title><secondary-title>Toxicol Sci</secondary-title></title><periodical><full-title>Toxicol Sci</fulltitle></periodical><pages>75-86</pages><volume>132</volume><number>1</number><edition>2013/01/05</ed ition><keyword><keyword>Animals</keyword><keyword>Calibration</keyword ><keyword>Deficiency Diseases/ physiopathology</keyword><keyword>Dose-Response Relationship, Drug</keyword><keyword>Hypothalamo-Hypophyseal System</keyword><keyword>Iodides/ metabolism</keyword><keyword>Lactation</keyword><keyword>Models, Biological</keyword><keyword>Rats</keyword><keyword>Thyroid Gland/physiopathology</keyword><keyword>Thyroid Hormones/blood</keyword></keywords><dates><year>2013</year><pubdates><date>Mar</date></pub-dates></dates><isbn>1096-0929 (Electronic)
1096-0929 (Linking)</isbn><accessionnum>23288054</accession-num><urls></urls><electronic-resourcenum>10.1093/toxsci/kfs336</electronic-resource-num><remote-databaseprovider>NLM</remote-databaseprovider><language>eng</language></record></Cite></EndNote>] model was based on pregnant women living in a Chilean city with high levels of perchlorate in

standard deviation of the concentration of perchlorate in urine was similar to, or in one instance, greater than the mean urinary perchlorate concentration, indicating

[PAGE \\* MERGEFORMAT]

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drinking water [ ADDIN EN.CITE

extreme variability. Furthermore, data used in the calibration from the Chilean city in Telléz Telléz et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA | did not reflect a low-iodine intake population. Urinary iodide for the 37 subjects at the second prenatal visit was, on average, 217 μg/L, with a standard deviation of 109 μg/L" (p. 3-16).

In addition, "toxicodynamic aspects such as competitive inhibition at the NIS, depletion of iodide stores under different iodine intake levels and physiological states, and the ability of the TSH feedback loop to compensate for perturbations in thyroid function each have their own uncertain features. ... The degree to which there is inter-individual variation in NIS structure and/or regulation contributing to variance across the population in the impact of perchlorate exposure on thyroid function is currently unknown. ... Given the limited data available for model evaluation, perchlorate toxicokinetics and toxicodynamics in pregnancy remain a source of uncertainty. This could impact the resulting fT4 predictions of neurodevelopmental effects" (p. 3-18).

There are also uncertainties in linking maternal fT4 levels to an offspring's IQ. These uncertainties include the population for which dose-response information is available (i.e., no study is U.S.-based), a lack of study information on the iodine intake status for the population for which the dose-response information is available, uncertainties around the methods used to assess maternal fT4 measurement during pregnancy, and uncertainties related to the true distribution of fT4 for a given iodine intake.<sup>27</sup>

Although human data were used to calibrate the BBDR model, these data are not specific to the most sensitive individuals (i.e., the fetuses of hypothyroxinemic pregnant women). Further, as outlined in Section 2, the EPA believes that protecting the fetus of a hypothyroxinemic woman will protect other identified sensitive life stages. However, there is some uncertainty due to the lack of information linking incremental changes in infant thyroid hormone levels to adverse neurodevelopmental outcomes.

Subsequently, the EPA has determined that it is appropriate to employ an uncertainty factor of 3. The value of 3 was selected instead of the full 10 because the EPA has specifically attempted to model the most sensitive individuals (i.e., muted TSH feedback, low fT4 values, low-iodine intake). However, as outlined above, there remains important pieces of data which are still uncertain. Therefore, a UF of 3 is necessary to ensure protection of the most vulnerable populations with an adequate margin of safety and to account for differences at other life stages (e.g., bottle-fed vs. breast-fed infants).

- (2) Uncertainty in extrapolating animal data to humans (i.e., interspecies uncertainty) (uncertainty factor, animal-to-human, UF<sub>A</sub>)
  - a. Not applicable. Animal studies were not used to develop the BBDR model nor were they used to relate alternations in maternal fT4 to IQ. Therefore, the EPA assigns this UF a value of 1.

<sup>&</sup>lt;sup>27</sup> For more details, see Section 6.5 in the Approaches Report.

- (3) Uncertainty in extrapolating data obtained in a study with less-than-lifetime exposure to lifetime exposure (i.e., extrapolating from subchronic to chronic exposure) (uncertainty factor, subchronic to subchronic, UF<sub>s</sub>)
  - a. *Not applicable*. Extrapolating from subchronic to chronic exposures did not occur given that the BBDR model was "designed to assess long-term steady-state conditions in the non-pregnant woman and week-to-week variation in pregnancy, rather than short-term (hour-to-hour or day-to-day) fluctuations" (U.S. EPA, 2018, p. 3-3). Therefore, the EPA assigns this UF a value of 1.
- (4) Uncertainty in extrapolating from a LOAEL rather than from a NOAEL (uncertainty factor, LOAEL-to-NOAEL, UF<sub>L</sub>)
  - a. *Not applicable*. LOAELs and NOAELs were not identified or used in this approach. Therefore, the EPA assigns this UF a value of 1.
- (5) A database UF is applied, when needed, to address the potential for deriving an inadequately protective RfD in the instance where the available database is incomplete as a result of an incomplete characterization of the chemical's toxicity (uncertainty factor, database deficiency, UF<sub>D</sub>) US EPA, 2002).
  - a. Not applicable. As described in Section 2.4 of the SAB stated, "the mode of action of perchlorate toxicity is well understood" (SAB, 2013, p. 2) Based on the findings of a 2005 NAS report, the EPA has previously concluded that a UF $_{\rm D}$  was not needed for deficiencies in the perchlorate database (NRC, 2005; US EPA, 2005) . Therefore, the EPA assigns a UF $_{\rm D}$  value of 1.

#### 11.2 RfD

Using the POD of 3.1  $\mu$ g/kg/day based on a 1 percent change in the standardized mean IQ from the EPA's independent analysis of the Korevaar et al. (2016) data (Section [ REF \_Ref516142958 \r \h ]), the EPA can derive an RfD by incorporating the UF<sub>H</sub>. This results in the following:

$$RfD = \frac{POD}{UF_H} = \frac{3.1}{3} = 1.03 \frac{\mu g/kg}{day}$$

Using the POD of 6.7  $\mu$ g/kg/day based on a 2 percent change in the standardized mean IQ from the EPA's independent analysis of the Korevaar et al. (2016) data (Section [ REF\_Ref516142958 \r \h ]), the EPA can derive a second RfD by incorporating the UF<sub>H</sub>, which results in the following:

$$RfD = \frac{POD}{UF_{H}} = \frac{6.7}{3} = 2.23 \frac{\mu g/kg}{day}$$

# 12. Translate RfD to a Drinking Water Equivalent Level and Identify the MCLG

To translate the RfD ( $\mu$ g/kg/day) to an MCLG in drinking water ( $\mu$ g/L), the EPA will use the following equation:

$$W\left(\frac{\mu g}{L}\right) = \frac{RfD}{DWI} \times RSC_w$$

where:

W = drinking water concentration of perchlorate;

RfD = reference dose (1.03 µg/kg/day for a 1 percent change or 2.23 µg/kg/day for a 2 percent change in IQ);

DWI = bodyweight-adjusted drinking water ingestion rate (L/kg/day); and

 $RSC_w$  = relative source contribution.

This section walks the reader through each input in the above equation (with the exception of the RfD derivation, which was described previously) and presents a justification for its use in estimating an MCLG for perchlorate in drinking water.

#### 12.1 Drinking Water Intake

Given the above equation for translating the RfD to a concentration, it is necessary to determine the most appropriate bodyweight-adjusted drinking water ingestion rate for the target population of first-trimester pregnant women. The EPA Exposure Factors Handbook (EFH) [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2011</Year><RecNum>1740</RecNum><DisplayText>(U.S. EPA, 2011b)</br/>DisplayText><record><rec-number>1740</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206447">1740</key></foreignkeys><ref-type name="Report">27</ref-type><contributors><authors><author>U.S. EPA,</author></authors></contributors><title>Exposure factors handbook: 2011 edition (final)</title></title></title></dates></exar>2011</year></dates></publication>Washington, DC</publication> location><publisher>U.S. Environmental Protection Agency, Office of Research and Development, National Center for Environmental Assessment</publisher><isbn>EPA/600/R-090/052F</isbn><label>786546</label><work-type>EPA Report</work-type><urls><relatedurls><url>http://cfpub.epa.gov/ncea/cfm/recordisplay.cfm?deid=236252</url></relatedurls></urls><language>English</language><modified-date>U.S. Environmental Protection Agency</modified-date></record></Cite></EndNote>] reports mean, 90th, and 95th percentile bodyweight-adjusted drinking water intakes for pregnant, lactating, and non-pregnant non-lactating women of childbearing age from the [HYPERLINK \1"\_ENREF\_62" \0 "Kahn, 2008 #2002"] study based on the Continuing Survey of Food Intakes by Individuals (CSFII) data from 1994 to 1996 and 1998. This study estimated bodyweight-adjusted drinking water intake rates for direct and indirect community water ingestion, as well as for direct and indirect water intake from all sources, on both a per-capita and consumers-only basis. As the proposed MCLG is specific for the offspring of pregnant women consuming community drinking water, the EPA chose to focus on community drinking water

intake estimates on a consumers-only basis as potential inputs. These estimates are reported in Table 3-81 in EFH [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2011</Year><RecNum>1740</RecNum><DisplayText>(U.S. EPA, 2011b)</DisplayText><record><rec-number>1740</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206447">1740</key></foreign-keys><ref-type name="Report">27</ref-type><contributors><authors><authors><author>U.S. EPA,</author></author></authors></contributors><titles><title>Exposure factors handbook: 2011 edition (final)</title></title></dates><year>2011</year></dates><pub-location>Washington, DC</pub-location><publisher>U.S. Environmental Protection Agency, Office of Research and Development, National Center for Environmental Assessment</publisher><isbn>EPA/600/R-090/052F</isbn><label>786546</label><work-type>EPA Report</work-type><urls><related-urls><url>http://cfpub.epa.gov/ncea/cfm/recordisplay.cfm?deid=236252</url></re></re></rr>Vurls><language>English</language><modified-date>U.S. Environmental Protection Agency</modified-date></re></rr>Agency</modified-date></record></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite></cite><

Table [ SEQ Table \\* ARABIC ]. Consumers-Only Estimated Direct and Indirect Community Water Ingestion Rates from Kahn and Stralka (2008) (L/kg/day)

Women Categories	Sample Size	Mean	90th Percentile	95th
Pregnant	65	0.014ª	0.033ª	
Lactating	33	0.026a	0.054ª	
Non-pregnant, non-lactating, 15 to 44 years of age	2.028	0.015	0.032	

<sup>&</sup>lt;sup>a</sup> The sample size does not meet minimum reporting requirements to make statistically reliable estimates as described in the *Third Report on Nutrition Monitorins* 1994-1996 [ ADDIN EN.CITE

<EndNote><Cite><Author>FASEB/LSRO</Author><Year>1995</Year><RecNum>2003</RecNum><DisplayText>1995)</DisplayText><record><rec-number>2003</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1529955869">2003</key></foreign-keys><ref-type</td>

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1529955869">2003</key></foreign-keys><ref-type name="Report">27</ref-

type><contributors><author>FASEB/LSRO</author></authors></contributors><title>Third report monitoring in the United States</title></title></title></column=1</column=1</td>

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size="100%">https://www.cdc.gov/nchs/data/misc/nutri95\_lacc.pdf</style></url></related-urls></urls></record></Cite></EndNote>].

The EPA chose to use exposure factor estimates specific to women of childbearing age in deriving an MCLG (i.e., non-pregnant, non-lactating, 15–44 years of age). This determination was reached as the analysis in the development of an RSC was performed using a population of women of childbearing age from NHANES (20 – 44 years of age). The EPA acknowledges there is a difference in the age group defining women of child bearing age used to develop the drinking water intake rate compared to what is used to develop the relative source contribution, which was based on the age group used in developing the BBDR model. However, the EFH reports this drinking water intake to be applicable for women of childbearing age, and the age range of the BBDR model fits within the age range used to estimate this recommended drinking water intake. Subsequently the Agency believes the difference in the age groups will have minimal impact on the resulting MCLG analysis. The EPA has chosen to apply the 90th percentile drinking water intake rate, in order to account for variability in drinking

water intakes. Thus, according to Table 3-81 in the EFH [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2011</Year><RecNum>1740</RecNum><DisplayText>(U.S. EPA, 2011b)</DisplayText><record><rec-number>1740</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206447">1740</key></foreign-keys><ref-type name="Report">27</ref-type><contributors><authors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><autho

#### 12.2 Relative Source Contribution

The EPA believes that dietary ingestion is the only significant pathway for non-drinking-water perchlorate exposure. Thus, EPA estimates the RSC for drinking water, or the percentage of the RfD that is left for drinking water when accounting for all other routes of exposure, from the following equation:

$$RSC = \frac{RfD - Food}{RfD} \times 100\%$$

In order to estimate the RSC from this equation, it was necessary to produce an estimate for the dose of perchlorate coming from food in women of childbearing age. To estimate this figure the EPA mimic the approach taken by Abt et al. (2016; See Section [ REF\_Ref530562089 \r \h ]). However, for the purposes of estimating perchlorate dose for informing the MCLG, the EPA also wanted to obtain distributional information on perchlorate in food, such as the 95th percentile intake rate for women of childbearing age. Additionally, the EPA examined the same age group that was used in the BBDR model for other parameters, to match the perchlorate doses from the BBDR model. The EPA produced these estimates by following the methods for estimating perchlorate consumption that were published in Abt et al. (2016), and subsequently summarized distributional information for perchlorate in food, specifically for the same age group used in the BBDR model for other parameters (women 20 to 44 years of age).

To begin EPA combined data from the following sources:

• The NHANES is run by the Centers for Disease Control and Prevention's National Center for Health Statistics and was designed to collect information on the health and nutritional status of the U.S. civilian, non-institutionalized population through in-home interviews and physical examinations [ ADDIN EN.CITE <EndNote><Cite><Author>CDC</Author><Year>2007-2008, 2009-2010, 2011-2012</Year><RecNum>2005</RecNum><DisplayText>(CDC & DisplayText><CDC & DisplayText><COC &

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tributors><title>National health and nutrition examination survey data</title></title></title></dates><year>2007-2008, 2009-2010, 2011-2012</year></dates><urls><related-urls><url>http://www.cdc.gov/nchs/nhanes.htm</url></related-urls></urls></record></Cite></EndNote>].
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• What We Eat in America (WWEIA) provides 24-hour food diary data that are collected as part of NHANES and serves as the dietary intake measurement component of the survey. Dietary data are collected for up to two days for each respondent [ ADDIN EN.CITE <EndNote><Cite><Author>CDC</Author><Year>2007-2008, 2009-2010, 2011-2012</Year><RecNum>2005</RecNum><DisplayText>(CDC & Description of the survey). With the component of the survey. We have a possible of the survey of the survey. Dietary data are collected for up to two days for each respondent [ ADDIN EN.CITE <EndNote><Cite><Author>CDC</Author>CDC & Dietary (CDC & Dietary (CDC

The Food and Drug Administration's (FDA's) Total Diet Study (TDS) is an ongoing FDA program that collects information on levels of various contaminants, including perchlorate, which occur in food and beverages commonly consumed by the U.S. population [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

FDA</Author><Year>2015</Year><RecNum>2006</RecNum><DisplayText>(U.S. FDA, 2015)</DisplayText><record><rec-number>2006</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmedxc5vxfpkax2vzp0ftv29"

timestamp="1530038884">2006</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><author>U.S.

FDA,</author></contributors></title>Total diet study — study design</title></title></title></title></title></title></title></title>

urls><url>http://www.fda.gov/food/foodscienceresearch/totaldietstudy/ucm184232.htm#</url></re>related-urls></urls></record></EndNote>]. To estimate the levels of contaminants in these food products, the FDA buys these foods as a consumer would, prepares them as directed, and then analyzes the prepared foods for levels of the contaminants of interest. This process yields nationally representative estimates of contaminant levels in more than 280 kinds of food and beverages.

The analysis plan followed by the EPA to estimate food consumption estimates can be divided into eight main steps:

- 1. Aggregate TDS data cycles from 2008 to 2012 to maximize the samples of perchlorate contaminants in food.
- 2. Account for values of perchlorate that fall below the level of detection (LOD).

<sup>&</sup>lt;sup>28</sup> The FDA prepares their samples using deionized water.

- 3. Aggregate WWEIA data cycles from 2007 to 2012 to maximize the number of women of childbearing age with consumption data.
- 4. Match the Xue, Zartarian, Wang, Liu, & Georgopoulos, [ADDIN EN.CITE ADDIN EN.CITE.DATA ] method and the World Health Organization (WHO) method [ADDIN EN.CITE <EndNote><Cite><Author>European Food Safety
  Authority</Author><Year>2010</Year><RecNum>2007</RecNum><DisplayText>(Authority, 2010)</DisplayText><record><rec-number>2007</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1530039022">2007</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>European Food Safety
  Authority</author></authors></contributors><titles><title>Management of left-censored data in dietary exposure assessment of chemical substances</title><secondary-title>EFSA
  Journal</secondary-title></title></periodical><full-title>EFSA Journal
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- 5. Create summary perchlorate consumption information from the WWEIA/TDS data file, based on those women with two days of dietary data.
- 6. Aggregate the NHANES demographic data to correspond with the WWEIA participants.

2017 from FDA. This is the same matching file that was used in Abt et al. (2016).

- 7. Merge the summary perchlorate consumption information with the NHANES demographic variables.
- 8. Estimate the relevant summary statistics.

Multiple years of TDS data on perchlorate in food are available from FDA. The EPA combined perchlorate in food data collected in TDS for the years spanning 2008–2012. With four samples taken for each food for each year of data, the EPA had a total of 20 perchlorate samples for each of the 282 foods in the TDS. To account for non-detects in the TDS data, the EPA used three methods when summarizing the data for each food. The EPA first used a method suggested by WHO to calculate both the lower and upper bound concentrations in the foods by substituting all values below the LOD with zero to estimate the lower bound, and substituting all values below the LOD with the LOD to estimate the upper bound [ ADDIN EN.CITE < EndNote > < Cite > < Author > European Food Safety Authority</Author><Year>2010</Year><RecNum>2007</RecNum><DisplayText>(Authority, 2010)</DisplayText><record><rec-number>2007</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1530039022">2007</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>European Food Safety Authority</author></authors></contributors><title>Management of leftcensored data in dietary exposure assessment of chemical substances</title><secondary-title>EFSA Journal</secondary-title></title><periodical><full-title>EFSA Journal</fulltitle></periodical><pages>1-

96</pages><number>8</number><dates><year>2010</year></dates><urls></urls></record></Cite></EndNote>]. This method was used for the perchlorate analysis conducted in Abt et al. (2016). Additionally, the EPA used an approach based on Xue et al. (2010), which assigns half the LOD to values below the LOD if there is at least one detection among the multiple samples taken of each commodity; otherwise a value of zero is assigned. The WHO lower and upper bounds provide bounding on the summarized perchlorate concentration estimates calculated from the TDS data; and

results calculated with other substitution methods to account for the non-detects, including the Xue LOD method, would fall within these bounds.

The EPA calculated both the mean and the 95th percentile of the perchlorate levels in each food based on the TDS data. In order to estimate the 95th percentile from the 20 samples of each food in the TDS data, the EPA used the second-highest test result in each food to represent the 95th percentile concentration. While simple, this method avoids the need to assume a distributional shape for the samples, and has been used in recent publications of TDS data for iodine [ ADDIN EN.CITE <EndNote><Cite><Author>Carriquiry</Author><Year>2016</Year><RecNum>2008</RecNum>< DisplayText>(Carriquiry et al., 2016)</DisplayText><record><rec-number>2008</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1530039524">2008</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Carriquiry, A. L. </author><author>Spungen, J. H.</author><author>Murphy, S. P.</author><author>Pehrsson, P. R.</author><author>Dwyer, J. T.</author><author>Juan, W.</author><author>Wirtz, M. S. </author></authors></contributors><title>Variation in the iodine concentrations of foods: considerations for dietary assessment</title><secondary-title>The American Journal of Clinical Nutrition</secondary-title></titles><periodical><full-title>The American Journal of Clinical Nutrition</full-title></periodical><pages>877S-887S</pages><number>104(Suppl 3)</number><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>].

In order to understand food consumption patterns in women of child-bearing age, the EPA combined WWEIA food diary data from 2007 to 2012 into one large dataset to maximize the number of women. These food diary data include each individual food that was consumed during each 24-hour study period, as well as the amount of each food consumed measured in grams. For a large subset of respondents, food diary data are available for two days. For this analysis, the EPA included only participants with two days of dietary survey data and used the appropriate NHANES survey weights for this segment of the dataset.

The EPA then matched each food consumed with both its mean and its estimated 95th percentile perchlorate contamination amount from the TDS. Foods that are included in the TDS are not characterized and coded the same way as the foods in the WWEIA food diaries. Therefore, it is necessary to use expert judgement to match the TDS foods with the WWEIA foods. The study authors of Abt et al. (2016) provided the EPA with a TDS/WWEIA matching file<sup>29</sup> that is relevant for the years 2003–2012.<sup>30</sup> For this reason, the EPA did not include the most recent WWEIA data (2013–2014) in the analysis as some of the WWEIA food codes have changed since the creation of the mapping file.

Using the Abt et al. (2016) matching file, the EPA matched relevant TDS perchlorate contamination information with each food present in the WWEIA consumption data. The EPA then calculated each participant's daily perchlorate ingestion by estimating the average of the two days of estimated

<sup>&</sup>lt;sup>29</sup> The matching file, "TDSMapping\_2003-2012.xlsx," was emailed to EPA by Judith Spungen of FDA on May 26, 2017.

<sup>&</sup>lt;sup>30</sup> The development of the mapping file relied on some expert judgement by FDA as it relates to matching food groups and consumption data.

perchlorate consumption. The EPA divided each NHANES participant's estimated daily perchlorate consumption estimate by their individual-specific bodyweight to arrive at their individual bodyweight adjusted daily perchlorate consumption. The EPA then calculated perchlorate consumption distributional information from these individual bodyweight adjusted daily perchlorate intakes.

The EPA first replicated the mean food consumption results in Abt et al. (2016), matching the mean perchlorate consumption estimates using WHO lower and upper bounds reported in the paper (using the mean value for each food from TDS). The EPA then restricted the sample to only include women 20 to 44 years of age in order to focus on women of childbearing age. To estimate the food consumption distributional information, the EPA paired both the mean and the 95th percentile estimates of perchlorate in food from the TDS with the food consumption data based on the two-day intake diaries in WWEIA, and adjusted for each individual's bodyweight to produce the estimates in [REF\_Ref517684101\h\_\\*MERGEFORMAT].

Table [ SEQ Table \\* ARABIC ]. Perchlorate Consumption from Food (μg/kg/day) in U.S. Women Aged 20–44 Using the Mean and 95th Percentile TDS Results<sup>a</sup>

Level of Bodyweight Adjusted Perchlorate Consumption from Population Distribution	Perchlorate Dose in Food Based on Mean Concentrations of TDS data (µg/kg/day)	Perchlorate Dose in Food Based on 95th Percentile Concentrations of TDS data (µg/kg/day)
Mean	0.09-0.12	0.23–0.24
50th Percentile	0.08-0.10	0.17–0.19
90th Percentile	0.18–0.21	0.45
95th Percentile	0.23-0.26	0.59-0.60
97.5 Percentile	0.27-0.30	0.80-0.81
99th Percentile	0.33-0.38	1.16–1.17

<sup>&</sup>lt;sup>a</sup> Ranges are due to various approaches for handling values < LOD. If no range is presented, all approaches resulted in the same value.

Given these results, the EPA chose to use the 90th percentile bodyweight-adjusted perchlorate consumption from food using the 95th percentile TDS results to estimate the perchlorate RSC from drinking water. The EPA believes this is the most appropriate value for perchlorate consumption from food to ensure the protection of potentially highly exposed individuals. Given the range of perchlorate concentrations in food, and that food is the only other exposure source being considered in the RSC analysis, the EPA believes it is sufficiently protective to estimate the MCLG for drinking water using the 90th percentile bodyweight-adjusted perchlorate consumption based on the 95th percentile concentrations in TDS. This assures that highly exposed individuals from this most sensitive population are considered in the evaluation of whether perchlorate is found at levels of health concern.

The RSC from drinking water can be calculated as follows by using a 1 percent change in IQ as the basis for the RfD:

$$RSC = \frac{1.03 - 0.45}{1.03} \times 100\% = 56\%,$$

The **bolded** value represents the selected value

or a 2 percent change in IQ as the basis for the RfD:

$$RSC = \frac{2.23 - 0.45}{2.23} \times 100\% = 80\%.$$

Using these RSCs in the standard equation to develop the MCLG, along with the previously used bodyweight-adjusted drinking water intake rate for women of childbearing age of 0.032 L/kg/day, results in the following estimates for the MCLG:

$$W\left(\frac{\mu g}{L}\right) = \frac{RfD\left(\mu g/kg/day\right)}{DWI\left(L/kg/day\right)} \times RSC_w.$$

A 1 percent change in IQ as the basis for the RfD results in:

$$W\left(\frac{\mu g}{L}\right) = \frac{1.03 \,\mu g/kg/day}{0.032 \,L/kg/day} \times 56\% = 18.03 \,\mu g/L.$$

A 2 percent change in IQ as the basis for the RfD results in:

$$W\left(\frac{\mu g}{L}\right) = \frac{2.23 \,\mu g/kg/day}{0.032 \,\text{L/kg/day}} \times 80\% = 55.7 \,\mu g/L.$$

#### 12.3 Evaluation of Uncertainty and Variability in the MCLG

As indicated throughout this document, each input into the analysis to inform the MCLG is a decision point. The EPA has attempted to select the most appropriate inputs to protect the most sensitive population with an adequate margin of safety. Specifically, the EPA evaluated both a 1 percent and a 2 percent change in IQ. The EPA conducted analyses to determine the effect that varying specific inputs will have. The EPA tested the effect of varying the drinking water intake by assuming the drinking water intake rate for pregnant women, or 0.033 L/kg/day. The EPA also evaluated the impact of using the median fT4 estimates to inform the POD as opposed to the 10th percentile. The following subsections will demonstrate some of the uncertainty and variability in the resulting MCLG values based on varying inputs, considering a 1 percent and a 2 percent change in IQ.

#### 12.3.1 Using a POD for a 1 Percent Change in IQ

When considering a POD based on a 1 percent change in IQ, the EPA used the following inputs to calculate the MCLG:

- POD = 3.1  $\mu$ g/kg/day, based on BBDR model output from the thirtcenth week of gestation, pTSH = 0.398, at 75  $\mu$ g/day iodine intake, at the 10th percentile of baseline fT4, matched with the function relating changes in maternal fT4 to child IQ from the EPA reanalysis of Korevaar et al. (2016).<sup>31</sup>
- RSC = 56 percent, based on subtracting the 90th percentile of perchlorate ingestion from food from the calculated RfD of 1.03 μg/kg/day based on the 3.1 μg/kg/day POD, assuming the 95th percentile concentration of perchlorate in each food.
- Uncertainty factor = 3, to account for intraspecies variability in sensitivity to perchlorate.

<sup>&</sup>lt;sup>31</sup> For justification, see Section [ REF Ref529865790 \r \h ].

• Drinking water ingestion (DWI) = 0.032 L/kg/day, based on the 90th percentile of consumersonly bodyweight-adjusted direct and indirect community drinking water intake for women of childbearing age from Kahn and Stralka (2008), as presented in Table 3-81 of the EFH (2011).

If using the median fT4 values, the POD would be equal to 4.2  $\mu g/kg/day$  based on the EPA's independent analysis of the Korevaar et al. (2016) data (Section [ REF \_Ref515367288 \r \h ]) and, thus, the RfD would equal 1.40  $\mu g/kg/day$  after applying the UF of 3. The RSC based on this new RfD of 1.40  $\mu g/kg/day$  would be 68 percent, based on subtracting the 90th percentile of perchlorate ingestion from food from the RfD, assuming the 95th percentile concentration of perchlorate in each food. The impact of varying each of these parameters is summarized in [ REF \_Ref530562482 \h ].

Table [ SEQ Table \\* ARABIC ]. Impact of Selecting Various fT4 Percentiles and DWI Rates on Inputs, and Calculated MCLG Based on a 1 Percent Change in IQ

	400 D 10 FT4 F00 D 10 10	
	10th Percentile fT4	50th Percentile fT4
MCLG using women of childbearing age DWI (0.032 L/kg/day)	18.0	29.8
MCLG using pregnant women DWI (0.033 L/kg/day)	17.5	28.9
Inputs		
POD (μg/kg/day)	3.1	4.2
RfD (µg/kg/day)	1.03	1.40
RSC	56%	68%

#### 12.3.2 Using a POD for a 2 Percent Change in IQ

When considering a POD based on a 2 percent change in IQ, the EPA used the following inputs to calculate the MCLG:

- POD =  $6.7 \,\mu\text{g/kg/day}$ , based on the BBDR model output from the thirteenth week of gestation, pTSH = 0.398, at 75  $\,\mu\text{g/day}$  iodine intake, at the 10th percentile of baseline fT4, matched with the function relating changes in maternal fT4 to child IQ from the EPA reanalysis of Korevaar et al. (2016).<sup>32</sup>
- RSC = 80 percent, based on subtracting the 90th percentile of perchlorate ingestion from food from the calculated RfD of 2.23  $\mu$ g/kg/day, based on the 6.7  $\mu$ g/kg/day POD, assuming the 95th percentile concentration of perchlorate in each food.
- Uncertainty factor = 3, to account for intraspecies variability in sensitivity to perchlorate.
- DWI = 0.032 L/kg/day, based on the 90th percentile of consumers-only bodyweight-adjusted direct and indirect community drinking water intake for women of childbearing age from Kahn and Stralka (2008), as presented in Table 3-81 of the EFH (2011).

If using the median fT4 values, the POD would be equal to 9.3  $\mu$ g/kg/day based on the EPA's independent analysis of the Korevaar et al. (2016) data (Section [ REF \_Ref515367288 \r \h \\* MERGEFORMAT ]) and, thus, the RfD would equal 3.10  $\mu$ g/kg/day after applying the UF of 3. The RSC, based on this new RfD of 3.10  $\mu$ g/kg/day would be 80 percent, as subtracting the 90th

<sup>&</sup>lt;sup>32</sup> For justification, see Section [ REF \_Ref529865904 \r \h ].

percentile of perchlorate ingestion from food from the RfD, assuming the 95th percentile concentration of perchlorate in each food yields an RSC of 85 percent, which is greater than the EPA suggested RSC "ceiling" of 80 percent [ ADDIN EN.CITE

<EndNote><Cite><Author>EPA</Author><Year>2000</Year><RecNum>2026</RecNum><Displa yText>(U.S. EPA, 2000)</DisplayText><record><rec-number>2026</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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EPA,</author></authors></contributors><titles><title>Methodology for Deriving Ambient Water Quality Criteria for the Protection of Human Health

(2000) < title > dates > dates > dates > dates > pub-location > Washington, DC < pub-location > isbn > EPA-822-B-00-004 < isbn > curls > related-

 $urls > <\!\!url\!>\!\!https://www.nj.gov/drbc/library/documents/EPA\_human-health-$ 

criteria2000.pdf</url></related-urls></urls></record></Cite></EndNote>]. The impact of varying each of these parameters is summarized in [ REF Ref519031384 \h \\* MERGEFORMAT ].

Table [ SEQ Table \\* ARABIC ]. Impact of Selecting Various fT4 Percentiles and DWI Rates on Inputs, and Calculated MCLG Based on a 2 Percent Change in IQ

	MCLG	
	10th Percentile fT4	50th Percentile fT4
MCLG using women of childbearing age DWI (0.032 L/kg/day)	56	78
MCLG using pregnant women DWI (0.033 L/kg/day)	54	75
Inputs		
POD (µg/kg/day)	6.7	9.3
RfD (µg/kg/day)	2.23	3.10
RSC	80%	80%ª

<sup>&</sup>lt;sup>a</sup> For a 2 IQ point change at the 50th percentile fT4 the calculated RSC is 85%. The EPA therefore applied the 80% "ceiling" instead of this calculated value

The **bolded** value indicates the selected value.

[ REF \_Ref530562482 \h ] and [ REF \_Ref519031384 \h ] show that the selected DWI rate has very little impact on the estimated MCLG. Using the median fT4 value results in a higher POD, a higher RSC, and subsequently also higher potential MCLGs. Since the EPA is aiming to protect the most sensitive population, the use of the lower percentile fT4 value is appropriate.

#### 12.4 Conclusion

The EPA identified potential MCLGs for perchlorate in drinking water in the range of  $18.0 \mu g/L$  to  $56 \mu g/L$ , informed by an evaluation using the best available peer-reviewed science and methods.

## 13. References

[ ADDIN EN.REFLIST ]



# Health Risk Reduction and Cost Analysis of the Proposed Perchlorate National Primary Drinking Water Regulation

Office of Water (4607M) EPA XXX-X-XXX January 2019 [ HYPERLINK "http://www.epa.gov/safewater" ]

# **Table of Contents**

[ TOC \o "1-1" \h \z \t "Heading 2,2,Heading 3,3,Heading 5,1" ]

# **Table of Exhibits**

[ TOC \t "Caption" \c ]

# **Abbreviations and Acronyms**

ACS American Community Survey

ADHD attention-deficit/hyperactive disorder
Agency U.S. Environmental Protection Agency
ANSI American National Standards Institute
ASES Annual Social and Economic Supplement

BBDR biologically based dose response

BLS Bureau of Labor Statistics
CCL Contaminant Candidate List
CCR Consumer Confidence Report

CDC Centers for Disease Control and Prevention

CFR Code of Federal Regulations

Council National Drinking Water Advisory Council

CPS Current Population Survey

CSFII Continuing Survey of Food Intakes by Individuals

CVD cardiovascular disease CWS community water system

CWSS Community Water System Survey

DWI drinking water intake

EFH Exposure Factors Handbook

EPA U.S. Environmental Protection Agency FDA U.S. Food and Drug Administration

fT4 free thyroxine
GW gestational week
UPL Health Performance

HRL Health Reference Level

HRRCA Health Risk Reduction and Cost Analysis

ICC intraclass correlation coefficient

IQintelligence quotientLDLlow-density lipoproteinMCLmaximum contaminant levelMCLGmaximum contaminant level goal

MHI mean household income μg/day micrograms per day

μg/kg/day micrograms per kilogram per day

μg/L micrograms per liter
MGD million gallons per day

MOA mode of action

MRL minimum reporting level

NCES National Center for Education Statistics NCOD National Contaminant Occurrence Database

NCWS non-community water system

NDWAC National Drinking Water Advisory Council

NHANES National Health and Nutrition Examination Survey

NIS sodium-iodide symporter

NLSY National Longitudinal Survey of Youth

NPDWR national primary drinking water regulation

NRC National Research Council NSF National Science Foundation

NTNCWS non-transient non-community water system

NTTAA National Technology Transfer and Advancement Act

O&M operating and maintenance

OMB Office of Management and Budget

ClO<sub>4</sub> Perchlorate

PBPK physiologically based pharmacokinetic

PD pharmacodynamic pmol/L picomoles per litre

POTW publicly owned treatment works

POU point-of-use

pph persons per household

PUMS Public Use Microdate Sample

PWS public water system
RFA Regulatory Flexibility Act

RfD reference dose

SAB Science Advisory Board

SBA Small Business Administration

SBREFA Small Business Regulatory Enforcement Fairness Act

SDWA Safe Drinking Water Act

SDWIS Safe Drinking Water Information System

SDWIS/FED Federal Safe Drinking Water Information System

SSA Social Security Administration

SSCT small system compliance technology

T3 triiodothyronine T4 thyroxine

TNCWS transient non-community water system

TSH thyroid stimulating hormone

UCMR Unregulated Contaminant Monitoring Rule
UCMR 1 Unregulated Contaminant Monitoring Rule (first)

UMRA Unfunded Mandates Reform Act

WBS work breakdown structure

# 1 Introduction

The U.S. Environmental Protection Agency (EPA or the Agency) is proposing to regulate perchlorate in drinking water distributed by certain public water systems (PWSs). In 2011, the EPA determined that a national primary drinking water regulation (NPDWR) for perchlorate would result in a meaningful opportunity to reduce health risks [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"xCfQfPLZ","properties": {"formattedCitation":"(USEPA, 2011a)","plainCitation":"(USEPA, 2011a)", "noteIndex":0}, "citationItems":[{"id":191, "uris":["http://zotero.org/groups/945096/item s/4EFIN6HN"],"uri":["http://zotero.org/groups/945096/items/4EFIN6HN"],"itemData":{"id":19 1,"type":"webpage", "title": "Drinking Water: Regulatory Determination on Perchlorate. Federal Register Notice. 76 FR No. 29. Pages 7762-7767. (February 11, 2011) (to be codified at 40 C.F.R pt. 141).","URL":"https://www.federalregister.gov/articles/2011/02/11/2011-2603/drinking-water-regulatory-determination-on-perchlorate", "shortTitle": "Drinking Water: Regulatory Determination on Perchlorate. Federal Register Notice. 76 FR No. 29. Pages 7762-7767. (February 11, 2011) (to be codified at 40 C.F.R pt. 141).", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2011"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Based on the best available scientific information on the health effects of perchlorate, the EPA is proposing a maximum contaminant level goal (MCLG) of 56 micrograms per liter (µg/L). The EPA is also proposing an enforceable maximum contaminant level (MCL) of 56 µg/L and compliance monitoring requirements consistent with the Standardized Monitoring Framework for inorganic contaminants in the Title 40 Code of Federal Regulations (CFR) 141.23(c).

# 1.1 Purpose of Economic Analysis

The purpose of this economic analysis is to provide a description of the potential benefits and costs of the proposed perchlorate NPDWR. An economic analysis is required for all significant rules under Executive Order (EO) 12866 (*Regulatory Planning and Review*). In addition, Section 1412(b)(3)(C) of the 1996 Amendments to the Safe Drinking Water Act (SDWA) requires the EPA to prepare a Health Risk Reduction and Cost Analysis (HRRCA) in support of any NPDWR that includes an MCL. The analysis in this document addresses these and other regulatory reporting requirements. With respect to the HRRCA requirements, this document provides the following:

- Quantifiable and non-quantifiable health risk reduction benefits for which there is a factual basis in the rulemaking record to conclude that such benefits are likely to occur as the result of treatment to comply with each level (Chapter [ REF \_Ref523452255 \r \h ]);
- Quantifiable and non-quantifiable health risk reduction benefits for which there is a
  factual basis in the rulemaking record to conclude that such benefits are likely to occur
  from reductions in co-occurring contaminants that may be attributed solely to compliance
  with the MCL, excluding benefits resulting from compliance with other proposed or
  promulgated regulations (Chapter [ REF \_Ref523452255 \r \h ]);
- Quantifiable and non-quantifiable costs for which there is a factual basis in the rulemaking record to conclude that such costs are likely to occur solely as a result of

- compliance with the MCL, including monitoring, treatment, and other costs, and excluding costs resulting from compliance with other proposed or promulgated regulations (Chapter [ REF Ref525051776 \r \h ]);
- Incremental costs and benefits associated with each alternative MCL considered (Chapter [ REF Ref535310132 \r \h ]);
- Effects of the contaminant on the general population and on groups within the general population, such as infants, children, pregnant women, the elderly, individuals with a history of serious illness, or other sub-populations that are identified as likely to be at greater risk of adverse health effects due to exposure to contaminants in drinking water than the general population (Chapter [ REF Ref523452255 \r \h ]);
- Any increased health risk that may occur as the result of compliance, including risks associated with co-occurring contaminants (Chapter [ REF Ref523452255 \r \h ]); and
- Other relevant factors, including the quality and extent of the information, the uncertainties in the analysis, and factors with respect to the degree and nature of the risk (Chapters [ REF \_Ref523452255 \r \h ] to [ REF \_Ref535310160 \r \h ]).

#### 1.2 Outline

This document contains the following:

- Chapter [ REF \_Ref535310173 \r \h ] [ REF \_Ref535310189 \h ] describes the proposed NPDWR and alternatives the Agency considered;
- Chapter [ REF \_Ref535310292 \r \h ] [ REF \_Ref535310201 \h ] provides key information about current conditions that form the baseline for the subsequent benefit and cost analysis, including a description of perchlorate occurrence in drinking water and the potentially affected entities;
- Chapter [ REF \_Ref523452255 \r \h ] [ REF \_Ref523452255 \h ] provides a summary of the health effects of concern, the basis for the proposed MCLG, and the method that the Agency used to estimate the health risk reductions of proposing an enforceable MCL;
- Chapter [ REF \_Ref525051776 \r \h ] [ REF \_Ref525051776 \h ] describes the potentially affected entities and the basis for estimating costs to implement the proposed rule and comply with the MCL;
- Chapter [ REF \_Ref535310268 \r \h ] [ REF \_Ref535310229 \h ] provides side-by-side comparison of the benefits and costs by the proposed rule alternative;
- Chapter [ REF \_Ref535310259 \r \h ] [ REF \_Ref535310237 \h ] addresses several reporting requirements under various statutes and Executive Orders; and
- Appendices provide additional details for selected topics in the main document.

Information in the chapters often summarizes more detailed technical support documents, which are cited throughout the text.

#### 1.3 Public Health Concerns to Be Addressed

Perchlorate is an anion containing one chlorine atom bound to four oxygen atoms (ClO<sub>4</sub><sup>-</sup>). It combines with cations to form salts including ammonium perchlorate and potassium perchlorate. Each salt has different chemical properties including molecular weight, density, boiling/melting

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points, and solubility [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"MwgdYrvr","properties": {"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA ***-**-**,"author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].
```

Perchlorate ingestion – via drinking water or food – can adversely affect human health. The main target organ for perchlorate toxicity is the thyroid gland, where perchlorate competes with iodide for transport and, therefore, may lead to decreases in iodide uptake, which can adversely affect hormone production levels. These changes in thyroid hormones in a pregnant or nursing woman can be linked to changes in the intelligence quotient (IQ) in her offspring using dose-response functions derived from the peer-reviewed literature. Chapter 4 describes these relationships in more detail because they form the basis for a quantitative benefits analysis. Chapter 4 also identifies other adverse neurological effects and cardiovascular disease (CVD) that may be linked to perchlorate exposure.

#### 1.3.1 Rule Objectives and Public Health Concerns

The proposed rule will reduce perchlorate concentrations in the drinking water distributed by PWSs that exceed the proposed MCL. These reductions will ultimately reduce the incidence of IQ impacts among offspring born to women exposed to high perchlorate concentrations under baseline occurrence conditions. The proposed rule also includes requirements for every affected drinking water system to conduct initial monitoring to determine whether perchlorate exceeds the proposed MCL. Customers of these systems will benefit from knowing whether they are exposed to perchlorate in excess of the MCL and, if so, what measures they can take to protect themselves and their families until the treatment or other control efforts reduce perchlorate to a compliant level.

#### 1.3.2 Sources and Mechanisms of Exposure

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The Agency's Perchlorate Occurrence and Monitoring Report technical support document [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"TKLDxDYY","properties": {"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA ***-***","author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] provides a review of perchlorate sources, fate, and transport. The following discussion provides a brief summary of that discussion.
```

Perchlorate is naturally occurring and man-made. Natural sources include geologic materials such as potash ore from New Mexico and sodium nitrate-rich soils in Chile, which have been

used to produce fertilizers applied in the United States. Man-made sources include the use of ammonium perchlorate as an oxidizer in solid fuels for rockets and fireworks. Major production sites include facilities in Nevada and Utah.

Perchlorate is likely to be mobile in soil and aqueous media because perchlorate salts are highly soluble and unlikely to sorb to minerals or organic matter. Dissolved in water, the perchlorate ion is unlikely to undergo reduction, hydrolysis, or direct photolysis; form insoluble metal complexes; or volatilize from water. Therefore, perchlorate is likely to persist in water absent biological removal or uptake processes. Releases to air are likely to result in eventual deposition to soil or water. Thus, once perchlorate reaches surface or ground water sources of drinking water, these characteristics suggest it is likely to persist in the water. These same characteristics indicate that effective treatment processes include biological removal and anion exchange.

## 1.4 Regulatory History and Background

This section describes the process that led the Agency to propose an NPDWR for perchlorate. The SDWA requires the EPA to make determinations every five years of whether to regulate at least five contaminants on the Contaminant Candidate List (CCL). The CCL is a list of drinking water contaminants that are known to occur in PWSs and are not currently subject to the EPA drinking water regulations. Contaminants listed on the CCL may require future regulation under the SDWA. The EPA uses the CCL to identify priority contaminants for regulatory decision-making and information collection. The EPA included perchlorate on the first, second, and third CCLs published in 1998, 2005, and 2009, respectively.

After including contaminants on the CCL, the Agency continues to collect data and encourage further research on listed contaminants to better understand potential health effects and at what levels they occur in drinking water. One avenue for collecting occurrence data for contaminants that are suspected to be present in drinking water and do not have health-based standards set under the SDWA is through the Unregulated Contaminant Monitoring Rule (UCMR). The UCMR provides for monitoring of up to 30 contaminants every 5 years at large systems and a representative sample of small PWSs serving less than or equal to 10,000 people. Analytical results are stored in a National Contaminant Occurrence Database (NCOD). The first UCMR cycle (UCMR 1) required monitoring for 26 contaminants, including perchlorate, between 2001 and 2003.

In 2005, at the request of the EPA and other federal agencies, the National Research Council (NRC) evaluated the health implications of perchlorate ingestion. The NRC concluded that perchlorate exposure could inhibit the transport of iodide into the thyroid, leading to thyroid hormone deficiency [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"PINY0WnN","properties": {"formattedCitation":"(NRC, 2005)","plainCitation":"(NRC, 2005)","noteIndex":0},"citationItems":[{"id":349,"uris":["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData": {"id":349,"type":"book","title":"Health Implications of Perchlorate Ingestion","publisher":"National Academies Press","publisher-place":"Washington, DC","event-place":"Washington, DC","author":[{"literal":"NRC"}],"issued": {"date-parts":[["2005"]]}}}],"schema":"https://github.com/citation-style-

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language/schema/raw/master/csl-citation.json"} ]. Based on NRC's recommendations, the EPA
adopted a reference dose (RfD) of 0.7 micrograms per kilogram per day (µg/kg/day) based on a
study [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"a3u94lt6me","properties":{"formattedCitation":"(Greer et al.,
2002)", "plainCitation": "(Greer et al.,
2002)", "noteIndex":0}, "citationItems":[{"id":387, "uris":["http://zotero.org/groups/945096/items/
6AKUNIX6"], "uri": ["http://zotero.org/groups/945096/items/6AKUNIX6"], "itemData": {"id":387
"type":"article-journal", "title": "Health effects assessment for environmental perchlorate,"
contamination: the dose response for inhibition of thyroidal radioiodine uptake in
humans", "container-title": "Environmental Health
Perspectives", "page": "927", "volume": "110", "issue": "9", "author": [{"family": "Greer", "given": "M
onte A."}, {"family": "Goodman", "given": "Gay"}, {"family": "Pleus", "given": "Richard
C."},{"family":"Greer","given":"Susan E."}],"issued":{"date-
parts":[["2002"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] in healthy adults for the inhibition of
radioactive iodide uptake and the application of an uncertainty factor of 10 for intraspecies
variability [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID": "0oHz805e", "properties": {"formattedCitation": "(USEPA,
2005b)","plainCitation":"(USEPA,
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1007, "type": "article", "title": "Integrated Risk Information System (IRIS) Chemical Assessment
Summary: Perchlorate (ClO4-) and Perchlorate Salts", "publisher": "USEPA National Center for
Environmental Assessment", "author": [{"literal": "USEPA"}], "issued": {"date-
parts":[["2005"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"}].
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In October 2008, the EPA published a preliminary regulatory determination not to regulate perchlorate in drinking water using a Health Reference Level (HRL) of 15 µg/L based on the RfD of 0.7 µg/kg/day [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID": "FZ6WMtAv", "properties": {"formattedCitation": "(USEPA, 2008a)", "plainCitation": "(USEPA, 2008a)", "noteIndex":0}, "citationItems":[{"id":934, "uris":["http://zotero.org/groups/945096/item s/HBX88QM9"],"uri":["http://zotero.org/groups/945096/items/HBX88QM9"],"itemData":{"id": 934,"type":"article-journal","title":"Drinking water: Preliminary regulatory determination on perchlorate", "container-title": "Federal Register", "volume": "73", "issue": "198", "abstract": "SUMMARY: This action presents EPA's preliminary regulatory determination for perchlorate in accordance with the Safe Drinking Water Act (SDWA). The Agency has determined that a national primary drinking water regulation (NPDWR) for perchlorate would not present \"a meaningful opportunity for health risk reduction for persons served by public water systems." The SDWA requires EPA to make determinations every five years of whether to regulate at least five contaminants on the Contaminant Candidate List (CCL). EPA included perchlorate on the first and second CCLs that were published in the Federal Register on March 2, 1998 and February 24, 2005. Most recently, EPA presented final regulatory determinations regarding 11 contaminants on the second CCL in a notice published in the Federal Register on July 30, 2008. In today's action, EPA presents supporting rationale and requests public comment on its preliminary regulatory determination for perchlorate. EPA will

make a final regulatory determination for perchlorate after considering comments and information provided in the 30-day comment period following this notice. EPA plans to publish a health advisory for perchlorate at the time the Agency publishes its final regulatory determination to provide State and local public health officials with technical information that they may use in addressing local contamination.","ISSN":"ISSN 0097-6326 EISSN 2167-2520","shortTitle":"Federal Register","journalAbbreviation":"Fed.

Reg.","language":"English","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2008"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The preliminary determination found that perchlorate did not occur with a frequency and at levels of public health concern, and that development of a regulation did not present a meaningful opportunity for health risk reduction for persons served by PWSs.

In August 2009, the EPA published a supplemental request for comment with a new analysis that derived potential alternative HRLs for 14 life stages, including infants and children. The analysis used the RfD of 0.7  $\mu$ g/kg/day and life stage-specific bodyweight and exposure information [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"XzGawKtq","properties": {"formattedCitation":"(USEPA, 2009a)","plainCitation":"(USEPA, 2009a)","noteIndex":0},"citationItems":[{"id":309,"uris":["http://zotero.org/groups/945096/items/ILBCYL66"],"itemData": {"id":309,"type":"report","title":"Inhibition of the Sodium-Iodide Symporter By Perchlorate: An Evaluation of Lifestage Sensitivity Using Physiologically Based Pharmacokinetic (PBPK) Modeling (Final Report)","collection-title":"EPA/600/R-08/106A","publisher-place":"Washington, D.C.","event-place":"Washington, D.C.","author":[{"family":"USEPA","given":""}],"issued":{"date-parts":[["2009"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

After considering comments on the October 2008 and August 2009 notices, the EPA made a final determination in February 2011 to regulate perchlorate in drinking water [ADDIN ZOTERO ITEM CSL CITATION {"citationID": "E1I1fcIP", "properties": {"formattedCitation": "(USEPA, 2011a)","plainCitation":"(USEPA, 2011a)","noteIndex":0},"citationItems":[{"id":191,"uris":["http://zotero.org/groups/945096/item s/4EFIN6HN"], "uri": ["http://zotero.org/groups/945096/items/4EFIN6HN"], "itemData": {"id":19 1,"type":"webpage","title":"Drinking Water: Regulatory Determination on Perchlorate. Federal Register Notice. 76 FR No. 29. Pages 7762-7767. (February 11, 2011) (to be codified at 40 C.F.R pt. 141).","URL":"https://www.federalregister.gov/articles/2011/02/11/2011-2603/drinking-water-regulatory-determination-on-perchlorate", "shortTitle": "Drinking Water: Regulatory Determination on Perchlorate. Federal Register Notice. 76 FR No. 29. Pages 7762-7767. (February 11, 2011) (to be codified at 40 C.F.R pt. 141).", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2011"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. The Agency found that perchlorate may have an adverse effect on the health of persons and is known to occur in public drinking water systems with a frequency and at levels that present a public health concern. As a result of the

determination, the EPA initiated the process to develop an MCLG and NPDWR for perchlorate under the SDWA.

Since then, the EPA has developed a novel approach for deriving an MCLG for perchlorate in drinking water based on feedback from the EPA's Science Advisory Board (SAB), including the use of a biologically based dose response (BBDR) model [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"IWXxamER","properties":{"formattedCitation":"(USEPA, 2018e)","plainCitation":"(USEPA, 2018e)","noteIndex":0},"citationItems":[{"id":926,"uris":["http://zotero.org/groups/945096/items/ANBBTDKU"],"itemData":{"id ":926,"type":"report","title":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","shortTitle":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","author":[{"literal":"USEPA"}],"issued":{"dateparts":[["2018"]]}}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

### 1.5 Rationale for the Proposed Regulation

This section provides the statutory and economic rationales for choosing a regulatory approach to address the public health consequences of drinking water contamination.

#### 1.5.1 Statutory Authority

Section 1412(b)(1)(A) of the SDWA requires the EPA to establish NPDWRs for contaminants that may have an adverse public health effect; that are known to occur or that present a substantial likelihood of occurring once in PWSs, at a frequency and level of public concern; and that present a meaningful opportunity for health risk reduction for persons served by PWSs.

#### 1.5.2 Economic Rationale for Regulation

The Office of Management and Budget (OMB) Circular A-4 (1996 and 2003) states that "in order to establish the need for the proposed action, the analysis should discuss whether the problem constitutes a significant market failure." This section describes the types of market failures that NPDWRs address.

In a perfectly competitive market, market forces guide buyers and sellers to attain the best possible social outcome. A perfectly competitive market occurs when there are many producers of a product, many buyers for the product, and both producers and buyers have complete knowledge regarding the products of each firm. Also, there must not be any barriers to entry into the industry, and producers in the industry must not have any advantage over potential new producers. Several factors in the public water supply industry do not satisfy the requirements for a perfect market and lead to market failures that may require regulation.

First, water utilities are natural monopolies. A natural monopoly exists because it is not economically efficient to have multiple suppliers competing to build multiple systems of pipelines, reservoirs, wells, and other facilities. Instead, a single firm or government entity performs these functions generally under public control. Under monopoly conditions, consumers are provided only one level of service with respect to the quality of the product, in this case

drinking water quality. If consumers do not believe that the market of safety in public health production is adequate, they cannot simply switch to another water utility or a perceived higher-quality source of supply (e.g., bottled water) without incurring additional cost.

Second, high information and transaction costs impede the public's understanding of health and safety issues concerning drinking water quality. The types of health risks potentially posed by trace quantities of drinking water contaminants involve the analysis and distillation of complex toxicological data and health sciences. The EPA developed the Consumer Confidence Report (CCR) rule to make water quality information more easily available to consumers. The CCR rule requires community water systems (CWSs) to mail their customers an annual report on local drinking water quality. However, consumers still have to analyze this information for its health risk implications. Even if informed consumers are able to engage utilities regarding these health issues, the costs of such engagement, known as "transaction costs" (in this case measured in personal time and commitment), present another significant impediment to consumer expression of risk preference.

SDWA regulations are intended to provide a level of protection from exposure to drinking water contaminants that would not otherwise occur in the existing market environment of public water supply. The regulations set minimum performance requirements for all public water supplies in order to reduce the risk confronted by all consumers from exposure to drinking water contaminants. SDWA regulations are not intended to restructure market mechanisms or to establish competition in supply. Rather, SDWA standards establish the level of service to be provided in order to better reflect the public's preference for safety. The Federal regulations remove the high information and transaction costs by acting on behalf of all consumers in balancing the risk reduction and the social costs of achieving this reduction.

# 2 Consideration of Regulatory Alternatives

The Agency's proposed rule comprises the following elements: an MCLG, an MCL, and monitoring and reporting requirements. This section describes each element of the proposed rule and identifies the alternatives that the Agency considered during the rule-making process.

#### **2.1 MCLG**

Section 1412 (b)(4)(A) of the SDWA requires that – when regulating a contaminant – the EPA first sets an MCLG "at the level at which no known or anticipated adverse effects on the health of persons occur and which allows an adequate margin of safety." MCLGs are non-enforceable health goals. For this rulemaking, the EPA is proposing to set an MCLG of  $56 \mu g/L$  for perchlorate based on a 2 IQ point decrease. Section 4 describes the basis for this MCLG. The Agency also considered an alternative MCLG of  $18 \mu g/L$  based on a 1 IQ point decrease.

#### 2.2 MCL Alternatives

Section 1412 (b)(4)(B) of the SDWA requires that when the EPA sets an enforceable MCL, it is "as close to the maximum contaminant level goal as is feasible." Section 1412 (b)(4)(D) defines feasible as follows: "feasible with the use of the best technology, treatment techniques and other means which the Administrator finds, after examination for efficacy under field conditions and not solely under laboratory conditions, are available (taking cost into consideration)." Furthermore, Section 1401 (1)(A)(i) defines an MCL as feasible when "it is economically and technologically feasible to ascertain the level of such contaminant in water in public water systems." Finally, under Section 1412 (b)(6), the Administrator can determine that the benefits of an MCL as close as feasible to the MCLG "would not justify the costs of complying with the level" and promulgate an MCL "that maximizes health risk reduction benefits at a cost that is justified by the benefits." The EPA determined that an MCL of 56  $\mu$ g/L is feasible and, therefore, is proposing to set the MCL equal to the MCLG of 56  $\mu$ g/L. For the alternative MCLG of 18  $\mu$ g/L, the Agency determined that setting the MCL equal to the MCLG was feasible.

The proposed rule applies to certain PWSs. A PWS is a system that provides water for human consumption to the public through pipes or other constructed conveyances and has at least 15 service connections or regularly serves at least 25 individuals for at least 60 days per year [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"a15poirtvap","properties": {"formattedCitation":"(USEPA, 2017b)","plainCitation":"(USEPA, "Color,") | "Color," | "

2017b)","noteIndex":0},"citationItems":[{"id":922,"uris":["http://zotero.org/groups/945096/items/X6GF6JJW"],"uri":["http://zotero.org/groups/945096/items/X6GF6JJW"],"itemData":{"id":922,"type":"webpage","title":"Information about Public Water

 $Systems", "URL": "https://www.epa.gov/dwreginfo/information-about-public-water-systems", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2017"]]}, "accessed": {"date-parts": [["2018",8,17]]}}}], "schema": "https://github.com/citation-style-$ 

language/schema/raw/master/csl-citation.json"} ]. PWSs may be publicly or privately owned. Types of PWSs include CWSs and non-community water systems (NCWSs), which may be transient or non-transient. The PWS types are defined as follows [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"BDOZjmky","properties": {"formattedCitation":"(USEPA,

Preliminary draft for internal EPA review[ PAGE \\* MERGEFORMAT ]
Do not cite or quote

2009b; 2017b)", "plainCitation":"(USEPA, 2009b; 2017b)", "noteIndex":0}, "citationItems":[{"id":925, "uris":["http://zotero.org/groups/945096/items/QPD46SEA"], "itemData": {"id":925, "type":"article", "title": "2006 Community Water System Survey - Volume I: Overview", "URL": "https://www.epa.gov/dwstandardsregulations/community-water-system-survey", "author":[{"literal": "USEPA"}], "issued": {"date-parts":[["2009",2]]}, "accessed": {"date-parts":[["2018",8,17]]}}}, {"id":922, "uris":["http://zotero.org/groups/945096/items/X6GF6JJW"], "uri":["http://zotero.org/groups/945096/items/X6GF6JJW"], "itemData": {"id":922, "type": "webpage", "title": "Information about Public Water Systems", "URL": "https://www.epa.gov/dwreginfo/information-about-public-water-systems", "author":[{"literal": "USEPA"}], "issued": {"date-parts":[["2017"]]}, "accessed": {"date-parts":[["2018",8,17]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]:

- CWSs supply water to at least 15 service connections used by year-round residents or regularly serve at least 25 year-round residents.
- NCWSs do not serve year-round residents. These water systems serve areas where the people do not stay for long periods of time or where the same population is served less than year-round. Two categories of NCWSs are:
  - Non-transient non-community water systems (NTNCWSs) serve at least 25 of the same people for at least 6 months of the year. Examples include schools and office buildings.
  - Transient non-community water systems (TNCWSs) serve fewer than 25 of the same people over 6 months of the year. Examples include gas stations and campgrounds.

The EPA proposes to regulate perchlorate at CWSs and NTNCWSs.

# 2.3 Monitoring Requirements

The EPA is proposing the following monitoring requirements for perchlorate:

- Initial monitoring one year of monitoring by all affected systems to determine compliance with the proposed MCL; and
- Long-term monitoring consistent with the Standardized Monitoring Framework for inorganic contaminants.

#### 2.3.1 Initial Monitoring Requirements

The Agency is proposing that all CWSs and NTNCWSs conduct one year of initial monitoring. Large CWSs, which serve more than 10,000 people, will conduct initial monitoring within the first 3 years after the effective date. Small CWSs, which serve up to 10,000 people, and all NTNCWSs have 6 years after the effective date to conduct initial monitoring. To meet the initial monitoring requirement, all water systems must collect four quarterly samples.

#### 2.3.2 Long-Term Monitoring Requirements

Following their respective initial monitoring period, CWS and NTNCWSs will conduct long-term monitoring according to the Standardized Monitoring Framework [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"aGOrxjKr","properties": {"formattedCitation":"(USEPA, 2004)","plainCitation":"(USEPA,

2004)","noteIndex":0},"citationItems":[{"id":993,"uris":["http://zotero.org/groups/945096/items/XIN79HTB"],"uri":["http://zotero.org/groups/945096/items/XIN79HTB"],"itemData":{"id":993, "type":"article","title":"The Standardized Monitoring Framework: A Quick Reference Guide","publisher":"EPA-816-F-04-

010","author":[{"family":"USEPA","given":""}],"issued":{"date-parts":[["2004"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. Monitoring frequency depends on the water source and whether a system qualifies for a perchlorate monitoring waiver or exceeds the MCL. [ REF Ref525846605 \h ] shows the proposed monitoring frequencies.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Long-Term Perchlorate Monitoring Requirements

Water Source	Waiver	No Waiver, Reliably and Consistently ≤ MCL	> MCL or Not Reliably and Consistently ≤ MCL
Groundwater	1 sample every 9 years	1 sample every 3 years	Quarterly samples
Surface water	1 sample every 9 years	Annual sample	Quarterly samples

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"IJ22Yulq","properties": {"formattedCitation":"(2004)","plainCitation":"(2004)","noteIndex":0},"citationItems":[{"id":993,"uris":["http://zotero.org/groups/945096/items/XIN79HTB"],"uri":["http://zotero.org/groups/945096/items/XIN79HTB"],"itemData": {"id":993,"type":"article","title":"The Standardized Monitoring Framework: A Quick Reference Guide","publisher":"EPA-816-F-04-

010", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-parts": [["2004"]]}}, "suppressauthor": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

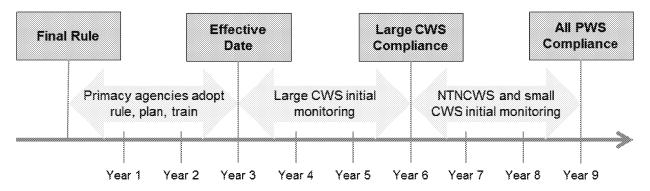
# 2.4 Reporting Requirements

The proposed rule includes several reporting requirements. Water systems must provide monitoring results to primacy agencies. These agencies must report violation-related information to the EPA. Systems may also include perchlorate monitoring information in their annual CCR. Finally, systems will have public notification requirements in the event of an MCL violation.

# 2.5 Implementation Schedule

The EPA is proposing effective dates that vary by system size and type. [REF\_Ref529978990 \h ] provides an overview of the implementation schedule. The effective date of the rule is three years after the rule finalization. Then, large CWSs will have three years (i.e. the first compliance period after the effective date) to complete their initial monitoring and install needed controls. NTNCWSs and small CWSs will complete their initial monitoring in the subsequent 3-year compliance period (i.e., years six to nine). All covered systems will be in compliance by year nine.

#### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Proposed Rule Timeline



# 2.6 Summary of Regulatory Alternatives

[ REF \_Ref525846788 \h ] provides a summary of the regulatory alternatives that EPA considered for the proposed perchlorate rule.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Long-Term Perchlorate Monitoring Requirements

Element	Proposed Rule – Preferred Alternative	Alternative A
MCLG	56 μg/L	18 μg/L
MCL	56 μg/L	18 μg/L
Monitoring	Standard Monitoring Framework	Standard Monitoring Framework

### 3 Baseline Analysis

### 3.1 Introduction

This chapter presents the data and assumptions used to establish the baseline for calculating the costs, benefits, and economic impacts of the proposed perchlorate rule. The baseline is the EPA's expectation of the conditions that would exist in the absence of perchlorate regulation. The baseline includes a profile of the PWSs that are potentially affected by a perchlorate regulation and current demographic information to characterize the exposed population. The baseline also includes estimates of perchlorate occurrence and exposure, which affect the Agency's analysis of costs and benefits of the proposed rule.

### 3.1.1 Background and Purpose

The purpose of the baseline analysis is to describe the data available to develop baseline characterization of the water supply industry, prior to the promulgation and implementation of an NPDWR for perchlorate. In the baseline analysis, the EPA defines the various types of water systems and provides information on the number and size of these water systems. The EPA also presents characteristics of the water systems, including population served, number of entry points, current treatment technologies that are in place, and the amount of production. The baseline analysis also discusses water consumption per household, estimates of current perchlorate occurrence, and population groups that may be susceptible to the health effects of perchlorate exposure.

### 3.1.2 Chapter Organization

The remainder of this chapter is organized into three sections. Section [ REF \_Ref523471222 \r \h ] provides a description of the data sources used in the baseline analysis. Section [ REF \_Ref523415085 \r \h ] characterizes the water supply industry as outlined above; this section also includes assumptions made in the analysis. The EPA then discusses current perchlorate occurrence in Section [ REF \_Ref523404705 \r \h ], and sensitive life stages and other populations in Section [ REF \_Ref523459825 \r \h ].

### 3.2 Data Sources

Data sources for the baseline analysis include data specific to the water supply industry and other information needed to characterize baseline conditions. Specifically, the EPA characterizes water systems using the EPA's Safe Drinking Water Information System (SDWIS) database and the 2006 Community Water System Survey (CWSS). The Agency uses perchlorate monitoring data from the UCMR 1 and comparable, but more recent, data sources to characterize baseline exposure to perchlorate in drinking water. The data sources are described in detail below.

#### 3.2.1 SDWIS/FED and Other Sources for Water System Data

The EPA uses the SDWIS Fed Data Warehouse (SDWIS/FED) to characterize the universe of affected systems. SDWIS/FED contains information on the more than 146,000 active PWSs in the United States, as reported by states and the EPA Regions [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"QA4gnxdO","properties": {"formattedCitation":"(USEPA, 2018f)","plainCitation":"(USEPA,

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2018f)", "noteIndex":0}, "citationItems":[{"id":923, "uris":["http://zotero.org/groups/945096/items
/CW34PNAZ"],"uri":["http://zotero.org/groups/945096/items/CW34PNAZ"],"itemData":{"id":9
23,"type":"article","title":"Safe Drinking Water Information System Federal Reports
Search","URL":"https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:....","author":[{"literal":"USE
PA"}],"issued":{"date-parts":[["2018"]]},"accessed":{"date-
parts":[["2018",8,28]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. Information reported includes basic
information on each water system, details on each PWS's compliance and violation history, and
states' actions to enforce drinking water regulations [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID": "PNHTPwkH", "properties": {"formattedCitation": "(USEPA,
2018f)", "plainCitation": "(USEPA,
2018f)", "noteIndex":0}, "citationItems":[{"id":923, "uris":["http://zotero.org/groups/945096/items
/CW34PNAZ"],"uri":["http://zotero.org/groups/945096/items/CW34PNAZ"],"itemData":{"id":9
23,"type": "article", "title": "Safe Drinking Water Information System Federal Reports
Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:...:", "author": [{"literal": "USE
PA"}],"issued":{"date-parts":[["2018"]]},"accessed":{"date-
parts":[["2018",8,28]]}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. Basic information about PWSs includes
location, number of people served, system type (e.g., CWS or NTNCWS), operation schedule
(year-round or seasonal), ownership type (public or private), and characteristics of the source
water. To characterize baseline conditions for this analysis, the EPA downloaded data from
SDWIS/FED in August 2018.1
Another data source specific to the water supply industry is the 2006 CWSS report [ ADDIN
ZOTERO ITEM CSL CITATION
{"citationID":"kFw5Lm0h","properties":{"formattedCitation":"(USEPA,
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ZOTERO\_ITEM CSL\_CITATION {"citationID":"kFw5Lm0h","properties": {"formattedCitation":"(USEPA, 2009b)","plainCitation":"(USEPA, 2009b)","noteIndex":0},"citationItems": [{"id":925,"uris": ["http://zotero.org/groups/945096/items/QPD46SEA"],"uri": ["http://zotero.org/groups/945096/items/QPD46SEA"],"itemData": {"id":925,"type":"article","title":"2006 Community Water System Survey - Volume I: Overview","URL":"https://www.epa.gov/dwstandardsregulations/community-water-system-survey", "author": [{"literal":"USEPA"}],"issued": {"date-parts": [["2009",2]]}, "accessed": {"date-parts": [["2018",8,17]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. This survey has been administered periodically since 1976 and the most recent version is the 2006 CWSS. In the survey, the EPA collected information on major operational and financial characteristics of CWSs. CWSs are PWSs that have at least 15 service connections used year-round or regularly serve at least 25 people year-round [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"a28flh1e4jm","properties": {"formattedCitation":"(USEPA, 2009b)","plainCitation":"(USEPA,

2009b)", "noteIndex":0}, "citationItems":[{"id":925, "uris":["http://zotero.org/groups/945096/items/QPD46SEA"], "uri":["http://zotero.org/groups/945096/items/QPD46SEA"], "itemData": {"id":925, "type":"article", "title": "2006 Community Water System Survey - Volume I:

Overview", "URL": "https://www.epa.gov/dwstandardsregulations/community-water-system-

<sup>&</sup>lt;sup>1</sup> [ HYPERLINK "https://ofmpub.epa.gov/apex/sfdw/f?p=108:1:::NO:1" ], data extracted August 14, 2018.

survey","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2009",2]]},"accessed":{"date-parts":[["2018",8,17]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

#### 3.2.2 Other Data and Information Used

The EPA's *Perchlorate Occurrence and Monitoring Report* is the source for national perchlorate occurrence data [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"uKDF3cH2","properties": {"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. This document describes the available data and analytical approaches that the EPA used to assess the baseline perchlorate occurrence in CWSs and NTNCWSs. The document also provides detailed estimates of perchlorate occurrence in drinking water.

Additional data and information used in analyzing the baseline conditions include data on the number of people in population groups that may be particularly susceptible to the health effects of perchlorate exposure. These data include U.S. Census Bureau population data and the National Center for Health Statistics annual live birth rates.

### 3.3 Baseline Profile

The proposed rule for perchlorate will apply to CWSs and NTNCWSs, and will not affect TNCWSs. Key characteristics of the affected systems are their service population size, type, ownership, and source water. This section provides details on these systems grouped by these characteristics.

#### 3.3.1 Number and Size of PWSs

The proposed rule will affect CWSs and NTNCWSs. The EPA uses a SDWIS data extract from August 2018 to determine the number of these water systems in the United States by system size category, source water, ownership, and system type. Both CWSs and NTNCWSs have more groundwater sources than surface water sources ([REF\_Ref523343598 \h]). CWSs make up approximately three-fourths (49,879 out of 67,497) of regulated water systems. NTNCWSs are more likely to be privately owned than CWSs ([REF\_Ref523343749 \h]) and be a small system ([REF\_Ref523343805 \h]). The exhibits show national totals and revised national totals that exclude systems in California and Massachusetts, which are the subset of regulated systems that would incur costs and benefits under the proposed rule. As noted in Section [REF\_Ref523347586 \r \h \\* MERGEFORMAT ], California and Massachusetts regulations supersede the proposed MCL. The systems in these states must comply with MCLs that are less than the proposed MCL or alternative MCL. Therefore, the EPA assumed that the proposed rule would not impose any incremental compliance costs on these systems.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Distribution of Affected Systems by Water Source and System Type

Water Source	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	Total Systems (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)	Total Systems (excluding CA and MA)
Groundwater	38,202	16,860	55,062	35,568	15,220	50,788
Surface water	11,677	758	12,435	10,634	654	11,288
Total	49,879	17,618	67,497	46,202	15,874	62,076

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"FXSyRdII","properties":{"formattedCitation":"(USEPA, 2018f)","plainCitation":"(USEPA, 2018f)","noteIndex":0},"citationItems":[{"id":923,"uris":["http://zotero.org/groups/945096/items/CW34PNAZ"],"uris":["http://zotero.org/groups/945096/items/CW34PNAZ"],"itemData":{"id":923,"type":"article","title":"Safe Drinking Water Information System Federal Reports

 $Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:::::", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}, "accessed": {"date-parts": [["2018",8,28]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].$ 

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Distribution of Affected Systems by Ownership and System Type

Ownership	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	Total Systems (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)	Total Systems (excluding CA and MA)
Private	22,865	12,266	35,131	20,742	11,153	31,895
Public/other	27,014	5,352	32,366	25,460	4,721	30,181
Total	49,879	17,618	67,497	46,202	15,874	62,076

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"gsHmnekE","properties": {"formattedCitation":"(USEPA, 2018f)", "plainCitation":"(USEPA, 2018f)", "noteIndex":0}, "citationItems": [{"id":923,"uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"],"uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "itemData": {"id":923, "type": "article", "title": "Safe Drinking Water Information System Federal Reports

Search","URL":"https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:::::","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]},"accessed":{"date-parts":[["2018",8,28]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Distribution of Affected Systems by Size and System Type

Service Population	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	Total Systems (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)	Total Systems (excluding CA and MA)
Small (≤ 10,000)	45,553	17,585	63,138	42,481	15,844	58,325
Large (> 10,000)	4,326	33	4,359	3,721	30	3,751
Total	49,879	17,618	67,497	46,202	15,874	62,076

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"ofFRpGY2","properties": {"formattedCitation":"(USEPA, 2018f)", "plainCitation":"(USEPA, 2018f)", "noteIndex":0}, "citationItems": [{"id":923,"uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"],"uri": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "itemData": {"id":923,"type": "article", "title": "Safe Drinking Water Information System Federal Reports
Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:::::", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}, "accessed": {"date-parts": [["2018",8,28]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

### 3.3.2 System Size and Population Served

SDWIS/FED also provides information on the retail population served by each system. Baseline health risks are a function of service populations. National population served estimates can also be disaggregated by source water ([ REF \_Ref523344075 \h ]), system type ([ REF \_Ref523344088 \h ]), and system size category ([ REF \_Ref523344097 \h ]).² CWS service populations are substantially larger than NTNCWS service populations. Among CWSs, publicly owned systems and large systems account for a large majority of the total population served. The exhibits include total national estimates as well as national estimates excluding the service populations of systems in California and Massachusetts. Systems in these states must meet state MCLs that are lower than the proposed MCL, so there will be no perchlorate reductions associated with the proposed rule and, therefore, no benefits.

<sup>&</sup>lt;sup>2</sup> The populations across CWSs and NTNCWSs are not additive because the population served by an NTNCWS can also be served by a CWS. For example, students at a school that is served by an NTNCWS may also consume water provided by a CWS at home.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Aggregate Service Population by Water Source and System Type

Water Source <sup>a</sup>	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)
Groundwater	89,232,288	5,216,951	80,367,122	4,804,167
Surface water	219,408,388	1,289,825	176,786,648	1,175,137
Total	308,640,676	6,506,776	257,153,770	5,979,304

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"4FPuHD93","properties": {"formattedCitation":"(USEPA, 2018f)","plainCitation":"(USEPA, 2018f)","noteIndex":0},"citationItems": [{"id":923,"uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"],"uri": ["http://zotero.org/groups/945096/items/CW34PNAZ"],"itemData": {"id":923,"type":"article","title":"Safe Drinking Water Information System Federal Reports

 $Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:::::", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}, "accessed": {"date-parts": [["2018",8,28]]}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].$ 

a. Forty-four systems serving 41,316 people (24 systems serving 37,635 people in included states) have unspecified source water; for the purpose of this analysis, the EPA assumed that these systems use groundwater.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Aggregate Service Population

by Ownership and System Type

Ownership	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)
Private	36,740,435	3,481,164	29,240,150	3,286,132
Public	271,900,241	3,025,612	227,913,620	2,693,172
Total	308,640,676	6,506,776	257,153,770	5,979,304

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"CyEAJ7uL","properties": {"formattedCitation":"(USEPA, 2018f)", "plainCitation":"(USEPA, 2018f)", "noteIndex":0}, "citationItems": [{"id":923, "uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "itemData": {"id":923, "type": "article", "title": "Safe Drinking Water Information System Federal Reports

 $Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200: ....", "author": [\{"literal": "USEPA"\}], "issued": \{"date-parts": [["2018"]]\}, "accessed": \{"date-parts": [["2018",8,28]]\}\}\}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"\}]. [ADDIN ZOTERO_ITEM CSL_CITATION]$ 

{"citationID":"I4ASsYq1","properties":{"formattedCitation":"(USEPA, 2018c)","plainCitation":"(USEPA, 2018c)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*\*-

\*\*\*","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Aggregate Service Population by Size and System Type

Service Population	CWSs (including CA and MA)	NTNCWSs (including CA and MA)	CWSs (excluding CA and MA)	NTNCWSs (excluding CA and MA)
Small (≤ 10,000)	53,121,502	5,573,773	53,121,502	5,573,773
Large (> 10,000)	255,519,174	933,003	204,032,268	405,531
Total	308,640,676	6,506,776	257,153,770	5,979,304

Source: SDWIS/FED 2018 data extract [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"qBR3UBpF","properties": {"formattedCitation":"(USEPA, 2018f)", "plainCitation":"(USEPA, 2018f)", "noteIndex":0}, "citationItems": [{"id":923, "uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "uris": ["http://zotero.org/groups/945096/items/CW34PNAZ"], "itemData": {"id":923, "type": "article", "title": "Safe Drinking Water Information System Federal Reports
Search", "URL": "https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:::::", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}, "accessed": {"date-parts": [["2018",8,28]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

The EPA did not incorporate population growth factors in the benefit and cost analysis. Although the overall national population will tend to grow over time, the Agency does not assume that the same growth rate applies individually to each CWS and NTNCWS affected by the rule, in particular those systems that need to implement treatment changes to meet the proposed MCL. Some systems may actually be experiencing population declines. Therefore, the analysis of benefits and costs will be based on current population estimates.

### 3.3.3 Consumption per Household

The Agency's analysis of distributional impacts include per-household cost estimates based on the assumption that systems pass along their incremental compliance costs through customer utility bills. Although systems use a variety of approaches to billing for water costs, the EPA assumed that the incremental costs would be billed at a constant dollar-per-thousand gallons. The incremental cost per household varies with the annual consumption rate. The 2006 CWSS report provides estimates of water consumption per residential connection. In Section 7, the EPA uses data from the 2006 CWSS report to describe water consumption per residential connection by system size category and ownership type.

#### 3.3.4 Production Profile

As noted above, the SDWIS/FED data contain estimates of service population. Treatment costs, however, are based on the volume of water treated per day, expressed as either design flow (a maximum daily treatment capacity) or average flow (the average daily production rate). The EPA uses equations to translate population served (*pop*) into design flow and average flow estimates [ ADDIN ZOTERO\_ITEM CSL\_CITATION

("First in a ID" "TO PK STATE" "To PK STATE" "TO PK STATE "

{"citationID":"mBK5bZJh","properties":{"formattedCitation":"(USEPA, 2000)","plainCitation":"(USEPA,

Average Flow = 
$$a_A(pop)^{b_A}$$
 ([ SEQ Eq. \\* ARABIC ])  
Design Flow =  $a_D(pop)^{b_D}$  ([ SEQ Eq. \\* ARABIC ])

2000)","noteIndex":0},"citationItems":[{"id":153,"uris":["http://zotero.org/groups/945096/items/DL7WBKI6"],"uri":["http://zotero.org/groups/945096/items/DL7WBKI6"],"itemData":{"id":15 3,"type":"article","title":"Geometries and Characteristics of Public Water Systems","publisher":"EPA 815-R-00-024","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2000"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The functional forms of the flow equations (measuring flow in thousands of gallons per day) are as follows:

Parameters  $a_A$ ,  $b_A$ ,  $a_D$ , and  $b_D$  are estimated regression function coefficients that define the relationship between the population served by the water system and flow. The point values for the parameters in the flow equations vary by source water ([REF\_Ref523346497 \h]). For the design flow, the EPA selected the maximum of either Equation 2 or two times the calculated average flow.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Flow Parameters by Water Source

Water Source	a <sub>A</sub>	b <sub>A</sub>	ao	b <sub>D</sub>
Groundwater	0.08575	1.05839	0.54992	0.95538
Surface water	0.14004	0.99703	0.59028	0.94573

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

### 3.4 Occurrence of Perchlorate

EPA's *Perchlorate Occurrence and Monitoring Report* provides estimates of the baseline perchlorate occurrence in PWSs [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"qUTie4VE","properties": {"formattedCitation":"(USEPA, 2018d)", "plainCitation":"(USEPA,

2018d)", "noteIndex":0}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/items/YERQWPRZ"], "uri":["http://zotero.org/groups/945096/items/YERQWPRZ"], "itemData": {"id":969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA \*\*\*-\*-\*\*, "author":[{"literal": "USEPA"}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. After reviewing the available data on perchlorate in drinking water, the EPA determined that the best nationally representative source is data from the UCMR 1.

This section summarizes the EPA's perchlorate occurrence analysis [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"UCKVzf9d","properties": {"formattedCitation":"(USEPA, 2018d)", "plainCitation":"(USEPA,

2018d)", "noteIndex":0}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/items/YERQWPRZ"], "uri":["http://zotero.org/groups/945096/items/YERQWPRZ"], "itemData":{"id":969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA \*\*\*-\*\*-\*\*, "author":[{"literal": "USEPA"}], "issued": {"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. Section [ REF \_Ref523347586 \r \h ] provides an overview of UCMR 1 and its perchlorate occurrence data, Section [ REF \_Ref523347601 \r \h ] summarizes the EPA's analysis of the UMCR 1 data, and Section [ REF \_Ref523347648 \r \h ] summarizes the national perchlorate occurrence estimates used in the cost and benefit analyses.

#### 3.4.1 Overview of UCMR 1 Data

The UCMR is a national drinking water monitoring program administered by the EPA. The UCMR 1 monitoring cycle included a census of all large CWSs and NTNCWSs (i.e., those serving more than 10,000 people), and a statistical sample of 800 small CWSs and NTNCWSs (i.e., those serving 10,000 people or fewer) [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"sOBUvFhD","properties": {"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*\*\_\*\*\*\*,"author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The UMCR1 cycle was from 2001 to 2005, and most of the data collection occurred between 2001 and 2003.

The UCMR 1 data comprise perchlorate monitoring samples from systems in all 50 states, the District of Columbia, the Tribal Nations, and 4 U.S. territories (Puerto Rico, Virgin Islands, Guam, and the Northern Mariana Islands). Response rates were high: 99.6 percent of small systems and 99.0 percent of large systems provided data [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"45nk8Bn9","properties":{"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*-\*\*\*","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Systems collected samples at each entry point to their customer distribution system.<sup>3</sup> Entry points are the point of compliance for the proposed rule and systems can have multiple entry points.

<sup>&</sup>lt;sup>3</sup> In response to comments on UCMR 1 data quality [ADDIN ZOTERO ITEM CSL CITATION {"citationID":"6awaaIIF", "properties": {"formattedCitation":"(U.S. Chamber of Commerce, 2012)", "plainCitation": "(U.S. Chamber of Commerce, 2012)", "noteIndex":3}, "citationItems":[{"id":152, "uris":["http://zotero.org/groups/945096/items/CMDYPKE9"], "uris":["http://zotero.org/groups/945096/items/groups/94509/items/groups/94509/items/groups/94509/items/groups/groups/94509/items/groups/groups/94509/items/groups/gr ":["http://zotero.org/groups/945096/items/CMDYPKE9"],"itemData":{"id":152,"type":"article","title":"Information Quality Guidelines (IQG) Request for Correction", "publisher": "Letter to USEPA. September.", "author": [{"family": "U.S. Chamber of Commerce", "given": ""}], "issued": {"dateparts":[["2012"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], the EPA reviewed the UCMR 1 data to identify instances where source water monitoring samples were accompanied by corresponding "downstream" entry point monitoring samples. In these instances, only the entry point samples provide the perchlorate concentration in water delivered to customers. Therefore, the 2013 version of the UCMR 1 dataset excludes these types of source water samples [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"88kDnU3n","properties": {"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA, 2018d)", "noteIndex":3}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/items/YERQWPRZ"], "uris":["http://zotero.org/groups/945096/items/yergwpraz"], "uris":["http://zotero.org/groups/945096/items/yergwpraz"], "uris":["http://zotero.org/groups/94509/item ri":["http://zotero.org/groups/945096/items/YEROWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorat e Occurrence and Monitoring Report", "publisher": "EPA \*\*\*-\*-\*\*-

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The sampling frequency varied by source water: four quarterly samples in a one-year period for
surface water systems, and two samples at least six months apart for groundwater systems [
ADDIN ZOTERO ITEM CSL CITATION
{"citationID": "8NSuLNK0", "properties": {"formattedCitation": "(USEPA,
2018d)", "plainCitation": "(USEPA,
2018d)", "noteIndex":0}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/item
s/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id"
:969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA
***-*-**,"author":[{"literal":"USEPA"}],"issued":{"date-
parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. The minimum reporting level (MRL) was 4
µg/L [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"k4YEKduX", "properties": {"formattedCitation":"(USEPA,
2018d)","plainCitation":"(USEPA,
2018d)", "noteIndex":0}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/item
s/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id"
:969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA
***-*-***,"author":[{"literal":"USEPA"}],"issued":{"date-
parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"}].
```

The summary statistics in [ REF \_Ref523358120 \h ] show total samples, entry points, and systems in the UCMR 1 perchlorate dataset. It also shows the number of reported perchlorate detections ( $\geq$  4  $\mu g/L$ ) along with the corresponding number of entry points and systems reporting those results.

\*\*\*","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: UCMR 1 Data Summary Statistics

Item	Small System Sample <sup>a</sup>	Large System Census	Total
Total samples	3,295	30,837	34,132
Measurements ≥ 4 µg/L	15	525	540
Total entry points	1,454	13,482	14,936
Measurements ≥ 4 µg/L	8	328	336
Total systems	797	3,068	3,865
Measurements ≥ 4 µg/L	8	141	149

Source: USEPA [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"UdoRmqu2","properties": {"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0}," citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*\*-","author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]

[ REF \_Ref523358221 \h ] shows the populations that correspond with the occurrence summary in [ REF \_Ref523358120 \h ]. The entry point population estimates reflect the assumption that system population is uniformly distributed across entry points (e.g., the entry point population for a system with two entry points is one-half the total system population).

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: UCMR 1 Data Service Population Summary Statistics

ltem	Small System Sample <sup>a</sup>	Large System Census	Total
Total entry point population	2,760,570	222,853,101	225,613,671
Measurements ≥ 4 µg/L	9,484	4,281,937	4,291,420
Total system population	2,760,570	222,853,101	225,613,671
Measurements ≥4 µg/L	13,483	16,159,082	16,172,565

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"LtcBtcnO", "properties": {"formattedCitation":"(2018d)", "plainCitation":"(2018d)", "noteIndex":0}, "citationItems": [{"id":969, "uris": ["http://zotero.org/groups/945096/items/YERQWPRZ"], "uri": ["http://zotero.org/groups/945096/items/YERQWPRZ"], "itemData": {"id":969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA \*\*\*-\*\*-\*\*\*, "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}}, "suppress-author": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]

Because the UCMR 1 data are well over a decade old, the EPA considered potential sources of uncertainty because of changes between current conditions and conditions at the time of data collection. One important change is the adoption of perchlorate drinking water limits in two states: Massachusetts adopted a drinking water standard for perchlorate of 2  $\mu$ g/L in 2006 [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"8DPpSrv3","properties": {"formattedCitation":"(Masschusetts Department of Environmental Protection (MassDEP), 2006)", "plainCitation": "(Masschusetts Department of

a. The small system values shown are sample results that have not been extrapolated to national estimates.

a. The small system values shown are sample results that have not been extrapolated to national estimates.

Environmental Protection (MassDEP), 2006)", "dontUpdate":true, "noteIndex":0}, "citationItems":[{"id":151, "uris":["http://zotero.org/gr oups/945096/items/9893MBZH"],"uri":["http://zotero.org/groups/945096/items/9893MBZH"],"i temData": {"id":151, "type": "personal communication", "title": "Letter to Public Water Suppliers concerning new perchlorate regulations", "URL": "https://www.mass.gov/lists/perchloratebackground-information-and-standards#perchlorate---final-standards-","author":[{"literal":"Massachusetts Department of Environmental Protection (MassDEP)"}],"issued":{"date-parts":[["2006"]]}}}],"schema":"https://github.com/citationstyle-language/schema/raw/master/csl-citation.json"}], and California promulgated a drinking water standard of 6 µg/L in 2007 [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"cfr6HNhg","properties":{"formattedCitation":"(California Department of Public Health, 2007)", "plainCitation": "(California Department of Public Health, 2007)","noteIndex":0},"citationItems":[{"id":150,"uris":["http://zotero.org/groups/945096/items/ RA45NKLQ"],"uri":["http://zotero.org/groups/945096/items/RA45NKLQ"],"itemData":{"id":15 0,"type":"personal communication","title":"State Adoption of a Perchlorate Standard", "URL": "https://www.waterboards.ca.gov/drinking\_water/certlic/drinkingwater/docum ents/perchlorate/AdoptionMemotoWaterSystems-10-2007.pdf", "author":[{"literal":"California Department of Public Health"}],"issued":{"dateparts":[["2007"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Systems in these states cannot exceed these limits, which are lower than the proposed federal MCL and alternative MCL. Therefore, any exceedances in the UCMR 1 data in these states overstate baseline occurrence and exposure under current conditions.

For the purpose of estimating the costs and benefits of the proposed rule, the EPA assumed that systems in California and Massachusetts comply with baseline perchlorate MCLs. Therefore, these systems will not incur incremental control costs to comply with the proposed rule. [ REF \_Ref525889374 \h ] summarizes the UCMR 1 data pursuant to this assumption, including information about sample measurements exceeding 18  $\mu g/L$  and 56  $\mu g/L$ .

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: UCMR 1 Data Summary Statistics, Excluding California and Massachusetts

Item	Small System Sample <sup>a</sup>	Large System Census	Total
Total samples	2,984	21,128	24,112
➤ Measurements ≥ 4 μg/L	13	206	219
➤ Measurements ≥ 18 μg/L	1	16	17
➤ Measurements ≥ 56 μg/L	0	2	2
Total entry points	1,327	9,118	10,445
≽ Measurements ≥ 4 μg/L	7	159	166
➤ Measurements ≥ 18 μg/L	1	16	17
➤ Measurements ≥ 56 µg/L	0	2	2
Total systems	737	2,591	3,328
≻ Measurements ≥ 4 μg/L	7	91	98
➤ Measurements ≥ 18 μg/L	1	14	15
≻ Measurements ≥ 56 μg/L	0	2	2
Total entry point population	2,537,888	183,525,431	186,063,319
≽ Measurements ≥ 4 μg/L	5,430	2,380,918	2,386,348
➤ Measurements ≥ 18 µg/L	2,155	618,406	620,561
➤ Measurements ≥ 56 µg/L	0	32,432	32,432
Total system population	2,537,888	183,525,431	186,063,319
≽ Measurements ≥ 4 μg/L	9,429	7,762,593	7,772,022
➤ Measurements ≥ 18 µg/L	4,309	696,871	701,180
≽ Measurements ≥ 56 μg/L	0	64,733	64,733

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"wyoUjFTz","properties":{"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0},"c itationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*,"author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

The age of the UCMR 1 data introduces additional sources of uncertainty. One is the effect of remediation efforts to reduce the sources of perchlorate in drinking water. The *Perchlorate Occurrence and Monitoring Report* [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"NaOQHInh","properties": {"formattedCitation":"(USEPA, 2018d)", "plainCitation":"(USEPA,

2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*\*","author":[{"literal":"USEPA"}],"issued":{"date-

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ] describes remediation efforts that have effectively reduced perchlorate levels in the Colorado River water from a range of 4  $\mu$ g/L to 9  $\mu$ g/L during the UCMR 1 data collection period to 1  $\mu$ g/L to 2  $\mu$ g/L after 2009. Systems that use the Colorado River as a water source may have lower concentrations at entry points than the values reported in the UCMR 1. Another type of change that has an uncertain impact on occurrence is the change in the universe of systems over time. Some systems operating during Preliminary draft for internal EPA review[PAGE \ \text{\* MERGEFORMAT }]

a. The values shown are sample results that have not been extrapolated to national estimates.

the UCMR 1 data collection period are now inactive. There are also new systems that were not operating during the UCMR 1 period. Such changes over time have an uncertain impact on perchlorate occurrence and exposure.

### 3.4.2 Summary of the EPA's Analysis of UCMR 1 Data

Preliminary draft for internal EPA review[ PAGE \\* MERGEFORMAT ]

Do not cite or quote

The analytical approach that the EPA used to estimate perchlorate occurrence and exposure is the approach that it used to evaluate the national contaminant occurrence analyses for the six-year reviews of NPDWRs [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"xqeVN1Mh", "properties": {"formattedCitation":"(USEPA, 2003b)", "plainCitation": "(USEPA, 2003b)", "noteIndex":0}, "citationItems":[{"id":1012, "uris":["http://zotero.org/groups/945096/ite ms/775HKG64"],"uri":["http://zotero.org/groups/945096/items/775HKG64"],"itemData":{"id":1 012,"type":"article","title":"Occurrence Estimation Methodology and Occurrence Findings Report for the Six-Year Review of Existing National Primary Drinking Water Regulations", "URL": "https://www.epa.gov/sites/production/files/2014-12/documents/815r03006.pdf", "note": "EPA-815-R-03-006", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2003",6]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}], the UCMR 1 data [ADDIN ZOTERO ITEM CSL CITATION {"citationID":"QNSihtwR", "properties": {"formattedCitation":"(USEPA, 2018d)", "plainCitation": "(USEPA, 2018d)", "noteIndex":0}, "citationItems":[{"id":969, "uris":["http://zotero.org/groups/945096/item s/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id" :969, "type": "article", "title": "Perchlorate Occurrence and Monitoring Report", "publisher": "EPA \*\*\*-\*-\*\*,"author":[{"literal":"USEPA"}],"issued":{"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}], and prior regulatory determinations [ADDIN ZOTERO ITEM CSL CITATION {"citationID":"fGy4pkzB","properties":{"formattedCitation":"(USEPA, 2003a; 2008a)", "plainCitation": "(USEPA, 2003a; 2008a)", "dontUpdate": true, "noteIndex": 0}, "citationItems": [{"id":1004, "uris": ["http://zotero.org/ groups/945096/items/WAS6UYUI"],"uri":["http://zotero.org/groups/945096/items/WAS6UYUI" ],"itemData": {"id":1004,"type":"article", "title":"Announcement of Regulatory Determinations for Priority Contaminants on the Drinking Water Contaminant Candidate List", "publisher": "68 FR 138", "shortTitle": "Federal Register", "language": "English", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2003"]]}}},{"id":934,"uris":["http://zotero.org/groups/945096/items/HBX88QM9"],"ur i":["http://zotero.org/groups/945096/items/HBX88OM9"], "itemData": {"id":934, "type": "articlejournal", "title": "Drinking water: Preliminary regulatory determination on perchlorate", "container-title": "Federal Register", "volume": "73", "issue": "198", "abstract": "SUMMARY: This action presents EPA's preliminary regulatory determination for perchlorate in accordance with the Safe Drinking Water Act (SDWA). The Agency has determined that a national primary drinking water regulation (NPDWR) for perchlorate would not present \"a meaningful opportunity for health risk reduction for persons served by public water systems." The SDWA requires EPA to make determinations every five years of whether to regulate at least five contaminants on the Contaminant Candidate

List (CCL). EPA included perchlorate on the first and second CCLs that were published in the Federal Register on March 2, 1998 and February 24, 2005. Most recently, EPA presented final regulatory determinations regarding 11 contaminants on the second CCL in a notice published in the Federal Register on July 30, 2008. In today's action, EPA presents supporting rationale and requests public comment on its preliminary regulatory determination for perchlorate. EPA will make a final regulatory determination for perchlorate after considering comments and information provided in the 30-day comment period following this notice. EPA plans to publish a health advisory for perchlorate at the time the Agency publishes its final regulatory determination to provide State and local public health officials with technical information that they may use in addressing local contamination.", "ISSN": "ISSN 0097-6326 EISSN 2167-2520", "shortTitle": "Federal Register", "journalAbbreviation": "Fed. Reg.", "language": "English", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2008"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. In each case, the data analysis process and presentation were peer-reviewed and subject to public and stakeholder review and comment. The approach relevant for the proposed perchlorate MCL and alternative MCL is the "Stage 1" analysis, for which an exceedance occurs if a single sample concentration is greater than a threshold such as the MCL.

EPA conducted the Stage 1 analysis at the entry-point level to derive estimates of benefits and costs that reflect the fact that a system may not have exceedances at all entry points and, therefore, benefits and costs should reflect a population smaller than the total system population.

### 3.4.3 Summary of National Perchlorate Occurrence

The EPA estimated Stage 1 occurrence for the proposed MCL of 56  $\mu$ g/L and the alternative MCL of 18  $\mu$ g/L. The results in [ REF \_Ref523449433 \h ] and [ REF \_Ref523448716 \h ] show the number of entry points and systems at which the highest perchlorate concentration exceeds these respective values, along with corresponding entry point service populations. Regardless of the threshold, there are exceedances at relatively few entry points or systems.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Expected Stage 1 Perchlorate Occurrence Greater than  $56 \mu g/L$ 

Affected Entity	Small Systems	Large Systems	Total Systems
Entry points	0	2	2
Population served	0	32,432	32,432
Water systems	0	2	2
Population served	0	64,733	64,733

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"JZbvtgrP","properties": {"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0},"cit ationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*,"author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Expected Stage 1 Perchlorate Occurrence Greater than 18 µg/L

Affected Entity	Small Systems <sup>a</sup>	Large Systems	Total Systems
Entry points	1	16	17
Population served	2,155	605,485	607,640
Water systems	1	14	15
Population served	4,309	696,871	701,180

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"0tmWzc6h","properties": {"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0}," citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*\*-","author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

a. These estimates reflect the sample data. The EPA also applied the statistical sampling weights to the results to extrapolate them to national results. The entry point at which a measurement exceeds  $18~\mu g/L$  is 1~of~20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the 6~UCMR~1 systems in the stratum. Overall, the stratum population served accounts for 1.32% of the national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers may be exposed to perchlorate greater than  $18~\mu g/L$ .

### 3.4.4 Number of Entry Points

The point of compliance for the proposed perchlorate MCL is the entry point to the distribution system, which can have one or more entry points. The number of entry points for CWSs are identified in the occurrence data. For systems that are not included in the occurrence data, the EPA assigned the number of entry points based on the population size classification of the system, calculated from the occurrence data. [ REF \_Ref524939687 \h ] summarizes the number of entry points by source and population served, based on the occurrence data.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Average Number of Entry Points per PWS Based on Population Served

Size Category	Maximum Population	Number of PWSs		Number of Entry Points per PWS		
		Occurrence Data	Total	Groundwater	Surface Water	
Very small	500	163	38,411	1.2	1.1	
Small	3,300	290	15,104	1.8	1.0	
Medium	10,000	344	4,810	2.7	1.2	
Large	1,000,000	2,369	3,734	4.6	2.1	
Very large	8,271,000	699	17	14.7	5.5	

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"XcGVSxov","properties": {"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0}," citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*,"author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

### 3.5 Sensitive Life Stages and Other Subpopulations

According to the 1996 Amendments to the SDWA, sensitive populations include "infants, children, pregnant women, the elderly, individuals with a history of serious illness, and other subpopulations" that may have a greater risk of adverse health effects due to exposure to contaminants in drinking water than the general population [SDWA Section 1458 (a)(1)]. In 2005, the EPA finalized guidance to differentiate health risk across life stages associated with development and growth (e.g., childhood age groups, pregnancy, and nursing) [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"ZWU3ZKsJ","properties":{"formattedCitation":"(USEPA, 2005a)", "plainCitation": "(USEPA, 2005a)","noteIndex":0},"citationItems":[{"id":149,"uris":["http://zotero.org/groups/945096/item s/YKZTCBGM"], "uri": ["http://zotero.org/groups/945096/items/YKZTCBGM"], "itemData": {"id ":149,"type":"article", "title":"Guidance on Selecting Age Groups for Monitoring and Assessing Childhood Exposures to Environmental Contaminants", "publisher": "EPA/630/P-03/003F", "author": [{"family": "USEPA", "given": ""}], "issued": {"dateparts":[["2005"]]}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. In this analysis, the EPA refers to subpopulations (as defined by the SDWA) that are associated with development and growth as life stages in recognition that all humans were once occupants of early developmental life stages. Chapter [ REF Ref523452255 \r \h ] identifies the sensitive life stages with respect to perchlorate ingestion. In addition, subpopulations are susceptible to perchlorate health effects due to low iodine intake and other diseases or conditions. These subpopulations include people with thyroid diseases and those with high levels of thiocyanate ingestion. There are approximately 20 million Americans with thyroid diseases [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"Z1WQPRxB","properties": {"formattedCitation":"(American Thyroid Association, undated)", "plainCitation": "(American Thyroid Association, undated)", "noteIndex":0}, "citationItems":[{"id":148, "uris":["http://zotero.org/groups/945096/ite ms/U683NFTS"],"uri":["http://zotero.org/groups/945096/items/U683NFTS"],"itemData":{"id":1 48,"type":"post-weblog","title":"General Information/Press Room", "URL": "https://www.thyroid.org/media-main/pressroom/", "author": [{"family": "American Thyroid Association", "given": ""}], "issued": {"literal": "undated"}}}], "schema": "https://github.com/citatio n-style-language/schema/raw/master/csl-citation.json"}].

### 4 Health Effects and Benefits Analysis

This chapter describes the quantifiable benefits of regulating perchlorate in drinking water, which mainly arise from reduced adverse health impacts. An overview of health effects associated with exposure to perchlorate is presented (Section [ REF \_Ref523327567 \n \h ]), which is followed by a presentation of the approach and results for quantifying the impact of reduced perchlorate exposure from drinking water based on the subsequent reduction in lost IQ points (Section [ REF \_Ref523327605 \n \h ]). Next, a brief discussion of additional potential benefits from reducing perchlorate exposure from drinking water that could not be quantified at this time is discussed (Section [ REF \_Ref523397403 \n \h ]). Because many of the potential health effects of perchlorate exposure cannot be accurately quantified, the estimated benefits associated only with avoidance of lost IQ are likely an underestimate of the total benefits of a reduction of perchlorate in drinking water.

### 4.1 Overview of the Health Effects of Perchlorate Exposure

The main target organ for perchlorate's toxicity is the thyroid gland. Specifically, perchlorate competes with iodide<sup>4</sup> for transport through the sodium-iodide symporter (NIS) into the thyroid gland. Transporting iodide into the thyroid gland is a necessary step in the production of thyroid hormones triiodothyronine (T3) and thyroxine (T4) [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "CNjLrLOp", "properties": {"formattedCitation": "(NRC, 2005)", "plainCitation": "(NRC, 2005)", "noteIndex":0}, "citationItems":[{"id":349, "uris":["http://zotero.org/groups/945096/items/ TN6HMC9D"], "uri": ["http://zotero.org/groups/945096/items/TN6HMC9D"], "itemData": {"id":3 49,"type":"book","title":"Health Implications of Perchlorate Ingestion","publisher":"National Academies Press", "publisher-place": "Washington, DC", "event-place": "Washington, DC", "author": [{"literal": "NRC"}], "issued": {"dateparts":[["2005"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Therefore, perchlorate may lead to decreases in levels of these hormones by decreasing iodide uptake [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "3rPSIFZZ", "properties": {"formattedCitation": "(Agency for Toxic Substances and Disease Registry (ATSDR), 2008; Blount et al., 2006; NRC, 2005; Steinmaus et al., 2013; 2016)", "plainCitation": "(Agency for Toxic Substances and Disease Registry (ATSDR), 2008; Blount et al., 2006; NRC, 2005; Steinmaus et al., 2013; 2016)", "noteIndex":0}, "citationItems":[{"id":428, "uris":["http://zotero.org/groups/945096/items/ UIANA947"], "uri": ["http://zotero.org/groups/945096/items/UIANA947"], "itemData": {"id": 428, "type":"bill", "title": "Toxicological Profile for Perchlorates", "author": [{"literal": "Agency for Toxic Substances and Disease Registry (ATSDR)"}],"issued":{"dateparts":[["2008"]]}},"suppress-author":true,"prefix":"Agency for Toxic Substances and Disease (ATSDR),"}, {"id":203,"uris":["http://zotero.org/groups/945096/items/UW4TFPNI"],"urit":["http://zotero.org/groups/945096/items/UW4TFPNI"],"urit":["http://zotero.org/ //zotero.org/groups/945096/items/UW4TFPNI"], "itemData": {"id":203, "type": "article-

journal", "title": "Urinary perchlorate and thyroid hormone levels in adolescent and adult men and

<sup>&</sup>lt;sup>4</sup> For the purposes of this report, the term "iodine" will be used to refer to dietary intake before entering the body. Once in the body, "iodide" will be used to refer to the ionic form.

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thiocyanate, and low iodine intake can all decrease iodide intake into the thyroid gland. This can
reduce thyroid hormone production since iodide is a key component of thyroid hormone.
Previous research has suggested that each of these factors
...","URL":"https://www.ncbi.nlm.nih.gov/pmc/articles/PMC3857960/","DOI":"10.1016/j.envres
.2013.01.005", "note": "PMID:
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867", "volume": "124", "issue": "6", "source": "PubMed", "abstract": "BACKGROUND: Findings
from national surveys suggest that everyone in the United States is exposed to perchlorate. At
high doses, perchlorate, thiocyanate, and nitrate inhibit iodide uptake into the thyroid and
decrease thyroid hormone production. Small changes in thyroid hormones during pregnancy,
including changes within normal reference ranges, have been linked to cognitive function
declines in the offspring.\nOBJECTIVES: We evaluated the potential effects of low
environmental exposures to perchlorate on thyroid function.\nMETHODS: Serum thyroid
hormones and anti-thyroid antibodies and urinary perchlorate, thiocyanate, nitrate, and iodide
concentrations were measured in 1,880 pregnant women from San Diego County, California,
during 2000-2003, a period when much of the area's water supply was contaminated from an
industrial plant with perchlorate at levels near the 2007 California regulatory standard of 6 µg/L.
Linear regression was used to evaluate associations between urinary perchlorate and serum
thyroid hormone concentrations in models adjusted for urinary creatinine and thiocyanate,
maternal age and education, ethnicity, and gestational age at serum collection.\nRESULTS: The
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Preliminary draft for internal EPA review[ PAGE \\* MERGEFORMAT ]

median urinary perchlorate concentration was 6.5 µg/L, about two times higher than in the general U.S.\nPOPULATION: Adjusted associations were identified between increasing log10 perchlorate and decreasing total thyroxine (T4) [regression coefficient ( $\beta$ ) = -0.70; 95% CI: -1.06, -0.34], decreasing free thyroxine (fT4) ( $\beta$  = -0.053; 95% CI: -0.092, -0.013), and increasing log10 thyroid-stimulating hormone ( $\beta = 0.071$ ; 95% CI: 0.008, 0.133).\nCONCLUSIONS: These results suggest that environmental perchlorate exposures may affect thyroid hormone production during pregnancy. This could have implications for public health given widespread perchlorate exposure and the importance of thyroid hormone in fetal neurodevelopment.\nCITATION: Steinmaus C, Pearl M, Kharrazi M, Blount BC, Miller MD, Pearce EN, Valentin-Blasini L, DeLorenze G, Hoofnagle AN, Liaw J. 2016. Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California. Environ Health Perspect 124:861-867: http://dx.doi.org/10.1289/ehp.1409614.","DOI":"10.1289/ehp.1409614","ISSN":"1552-9924", "note": "PMID: 26485730\nPMCID: PMC4892913", "journal Abbreviation": "Environ. Perspect.", "language": "eng", "author": [{"family": "Steinmaus", "given": "Craig"}, {"family": "Pearl ","given":"Michelle"},{"family":"Kharrazi","given":"Martin"},{"family":"Blount","given":"Benj amin C."}, {"family": "Miller", "given": "Mark D."}, {"family": "Pearce", "given": "Elizabeth N."}, {"family": "Valentin-Blasini", "given": "Liza", {"family": "DeLorenze", "given": "Gerald", {"family": "Hoofnagle", "given": "Gerald", {"family": "Gerald", given": "Gerald", {"family": "Gerald", given": "Gerald", given: "Gerald", give en":"Andrew N."}, {"family":"Liaw", "given": "Jane"}], "issued": {"dateparts":[["2016",6]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. A reduction in T3 and T4 may also result in an increase in the thyroid stimulating hormone (TSH) as this hormone acts on the thyroid to increase the uptake of iodine in an effort to create more T3 and T4 based on a regulatory feedback mechanism. This mode of action (MOA) is supported by several advisory committee reports on perchlorate [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "RePjCjFr", "properties": {"formattedCitation": "(NRC, 2005; SAB and USEPA, 2013)", "plainCitation": "(NRC, 2005; SAB and USEPA, 2013)","noteIndex":0},"citationItems":[{"id":349,"uris":["http://zotero.org/groups/945096/items/ TN6HMC9D"],"uri":["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData": {"id":3 49,"type":"book","title":"Health Implications of Perchlorate Ingestion","publisher":"National Academies Press", "publisher-place": "Washington, DC", "event-place": "Washington, DC", "author": [{"literal": "NRC"}], "issued": {"date-parts": [["2005"]]}}, "suppressauthor":true,"prefix":"NRC,"},{"id":1263,"uris":["http://zotero.org/groups/945096/items/3MNU 7GPK"], "uri": ["http://zotero.org/groups/945096/items/3MNU7GPK"], "itemData": {"id":1263, "ty pe":"article", "title": "SAB Advice on Approaches to Derive a Maximum Contaminant Level Goal for Perchlorate. EPA-SAB-13-004", "author": [{"literal": "SAB"}, {"family": "USEPA", "given": ""}], "issued": {"dateparts":[["2013"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ].

In 2005, the NRC evaluated the health implications of perchlorate exposure at the request of several Federal agencies. In their assessment, the NRC proposed an MOA that followed perchlorate exposure to reduced iodine intake, reduced serum T3 and T4, increased TSH, and subsequently to the adverse effect of hypothyroidism. Hypothyroidism is defined as increased TSH and decreased T4 to concentrations outside of their reference ranges. In support of the EPA Preliminary draft for internal EPA review[PAGE \ MERGEFORMAT ]

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Office of Water's analyses related to perchlorate, the EPA's SAB reviewed the current state of the science on perchlorate and thyroid physiology in 2013 and largely supported the NRC's MOA, but revised it to include hypothyroxinemia as an additional outcome to consider [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "CdXI5Naz", "properties": {"formattedCitation": "(SAB for the U.S. EPA & U.S. Environmental Protection Agency, 2013)", "plainCitation": "(SAB for the U.S. EPA & U.S. Environmental Protection Agency, 2013)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":"5Cn20MPv/gQaELIao","uris":[ "http://zotero.org/groups/945096/items/RVTWTXQR"],"uri":["http://zotero.org/groups/945096/i tems/RVTWTXQR"], "itemData": {"id":336, "type": "article", "title": "SAB Advice on Approaches to Derive a Maximum Contaminant Level Goal for Perchlorate. EPA-SAB-13-004", "author": [{"literal": "SAB for the U.S. EPA"}, {"family": "U.S. Environmental Protection Agency", "given": ""}], "issued": {"dateparts":[["2013"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} | ADDIN ZOTERO ITEM CSL CITATION {"citationID": "gRrbuWUo", "properties": {"formattedCitation": "(SAB and USEPA, 2013)", "plainCitation": "(SAB and USEPA, 2013)","noteIndex":0},"citationItems":[{"id":1263,"uris":["http://zotero.org/groups/945096/item s/3MNU7GPK"],"uri":["http://zotero.org/groups/945096/items/3MNU7GPK"],"itemData":{"id": 1263, "type": "article", "title": "SAB Advice on Approaches to Derive a Maximum Contaminant Level Goal for Perchlorate. EPA-SAB-13-004", "author": [{"literal": "SAB"}, {"family": "USEPA", "given": ""}], "issued": {"dateparts":[["2013"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Hypothyroxinemia is a condition that is defined as free T4 (fT4; T4 that is not bound to a protein) levels in the lower end of the reference range and normal TSH concentrations; it is not currently a medically treated condition. Both maternal hypothyroxinemia and maternal hypothyroidism during pregnancy have been associated with adverse neurodevelopmental outcomes (i.e., abnormal fetal and child

When the NRC conducted its review of the health effects of perchlorate exposure in 2005, it stated that although none of the steps that follow iodine uptake inhibition in the MOA have been clearly observed in humans, they are biologically plausible (these are demonstrated with the dotted arrows in [ REF \_Ref522634718 \h \\* MERGEFORMAT ]). Since the 2005 NRC report was published, however, several epidemiological studies have demonstrated an association between perchlorate exposure and changes in serum thyroid hormone levels [ ADDIN ZOTERO ITEM CSL CITATION

development). [ REF Ref522634718 \h ] presents both the NRC original and SAB modified

{"citationID":"qZgStztL","properties": {"formattedCitation":"(Blount et al., 2006; Steinmaus et al., 2013; 2016)", "plainCitation":"(Blount et al., 2006; Steinmaus et al., 2013; 2016)", "noteIndex":0}, "citationItems": [{"id":203, "uris": ["http://zotero.org/groups/945096/items/UW4TFPNI"], "uri": ["http://zotero.org/groups/945096/items/UW4TFPNI"], "itemData": {"id":203, "type": "article-journal", "title": "Urinary perchlorate and thyroid hormone levels in adolescent and adult men and women living in the United States", "container-title": "Environmental Health Perspectives", "page": "1865-

Preliminary draft for internal EPA review[ PAGE \\* MERGEFORMAT ]

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MOA of perchlorate's thyroid-related adverse health effects.

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thiocyanate, and low iodine intake can all decrease iodide intake into the thyroid gland. This can
reduce thyroid hormone production since iodide is a key component of thyroid hormone.
Previous research has suggested that each of these factors
...","URL":"https://www.ncbi.nlm.nih.gov/pmc/articles/PMC3857960/","DOI":"10.1016/j.envres
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867", "volume": "124", "issue": "6", "source": "PubMed", "abstract": "BACKGROUND: Findings
from national surveys suggest that everyone in the United States is exposed to perchlorate. At
high doses, perchlorate, thiocyanate, and nitrate inhibit iodide uptake into the thyroid and
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Linear regression was used to evaluate associations between urinary perchlorate and serum
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1.06, -0.34], decreasing free thyroxine (fT4) (\beta = -0.053; 95% CI: -0.092, -0.013), and increasing
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results suggest that environmental perchlorate exposures may affect thyroid hormone production
during pregnancy. This could have implications for public health given widespread perchlorate
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exposure and the importance of thyroid hormone in fetal neurodevelopment.\nCITATION: Steinmaus C, Pearl M, Kharrazi M, Blount BC, Miller MD, Pearce EN, Valentin-Blasini L, DeLorenze G, Hoofnagle AN, Liaw J. 2016. Thyroid hormones and moderate exposure to perchlorate during pregnancy in women in Southern California. Environ Health Perspect 124:861-867;

http://dx.doi.org/10.1289/ehp.1409614.","DOI":"10.1289/ehp.1409614","ISSN":"1552-9924","note":"PMID: 26485730\nPMCID: PMC4892913","journalAbbreviation":"Environ. Health

Perspect.","language":"eng","author":[{"family":"Steinmaus","given":"Craig"},{"family":"Pearl ","given":"Michelle"},{"family":"Kharrazi","given":"Martin"},{"family":"Blount","given":"Benj amin C."},{"family":"Miller","given":"Mark D."},{"family":"Pearce","given":"Elizabeth N."},{"family":"Valentin-

Blasini", "given": "Liza"}, {"family": "DeLorenze", "given": "Gerald"}, {"family": "Hoofnagle", "given": "Andrew N."}, {"family": "Liaw", "given": "Jane"}], "issued": {"date-parts": [["2016", 6]]}}}], "schema": "https://github.com/citation-style-

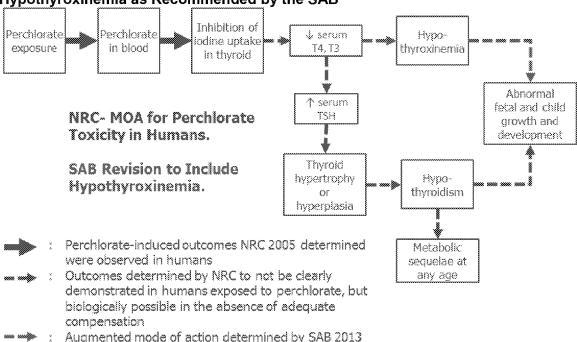
language/schema/raw/master/csl-citation.json"} ]. Additionally, Taylor et al. [ ADDIN ZOTERO\_ITEM CSL\_CITATION

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E."},{"family":"Iaylor","given":"P."},{"family":"Okosieme","given":"Onyebuchi E."},{"family":"Murphy","given":"Rhian"},{"family":"Hales","given":"C."},{"family":"Chiusan o","given":"Elisabetta"},{"family":"Maina","given":"Aldo"},{"family":"Joomun","given":"Moh amed"},{"family":"Bestwick","given":"Jonathan

P."}, {"family": "Smyth", "given": "Peter"}, {"family": "Paradice", "given": "Ruth"}], "issued": {"date-parts": [["2014"]]}}, "suppress-author": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] demonstrated the association between high maternal perchlorate exposure and risk of low IQ in offspring.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Modified Representation of NRC's Suggested MOA for Perchlorate Toxicity in Humans Indicating First Adverse Effect in the Continuum of Perchlorate Exposure to Effect, Revised to Include Hypothyroxinemia as Recommended by the SAB



As such, the EPA drew on the MOA proposed by the SAB to estimate benefits from reduced perchlorate exposure as a result of an NPDWR, which is now supported by several epidemiological studies [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"tvXUnpqQ","properties": {"formattedCitation":"(e.g., Blount et al., 2007; Steinmaus et al., 2016)", "plainCitation": "(e.g., Blount et al., 2007; Steinmaus et al., 2016)", "noteIndex":0}, "citationItems":[{"id":264, "uris":["http://zotero.org/groups/945096/items/ R9T25QKS"], "uri": ["http://zotero.org/groups/945096/items/R9T25QKS"], "itemData": {"id":264, "type": "article-journal", "title": "Perchlorate exposure of the US population, 2001– 2002", "container-title": "Journal of Exposure Science and Environmental Epidemiology", "page": "400-407", "volume": "17", "issue": "4", "source": "Google Scholar", "author": [{"family": "Blount", "given": "Benjamin C."}, {"family": "Valentin-Blasini", "given": "Liza" \, { "family": "Osterloh", "given": "John D."}, {"family":"Mauldin", "given":"Joshua P."}, {"family":"Pirkle", "given":"James L."}],"issued":{"date-parts":[["2007"]]}},"prefix":"e.g., "},{"id":211,"uris":["http://zotero.org/groups/945096/items/H4FH49VS"],"uri":["http://zotero.or g/groups/945096/items/H4FH49VS"],"itemData": {"id":211,"type": "articlejournal", "title": "Thyroid hormones and moderate exposure toperchlorate during pregnancy in women in southern California", "container-title": "Environmental Health Perspectives", "page": "861-867", "volume": "124", "issue": "6", "source": "PubMed", "abstract": "BACKGROUND: Findings from national surveys suggest that everyone in the United States is exposed to perchlorate. At

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http://dx.doi.org/10.1289/ehp.1409614.","DOI":"10.1289/ehp.1409614","ISSN":"1552-9924","note":"PMID: 26485730\nPMCID: PMC4892913","journalAbbreviation":"Environ. Health

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Blasini", "given": "Liza"}, {"family": "DeLorenze", "given": "Gerald"}, {"family": "Hoofnagle", "given": "Andrew N."}, {"family": "Liaw", "given": "Jane"}], "issued": {"date-parts": [["2016",6]]}}}], "schema": "https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. The EPA took a two-step approach to relate perchlorate exposure to changes in a quantifiable health endpoint. Specifically, the EPA related changes in perchlorate to changes in thyroid hormones in a mother using a BBDR model, and applied the resulting changes in thyroid hormones to changes in IQ in her offspring using doseresponse functions derived from the peer-reviewed literature [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"9vxa6n5e","properties":{"formattedCitation":"(USEPA, 2018e)","plainCitation":"(USEPA,

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parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

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language/schema/raw/master/csl-citation.json"} ]. This analysis is underpinned by the
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from animal and human studies indicates that thyroid hormone deficiency during early gestation
alters brain development. As schizophrenia is associated with prenatal brain insults and
premorbid cognitive deficits, we tested the a priori hypothesis that serologically defined maternal
thyroid deficiency during early gestation to mid-gestation is associated with schizophrenia in
offspring.\nMETHODS: The investigation is based on the Finnish Prenatal Study of
Schizophrenia, a nested case-control study that included archived maternal sera from virtually all
pregnancies since 1983 (N = > 1 million). We identified all offspring in the cohort with a
diagnosis of schizophrenia based on the national inpatient and outpatient register and matched
them on sex, date of birth, and residence in Finland at time of onset of the case to comparison
subjects (1:1) from the cohort. Maternal sera of 1010 case-control pairs were assessed for free
thyroxine, and sera of 948 case-control pairs were assessed for thyroid-stimulating
hormone.\nRESULTS: Maternal hypothyroxinemia (free thyroxine ≤10th percentile, normal
thyroid-stimulating hormone) was associated with an increased odds of schizophrenia (odds ratio
= 1.75, 95% confidence interval = 1.22-2.50, p = .002). When adjusted for maternal psychiatric
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history, province of birth, and maternal smoking during pregnancy, the association remained
significant (odds ratio = 1.70, 95% confidence interval = 1.13-2.55, p =
.010).\nCONCLUSIONS: In a large, national birth cohort, prospectively documented
hypothyroxinemia during early gestation to mid-gestation was associated with increased odds of
schizophrenia in offspring. This information can inform translational studies of maternal
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hypothyroidism in pregnant women is associated with a lower intelligence quotient in their
children. More recently, subtle decreases in maternal thyroid function have also been associated
with neurodevelopmental impairment in offspring. We tested the effect of hypothyroxinaemia
during early pregnancy on school performance.\nDESIGN: This was a longitudinal study that
included the data of 1196 mother-child pairs from the Amsterdam Born Children and Their
Development study.\nMETHODS: Maternal serum free thyroxine (T4) and TSH were obtained
at a median gestational age of 12.9 (interquartile range: 11.9-14.3) weeks. School performance
was assessed at age 5 years and based on scores obtained in arithmetic and language tests from
the national monitoring and evaluation system. Poor school performance was defined as a test
result <25th percentile and subnormal school performance as a result <50th percentile of the
norm population. To estimate the impact of possible non-response bias, we conducted inverse-
probability weighted analyses.\nRESULTS: Maternal hypothyroxinaemia (i.e., a maternal free
T4 in the lowest 10% of distribution) was associated with a 1.61 (95% CI: 1.05-2.47) -fold
increased odds of subnormal arithmetic performance after adjustment for confounders (P=0.03).
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However, the odds ratio dropped to 1.48 (95% CI: 0.94-2.32) after inverse-probability weighting (P=0.09). No such relations were found with TSH.\nCONCLUSIONS: Maternal hypothyroxinaemia at the end of the first trimester was associated with reduced performance in an arithmetic test, but not in a language test, in 5-year-old offspring. However, our results should be interpreted carefully because of possible non-response bias.","DOI":"10.1530/EJE-15-0397", "ISSN": "1479-683X", "note": "PMID: 26306579", "journal Abbreviation": "Eur. J. Endocrinol.", "language": "eng", "author": [{"family": "Noten", "given": "Anna M. E."},{"family":"Loomans","given":"Eva M."},{"family":"Vrijkotte","given":"Tanja G. M."},{"family":"Ven","given":"Peter M.","non-dropping-particle":"van de"},{"family":"Trotsenburg","given":"A. S. Paul","non-droppingparticle":"van"},{"family":"Rotteveel","given":"Joost"},{"family":"Eijsden","given":"Manon"," non-dropping-particle":"van"},{"family":"Finken","given":"Martijn J. J."}],"issued":{"dateparts":[["2015",11]]}}},{"id":198,"uris":["http://zotero.org/groups/945096/items/WKCD55DW" ],"uri":["http://zotero.org/groups/945096/items/WKCD55DW"],"itemData": {"id":198,"type":"art icle-journal", "title": "Maternal hypothyroxinaemia during early pregnancy and subsequent child development: a 3-year follow-up study", "container-title": "Clinical Endocrinology", "page": "282-288", "volume": "59", "issue": "3", "source": "PubMed", "abstract": "OBJECTIVE: To evaluate the impact of maternal hypothyroxinaemia during early gestation (fT4 below the lowest tenth percentile and TSH within the reference range: 0.15-2.0 mIU/l) on infant development, together with any subsequent changes in fT4 during gestation.\nDESIGN: A prospective 3-year follow-up study of pregnant women and their children up to the age of 2 years.\nMEASUREMENTS: Child development was assessed by means of the Bayley Scales of Infant Development in children of women with hypothyroxinaemia (fT4 below the tenth percentile at 12 weeks' gestation) at 12 weeks' gestation (cases), and in children of women with fT4 between the 50th and 90th percentiles at 12 weeks' gestation, matched for parity and gravidity (controls). Maternal thyroid function (fT4 and TSH) was assessed at 12, 24 and 32 weeks' gestation. The mental and motor function of 63 cases and 62 controls was compared at the age of 1 year, and of 57 cases and 58 controls at the age of 2 years.\nRESULTS: Children of women with hypothyroxinaemia at 12 weeks' gestation had delayed mental and motor function compared to controls: 10 index points on the mental scale (95% CI: 4.5-15 points, P = 0.003) and eight on the motor scale at the age of 1 year (95% CI: 2.3-12.8 points, P = 0.02), as well as eight index points on the mental (95% CI: 4-12 points, P = 0.02), and 10 on the motor scale (95% CI: 6-16 points, P = 0.005) at the age of 2 years. Children of hypothyroxinaemic women in whom the fT4 concentration was increased at 24 and 32 weeks' gestation had similar scores to controls, while in the controls, the developmental scores were not influenced by further declines in maternal fT4 at 24 and 32 weeks' gestation.\nCONCLUSIONS: Maternal hypothyroxinaemia during early gestation is an independent determinant of a delay in infant neurodevelopment. However, when fT4 concentrations increase during pregnancy in women who are hypothyroxinaemic during early gestation, infant development appears not to be adversely affected.", "ISSN": "0300-0664","note":"PMID: 12919150","shortTitle":"Maternal hypothyroxinaemia during early pregnancy and subsequent child development", "journal Abbreviation": "Clin. Endocrinol. (Oxf)", "language": "eng", "author": [{"family": "Pop", "given": "Victor J."}, {"family": "Brouwers", "given": "Evelien P."}, {"family": "Vader", "given": "Huib L."}, {"family": "Vulsma", "given": "Thomas"}, {"family": "Baar", "given": "Anneloes L.", "nondropping-particle": "van" }, { "family": "Vijlder", "given": "Jan J.", "non-droppingparticle":"de"}],"issued":{"date-

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thyroid function during early pregnancy is an important determinant of early fetal brain
development because the fetal thyroid is unable to produce any T4 before 12-14 weeks'
gestation. Overt maternal hypothyroidism as seen in severe iodine-deficient areas is associated
with severely impaired neurological development of the offspring. At present, it is not known
whether low free T4 (fT4) levels during pregnancy in healthy women from iodine sufficient
areas may affect fetal neurodevelopment.\nMETHODS: Neurodevelopment was assessed at 10
months of age in a cohort of 220 healthy children, born after uncomplicated pregnancies and
deliveries, using the Bayley Scales of Infant Development. Maternal TSH, fT4 and TPO
antibody status were assessed at 12 and 32 weeks' gestation. Maternal gestational fT4
concentration was defined as an independent parameter for child development.\nRESULTS:
Children of women with fT4 levels below the 5th (\leq 9.8 \text{ pmol/l}, n = 11) and 10th (\leq 10.4 \text{ pmol/l},
n = 22) percentiles at 12 weeks' gestation had significantly lower scores on the Bayley
Psychomotor Developmental Index (PDI) scale at 10 months of age, compared to children of
mothers with higher fT4 values (t test, mean difference: 14.1, 95% confidence interval (CI): 5.9-
22 and 7.4, 95% CI: 1.1-13.9, respectively). At 32 weeks' gestation, no significant differences
were found. In the group of women with the lowest 10th percentile fT4 concentrations at 12
weeks' gestation, a positive correlation was found between the mothers' fT4 concentration and
children's PDI scores (linear regression, R: 0.46, P = 0.03). After correction for confounding
variables, a fT4 concentration below the 10th percentile at 12 weeks' gestation was a significant
risk factor for impaired psychomotor development (RR): 5.8, 95% CI: 1.3-
12.6).\nCONCLUSIONS: Low maternal plasma fT4 concentrations during early pregnancy may
be an important risk factor for impaired infant development.", "ISSN": "0300-
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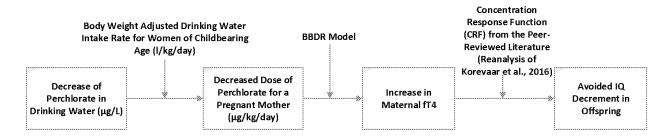
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P."},{"family":"Tiemeier","given":"Henning"}],"issued":{"date-
parts":[["2012"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. This evidence demonstrates that minor
perturbations in maternal thyroid hormones, in what would be considered a "normal range," can
result in altered neurodevelopmental outcomes in offspring, including decreased IQ. This multi-
step approach was evaluated by a peer-review panel in the context of setting an MCLG in
January 2018 and was well-received [ ADDIN ZOTERO ITEM CSL CITATION
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[ REF \_Ref522614073 \h ] summarizes the approach taken to evaluate the benefits of changes in IQ as a result of a reduction in perchlorate exposure due to the proposed MCLs. IQ is the only endpoint currently being monetized in this economic analysis. The EPA has deemed perchlorate not likely to be carcinogenic to humans [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"UtjLzmMG","properties": {"formattedCitation":"(2005b)","plainCitation":"(2005b)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":1007,"uris":["http://zotero.org/groups/945096/items/LHANJBR6"],"item Data": {"id":1007,"type":"article","title":"Integrated Risk Information System (IRIS) Chemical Assessment Summary: Perchlorate (ClO4-) and Perchlorate Salts","publisher":"USEPA National Center for Environmental Assessment","author":[{"literal":"USEPA"}],"issued": {"dateparts":[["2005"]]}}, "suppress-author":true}], "schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], and subsequently no cancer endpoints have

been assessed. Additional benefits that may arise from reducing perchlorate exposure but could not be monetized are discussed in Section [ REF Ref523397403 \n \h ].

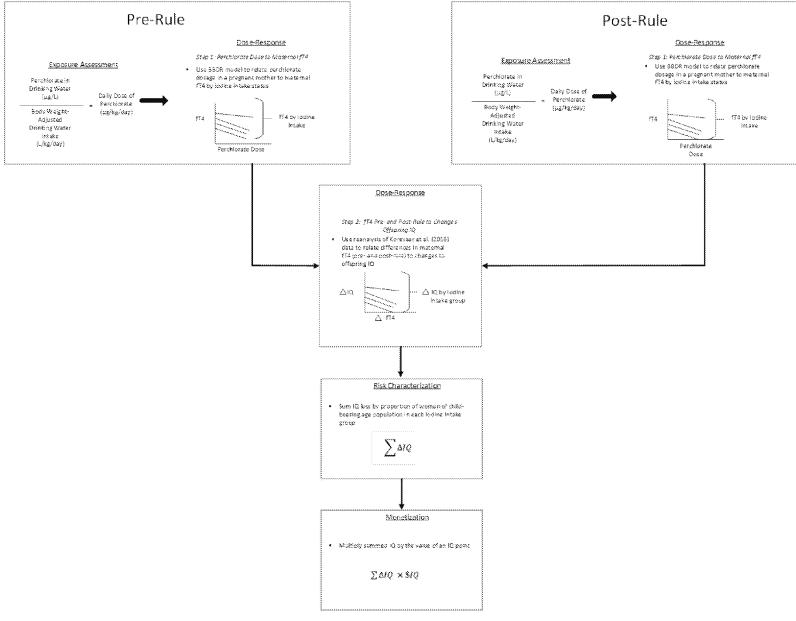
Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Overview of the Approach for Estimating Changes in IQ in Offspring from Reduced Perchlorate Exposure in a Pregnant Mother



### 4.2 Quantitative Benefits Assessment

As outlined in [REF\_Ref522614073 \h ], the EPA is utilizing a multi-step process to evaluate the human health-related impacts of perchlorate. [REF\_Ref522697939 \h ][REF\_Ref523393377 \h ] elaborates more on this process. The figure demonstrates that an analysis to determine pre- and post-rule exposures to perchlorate is conducted first, followed by a two-step dose-response analysis that relates pre- and post-rule perchlorate exposure to pre- and post-rule maternal fT4 levels, which is then translated to a change in offspring IQ and monetized. The dose-response relationship between perchlorate exposure and maternal fT4 is dependent on maternal iodine intake status, and as such this analysis is repeated for several categories of iodine intake. Ultimately, the change in IQ for all iodine intake groups is averaged based on the proportion of individuals in each iodine intake category. The total avoided IQ decrements on an annual basis is estimated over a 35-year timeframe used for the cost analysis described in Section 5. Additional details on each step of the benefits analysis are in subsequent sections.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Overview of Process to Estimate Benefits from Perchlorate Exposure



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### 4.2.1 Exposure Assessment

To assess the benefits of reduced exposure to perchlorate in drinking water, the EPA must first define who is exposed and at what levels. Additionally, the EPA must also define and enumerate the population experiencing quantitative benefits due to the reduction in exposure. Therefore in this section the key inputs to understand who will be exposed, who will experience quantitative benefits, and the amount of perchlorate exposure in both pre- and post-rule scenarios are outlined.

### 4.2.1.1 Perchlorate Occurrence in U.S. Drinking Water

Section [ REF \_Ref523404705 \n \h ] reports the current occurrence of perchlorate. As a reminder, [ REF \_Ref523449433 \h ] and [ REF \_Ref523448716 \h ] are repeated as [ REF \_Ref523404751 \h ] and [ REF \_Ref523404750 \h ]. These tables summarize the systems impacted when considering maximum perchlorate concentrations at 56  $\mu$ g/L or 18  $\mu$ g/L. The concentrations of perchlorate in these systems are the pre-rule exposure concentrations.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Expected Stage 1 Perchlorate Occurrence Greater than  $56 \mu g/L$ 

Affected Entity	Small Systems	Large Systems	Total Systems
Entry points	0	2	2
Population served	0	32,432	32,432
Water systems	0	2	2
Population served	0	64,733	64,733

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

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Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Expected Stage 1 Perchlorate Occurrence Greater than 18  $\mu$ g/L

Affected Entity	Small Systems	Large Systems	Total Systems
Entry points	1	16	17
Population served	2,155	618,406	620,560
Water systems	1	14	15
Population served	4,309	696,871	701,180

Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"sYgCheHc","properties": {"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0}," citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData": {"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*,"author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

For post-rule, the EPA assumed that any system exceeding an MCL will include a safety factor of 80 percent when implementing treatment (i.e., treat to  $0.8 \times MCL$ ) to avoid future MCL exceedances. Therefore, when considering systems that are at or above 18  $\mu$ g/L, the system is assumed to design and implement treatment to achieve a target of 14.4  $\mu$ g/L. If the MCL is 56  $\mu$ g/L, the system is assumed to achieve a target of 44.8  $\mu$ g/L.

### 4.2.1.2 Drinking Water Consumption Rates

The EPA Exposure Factors Handbook (EFH) [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"EmVsXCy9","properties": {"formattedCitation":"(EPA, 2011b)","plainCitation":"(EPA,

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 $(EPA/600/R","URL":"https://cfpub.epa.gov/ncea/risk/recordisplay.cfm?deid=236252","language ":"en","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2011"]]},"accessed":{"date-parts":[["2017",5,3]]}},"suppress-$ 

author":true,"prefix":"EPA,"}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] reports mean, 90th, and 95th percentile bodyweight-adjusted drinking water intakes for pregnant, lactating, and non-pregnant non-lactating women of childbearing age. These figures are reported from the Kahn and Stralka [ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"rE57YWmU","properties": {"formattedCitation":"(2008)","plainCitation":"(2008) ","noteIndex":0},"citationItems": [{"id":188,"uris": ["http://zotero.org/groups/945096/items/JZJAVK6M"],"uri": ["http://zotero.org/groups/945096/items/JZJAVK6M"],"itemData": {"id":188,"type":"article-journal", "title": "Estimates of water ingestion for women in pregnant, lactating, and non-pregnant and non-lactating child-bearing age groups based on USDA's 1994–96, 1998 continuing survey of food intake by individuals", "container-title": "Human and Ecological Risk Assessment: An International Journal", "page": "1273-

1290", "volume": "14", "issue": "6", "DOI": "10.1080/10807030802494618", "ISSN": "1080-7039", "shortTitle": "Estimates of water ingestion for women in pregnant, lactating, and non-pregnant and non-lactating child-bearing age groups based on USDA's 1994–96, 1998 continuing survey of food intake by individuals", "author": [{"family": "Kahn", "given": "Henry

<sup>&</sup>lt;sup>5</sup> Safety factors are commonly used, but the values can vary. For example, the 10 States Standards [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"zykqAnZL","properties": {"formattedCitation":"(Great Lakes - Upper Mississippi River Board of State and Provincial and Public Health and Environmental Managers, 2012)", "plainCitation":"(Great Lakes - Upper Mississippi River Board of State and Provincial and Public Health and Environmental Managers,

<sup>2012)&</sup>quot;,"noteIndex":5},"citationItems":[{"id":1262,"uris":["http://zotero.org/groups/945096/items/GNKJ3I8R"],"uri ":["http://zotero.org/groups/945096/items/GNKJ3I8R"],"itemData":{"id":1262,"type":"article","title":"Recommende d Standards for Water Works, 2012 ed.","author":[{"literal":"Great Lakes - Upper Mississippi River Board of State and Provincial"}, {"literal":"Public Health and Environmental Managers"}],"issued":{"dateparts":[["2012"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] recommends design targets below MCLs, as low as 50 percent for arsenic treatment. The EPA selected 80 percent, which allows a 25 percent excursion from design performance while still achieving compliance.

D."}, {"family": "Stralka", "given": "Kathleen"}], "issued": {"dateparts":[["2008",11,21]]}}, "suppress-author":true}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] study, which is based on the Continuing Survey of Food Intakes by Individuals (CSFII) data collected from 1994 to 1996 and 1998. This study estimated bodyweight-adjusted drinking water intake rates for direct and indirect community water ingestion, as well as for direct and indirect water intake from all sources, on both a per-capita and consumers-only basis. As the potential MCLs are specific for the offspring of pregnant women consuming community drinking water, the EPA chose to focus on community drinking water intake estimates on a consumers-only basis as potential inputs. These estimates are reported in Table 3-81 in the EFH [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"43y7oVIa","properties":{"formattedCitation":"(USEPA. 2011b)", "plainCitation": "(USEPA, 2011b)", "noteIndex":0}, "citationItems":[{"id":162, "uris":["http://zotero.org/groups/945096/item s/8AZKGTHH"],"uri":["http://zotero.org/groups/945096/items/8AZKGTHH"],"itemData":{"id": 162, "type": "report", "title": "Exposure Factors Handbook 2011 Edition (Final Report)", "page": "Chapter 8", "abstract": "EPA announced the release of the final report, <i>Exposure Factors Handbook: 2011 Edition (EPA/600/R", "URL": "https://cfpub.epa.gov/ncea/risk/recordisplay.cfm?deid=236252", "language ":"en","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2011"]]},"accessed":{"dateparts":[["2017",5,3]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] and are reproduced in [ REF Ref522614193

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Consumers-Only Estimated Direct and Indirect Community Water Ingestion Rates from Kahn and Stralka (2008) (L/kg/day)

Women Categories	Sample Size	Mean	90th Percentile	95th Percentile
Pregnant <sup>a</sup>	65	0.014	0.033	0.043
Lactating <sup>a</sup>	33	0.026	0.054	0.055
Non-pregnant, non-lactating, 15 to 44 years of age	2,028	0.015	0.032	0.038

a. The sample size does not meet minimum reporting requirements to make statistically reliable estimates as described in the *Third Report on Nutrition Monitoring in the United States*, which covers 1994–1996 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"159hN8ZE","properties": {"formattedCitation":"(FASEB/LSRO, 1995)", "plainCitation":"(FASEB/LSRO,

The EPA chose to use exposure factor estimates specific to women of childbearing age in conducting this benefits assessment for the proposed MCL (i.e., non-pregnant, non-lactating, 15–44 years of age). This determination was reached as the calibration of the BBDR model was performed using a population of women of childbearing age from the National Health and

\h ].

<sup>1995)&</sup>quot;, "noteIndex":0}, "citationItems":[{"id":1040, "uris":["http://zotero.org/groups/945096/items/MN5AETQ4"], "uri":["http://zotero.org/groups/945096/items/MN5AETQ4"], "itemData":{"id":1040, "type": "report", "title": "Third report on nutrition monitoring in the United States", "publisher-place": "Washington, DC", "event-place": "Washington,

DC","URL":"https://www.cdc.gov/nchs/data/misc/nutri95\_1acc.pdf","author":[{"literal":"FASEB/LSRO"}],"issued ":{"date-parts":[["1995"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Nutrition Examination Survey (NHANES). The EPA has chosen to apply the 90th percentile drinking water intake rate in order to remain consistent with the assumption made in supporting the derivation of the MCLG. Thus, according to Table 3-81 in the EFH [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"3qPpsryF","properties":{"formattedCitation":"(EPA, 2011b)","plainCitation":"(EPA,

2011b)", "noteIndex":0}, "citationItems":[{"id":162, "uris":["http://zotero.org/groups/945096/items/8AZKGTHH"], "uri":["http://zotero.org/groups/945096/items/8AZKGTHH"], "itemData":{"id":162, "type":"report", "title":"Exposure Factors Handbook 2011 Edition (Final

Report)", "page": "Chapter 8", "abstract": "EPA announced the release of the final report, <i>Exposure Factors Handbook: 2011 Edition

 $(EPA/600/R","URL":"https://cfpub.epa.gov/ncea/risk/recordisplay.cfm?deid=236252","language ":"en","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2011"]]},"accessed":{"date-parts":[["2017",5,3]]}},"suppress-$ 

author":true,"prefix":"EPA,"}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], the bodyweight-adjusted drinking water intake rate for women of childbearing age is 0.032 L/kg/day.

### 4.2.2 Daily Dose of Perchlorate

By combining the information on the concentration of perchlorate in drinking water from UCMR 1 and the specified drinking water intake rate, it is possible to estimate the daily dose of perchlorate being consumed at each PWS. This dose is estimated by the following equation:

$$D = \frac{DW_{ClO_4^-}}{DWI},$$

where:

D = dose of perchlorate ( $\mu$ g/kg/day)  $DW_{ClO_4^-}$  = the concentration of perchlorate in drinking water in  $\mu$ g/L DWI = the bodyweight-adjusted drinking water intake rate in L/kg/day.

This dose is calculated based on both the pre- and post-rule concentration of perchlorate in drinking water.

### 4.2.3 Population Impacted

The population impacted by the rule for which benefits can be quantified is specific to live births from mothers who were served by a PWS with perchlorate concentrations at or above the potential MCLs. To determine the nationwide population of children that will experience a quantifiable benefit of avoided IQ decrements from reducing maternal perchlorate exposure during pregnancy, the EPA first estimated the total population being served by systems at or above the MCL based on data from UCMR 1. The EPA then multiplied the total population served for each effected PWS by the proportion of women of childbearing age (aged 15–44) in the United States, which is 19.7 percent [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"0Is16I8T","properties": {"formattedCitation":"(U.S. Census Bureau, 2017b)", "plainCitation":"(U.S. Census Bureau, 2017b)", "noteIndex":0}, "citationItems": [{"id":945, "uris": ["http://zotero.org/groups/945096/item

s/ZM7S6H44"],"uri":["http://zotero.org/groups/945096/items/ZM7S6H44"],"itemData":{"id":94 5,"type":"article", "title": "Annual estimates of the resident population by single year of age and sex for the United States: April 1, 2010 to July 1, 2016.", "URL": "https://www.census.gov/data/datasets/2016/demo/popest/nationdetail.html#ds", "author":[{"literal":"U.S. Census Bureau"}], "issued": {"dateparts":[["2017"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}]. The number of women of child-bearing age for each PWS was then multiplied by the annual number of live births in the United States, or 62 births per 1,000 women (6.2 percent) [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"oSRGsU5q","properties": {"formattedCitation":"(Martin et al., 2017)", "plainCitation": "(Martin et al., 2017)", "noteIndex":0}, "citationItems":[{"id":186, "uris":["http://zotero.org/groups/945096/items/ MY6LPDKD"], "uri": ["http://zotero.org/groups/945096/items/MY6LPDKD"], "itemData": {"id":1 86,"type": "article", "title": "Births in the United States, 2016. NCHS Data Brief No. 287","URL":"https://www.cdc.gov/nchs/data/databriefs/db287.pdf","author":[{"family":"Martin" ","given":"J.A.", {"family":"Hamilton", "given":"B.E."}, {"family":"Osterman", "given":"M.J.K"}] ,"issued":{"date-parts":[["2017"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. The resulting impacted population characterized for the monetized benefits is summarized in [ REF Ref522698332 \h ].

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Size of the Total U.S. Population, Women of Child-Bearing Age and Live Births Exposed at or above the Potential MCLs to Perchlorate in Drinking Water

	Total U.S. Population			
Potential MCL (µg/L)	Total at or above MCL (A)	Women of Child-Bearing Age at or above the MCL (B = A x 19.7%)	pe Annual Live Births Occurring at or above the MCL (C = B x 6.2%)	
56	32,432	6,385	396	
18 (UCMR 1)	620,560	122,168	7,574	
18 (national) <sup>a</sup>	659,547	129,843	8,050	

a. EPA applied statistical sampling weights to the results to extrapolate small system results to national results. The entry point at which a measurement exceeds  $18 \mu g/L$  is 1 of 20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the 6 UCMR 1 systems in the stratum (40,574). Currently, the stratum population of 774,780 accounts for 1.32% of the 58.7 million national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers (approximately 41,100) may be exposed to perchlorate greater than  $18 \mu g/L$ . The EPA assumed that this population would incur benefits equivalent to the sampled entry point's population.

These resulting populations were then further divided into the proportion of live births born to women of each level of iodine intake, given that the dose-response relationship between perchlorate and altered maternal fT4 is dependent on the level of daily iodine intake (see Section [ REF \_Ref522534193 \r \h \\* MERGEFORMAT ]).

The EPA estimated the distribution of iodine intakes of non-pregnant women of childbearing age (20–44) using data from NHANES 2011–2014 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"JsLCoBF1","properties": {"formattedCitation":"(CDC and NCHS, 2011; 2013)","plainCitation":"(CDC and NCHS, 2011;

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2013)","noteIndex":0},"citationItems":[{"id":261,"uris":["http://zotero.org/groups/945096/items/
G49EFPQ4"], "uri": ["http://zotero.org/groups/945096/items/G49EFPQ4"], "itemData": {"id":261,
"type": "report", "title": "National Health and Nutrition Examination Survey
Data", "publisher": "U.S. Department of Health and Human Services, Centers for Disease Control
and Prevention", "publisher-place": "Hyattsville, MD", "event-place": "Hyattsville,
MD", "URL": "https://wwwn.cdc.gov/nchs/nhanes/Search/Nhanes11 12.aspx", "author": [{"literal"
:"CDC"},{"literal":"NCHS"}],"issued":{"date-
parts":[["2011"]],"season":"2012"}}},{"id":"5Cn20MPv/ji5wASja","uris":["http://zotero.org/gro
ups/635290/items/H5BG78PS"],"uri":["http://zotero.org/groups/635290/items/H5BG78PS"],"ite
mData": {"id":1260, "type": "report", "title": "National Health and Nutrition Examination Survey
Data", "publisher": "U.S. Department of Health and Human Services, Centers for Disease Control
and Prevention", "publisher-place": "Hyattsville, MD", "event-place": "Hyattsville,
MD", "URL": "https://wwwn.cdc.gov/nchs/nhanes/Search/Nhanes13 14.aspx", "author": [{"literal"
:"CDC"},{"literal":"NCHS"}],"issued":{"date-
parts":[["2013"]],"season":"2014"}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. This was accomplished by first taking each
participant's urinary iodine sample concentration from the NHANES and implementing the data
smoothing technique outlined in Pleil and Sobus [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID": "8Adyiuja", "properties": {"formattedCitation": "(2016)", "plainCitation": "(2016)", "n
oteIndex":0}, "citationItems":[{"id":269, "uris":["http://zotero.org/groups/945096/items/W6ZWA
4ES"], "uri": ["http://zotero.org/groups/945096/items/W6ZWA4ES"], "itemData": {"id":269, "type"
:"article-journal", "title": "Estimating central tendency from a single spot measure: A closed-form
solution for lognormally distributed biomarker data for risk assessment at the individual
level", "container-title": "Journal of Toxicology and Environmental Health, Part A", "page": "837-
847","volume":"79","issue":"18","source":"CrossRef","DOI":"10.1080/15287394.2016.1193108
","ISSN":"1528-7394, 1087-2620","shortTitle":"Estimating central tendency from a single spot
measure", "language": "en", "author": [{"family": "Pleil", "given": "Joachim
D."},{"family":"Sobus","given":"Jon R."}],"issued":{"date-parts":[["2016",9,16]]}},"suppress-
author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-
citation.json" \ ]. This technique accounts for the relative representativeness of a single spot
measurement of iodine by predicting an intra-individual geometric mean concentration for each
participant. The EPA applied the average of the published intraclass correlation coefficients
(ICCs) for urinary spot measurements of iodine from Amouzegar and Azizi [ ADDIN
ZOTERO ITEM CSL CITATION
{"citationID":"EXM8M7pr","properties": {"formattedCitation":"(2013)", "plainCitation":"(2013)"
"noteIndex":0}, "citationItems":[{"id":268, "uris":["http://zotero.org/groups/945096/items/4894G,"
8UH"], "uri": ["http://zotero.org/groups/945096/items/4894G8UH"], "itemData": {"id":268, "type":
"article-journal", "title": "Variations of urinary iodine during the first trimester of pregnancy in an
iodine-replete area. Comparison with non-pregnant women", "container-
title":"Hormones","page":"111-18","volume":"12","issue":"1","source":"Google
Scholar", "author": [{"family": "Amouzegar", "given": "Atieh"}, {"family": "Azizi", "given": "Fereido
un"}],"issued":{"date-parts":[["2013"]]}},"suppress-
author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-
citation.json"}] as the basis for this smoothing technique. The EPA then multiplied each
participant's predicted geometric mean urinary iodine concentration by their estimated daily
urinary output based on their NHANES measured urinary output data. When urinary output data
```

were not available, each missing value was replaced with median urinary output, or 1.33 L/day. The EPA then estimated the proportion of the population of non-pregnant women of childbearing age that fell into each daily iodine intake rate category, as summarized in [ REF \_Ref522614353 \h ].

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Proportion of Population Based on Maternal Iodine Intake Status

lodine Intake Range (μg/day) Used for Benefits Analysis	Proportion of the Population
0 to < 55	7.14%
55 to < 60	2.15%
60 to < 65	1.06%
65 to < 70	1.86%
70 to < 75	1.31%
75 to < 80	3.10%
80 to < 85	2.62%
85 to < 90	1.20%
90 to < 95	1.83%
95 to < 100	2.94%
100 to < 125	13.56%
125 to < 150	9.08%
150 to < 170	10.31%
170 to < 300	24.47%
≥ 300	17.36%

#### 4.2.4 Dose-Response

The process of connecting maternal perchlorate exposure to offspring IQ decrements requires two steps. The first step relates perchlorate exposure with changes in maternal fT4, and the second step relates the changes in maternal fT4 from the pre- to post-rule to changes in offspring IQ. Each step is described more thoroughly in the subsequent sections.

#### 4.2.4.1 Step 1: Perchlorate Exposure to Changes in Maternal fT4

The EPA developed a BBDR model to describe the impact of perchlorate exposure on maternal fT4 levels in early pregnancy [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"zdqBNCcO","properties": {"formattedCitation":"(USEPA, 2018e)","plainCitation":"(USEPA, 2018e)","noteIndex":0},"citationItems":[{"id":926,"uris":["http://zotero.org/groups/945096/items/ANBBTDKU"],"uri":["http://zotero.org/groups/945096/items/ANBBTDKU"],"itemData": {"id ":926,"type":"report","title":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","shortTitle":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. This model has two main components: (1) a pharmacokinetic model for perchlorate and iodide, which describes chemical absorption,

distribution, metabolism, and excretion of these two anions; and (2) a pharmacodynamic (PD) model, which describes the combined effect of varying perchlorate and iodide blood concentrations on the thyroidal uptake of iodide and subsequent production of thyroid hormones, most significantly T4. The pharmacokinetic portion contains a physiological description (e.g., organ volumes, blood flows) and chemical-specific information (e.g., partition coefficients; volume of distribution; rate constants for transport, metabolism, and elimination) that enable a prediction of perchlorate and iodide internal concentration at the critical target [i.e., thyroidal NIS in association with a particular exposure scenario (route of exposure, age, dose level)]. This portion of the model is similar to other physiologically based pharmacokinetic (PBPK) models and for perchlorate is simplified by the absence of metabolism. The PD portion of the model uses this internal concentration to simulate how the chemical will act within a known mechanism of action to perturb host systems and lead to a toxic effect. Thus, BBDR modeling attempts to predict the internal dose of a chemical associated with a particular exposure scenario, and the perturbation this internal dose can have on host systems.<sup>6</sup>

The BBDR model predicts serum thyroid hormone levels at:

- Specific gestational weeks. The EPA data that connect maternal fT4 concentrations to offspring IQ have a mean week of fT4 collection of approximately 13 gestational weeks (GWs). Subsequently in this benefits analysis, GW 13 data from the BBDR model have been used.
- A specific TSH feedback loop strength. As the benefits of reducing perchlorate exposure
  may extend to the entire population exposed, the BBDR model was run with the TSH
  feedback loop adjusted to its median level.
- Specific levels of iodine intake. Given that the dose-response relationship between perchlorate and fT4 is dependent on daily iodine intake concentrations, the EPA derived dose-response relationships based on BBDR model output for 13 different levels of iodine intake spanning from 50 μg/day to 300 μg/day.

To derive the dose-response functions for each level of daily iodine intake, the EPA first converted each dose of perchlorate that was input into the BBDR model to its equivalent drinking water concentration using the defined drinking water intake rate (see Section [ REF \_Ref522698474 \n \h ]). Then, the EPA estimated a linear regression function of perchlorate concentration in drinking water on maternal fT4. This was done in order to derive a function specific to each iodine intake group that could be used to estimate the benefits of reduced

<sup>&</sup>lt;sup>6</sup> For additional information on the BBDR model, refer to Chapter 3 and Appendix A in the *Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water* report [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"ahcggbXk","properties": {"formattedCitation":"(EPA, 2018e)", "plainCitation":"(EPA,

<sup>2018</sup>e)", "noteIndex":6}, "citationItems":[{"id":926,"uris":["http://zotero.org/groups/945096/items/ANBBTDKU"], "uri":["http://zotero.org/groups/945096/items/ANBBTDKU"], "itemData": {"id":926, "type": "report", "title": "Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water", "shortTitle": "Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}}, "suppressauthor": true, "prefix": "EPA,"}], "schema": "https://github.com/citation-style-language/schema/raw/master/cslcitation.json"}].

exposure to perchlorate at any possible water concentration, as opposed to just the water concentrations input into the BBDR model. The  $R^2$  values of each regression analysis were evaluated to confirm the fit of the linear functional relationship between perchlorate and fT4. In all instances the  $R^2$  values were 0.98 or greater, confirming a reasonable predictive power for values in-between the original perchlorate doses input into the BBDR model. The high  $R^2$  values also confirm the linearity of the relationship between fT4 and perchlorate for each iodine intake level in the dose-range evaluated (0 to 10  $\mu g/kg/day$ ).

The BBDR model can only be run at discrete iodine intake levels and, therefore, the results for a single value of iodine intake were assigned to a range of iodine intake values for the purposes of informing a benefits assessment. The functions relating perchlorate to fT4 are summarized in [ REF \_Ref522614369 \h ]. The raw output from the BBDR model is presented in [ REF \_Ref523405532 \n \h ].

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Effect Estimates Relating Perchlorate in Drinking Water to Changes in fT4 Based on Analysis of BBDR Model Results

lodine Intake Used for BBDR Model (μg/day)	lodine Intake Range (μg/day) Used for Benefits Analysis	Intercept	β Relating Perchlorate in Drinking Water (μg/L) to fT4 (pmol/L)	
50	0 to < 55	8.26	-0.0008	
55	55 to < 60	8.46	-0.0009	
60	60 to < 65	8.66	-0.0010	
65	65 to < 70	8.85	-0.0011	
70	70 to < 75	9.05	-0.0012	
75	75 to < 80	9.25	-0.0013	
80	80 to < 85	9.45	-0.0015	
85	85 to < 90	9.66	-0.0016	
90	90 to < 95	9.86	-0.0017	
95	95 to < 100	10.04	-0.0017	
100	100 to < 125	10.19	-0.0016	
125	125 to < 150	10.41	-0.0004	
150	150 to < 170	10.47	-0.0002	
170	170 to < 300	10.50	-0.0001	
300	≥ 300	10.57	-0.00003	

Using the inputs in [ REF \_Ref522614369 \h ], pre- and post-rule maternal fT4 values are estimated using the following equation:

$$fT4_{i,p} = \beta_i \times P_p + I_i$$
,

where:

 $fT4_{i,p}$  = fT4 concentration for population p at iodine intake range i (in picomoles per litre, pmol/L)

 $\beta_i = \beta$  from [ REF Ref522614369 \h ] for iodine intake range i

 $P_p$  = perchlorate concentration in drinking water for population p (µg/L)

 $I_i$  = intercept from [ REF\_Ref522614369 \h ] for iodine intake range i.

#### 4.2.4.2 Step 2: Maternal fT4 to Offspring IQ

Following the SAB's recommendation, the EPA conducted a literature review to evaluate the most rigorous study or studies to use that associated changes in maternal fT4 to changes in offspring IQ.<sup>7</sup> To identify studies that connected incremental changes in maternal T4 or fT4 to incremental changes in offspring neurodevelopment, the EPA assessed 71 epidemiological studies using a 4-step approach and also assessed the feasibility of conducting de novo analysis on available datasets. Ultimately, the EPA selected its own reanalysis of the Korevaar et al. [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID": "Chy7rAPY", "properties": {"formattedCitation": "(2016)", "plainCitation": "(2016)", "noteIndex":0}, "citationItems":[{"id":43, "uris":["http://zotero.org/groups/945096/items/B968J6 XI"],"uri":["http://zotero.org/groups/945096/items/B968J6XI"],"itemData":{"id":43,"type":"artic le-journal", "title": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study", "container-title": "The Lancet Diabetes & Endocrinology", "page": "35-43","volume":"4","issue":"1","source":"ScienceDirect","abstract":"SummaryBackground\nThyro id hormone is involved in the regulation of early brain development. Since the fetal thyroid gland is not fully functional until week 18-20 of pregnancy, neuronal migration and other crucial early stages of intrauterine brain development largely depend on the supply of maternal thyroid hormone. Current clinical practice mostly focuses on preventing the negative consequences of low thyroid hormone concentrations, but data from animal studies have shown that both low and high concentrations of thyroid hormone have negative effects on offspring brain development. We aimed to investigate the association of maternal thyroid function with child intelligence quotient (IQ) and brain morphology.\nMethods\nIn this population-based prospective cohort study, embedded within the Generation R Study (Rotterdam, Netherlands), we investigated the association of maternal thyroid function with child IQ (assessed by non-verbal intelligence tests) and brain morphology (assessed on brain MRI scans). Eligible women were those living in the study area at their delivery date, which had to be between April 1, 2002, and Jan 1, 2006. For this study, women with available serum samples who presented in early pregnancy (<18 weeks) were included. Data for maternal thyroid-stimulating hormone, free thyroxine, thyroid peroxidase antibodies (at weeks 9–18 of pregnancy), and child IQ (assessed at a median of 6.0 years of age [95% range 5.6–7.9 years]) or brain MRI scans (done at a median of 8.0 years of age [6·2–10·0]) were obtained. Analyses were adjusted for potential confounders including concentrations of human chorionic gonadotropin and child thyroid-stimulating hormone and free

citation.json"} ].

<sup>&</sup>lt;sup>7</sup> This process, analysis, and justification of the ultimate selection is presented in Chapters 5 and 6 of the *Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water* report [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"UnXosuZ9","properties":{"formattedCitation":"(EPA, 2018e)","plainCitation":"(EPA, 2018e)","noteIndex":7},"citationItems":[{"id":926,"uris":["http://zotero.org/groups/945096/items/ANBBTDKU"],"uri":["http://zotero.org/groups/945096/items/ANBBTDKU"],"itemData":{"id":926,"type":"report","title":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","shortTitle":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}},"suppress-author":true,"prefix":"EPA,"}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-

thyroxine.\nFindings\nData for child IQ were available for 3839 mother-child pairs, and MRI scans were available from 646 children. Maternal free thyroxine concentrations showed an inverted U-shaped association with child IQ (p=0.0044), child grey matter volume (p=0.0062), and cortex volume (p=0.0011). For both low and high maternal free thyroxine concentrations, this association corresponded to a 1.4-3.8 points reduction in mean child IQ. Maternal thyroidstimulating hormone was not associated with child IQ or brain morphology. All associations remained similar after the exclusion of women with overt hypothyroidism and overt hyperthyroidism, and after adjustment for concentrations of human chorionic gonadotropin, child thyroid-stimulating hormone and free thyroxine or thyroid peroxidase antibodies (continuous or positivity).\nInterpretation\nBoth low and high maternal free thyroxine concentrations during pregnancy were associated with lower child IQ and lower grey matter and cortex volume. The association between high maternal free thyroxine and low child IQ suggests that levothyroxine therapy during pregnancy, which is often initiated in women with subclinical hypothyroidism during pregnancy, might carry the potential risk of adverse child neurodevelopment outcomes when the aim of treatment is to achieve high-normal thyroid function test results.\nFunding\nThe Netherlands Organisation for Health Research and Development (ZonMw) and the European Community's Seventh Framework Programme.","DOI":"10.1016/S2213-8587(15)00327-7", "ISSN": "2213-8587", "shortTitle": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood", "journal Abbreviation": "The Lancet Diabetes & Endocrinology", "author": [{"family": "Korevaar", "given": "Tim I M"},{"family":"Muetzel","given":"Ryan"},{"family":"Medici","given":"Marco"},{"family":"Ch aker","given":"Layal"},{"family":"Jaddoe","given":"Vincent W V"},{"family":"Rijke","given":"Yolanda B","non-droppingparticle":"de"},{"family":"Steegers","given":"Eric A P"},{"family":"Visser","given":"Theo J"},{"family":"White","given":"Tonya"},{"family":"Tiemeier","given":"Henning"},{"family":"P eeters", "given": "Robin P"}], "issued": {"date-parts": [["2016",1]]}}, "suppressauthor":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json"} dataset to be the basis for the function relating maternal fT4 to offspring IQ. This selection was based on the large sample size of the analysis dataset compared to the other studies; the ability to control for an appropriate set of confounders; the feasibility of assessing the appropriate dose-response relationship; and the ability to develop a function that is specific to particular ranges of fT4, but span the entire distribution of possible fT4 levels. Based on the EPA's reanalysis, the concentration-response function relating fT4 to IQ is as follows:

<sup>8</sup> See Chapter 6 of the *Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water* report [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"JlnOzgLX","properties": {"formattedCitation":"(USEPA, 2018e)","plainCitation":"(USEPA, 2018e)","noteIndex":8},"citationItems":[{"id":926,"uris":["http://zotero.org/groups/945096/items/ANBBTDKU"],"uri":["http://zotero.org/groups/945096/items/ANBBTDKU"],"itemData": {"id":926,"type":"report","title":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","shortTitle":"Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

When  $fT4_1$  and  $fT4_0$  are both less than or equal to 11.76 pmol/L<sup>9</sup>:

$$\Delta IQ = (\gamma \times \ln(fT4_1)) - (\gamma \times \ln(fT4_0)),$$

where:

 $\gamma$  = coefficient from the EPA reanalysis of Korevaar et al. (2016), or 17.26 (3.77, 30.75)  $fT4_1$  = maternal fT4 under the perchlorate rule option in drinking water scenario  $fT4_0$  = maternal fT4 under the baseline perchlorate concentration in drinking water scenario.

When the fT4 values exceed 11.76, it is assumed that no benefits will accrue. This is based on the fact that the function derived for fT4 levels between 11.76 and 18.94 [the 10th to 90th percentiles from the Korevaar et al. [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"CIzJ3rfd", "properties": {"formattedCitation":"(2016)", "plainCitation":"(2016)", "n oteIndex":0}, "citationItems":[{"id":43,"uris":["http://zotero.org/groups/945096/items/B968J6XI" ],"uri":["http://zotero.org/groups/945096/items/B968J6XI"],"itemData":{"id":43,"type":"articlejournal". "title": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study", "containertitle":"The Lancet Diabetes & Endocrinology", "page": "35-43", "volume": "4", "issue": "1", "source": "ScienceDirect", "abstract": "SummaryBackground\nThyro id hormone is involved in the regulation of early brain development. Since the fetal thyroid gland is not fully functional until week 18–20 of pregnancy, neuronal migration and other crucial early stages of intrauterine brain development largely depend on the supply of maternal thyroid hormone. Current clinical practice mostly focuses on preventing the negative consequences of low thyroid hormone concentrations, but data from animal studies have shown that both low and high concentrations of thyroid hormone have negative effects on offspring brain development. We aimed to investigate the association of maternal thyroid function with child intelligence quotient (IQ) and brain morphology.\nMethods\nIn this population-based prospective cohort study, embedded within the Generation R Study (Rotterdam, Netherlands), we investigated the association of maternal thyroid function with child IQ (assessed by non-verbal intelligence tests) and brain morphology (assessed on brain MRI scans). Eligible women were those living in the study area at their delivery date, which had to be between April 1, 2002, and Jan 1, 2006. For this study, women with available serum samples who presented in early pregnancy (<18 weeks) were included. Data for maternal thyroid-stimulating hormone, free thyroxine, thyroid peroxidase antibodies (at weeks 9–18 of pregnancy), and child IQ (assessed at a median of 6.0 years of age [95% range 5.6–7.9 years]) or brain MRI scans (done at a median of 8.0 years of age [6·2–10·0]) were obtained. Analyses were adjusted for potential confounders including concentrations of human chorionic gonadotropin and child thyroid-stimulating hormone and free thyroxine.\nFindings\nData for child IQ were available for 3839 mother-child pairs, and MRI scans were available from 646 children. Maternal free thyroxine concentrations showed an inverted U-shaped association with child IQ (p=0.0044), child grey matter volume (p=0.0062), and cortex volume (p=0.0011). For both low and high maternal free thyroxine concentrations,

<sup>&</sup>lt;sup>9</sup> This fT4 value represents untransformed values of the ln(fT4) values at each knot of the spline. The values were obtained by calculating exp(ln(fT4)).

this association corresponded to a 1·4–3·8 points reduction in mean child IQ. Maternal thyroidstimulating hormone was not associated with child IQ or brain morphology. All associations remained similar after the exclusion of women with overt hypothyroidism and overt hyperthyroidism, and after adjustment for concentrations of human chorionic gonadotropin, child thyroid-stimulating hormone and free thyroxine or thyroid peroxidase antibodies (continuous or positivity).\nInterpretation\nBoth low and high maternal free thyroxine concentrations during pregnancy were associated with lower child IQ and lower grey matter and cortex volume. The association between high maternal free thyroxine and low child IQ suggests that levothyroxine therapy during pregnancy, which is often initiated in women with subclinical hypothyroidism during pregnancy, might carry the potential risk of adverse child neurodevelopment outcomes when the aim of treatment is to achieve high-normal thyroid function test results.\nFunding\nThe Netherlands Organisation for Health Research and Development (ZonMw) and the European Community's Seventh Framework Programme.","DOI":"10.1016/S2213-8587(15)00327-7", "ISSN": "2213-8587", "shortTitle": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood", "journal Abbreviation": "The Lancet Diabetes & Endocrinology", "author": [{"family": "Korevaar", "given": "Tim I M"},{"family":"Muetzel","given":"Ryan"},{"family":"Medici","given":"Marco"},{"family":"Ch aker", "given": "Layal"}, {"family": "Jaddoe", "given": "Vincent W V"}, {"family": "Rijke", "given": "Yolanda B", "non-droppingparticle":"de"},{"family":"Steegers","given":"Eric A P"},{"family":"Visser","given":"Theo J"},{"family":"White","given":"Tonya"},{"family":"Tiemeier","given":"Henning"},{"family":"P eeters", "given": "Robin P"}], "issued": {"date-parts": [["2016", 1]]}}, "suppressauthor":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json"} dataset] does not demonstrate a statistically significant relationship between fT4 and IQ (p-values greater than 0.8). However, given that the BBDR model results estimate the median fT4 to be 10.7 pmol/L with adequate iodine (i.e., iodine intake = 170  $\mu$ g/day), it is unlikely that any significant benefits based on this cutpoint will be missed.

Combining the fT4 equation in Step 1 of the dose-response function with the  $\Delta$ IQ equation in Step 2 for each iodine intake range group yields the following full dose-response equation:

$$\Delta IQ = \sum_{i=1}^{15} \left( \left( \left( \gamma \times \ln \left( (\beta_i P_{1p} + I_i) \right) \right) - \left( \gamma \times \ln \left( (\beta_i P_{0p} + I_i) \right) \right) \right) * IProp \right),$$

where:

 $\gamma$  = coefficient from the EPA reanalysis of Korevaar et al. [ ADDIN ZOTERO\_ITEM CSL CITATION

{"citationID":"I3Kk2AXD","properties":{"formattedCitation":"(2016)","plainCitation":"(2016)","noteIndex":0},"citationItems":[{"id":43,"uris":["http://zotero.org/groups/945096/items/B968J6XI"],"uri":["http://zotero.org/groups/945096/items/B968J6XI"],"itemData":{"id":43,"type":"article-journal","title":"Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study","container-title":"The Lancet Diabetes & Endocrinology","page":"35-

43", "volume": "4", "issue": "1", "source": "ScienceDirect", "abstract": "SummaryBackg round \nThyroid hormone is involved in the regulation of early brain development.

Since the fetal thyroid gland is not fully functional until week 18–20 of pregnancy, neuronal migration and other crucial early stages of intrauterine brain development largely depend on the supply of maternal thyroid hormone. Current clinical practice mostly focuses on preventing the negative consequences of low thyroid hormone concentrations, but data from animal studies have shown that both low and high concentrations of thyroid hormone have negative effects on offspring brain development. We aimed to investigate the association of maternal thyroid function with child intelligence quotient (IQ) and brain morphology.\nMethods\nIn this population-based prospective cohort study, embedded within the Generation R Study (Rotterdam, Netherlands), we investigated the association of maternal thyroid function with child IQ (assessed by non-verbal intelligence tests) and brain morphology (assessed on brain MRI scans). Eligible women were those living in the study area at their delivery date, which had to be between April 1, 2002, and Jan 1, 2006. For this study, women with available serum samples who presented in early pregnancy (<18 weeks) were included. Data for maternal thyroidstimulating hormone, free thyroxine, thyroid peroxidase antibodies (at weeks 9–18 of pregnancy), and child IQ (assessed at a median of 6.0 years of age [95% range 5.6-7.9 years]) or brain MRI scans (done at a median of 8.0 years of age [6.2-10.0]) were obtained. Analyses were adjusted for potential confounders including concentrations of human chorionic gonadotropin and child thyroid-stimulating hormone and free thyroxine.\nFindings\nData for child IQ were available for 3839 mother-child pairs, and MRI scans were available from 646 children. Maternal free thyroxine concentrations showed an inverted U-shaped association with child IQ (p=0.0044), child grey matter volume (p=0.0062), and cortex volume (p=0.0011). For both low and high maternal free thyroxine concentrations, this association corresponded to a 1·4-3·8 points reduction in mean child IQ. Maternal thyroidstimulating hormone was not associated with child IQ or brain morphology. All associations remained similar after the exclusion of women with overt hypothyroidism and overt hyperthyroidism, and after adjustment for concentrations of human chorionic gonadotropin, child thyroid-stimulating hormone and free thyroxine or thyroid peroxidase antibodies (continuous or positivity).\nInterpretation\nBoth low and high maternal free thyroxine concentrations during pregnancy were associated with lower child IQ and lower grey matter and cortex volume. The association between high maternal free thyroxine and low child IQ suggests that levothyroxine therapy during pregnancy, which is often initiated in women with subclinical hypothyroidism during pregnancy, might carry the potential risk of adverse child neurodevelopment outcomes when the aim of treatment is to achieve high-normal thyroid function test results.\nFunding\nThe Netherlands Organisation for Health Research and Development (ZonMw) and the European Community's Seventh Framework Programme.","DOI":"10.1016/S2213-8587(15)00327-7","ISSN":"2213-8587", "shortTitle": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood", "journal Abbreviation": "The Lancet Diabetes & Endocrinology", "author": [{"family": "Korevaar", "given": "Tim I M"}, {"family": "Muetzel", "given": "Ryan"}, {"family": "Medici", "given": "Marco"}, {

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"family": "Chaker", "given": "Layal" \, \ "family": "Jaddoe", "given": "Vincent W
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      parts":[["2016",1]]}},"suppress-
      author":true}],"schema":"https://github.com/citation-style-
      language/schema/raw/master/csl-citation.json"} ], or 17.26 (3.77, 30.75)
\beta_i = \beta from [ REF Ref522614369 \h \* MERGEFORMAT ] for iodine intake range i
P_{1n} = perchlorate concentration in drinking water (µg/L) under the rule scenario for
      population p
P_{0n} = perchlorate concentration in drinking water (\mug/L) under the baseline scenario for
      population p
I_i = intercept from [ REF Ref522614369 \h \* MERGEFORMAT ] for iodine intake
      range i
i = iodine intake range group
IProp = proportion of population from [ REF Ref522614353 \h \* MERGEFORMAT ]
      in a particular iodine intake range.
```

#### 4.2.5 Value of an IQ point

To determine the value of avoided IQ losses, the EPA used estimates of the change in a child's future expected lifetime earnings per one IQ point reduction. Based on methods developed by Salkever [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"isjNxsEc","properties": {"formattedCitation":"(1995)","plainCitation":"(1995)","n oteIndex":0},"citationItems":[{"id":182,"uris":["http://zotero.org/groups/945096/items/PMXG4BGW"],"uri":["http://zotero.org/groups/945096/items/PMXG4BGW"],"itemData": {"id":182,"ty pe":"article-journal","title":"Updated estimates of earnings benefits from reduced exposure of children to environmental lead","container-title":"Environmental Research","page":"1-6","volume":"70","issue":"1","source":"PubMed","abstract":"The recent and important study by Schwartz found that almost three-fourths of the benefits of reduced lead exposure in children are in the form of earnings gains (earnings losses avoided). New data on recent trends in returns to education and cognitive skills in the labor market suggest a need to revise this estimate upward. Based on an analysis of data from the National Longitudinal Survey of Youth, the present study estimates that an upward revision of at least 50% (or \$2.5 billion per annual birth cohort) is indicated. The study also finds evidence that percentage earnings gains are considerably larger for females than for males.","DOI":"10.1006/enrs.1995.1038","ISSN":"0013-9351","note":"PMID: 8603652","journalAbbreviation":"Environ.

Res.","language":"eng","author":[{"family":"Salkever","given":"D. S."}],"issued":{"date-parts":[["1995",7]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], the EPA estimates that a one point change in IQ results in a 1.865 percent change in lifetime earnings for males and a 3.397 percent change in lifetime earnings for females. The EPA estimated lifetime earnings separately for males and females using average education enrollment and annual earnings from 10 American Community Survey (ACS) Public Use Microdata Sample (PUMS) single-year samples (2008 to 2017). Then, the EPA weighted the male and female lifetime earnings by the proportion of the adult

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population that is male and female, based on life tables from the Social Security Administration (SSA) [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"P2VLrPmO","properties": {"formattedCitation":"(U.S. Census Bureau, 2017a)","plainCitation":"(U.S. Census Bureau, 2017a)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":951,"uris":["http://zotero.org/groups/945096/items/MRQCZ8P9"],"uri":["http://zotero.org/groups/945096/items/MRQCZ8P9"],"itemData":{"id":951,"type":"article","title":"American Community Survey","author":[{"literal":"U.S. Census Bureau"}],"issued":{"date-parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].
```

Additionally, the EPA adjusted the change in lifetime earnings to account for the decreased average length of education associated with an IQ decrement. Also based on the methods developed by Salkever (1995), the EPA estimated that a one IQ point reduction leads to an average reduction in schooling of 0.0811 years for males and 0.0916 years for females. To estimate the costs associated with the change in educational attainment, the EPA used ACS PUMS enrollment data to determine the level at which the change occurs (i.e. secondary school versus postsecondary school) together with data on educational costs from the Digest of Education Statistics (Snyder and Dillow, 2016). The EPA also adjusted the values to account for lost earnings during the additional educational enrollment.

The net monetized value of a one IQ point change is change in lifetime earnings, net of the change in educational costs and foregone earnings. See [ REF \_Ref514841528 \n \h ] for a description of the methodology.

```
REF Ref522614422 \h | summarizes the net value of an IQ point. Estimates are presented in
2017 dollars and are discounted using both a 3 percent and a 7 percent discount rate. The original
analysis of the discounted present value of lifetime income differentials associated with a one-
point IQ loss [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"B1LFmv0q","properties": {"formattedCitation":"(USEPA,
2018h)", "plainCitation": "(USEPA,
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d":1009, "type": "article", "title": "Economic Analysis of the Proposed Rule to Revise the TSCA
Dust Lead Hazard Standards", "author": [{"literal": "USEPA"}], "issued": {"date-
parts":[["2018",6]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] was discounted to the third year of life to
reflect a typical exposure age to lead in dust. For the proposed perchlorate rule, the EPA further
discounted the present value of lifetime income differentials three additional years to align the
benefits of reduced perchlorate exposure more closely with prenatal exposure. This adjustment
does not affect the total value of an IQ point over a lifetime but rather reflects the present value
at birth rather than at age three. These revised values are presented in the bolded row in REF
Ref522614422 h].
```

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Average Effects of a One-Point Change in IQ on Earnings (2017\$)

Fatimate Barranta	Discount Rate	
Estimate Parameter	3%	7%
Present value of lifetime earnings	\$884,342	\$205,144
IQ value	\$23,269	\$5,398
Additional education costs and lost earnings	\$1,592	\$691
Net value of an IQ point (IQ value less additional education costs and lost earnings) discounted to the third year of life	\$21,677	\$4,707
Net value of an IQ point, discounted to birth	\$19,838	\$3,842

#### 4.2.6 Summary of Benefits due to Avoided IQ Decrements

Following the approach laid out in [ REF \_Ref523393377 \h ] and the inputs laid out in the preceding sections, the EPA estimates between 30 and 243 points of lost IQ will be avoided in the affected populations each year, after full implementation. The EPA estimated the value of these benefits over a 35-year analysis period and accounted for a phase-in of control technology in year 6 for large CWSs and year 9 for all other systems (see Section [ REF \_Ref523415193 \r \h \\* MERGEFORMAT ]). As such, benefits begin to accrue in year 6.

The EPA calculated the present value of total benefits in each year of the analysis period and discounted benefits to year 1 using both a 3 percent and 7 percent discount rate. [ REF \_Ref523390582 \h \\* MERGEFORMAT ] and [ REF \_Ref525891578 \h \\* MERGEFORMAT ] summarize the results at MCLs of 56  $\mu$ g/L and 18  $\mu$ g/L, respectively.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Annualized Benefits of Avoided IQ Decrements at an MCL of 56  $\mu$ g/L (millions 2017\$)

	Annual Delta IQ <sup>a</sup>	Annual Benefits <sup>b</sup>		
	Allitual Della IQ	3% Discount	7% Discount	
Upper	243	\$3.79	\$0.64	
Central	136	\$2.12	\$0.36	
Lower	30	\$0.46	\$0.08	

a. Annual change in IQ points in affected population is after full implementation.

b. Annualized benefits are calculated over a 35-year period and account for a phase-in of benefits corresponding to compliance in year 6 for large CWSs and in year 9 for all other systems.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Annualized Benefits of Avoided IQ Decrements at an MCL of 18  $\mu$ g/L (millions 2017\$)

Annual Delta IQ <sup>a</sup>		UCMR 1 <sup>c</sup> National <sup>b,c</sup>			nal <sup>b,c</sup>	
	UCMR 1	National <sup>o</sup>	3% Discount	7% Discount	3% Discount	7% Discount
Upper	442	447	\$6.90	\$1.16	\$6.97	\$1.17
Central	248	251	\$3.87	\$0.65	\$3.91	\$0.66
Lower	54	55	\$0.85	\$0.14	\$0.85	\$0.14

a. Annual change in IQ points in affected population is after full implementation.

#### 4.2.7 Limitations to Benefits Assessment

The quantitative benefits analysis has several limitations. The primary limitation is that it includes only one health endpoint and, therefore, excludes benefits of avoiding other types of adverse health effects of perchlorate exposure. Section 4.3 provides a qualitative discussion of other health effects. Other limitations include the health risks based on maximum recorded concentration estimates do not account for the possibility of exposure to concentrations greater than or less than this maximum concentration assuming that:

- Baseline fT4 is equal to the median likely underestimates disease benefits as the logarithmic relationship between maternal fT4 and child IQ leads to larger relative changes in fT4 with increasing levels of perchlorate with lower levels of baseline fT4, and
- A median TSH feedback loop strength for the exposed population does not incorporate the variability in the feedback mechanism of the body's creation of TSH in response to decreasing fT4.

### 4.3 Non-Monetized Benefits of Reduced Perchlorate Exposure

The monetized benefits do not include several types of non-quantifiable benefits of reduced perchlorate exposure. These consist of other health effects associated with perchlorate due to its alteration of iodine and thyroid hormone levels (see Section [ REF \_Ref523470708 \r \h \\* MERGEFORMAT ]), improved perception of water quality (see Section [ REF \_Ref325447503 \r \h \\* MERGEFORMAT ]), and the possibility of reducing other contaminants if perchlorate is reduced (see Section [ REF \_ Ref325447528 \r \h \\* MERGEFORMAT ]).

b. The EPA applied statistical sampling weights to the results to extrapolate small system results to national results. The entry point at which a measurement exceeds  $18~\mu g/L$  is 1~of~20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the 6 UCMR 1 systems in the stratum (40,574). Currently, the stratum population of 774,780 accounts for 1.32% of the 58.7 million national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers (approximately 41,100) may be exposed to perchlorate greater than  $18~\mu g/L$ . The EPA assumed that this population would incur benefits equivalent to the sampled entry point's population.

c. Annualized benefits are calculated over a 35-year period and account for a phase-in of benefits corresponding to compliance in year 6 for large CWSs and in year 9 for all other systems.

# 4.3.1 Additional Neurological Endpoints Associated with Reduced Iodine and Altered Thyroid Hormones

Given the evidence that perchlorate can alter iodine uptake, evidence examining the impact of reduced or low iodine are relevant in understanding the potential impacts of perchlorate exposure. One such study, a review by Bleichrodt and Born [ ADDIN ZOTERO\_ITEM CSL CITATION

{"citationID":"CpSKIcOJ","properties":{"formattedCitation":"(1994)","plainCitation":"(1994)"," noteIndex":0},"citationItems":[{"id":178,"uris":["http://zotero.org/groups/945096/items/DDKM KVDC"],"uri":["http://zotero.org/groups/945096/items/DDKMKVDC"],"itemData":{"id":178,"t ype":"chapter","title":"A metaanalysis of research on iodine and its relationship to cognitive development","container-title":"The Damaged Brain of Iodine Deficiency","publisher":"Cognizant Communication","publisher-place":"New York, NY","page":"195-200","event-place":"New York, NY","author":[{"family":"Bleichrodt","given":"N"},{"family":"Born","given":"M."}],"issued":{"date-parts":[["1994"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], looked at 18 studies of iodine deficiency and

"date-parts":[["1994"]]}}, "suppress-author":true}], "schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], looked at 18 studies of iodine deficiency and mental development. In a meta-analysis, the authors calculated an effective size of 0.90 for iodine deficiency on cognitive development, indicating the mean scores for the two groups (the iodine-deficient group and the non-iodine-deficient group) were 0.90 of a standard deviation apart, or 13.5 IQ points.

Further, many studies evaluate the relationship between hypothyroxinemia and altered neurodevelopmental outcomes. The body of literature evaluates different populations, at different ages for neurodevelopmental assessment, and at various cut points for fT4 to define hypothyroxinemia, and finds a significant difference in performance on global cognitive tests when comparing the offspring of hypothyroxinemic women to those of non-hypothyroxinemic women [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"HEa1BIKe", "properties": {"formattedCitation":"(Costeira et al., 2011; Ghassabian et al., 2014; J\\uc0\\u250{} lvez et al., 2013; Korevaar et al., 2016; Li et al., 2010; Pop et al., 2003; 1999)", "plainCitation": "(Costeira et al., 2011; Ghassabian et al., 2014; Júlvez et al., 2013; Korevaar et al., 2016; Li et al., 2010; Pop et al., 2003; 1999)", "noteIndex":0}, "citationItems":[{"id":230, "uris":["http://zotero.org/groups/945096/items/ P5AYCR9S"], "uri": ["http://zotero.org/groups/945096/items/P5AYCR9S"], "itemData": {"id":230 "type":"article-journal", "title": "Psychomotor development of children from an iodine-deficient," region", "container-title": "Journal of Pediatrics", "page": "447-453", "volume": "159", "issue": "3", "DOI": "10.1016/j.jpeds.2011.02.034", "ISSN": "ISSN 0022-3476 EISSN 1097-6833", "language": "English", "author": [{"family": "Costeira", "given": "M. J."}, {"family": "Oliveira", "given": "P."}, {"family": "Santos", "given": "N. C."},{"family":"Ares","given":"S."},{"family":"Sáenz-Rico", "given": "B." }, { "family": "Escobar", "given": "G. M.", "non-droppingparticle":"de"},{"family":"Palha","given":"J. A."}],"issued":{"dateparts":[["2011"]]}}},{"id":391,"uris":["http://zotero.org/groups/945096/items/CZUXXSSN"],"ur i":["http://zotero.org/groups/945096/items/CZUXXSSN"],"itemData":{"id":391,"type":"articlejournal", "title": "Downstream effects of maternal hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age children", "container-title": "Journal of Clinical Endocrinology and Metabolism", "page": "2383-

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id hormone is involved in the regulation of early brain development. Since the fetal thyroid gland
is not fully functional until week 18–20 of pregnancy, neuronal migration and other crucial early
stages of intrauterine brain development largely depend on the supply of maternal thyroid
hormone. Current clinical practice mostly focuses on preventing the negative consequences of
low thyroid hormone concentrations, but data from animal studies have shown that both low and
high concentrations of thyroid hormone have negative effects on offspring brain development.
We aimed to investigate the association of maternal thyroid function with child intelligence
quotient (IQ) and brain morphology.\nMethods\nIn this population-based prospective cohort
study, embedded within the Generation R Study (Rotterdam, Netherlands), we investigated the
association of maternal thyroid function with child IQ (assessed by non-verbal intelligence tests)
and brain morphology (assessed on brain MRI scans). Eligible women were those living in the
study area at their delivery date, which had to be between April 1, 2002, and Jan 1, 2006. For this
study, women with available serum samples who presented in early pregnancy (&lt:18 weeks)
were included. Data for maternal thyroid-stimulating hormone, free thyroxine, thyroid
peroxidase antibodies (at weeks 9-18 of pregnancy), and child IQ (assessed at a median of 6.0
years of age [95% range 5.6-7.9 years]) or brain MRI scans (done at a median of 8.0 years of
age [6·2–10·0]) were obtained. Analyses were adjusted for potential confounders including
concentrations of human chorionic gonadotropin and child thyroid-stimulating hormone and free
thyroxine.\nFindings\nData for child IQ were available for 3839 mother-child pairs, and MRI
scans were available from 646 children. Maternal free thyroxine concentrations showed an
inverted U-shaped association with child IQ (p=0.0044), child grey matter volume (p=0.0062),
and cortex volume (p=0.0011). For both low and high maternal free thyroxine concentrations.
this association corresponded to a 1.4-3.8 points reduction in mean child IQ. Maternal thyroid-
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stimulating hormone was not associated with child IQ or brain morphology. All associations remained similar after the exclusion of women with overt hypothyroidism and overt hyperthyroidism, and after adjustment for concentrations of human chorionic gonadotropin, child thyroid-stimulating hormone and free thyroxine or thyroid peroxidase antibodies (continuous or positivity).\nInterpretation\nBoth low and high maternal free thyroxine concentrations during pregnancy were associated with lower child IQ and lower grey matter and cortex volume. The association between high maternal free thyroxine and low child IQ suggests that levothyroxine therapy during pregnancy, which is often initiated in women with subclinical hypothyroidism during pregnancy, might carry the potential risk of adverse child neurodevelopment outcomes when the aim of treatment is to achieve high-normal thyroid function test results.\nFunding\nThe Netherlands Organisation for Health Research and Development (ZonMw) and the European Community's Seventh Framework Programme.","DOI":"10.1016/S2213-8587(15)00327-7", "ISSN": "2213-8587", "shortTitle": "Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood", "journal Abbreviation": "The Lancet Diabetes & Endocrinology", "author": [{"family": "Korevaar", "given": "Tim I M"}, {"family": "Muetzel", "given": "Ryan"}, {"family": "Medici", "given": "Marco"}, {"family": "Ch aker", "given": "Layal" }, { "family": "Jaddoe", "given": "Vincent W V"},{"family":"Rijke","given":"Yolanda B","non-droppingparticle":"de"},{"family":"Steegers","given":"Eric A P"},{"family":"Visser","given":"Theo J"},{"family":"White","given":"Tonya"},{"family":"Tiemeier","given":"Henning"},{"family":"P eeters", "given": "Robin P" ], "issued": {"dateparts":[["2016",1]]}}},{"id":366,"uris":["http://zotero.org/groups/945096/items/L3Y92VI4"],"ur i":["http://zotero.org/groups/945096/items/L3Y92VI4"],"itemData":{"id":366,"type":"articlejournal", "title": "Abnormalities of maternal thyroid function during pregnancy affect neuropsychological development of their children at 25-30 months", "container-title": "Clinical Endocrinology", "page": "825-829", "volume": "72", "DOI": "10.1111/j.1365-2265.2009.03743.x", "author": [{"family": "Li", "given": "Y."}, {"family": "Shan", "given": "Z."}, {"f amily":"Teng","given":"W."},{"family":"Yu","given":"X."},{"family":"Li","given":"Y."},{"fam ily":"Fan", "given":"C."}, {"family":"Teng", "given":"X."}, {"family":"Guo", "given":"R."}, {"famil y":"Wang", "given": "H." }, { "family": "Li", "given": "J." }, { "family": "Chen", "given": "Y." }, { "family": "Y." } ":"Wang", "given": "W."}, { "family": "Chawinga", "given": "M."}, { "family": "Zhang", "given": "L."} ,{"family":"Yang","given":"L."},{"family":"Zhao","given":"Y."},{"family":"Hua","given":"T."} ],"issued":{"dateparts":[["2010"]]}}},{"id":198,"uris":["http://zotero.org/groups/945096/items/WKCD55DW"],"u ri":["http://zotero.org/groups/945096/items/WKCD55DW"],"itemData":{"id":198,"type":"article -journal", "title": "Maternal hypothyroxinaemia during early pregnancy and subsequent child development: a 3-year follow-up study", "container-title": "Clinical Endocrinology", "page": "282-288", "volume": "59", "issue": "3", "source": "PubMed", "abstract": "OBJECTIVE: To evaluate the impact of maternal hypothyroxinaemia during early gestation (fT4 below the lowest tenth percentile and TSH within the reference range: 0.15-2.0 mIU/l) on infant development, together with any subsequent changes in fT4 during gestation.\nDESIGN: A prospective 3-year follow-up study of pregnant women and their children up to the age of 2 years.\nMEASUREMENTS: Child development was assessed by means of the Bayley Scales of Infant Development in children of women with hypothyroxinaemia (fT4 below the tenth percentile at 12 weeks' gestation) at 12 weeks' gestation (cases), and in children of women with fT4 between the 50th and 90th percentiles at 12 weeks' gestation, matched for parity and gravidity (controls). Maternal

thyroid function (fT4 and TSH) was assessed at 12, 24 and 32 weeks' gestation. The mental and motor function of 63 cases and 62 controls was compared at the age of 1 year, and of 57 cases and 58 controls at the age of 2 years.\nRESULTS: Children of women with hypothyroxinaemia at 12 weeks' gestation had delayed mental and motor function compared to controls: 10 index points on the mental scale (95% CI: 4.5-15 points, P = 0.003) and eight on the motor scale at the age of 1 year (95% CI: 2.3-12.8 points, P = 0.02), as well as eight index points on the mental (95% CI: 4-12 points, P = 0.02), and 10 on the motor scale (95% CI: 6-16 points, P = 0.005) at the age of 2 years. Children of hypothyroxinaemic women in whom the fT4 concentration was increased at 24 and 32 weeks' gestation had similar scores to controls, while in the controls, the developmental scores were not influenced by further declines in maternal fT4 at 24 and 32 weeks' gestation.\nCONCLUSIONS: Maternal hypothyroxinaemia during early gestation is an independent determinant of a delay in infant neurodevelopment. However, when fT4 concentrations increase during pregnancy in women who are hypothyroxinaemic during early gestation, infant development appears not to be adversely affected.","ISSN":"0300-0664", "note": "PMID: 12919150", "shortTitle": "Maternal hypothyroxinaemia during early pregnancy and subsequent child development", "journal Abbreviation": "Clin. Endocrinol. (Oxf)","language":"eng","author":[{"family":"Pop","given":"Victor J."}, {"family": "Brouwers", "given": "Evelien P."}, {"family": "Vader", "given": "Huib L."},{"family":"Vulsma","given":"Thomas"},{"family":"Baar","given":"Anneloes L.","nondropping-particle": "van" }, { "family": "Vijlder", "given": "Jan J.", "non-droppingparticle":"de"}],"issued":{"dateparts":[["2003",9]]}}},{"id":13,"uris":["http://zotero.org/groups/945096/items/VERVX5Q7"],"u ri":["http://zotero.org/groups/945096/items/VERVX5Q7"],"itemData":{"id":13,"type":"articlejournal", "title": "Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy", "container-title": "Clinical Endocrinology", "page": "149-155", "volume": "50", "issue": "2", "source": "PubMed", "abstract": "BACKGROUND: Maternal thyroid function during early pregnancy is an important determinant of early fetal brain development because the fetal thyroid is unable to produce any T4 before 12-14 weeks' gestation. Overt maternal hypothyroidism as seen in severe iodine-deficient areas is associated with severely impaired neurological development of the offspring. At present, it is not known whether low free T4 (fT4) levels during pregnancy in healthy women from iodine sufficient areas may affect fetal neurodevelopment.\nMETHODS: Neurodevelopment was assessed at 10 months of age in a cohort of 220 healthy children, born after uncomplicated pregnancies and deliveries, using the Bayley Scales of Infant Development. Maternal TSH, fT4 and TPO antibody status were assessed at 12 and 32 weeks' gestation. Maternal gestational fT4 concentration was defined as an independent parameter for child development.\nRESULTS: Children of women with fT4 levels below the 5th ( $\leq 9.8 \text{ pmol/l}$ , n = 11) and 10th ( $\leq 10.4 \text{ pmol/l}$ , n = 22) percentiles at 12 weeks' gestation had significantly lower scores on the Bayley Psychomotor Developmental Index (PDI) scale at 10 months of age, compared to children of mothers with higher fT4 values (t test, mean difference: 14.1, 95% confidence interval (CI): 5.9-22 and 7.4, 95% CI: 1.1-13.9, respectively). At 32 weeks' gestation, no significant differences were found. In the group of women with the lowest 10th percentile fT4 concentrations at 12 weeks' gestation, a positive correlation was found between the mothers' fT4 concentration and children's PDI scores (linear regression, R: 0.46, P = 0.03). After correction for confounding variables, a fT4 concentration below the 10th percentile at 12 weeks' gestation was a significant

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12.6).\nCONCLUSIONS: Low maternal plasma fT4 concentrations during early pregnancy may
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Sensitivity analysis by omitting other studies by turns showed that there was no obvious change
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delayed cognitive development in children (random RR = 3.08, 95 % CI 1.83-5.18, P < 0.001). When excluding the study with largest weight, there was no obvious between-study heterogeneity in the left studies (I2 = 47.6 %), and meta-analysis using random-effect model showed that low maternal thyroxine level was still significantly associated with increased risk of delayed cognitive development in children (random RR = 3.76, 95 % CI 2.14-6.58, P < 0.001). Sensitivity analysis by omitting other studies by turns showed that there was no obvious change in the pooled risk estimates, and all pooled RRs were statistically significant. Therefore, the findings from the meta-analysis provide strong evidence for the association between maternal thyroxine levels during pregnancy and cognitive development in children. Low maternal thyroxine level is significantly associated with a threefold risk of delayed cognitive development in children.","DOI":"10.1007/s12035-015-9189-z","ISSN":"0893-7648, 1559-1182","journalAbbreviation":"Mol

Neurobiol","language":"en","author":[{"family":"Wang","given":"Pingping"},{"family":"Gao"," given":"Jian"},{"family":"Zhao","given":"Shihua"},{"family":"Guo","given":"Yong"},{"family":"Wang","given":"Zengfang"},{"family":"Qi","given":"Feng"}],"issued":{"date-parts":[["2016",5,1]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] found that hypothyroxinemia was associated with a 5.7 point lower score on intelligence tests and a three-fold increased risk of delayed cognitive development in children, respectively. Thompson et al. [ ADDIN ZOTERO\_ITEM CSL CITATION

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2265","author":[{"family":"Thompson","given":"William"},{"family":"Russell","given":"Ginny "},{"family":"Baragwanath","given":"Genevieve"},{"family":"Matthews","given":"Justin"},{"family":"Vaidya","given":"Bijay"},{"family":"Thompson-Coon","given":"Jo"}],"issued":{"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] found maternal hypothyroxinemia to be associated with increased risk of cognitive delay, intellectual impairment, or lower scores on performance tests; but they did not find this association with attention-deficit/hyperactivity disorder (ADHD) or autism.

Additionally, studies have related maternal hypothyroxinemia with a plethora of other outcomes, including offspring's increased risk of schizophrenia [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"UZtr4rms","properties":{"formattedCitation":"(Gyllenberg et al., 2016)","plainCitation":"(Gyllenberg et al., 2016)","noteIndex":0},"citationItems":[{"id":24,"uris":["http://zotero.org/groups/945096/items/937P2NGT"],"uri":["http://zotero.org/groups/945096/items/937P2NGT"],"itemData":{"id":24,"ty pe":"article-journal","title":"Hypothyroxinemia during gestation and offspring schizophrenia in a national birth cohort","container-title":"Biological Psychiatry","page":"962-970","volume":"79","issue":"12","source":"PubMed","abstract":"BACKGROUND: Evidence from animal and human studies indicates that thyroid hormone deficiency during early gestation alters brain development. As schizophrenia is associated with prenatal brain insults and premorbid cognitive deficits, we tested the a priori hypothesis that serologically defined maternal

thyroid deficiency during early gestation to mid-gestation is associated with schizophrenia in offspring.\nMETHODS: The investigation is based on the Finnish Prenatal Study of Schizophrenia, a nested case-control study that included archived maternal sera from virtually all pregnancies since 1983 (N = > 1 million). We identified all offspring in the cohort with a diagnosis of schizophrenia based on the national inpatient and outpatient register and matched them on sex, date of birth, and residence in Finland at time of onset of the case to comparison subjects (1:1) from the cohort. Maternal sera of 1010 case-control pairs were assessed for free thyroxine, and sera of 948 case-control pairs were assessed for thyroid-stimulating hormone.\nRESULTS: Maternal hypothyroxinemia (free thyroxine ≤10th percentile, normal thyroid-stimulating hormone) was associated with an increased odds of schizophrenia (odds ratio = 1.75, 95% confidence interval = 1.22-2.50, p = .002). When adjusted for maternal psychiatric history, province of birth, and maternal smoking during pregnancy, the association remained significant (odds ratio = 1.70, 95% confidence interval = 1.13-2.55, p = .010).\nCONCLUSIONS: In a large, national birth cohort, prospectively documented hypothyroxinemia during early gestation to mid-gestation was associated with increased odds of schizophrenia in offspring. This information can inform translational studies of maternal hypothyroxinemia examining molecular and cellular deviations relevant to schizophrenia.", "DOI": "10.1016/j.biopsych.2015.06.014", "ISSN": "1873-2402", "note": "PMID: 26194598\nPMCID: PMC4684794","journalAbbreviation":"Biol. Psychiatry", "language": "eng", "author": [{"family": "Gyllenberg", "given": "David"}, {"family": "So urander", "given": "Andre"}, {"family": "Surcel", "given": "Heljä-Marja"}, {"family": "Hinkka-Yli-Salomäki", "given": "Susanna" }, { "family": "McKeague", "given": "Ian W."}, {"family": "Brown", "given": "Alan S."}], "issued": {"dateparts":[["2016",6,15]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ], ADHD [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "aNEaI0ec", "properties": {"formattedCitation": "(Modesto et al., 2015)", "plainCitation": "(Modesto et al., 2015)", "noteIndex":0}, "citationItems":[{"id":242, "uris":["http://zotero.org/groups/945096/items/ K4FG7Q2X"], "uri": ["http://zotero.org/groups/945096/items/K4FG7Q2X"], "itemData": {"id":242 "type": "article-journal", "title": "Maternal mild thyroid hormone insufficiency in early pregnancy, and attention-deficit/hyperactivity disorder symptoms in children", "container-title": "JAMA pediatrics", "page": "838-845", "volume": "169", "issue": "9", "source": "PubMed", "abstract": "IMPORTANCE: Maternal thyroid hormone insufficiency during pregnancy can affect children's cognitive development. Nevertheless, the behavioral outcomes of children exposed prenatally to mild thyroid hormone insufficiency are understudied.\nOBJECTIVE: To examine whether exposure to maternal mild thyroid hormone insufficiency in early pregnancy was related to symptoms of attentiondeficit/hyperactivity disorder (ADHD) in children at 8 years of age.\nDESIGN, SETTING, AND PARTICIPANTS: The study was embedded within the Generation R, a population-based birth cohort in the Netherlands. Children in the Generation R Study are followed up from birth (April 1, 2002, through January 31, 2006) until young adulthood. Of the 4997 eligible mother-child pairs with data on maternal thyroid levels (excluding twins), 3873 pairs of children and caregivers (77.5%) visited the Generation R research center for in-depth assessments and were included in the main analyses. Data collection in Generation R started December 1, 2001 (enrollment of pregnant women), and is ongoing. For this study, we used the data that were collected until January 1, 2014. Data analyses started on January 31 and finished June 30,

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2014.\nMAIN OUTCOMES AND MEASURES: Maternal hypothyroxinemia, characterized by
low levels of free thyroxine coexisting with reference thyrotropin levels, and children's
symptoms of ADHD. Maternal thyroid hormone levels (thyrotropin, free thyroxine, thyroid
peroxidase antibodies) were measured at a mean (SD) of 13.6 (1.9) weeks of gestation.
Children's ADHD symptoms were assessed at 8 years of age using the Conners' Parent Rating
Scale-Revised Short Form; higher scores indicate more ADHD symptoms (possible range, 0-
36).\nRESULTS: Maternal hypothyroxinemia (n = 127) in early pregnancy was associated with
higher scores for ADHD symptoms in children at 8 years of age after adjustments for child and
maternal factors (ie, sex, ethnicity, maternal age, maternal educational level, and income)
(increase in ADHD scores, 7% [95% CI, 0.3%-15%]). The results remained essentially
unchanged when women with elevated levels of thyroid peroxidase antibodies were excluded
from the analyses (increase in ADHD scores, 8% [95% CI, 1%-16%]). Additional adjustment for
children's IQ or comorbid autistic symptoms attenuated the association (increase in ADHD
scores adjusted for autistic symptoms, 7% [95% CI, 1%-15%]; increase in ADHD scores
adjusted for IQ, 6% [95% CI, 1%-14%]).\nCONCLUSIONS AND RELEVANCE: Children
exposed to maternal hypothyroxinemia in early pregnancy had more ADHD symptoms,
independent of confounders. This finding suggests that intrauterine exposure to insufficient
thyroid hormone levels influences neurodevelopment in
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hypothyroidism in pregnant women is associated with a lower intelligence quotient in their
children. More recently, subtle decreases in maternal thyroid function have also been associated
with neurodevelopmental impairment in offspring. We tested the effect of hypothyroxinaemia
during early pregnancy on school performance.\nDESIGN: This was a longitudinal study that
included the data of 1196 mother-child pairs from the Amsterdam Born Children and Their
Development study.\nMETHODS: Maternal serum free thyroxine (T4) and TSH were obtained
at a median gestational age of 12.9 (interquartile range: 11.9-14.3) weeks. School performance
was assessed at age 5 years and based on scores obtained in arithmetic and language tests from
the national monitoring and evaluation system. Poor school performance was defined as a test
result <25th percentile and subnormal school performance as a result <50th percentile of the
norm population. To estimate the impact of possible non-response bias, we conducted inverse-
probability weighted analyses.\nRESULTS: Maternal hypothyroxinaemia (i.e., a maternal free
T4 in the lowest 10% of distribution) was associated with a 1.61 (95% CI: 1.05-2.47) -fold
increased odds of subnormal arithmetic performance after adjustment for confounders (P=0.03).
However, the odds ratio dropped to 1.48 (95% CI: 0.94-2.32) after inverse-probability weighting
(P=0.09). No such relations were found with TSH.\nCONCLUSIONS: Maternal
hypothyroxinaemia at the end of the first trimester was associated with reduced performance in
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hypothyroidism in pregnant women is associated with a lower intelligence quotient in their
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with neurodevelopmental impairment in offspring. We tested the effect of hypothyroxinaemia
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Development study.\nMETHODS: Maternal serum free thyroxine (T4) and TSH were obtained
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was assessed at age 5 years and based on scores obtained in arithmetic and language tests from
the national monitoring and evaluation system. Poor school performance was defined as a test
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result <25th percentile and subnormal school performance as a result <50th percentile of the
norm population. To estimate the impact of possible non-response bias, we conducted inverse-
probability weighted analyses.\nRESULTS: Maternal hypothyroxinaemia (i.e., a maternal free
T4 in the lowest 10% of distribution) was associated with a 1.61 (95% CI: 1.05-2.47) -fold
increased odds of subnormal arithmetic performance after adjustment for confounders (P=0.03).
However, the odds ratio dropped to 1.48 (95% CI: 0.94-2.32) after inverse-probability weighting
(P=0.09). No such relations were found with TSH.\nCONCLUSIONS: Maternal
hypothyroxinaemia at the end of the first trimester was associated with reduced performance in
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#### 4.3.2 Thyroid Hormone Levels and CVD, and Conclusion on Benefits Analysis

Multiple studies have established an association between overt thyroid disorders and CVD risk [ ADDIN ZOTERO ITEM CSL CITATION

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ie", "given": "P. A." }], "issued": {"date-
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ZOTERO ITEM CSL CITATION
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noteIndex":0},"citationItems":[{"id":418,"uris":["http://zotero.org/groups/945096/items/C3P4K
CLI"], "uri": ["http://zotero.org/groups/945096/items/C3P4KCLI"], "itemData": {"id":418, "type":"
article-journal", "title": "The Colorado Thyroid Disease Prevalence Study", "container-title": "Arch
Intern Med", "page": "526-534", "volume": "160", "author": [{"family": "Canaris", "given": "G.
J."}, {"family": "Manowitz", "given": "N.
R."},{"family":"Mayor","given":"G."},{"family":"Ridgway","given":"E. C."}],"issued":{"date-
parts":[["2000"]]}}, "suppress-author":true}], "schema": "https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] found that there are statistically significant
trends when examining a person's TSH and T4 levels compared to total cholesterol, low-density
lipoprotein (LDL) cholesterol, and triglycerides. Further, Asvold et al. [ ADDIN
ZOTERO ITEM CSL CITATION
{"citationID": "ggh7yLw1", "properties": {"formattedCitation": "(2007)", "plainCitation": "(2007)", "
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QK"],"uri":["http://zotero.org/groups/945096/items/LN473TQK"],"itemData":{"id":431,"type":"
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concentrations in a population-based study. The HUNT study", "container-title": "European
Journal of Endocrinology", "page": "181-
186", "volume": "156", "author": [{"family": "Asvold", "given": "B."}, {"family": "Vatten", "given": "L
. J."}, {"family":"Nilsen", "given":"T."}, {"family":"Bjoro", "given":"T."}], "issued": {"date-
parts":[["2007"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] found that increases in TSH, even within the
normal range, can increase an individual's risk of fatal coronary heart disease.
```

#### 4.3.3 Reduction of Other Co-Occurring Contaminants

Many of the treatment techniques used to remove perchlorate from drinking water could also potentially remove nitrate, a co-occurring contaminant and a goitrogen. In an analysis of California drinking water sources, perchlorate was found to co-occur with nitrate, although nitrate concentrations were also found to be significantly higher and to occur over a broader geographic area [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"dEZWePMn","properties": {"formattedCitation":"(Kimbrough and Parekh, 2007)","plainCitation":"(Kimbrough and Parekh,

<sup>&</sup>lt;sup>10</sup> Goitrogens are substances that suppress the function of the [ HYPERLINK "http://en.wikipedia.org/wiki/Thyroid" \o "Thyroid" ] by interfering with [ HYPERLINK "http://en.wikipedia.org/wiki/Iodine" \o "Iodine" ] uptake, which can cause an enlargement of the thyroid (e.g., a goiter is a swelling of the thyroid).

2007)","noteIndex":0},"citationItems":[{"id":174,"uris":["http://zotero.org/groups/945096/items/68BUPB77"],"uri":["http://zotero.org/groups/945096/items/68BUPB77"],"itemData":{"id":174," type":"article-journal","title":"Occurrence and co-occurrence of perchlorate and nitrate in California drinking water sources","container-title":"Journal (American Water Works Association)","page":"126-132","volume":"99","issue":"9","source":"JSTOR","ISSN":"0003-150X","author":[{"family":"Kimbrough","given":"D.E."},{"family":"Parekh","given":"P."}],"issued":{"date-parts":[["2007"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Additionally, in a study of 326 groundwater samples from "pristine" locations across the contiguous United States, a highly significant positive correlation was found between perchlorate and nitrate concentrations [ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"Hhrd7B9k","properties": {"formattedCitation":"(Parker et al., 2008)", "plainCitation":"(Parker et al.,

2008)","noteIndex":0},"citationItems":[{"id":173,"uris":["http://zotero.org/groups/945096/items/6D2GLW89"],"uri":["http://zotero.org/groups/945096/items/6D2GLW89"],"itemData":{"id":173,"type":"article-journal","title":"Perchlorate in groundwater: a synoptic survey of \"pristine\" sites in the coterminous United States","container-title":"Environmental Science & Technology","page":"1465-

1471", "volume": "42", "issue": "5", "source": "PubMed", "abstract": "Perchlorate is widely used as an oxidant in solid rocket propellants and energetic applications, and it has frequently been detected in groundwaters at concentrations relevant to human health. The possibility of naturally occurring perchlorate has only recently received significant attention. Relying primarily on domestic, agricultural, and recreational wells, we utilized a network of volunteers to help collect 326 groundwater samples from across the coterminous United States. Care was taken to avoid known, USEPA-documented sites of perchlorate use or release, as well as perchlorate contamination due to disinfection using hypochlorite. Using IC-ESI-MS and a Cl18O4- internal standard, we achieved a method detection limit (MDL) of 40 ng/L perchlorate and a minimum reporting level (MRL) of 120 ng/L. Of the 326 samples, 147 (45%) were below the MDL, while 42 (13%) were between the MDL and the MRL. Of the 137 samples that could be quantified, most (109) contained < 1000 ng/L perchlorate; the remaining 28 samples contained from 1000 to 10400 ng/L. Our results support the notion that perchlorate occurs naturally in many groundwaters, but the unusually high concentrations (> 10000 ng/L) previously reported for the west-central Texas area appear to be anomalous. Perchlorate concentrations were positively correlated with nitrate levels (P < 0.001) but not with chloride concentrations. Opportunities exist for follow-up studies of perchlorate's origins using isotope forensics and for further elucidation of the role of atmospheric processes in the formation or transport of perchlorate.","ISSN":"0013-936X","note":"PMID: 18441789","shortTitle":"Perchlorate in groundwater", "journal Abbreviation": "Environ. Sci.

Technol.","language":"eng","author":[{"family":"Parker","given":"David R."},{"family":"Seyfferth","given":"Angelia L."},{"family":"Reese","given":"Brandi Kiel"}],"issued":{"date-parts":[["2008",3,1]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Each of the perchlorate treatment technologies evaluated in this analysis (ion exchange, biological treatment, and reverse osmosis) can also remove co-occurring nitrate [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID": "cS5tQE6t", "properties": {"formattedCitation": "(Water Research Foundation,

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2014)", "plainCitation": "(Water Research Foundation,
2014)", "noteIndex":0}, "citationItems":[{"id":172, "uris":["http://zotero.org/groups/945096/items/
YB3TVMYJ"],"uri":["http://zotero.org/groups/945096/items/YB3TVMYJ"],"itemData":{"id":17
2,"type":"article-journal","title":"Perchlorate in Drinking Water: Regulatory Update and
Treatment Options", "page": "3", "source": "Zotero", "language": "en", "author": [{"literal": "Water
Research Foundation"}],"issued":{"date-
parts":[["2014"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. For biological treatment and reverse osmosis,
nitrate removal would be continuous along with perchlorate throughout operation. Ion exchange
run to full exhaustion for perchlorate without consideration of nitrate breakthrough, however,
could result in a "peaking" situation. Peaking can occur when nitrate adsorbed early in the resin's
life is displaced by competing perchlorate, resulting in a treated water concentration of nitrate
greater than the influent concentration. Therefore, perchlorate-selective ion exchange would
require careful operation to maintain nitrate removal throughout the process. A quantitative
evaluation of the potential benefits associated with the added advantage of nitrate removal was
not included in this benefits assessment.
In the Office of Inspector General: Scientific Analysis of Perchlorate [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID": "rJD7MSTE", "properties": {"formattedCitation": "(U.S. EPA,
2008b)", "plainCitation": "(U.S. EPA,
2008b)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":966,"uris":["http://zotero.org/g
roups/945096/items/IQ4YKBMA"],"uri":["http://zotero.org/groups/945096/items/IQ4YKBMA"]
"itemData": {"id":966, "type": "article-journal", "title": "OIG Scientific Analysis of Perchlorate,"
External Review
Draft", "page": "213", "source": "Zotero", "language": "en", "author": [{"literal": "USEPA"}], "issued":
{"date-parts":[["2008"]]}}}], "schema": "https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ], the EPA suggested that the best approach to
conduct a risk assessment for perchlorate would include all four NIS stressors acting on the
thyroid: thiocyanate, nitrate, perchlorate, and lack of iodide. Perchlorate is a strong NIS
inhibitor; however, exposure to humans may be relatively low. In contrast, nitrate and
thiocyanate are weak NIS inhibitors; however, exposure levels to these two chemicals are much
greater than perchlorate [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"i8CWv09R","properties":{"formattedCitation":"(U.S. EPA,
2008b)", "plainCitation": "(U.S. EPA,
2008b)", "dontUpdate":true, "noteIndex":0}, "citationItems":[{"id":966, "uris":["http://zotero.org/g
roups/945096/items/IQ4YKBMA"],"uri":["http://zotero.org/groups/945096/items/IQ4YKBMA"]
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External Review
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Draft", "page": "213", "source": "Zotero", "language": "en", "author": [{"literal": "USEPA"}], "issued":

language/schema/raw/master/csl-citation.json"} ]. Consequently, reductions of the co-occurring

Preliminary draft for internal EPA review[ PAGE \\* MERGEFORMAT ]

Do not cite or quote

contaminant nitrate could lead to additional health benefits.

{"date-parts":[["2008"]]}}}], "schema": "https://github.com/citation-style-

#### 4.3.4 Improved Public Perception of Water Quality

PWS customers may avoid using tap water when they believe it is contaminated and poses health risks. When the public perception of water quality declines, consumers purchase bottled water or point-of-use (POU) filters if they have the means to do so.

In addition or as an alternative, they may avoid the use of tap water, ingesting and cooking with other liquids, substituting pre-mixed baby formula, and using other strategies to limit ingestion. Consumer avoidance of tap water sources usually results in costs to the consumers, either in the cost of obtaining substitute fluids or in potential health impacts of reduced fluid intake.

The relationship between perchlorate in tap water and changes in consumer behavior is a complex one. Factors that impact the choice to avoid tap water depend on public information that is provided on levels of the contamination, potential health effects, individual aversions to risk taking, and other considerations. A quantitative evaluation of these responses and the potential benefits of avoiding associated costs to the consumer or governments is not included in this benefits assessment. Nevertheless, consumers purchase bottled water or invest in other methods of improving drinking water quality, such as POU devices, specifically to avoid ingestion of contaminants such as perchlorate. Thus, it is possible that a reduction in perchlorate contamination may reduce mitigation expenditures.

## 5 Economic Impact and Cost Analysis

This chapter presents estimates of the total national costs of the proposed perchlorate rule. To estimate the national costs of the rule, the EPA calculated the incremental costs of rule components associated with the proposed rule compared to the current requirements under the SDWA. Specifically, the costs associated with the proposed rule include (1) costs borne by water systems to comply with the new NPDWR, and (2) costs to the primacy agencies to implement and enforce the NPDWR.

Costs to water systems include monitoring costs, treatment costs, and administrative costs. For treatment costs, the EPA identified the treatment technologies that will be used to comply with the MCL and estimates capital costs and operating and maintenance (O&M) costs for these technologies [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"ZWJdzdEq","properties": {"formattedCitation":"(USEPA, 2018g)","plainCitation":"(USEPA, 2018g)","noteIndex":0},"citationItems":[{"id":977,"uris":["http://zotero.org/groups/945096/items/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData": {"id":977,"type":"report","title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters","publisher":"EPA \*\*\*-\*-\*\*\*\*,"author":[{"literal":"USEPA"}],"issued": {"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Administrative costs to water systems include one-time costs to understand the rule and provide training, as well as ongoing costs for activities associated with compliance monitoring (i.e., sampling, applying for waivers, and reporting of compliance results).

Similarly, primacy agencies (states, territories, and tribal nations) incur one-time administrative costs for reading and understanding the rule, and modifying existing regulations. Ongoing administrative costs to primacy agencies include labor costs for reviewing compliance monitoring reports, making determinations on monitoring waivers, and reviewing proposed changes in treatment.

For the analysis of costs, the EPA used the PWS and occurrence data described in Sections [ REF \_Ref523415085 \r \h ] and [ REF \_Ref523404705 \r \h ], with exceptions as noted.

Section [ REF \_Ref523415174 \r \h ] summarizes methods used to identify systems that may need to control perchlorate to meet the MCL and estimate associated control costs. Section [ REF \_Ref523415193 \r \h ] describes the methods used to estimate the monitoring and administration costs for PWSs and primacy agencies. Section [ REF \_Ref535302370 \r \h ] presents the total cost results, and Section [ REF \_Ref535302385 \r \h ] summarizes household-level costs for affected systems with control costs.

#### 5.1 Control Costs Method

The EPA assumed that the entry points with exceedances will need to implement a control technology to comply with the proposed MCL or alternative MCL. The cost method overview in this section includes brief technology descriptions (Section [ REF \_Ref523455076 \r \h \\*

MERGEFORMAT]), a brief discussion of the Agency's treatment cost estimating tools, and assumptions about compliance options (Section [ REF Ref535311709 \r \h ]).

#### 5.1.1 Description of Available Control Technologies

The EPA has identified the following technologies as effective for the removal of perchlorate from drinking water:

- Ion exchange;
- Biological treatment;
- Centralized reverse osmosis; and
- POU reverse osmosis.

In addition, there are non-treatment options such and changing source water that might be used in lieu of treatment to comply with a perchlorate standard. The sections below describe each of the technologies and non-treatment options, along with key assumptions and scenarios used to estimate unit costs for each. The document *Technologies and Costs for Treating Perchlorate-Contaminated Waters* [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"wyr7sZ4c","properties": {"formattedCitation":"(USEPA, 2018g)","plainCitation":"(USEPA, 2018g)","noteIndex":0},"citationItems":[{"id":977,"uris":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData": {"id":977,"type":"report","title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters","publisher":"EPA \*\*\*-\*-\*\*\*","author":[{"literal":"USEPA"}],"issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] contains a more complete discussion of the technologies and cost estimating method.

#### 5.1.1.1 Ion Exchange

Ion exchange is a physical/chemical separation process in which an ion (such as perchlorate) in the feed water is exchanged for an ion (typically chloride) on a resin generally made of synthetic beads or gel. A variety of resin types have been tested for perchlorate removal. These resin types include strong-base polyacrylic, strong-base polystyrenic (including nitrate-selective), weak-base polyacrylic, weak-base polystyrenic, and perchlorate-selective.<sup>11</sup>

available.

<sup>11</sup> While Tripp et al. [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"SGLaAt6R","properties": {"formattedCitation":"(2003)","plainCitation":"(2003)","noteIndex":11},"citationItems": [{"id":988,"uris": ["http://zotero.org/groups/945096/items/J24BWNQH"],"uri": ["http://zotero.org/groups/945096/items/J24BWNQH"],"itemData": {"id":988,"type":"article","title":"Treatment of perchlorate in groundwater by ion exchange technology", "publisher": "AWWA Research Foundation", "call-number": "Publication No.

<sup>909&</sup>quot;, "author": [{"family": "Tripp", "given": "A.R."}, {"family": "Clifford", "given": "D."}, {"family": "Roberts", "given": "D.J."}, {"family": "Cang", "given": "Y."}, {"family": "Aldridge", "given": "L."}, {"family": "Gillogly", "given": "T."}, {"family": "Boulos", "given": "L."}], "issued": {"date-parts": [["2003"]]}}, "suppressauthor": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] also examined strong base polyvinylpyridine resins, comparable quantitative data on their removal efficiency are not

In application, feed water passes through a bed of resin in a vessel or column. The operation typically continues until the resin is exhausted, meaning that the chloride on enough of the resin's available exchange sites has been replaced with ions from the feed water so that the resin is no longer effective for removing the ion. At this point, the resin may be disposed of and replaced or regenerated. The length of time until resin exhaustion and replacement or regeneration is a critical factor in the cost-effectiveness of ion exchange as a treatment technology. It is typically measured by the number of bed volumes of water that can be treated before the breakthrough of perchlorate and can vary based on a variety of factors, including the type of resin used.

Based on data from full-scale operations, it is likely that most systems using ion exchange to comply with a perchlorate MCL would use a perchlorate-selective resin that would be disposed of, rather than regenerated, when exhausted. The ion exchange unit costs presented here assume the use of perchlorate-selective resin in two scenarios. The first scenario assumes the resin bed treats 250,000 bed volumes before disposal and replacement. The second scenario assumes 170,000 bed volumes. Both scenarios assume disposal of the spent resin by incineration, although some systems might have the slightly cheaper option of landfill disposal available. Perchlorate-selective resins are unlikely to remove substantial amounts of co-occurring contaminants because the resins maximize perchlorate removal at the expense of removing co-occurring contaminants like nitrate. Therefore, ancillary benefits are unlikely to occur.

#### 5.1.1.2 Biological Treatment

Biological treatment of perchlorate uses bacteria to reduce perchlorate to chlorate, chlorite, chloride, and oxygen. Biological treatment offers complete destruction of the perchlorate ion, eliminating the need for management of perchlorate-bearing waste streams. Although biological treatment is a relatively new technology for the treatment of drinking water in the United States, the State of California has identified biological treatment (along with ion exchange) as one of two best available technologies for achieving compliance with its standard for perchlorate in drinking water (California Code of Regulations, Title 22, Chapter 15, Section 64447.2). The first full-scale facility using biological treatment of perchlorate to supply municipal drinking water began operation in 2016 [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"sMXmpIso","properties": {"formattedCitation":"(Webster and Crowley, 2016; Webster and Litchfield, 2017)","plainCitation":"(Webster and Crowley, 2016; Webster and Litchfield,

2017)","noteIndex":0},"citationItems":[{"id":989,"uris":["http://zotero.org/groups/945096/items/BI5LYMZP"],"uri":["http://zotero.org/groups/945096/items/BI5LYMZP"],"itemData":{"id":989,"type":"speech","title":"Biological treatment of perchlorate in groundwater.","event":"AWWA Annual Conference and

Exposition", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Crowley", "given": "T.J."}], "issued": {"date-

parts":[["2016",6,21]]}}}, {"id":990,"uris":["http://zotero.org/groups/945096/items/64HZKA2M"],"uri":["http://zotero.org/groups/945096/items/64HZKA2M"],"itemData":{"id":990,"type":"article-journal","title":"Full-scale biological treatment of nitrate and perchlorate for potable water production", "container-title":"Journal AWWA", "page":"30-

40","volume":"109","issue":"5","author":[{"family":"Webster","given":"T.D."},{"family":"Litch field","given":"M.H."}],"issued":{"date-

parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

The most promising designs for biological treatment of perchlorate at drinking water facilities are those that operate either in a fixed bed or a fluidized bed configuration. Both fixed bed and fluidized bed designs involve a media bed that provides a surface on which perchlorate-reducing bacteria grow. For fixed bed reactors, influent water is typically passed under pressure through a static media bed located in a vessel. An alternative fixed bed design uses a gravity-fed concrete basin to hold the biologically active media. Fluidized bed bioreactor designs use vessels where high influent rates in an up-flow design fluidize the media bed, allowing for more surface area for biomass growth. The biological treatment unit costs presented here consider three design scenarios: a fixed bed using pressure vessels, a fixed bed using gravity basins, and a fluidized bed.

#### 5.1.1.3 Centralized Reverse Osmosis

Membrane filtration processes physically remove perchlorate ions from drinking water. These processes separate a solute such as perchlorate ions from a solution by forcing the solvent to flow through a membrane at a pressure greater than the normal osmotic pressure. The membrane is semi-permeable, transporting different molecular species at different rates. Water and low-molecular weight solutes pass through the membrane and are removed as permeate, or filtrate. Dissolved and suspended solids are rejected by the membrane and are removed as concentrate, or reject. This technique does not destroy the perchlorate ion and, therefore, creates a subsequent need for disposal or treatment of the perchlorate-contaminated waste (the concentrate).

Membranes may remove ions from feed water by a sieving action (called steric exclusion), or by the electrostatic repulsion of ions from the charged membrane surface. Membrane filtration technologies evaluated for perchlorate treatment include reverse osmosis, nanofiltration, and ultrafiltration. Bench studies of nanofiltration and ultrafiltration membranes show significant variability in these membranes' abilities to remove perchlorate, depending on other constituents of the source water. Across multiple studies, however, reverse osmosis membranes consistently achieve perchlorate removal by up to 95-98 percent [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"dFA9xEYZ","properties": {"formattedCitation":"(Liang et al., 1998; Nam et al., 2005; Sanyal et al., 2015; Yoon, Amy, et al., 2005; Yoon, Yoon, et al., 2005)", "plainCitation": "(Liang et al., 1998; Nam et al., 2005; Sanyal et al., 2015; Yoon, Amy, et al., 2005; Yoon, Yoon, et al., 2005)","noteIndex":0},"citationItems":[{"id":985,"uris":["http://zotero.org/groups/945096/items/ IQVVPD73"], "uri": ["http://zotero.org/groups/945096/items/IQVVPD73"], "itemData": {"id":985, "type": "paper-conference", "title": "Investigation of Treatment Options for Perchlorate Removal.", "publisher": "La Verne, CA: Metropolitan Water District of Southern California", "publisher-place": "San Diego, CA", "event": "AWWA Water Quality Technology Conference", "event-place": "San Diego,  $CA", "author": [\{"family": "Liang", "given": "S."\}, \{"family": "Scott", "given": "K.N."\}, \{"family": "Pathor": [\{"family": "Liang", "given": "K.N."\}, \{"family": "Pathor": [\{"family": "Ramily": "$ lencia", "given": "L.S."}, {"family": "Bruno", "given": "J."}], "issued": {"dateparts":[["1998"]]}}},{"id":986,"uris":["http://zotero.org/groups/945096/items/YHEV76YW"],"u ri":["http://zotero.org/groups/945096/items/YHEV76YW"], "itemData": {"id":986, "type": "paperconference", "title": "Perchlorate Rejection by High-Pressure Membranes and Brine Stream Treatment by Chemical and Biological Processes.", "publisher-place": "Phoenix,

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journal", "title": "Design of ultrathin nanostructured polyelectrolyte-based membranes with higher
perchlorate rejection and high permeability", "container-title": "Separation and Purification
Technology", "volume": "145", "issue": "113-
119", "author": [{"family": "Sanyal", "given": "O."}, {"family": "Sommerfeld", "given": "A.N."}, {"fa
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ri":["http://zotero.org/groups/945096/items/HPHVBSWB"],"itemData":{"id":992,"type":"article-
journal", "title": "Transport of target anions, chromate (Cr (VI)), arsenate (As (V)), and
perchlorate (ClO4), through RO, NF, and UF membranes.", "container-title": "Water Science and
Technology", "page": "327-334", "volume": "51", "issue": "6-
7","author":[{"family":"Yoon","given":"J."},{"family":"Amy","given":"G."},{"family":"Yoon",
"given": "Y." } ], "issued": { "date-
parts":[["2005"]]}}},{"id":991,"uris":["http://zotero.org/groups/945096/items/IIJW6E8Q"],"uri":
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journal", "title": "Determination of perchlorate rejection and associated inorganic fouling (scaling)
for reverse osmosis and nanofiltration membranes under various operating
conditions", "container-title": "Journal of Environmental Engineering", "page": "726-
733", "author": [{"family": "Yoon", "given": "J."}, {"family": "Yoon", "given": "Y."}, {"family": "Am
y", "given": "G."}, {"family": "Her", "given": "N."}], "issued": {"date-
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language/schema/raw/master/csl-citation.json"}].
```

The centralized reverse osmosis unit costs presented here assume the use of relatively low-pressure reverse osmosis membrane elements (consistent with the type of elements shown to be effective in the literature) to remove perchlorate. The designs on which the unit costs are based achieve recovery rates from 70 percent to 85 percent, meaning 15 to 30 percent of the influent water becomes perchlorate-laden concentrate. The unit costs assume discharge of this concentrate to a publicly owned treatment works (POTW). Although it might be impractical for most POTWs to treat very large concentrate flows, this assumption results in more conservative estimates (i.e., erring on the side of higher costs) than surface water (ocean) discharge or deep well injection, options that might be available to a limited number of systems.

#### 5.1.1.4 POU Reverse Osmosis

For perchlorate removal, the National Science Foundation (NSF) Joint Committee on Drinking Water Treatment Units has added a protocol to the *NSF/ANSI Standard 58: Reverse Osmosis Drinking Water Treatment Systems* that requires a reverse osmosis unit to be able to reduce perchlorate from a challenge level of 130 µg/L to a target level of 4 µg/L [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"0rSQe3eF","properties": {"formattedCitation":"(NSF International,

```
2004)","plainCitation":"(NSF International, 2004)","noteIndex":0},"citationItems":[{"id":997,"uris":["http://zotero.org/groups/945096/items/9EAEK7S4"],"uri":["http://zotero.org/groups/945096/items/9EAEK7S4"],"itemData":{"id":997, "type":"webpage","title":"Perchlorate Reduction","URL":"http://www.nsf.org/consumer/drinking_water/perchlorate_reduction.asp","a uthor":[{"literal":"NSF International"}],"issued":{"date-parts":[["2004"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. NSF International, the Underwriters Laboratories, and the Water Quality Association provide third-party testing and certification that POU devices meet drinking water treatment standards. There are no perchlorate certification standards for other types of POU devices such as those using ion exchange media.
```

The operating principle for POU reverse osmosis devices is the same as centralized reverse osmosis: steric exclusion and electrostatic repulsion of ions from the charged membrane surface. In addition to a reverse osmosis membrane for dissolved ion removal, POU reverse osmosis devices often have a sediment pre-filter and a carbon filter in front of the reverse osmosis membrane, a 3- to 5-gallon treated water storage tank, and a carbon filter between the tank and the tap.

```
The POU reverse osmosis unit costs the EPA developed [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID":"mVTmCCMo","properties": {"formattedCitation":"(USEPA,
2018g)","plainCitation":"(USEPA,
2018g)", "noteIndex":0}, "citationItems":[{"id":977, "uris":["http://zotero.org/groups/945096/item
s/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData":{"id":9
77, "type": "report", "title": "Technologies and Costs for Treating Perchlorate-Contaminated
Waters", "publisher": "EPA ***-*-*-**", "author": [{"literal": "USEPA"}], "issued": {"date-
parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] assume small drinking water systems would
purchase, install, and maintain certified POU devices for all customers. When a system installs,
controls (i.e., owns), and maintains POU devices at all customer locations where water is
consumed (e.g., residences), it can forego centralized treatment [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID": "hwkrqREN", "properties": {"formattedCitation": "(USEPA,
2006a)", "plainCitation": "(USEPA,
2006a)","noteIndex":0},"citationItems":[{"id":987,"uris":["http://zotero.org/groups/945096/item
s/SAEYMT7G"],"uri":["http://zotero.org/groups/945096/items/SAEYMT7G"],"itemData":{"id":
987, "type": "report", "title": "Point-of-Use or Point-of-Entry Treatment Options for Small
Drinking Water Systems", "collection-title": "EPA-815-R-06-010", "publisher": "USEPA, Office of
Ground Water and Drinking Water, Standards and Risk Management Division", "publisher-
place":"Washington, D.C.", "event-place": "Washington,
D.C.", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-
parts":[["2006"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. The costs also include development of a
public education program and monitoring of the POU devices.
```

#### 5.1.1.5 Non-Treatment Alternatives

For small water utilities that lack the financial and/or technical capacity to implement a new treatment-based compliance strategy, non-treatment options may offer a more cost-effective path

to compliance. Non-treatment options essentially replace the contaminated water source with water that meets drinking water standards, including a new standard for perchlorate.

Non-treatment solutions for drinking water compliance include well rehabilitation, contaminant source elimination, new well construction, and interconnecting with another system to purchase water [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"lK26Qjmg","properties": {"formattedCitation":"(USEPA, 2006b)", "plainCitation": "(USEPA, 2006b)", "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/item s/3SWI97AY"],"uri":["http://zotero.org/groups/945096/items/3SWI97AY"],"itemData":{"id":98 6,"type":"article","title":"Technology and Cost Document for the Final Ground Water Rule", "publisher": "EPA-815-R-06-015", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2006"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. The feasible non-treatment options will depend on site-specific circumstances such as system size, source water type, contaminant reduction needs, and proximity to alternative water sources. For small systems, neither the well rehabilitation for contaminated groundwater sources nor source elimination (e.g., remediation of perchlorate-contaminated sediments or groundwater) is likely to be a feasible and cost-effective solution. Another option – blending water from existing wells – may be a feasible, low-cost option for systems with multiple wells, including some for which perchlorate does not exceed the proposed perchlorate standard. For systems that cannot blend source water to comply with the proposed standard, two feasible non-treatment options include a new well to replace the contaminated source water and an interconnection to purchase water from a supplier. These two options are likely to have higher costs than the other options [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "nVsZ7M6v", "properties": {"formattedCitation": "(USEPA, 2006b)", "plainCitation": "(USEPA, 2006b)", "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/item s/3SWI97AY"], "uri": ["http://zotero.org/groups/945096/items/3SWI97AY"], "itemData": {"id":98 6,"type":"article", "title": "Technology and Cost Document for the Final Ground Water Rule", "publisher": "EPA-815-R-06-015", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2006"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}].

The non-treatment unit costs presented here consider two scenarios: interconnecting with another system and drilling a new well to replace a contaminated one. The costs associated with drilling a new well include well casing, screens, plugs, and pumps; well installation; buried piping and valves to connect the new well to the system; and operator labor, materials, and energy for operating and maintaining the well pumps and valves. The interconnection option involves laying a pipeline to connect the affected system to the distribution network of a neighboring system that can provide adequate water that meets all applicable drinking water standards. Costs include construction of a buried interconnecting pipeline and valves, the cost of purchased water, and maintenance of the pipeline.

#### 5.1.2 Treatment Assumptions and Unit Costs

To generate unit costs for the treatment technologies and non-treatment alternatives discussed above, the EPA used its work breakdown structure (WBS) cost-estimating models. The WBS models are spreadsheet-based engineering models for individual treatment technologies that are

linked to a central database of component unit costs. Each WBS model contains the work breakdown for a particular treatment process, and preprogrammed engineering criteria and equations that estimate equipment requirements for user-specified design requirements (e.g., system size and influent water quality). Based on these user-specific inputs, each model generates outputs that include total capital cost and annual O&M cost.

The EPA used the WBS models to generate total capital and O&M cost estimates for each technology and non-treatment option for up to 49 different system flow rates. The EPA generated separate estimates that correspond to different water sources (groundwater or surface water), three different cost levels (low, mid, and high), and different technology-specific scenarios (e.g., 250,000 or 170,000 bed volumes for ion exchange). The EPA then fit cost equations to the resulting WBS estimates for each scenario modeled, and separately for total capital and for O&M costs. The cost equations for total capital costs depend on system peak production or design flow, measured in million gallons per day (MGD). The equations for O&M cost depend on average daily flow in MGD. For each scenario, the EPA fit up to three curves: one covering small systems (less than 1 MGD design flow), one covering medium systems (1 MGD to less than 10 MGD design flow), and one covering large systems (10 MGD design flow and greater). The document Technologies and Costs for Treating Perchlorate-Contaminated Waters [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"aKrrsfCv","properties":{"formattedCitation":"(USEPA, 2018g)", "plainCitation": "(USEPA, 2018g)", "noteIndex":0}, "citationItems":[{"id":977, "uris":["http://zotero.org/groups/945096/item s/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData":{"id":9 77,"type": "report", "title": "Technologies and Costs for Treating Perchlorate-Contaminated Waters","publisher":"EPA \*\*\*-\*-\*\*\*\*","author":[{"literal":"USEPA"}],"issued":{"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] contains a more complete discussion of the WBS models and the cost-estimating approach.

For each entry point in the UCMR 1 dataset, the EPA compared the maximum perchlorate concentration to the MCL, and identified those that have a concentration that exceeds the MCL. These entry points may incur some control costs to comply with the proposed rule. The EPA estimated design and average flows based on entry point populations using the method described in section [ REF \_Ref536806570 \r \h ] and a blending ratio. Based on the flows, the EPA used the cost curves to compare costs across the technologies, which indicated that ion exchange with perchlorate-selective resin was the most cost-effective option. This outcome is consistent with the treatability literature, which contains substantially more full-scale ion exchange treatment plants compared to biological treatment or reverse osmosis. Therefore, the EPA used the capital cost and O&M cost curves to estimate treatment costs using the following assumptions:

- perchlorate selective resin with 170,000 bed volumes;
- 95 percent removal effectiveness:
- 80 percent safety factor (e.g., treatment target 14.4  $\mu$ g/L for an MCL of 18  $\mu$ g/L); and
- Design and average flow rates reflect the use of blending of treated and non-treated water to meet the treatment target.

[ REF \_Ref535312062 \n \h ] provides the cost curves used in this analysis. It also contains the formula used to calculate a blending ratio.

#### 5.2 Administrative and Monitoring Costs Method

Although added treatment improves drinking water quality and, therefore, health risk reductions and benefits, there are additional costs of the proposed rule. There are several implementation activities that must occur before systems adopt treatment to comply with an MCL. These activities result in the following costs:

- State administrative costs. States incur costs associated with adopting and enforcing the NPDWR and administering compliance monitoring programs. Each state must conduct one-time activities: reading and understanding the rule, and modifying existing state regulations. Other state costs are per entry point, and include reviewing compliance monitoring reports and making determinations on monitoring waivers.
- Sampling costs. PWSs incur O&M and labor costs for taking and analyzing a single water sample. The EPA assumes that all PWS size categories are subject to the same persample costs. Since these costs are per entry point, the cost per PWS may reflect multiple entry points.
- Other PWS administrative costs. Administrative costs may be per PWS or per entry point. Activities that result in administrative costs per PWS include reading the rule and providing training. Administrative costs per entry point include those for applying to the state for a monitoring waiver and reporting of compliance monitoring results.

As noted in Section [ REF \_Ref523347586 \r \h ], California and Massachusetts are excluded from the analysis of treatment-related costs and benefits. However, the EPA expects that California and Massachusetts will incur one-time administrative costs to read and understand the rule. As such, these states are included in the counts and costs for one-time primacy agency costs, but system counts, monitoring, waiver, and control costs exclude these states.

#### 5.2.1 Labor Rates

Because the administrative and monitoring activities primarily require labor time, the EPA estimated current labor rates for the cost analysis. State labor rates are based on the mean hourly wage rate from the BLS Standard Occupational Classification code 19-2041 (State Government – Environmental Scientists and Specialists, Including Health). Wages are loaded using a factor calculated from the BLS Employer Costs for Employee Compensation report [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"AyQYarRv","properties": {"formattedCitation":"(Bureau of Labor Statistics (BLS), 2016 Table 3)","plainCitation":"(Bureau of Labor Statistics (BLS), 2016 Table 3)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":984,"uris":["http://zotero.org/groups/945096/items/L8X3BDZ9"],"uri":["http://zotero.org/groups/945096/items/L8X3BDZ9"],"itemD ata": {"id":984,"type":"webpage","title":"Employer Cost for Employee Compensation -- September 2016","author":[{"literal":"Bureau of Labor Statistics (BLS)"}],"issued":{"date-parts":[["2016"]]}},"label":"book","suffix":"Table 3"}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], for a fully loaded hourly wage rate of \$50.67 for states.

To estimate costs to PWSs associated with compliance monitoring and other administrative costs, the EPA used the labor rates in the WBS models for technical staff (i.e., treatment plant operators) and managerial staff (i.e., utility managers for smaller systems and environmental managers for larger systems). The labor rates are in 2017 dollars and include wages and benefits. They vary by occupation and by water system size: technical rates range from \$31.91 per hour for systems up to 3,300 people to \$43.84 per hour for systems serving more than 100,000 people; and managerial rates range from \$45.24 to \$71.85 across system size categories. The EPA estimated a weighted average wage rate of \$34.71. This average rate incorporates within-size category weights for a mix of managerial and technical labor time based on employment data in the 2006 Community Water System Survey [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"vMhubruh","properties": {"formattedCitation":"(USEPA, 2009c)","plainCitation":"(USEPA,

2009c)","noteIndex":0},"citationItems":[{"id":924,"uris":["http://zotero.org/groups/945096/items/DZNAAV6M"],"uri":["http://zotero.org/groups/945096/items/DZNAAV6M"],"itemData":{"id ":924,"type":"article","title":"2006 Community Water System Survey - Volume II: Detailed Tables and Survey

Methodology","URL":"https://www.epa.gov/dwstandardsregulations/community-water-system-survey","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2009",5]]},"accessed":{"date-parts":[["2018",8,17]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], and across-size category weights based on the number of systems. [ REF Ref535312062 \n \h ] provides data and calculation details.

#### 5.2.2 Labor Hours

As described above, PWSs and states will both include monitoring and administrative costs, including one-time costs, and recurring monitoring and waiver application costs. [ REF \_Ref523410484 \h ] shows the activities, frequency, and hours per activity for primacy agencies (including 49 states, 1 tribal nation, and 5 territories). [ REF \_Ref536806569 \h ] itemizes the administrative and monitoring activities for PWSs, and shows the frequency and number of hours for each activity.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Labor Hours for Primacy Agency Administrative Requirements

Activity	Frequency	Hours
Read and understand the rule, adopt regulatory changes <sup>a</sup>	One time per state	416
Provide training and assistance to PWSs	One time per state	2,080
Provide training to staff and systems	One time per state	250
Review waiver applications	Once every 9 years per eligible system	8
Review monitoring reports	Per monitoring event <sup>b</sup>	1

a. The EPA assumed that two states that already regulate perchlorate in drinking water would not incur most of the burdens listed, but included 40 hours for this activity.

b. See Section [ REF Ref532278116 \r \h ] for monitoring frequency requirements.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Labor Hours for Drinking Water Systems' Administrative and Monitoring Requirements

Activity	Frequency	Hours		
Activity	Frequency	Small Systems	Large Systems	
Read the rule	One time per system	4	4	
Provide training	One time per system	16	32	
Apply to state for monitoring waiver	Once every 9 years per eligible system	16	16	
Take and analyze a single finished water sample	Per monitoring event <sup>a</sup>	1	1	

a. See Section [ REF \_Ref532278116 \r \h ] for monitoring frequency requirements.

#### 5.2.3 Analytical Costs for Monitoring

PWSs will also incur analytical costs for each water sample. The EPA assumes that these costs will be in the range of \$64 per sample based on an average of costs per sample from laboratories [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"Ftv0lunK","properties": {"formattedCitation":"(Eurofins, 2016; TestAmerica Laboratories, 2015; Utah Department of Environmental Quality (Utah DEQ),

2016)", "plainCitation": "(Eurofins, 2016; TestAmerica Laboratories, 2015; Utah Department of Environmental Quality (Utah DEQ),

2016)","dontUpdate":true,"noteIndex":0},"citationItems":[{"id":1003,"uris":["http://zotero.org/g roups/945096/items/EZMGVZEJ"],"uri":["http://zotero.org/groups/945096/items/EZMGVZEJ"], "itemData":{"id":1003,"type":"article","title":"Seafood Products Association (SPA) -- Eurofins Pricing Document","author":[{"family":"Eurofins","given":""}],"issued":{"date-parts":[["2016"]]}}},{"id":1005,"uris":["http://zotero.org/groups/945096/items/S8P2VE4W"],"uri":["http://zotero.org/groups/945096/items/S8P2VE4W"],"itemData":{"id":1005,"type":"article", "title":"General Services Administration Federal Acquisition Services (FAS) Authorized Federal List Price","author":[{"family":"TestAmerica Laboratories","given":""}],"issued":{"date-parts":[["2015"]]}}},{"id":1004,"uris":["http://zotero.org/groups/945096/items/WXXSZDNZ"],"uri":["http://zotero.org/groups/945096/items/WXXSZDNZ"],"itemData":{"id":1004,"type":"article","title":"Laboratory Costs and Sampling Procedures","author":[{"literal":"Utah Department of Environmental Quality (Utah DEQ)"}],"issued":{"date-parts":[["2016"]]}}}]."schema":"https://github.com/citation-style-

parts":[["2016"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

#### 5.2.4 Costs per Administrative/Monitoring Event

Based on the hours per activity, the labor rates for PWSs and primacy agencies, and the analytical costs for monitoring, the EPA calculated a per-activity cost for PWSs and primacy agencies, as shown in [REF\_Ref523411577 \h]. As noted above, California and Massachusetts are assumed to incur upfront costs to read and understand the rule, but will not incur incremental costs to provide training, or to review waiver applications or monitoring reports.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Costs per Administrative/Monitoring Event for PWSs and Primacy Agencies

	PWS		Primacy Agencies		
Event	Small	Large	Included	Excluded	
Monitoring	\$99	\$99	\$51	\$0	
Waiver application	\$555	\$555	\$405	\$0	
Read the rule, provide training, adopt regulatory or programmatic changes	\$694	\$1,249	\$139,140	\$2,027	

#### 5.2.5 Number of Monitoring Events

The EPA estimated costs for two phases of perchlorate monitoring: the initial monitoring and long-term monitoring. The Agency assumed that initial perchlorate monitoring requirements will be phased in over a period of six years following the effective date. Monitoring requirements vary by size and type of system. Large CWSs will collect initial monitoring samples during the first three years following the effective date (i.e., years four to six of the analysis period); the EPA assumed that one-third of these systems would collect samples per year. Small CWSs and NTNCWSs would collect initial monitoring samples in the subsequent three-year period (i.e., years seven to nine of the analysis period). The EPA assumed that, within these periods, all systems would conduct initial monitoring – one year of quarterly monitoring to determine whether perchlorate concentrations are consistently and reliably below the proposed MCL.

The schedule for long-term monitoring is based on the EPA's Standardized Monitoring Framework for drinking water contaminants [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"HIJYZ8nR","properties": {"formattedCitation":"(USEPA, 1991; 2004)","plainCitation":"(USEPA, 1991;

2004)","noteIndex":0},"citationItems":[{"id":169,"uris":["http://zotero.org/groups/945096/items/ZSHKRHSY"],"uri":["http://zotero.org/groups/945096/items/ZSHKRHSY"],"itemData":{"id":169,"type":"article","title":"Standardized Monitoring

Framework", "author": [{"literal": "USEPA"}], "issued": {"date-

parts":[["1991",2]]}}},{"id":993,"uris":["http://zotero.org/groups/945096/items/XIN79HTB"],"uri":["http://zotero.org/groups/945096/items/XIN79HTB"],"itemData":{"id":993,"type":"article","title":"The Standardized Monitoring Framework: A Quick Reference Guide","publisher":"EPA-816-F-04-010","author":[{"family":"USEPA","given":""}],"issued":{"date-parts":[["2004"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. Under this framework, systems with MCL exceedances would continue to monitor quarterly, while systems below the MCL that obtain waivers will monitor annually for three years (surface water systems) or triennially for nine years (groundwater systems), and then incur costs for a waiver application. Thereafter, these systems will continue reduced monitoring – once every nine years – under subsequent waivers. Systems that are below the MCL without waivers will monitor once yearly (surface water systems) or once every three years (groundwater).

For other inorganic contaminants with MCLs currently in place (e.g., mercury), the EPA assumes that 90 percent of eligible systems with groundwater sources apply for and receive system-specific waivers; for eligible systems with surface water sources, this proportion is

40 percent [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"nRvPG48F","properties": {"formattedCitation":"(USEPA, 2008c)", "plainCitation":"(USEPA, 2008c)", "noteIndex":0}, "citationItems": [{"id":989, "uris": ["http://zotero.org/groups/945096/items/QSXYHBID"], "uri": ["http://zotero.org/groups/945096/items/QSXYHBID"], "itemData": {"id":989, "type": "article", "title": "Draft Information Collection Request for the Disinfectants/Disinfection Byproducts, Chemical, and Radionuclides Rule", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-parts": [["2008",6]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. The EPA also applied this assumption in this analysis.

[ REF \_Ref530148157 \h \\* MERGEFORMAT ] summarizes the timing of expected activities for affected entities based on the requirements.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Schedule of Administrative Requirements

Year	Primacy Agencies	Large CWS	NTNCWS and Small CWS		
1	Dood/understand mile adent				
2	Read/understand rule, adopt rule, and training				
3	Tuic, and training				
4	Review monitoring reports	0			
5	and waiver applications	Conduct initial monitoring; implement treatment			
6	]	implement deathert			
7		Monitor per Standardized	Candinat initial manitarians		
8		Monitoring Framework	Conduct initial monitoring implement treatment		
9		1	implement treatment		
10			Monitor per Standardized		
11			Monitoring Framework		
12					
13					
14					
15					
16	]				
17					
18					
19					
20 and after	₩	<b>\</b>	↓		

The monitoring schedule based on the Standardized Monitoring Framework is per entry point. To estimate the number of monitoring events per PWS, the EPA estimated the average number of entry points for very small, small, medium, large, and very large PWSs (with categories based on the population served) based on the occurrence data [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"NoXLDR17","properties":{"formattedCitation":"(USEPA, 2018d)","plainCitation":"(USEPA,

2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"uri":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlorate Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*\*\*,"author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], as described in Section [ REF \_Ref524938747 \r \h ]. [ REF \_Ref524939687 \h ] summarizes the average number of entry points per PWS based on population served. For PWSs that were included in the occurrence data, EPA used the number of entry points from that dataset. For PWSs not included in the occurrence data, EPA used the average number of entry points based on the population served from [ REF \_Ref524939687 \h ].

#### 5.3 Total Cost Results

The EPA estimated the costs over a 35-year analysis period and assumed that control technologies would be implemented by the end of the initial monitoring phase (e.g., by year six for large CWSs; if small CWSs or NTNCWSs incurred control costs, those costs would be phased in by year nine). The EPA calculated the present value of total costs in each year of the analysis period and discounted to rule finalization using both a 3 percent and 7 percent discount rate. [ REF \_Ref523412826 \h ] and [ REF \_Ref525891735 \h ] summarize the results at MCLs of 56  $\mu$ g/L and 18  $\mu$ g/L, respectively.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Summary of Total Annualized Costs at an MCL of  $56 \mu g/L$  (millions 2017\$)

Cost Component	3% Discount	7% Discount
Drinking water systems costs		
Treatment costs <sup>a</sup>	\$0.65	\$0.70
Monitoring and administration <sup>b</sup>	\$5.93	\$6.38
Drinking water systems total	\$6.58	\$7.07
State costs		
Administration	\$3.09	\$3.20
Total costs	\$9.67	\$10.28

a. The values shown are the mid-cost estimates. Low-cost estimates are 1% to 2% lower than the mid-cost estimates at the 3% and 7% discount rates, respectively; the high-cost estimates are 13% to 20% higher across discount rates. b. Costs include monitoring for all CWSs and NTNCWSs. Some consecutive systems that purchase 100% of their water from wholesale systems may not be required to monitor for perchlorate, provided the states allow integrated system agreements to include perchlorate among the monitoring requirements that the wholesale system fulfills for the consecutive system. The potential number of consecutive systems excluded from perchlorate monitoring depends on the system and state decisions and, therefore, is unknown. Excluding monitoring costs for approximately 8,400 consecutive systems that do not report a water source facility (e.g., well or intake) in SDWIS/FED from the monitoring cost analysis reduces annualized monitoring costs by \$0.8 million.

Note: Totals may not sum because of rounding.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Summary of Total Annualized Costs at an MCL of 18 µg/L (millions 2017\$)

Cost Component	UCN	UCMR 1		National <sup>a</sup>	
Cost Component	3% Discount	7% Discount	3% Discount	7% Discount	
Drinking water systems costs					
Treatment costs <sup>b</sup>	\$6.92	\$7.29	\$7.92	\$8.37	
Monitoring and administration <sup>c</sup>	\$5.94	\$6.38	\$5.94	\$6.38	
Drinking water systems total	\$12.85	\$13.67	\$13.86	\$14.75	
State costs					
Administration	\$3.09	\$3.21	\$3.09	\$3.21	
Total costs	\$15.95	\$16.88	\$16.95	\$17.96	

a. The EPA applied statistical sampling weights to the results to extrapolate small system results to national results. The entry point at which a measurement exceeds  $18~\mu g/L$  is 1~of~20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the six UCMR 1 systems in the stratum. Overall, the stratum population served accounts for 1.32% of the national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers may be exposed to perchlorate greater than  $18~\mu g/L$ . Based on this population estimate, the EPA calculated per-capita costs for the system and extrapolated them to national levels.

Note: Totals may not sum because of rounding.

#### 5.4 Household Costs

Water systems typically recover control costs through increased household rates, resulting in increased costs at the household level. To calculate the magnitude of this cost increase, the EPA first estimated the number of households that may incur costs as a result of the rule based on the population served by affected PWSs and the state-specific average household size [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"fs2jK6J2","properties": {"formattedCitation":"(U.S. Census Bureau, 2017c)","plainCitation":"(U.S. Census Bureau,

2017c)","noteIndex":0},"citationItems":[{"id":985,"uris":["http://zotero.org/groups/945096/items/CGU3LT9N"],"uri":["http://zotero.org/groups/945096/items/CGU3LT9N"],"itemData":{"id":985,"type":"article","title":"Average Household Size of Occupied Housing Units by Tenure. American Community Survey 1-Year Estimates: Table B25010","author":[{"family":"U.S.

Census Bureau", "given": ""}], "issued": {"date-

parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. For PWSs that are expected to incur control costs, EPA estimates that approximately 23,893 households will bear increase water rates under an MCL of 56  $\mu$ g/L. Under an MCL of 18  $\mu$ g/L, 264,361 households would incur control costs.

b. The values shown are the mid-cost estimates. Low-cost estimates are 1% to 2% lower than the mid-cost estimates at the 3% and 7% discount rates, respectively; the high-cost estimates are 13% to 20% higher across discount rates. c. Costs include monitoring for all CWSs and NTNCWSs. Some consecutive systems that purchase 100% of their water from wholesale systems may not be required to monitor for perchlorate, provided the states allow integrated system agreements to include perchlorate among the monitoring requirements that the wholesale system fulfills for the consecutive system. The potential number of consecutive systems excluded from perchlorate monitoring depends on the system and state decisions and, therefore, is unknown. Excluding monitoring costs for approximately 8,400 consecutive systems that do not report a water source facility (e.g., well or intake) in SDWIS/FED from the monitoring cost analysis reduces annualized monitoring costs by \$0.8 million.

The EPA divided the total annual PWS-level costs by the number of households served by the system.

[ REF \_Ref525046550 \h ] summarizes the results. [ REF \_Ref535312062 \n \h ] provides this calculation for each entry point expected to incur control costs. This approach may result in an overestimation of household costs because it assumes that all control costs will be passed to residential customers, although some costs may accrue to industrial or commercial customers.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Summary of Household-Level Annual Control Costs (2017\$)

	3% Discount <sup>a</sup>	7% Discount <sup>a</sup>
$MCL = 56 \mu g/L$		
Minimum	\$11	\$14
Average	\$40	\$47
Maximum	\$69	\$80
MCL = 18 μg/L <sup>b</sup>		
Minimum	\$18	\$24
Average	\$38	\$46
Maximum	\$72	\$84

a. See [ REF \_Ref535312080 \n \h ] for detailed calculations for all systems.

b. Note that the household-level costs are the same whether or not the small systems costs are extrapolated because the extrapolation is based on per-capita estimated costs.

#### 6 Comparison of Benefits and Costs

#### 6.1 Introduction

This chapter provides a comparison of benefits and costs for each of the regulatory alternatives. The Agency analyzed the costs and benefits of regulating perchlorate concentrations in drinking water to two different MCL standards. In both instances, the MCL is equal to the corresponding MCLG. The alternative MCLG values reflect two approaches to establishing a perchlorate level that is protective of human health.

#### 6.2 Summary of National Costs and Benefits

#### 6.2.1 National Cost Estimates

National compliance costs to PWSs for treatment (both annualized capital and O&M costs); monitoring and administrative activities; and costs to states, including any one-time start-up costs, for regulatory implementation and enforcement, were estimated and described in Chapter 5. [ REF \_Ref525876134 \h ] provides a summary of the national costs for PWSs to comply with the two MCL alternatives for two alternative discount rates.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Summary of Total Annual Costs by Alternative (2017\$)

.,,	onala			
MCL Alternative	3% Discount	7% Discount	3% Discount	7% Discount
Preferred MCL (56 μg/L)	\$9.67	\$10.28	\$9.67	\$10.28
Alternative MCL (18 μg/L)	\$15.95	\$16.88	\$16.95	\$17.96

a. For the proposed MCL of  $56~\mu g/L$ , the national estimates are the same as the estimates based on UCMR 1 data because there were no small system sample results to extrapolate to national small system estimates. For an MCL of  $18~\mu g/L$ , the EPA applied statistical sampling weights to the results to extrapolate small system results to national results. The entry point at which a measurement exceeds  $18~\mu g/L$  is 1~of~20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the 6~UCMR~1 systems in the stratum (40,574). Currently, the stratum population of 775,000 accounts for 1.32% of the 58.7 million national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers (approximately 41,100) may be exposed to perchlorate greater than  $18~\mu g/L$ . Based on this population estimate, the EPA calculated per-capita costs for the system and extrapolated them to national levels.

#### **6.2.2 National Benefits Estimates**

Chapter 4 provided a description of the expected health effects benefits of regulating perchlorate and described a method to quantify avoided IQ decrements in offspring born to pregnant women exposed to perchlorate above alternative MCLs. The Agency estimated national benefits based on discounted lifetime differential earnings estimates for one-point IQ decrements. [ REF \_Ref525876235 \h ] provides a summary of the national benefits for PWSs to comply with the two MCL alternatives for two alternative discount rates.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Summary of Annual Control Benefits by Alternative (Central Estimate; 2017\$)

MOI Allanastina	UCN	MR 1	Natio	onal <sup>a</sup>
MCL Alternative	3% Discount	7% Discount	3% Discount	7% Discount
Preferred MCL (56 μg/L)	\$2.12	\$0.36	\$2.12	\$0.36
Alternative MCL (18 μg/L)	\$3.87	\$0.65	\$3.91	\$0.66

a. For the proposed MCL of  $56 \,\mu g/L$ , the national estimates are the same as the estimates based on UCMR 1 data because there were no small system sample results to extrapolate to national small system estimates. For an MCL of  $18 \,\mu g/L$ , the EPA applied statistical sampling weights to the results to extrapolate small system results to national results. The entry point at which a measurement exceeds  $18 \,\mu g/L$  is 1 of 20 in its sample stratum; no other sample in the stratum had a measurement of perchlorate greater than the minimum reporting level. The entry point population of 2,155 represents 5.31% of the total population served by the 6 UCMR 1 systems in the stratum (40,574). Currently, the stratum population of 775,000 accounts for 1.32% of the 58.7 million national population served by small systems. Thus, the UCMR 1 results indicate that 0.07% (5.31% x 1.32%) of small system customers (approximately 41,100) may be exposed to perchlorate greater than  $18 \,\mu g/L$ .

#### 6.3 Comparison of Benefits and Costs

This section provides three comparisons of benefits and costs: a direct comparison of national incremental costs and benefits, a cost-effectiveness analysis, and a break-even analysis. The comparisons include benefits and costs for the proposed MCL of 56  $\mu$ g/L and the alternative MCL of 18  $\mu$ g/L.

#### 6.3.1 National Benefit-Cost Comparison

The impact of the proposed rule on national benefits and costs indicates that promulgating an NPDWR for perchlorate is unlikely to result in positive net benefits. [ REF \_Ref530092144 \h ] shows costs, benefits, and net benefits for both MCLs. In both instances, net benefits are negative because costs exceed benefits. The exhibit also shows the incremental costs, benefits, and net benefits between the two MCLs.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Comparison of Incremental Costs and Benefits for Proposed Rule (millions 2017\$)

	UCN	AR 1	National <sup>a</sup>	
Item	3% Discount	7% Discount	3% Discount	7% Discount
MCL = 56 μg/L				
Total annual costs	\$9.67	\$10.28	\$9.67	\$10.28
Total annual quantified benefits	\$2.12	\$0.36	\$2.12	\$0.36
Total annual quantified net benefits	-\$7.55	-\$9.92	-\$7.55	-\$9.92
$MCL = 18 \mu g/L$				
Total annual costs	\$15.95	\$16.88	\$16.95	\$17.96
Total annual quantified benefits	\$3.87	\$0.65	\$3.91	\$0.66
Total annual quantified net benefits	-\$12.07	-\$16.23	-\$13.04	-\$17.30
Incremental between 56 and 18 µg/L				
Incremental annual costs	\$6.28	\$6.60	\$7.28	\$7.68
Incremental annual quantified benefits	\$1.75	\$0.29	\$1.79	\$0.30
Incremental annual quantified net benefits	-\$4.53	-\$6.31	-\$5.49	-\$7.38

a. For the proposed MCL of  $56~\mu g/L$ , the national estimates are the same as the estimates based on the UCMR 1 data because there were no small system sample results to extrapolate to national small system estimates. At an MCL of  $18~\mu g/L$ , national estimates include extrapolation for 1 small system entry point to national estimates based on sampling weights, described above.

Note: There are some slight variations in net benefit differences due to rounding.

#### 6.3.2 Cost-Effectiveness Analysis

Cost-effectiveness analysis provides an alternative way of evaluating costs and benefits. Typically, the analysis involves dividing costs by the quantified benefits such as avoided cases of morbidity. For perchlorate, the EPA considered the cost per avoided IQ decrement. [ REF \_Ref530093447 \h ] shows the cost-effectiveness inputs – total annual costs and annual avoided IQ decrements – and the results for the proposed MCL, the alternative MCL, and the incremental impact between the two MCLs.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Costs per IQ Decrement Avoided (millions 2017\$)

ltom	UCMR 1		National <sup>a</sup>					
Item	3% Discount	7% Discount	3% Discount	7% Discount				
MCL = 56 μg/L	$MCL = 56 \mu g/L$							
Total annual costs	\$9.67	\$10.28	\$9.67	\$10.28				
Annual avoided IQ decrement (central)	136.12	\$136.12	136.12	136.12				
Cost per avoided IQ decrement	\$0.07	\$0.08	\$0.07	\$0.08				
$MCL = 18 \mu g/L$								
Total annual costs	\$15.95	\$16.88	\$16.95	\$17.96				
Annual avoided IQ decrement (central)	248.20	248.20	250.90	250.90				
Cost per avoided IQ decrement	\$0.06	\$0.07	\$0.07	\$0.07				
Incremental between 56 and 18 µg/L								
Total annual costs	\$6.28	\$6.60	\$7.28	\$7.68				
Annual avoided IQ decrement (central)	112.08	112.08	\$114.78	\$114.78				
Cost change per avoided IQ decrement	-\$0.01	-\$0.01	\$0.00	-\$0.01				

a. For the proposed MCL of  $56~\mu g/L$ , the national estimates are the same as the estimates based on the UCMR 1 data because there were no small system sample results to extrapolate to national small system estimates. At an MCL of  $18~\mu g/L$ , national estimates include extrapolation for 1 small system entry point to national estimates based on sampling weights, described above.

#### 6.3.3 Break-Even Analysis

When costs exceed benefits, a break-even analysis identifies what level of quantifiable health risk reduction would be needed to generate benefits that equal costs. For perchlorate, the quantifiable and monetizable health endpoint is avoided IQ decrements. [ REF \_Ref530094172 \h ] shows the inputs to this analysis are the annual costs and unit values for a one-point IQ loss reported in Section [ REF \_Ref530094326 \r \h ]. Dividing the annual cost by the unit value generates an estimate of the avoided IQ decrements that would be needed for benefits to equal costs. For the proposed MCL, these estimates are substantially higher than the estimated rule impact on avoided IQ decrements.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Break-Even Analysis Results (millions 2017\$)

	UCN	UCMR 1 Nati		ional <sup>a</sup>	
Item	3% Discount	7% Discount	3% Discount	7% Discount	
MCL = 56 μg/L					
Total annual costs	\$9.67	\$10.28	\$9.67	\$10.28	
Value per IQ point	\$0.020	\$0.004	\$0.020	\$0.004	
Break-even avoided IQ decrements	487.53	2,674.42	487.53	2,674.42	
$MCL = 18 \mu g/L$					
Total annual costs	\$15.95	\$16.88	\$16.95	\$17.96	
Value per IQ point	\$0.020	\$0.004	\$0.020	\$0.004	
Break-even avoided IQ decrements	803.79	4,393.37	854.23	4,674.45	
Incremental between 56 and 18 µg/L					
Total annual costs	\$6.28	\$6.60	\$7.28	\$7.68	
Value per IQ point	\$0.00	\$0.00	\$0.00	\$0.00	
Break-even avoided IQ decrements	316.26	1,718.95	366.70	2,000.03	

a. For the proposed MCL of  $56~\mu g/L$ , the national estimates are the same as the estimates based on the UCMR 1 data because there were no small system sample results to extrapolate to national small system estimates. At an MCL of  $18~\mu g/L$ , national estimates include extrapolation for 1 small system entry point to national estimates based on sampling weights, described above.

#### 6.3.4 Summary of Conclusions

The analysis of costs to implement the proposed rule primarily includes administrative and monitoring costs. The Agency expects very few systems to incur treatment costs to reduce perchlorate from baseline concentrations to comply with the proposed MCL of  $56~\mu g/L$  or the alternative MCL of  $18~\mu g/L$ . Because the proposed rule has little impact on drinking water quality, the corresponding health risk reductions are low and estimates of benefits are more than an order of magnitude less than costs.

# 6.4 Effect of Uncertainties and Non-Quantified Benefit/Cost Estimates on the Estimation of National Benefits and Costs

Estimates of regulatory benefits and costs are subject to a variety of limitations such as data availability and underlying variability. Section [ REF \_Ref530140173 \r \h ] provides an overview of several sources of uncertainty regarding the economic estimates reported above. Section [ REF \_Ref530140182 \r \h ] recognizes that the quantitative benefits analysis is limited to a single health endpoint.

#### 6.4.1 Summary of Major Uncertainties in the Cost and Benefit Analyses

Uncertainties regarding the economic analysis can be grouped into three general categories: baseline occurrence, benefits analysis, and cost analysis. [ REF \_Ref530140246 \h ] characterizes the uncertainties and their potential effects on estimated costs and benefits.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Sources of Uncertainty in Economic Analysis

Description	Potential Effect <sup>a</sup>
Baseline occurrence	
UCMR 1 data are more than one decade old; actual occurrence could be lower	± (benefits and costs
(e.g., because of contaminant cleanup) or higher (e.g., because new systems	will change in the
use perchlorate-contaminated source water)	same direction)
UCMR 1 data include a sample of small systems; the Stage 1 results (entry point	<ul> <li>(benefits and costs)</li> </ul>
maximums) indicate that no small systems would exceed 56 μg/L and that one	will change in the
small system would exceed 18 µg/L; it is possible that there are additional small	same direction)
systems where the baseline perchlorate is greater than either MCL	
The EPA assumed a uniform distribution of system population served across the	± (benefits and costs
entry points; the actual entry point service population could be greater than or	will change in the
less than the estimates	same direction)
The EPA estimated entry point maximum concentrations using the MRL of 4 µg/L	+ (benefits and costs
as a substitution value for nondetection sample results; the actual entry point	will change in the
mean concentrations could be lower	same direction)
Benefits analysis	·
The health risks and risk reductions are based on maximum recorded	
concentration estimates and thus do not account for exposures to concentrations	± (benefits only)
greater than or less than this recorded maximum	
The EPA assumed that baseline fT4 is equal to the median, which likely	
underestimates disease benefits as the logarithmic relationship between	- (benefits only)
maternal fT4 and child IQ leads to larger relative changes in fT4, with increasing	(Borionts orny)
levels of perchlorate and lower levels of baseline fT4	
The EPA assumed a median TSH feedback loop strength for the exposed	
population does not incorporate the variability in the feedback mechanism of the	± (benefits only)
body's creation of TSH in response to decreasing fT4	
The benefits analysis is based on a single health endpoint (see Section [ REF	- (benefits only)
_Ref535175449 \r \h ])	(201101110 0111)
Cost analysis	
The EPA assumed that systems requiring treatment would incorporate a safety	+ (benefits and costs
factor – treating to 80% of the proposed MCL or alternative MCL, which	will change in the
increases costs and benefits	same direction)
The EPA assumed that all systems requiring treatment would implement ion	± (costs only)
exchange, which may overestimate or underestimate costs	± (costs offig)
The EPA developed a monitoring schedule that assumed a uniform distribution of	± (costs only)
initial monitoring costs over three years; actual costs will vary	= (000to Offig)
The EPA assumed that long-term monitoring costs would occur in the last year of	
the applicable three-year monitoring period or nine-year monitoring cycle;	- (costs only)
systems may conduct monitoring in an earlier year of the period or cycle	
The EPA assumed that 90% of groundwater systems and 40% of surface water	
systems obtained perchlorate monitoring waivers; the actual percentages may	± (costs only)
vary	

a. A "—" symbol indicates that benefits and/or costs will tend to be underestimated. A "+" symbol indicates that benefits and/or costs will tend to be overestimated. A "±" symbol indicates an unknown direction of uncertainty (i.e., benefits and/or costs could be underestimated or overestimated).

#### 6.4.2 Summary of Non-Quantified Costs and Benefits

In addition to the monetized benefits, several other benefits of reducing perchlorate exposure have not been quantified. These consist of other health effects associated with perchlorate due to its alteration of iodine and thyroid hormone levels: additional neurological endpoints related maternal hypothyroxinemia such as the offspring's increased risk of schizophrenia [ADDIN]

#### ZOTERO ITEM CSL CITATION

{"citationID":"WNqAF5i1","properties":{"formattedCitation":"(Gyllenberg et al., 2016)", "plainCitation": "(Gyllenberg et al., 2016)", "noteIndex":0}, "citationItems":[{"id":24, "uris":["http://zotero.org/groups/945096/items/9 37P2NGT"], "uri": ["http://zotero.org/groups/945096/items/937P2NGT"], "itemData": {"id":24, "tv pe":"article-journal", "title": "Hypothyroxinemia during gestation and offspring schizophrenia in a national birth cohort", "container-title": "Biological Psychiatry", "page": "962-970", "volume": "79", "issue": "12", "source": "PubMed", "abstract": "BACKGROUND: Evidence from animal and human studies indicates that thyroid hormone deficiency during early gestation alters brain development. As schizophrenia is associated with prenatal brain insults and premorbid cognitive deficits, we tested the a priori hypothesis that serologically defined maternal thyroid deficiency during early gestation to mid-gestation is associated with schizophrenia in offspring.\nMETHODS: The investigation is based on the Finnish Prenatal Study of Schizophrenia, a nested case-control study that included archived maternal sera from virtually all pregnancies since 1983 (N = > 1 million). We identified all offspring in the cohort with a diagnosis of schizophrenia based on the national inpatient and outpatient register and matched them on sex, date of birth, and residence in Finland at time of onset of the case to comparison subjects (1:1) from the cohort. Maternal sera of 1010 case-control pairs were assessed for free thyroxine, and sera of 948 case-control pairs were assessed for thyroid-stimulating hormone.\nRESULTS: Maternal hypothyroxinemia (free thyroxine ≤10th percentile, normal thyroid-stimulating hormone) was associated with an increased odds of schizophrenia (odds ratio = 1.75, 95% confidence interval = 1.22-2.50, p = .002). When adjusted for maternal psychiatric history, province of birth, and maternal smoking during pregnancy, the association remained significant (odds ratio = 1.70, 95% confidence interval = 1.13-2.55, p = .010).\nCONCLUSIONS: In a large, national birth cohort, prospectively documented hypothyroxinemia during early gestation to mid-gestation was associated with increased odds of schizophrenia in offspring. This information can inform translational studies of maternal hypothyroxinemia examining molecular and cellular deviations relevant to schizophrenia.","DOI":"10.1016/j.biopsych.2015.06.014","ISSN":"1873-2402","note":"PMID: 26194598\nPMCID: PMC4684794", "journalAbbreviation": "Biol. Psychiatry", "language": "eng", "author": [{"family": "Gyllenberg", "given": "David"}, {"family": "So urander", "given": "Andre"}, {"family": "Surcel", "given": "Heljä-Marja"}, {"family": "Hinkka-Yli-Salomäki", "given": "Susanna" }, { "family": "McKeague", "given": "Ian W."}, {"family": "Brown", "given": "Alan S."}], "issued": {"dateparts":[["2016",6,15]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ], ADHD, and expressive language delay; and CVD. Section [ REF Ref523470708 \r \h \\* MERGEFORMAT ] provided some discussion regarding these endpoints.

Other benefits that are not associated with health risk reductions include improved perception of water quality (see Section [ REF \_Ref325447503 \r \h \\* MERGEFORMAT ]) and the possibility of reducing other contaminants if perchlorate is reduced (see Section [ REF \_Ref325447528 \r \h \\* MERGEFORMAT ]). Given the relatively few systems that might need to remove perchlorate, the potential benefits of these impacts are also minimal.

#### 7 Administrative Requirements

This section provides information required by several federal statutes and Executive Orders.

# 7.1 Executive Order 12866: Regulatory Planning and Review and Executive Order 13563: Improving Regulation and Regulatory Review

Under Executive Order 12866, entitled "Regulatory Planning and Review" (58 FR 51735, October 4, 1993) and reaffirmed by Executive Order 13563, entitled "Improving Regulation and Regulatory Review" (76 FR 3821, January 21, 2011), the Agency must determine whether the regulatory action is "significant" and therefore subject to OMB review and the requirements of the Executive Order. The Order defines "significant regulatory action" as one that is likely to result in a rule that may: (1) Have an annual effect on the economy of \$100 million or more or adversely affect in a material way the economy, a sector of the economy, productivity, competition, jobs, the environment, public health or safety, or State, local, or tribal governments or communities; (2) Create a serious inconsistency or otherwise interfere with an action taken or planned by another agency; (3) Materially alter the budgetary impact of entitlements, grants, user fees, or loan programs or the rights and obligations of recipients thereof; or (4) Raise novel legal or policy issues arising out of legal mandates, the President's priorities, or the principles set forth in the Executive Order. Although the cost estimates in Chapter 5 do not exceed the annual cost threshold, the EPA determined that this rule is a "significant regulatory action" because it raises novel legal or policy issues. Accordingly, the EPA submitted this action to OMB for review. Changes made in response to OMB recommendations have been documented in the docket for this action.

#### 7.2 Paperwork Reduction Act

The information collection requirements in this proposed rule have been submitted for approval to OMB under the Paperwork Reduction Act, 44 U.S.C. 3501 *et seq*. The information collection requirements are not enforceable until OMB approves them.

The information collected as a result of this rule will allow the States and the EPA to evaluate compliance with the rule. For the first 3-year period following rule promulgation, the major information requirements concern primacy agency activities to implement the rule. Compliance actions for drinking water systems (including monitoring, administration, and treatment costs) do not begin until after Year 3.

The estimate of annual average burden hours for the proposed rule during the first three years following promulgation is 48,539 hours. The annual average cost estimate is \$7.4 million for labor. The burden hours per response is 2,648 hours and the cost per response is \$134,154. The frequency of response (average responses per respondent) is one for primacy agencies, annually (for upfront administrative activities to implement the rule). The estimated number of likely respondents is 55 over the three year period (for an average of 18.3 each year).

Burden means the total time, effort, or financial resources expended by persons to generate, maintain, retain, or disclose or provide information to or for a federal agency. This includes the

time needed to review instructions; develop, acquire, install, and utilize technology and systems for the purposes of collecting, validating, and verifying information, processing and maintaining information, and disclosing and providing information; adjust the existing ways to comply with any previously applicable instructions and requirements; train personnel to be able to respond to a collection of information; search data sources; complete and review the collection of information; and transmit or otherwise disclose the information.

An agency may not conduct or sponsor, and a person is not required to respond to a collection of information unless it displays a currently valid OMB control number. The OMB control numbers for EPA's regulations in 40 CFR are listed in 40 CFR part 9.

#### 7.3 Regulatory Flexibility Act

The Regulatory Flexibility Act (RFA) of 1980, as amended by the Small Business Regulatory Enforcement Fairness Act (SBREFA) of 1996, generally requires an agency to prepare an initial regulatory flexibility analysis for any proposed rule subject to notice and comment rulemaking requirements under the Administrative Procedure Act or other statute unless the agency certifies that the rule will not have a significant economic impact on a substantial number of small entities. Small entities include small businesses, small organizations, and small governmental jurisdictions.

For purposes of assessing the impacts of today's proposed rule on small entities, the EPA considered small entities to be public water systems serving 10,000 or fewer persons. This is the threshold specified by Congress in the 1996 Amendments to the Safe Drinking Water Act for small system flexibility provisions. In accordance with the RFA requirements, the EPA proposed using this alternative definition in the *Federal Register*, (63 FR 7620, February 13, 1998), requested public comment, consulted with the Small Business Administration (SBA), and expressed its intention to use the alternative definition for all future drinking water regulations in the Consumer Confidence Reports regulation (63 FR 44511, August 19, 1998). As stated in that final rule, the alternative definition is applied to this proposed regulation.

The proposed rule contains provisions that will affect 58,325 CWS and NTNCWS serving 10,000 or fewer people. To meet the proposed rule requirements, all of these systems will need to conduct perchlorate monitoring. At the proposed MCL of 56  $\mu$ g/L, the UCMR 1 monitoring data indicate that no small systems would incur costs to reduce the levels of perchlorate in drinking water. Therefore, all small PWSs will incur monitoring costs only, as shown in [ REF Ref535312437 \h ].

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Number of PWSs by Size and Ownership

Owner Type	Numb	er of Sys	stems	With	Control C	osts	Monitoring Costs Only			
Owner Type	Total	Large	Small	Total Large Small			Total Large		Small	
Public/other <sup>a</sup>	30,181	3,366	26,815	2	2	0	30,179	3,364	26,815	
Private	31,895	385	31,510	0	0	0	31,895	385	31,510	
Total	62,076	3,751	58,325	2	2	0	62,074	3,749	58,325	

a. Includes the following owner types: local, state, and federal government, Native American, public/private, and missing owner type for some NTNCWS.

Total annual monitoring and administrative costs for PWSs are approximately \$6.6 million to \$7.1 million annually ([REF\_Ref523412826 \h]), with \$5.1 million to \$5.5 million accruing to small PWSs. Based on 58,325 small systems, this yields an average annual per-system cost of \$88 (3% discount rate) to \$94 (7% discount rate).

The EPA estimated the average annual revenues for PWSs based on the 2006 Community Water System Survey [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"WbV8Ae9u","properties":{"formattedCitation":"(USEPA, 2009c)","plainCitation":"(USEPA,

2009c)","noteIndex":0},"citationItems":[{"id":924,"uris":["http://zotero.org/groups/945096/items/DZNAAV6M"],"uri":["http://zotero.org/groups/945096/items/DZNAAV6M"],"itemData":{"id ":924,"type":"article","title":"2006 Community Water System Survey - Volume II: Detailed Tables and Survey

Methodology","URL":"https://www.epa.gov/dwstandardsregulations/community-water-system-survey","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2009",5]]},"accessed":{"date-parts":[["2018",8,17]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"}], which includes data on revenues for private and public PWSs, and updated all values to 2017\$ based on BLS [ADDIN ZOTERO\_ITEM CSL CITATION

{"citationID":"IadU1bbg","properties": {"formattedCitation":"(2018a)","plainCitation":"(2018a)", "noteIndex":0}, "citationItems":[{"id":984,"uris":["http://zotero.org/groups/945096/items/E3I7H RK8"],"uri":["http://zotero.org/groups/945096/items/E3I7HRK8"],"itemData": {"id":984,"type":" article","title":"Chained consumer price index for fuels and utilities in U.S. city average, all urban consumers, 2000 to 2018","author":[{"literal":"Bureau of Labor Statistics (BLS)"}],"issued": {"date-parts":[["2018"]]}}, "suppress-

author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Based on a comparison of annual costs to revenues shown in [ REF \_Ref535312451 \h ], the EPA does not expect small entities to incur costs that exceed one percent of revenue.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Annual Monitoring and Administrative Costs as a Percentage of Average Annual Revenue for Small PWSs

Average Annual

**Average Annualized Costs** 

Small	Average Annual	(as a percent of revenue)						
Systems	Revenues	3% Discount	7% Discount					
22,716	\$1,409,252	\$88 (0.006%)	\$94 (0.007%)					
31,510	\$518,439	\$88 (0.017%)	\$94 (0.018%)					
762	\$1,409,252	\$88 (0.006%)	\$94 (0.007%)					
600	\$1,409,252	\$88 (0.006%)	\$94 (0.007%)					
751	\$1,409,252	\$88 (0.006%)	\$94 (0.007%)					
1,825	\$518,439	\$88 (0.017%)	\$94 (0.018%)					
161	\$518,439	\$88 (0.017%)	\$94 (0.018%)					
58,325	\$838,719	\$88 (0.010%)	\$94 (0.010%)					
"properties": {"for PA, 2009c Table onItems": [ {"id": 92 45096/items/DZN Survey - Volume os://www.epa.gov/ :"USEPA" } ], "issu uffix": "Table 65" } er/csl-citation.jsor S. city average, all	mattedCitation":"(USE 24,"uris":["http://zotero [AAV6M"],"itemData" II: Detailed Tables and (dwstandardsregulation led":{"date-parts":[["26],"schema":"https://git ""}] and updated to 20 urban consumers [ AD	EPA, 2009c Table  o.org/groups/945096/items/l ":{"id":924,"type":"article", l Survey as/community-water-system 009",5]]},"accessed":{"date hub.com/citation-style- 17\$ based on the chained co DDIN ZOTERO_ITEM CSI	"title":"2006 c- consumer price indexCITATION					
	22,716 31,510 762 600 751 1,825 161 58,325 DDIN ZOTERO_I "properties": { "for PA, 2009c Table confitems": [ { "id":92 45096/items/DZN Survey - Volume os://www.epa.gov/:"USEPA" } ],"issu uffix": "Table 65" } er/csl-citation.jsor S. city average, all	Systems         Revenues <sup>4</sup> 22,716         \$1,409,252           31,510         \$518,439           762         \$1,409,252           600         \$1,409,252           751         \$1,409,252           1,825         \$518,439           161         \$518,439           58,325         \$838,719           DDIN ZOTERO_ITEM CSL_CITATION, "properties": {"formattedCitation": "(USI PA, 2009c Table onlitems": {"id":924,"uris": ["http://zotero/45096/items/DZNAAV6M"], "itemData' Survey - Volume II: Detailed Tables and os://www.epa.gov/dwstandardsregulation: :"USEPA"}], "issued": {"date-parts": [["20, 20, 20, 20, 20, 20, 20, 20, 20, 20,	Small Systems         Average Annual Revenues <sup>a</sup> (as a percent of a perce					

### 7.4 Unfunded Mandates Reform Act

Title II of the Unfunded Mandates Reform Act of 1995 (UMRA), Public Law 104–4, establishes requirements for federal agencies to assess the effects of their regulatory actions on State, local, and tribal governments and the private sector. Under section 202 of UMRA, the EPA generally must prepare a written statement, including a cost-benefit analysis, for proposed and final rules with "federal mandates" that may result in expenditures to State, local, and tribal governments, in the aggregate, or to the private sector of \$100 million or more in any one year, adjusted annually for inflation, or \$156 million based on the most recent guidance [ ADDIN ZOTERO\_ITEM CSL CITATION

2018a)", "noteIndex": 0}, "citationItems": [{"id":984, "uris": ["http://zotero.org/groups/945096/items/E3I7HRK8"], "uri" :["http://zotero.org/groups/945096/items/E3I7HRK8"],"itemData":{"id":984,"type":"article","title":"Chained

parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

consumer price index for fuels and utilities in U.S. city average, all urban consumers, 2000 to

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{"citationID": "hX46uSy1", "properties": {"formattedCitation": "(U.S. Department of Transportation, 2016)", "plainCitation": "(U.S. Department of Transportation, 2016)", "noteIndex":0}, "citationItems":[{"id":1269, "uris":["http://zotero.org/groups/945096/item s/GV8GHL44"],"uri":["http://zotero.org/groups/945096/items/GV8GHL44"],"itemData":{"id":1 269, "type": "article", "title": "Departmental Guidance: Threshold of Significant Regulatory Actions Under the Unfunded Mandates Report Act of

1995.","URL":"https://www.transportation.gov/sites/dot.gov/files/docs/Guidance%20-%20Threshold%20of%20Significant%20Regulatory%20Actions%20Under%20the%20Unfunde d%20Mandates%20Reform%20Act%20of%201995.pdf","author":[{"literal":"U.S. Department of Transportation"}],"issued":{"date-parts":[["2016"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Based on the cost estimates detailed in Section [ REF \_Ref525051776 \r \h ], the EPA determined that compliance costs in any given year will be below the threshold set in UMRA, with maximum single-year costs of approximately \$16 million. The EPA has determined that this rule contains a federal mandate that will not result in expenditures of \$100 million or more for State, local, and tribal governments, in the aggregate, or the private sector in any one year.

#### 7.5 Executive Order 12898: Federal Actions to Address Environmental Justice in Minority Populations and Low-Income Populations

Executive Order 12898, entitled "Federal Actions to Address Environmental Justice in Minority Populations and Low-Income Populations" (59 FR 7629, February 16, 1994), establishes federal executive policy on environmental justice. Its main provision directs federal agencies, to the greatest extent practicable and permitted by law, to make environmental justice part of their mission by identifying and addressing, as appropriate, disproportionately high and adverse human health or environmental effects of their programs, policies, and activities on minority and low-income populations in the United States.

The EPA has determined that this proposed rule will not have a disproportionately high and adverse human health or environmental effects on minority or low-income populations because it increases the level considered environmental protection for all affected populations without having any disproportionately high and adverse human health or environmental effects on any population including any minority or low-income population.

# 7.6 Executive Order 13045: Protection of Children from Environmental Health Risks and Safety Risks

Executive Order 13045, entitled "Protection of Children from Environmental Health Risks and Safety Risks" (62 FR 19885, April 23, 1997), applies to any rule that: (1) Is determined to be "economically significant" as defined under Executive Order 12866, and (2) concerns an environmental health or safety risk that the EPA has reason to believe may have a disproportionate effect on children. If the regulatory action meets both criteria, the Agency must evaluate the environmental health or safety effects of the planned rule on children and explain why the planned regulation is preferable to other potentially effective and reasonably feasible alternatives considered by the Agency.

This proposed rule is not "economically significant" as defined under Executive Order 12866; however, the environmental health risk addressed by this action may have a disproportionate effect on children. Accordingly, and consistent with Executive Order 13045 and the EPA's Policy on Evaluating Health Risks to Children [ ADDIN ZOTERO ITEM CSL CITATION

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2018c)", "plainCitation": "(USEPA,
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"id":1256, "type": "article", "title": "Memorandum: Reaffirmation of EPA's 1995 Policy on
Evaluating Health Risks to
Children", "author": [{"family": "USEPA", "given": ""}], "issued": {"date-
parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ], the EPA evaluated the environmental health or
safety effects of perchlorate on children. The results of this evaluation are contained in the
Health Effects Technical Support Document [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"WJ6JKkg7","properties":{"formattedCitation":"(SAB and USEPA,
2013)", "plainCitation": "(SAB and USEPA,
2013)", "noteIndex":0}, "citationItems":[{"id":1263, "uris":["http://zotero.org/groups/945096/item
s/3MNU7GPK"], "uri": ["http://zotero.org/groups/945096/items/3MNU7GPK"], "itemData": {"id":
1263, "type": "article", "title": "SAB Advice on Approaches to Derive a Maximum Contaminant
Level Goal for Perchlorate. EPA-SAB-13-
004", "author": [{"literal": "SAB"}, {"family": "USEPA", "given": ""}], "issued": {"date-
parts":[["2013"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] and described in Chapter 4. The EPA has
evaluated the risk associated with perchlorate in drinking water for the sensitive population –
offspring of pregnant women exposed to perchlorate during the first trimester – and developed a
proposed MCLG that is protective of this population as well as other children. The EPA has also
estimated the health risk reduction of the proposed and alternative MCLs. This analysis is
described in Chapter [ REF Ref523452255 \r \h ].
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#### 7.7 Executive Order 13132: Federalism

Executive Order 13132, entitled "Federalism" (64 FR 43255, August 10, 1999) requires the EPA to develop an accountable process to ensure "meaningful and timely input by State and local officials in the development of regulatory policies that have federalism implications." "Policies that have federalism implications" is defined in the Executive Order to include regulations that have "substantial direct effects on the States, on the relationship between the national government and the States, or on the distribution of power and responsibilities among the various levels of government."

Under Executive Order 13132, the EPA may not issue a regulation that has federalism implications, that imposes substantial direct compliance costs, and that is not required by statute, unless the federal government provides the funds necessary to pay the direct compliance costs incurred by State and local governments, or the EPA consults with State and local officials early in the process of developing the proposed regulation.

The EPA has concluded that this proposed rule may have federalism implications, because it may impose direct compliance costs on State or local governments, and the federal government will not provide the funds necessary to pay those costs. However, the EPA estimates that the proposed rule will not result in substantial expenditures (i.e., in excess of \$100 million in any one year) by State and local governments. Annual costs are estimated to range from \$9.7 million

at a 3 percent discount rate to \$10.3 million using a 7 percent, with \$6.8 million to \$7.2 million annually accruing to public entities.

# 7.8 Executive Order 13175: Consultation and Coordination with Indian Tribal Governments

Executive Order 13175, entitled "Consultation and Coordination with Indian Tribal Governments" (65 FR 67249, November 9, 2000) requires the EPA to develop "an accountable process to ensure meaningful and timely input by Tribal officials in the development of regulatory policies that have Tribal implications." The definition of "policies that have Tribal implications" includes regulations that have "substantial direct effects on one or more Indian tribes, on the relationship between the federal government and the Indian Tribes, or on the distribution of power and responsibilities between the Federal government and Indian Tribes." Under Executive Order 13175, the EPA may not issue a regulation that has Tribal implications, that imposes substantial direct compliance costs, and that is not required by statute, unless the federal government provides the funds necessary to pay the direct compliance costs incurred by Tribal governments, or the EPA consults with Tribal officials early in the process of developing the proposed regulation and develops a Tribal summary impact statement.

The EPA has concluded that this proposed rule may have Tribal implications, because it may impose direct compliance costs on Tribal governments, and the federal government will not provide the funds necessary to pay those costs. The EPA has identified 768 water systems with 1,167 entry points under Native American ownership that may be subject to the proposed rule. They will bear an estimated total annualized cost of \$74,095 at a 3 percent discount rate (\$79,620 at 7 percent) to implement this rule as proposed, with all costs attributable to monitoring and administrative costs. Estimated average annualized cost per system ranges from \$96 at a 3 percent discount rate to \$104 at a 7 percent discount rate.

Accordingly, the EPA provides the following Tribal summary impact statement as required by section 5(b) of Executive Order 13175. The EPA consulted with representatives of Tribal officials early in the process of developing this proposed regulation to permit them to have meaningful and timely input into its development. The EPA conducted consultation with Indian Tribes, which included a public meeting in February 28, 2012, to request input and provide rulemaking information to interested parties. A meeting summary report is available on the docket for public inspection (USEPA, 2012). The EPA notes that 751 of the 768 Tribal systems identified by the Agency as subject to the proposed rule are small systems that are expected to incur only monitoring costs. Due to the health risks associated with perchlorate, capital expenditures needed for compliance with the rule will be eligible for federal funding sources, specifically the Drinking Water State Revolving Fund. In the spirit of Executive Order 13175, and consistent with the EPA's policy to promote communications between the EPA and Tribal governments, the EPA specifically solicits additional comment on this proposed rule from Tribal officials.

# 7.9 Executive Order 13211: Actions Concerning Regulations that Significantly Affect Energy Supply, Distribution, or Use

This rule is not a "significant energy action" as defined in Executive Order 13211, entitled "Actions Concerning Regulations That Significantly Affect Energy Supply, Distribution, or Use" (66 FR 28355, May 22, 2001), because it is not likely to have a significant adverse effect on the supply, distribution, or use of energy. This determination is based on the following analysis.

The first consideration is whether the proposed rule would adversely affect the supply of energy. The proposed rule does not regulate power generation, either directly or indirectly. The public and private water systems that the proposed rule regulates do not generate power. Further, the cost increases borne by customers of water utilities as a result of the proposed rule are a low percentage of the total cost of water, except for a few water systems that might install treatment technologies and would likely spread that cost over their customer base. In sum, the proposed rule does not regulate the supply of energy, does not generally regulate the utilities that supply energy, and is unlikely to affect significantly the customer base of energy suppliers. Thus, the proposed rule would not translate into adverse effects on the supply of energy.

The second consideration is whether the proposed rule would adversely affect the distribution of energy. The proposed rule does not regulate any aspect of energy distribution. The water systems that are regulated by the proposed rule already have electrical service. At the proposed MCL, one entry point at one system may require incremental power to operate new treatment processes. The increase in peak electricity demand at water utilities is negligible. Therefore, the EPA estimates that the existing connections are adequate and that the proposed rule has no discernable adverse effect on energy distribution.

The third consideration is whether the proposed rule would adversely affect the use of energy. Because only one system is expected to add treatment technologies that use electrical power, this potential impact on sector demand or overall national demand for power is negligible.

Based on its analysis of these considerations, the EPA has concluded that proposed rule is not likely to have a significant adverse effect on the supply, distribution, or use of energy.

#### 7.10 National Technology Transfer and Advancement Act of 1995

Section 12(d) of the National Technology Transfer and Advancement Act of 1995 (NTTAA), Public Law 104–113, 12(d) (15 U.S.C. 272 note) directs the EPA to use voluntary consensus standards in its regulatory activities unless to do so would be inconsistent with applicable law or otherwise impractical. Voluntary consensus standards are technical standards (e.g., materials specifications, test methods, sampling procedures, and business practices) that are developed or adopted by voluntary consensus standards bodies. The NTTAA directs the EPA to provide Congress, through OMB, explanations when the Agency decides not to use available and applicable voluntary consensus standards. The proposed rulemaking involves technical standards. The proposed rule could involve voluntary consensus standards in that it would require monitoring for perchlorate. The EPA proposed five analytical methods for the identification and quantification of perchlorate in drinking water. EPA methods 314.0, 314.1,

314.2, 331.0, and 332.0 incorporate quality control criteria that allow accurate quantitation of perchlorate.

The EPA's monitoring and sampling protocols generally include voluntary consensus standards developed by agencies such as ASTM International, Standard Methods and other such bodies wherever the EPA deems these methodologies appropriate for compliance monitoring.

#### 7.11 Impacts on Sensitive Subpopulations and Life Stages

Section 1412(b)(3)(C)(i) of the SDWA requires that the EPA evaluate the effects of a contaminant on the general population and on potentially sensitive subpopulations such as infants, children, pregnant women, the elderly, individuals with a history of serious illness, or other sub-populations that are identified as likely to be at greater risk of adverse health effects due to exposure to contaminants in drinking water than the general population. For the proposed perchlorate rule, the EPA based the proposed MCLG and the benefits analysis in Chapter 4 on a sensitive life stage – the offspring of pregnant women exposed to perchlorate during their first trimester. The Agency determined that an MCLG protective of this sensitive life stage would also be protective of other sensitive life stages. See Chapter 4 for a discussion of the MCLG development method.

# 7.12 Consultations with the Science Advisory Board, National Drinking Water Advisory Council, and the Secretary of Health and Human Services

In accordance with sections 1412 (d) and 1412 (e) of the SDWA, the Agency consulted with the National Drinking Water Advisory Council (NDWAC or the Council); the Secretary of Health and Human Services (HHS); and with the EPA SAB. The Agency consulted with NDWAC during the Council's October 4-5, 2012 meeting. A summary of the NDWAC recommendations is available in the Fall 2012 Meeting Summary Report [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"mQCvFjdz","properties": {"formattedCitation":"(NDWAC, 2012)", "plainCitation":"(NDWAC,

2012)","noteIndex":0},"citationItems":[{"id":1268,"uris":["http://zotero.org/groups/945096/items/UIZPA9CD"],"uri":["http://zotero.org/groups/945096/items/UIZPA9CD"],"itemData":{"id":1268,"type":"article","title":"Meeting Summary, October 4-5,

2012.","URL":"https://www.epa.gov/sites/production/files/2015-

10/documents/3ndwac2012oct4-

5summary.pdf","author":[{"literal":"NDWAC"}],"issued":{"date-parts":[["2012"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ] and the docket for this proposed rule. The EPA carefully considered NDWAC recommendations during development of a proposed drinking water rule for perchlorate.

On May 29, 2012, the EPA sought guidance from the EPA SAB on how best to consider and interpret life stage information, epidemiological and biomonitoring data since the publication of the National Research Council 2005 report, the Agency's PBPK analyses, and the totality of perchlorate health information to derive an MCLG for perchlorate [ ADDIN ZOTERO\_ITEM CSL CITATION {"citationID":"b0Nazc6r","properties": {"formattedCitation":"(NRC, 2005;

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USEPA, 2012)", "plainCitation": "(NRC, 2005; USEPA,
2012)", "noteIndex":0}, "citationItems":[{"id":349, "uris":["http://zotero.org/groups/945096/items/
TN6HMC9D"],"uri":["http://zotero.org/groups/945096/items/TN6HMC9D"],"itemData": {"id":3
49, "type": "book", "title": "Health Implications of Perchlorate Ingestion", "publisher": "National
Academies Press", "publisher-place": "Washington, DC", "event-place": "Washington,
DC", "author": [{"literal": "NRC"}], "issued": {"date-
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uri":["http://zotero.org/groups/945096/items/WS4SC2GM"],"itemData":{"id":1088,"type":"articl
e","title":"Benchmark dose technical guidance","author":[{"literal":"USEPA"}],"issued":{"date-
parts":[["2012"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. On May 29, 2013, the EPA received
significant input from SAB, summarized in the report, SAB Advice on Approaches to Derive a
Maximum Contaminant Level Goal for Perchlorate[ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"pq8BBx4X","properties": {"formattedCitation":"(USEPA,
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Maximum Contaminant Level Goal for Perchlorate. EPA-SAB-13-
004", "author": [{"literal": "SAB"}, {"family": "USEPA", "given": ""}], "issued": {"date-
parts":[["2013"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] [ ADDIN ZOTERO ITEM CSL CITATION
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004", "author": [{"literal": "SAB"}, {"family": "USEPA", "given": ""}], "issued": {"date-
parts":[["2013"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"}].
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On July 15, 2013, the EPA responded by stating that the Agency would consider all the recommendations from the SAB, as it continued working on the development of the rulemaking process for perchlorate (USEPA, 2013). To address SAB recommendations, the EPA collaborated with Food and Drug Administration (FDA) scientists to develop PBPK/PD, or BBDR, models that incorporate all available health-related information on perchlorate to predict changes in thyroid hormones in a sensitive life stages exposed to different dietary iodide and perchlorate levels [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"yl7M4vOP","properties":{"formattedCitation":"(USEPA, 2017a)","plainCitation":"(USEPA, 2017a)","plainCitation":"(USEPA, 2017a)","noteIndex":0},"citationItems":[{"id":50,"uris":["http://zotero.org/groups/945096/items/NH2URBTL"],"itemData":{"id":50,"type":"report","title":"Draft Report: Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water", "author":[{"literal":"USEPA"}],"issued":{"dateparts":[["2017",9]]}}}],"schema":"https://github.com/citation-style-

```
language/schema/raw/master/csl-citation.json"}]. As recommended by SAB, the EPA developed these models based upon perchlorate's MOA (i.e., iodide uptake inhibition by the thyroid; USEPA, 2013). Additional details are in the Health Effects Technical Support Document[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"LfAWm4FT","properties": {"formattedCitation":"(USEPA, 2018b)","plainCitation":"(USEPA, 2018b)","noteIndex":0}, "citationItems":[{"id":1201,"uris":["http://zotero.org/groups/945096/items/MT6BJDE3"],"uri":["http://zotero.org/groups/945096/items/MT6BJDE3"],"itemData": {"id": 1201,"type":"report", "title":"Health Effects Technical Support Document","author":[{"literal":"USEPA"}],"issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] and described above in chapter 4.
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In accordance with SAB recommendations, the EPA developed a two-stage approach to integrate BBDR model results with data on neurodevelopmental outcomes from epidemiological studies, this approach allowed the Agency to link maternal thyroid hormones levels as a result of low iodine and perchlorate exposure, to derive an MCLG that directly addresses the most sensitive life stage (USEPA, 2013).

On [INSERT DATE], the EPA consulted with the HHS. The EPA received a favorable response to the Agency's novel approach and development of the proposed MCLG, no objection was raised as a result of the consultation.

# 7.13 Technical, Financial, and Managerial Capacity of Public Water Systems

The EPA considered whether the regulated CWS and NTNCWS would have the technical, financial, and managerial capacity to implement the proposed rule as required by Section 1420(d)(3) of the SDWA. Because the vast majority of the systems would only be required to conduct periodic monitoring for perchlorate, the Agency determined that the affected systems should have the capacity to comply with the rule. Very few systems are expected to require additional treatment to meet the proposed MCL. All of these systems are large systems, which are more likely than small systems to have the capacity to implement treatment.

#### 7.14 National Affordability Determination

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The EPA determined that there are several affordable treatment technologies for small systems. The determination, documented in *Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water* [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"JNSHWgHz","properties": {"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA, 2018a)","noteIndex":0},"citationItems":[{"id":1189,"uris":["http://zotero.org/groups/945096/items/QBLZF9AR"],"uri":["http://zotero.org/groups/945096/items/QBLZF9AR"],"itemData": {"id":1189,"type":"article", "title":"Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water.","publisher":"EPA \*\*\*-\*-\*\*\*\*\*,"author":[{"literal":"USEPA"}],"issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ], compared the estimated incremental treatment costs per household with a baseline expenditure margin that equals 2.5 percent of median household income minus baseline drinking water utility per household. [ REF \_Ref535312503 \h ] shows which technologies satisfy the affordability criterion for three small system size categories. For the smallest system size category, ion exchange and point-of-use reverse osmosis are affordable technologies, but biological treatment and centralized reverse osmosis are not.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Proposed Small System Compliance Technologies for Perchlorate Removal

System Size (population served)	Ion Exchange	Biological Treatment	Reverse Osmosis	Point-of-Use Reverse Osmosis
25–500	Yes	No	No	Yes
501–3,300	Yes	Yes	Yes	Yes
3,301–10,000	Yes	Yes	Yes	Not applicable <sup>a</sup>

a. EPA's WBS cost model for POU treatment does not cover systems larger than 3,300 people (greater than 1 MGD design flow).because implementing and maintaining a large-scale POU program in lieu of central treatment for perchlorate is likely to be impractical.

The small system compliance technology (SSCT) listed in [ REF \_Ref532979324 \h ] include a POU version of reverse osmosis. Although this technology is not a proposed BAT, it can meet the proposed MCL and, therefore, meets the effectiveness requirement for an SSCT. For perchlorate removal, the NSF Joint Committee on Drinking Water Treatment Units added a protocol to NSF/ANSI Standard 58: Reverse Osmosis Drinking Water Treatment Systems that requires a reverse osmosis unit to be able to reduce perchlorate from a challenge level of 130  $\mu$ g/L to a target level of 4  $\mu$ g/L [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"NYqGuVQI","properties":{"formattedCitation":"(NSF International, 2004)","plainCitation":"(NSF International,

2004)","noteIndex":0},"citationItems":[{"id":997,"uris":["http://zotero.org/groups/945096/items/9EAEK7S4"],"uri":["http://zotero.org/groups/945096/items/9EAEK7S4"],"itemData":{"id":997, "type":"webpage","title":"Perchlorate

 $Reduction", "URL": "http://www.nsf.org/consumer/drinking_water/perchlorate_reduction.asp", "a uthor": [{"literal": "NSF International"}], "issued": {"date-$ 

parts":[["2004"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. Organizations (e.g., NSF International, Underwriters Laboratories, Water Quality Association) provide third-party testing and certification that POU devices meet drinking water treatment standards. There are no perchlorate certification standards for other types of POU devices such as those using ion exchange media.

The operating principle for POU reverse osmosis devices is the same as centralized reverse osmosis: steric exclusion and electrostatic repulsion of ions from the charged membrane surface. In addition to a reverse osmosis membrane for dissolved ion removal, POU reverse osmosis devices often have a sediment pre-filter and a carbon filter in front of the reverse osmosis membrane, a 3- to 5-gallon treated water storage tank, and a carbon filter between the tank and the tap.

The EPA identified the SSCT using the affordability criteria it developed for drinking water rules [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"LHgBHn5b","properties":{"formattedCitation":"(USEPA, 1998)","plainCitation":"(USEPA,

1998)", "noteIndex":0}, "citationItems":[{"id":1215, "uris":["http://zotero.org/groups/945096/items/399QNBY4"], "uri":["http://zotero.org/groups/945096/items/399QNBY4"], "itemData":{"id":1215, "type":"article", "title":"Variance Technology Findings for Contaminants Regulated Before 1996", "publisher": "EPA 815–R–98–003.

September", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["1998"]]}}}], "schema": "https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. The analysis method is a comparison of estimated incremental household costs for perchlorate treatment to an expenditure margin, which is the difference between baseline household water costs and a threshold equal to 2.5 percent of median household income. [ REF \_Ref535312582 \h ] shows the expenditure margins derived for the analysis.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Expenditure Margins for SSCT Affordability Analysis

System Size (population served)	Median Household Income <sup>a</sup> (a)	Affordability Threshold <sup>b</sup> (b) = 2.5% x a	Baseline Water Cost <sup>c</sup> (c)	Expenditure Margin (d) = b - c
25-500	\$52,791	\$1,320	\$341	\$979
501-3,300	\$51,093	\$1,277	\$395	\$883
3,301-10,000	\$55,975	\$1,399	\$412	\$987

Source: Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"2scXqvv0","properties": {"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA, 2018a)","noteIndex":0},"citationItems": [{"id":1189,"uris": ["http://zotero.org/groups/945096/items/QBLZF9AR"],"uris": ["http://zotero.org/groups/945096/items/QBLZF9AR"],"itemData": {"id":1189,"type":"article","title":"Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water.", "publisher": "EPA \*\*\*-\*-\*\*\*\*, "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] a. Mean household income (MHI) is based on U.S. Census 2010 ACS 5-year estimates [ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID": "x096Tc8Y", "properties": {"formattedCitation": "(U.S. Census Bureau, 2010)", "plainCitation": "(U.S. Census Bureau,

 $2010)", "noteIndex":0\}, "citationItems":[{"id":1225,"uris":["http://zotero.org/groups/945096/items/WJ35QNBT"],"uris":["http://zotero.org/groups/945096/items/WJ35QNBT"], "itemData":{"id":1225,"type":"article", "title":"American Community Survey, 5-year Estimates (2006-2010)", "author":[{"family":"U.S. Census Bureau", "given":""}], "issued":{"date-parts":[["2010"]]}}}], "schema":"https://github.com/citation-style-$ 

Bureau", "given": ""}], "issued": {"date-parts": [["2010"]]}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] stated in 2010 dollars, adjusted to 2017 dollars using the CPI (for all items) for areas under 50,000 persons [ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID": "7Rg9m81J", "properties": {"formattedCitation": "(Bureau of Labor Statistics (BLS),

{"citationID":"7Rg9m81J","properties": {"formattedCitation":"(Bureau of Labor Statistics (BLS 2018b)","plainCitation":"(Bureau of Labor Statistics (BLS),

2018b)", "noteIndex":0}, "citationItems":[{"id":1218, "uris":["http://zotero.org/groups/945096/items/GTI7H6YK"], "uri":["http://zotero.org/groups/945096/items/GTI7H6YK"], "itemData":{"id":1218, "type": "article", "title": "CPI--All Urban Consumers (all items), for areas under 50,000 persons", "author":[{"family": "Bureau of Labor Statistics (BLS)", "given": ""}], "issued": {"date-parts":[["2018"]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

b. Affordability threshold equals 2.5 percent of MHI.

c. Household water costs derived from 2006 Community Water System Survey [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"fS2Ibu6t","properties": {"formattedCitation":"(USEPA, 2009c)","plainCitation":"(USEPA,

2009c)","noteIndex":0},"citationItems":[{"id":924,"uris":["http://zotero.org/groups/945096/items/DZNAAV6M"],"uri":["http://zotero.org/groups/945096/items/DZNAAV6M"],"itemData":{"id":924,"type":"article","title":"2006 Community Water System Survey - Volume II: Detailed Tables and Survey

Methodology","URL":"https://www.epa.gov/dwstandardsregulations/community-water-system-survey","author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2009",5]]},"accessed":{"date-parts":[["2018",8,17]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], based on residential revenue per connection within each size category, adjusted to 2017 dollars based on the CPI (for all items) for areas under 50,000 persons.

[ REF \_Ref535312595 \h ] shows the estimates of per-household costs by treatment technology and size category generated using the treatment cost method described in section XI.B as well as Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"z6GYvRh1","properties": {"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA, 2018a)","noteIndex":0}, "citationItems":[{"id":1189,"uris":["http://zotero.org/groups/945096/ite

ms/QBLZF9AR"],"uri":["http://zotero.org/groups/945096/items/QBLZF9AR"],"itemData":{"id" :1189, "type": "article", "title": "Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water.", "publisher": "EPA \*\*\*-\*-\*\*\*\*","author":[{"literal":"USEPA"}],"issued":{"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] and Technologies and Costs for Treating Perchlorate-Contaminated Waters [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"18aKvRLD","properties":{"formattedCitation":"(USEPA, 2018g)", "plainCitation": "(USEPA, 2018g)", "noteIndex":0}, "citationItems":[{"id":977, "uris":["http://zotero.org/groups/945096/item s/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData":{"id":9 77,"type":"report","title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters", "publisher": "EPA \*\*\*-\*-\*-\*\*\*", "author": [{"literal": "USEPA"}], "issued": {"dateparts":[["2018"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Costs in bold font do not exceed the corresponding expenditure margin and, therefore, meet the SSCT affordability criterion.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]: Annual Incremental Cost Estimates for SSCT Affordability Analysis

System Size	lon	Biological	Reverse	Point-of-Use
(population served)	Exchange	Treatment	Osmosis	Reverse Osmosis
25–500	\$378 to \$610	\$2,146 to \$3,709	\$2,272 to \$2,671	\$265 to \$271
501–3,300	\$98 to \$148	\$324 to \$566	\$561 to \$688	\$250 to \$251
3,301–10,000	\$104 to \$153	\$211 to \$315	\$431 to \$493	Not applicable <sup>a</sup>

Source: Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"8y1WSJT4","properties":{"formattedCitation":"(USEPA, 2018a)","plainCitation":"(USEPA, 2018a)","noteIndex":0},"citationItems":[{"id":1189,"uris":["http://zotero.org/groups/945096/items/QBLZF9AR"],"uris":["http://zotero.org/groups/945096/items/QBLZF9AR"],"itemData":{"id":1189,"type":"article","title":"Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water.","publisher":"EPA \*\*\*-\*-\*-\*\*\*\*,"author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]; bold font indicates cost estimates that do not exceed the corresponding expenditure margin.

a. EPA's WBS model for POU treatment does not cover systems larger than 3,300 people (greater than 1 MGD design flow), because implementing and maintaining a large-scale POU program is likely to be impractical.

8 References
[ ADDIN ZOTERO_BIBL {"uncited":[],"omitted":[],"custom":[]} CSL_BIBLIOGRAPHY ]
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Appendix A. Output from the BBDR Model Summarizing Maternal fT4 Levels Given Increasing Dose of Perchlorate

Perchlorate	lodine Intake Levels; fT4 (pmol/L)														
Dose (μg/kg/day)	50	55	60	65	70	75	80	85	90	95	100	125	150	170	300
0	8.27	8.47	8.67	8.87	9.06	9.26	9.47	9.67	9.87	10.05	10.18	10.40	10.47	10.50	10.57
0.1	8.27	8.47	8.66	8.86	9.06	9.26	9.46	9.67	9.87	10.04	10.17	10.40	10.47	10.50	10.57
0.2	8.26	8.46	8.66	8.86	9.05	9.25	9.46	9.66	9.86	10.04	10.17	10.40	10.47	10.50	10.57
0.3	8.26	8.46	8.66	8.85	9.05	9.25	9.45	9.65	9.85	10.03	10.17	10.40	10.46	10.50	10.57
0.4	8.26	8.46	8.65	8.85	9.04	9.24	9.45	9.65	9.85	10.03	10.16	10.40	10.46	10.50	10.57
0.5	8.25	8.45	8.65	8.84	9.04	9.24	9.44	9.64	9.84	10.02	10.16	10.40	10.46	10.50	10.57
0.6	8.25	8.45	8.65	8.84	9.04	9.23	9.43	9.64	9.84	10.02	10.15	10.40	10.46	10.50	10.57
0.7	8.25	8.45	8.64	8.84	9.03	9.23	9.43	9.63	9.83	10.01	10.15	10.39	10.46	10.50	10.57
0.8	8.24	8.44	8.64	8.83	9.03	9.22	9.42	9.63	9.82	10.00	10.15	10.39	10.46	10.50	10.57
0.9	8.24	8.44	8.64	8.83	9.02	9.22	9.42	9.62	9.82	10.00	10.14	10.39	10.46	10.50	10.57
1	8.24	8.44	8.63	8.82	9.02	9.21	9.41	9.61	9.81	9.99	10.14	10.39	10.46	10.50	10.57
1.1	8.24	8.43	8.63	8.82	9.01	9.21	9.41	9.61	9.81	9.99	10.13	10.39	10.46	10.49	10.57
1.2	8.23	8.43	8.62	8.82	9.01	9.20	9.40	9.60	9.80	9.98	10.13	10.39	10.46	10.49	10.57
1.3	8.23	8.43	8.62	8.81	9.00	9.20	9.40	9.60	9.79	9.98	10.13	10.39	10.46	10.49	10.57
1.4	8.23	8.42	8.62	8.81	9.00	9.19	9.39	9.59	9.79	9.97	10.12	10.39	10.46	10.49	10.57
1.5	8.22	8.42	8.61	8.80	9.00	9.19	9.39	9.58	9.78	9.97	10.12	10.39	10.46	10.49	10.57
1.6	8.22	8.42	8.61	8.80	8.99	9.18	9.38	9.58	9.78	9.96	10.11	10.39	10.46	10.49	10.57
1.7	8.22	8.42	8.61	8.80	8.99	9.18	9.38	9.57	9.77	9.96	10.11	10.39	10.46	10.49	10.57
1.8	8.22	8.41	8.60	8.79	8.98	9.17	9.37	9.57	9.76	9.95	10.10	10.39	10.46	10.49	10.57
1.9	8.21	8.41	8.60	8.79	8.98	9.17	9.36	9.56	9.76	9.94	10.10	10.38	10.46	10.49	10.57
2	8.21	8.41	8.60	8.79	8.97	9.17	9.36	9.56	9.75	9.94	10.09	10.38	10.46	10.49	10.57
2.1	8.21	8.40	8.59	8.78	8.97	9.16	9.35	9.55	9.75	9.93	10.09	10.38	10.46	10.49	10.57
2.2	8.20	8.40	8.59	8.78	8.97	9.16	9.35	9.55	9.74	9.93	10.08	10.38	10.46	10.49	10.57
2.3	8.20	8.40	8.59	8.77	8.96	9.15	9.34	9.54	9.74	9.92	10.08	10.38	10.46	10.49	10.57
2.4	8.20	8.39	8.58	8.77	8.96	9.15	9.34	9.53	9.73	9.92	10.08	10.38	10.45	10.49	10.57
2.5	8.20	8.39	8.58	8.77	8.95	9.14	9.33	9.53	9.72	9.91	10.07	10.38	10.45	10.49	10.57

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Perchlorate						lodir	ne Intake	Levels;	fT4 (pm	ol/L)					
Dose (μg/kg/day)	50	55	60	65	70	75	80	85	90	95	100	125	150	170	300
2.6	8.19	8.39	8.58	8.76	8.95	9.14	9.33	9.52	9.72	9.90	10.07	10.38	10.45	10.49	10.57
2.7	8.19	8.38	8.57	8.76	8.95	9.13	9.32	9.52	9.71	9.90	10.06	10.38	10.45	10.49	10.57
2.8	8.19	8.38	8.57	8.76	8.94	9.13	9.32	9.51	9.71	9.89	10.06	10.38	10.45	10.49	10.57
2.9	8.19	8.38	8.57	8.75	8.94	9.12	9.31	9.51	9.70	9.89	10.05	10.38	10.45	10.49	10.57
3	8.18	8.38	8.56	8.75	8.93	9.12	9.31	9.50	9.70	9.88	10.05	10.37	10.45	10.49	10.57
3.1	8.18	8.37	8.56	8.74	8.93	9.12	9.30	9.50	9.69	9.88	10.04	10.37	10.45	10.49	10.57
3.2	8.18	8.37	8.56	8.74	8.92	9.11	9.30	9.49	9.68	9.87	10.04	10.37	10.45	10.49	10.57
3.3	8.17	8.37	8.55	8.74	8.92	9.11	9.29	9.49	9.68	9.87	10.03	10.37	10.45	10.49	10.57
3.4	8.17	8.36	8.55	8.73	8.92	9.10	9.29	9.48	9.67	9.86	10.03	10.37	10.45	10.49	10.57
3.5	8.17	8.36	8.55	8.73	8.91	9.10	9.29	9.48	9.67	9.85	10.02	10.37	10.45	10.49	10.57
3.6	8.17	8.36	8.54	8.73	8.91	9.09	9.28	9.47	9.66	9.85	10.02	10.37	10.45	10.49	10.57
3.7	8.16	8.35	8.54	8.72	8.90	9.09	9.28	9.47	9.66	9.84	10.01	10.37	10.45	10.49	10.57
3.8	8.16	8.35	8.54	8.72	8.90	9.08	9.27	9.46	9.65	9.84	10.01	10.37	10.45	10.48	10.57
3.9	8.16	8.35	8.53	8.72	8.90	9.08	9.27	9.45	9.64	9.83	10.00	10.37	10.45	10.48	10.57
4	8.16	8.35	8.53	8.71	8.89	9.08	9.26	9.45	9.64	9.83	10.00	10.36	10.45	10.48	10.57
4.1	8.15	8.34	8.53	8.71	8.89	9.07	9.26	9.44	9.63	9.82	9.99	10.36	10.45	10.48	10.57
4.2	8.15	8.34	8.52	8.70	8.89	9.07	9.25	9.44	9.63	9.81	9.99	10.36	10.45	10.48	10.57
4.3	8.15	8.34	8.52	8.70	8.88	9.06	9.25	9.43	9.62	9.81	9.98	10.36	10.45	10.48	10.57
4.4	8.14	8.33	8.52	8.70	8.88	9.06	9.24	9.43	9.62	9.80	9.98	10.36	10.45	10.48	10.57
4.5	8.14	8.33	8.51	8.69	8.87	9.05	9.24	9.42	9.61	9.80	9.97	10.36	10.44	10.48	10.57
4.6	8.14	8.33	8.51	8.69	8.87	9.05	9.23	9.42	9.61	9.79	9.97	10.36	10.44	10.48	10.57
4.7	8.14	8.33	8.51	8.69	8.87	9.05	9.23	9.41	9.60	9.79	9.96	10.36	10.44	10.48	10.57
4.8	8.13	8.32	8.51	8.68	8.86	9.04	9.22	9.41	9.60	9.78	9.95	10.36	10.44	10.48	10.57
4.9	8.13	8.32	8.50	8.68	8.86	9.04	9.22	9.40	9.59	9.78	9.95	10.35	10.44	10.48	10.57
5	8.13	8.32	8.50	8.68	8.86	9.03	9.22	9.40	9.59	9.77	9.94	10.35	10.44	10.48	10.57
5.1	8.13	8.31	8.50	8.67	8.85	9.03	9.21	9.39	9.58	9.76	9.94	10.35	10.44	10.48	10.57
5.2	8.12	8.31	8.49	8.67	8.85	9.03	9.21	9.39	9.57	9.76	9.93	10.35	10.44	10.48	10.57
5.3	8.12	8.31	8.49	8.67	8.84	9.02	9.20	9.38	9.57	9.75	9.93	10.35	10.44	10.48	10.57
5.4	8.12	8.31	8.49	8.66	8.84	9.02	9.20	9.38	9.56	9.75	9.92	10.35	10.44	10.48	10.57

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Perchlorate						lodin	ie Intake	Levels;	fT4 (pm	ol/L)					
Dose (μg/kg/day)	50	55	60	65	70	75	80	85	90	95	100	125	150	170	300
5.5	8.12	8.30	8.48	8.66	8.84	9.01	9.19	9.37	9.56	9.74	9.92	10.35	10.44	10.48	10.57
5.6	8.11	8.30	8.48	8.66	8.83	9.01	9.19	9.37	9.55	9.74	9.91	10.35	10.44	10.48	10.57
5.7	8.11	8.30	8.48	8.65	8.83	9.01	9.18	9.36	9.55	9.73	9.91	10.35	10.44	10.48	10.57
5.8	8.11	8.30	8.47	8.65	8.83	9.00	9.18	9.36	9.54	9.73	9.90	10.34	10.44	10.48	10.57
5.9	8.11	8.29	8.47	8.65	8.82	9.00	9.18	9.36	9.54	9.72	9.90	10.34	10.44	10.48	10.57
6	8.10	8.29	8.47	8.64	8.82	8.99	9.17	9.35	9.53	9.72	9.89	10.34	10.44	10.48	10.57
6.1	8.10	8.29	8.47	8.64	8.81	8.99	9.17	9.35	9.53	9.71	9.89	10.34	10.44	10.48	10.57
6.2	8.10	8.28	8.46	8.64	8.81	8.99	9.16	9.34	9.52	9.70	9.88	10.34	10.44	10.48	10.57
6.3	8.10	8.28	8.46	8.63	8.81	8.98	9.16	9.34	9.52	9.70	9.88	10.34	10.44	10.48	10.56
6.4	8.09	8.28	8.46	8.63	8.80	8.98	9.15	9.33	9.51	9.69	9.87	10.34	10.43	10.47	10.56
6.5	8.09	8.28	8.45	8.63	8.80	8.97	9.15	9.33	9.51	9.69	9.86	10.34	10.43	10.47	10.56
6.6	8.09	8.27	8.45	8.62	8.80	8.97	9.15	9.32	9.50	9.68	9.86	10.33	10.43	10.47	10.56
6.7	8.09	8.27	8.45	8.62	8.79	8.97	9.14	9.32	9.50	9.68	9.85	10.33	10.43	10.47	10.56
6.8	8.08	8.27	8.45	8.62	8.79	8.96	9.14	9.31	9.49	9.67	9.85	10.33	10.43	10.47	10.56
6.9	8.08	8.27	8.44	8.62	8.79	8.96	9.13	9.31	9.49	9.67	9.84	10.33	10.43	10.47	10.56
7	8.08	8.26	8.44	8.61	8.78	8.95	9.13	9.30	9.48	9.66	9.84	10.33	10.43	10.47	10.56
7.1	8.08	8.26	8.44	8.61	8.78	8.95	9.12	9.30	9.48	9.66	9.83	10.33	10.43	10.47	10.56
7.2	8.07	8.26	8.43	8.61	8.78	8.95	9.12	9.30	9.47	9.65	9.83	10.33	10.43	10.47	10.56
7.3	8.07	8.26	8.43	8.60	8.77	8.94	9.12	9.29	9.47	9.65	9.82	10.32	10.43	10.47	10.56
7.4	8.07	8.25	8.43	8.60	8.77	8.94	9.11	9.29	9.46	9.64	9.82	10.32	10.43	10.47	10.56
7.5	8.07	8.25	8.43	8.60	8.77	8.94	9.11	9.28	9.46	9.64	9.81	10.32	10.43	10.47	10.56
7.6	8.07	8.25	8.42	8.59	8.76	8.93	9.10	9.28	9.45	9.63	9.81	10.32	10.43	10.47	10.56
7.7	8.06	8.24	8.42	8.59	8.76	8.93	9.10	9.27	9.45	9.63	9.80	10.32	10.43	10.47	10.56
7.8	8.06	8.24	8.42	8.59	8.76	8.93	9.10	9.27	9.44	9.62	9.80	10.32	10.43	10.47	10.56
7.9	8.06	8.24	8.41	8.58	8.75	8.92	9.09	9.26	9.44	9.62	9.79	10.32	10.43	10.47	10.56
8	8.06	8.24	8.41	8.58	8.75	8.92	9.09	9.26	9.43	9.61	9.79	10.31	10.43	10.47	10.56
8.1	8.05	8.23	8.41	8.58	8.75	8.91	9.08	9.26	9.43	9.61	9.78	10.31	10.42	10.47	10.56
8.2	8.05	8.23	8.41	8.58	8.74	8.91	9.08	9.25	9.42	9.60	9.78	10.31	10.42	10.47	10.56
8.3	8.05	8.23	8.40	8.57	8.74	8.91	9.08	9.25	9.42	9.60	9.77	10.31	10.42	10.47	10.56

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Perchlorate						lodir	ie Intake	Levels;	fT4 (pm	ol/L)					
Dose (μg/kg/day)	50	55	60	65	70	75	80	85	90	95	100	125	150	170	300
8.4	8.05	8.23	8.40	8.57	8.74	8.90	9.07	9.24	9.42	9.59	9.76	10.31	10.42	10.47	10.56
8.5	8.04	8.22	8.40	8.57	8.73	8.90	9.07	9.24	9.41	9.59	9.76	10.31	10.42	10.47	10.56
8.6	8.04	8.22	8.39	8.56	8.73	8.90	9.06	9.23	9.41	9.58	9.75	10.30	10.42	10.47	10.56
8.7	8.04	8.22	8.39	8.56	8.73	8.89	9.06	9.23	9.40	9.58	9.75	10.30	10.42	10.47	10.56
8.8	8.04	8.22	8.39	8.56	8.72	8.89	9.06	9.23	9.40	9.57	9.74	10.30	10.42	10.46	10.56
8.9	8.04	8.21	8.39	8.55	8.72	8.89	9.05	9.22	9.39	9.57	9.74	10.30	10.42	10.46	10.56
9	8.03	8.21	8.38	8.55	8.72	8.88	9.05	9.22	9.39	9.56	9.73	10.30	10.42	10.46	10.56
9.1	8.03	8.21	8.38	8.55	8.71	8.88	9.04	9.21	9.38	9.56	9.73	10.30	10.42	10.46	10.56
9.2	8.03	8.21	8.38	8.55	8.71	8.87	9.04	9.21	9.38	9.55	9.72	10.29	10.42	10.46	10.56
9.3	8.03	8.20	8.38	8.54	8.71	8.87	9.04	9.20	9.37	9.55	9.72	10.29	10.42	10.46	10.56
9.4	8.02	8.20	8.37	8.54	8.70	8.87	9.03	9.20	9.37	9.54	9.71	10.29	10.42	10.46	10.56
9.5	8.02	8.20	8.37	8.54	8.70	8.86	9.03	9.20	9.37	9.54	9.71	10.29	10.42	10.46	10.56
9.6	8.02	8.20	8.37	8.53	8.70	8.86	9.03	9.19	9.36	9.53	9.70	10.29	10.42	10.46	10.56
9.7	8.02	8.19	8.37	8.53	8.69	8.86	9.02	9.19	9.36	9.53	9.70	10.29	10.42	10.46	10.56
9.8	8.01	8.19	8.36	8.53	8.69	8.85	9.02	9.18	9.35	9.52	9.69	10.28	10.41	10.46	10.56
9.9	8.01	8.19	8.36	8.53	8.69	8.85	9.01	9.18	9.35	9.52	9.69	10.28	10.41	10.46	10.56
10	8.01	8.19	8.36	8.52	8.69	8.85	9.01	9.18	9.34	9.51	9.68	10.28	10.41	10.46	10.56

## Appendix B. Estimated Value of an IQ Point

Valuing changes in the intelligence quotient (IQ) is an important component to monetizing the effects of regulations expected to have significant impacts on IQ. The dollar value of a single IQ point is often equated to the change in lifetime earnings associated with an additional IQ point. This appendix describes the method used to estimate the average lifetime earnings of an individual born into a particular birth cohort and the associated effect of a one point change in IQ.

[ REF \_Ref532978150 \h ] summarizes this appendix's estimates of the IQ point dollar values produced in 2017 dollars. The values can be used to estimate the benefits of avoiding IQ decrements.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Summary of IQ Point Dollar Values (2017\$)

· /	Discount	Rate
Estimate Parameter	3%	7%
IQ value	\$23,269	\$5,398
Additional education costs and lost earnings	\$1,592	\$691
IQ value without additional education costs and lost earnings	\$21,677	\$4,707

[ REF \_Ref532978235 \h ] presents a summary of the changes incorporated into the method used in this analysis compared to methods used in previous EPA analyses. In addition to relying on more recent data, this analysis incorporates several methodological changes designed to improve the estimates for the value of an IQ point.

# Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Difference between the Current Methods and Previous Methods Used to Estimate Lifetime Earnings

,	To vious inclined Social to Estimate Elictime Eurimigs
Data and methods	Uses the average of ten ACS single-year Public Use Microdata Sample (PUMS; 2008 to 2017) in place of a single cross-section of the Current Population Survey (CPS). The ACS has a larger sample size than the CPS and data are independent from one year to the next. <sup>a</sup> A ten-year average has a smoothing effect on estimates, which diminishes economy-wide shocks to the productivity growth rate caused by cyclical and anomalous expansions and contractions of the economy. Incorporating data from the most recent decade produces estimates consistent with recent social, demographic, and economic trends affecting earnings, workforce participation, and educational attainment.
Life tables	Incorporates projected cohort life tables published by the SSA in the <i>Life tables underlying the SSA Trustees' 2015 annual report</i> ([ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Social Security Administration (SSA)/Author&gt;<year>2015/Year&gt;<recnum>219/RecNum&gt;<prefix>SSA /Prefix&gt;<displaytext>SSA 2015/DisplayText&gt;<record><rec-number>219</rec-number><foreign-keys><key app="EN" db-id="0a0wva05uf5wdvep9eex2dz052etaa2zdsxs" timestamp="1484253829">219/key&gt;</key></foreign-keys><ref-type name="Dataset">59</ref-type><contributors><author>&gt;cuthor&gt;&gt;cu</author></contributors></record></displaytext></prefix></recnum></year></author></cite></endnote>
IQ-effect	Uses an updated Salkever IQ effect that was estimated using more recent 1997 National Longitudinal Survey of Youth (NLSY) data; these updated IQ effects are 1.865 percent for males and 3.397 percent for females. The justification for this is presented in Section B.1.
Educational attainment	Models postsecondary educational attainment for the target cohort using enrollment statistics and degrees conferred. Previous estimates account for declining educational attainment observed in older cohorts by assuming that educational attainment cannot decline.
Lost earnings and education costs	Estimates lost earnings while in school by taking the difference of average earnings for individuals enrolled in school and individuals not enrolled in school. Lost earnings are weighted by simulated levels of educational attainment. Previously, this has been estimated by assuming that individuals enrolled in school work part-time and individuals not enrolled in school work full-time.  Weights education costs by school enrollment at each age rather than taking a straight average
	Weights education costs and lost earnings by the probability that an individual will attain additional education at a particular age rather than assuming that they accrue at the marginal education age  Discounts lifetime earnings to a baseline age of 3 rather than 0

a. The CPS Annual Social and Economic Supplement (ASEC) has a longitudinal component resulting in some overlap of individuals sampled from one year to the next.

#### **B.1** IQ Effect on Lifetime Earnings

For this analysis, the value of an IQ point as a percentage of lifetime earnings is taken from an EPA reanalysis of Salkever ([ADDIN EN.CITE ADDIN EN.CITE.DATA]), which follows the same approach as the original analysis but used more recent 1997 NLSY data instead of the original 1979 NLSY data. This EPA analysis indicates that a one point change in IQ results in a 1.865 percent and 3.397 percent change in lifetime earnings for males and females, respectively. The estimates incorporate the direct and indirect effects that changes in IQ have on earnings. Direct effects include the effect of IQ test scores on participation in employment and earnings with the years of schooling held constant. Indirect effects include the effect of IQ test scores on years of schooling attained, and the subsequent effect on participation and earnings.

Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]) also produces adjusted estimates that incorporate non-IQ related effects caused by lead exposure on schooling from Schwartz ([ ADDIN EN.CITE <EndNote><Cite

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provider></record></Cite></EndNote>]), yielding a 2.094 percent and 3.631 percent change in lifetime earnings for males and females, respectively. Non-IQ effects on schooling caused by lead exposure include effects on hearing, balance, hyperactivity, and perceptual and attention disorders ([ ADDIN EN.CITE

<EndNote><Cite><Author>Schwartz</Author><Year>1994</Year><RecNum>621</RecNum><DisplayText>Schwartz 1994</DisplayText><rec-number>621</rec-

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database-name><remote-database-provider>EBSCOhost</remote-database-
provider></record></Cite></EndNote>]) estimates lead exposure on schooling to be 0.131 years
per IQ point, 12 whereas Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]) estimates
the effect to be 0.1007 years per IQ point- the former estimate unsurprisingly larger given that it
captures IQ and non-IQ effects of lead exposure.
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In his meta-analysis, Salkever ([ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Salkever</Author><Year>2014</Year><RecNum>618</RecNum> <DisplayText>2014
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num>http://dx.doi.org/10.1016/j.envres.2014.03.018</electronic-resource-

num></record></Cite></EndNote>]) only defended his unadjusted results, implying that Salkever believes his unadjusted results to be the most defensible. For this reason, Salkever's ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]) adjusted estimates incorporating Schwartz's ([ ADDIN EN.CITE < EndNote > < Cite

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Because non-IQ effects on schooling are omitted, results presented here may be applied more generally to evaluate IQ decrements resulting from other toxins as well (e.g., mercury). However, if non-IQ effects on schooling are expected to be significant, the estimated benefits of preventing IQ decrements could be overly conservative.

#### **B.2** Critical Review of Salkever's Estimates

Salkever's estimates of the impacts of IQ decrements on future earnings are greater than those of other authors, which has led to claims that they overstate the true impacts. Most notably, Grosse ([ ADDIN EN.CITE <EndNote><Cite

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20</pages><volume>27</volume><dates><year>2007</year></dates><urls></record></Cite></EndNote>]) and Robinson ([ ADDIN EN.CITE <EndNote><Cite

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- 1. Measurement and statistical issues, and
- 2. Comparability of earnings impact results with recent findings in the labor economics literature.

Salkever ([ ADDIN EN.CITE < EndNote > < Cite

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date < /record < /Cite > < /EndNote > ]) by providing a meta-analysis of related studies. Salkever ([
ADDIN EN.CITE <EndNote><Cite
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num>http://dx.doi.org/10.1016/j.envres.2014.03.018</electronic-resource-
num></record></Cite></EndNote>]) provides convincing evidence in support of his original
estimates. For example, the author found that many of the studies only included males and/or
considered hourly wages instead of annual earnings with a participation effect. Because impacts
are lower for males ([ ADDIN EN.CITE
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consider the impact of IQ decrements on male wages will be biased downward. In reference to
the latter observation, Salkever ([ ADDIN EN.CITE < EndNote > < Cite
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urls></urls><electronic-resource-
num>http://dx.doi.org/10.1016/j.envres.2014.03.018</electronic-resource-
num></record></Cite></EndNote>]) concludes that studies examining impacts on hourly wages
omit the impact that IQ decrements have on workforce participation, again biasing the results
downward. In fact, Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]) is the only
analysis that separately estimates the IQ effect derived from average annual earnings and the
participation effect for both males and females. For a full discussion of Salkever's response to
critical review articles see Salkever ([ ADDIN EN.CITE < EndNote > < Cite
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num></record></Cite></EndNote>]).
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As of the date of this report, the estimates published in Salkever ([ADDIN EN.CITE ADDIN EN.CITE.DATA ]) remain the most appropriate for this analysis. Though the procedures and assumptions on which many of the related studies depend may be sound in themselves, they do not improve upon Salkever ([ADDIN EN.CITE ADDIN EN.CITE.DATA ]), at least insomuch as the present analysis is concerned.

#### **B.3** Method for Estimating Lifetime Earnings Stream

This section describes the methods used to estimate the average present value of the earnings stream of the affected birth cohort from age 16 to age 80. The universe of individuals from which

annual earnings and enrollment are estimated includes the civilian population only, excluding individuals living in institutional and non-institutional group quarters and unpaid workers. <sup>13</sup> Estimates are presented in 2017 dollars and are discounted using both a rate of 3 percent and of 7 percent.

Section [ REF \_Ref402355789 \r \h \\* MERGEFORMAT ] describes the method used to estimate the distribution of educational attainment associated with each age in the age profile. Section [ REF \_Ref402355963 \r \h \\* MERGEFORMAT ] describes the method used to estimate the annual earnings associated with each level of educational attainment for each age in the age profile. Section [ REF \_Ref402356068 \r \h \\* MERGEFORMAT ] describes the life tables used in the lifetime earnings estimation and the derivation of male-female population ratios from the life tables. Section [ REF \_Ref402356221 \r \h \\* MERGEFORMAT ] summarizes the lifetime earnings estimates. Section [ REF \_Ref402358029 \n \h \\* MERGEFORMAT ] describes the effect changes in IQ have on lifetime earnings. Section [ REF \_Ref480970863 \n \h ] presents a discussion of the limitations of this method.

#### **B.3.1** Educational Attainment

Two methods are used to estimate the distribution of educational attainment over the age profile of the target birth cohort. One method is used to estimate primary and secondary school attainment, with the exception of grade 12, and a second method is used to estimate postsecondary school attainment. Section [ REF \_Ref403393937 \n \h ] describes primary and secondary school attainment, and Section [ REF \_Ref402181341 \n \h ] describes postsecondary school attainment.

#### **B.3.1.1** Primary and Secondary School Attainment

Primary and secondary school attainment is estimated for ages 1 to 19 by estimating the average proportion of the population that has completed grades 1 to 11 for each age. Primary and secondary school attainment is estimated for ages 20 and older by estimating the average proportion of 20 to 25 year olds that have completed grades 1 to 11. ACS single-year PUMS from 2008 to 2017 are used to produce ten-year average primary and secondary school attainment estimates ([ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Census Bureau</Author><Year>2017</Year><RecNum>236</RecNum>CisplayText>U.S. Census Bureau 2017c</DisplayText><record><rec-number>236</rec-number><foreign-keys><key app="EN" db-id="9e2tazdvmef5axewx2ovx2xd92psp9ft9ets"

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<sup>&</sup>lt;sup>13</sup> According to the Census Bureau, people not living housing units (e.g., house, apartment, mobile home, rented room) are classified as living in group quarters. As of 2006, the ACS sample includes both institutional group quarters (e.g., correctional facilities, nursing homes, and mental hospitals) and non-institutional group quarters (e.g., college dormitories, military barracks, group homes, missions, and shelters; [ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Census

timestamp="1476715332">236</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>U.S. Census Bureau,,<author></author></authors></contributors><titles><title>American Community Survey Single-year Public Use Microdata Sample</title></title></title></title></dates><urls></article></contributors></article></dates></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article></article><

#### **B.3.1.2** Postsecondary School Attainment

Postsecondary educational attainment is estimated first by determining the age-sex distribution of degrees conferred and then by applying this distribution to the total number of degrees conferred. The age-sex distribution of enrollment in college and graduate school is used as a proxy for the age-sex distribution of degrees conferred. Lag times are used to stagger enrollment because of the length of time it takes to earn a degree. A lag time of one year is assigned to Associate's and Master's degrees, two years to Bachelor's degrees, and three years to doctoral degrees. For example, if 20 percent of males enrolled as undergraduates in college are 20 years old then it is assumed that 20 percent of Bachelor's degrees are awarded to 22 year old males.

The proportion of an age group receiving a degree is then calculated by dividing the number of degrees awarded at a particular age by the total number of people of the same age. To account for individuals receiving successive degrees from postsecondary institutions the proportion of individuals with a Bachelor's degree that previously received an Associate's degree (21 percent) is subtracted from the proportion of individuals with an Associate's degree <sup>14</sup> ([ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Department of

Education
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11/3/2014)</title></title></dates><urls></urls></record></Cite></EndNote>]); the proportion of individuals receiving a Master's degree or receiving a doctoral degree without previously receiving a Master's degree (25 percent; [ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Department of

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2003-04 Beginning Postsecondary
Students Longitudinal Study, Second Follow-up (Computation by NCES PowerStats Version 1.0 on
11/3/2014)
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).

<sup>&</sup>lt;sup>14</sup> Estimate based on the 2003-04 Beginning Postsecondary Students Longitudinal Study ([ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Department of

number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp" timestamp="1529590240">629</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author> U.S. Department of Education, National Center for Education Statistics (NCES)</author></contributors></title>Baccalaureate and Beyond Longitudinal Study 1993-2003 (Computation by NCES PowerStats Version 1.0 on 11/6/2014)</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></ti> EndNote>]) is subtracted from the proportion of individuals receiving a Bachelor's degree; and the proportion of individuals receiving a doctoral degree that previously received a Master's degree is subtracted from the proportion of individuals with a Master's degree (75 percent; [ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Department of Education</Author><Year>2003</Year><RecNum>629</RecNum><DisplayText>U.S. Department of Education 2003</DisplayText><record><rec-number>629</recnumber><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp" timestamp="1529590240">629</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author>U.S. Department of Education,, National Center for Education Statistics (NCES)</author></contributors></title>Baccalaureate and Beyond Longitudinal Study 1993-2003 (Computation by NCES PowerStats Version 1.0 on 11/6/2014)</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></ti> EndNote>]). 15, 16 The cumulative proportion of degrees conferred at a particular age represents the proportion of the population for whom the corresponding degree is the highest level of education attained.

Enrollment statistics are estimated from 2008 to 2017 using ACS single-year PUMS. Data describing Associate's, Bachelor's, Master's, and doctoral degrees conferred are taken from the 2016 Digest of Education Statistics, published by the National Center for Education Statistics (NCES) ([ ADDIN EN.CITE

<EndNote><Cite><Author>Snyder
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<sup>-</sup>

<sup>15</sup> Estimate based on the 1993 Baccalaureate and Beyond Longitudinal Study ([ADDIN EN.CITE <EndNote><Cite><Author>U.S. Department of Education</br>
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11/6/2014)
/title></title></fixed>
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/EndNote>]). The actual figure is 74.3.

<sup>&</sup>lt;sup>16</sup> Educational attainment does not account for individuals receiving a second degree within the same attainment group later in life. For example, a second Master's degree is treated as two individuals with a Master's degree.

U.S. Department of Education</publisher><urls><related-urls><url>https://nces.ed.gov/programs/digest/2016menu\_tables.asp</url></related-urls></urls><access-date>Oct. 06, 2016</access-date></record></Cite></EndNote>]). Data describing degrees conferred in the years 2016 and 2017 are projected by the NCES.

#### **B.3.2** Average Annual Earnings

Annual earnings for males and females are estimated from ACS single-year PUMS for each of the levels of educational attainment previously described for ages ranging from 16 to 80, for each year from 2008 to 2017. Average annual earnings for males and females are equal to the sum of annual earnings associated with each level of educational attainment weighted by the distribution of educational attainment for each year in the age profile of the target birth cohort. Earnings consist of pre-tax wages and salary.

#### B.3.2.1 Topcodes

For confidentiality reasons, the Census Bureau topcodes reported income components if it falls below or above a predetermined threshold before making the data publicly available. Amounts above or below the threshold are replaced with the state mean value above or below the threshold ([ADDIN EN.CITE <EndNote><Cite><Author>U.S. Census

Bureau</Author><Year>2013</Year><RecNum>626</RecNum><DisplayText>U.S. Census

Bureau 2013</DisplayText><record><rec-number>626</rec-number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp" timestamp="1529589874">626</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>U.S. Census

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#### **B.3.2.2** Real Earnings Growth Rate

Real earnings fluctuate from one year to the next in response to trends in the inflation rate and the economy-wide productivity growth rate ([ ADDIN EN.CITE 
<EndNote><Cite><Author>Thornton</Author><Year>1997</Year><RecNum>622</RecNum> 
OisplayText>Thornton et al. 1997</DisplayText><record><rec-number>622</rec-number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp" timestamp="1529358532">622</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Keys><ref-type name="Journal Article">1.</author><author>Rogers, James D.</author><author>Brookshire, Michael L.</author></authors></contributors><title>>on the Interpretations of Age-Earnings Profiles</title><secondary-title>Journal of Labor Research</secondary-title>
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into the annual earnings estimates. Grosse et al. ([ ADDIN EN.CITE < EndNote > < Cite
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control</keyword><keyword>Child</keyword>Cognition Disorders/chemically
induced</keyword><keyword>Cognition Disorders/prevention & Disorders
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#### **B.3.3** Life Tables and Male-female Ratios

#### B.3.3.1 Life Tables

Life tables present the survival rate or the probability that an individual will survive from age N to age N+1 (e.g. a 60 year old living to 61). Because some individuals die before retirement, the survival rate represents the probability of receiving earnings each year. As an individual ages, the probability declines. Future birth cohorts are expected to live longer, increasing the probability of living to age N+1, thereby increasing the average lifetime earnings for the cohort.

This analysis relies on the *Life tables underlying the SSA Trustees' 2015 annual report*, which are published on the website of Value Economics LLC ([ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Social Security Administration (SSA)</Author><Year>2015</Year><RecNum>243</RecNum><Prefix>SSA </Prefix><DisplayText>SSA 2015</DisplayText><record><rec-number>243</rec-number><foreign-keys><key app="EN" db-id="9e2tazdvmef5axewx2ovx2xd92psp9ft9ets" timestamp="1476715336">243</key></foreign-keys><ref-type name="Dataset">59</ref-type><contributors><author>Social Security Administration (SSA),,</author></author></author></author></authors><tittle><tittle>Life tables underlying the SSA Trustees&apos; 2015 annual report, in Excel format.</a></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></tit

LLC</publisher><urls><related-urls><url>http://www.valueeconomics.com/data.html</url></related-urls></urls></record></Cite></EndNote>]). The SSA develops projected life tables for future birth cohorts. In this analysis, the life tables associated with the 2015 birth cohort are incorporated to represent a recent cohort.

#### **B.3.3.2** Male-Female Ratios

Male-female ratios are needed to develop a weighted average of annual earnings. Historically, females have had higher survival rates than males which results in a female majority that increases as the population ages. Weighting earnings with male-female ratios allows average earnings to reflect the gender composition of the birth cohort at each age in the age profile.

Male-female ratios are forecasted by adjusting the male-female ratios present at age zero for each subsequent age using the survival rates from the life tables. The baseline male-female ratios are calculated from the 2014 National Population Projections dataset, 2016 birth cohort ([ ADDIN EN.CITE <EndNote><Cite><Author>U.S. Census

Bureau</Author><RecNum>239</RecNum><DisplayText>U.S. Census Bureau 2014a</DisplayText><record><rec-number>239</rec-number><foreign-keys><key app="EN" db-id="9e2tazdvmef5axewx2ovx2xd92psp9ft9ets"

timestamp="1476715335">239</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><authors>U.S. Census

Bureau,,</author></authors></contributors></title>></title>2014 National Population Projections</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

10</number><dates><year>2014</year><pub-dates><date>Apr. 6, 2017</date></pub-dates></dates><urls><related-urls><url>https://www2.census.gov/programs-surveys/popproj/datasets/2014/2014-popproj/np2014\_d1.csv</url></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></related-urls></re>

#### **B.3.4** Present Value of Lifetime Earnings

Age-earnings profiles are estimated for the years 2008 through 2017. Average earnings are calculated for each year, Step 1, and then the ten averages are averaged to produce the grand mean earnings estimates presented in this report, Step 2:

#### Step 1

$$NPV_{yr} = \sum_{N=3}^{80} S_m \left( \frac{Y_{N,m} P_{N,m} R_{N,m} (1+X)^{N-A+0.5}}{((1+r)^{N-A+1})} \right) + S_f \left( \frac{Y_{N,f} P_{N,f} R_{N,f} (1+X)^{N-A+0.5}}{((1+r)^{N-A+1})} \right)$$

where:

NPV = the net present value of earnings between ages A and 80; yr= survey year from 2008 to 2017 A = current age, which is 3 in this analysis;

<sup>&</sup>lt;sup>17</sup> Value Economics LLC requested the life tables from the SSA. The SSA does not publish the tables on their website.

S =share of total population that is male or female;

N = ages in the future (3, ..., 80);

Y = average annual earnings among earners for a particular age (N);

P = survival rate for a particular age (N);

R = percent of population with earnings;

m = male indicator;

f = female indicator;

X = productivity rate assumed at the midpoint of age N; and

r = discount rate for the beginning of age N.

#### Step 2

$$NPV_{average} = \sum_{yr=2007}^{2016} \frac{\left(NPV_{yr}\right)}{10}$$

where:

NPV = the net present value of earnings between ages A and 80; and yr = survey year;

Lifetime earnings are discounted at both 3 percent and 7 percent, and the model assumes that real earnings will increase by 1 percent per year. Earnings are assumed to be zero for individuals under the age of 16 and over the age of 80. [ REF \_Ref532978217 \h ] presents average lifetime earnings for males, females, and for the population from 2007 to 2016.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Lifetime Earnings (2017\$)

Earnings Group	Discount	Rate
Earnings Group	3%	7%
Population	\$884,342	\$205,144
Male	\$1,069,129	\$244,738
Female	\$695,243	\$164,27

#### **B.4** Effects on Earnings from Changes in IQ

The value of an IQ point is estimated by considering the direct and indirect effects of changes in IQ on earnings, as well as the costs associated with additional years of education. The cost of additional education is comprised of the direct cost of education and forgone earnings while in school.

Section [ REF \_Ref402356387 \r \h \\* MERGEFORMAT ] describes the parameter used to estimate the total value of an IQ point, Section [ REF \_Ref402356468 \r \h ] describes the methods used to estimate the components of costs associated with additional education, and Section [ REF \_Ref402356509 \r \h ] presents the estimate for the net value of an IQ point.

#### **B.4.1** Total Value of an IQ Point

The EPA's updated analysis based on Salkever ([ADDIN EN.CITE ADDIN EN.CITE.DATA]) yields a value of an IQ point equal to 1.865 percent and 3.397 percent of earnings for males and females, respectively. The average value of an IQ point is \$23,269 and \$5,398 discounted at a rate of 3 percent and 7 percent, respectively.

#### **B.4.2** Additional Education: Costs and Lost Earnings

The effect of lead exposure on schooling is estimated to be 0.0811 years per IQ point for males and 0.0916 years per IQ point for females in the EPA's reanalysis of Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]). Increases in educational attainment signify analogous increases in education costs and diminished earnings resulting from working fewer hours while in school. To accurately portray the benefits of a population with a higher average IQ, the average value of an IQ point is adjusted to reflect the cost of additional education and the opportunity cost of being in school.

ACS PUMS data on enrollment were used to estimate both: (1) the ages at which the marginal increases in educational attainment due to higher IQ would be realized, and (2) the levels of additional educational attainment that would be achieved.

The distribution of ages at which the last year of education was obtained was estimated based on changes in the percentages of the population enrolled in school by age. For example, if 95 percent of 16 year olds are enrolled in school and 90 percent of 17 year olds are enrolled in school, it was estimated that 5 percent (95 percent minus 90 percent) of the population attained their last year of education at age 16. Thus, the analysis assumes that for 5 percent of the population, the marginal increase in educational attainment would be realized at age 17 – one year after they achieved their highest level of education in the baseline.

The level of additional education attained is estimated based on the enrollment for the ages at which the additional education is realized. For example, if 30 percent of enrolled 18 year olds are enrolled in 12th grade and 70 percent of enrolled 18 year olds are enrolled in university, the analysis estimates that additional education attained at age 18 is 30 percent 12th grade and 70 percent university.

#### **B.4.2.1** Education Costs

Education costs associated with pursuing additional education are determined by the level of education being pursued. Three categories of education are considered in estimating costs:

- Public elementary and secondary schools (expenditures per pupil enrolled);
- Average undergraduate tuition (including fees, room, and board); and
- Average graduate tuition (including required fees).

Data describing the cost of education are taken from the 2016 version of the Digest of Education Statistics ([ ADDIN EN.CITE

<EndNote><Cite><Author>Snyder</Author><Year>2016</Year><RecNum>623</RecNum>< DisplayText>Snyder & DisplayText><record><rec-number>623</rec-

number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp" timestamp="1529359494">623</key></foreign-keys><ref-type name="Report">27</ref-type><contributors><author>>Contributors><author>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributors><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor><title>Contributor

For each age in the age profile an individual has a likelihood of pursuing additional education in one of the three categories listed above, and therefore of incurring corresponding education costs. Enrollment is used to determine each category's relative likelihood thereby allowing costs to be apportioned appropriately with respect to each age in the age profile. For example, at age 17, most individuals will be enrolled in secondary schools causing the cost of education to primarily reflect the costs associated with the first category of education.

The total cost of education for the target cohort is estimated by calculating the sum of education costs corresponding to each level of attainment, weighted by enrollment and the probability of pursuing additional education at age N. For example, suppose that 100 percent of the target cohort's females attain additional education at age 18 or age 19. Twenty-five percent of the target cohort females attain additional education at age 18 and 75 percent of the target cohort females attain additional education at age 19. At age 18, 30 percent of the target cohort currently enrolled in school is enrolled in 12th grade and 70 percent is enrolled in an undergraduate university. At age 19, 10 percent of the target cohort females currently enrolled in school are enrolled in 12th grade and 90 percent are enrolled in an undergraduate university. The annual cost of a secondary school and an undergraduate university is \$12,000 and \$20,000, respectively. As shown in [ REF \_Ref532978255 \h ], the weighted cost of education per IQ point (assuming 0.0916 additional years of education) for the target cohort females is \$1,722, \$403 and \$1,319 for 18 and 19 year olds, respectively.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Example of Education Cost Calculation

	Enro	llment	Annual C	ost of Ed.	Additional Ed.	Weighted	Cost per IQ
Age	12th Grade	University	12th Grade	University	Weight	Costs	Point
	а	b	С	d	e	f=((a·c)+(b·d))·e	f0916
18	30%	70%	£40.000	#20.000	25%	\$4,400	\$403
19	10%	90%	\$12,000	\$20,000	75%	\$14,400	\$1,319
Tota	<u> </u>			·		\$18,800	\$1,722

#### B.4.2.2 Lost Earnings

Lost earnings are calculated by separately estimating annual earnings for individuals enrolled in school and not enrolled in school according to the method detailed in Section [ REF \_Ref402355963 \n \h ], and taking the difference of the two estimates. The difference in earnings is assumed to be zero for a given age and sex combination if enrolled and/or not enrolled

earnings estimates rely on fewer than 150 observations. As with education costs, the differences in annual earnings are weighted by the probability that an individual will pursue additional education at their current age, N, as described at the beginning of this section. Lost earnings associated with an additional IQ point are generated by applying the EPA's IQ effect on schooling to the difference in annual earnings between the enrolled and not enrolled populations.

#### **B.4.2.3** Summary of Additional Education Costs

The average estimated costs for additional education (tuition plus lost earnings) are \$1,592 and \$691, discounted at a rate of 3 percent and 7 percent, respectively.

#### **B.4.3** Net Value of an IQ Point

The net value of an IQ point is equal to the total value minus the costs of attaining additional education. The estimates presented in [REF\_Ref532978264 \h] are averages produced from ten years of ACS single-year PUMS covering the years 2008 through 2017.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Effects of a One Point Change in IQ on Earnings (2017\$)

Estimate Parameter	Discount R	ate
Estimate Parameter	3%	7%
IQ value	\$23,269	\$5,398
Additional education costs and lost earnings	\$1,592	\$691
IQ value without education costs and lost earnings	\$21,677	\$4,707

#### **B.5** Discussion of Methods

This section includes a discussion of the methods used to estimate the lifetime earnings stream.

#### **B.5.1** Cross-Sectional Data

A single cross-section of data provides economic earnings data for individuals of different ages at one point in time. Thus, each age represented in a cross-section also represents a different birth cohort. Age-earnings profiles are often constructed with one cross-section of earnings data; however, there exists much controversy over this method. Even though this analysis averages the age-earnings profiles of 10 cross-sections of data, it is not a substitute for longitudinal data. The consensus is that age-earning profiles based on cross-sectional data underestimate earnings as individuals near retirement. If this is the case, lifetime earnings estimates based on cross-sectional data would be biased downward.

#### **B.5.2** School Enrollment

The ACS PUMS do not have a variable indicating whether individuals are enrolled in school as full-time or part-time students. Because an additional year of educational attainment ostensibly refers to a student enrolled in school full-time for one academic school year, inclusion of part-time students may bias lost earnings estimates downwards, thereby resulting in a higher valuation of an IQ point. However, lost earnings estimates are based primarily on individuals enrolled in secondary school and undergraduate degree programs, where full-time enrollment is more common.

#### **B.5.3** Labor Force Participation Rate

Average earnings are estimated as a function of average labor force participation rates from 2008 to 2017. The participation rates of some subpopulations will likely change over the course of the target cohort's lifespan, especially the female and elderly participation rates. Assuming the female and elderly participation rates continue to rise, estimates presented in this report would underestimate average lifetime earnings.

#### **B.6** Limitations

#### **B.6.1** Compensatory and Special Education

Compensatory education and special education costs are not reflected in the IQ point dollar value presented in this report. Compensatory education may be needed for individuals who experience high blood lead levels during infancy and special education may be needed for individuals with an IQ below 70. Reducing exposure to lead at an early age is expected to reduce the incidence of children requiring compensatory and/or special education, which would in turn lower associated costs. Though these costs are not anticipated to be a substantial component of the overall benefits, they do represent a potential benefit of reducing lead exposure.

#### **B.6.2** Nonmarket Work

The use of earnings is an incomplete measure of an individual's value to society. This is particularly true for individuals who choose to not participate in the labor force for all of their working years. If the opportunity cost of non-wage compensated work is assumed to be the average wage earned by persons of the same sex, age, and education, the average lifetime earnings estimates for these people would be significantly higher.

#### **B.6.3** Multiple Degrees

Individuals receiving a second degree within the same attainment group later in life are not taken into account. For example, a second Master's degree is treated as two individuals with a Master's degree. This would increase the proportion of individuals with degrees from postsecondary schooling institutions. Given that a minority of individuals would fall into this attainment category, the lifetime earnings estimate is not expected to be substantially impacted.

#### **B.6.4** Earnings Growth Rate

The earnings growth rate for females is substantially higher than the rate for males. A portion of the difference is likely due to females approaching equal pay to males over time. However, a portion of the difference is likely an artifact of rising female labor force participation, hours worked per year, and educational attainment. These trends have tapered off overtime and in all likelihood will continue to do so. Therefore, the average growth rate of real wages for the 2017 birth cohort may be closer to the earnings growth rate for males compared to older cohorts.

#### **B.6.5** Income vs. Earnings

Although earnings is a significant component of income for the middle quintiles of the distribution of income, Barth et al. ([ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Barth</Author><Year>1984</Year><RecNum>620</RecNum><D isplayText>1984</DisplayText><record><rec-number>620</rec-number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp"

timestamp="1529357814">620</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author>Barth, Michael C. </author><author>Janney III, Asa M.</author><author>Arnold, Frank</author><author>Sheiner, Louis</author></authors></contributors><title>A Survey of the Literature Regarding the Relationship between Measures of IQ and Income</title><secondary-title>EPA Contract 68-01-6614</secondary-title></title></dates><year>1984</year></dates><publisher>ICF, Washington, DC</publisher><urls></urls></record></Cite></EndNote>]) indicates that income may be substantially different from earnings for the tails of the income distribution. Therefore, earnings are assumed to be a good proxy for income in accordance with Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]).

#### **B.6.6** Benefits

Because benefits have become a greater portion of income, their exclusion could result in lifetime earnings to be underestimated.

#### **B.6.7** Impact of Ability on Earnings

Barth et al. ([ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Barth</Author><Year>1984</Year><RecNum>620</RecNum><D isplayText>1984</DisplayText><record><rec-number>620</rec-number><foreign-keys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp"

timestamp="1529357814">620</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author>Barth, Michael C. </author><author>Janney III, Asa M.</author><author>Arnold, Frank</author><author>Sheiner,

Louis</author></authors></contributors><title>A Survey of the Literature Regarding the Relationship between Measures of IQ and Income</title><secondary-title>EPA Contract 68-01-6614</secondary-title></title></dates><year>1984</year></dates><publisher>ICF, Washington, DC</publisher><urls></urls></record></Cite></EndNote>]) found that many studies concluded that the impact of ability on earnings increases with age. Therefore, estimates

of the impact of IQ on earnings derived from younger individuals may be smaller than if estimates were derived from individuals over the age of 30 ([ ADDIN EN.CITE

<EndNote><Cite><Author>Barth</Author><Year>1984</Year><RecNum>620</RecNum><D isplayText>Barth et al. 1984</DisplayText><record><rec-number>620</rec-number><foreignkeys><key app="EN" db-id="astfddffk2fvdhe99z7p9wreavxffwde9dwp"

timestamp="1529357814">620</key></foreign-keys><ref-type name="Report">27</reftype><contributors><author>Barth, Michael C. </author><author>Janney III, Asa M.</author><author>Arnold, Frank</author><author>Sheiner,

Louis</author></authors></contributors><title>> Survey of the Literature Regarding the Relationship between Measures of IQ and Income</title><secondary-title>EPA Contract 68-01-6614</secondary-title></title></dates><year>1984</year></dates><publisher>ICF, Washington, DC</publisher><urls></record></Cite></EndNote>]). The EPA reanalysis of Salkever ([ ADDIN EN.CITE ADDIN EN.CITE.DATA ]) relies on data where respondents range in age from 27 to 32. Thus, the estimate of the impact of IQ on earnings may underestimate the true impact of ability on earnings.

### Appendix C. Detailed Cost Calculations and Methods

This appendix provides detailed calculations for the cost analysis described in Section [ REF \_Ref525051776 \r \h ]. Section [ REF \_Ref536441817 \r \h ] provides detailed cost curves and the blending equation used to estimate control costs. Section [ REF \_Ref536441838 \r \h ] describes the sources and methods used to estimate PWS wage rates. Section [ REF \_Ref536441856 \r \h ] provides detailed calculations of the household-level control costs.

#### C.1 Capital and O&M Control Cost Details

To generate costs for the treatment technologies discussed in Section [ REF \_Ref523415174 \r \h ], the EPA used its WBS cost-estimating models. The WBS models are spreadsheet-based engineering models for individual treatment technologies that are linked to a central database of component unit costs.

For each scenario modeled and separately for total capital and for O&M costs, the EPA fit up to three curves: one covering small systems (less than 1 MGD design flow), one covering medium systems (1 MGD to less than 10 MGD design flow), and one covering large systems (10 MGD design flow and greater). For each curve fit, the EPA selected from among several possible equation forms: linear, quadratic, cubic, power, exponential, and logarithmic. The EPA chose the form that resulted in the best correlation coefficient (R<sup>2</sup>), subject to the requirement that the equation must be monotonically increasing over the appropriate range of flow rates (i.e., within the flow rate category, the equation must always result in higher estimated costs for higher flow systems than for lower flow systems).

For the selected technology (medium-cost perchlorate-selective 170,000 BV), the costs are calculated based on the following equation:

$$Cost = C7 \times Q^3 + C8 \times Q^2 + C9 \times Q + C10$$

where Q is treatment process design flow (MGD) for total capital costs or treatment process average flow (MGD) for annual O&M costs. [ REF \_Ref536441529 \h ] shows the values for C7 through C10 for the selected technology based on the water source (groundwater or surface water) and system size.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Capital and O&M Control Cost Curve Parameters

Source	Size	Cost Type	C7	C8	C9	C10	Useful Life
GW	Small	O&M	56351.5	-50187.5	122627.1	4364.815	17.69411765
GW	Medium	O&M	0	0	133861.3	29816.62	31.74
GW	Large	O&M	0	0	122526.4	48879.59	33.95882353
SW	Small	O&M	0	-18636.8	118095.4	4401.098	17.69411765
SW	Medium	O&M	0	0	136673.8	28097.73	31.74
SW	Large	O&M	0	0	123719.9	66316.38	33.97058824
GW	Small	Total capital	111665.4	-213210	435508.3	125329.7	17.69411765
GW	Medium	Total capital	802.1603	-17543.4	455922.7	687296.4	31.74
GW	Large	Total capital	0	0	287995.3	1130876	33.95882353
SW	Small	Total capital	111750.5	-213364	435584.9	125332.8	17.69411765
SW	Medium	Total capital	830.5377	-17999.3	457555.7	685792.9	31.74

Source	Size	Cost Type	C7	C8	C9	C10	Useful Life
SW	Large	Total capital	0	0	287250.1	1144962	33.97058824

Additionally, given the high perchlorate removal efficiencies achieved by the treatment technology, the EPA assumes that systems can blend treated water and untreated water to meet the MCL. As such, the EPA applied the above equations using treatment process flows that account for the blending rate, which is the proportion of influent water that must be treated. For example, a blending rate of 0.6 means 60 percent of the water is treated and then blended with 40 percent untreated water. This rate depends on baseline perchlorate concentration, the treatment target concentration, and the removal efficiency of the treatment process (i.e., the percent of baseline perchlorate removed during treatment). For a treatment efficiency of 95 percent (or 0.95), the following equation defines the treatment target concentration of perchlorate ( $P_t$ ) as a weighted average of the baseline concentration ( $P_b$ ) and the treated water concentration [ $P_b$  x (1-0.95)] where the weights – based on the blending rate (B) – are (B) for the untreated water and B for the treated water:

$$P_t = (1 - B) \times P_b + B \times (P_b \times (1 - 0.95)).$$

Rearranging terms to solve for B (the blending rate) shows that the blending rate increases when the baseline concentration increases or the treatment target concentration decreases.

$$B = \frac{(P_b - P_t)}{P_b \times 0.95}$$

In turn, the equation presented above uses treatment process flow as the independent variable. Treatment process flow can be calculated from entry point flow by incorporating the blending rate as follows:

Treatment Process Flow =  $B \times Entry Point Flow$ 

#### C.2 PWS Labor Rate Calculations

Section [ REF \_Ref536453331 \r \h ] summarizes the PWS labor rate used to estimate the administrative and reporting burden. This section describes the calculation of the labor rate in more detail. First, the EPA identified the labor rates in the WBS models for technical staff (i.e., a full-time treatment plant operators) and managerial staff (i.e., utility managers for smaller systems and environmental managers for larger systems). Then, EPA calculated a weighted wage rate for each system size range based on these labor rates and assumed division of labor among technical and managerial staff. [ REF \_Ref536453391 \h ] summarizes this calculation for each size range category; the weighted wage rate is calculated as follows:

$$Weighted = \sum_{i=1}^{7} CWS_i (R_{t,i} \times W_{t,i} + R_{m,i} \times W_{m,i}) / \sum_{i=1}^{7} CWS_i$$

Where:

Weighted = Overall system wage rate for labor associated with the proposed rule

R = hourly loaded wage rate

W = assigned weight for labor category

t = indicator for technical labor category

m = indicator for managerial labor category

s = indicator for size stratum

CWS = number of CWS by size stratum.

#### Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: PWS Wage Rate Assumptions (2017\$)

System Size	Technical	Managerial	Technical	Managerial	Weighted	Number of
(Population)	Wage Rate <sup>a</sup>	Wage Rate <sup>a</sup>	Weight	Weight	-	Systems <sup>c</sup>
0-100	\$31.91	\$45.24	90%	10%	\$33.24	10,838
101-500	\$31.91	\$45.24	90%	10%	\$33.24	14,166
501-1,000	\$31.91	\$45.24	90%	10%	\$33.24	5,117
1,001-3,300	\$31.91	\$45.24	90%	10%	\$33.24	7,697
3,301-10,000	\$34.05	\$51.74	70%	30%	\$39.36	4,663
10,001-50,000	\$35.94	\$57.64	70%	30%	\$42.45	2,937
50,001-100,000	\$37.52	\$67.25	70%	30%	\$46.44	445
> 100,000	\$43.84	\$71.85	70%	30%	\$52.24	339

a. Source: Based on EPA WBS models. For more information, see USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"p1kUSUMx","properties":{"formattedCitation":"(2018g)","plainCitation":"(2018g)","noteIndex":0}, "citationItems":[{"id":977,"uris":["http://zotero.org/groups/945096/items/VUJUPN7L"],"uri":["http://zotero.org/gro ups/945096/items/VUJUPN7L"], "itemData": {"id":977, "type": "report", "title": "Technologies and Costs for Treating Perchlorate-Contaminated Waters", "publisher": "EPA \*\*\*-\*-\*-\*\*\*, "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2018"]]}}, "suppress-

author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

b. Calculated as technical wage rate times technical weight plus managerial wage rate times managerial weight.

c. Source: USEPA [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"parC4dj0","properties": {"formattedCitation":"(2018f)","plainCitation":"(2018f)","noteIndex":0},"cit ationItems":[{"id":923,"uris":["http://zotero.org/groups/945096/items/CW34PNAZ"],"uri":["http://zotero.org/groups /945096/items/CW34PNAZ"],"itemData": {"id":923,"type":"article","title":"Safe Drinking Water Information System Federal Reports

Search","URL":"https://ofmpub.epa.gov/apex/sfdw/f?p=108:200:....","author":[{"literal":"USEPA"}],"issued":{"da te-parts":[["2018"]]}, "accessed": {"date-parts":[["2018",8,28]]}}, "suppress-

author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Includes community water systems.

For the systems serving more than 3,300, the EPA based the weight for technical and managerial labor on labor force data reported in [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "SBFUcslE", "properties": {"formattedCitation": "(USEPA, 2009c)", "plainCitation": "(USEPA,

2009c)","noteIndex":0},"citationItems":[{"id":924,"uris":["http://zotero.org/groups/945096/item s/DZNAAV6M"],"uri":["http://zotero.org/groups/945096/items/DZNAAV6M"],"itemData": {"id ":924,"type":"article", "title": "2006 Community Water System Survey - Volume II: Detailed Tables and Survey

Methodology", "URL": "https://www.epa.gov/dwstandardsregulations/community-water-system-

survey", "author": [{"literal": "USEPA"}], "issued": {"date-parts": [["2009",5]]}, "accessed": {"date-parts": [["2018",8,17]]}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Comparing the managerial and treatment plant operator labor hours for all employment classes indicates that operator labor accounts for 70 percent of aggregate manager and operator labor hours while managerial hours account for 30 percent. Although the implied labor hour split was similar for smaller systems, the number of smaller systems reporting managerial hours was approximately one-third of those reporting treatment plant operator hours. Assuming one-third of systems serving up to 3,300 people have managers account for 30 percent of labor time, but the other two-thirds have no managerial labor time, the overall average is 10 percent managerial time.

The EPA calculated an overall weighted average wage rate across CWS based on the number of CWS in each size range. This yields an overall wage rate of \$34.71 for PWSs. The EPA does not have wage rate data for NTNCWS. In the absence of data, the EPA used the CWS average wage rate as a proxy value for NTNCWS systems.

#### **C.3** Household-Level Control Cost Calculations

This section shows how the EPA calculated household-level annual costs based on total annualized treatment costs. The analysis focuses on treatment costs because the cost of initial and follow-up monitoring will have a negligible impact on households.

First, the EPA estimated the number of households served by dividing PWS population by the median persons per household (PPH) for the service area associated with the PWS based on U.S. Census Bureau data. Then, the EPA summed treatment costs across all of a system's entry points and divided the total cost by the number of households. [ REF \_Ref534206622 \h ] shows this calculation.

Exhibit [ STYLEREF 5 \s ]-[ SEQ Table \\* ARABIC \s 5 ]: Calculation of Household-Level Annual Control Costs

Entry Point	PWS	PWS	PPH	PWS	Total En	try Point ed Cost <sup>d</sup>		PWS ed Cost <sup>e</sup>	House Annua	
-		Pop <sup>a</sup>		Households	3%	7%	3%	7%	3%	7%
MCL = 56 μg/L					·					
FL6280250_004_08004	FL6280250	38,761	2.67	14,517	\$157,776	\$200,214	\$157,776	\$200,214	\$11	\$14
PR0002702_2702004_00004	PR0002702	25,972	2.77	9,376	\$644,017	\$751,577	\$644,017	\$751,577	\$69	\$80
$MCL = 18 \mu g/L$										
FL6280250_001_08001					\$197,908	\$246,137				
FL6280250_002_08002	FL6280250	38,761	2.67	14,517	\$208,039	\$257,730	\$630,485	\$780,477	\$43	\$54
FL6280250_004_08004					\$224,539	\$276,611				
FL6411132_POE1_08001	FL6411132	198,500	2.67	74,345	\$2,385,945	\$2,710,362	\$2,385,945	\$2,710,362	\$32	\$36
GA2190000_323_15152	GA2190000	15,231	2.72	5,600	\$142,776	\$181,117	\$142,776	\$181,117	\$25	\$32
LA1089001_3CAA-6_00001T	LA1089001	24,081	2.62	9,191	\$338,089	\$403,270	\$338,089	\$403,270	\$37	\$44
MD0120001_0100000_00001	MD0120001	13,800	2.68	5,149	\$199,068	\$245,137	\$199,068	\$245,137	\$39	\$48
MD0120002_0100000_00001	MD0120002	12,002	2.68	4,478	\$196,194	\$241,914	\$196,194	\$241,914	\$44	\$54
MS0750005_7500502_00004T	MS0750005	4,309	2.65	1,626	\$79,736	\$103,513	\$79,736	\$103,513	\$49	\$64
NM3528616_003_00003	NM3528616	16,500	2.66	6,203	\$111,127	\$146,997	\$111,127	\$146,997	\$18	\$24
NV0000076_EP04_00206	NV0000076	220,000	2.70	81,481	\$1,923,507	\$2,187,850	\$1,923,507	\$2,187,850	\$24	\$27
OH0900715_EP001_00008	OH0900715	42,097	2.43	17,324	\$347,464	\$414,047	\$347,464	\$414,047	\$20	\$24
OH4401612_EP001_00006	OH4401612	25,091	2.43	10,326	\$450,128	\$529,521	\$450,128	\$529,521	\$44	\$51
OK2002412_UCM0001_11032	OK2002412	35,031	2.60	13,473	\$563,740	\$656,196	\$563,740	\$656,196	\$42	\$49
PA6200036_00101E_00100	PA6200036	16,000	2.47	6,478	\$322,605	\$386,584	\$322,605	\$386,584	\$50	\$60
PR0002702_2702004_00004	PR0002702	25,972	2.77	9,376	\$677,224	\$789,394	\$677,224	\$789,394	\$72	\$84

Entry Point	PWS	PWS Pop <sup>a</sup>	PPH	PWS Households <sup>c</sup>	Total En		Total Annualiz		House Annua	
TX1100002_04003_04003	TX1100002	13,805	2.88	4,793	\$179,074	\$222,667	\$179,074	\$222,667	\$37	\$46

a. PWS population based on USEPA [ ADDIN ZOTERO ITEM CSL CITATION

- {"citationID":"AxwMJrpi","properties":{"formattedCitation":"(2018d)","plainCitation":"(2018d)","noteIndex":0},"citationItems":[{"id":969,"uris":["http://zotero.org/groups/945096/items/YERQWPRZ"],"itemData":{"id":969,"type":"article","title":"Perchlora te Occurrence and Monitoring Report","publisher":"EPA \*\*\*-\*\*-\*\*\*,"author":[{"literal":"USEPA"}],"issued":{"date-parts":[["2018"]]}},"suppressauthor":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].
- b. Persons per household (PPH) is state-specific average household size from U.S. Census Bureau [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"i17uUQKL","properties":{"formattedCitation":"(2017c)","plainCitation":"(2017c)","noteIndex":0},"citationItems":[{"id":985,"uris":["http://zotero.org/groups/945096/items/CGU3LT9N"],"uri":["http://zotero.org/groups/945096/items/CGU3LT9N"],"itemData":{"id":985,"type":"article","title":"Average Household Size of Occupied Housing Units by Tenure. American Community Survey 1-Year Estimates: Table B25010","author":[{"family":"U.S. Census Bureau","given":""}],"issued":{"date-parts":[["2017"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].
- c. Calculated as PWS population divided by PPH.
- d. Calculated based on method described in Section [ REF \_Ref523415174 \r \h ].
- e. Calculated as the sum of all entry point costs for the PWS; in cases where only one entry point incurs control costs, PWS costs equal entry point costs.
- f. Calculated as total PWS annualized costs divided by PWS households.



# Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water

Office of Water (4607M) EPA \*\*\*-\*-\*\*\* November 2018 www.epa.gov/safewater

# **Table of Contents**

[ TOC \o "1-2" \h \z \u ]

# **Table of Exhibits**

[ TOC \h \z \t "Exh Title,1" ]

# **Abbreviations and Acronyms**

AWWARF American Water Works Research Foundation

BAT best available technology
EBCT empty bed contact time

EPA Environmental Protection Agency

ESTCP Environmental Security Technology Certification Program

GAC granular activated carbon
MCL maximum contaminant limit
MHI median household income
MGD million gallons per day
mg/L milligrams per liter

NF nanofiltration

O&M operating and maintenance

POU point-of-use RO reverse osmosis

SDWA Safe Drinking Water Act

SSCT small system compliance technology

UF ultrafiltration

μg/L micrograms per liter

WBS Work Breakdown Structure

### 1 Introduction

The U.S. Environmental Protection Agency (EPA) is considering setting a federal maximum contaminant limit (MCL) for perchlorate in drinking water under the Safe Drinking Water Act (SDWA). This document addresses treatment technologies that drinking water systems could use to meet this potential new MCL. Specifically, it provides an evaluation of several technologies against predefined criteria to determine whether they might be considered best available technologies (BATs) to meet the potential MCL. In addition, it provides an evaluation of technologies for small systems against criteria to determine whether they can be designated small system compliance technologies (SSCT).

The three technologies included in the BAT evaluation are: ion exchange, biological treatment, and reverse osmosis (RO). [REF\_Ref330453648 \\* MERGEFORMAT] provides a list of the six major criteria considered for the BAT evaluation, along with specific evaluation questions. Sections [REF\_Ref330470839 \r ] through [REF\_Ref525297841 \r \h ] provide a discussion of the extent to which each technology meets the BAT criteria. Section [REF\_Ref525297883 \r \h ] provides a summary of the BAT evaluation results. The detailed discussion is based primarily on literature search information and technical analysis conducted during development of the document, *Technologies and Costs for Treating Perchlorate-Contaminated Water* [ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"ehEOwiCI","properties": {"formattedCitation":"(USEPA, 2018)", "plainCitation":"(USEPA,

2018)","noteIndex":0},"citationItems":[{"id":246,"uris":["http://zotero.org/groups/945096/items/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData":{"id":246,"type":"article","title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters","publisher":"EPA \*\*\*-\*-\*-

\*\*\*\*","author":[{"family":"USEPA","given":""}],"issued":{"date-parts":[["2018"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. That document contains more complete description of each technology and the state of science regarding their use for perchlorate treatment.

The SDWA, as amended in 1996, requires that EPA list technologies for small systems [Section 1412(b)(4)(E)(ii)]:

The Administrator shall include in the list any technology, treatment technique, or other means that is affordable, as determined by the Administrator in consultation with the States, for small public water systems serving -

- (I) a population of 10,000 or fewer but more than 3,300;
- (II) a population of 3,300 or fewer but more than 500; and
- (III) a population of 500 or fewer but more than 25;

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<sup>&</sup>lt;sup>1</sup> Granular activated carbon (GAC) is not included in this evaluation. Although there have been a few studies on the use of specially-modified GAC media for perchlorate removal, there have been no full-scale demonstrations of the technology and no apparent effort to certify the modified GAC media as safe for drinking water use.

and that achieves compliance with the MCL or treatment technique, including packaged or modular systems and point-of-entry or point-of-use treatment units (POU).

Section [ REF \_Ref330470934 \r ] of this document provides EPA's analysis to identify SSCTs for the proposed rule. Specifically, it evaluates four technologies against the affordability and compliance effectiveness criteria for SSCTs. The technologies are the three included in the BAT analysis and POU reverse osmosis. EPA's affordability criterion uses an affordability threshold of 2.5 percent of the median household income (MHI) of the median water system (as ranked by MHI) in each small system size category (i.e., systems serving populations of (1) 25 - 500; (2) 501 - 3,300; and (3) 3,301 - 10,000 people). As long as the sum of baseline expenditures on water (i.e., current costs excluding perchlorate treatment costs) and the incremental expenditures associated with a particular perchlorate treatment technology do not exceed 2.5 percent of MHI, then that technology meets the affordability criterion.

### Exhibit [ SEQ Exhibit \\* ARABIC ]. BAT Criteria for Perchlorate Technologies Evaluation

# 1. High Removal Efficiency 1.1. Have high removal efficiencies that achieve potential MCLs been documented? 1.2. Are the effects of water quality parameters on treatment effectiveness and reliability well-known?

- 1.3. Is the technology reliable enough to continuously meet a drinking water MCL?
- 1.4. Is additional research needed?

#### 2. History of Full-Scale Operation

- 2.1. Do existing studies include full-scale operations at drinking water treatment facilities?
- 2.2. Are there studies of full-scale treatment of residuals that fully characterize residual waste streams and disposal options?
- 2.3. Can the bench or pilot studies be scaled up to represent full-scale treatment, including residuals generation and handling?
- 2.4. Is additional research needed?

#### 3. General Geographic Applicability

- 3.1. What regions do the existing research studies represent?
- 3.2. Is it known that regional water quality variations will limit treatment effectiveness or reliability in some areas?
- 3.3. Are there any regional issues with respect to residuals handling or water resource use?
- 3.4. Is additional research needed?

#### 4. Compatibility with Other Treatment Processes

- 4.1. Have the effects (adverse or beneficial) of the treatment process on other processes likely to be present at existing plants been evaluated?
- 4.2. Will additional pre- or post-treatment be required for integration into an existing (or planned) treatment train?
- 4.3. Is additional research needed?

#### 5. Ability to Bring All of the Water System into Compliance

- 5.1. Will the treatment process adversely affect the distribution system or water resource decisions?
- 5.2. Might the treatment process, residuals handling, or pre- or post-treatment requirements raise new environmental quality concerns?
- 5.3. Is additional research needed?

#### 6. Reasonable Cost Basis for Large and Medium Systems

- 6.1. Is the technology currently used by medium and large systems (including uses for other treatment purposes)?
- 6.2. Do the treatment studies provide sufficient information on design assumptions to allow cost modeling?
- 6.3. Is additional research needed?

### 2 Best Available Technology Evaluation for Ion Exchange

The State of California has identified ion exchange (along with fluidized bed biological treatment) as one of two BATs for achieving compliance with its standard for perchlorate in drinking water (CCR, Title 22, Chapter 15, Section 64447.2). Ion exchange is a physical/chemical separation process in which an ion (such as perchlorate) in the feed water is exchanged for an ion (typically chloride) on a resin generally made of synthetic beads or gel. A variety of resin types have been tested for perchlorate removal. These resin types include strong-base polyacrylic, strong-base polystyrenic (including nitrate-selective), weak-base polyacrylic, weak-base polystyrenic, and perchlorate-selective.<sup>2</sup>

In application, feed water passes through a bed of resin in a vessel or column. The operation typically continues until the resin is exhausted, meaning that the chloride on enough of the resin's available exchange sites has been replaced with ions from the feed water that the resin is no longer effective for removing the ion. At this point, the resin may be disposed and replaced or regenerated. Based on data from full-scale operations (see below), it is likely that most systems using ion exchange to comply with a perchlorate MCL would use a perchlorate-selective resin that would be disposed, rather than regenerated, when exhausted. This resin choice has implications for technology feasibility, particularly with regard to residuals management, as discussed below.

### 2.1 High Removal Efficiency for Ion Exchange

### 2.1.1 Have high removal efficiencies that achieve potential MCLs been documented?

Yes. The literature documents perchlorate removal efficiencies for ion exchange that are typically in the high 90 percent range and to levels well below the potential MCLs, especially when using perchlorate-selective resin. This includes results from studies conducted in the laboratory, in the field at pilot scale, and in full-scale application. [ REF \_Ref296588903 \h ] summarizes the removal efficiencies and resulting concentrations reported in the literature. Ion exchange with various types of resin is capable of removing perchlorate to levels below 4 micrograms per liter ( $\mu g/L$ ), even given very high influent perchlorate concentrations. For perchlorate-selective resins, the research has shown that levels below 1 to 2  $\mu g/L$  are achievable.

<sup>&</sup>lt;sup>2</sup> While Tripp et al. (2003) also examined strong base polyvinylpyridine resins, comparable quantitative data on their removal efficiency are not available.

# Exhibit [ SEQ Exhibit \\* ARABIC ]. Perchlorate Effectiveness Results for Ion Exchange

		Resulting		
	Removal	Concentration	Study	
Resin Type <sup>1</sup>	Efficiency	(µg/L)	Scale <sup>2</sup>	Data Source(s)
SB	>77% to >94%	<4	Р	[ADDIN ZOTERO_ITEM CSL_CITATION  {"citationID":"jwZ3ZYct","properties":{"formattedCitation":"(Ground-Water Remediation Technologies Analysis Center (GWRTAC), 2001; Venkatesh, Klara, Jennings, & Wagner, 2000)", "plainCitation":"(Ground-Water Remediation Technologies Analysis Center (GWRTAC), 2001; Venkatesh, Klara, Jennings, & Wagner, 2000)", "dontUpdate":true,"noteIndex":0}, "citationItems":[{"id":1 145, "uris":["http://zotero.org/groups/945096/items/EKWDU7IW"],"uri":["http://zotero.org/groups/945096/items/EKWDU7IW"],"it emData":{"id":1145, "type":"report", "title":"Technology Status Report: Perchlorate Treatment Technologies, First Edition", "collection-title":"Technology Status Report TS-01-01", "author":[{"family":"Ground-Water Remediation Technologies Analysis Center (GWRTAC)", "given":""}], "issued":{"dateparts":[["2001"]]}}}, "id":1054, "uris":["http://zotero.org/groups/945096/items/3YQ6X7IZ"], "uri":["http://zotero.org/groups/945096/items/3YQ6X7IZ"], "itemData":{"id":1054, "type":"chapter", "title":" Removal and Destruction of Perchlorate and Other Anions from Ground Water Using the ISEP+™ System", "containertitle":"Perchlorate in the Environment", "publisher":"Kluwer Academic/Plenum", "publisher-place":"New York, NY", "event-place":"New York, NY", "author":[{"family":"Venkatesh", "given":"K.R."}, {"family":"Kl ara", "given":"S.M."}, {"family":"Jennings", "given":"D.L."}, {"family":"Kl ara", "given":"S.M."}, {"family":"Jennings", "given":"D.L."}, {"family":"Kl ara", "given":"N.J."}], "editor":[{"family":"Urbansky", "given": "E.T."}], "issued":{"date-parts":[["2000"]]}}}], "schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}]
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Deeler Tree -4	Removal	Resulting Concentration	Study	Data Carranta)
Resin Type <sup>1</sup>	Removal Efficiency >97.5% to >98.1%	Concentration (µg/L) <2,000	Study Scale <sup>2</sup>	Status Report TS-01-01", "author": [("family": "Ground-Water Remediation Technologies Analysis Center (GWRTAC)", "given": ""], "issued": ["date-parts": [["2001"]]]}}, ["id":1104, "uris": ["http://zotero.org/groups/94 5096/items/YJ33SNDL"], "uri": ["http://zotero.org/groups/945096/items/YJ33SNDL"], "itemData": ["id":1104, "type". "article", "title": "Treatment of Perchlorate in Water", "publisher": "EPA Region 9", "author": [["family": "Praskins", "given": "W."]], "issued": ["date-parts": [["2003", 5]]]}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"} ]  [ADDIN ZOTERO_ITEM CSL_CITATION ("citationID": "QjCZa1X1", "properties": ["formattedCitation": "(Ground-Water Remediation Technologies Analysis Center (GWRTAC), 2001; Praskins, 2003; Wagner & Drewry, 2000)", "plainCitation": "(Ground-Water Remediation Technologies Analysis Center (GWRTAC), 2001; Praskins, 2003; Wagner & Drewry, 2000)", "noteIndex": 0}, "citationItems": [["id":1145, "uris": ["http://zotero.org/groups/945096/items/EKWDU7IW"], "uri": ["http://zotero.org/groups/945096/items/EKWDU7IW"], "itemData": ("id":1145, "type": "report", "title": "Technology Status Report: Perchlorate Treatment Technologies, First Edition", "collection-title": "Technology Status Report TS-01-01", "author": [["family": "Ground-Water Remediation Technologies Analysis Center (GWRTAC), "given": ""], "issued": ["idte-parts": ["2001"]])}}, "id": 1104, "uris": ["http://zotero.org/groups/945096/items/YJ33SNDL"], "uri": ["http://zotero.org/groups/945096/items/YJ33SNDL"], "uri": ["http://zotero.org/groups/945096/items/AAA66WXV"], "uri": ["http://zotero.org/groups/945096/items/AAA66WXV"], "itemData": ["id":1104, "type": "article", "title": "Treatment of Perchlorate in Water", "publisher": "EPA Region 9", "author": ["family": "Praskins", "given": "W."], "issued": ["date-parts": ["2003", 5]]}}, "id": 1056, "uris": ["http://zotero.org/groups/945096/items/AAA66WXV"], "itemData": ["id": 11056, "type": "speech", "title": "Commercial Systems Treati
				Waste Management Conference & Exhibition", "event-place": "San Antonio, TX", "author": [{"family": "Wagner", "given": "N."}, {"family": "Drewry", "given": "C."}], "issued": {"date-parts": [["2000", 8]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]
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Desir T 4	Removal	Resulting Concentration	Study	Data Comments
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Resin Type <sup>1</sup>	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
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Resin Type¹	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
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			<u> </u>	Totalioning . log rooti , properties . Fromattegottation . (Hay

Resin Type <sup>1</sup>	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
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Resin Type¹	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
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Resin Type <sup>1</sup>	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
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#### Notes:

<sup>1.</sup> SB = strong-base; SB-S = strong-base polystyrenic; SB-A = strong-base polyacrylic; WB-S = weak-base polystyrenic; WB-A = weak-base polyacrylic; NS = nitrate-selective strong-base polystyrenic; PS = perchlorate selective

<sup>2.</sup> L = laboratory study; P = field pilot study; F = full-scale

### 2.1.2 Are the effects of water quality parameters on treatment effectiveness and reliability well-known?

Yes. Effectiveness varies depending on water quality, but for perchlorate-selective resins the effect is limited. The most significant raw water quality consideration in ion exchange perchlorate treatment is the concentration of competing anions (particularly sulfate, nitrate, bicarbonate, and chloride). The effect of these anions is to decrease a resin's longer-term capacity to adsorb perchlorate, as they compete with perchlorate for exchange sites. There are significant differences among resin types in terms of the relative impact of competing anions. This impact is related to the relative affinity of the resin for each anion present. The order of affinity for perchlorate-selective resins is as follows [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "UTFMr5pk", "properties": {"formattedCitation": "(Boodoo, 2003)", "plainCitation": "(Boodoo, 2003)", "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/items/ 2W4DAVDK"], "uri": ["http://zotero.org/groups/945096/items/2W4DAVDK"], "itemData": {"id": 986,"type":"article-journal","title":"POU/POE removal of perchlorate","container-title":"Water Conditioning and Purification", "volume": "45", "issue": "8", "author": [{"family": "Boodoo", "given": "F"}], "issued": {" date-parts":[["2003",8]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]:

perchlorate > nitrate > sulfate > chloride > bicarbonate.

In particular, the perchlorate affinity relative to nitrate affinity is nearly an order of magnitude greater [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"tVye3EEQ", "properties": {"formattedCitation":"(Boodoo, 2003)", "plainCitation": "(Boodoo, 2003)", "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/items/ 2W4DAVDK"], "uri": ["http://zotero.org/groups/945096/items/2W4DAVDK"], "itemData": {"id": 986, "type": "article-journal", "title": "POU/POE removal of perchlorate", "container-title": "Water Conditioning and Purification", "volume": "45", "issue": "8", "author": [{"family": "Boodoo", "given": "F"}], "issued": {" date-parts":[["2003",8]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}]. Although Boodoo [ADDIN ZOTERO ITEM CSL CITATION {"citationID":"1GQXVrQv","properties": {"formattedCitation":"(2003)", "plainCitation":"(2003)" "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/items/2W4D, "noteIndex":0}, "citationItems":[{"id":986, "uris":["http://zotero.org/groups/945096/items/2W4D, "uris":["http://zotero.org/groups/945096/items/2W4D] AVDK"], "uri": ["http://zotero.org/groups/945096/items/2W4DAVDK"], "itemData": {"id":986, "ty pe":"article-journal", "title": "POU/POE removal of perchlorate", "container-title": "Water Conditioning and Purification", "volume": "45", "issue": "8", "author": [{"family": "Boodoo", "given": "F"}], "issued": {" date-parts":[["2003",8]]}}, "suppress-author":true}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] suggests that perchlorate-selective resins would be negatively affected by high nitrate concentrations, a multitude of studies show that

these resins are not, in fact, very sensitive to competing anions. Perchlorate capacity remains

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high for a wide range of nitrate and sulfate concentrations [ ADDIN ZOTERO ITEM

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Lutes et al., 2010; J. Min, Boulos, & Brown, 2003; Russell et al., 2008; Tripp et al., 2003; Wu &
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e-parts":[["2010",3,31]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ].
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Although most investigators identify bicarbonate and chloride as other major competing anions, the affinity of ion exchange resins, particularly perchlorate-selective resins, for these anions is less than that for perchlorate, sulfate, and nitrate. Therefore, their impact on resin perchlorate capacity would be expected to be less than that of sulfate and nitrate. There are, however, no quantitative data in the literature on the effects of these major anions. Other co-contaminants that may affect perchlorate capacity include arsenic [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"WcuhRWxg","properties": {"formattedCitation":"(Berlien, 2003; Tripp et al., 2003)","plainCitation":"(Berlien, 2003; Tripp et al., 2003)","noteIndex":0},"citationItems":[{"id":992,"uris":["http://zotero.org/groups/945096/items/8PB22K95"],"uri":["http://zotero.org/groups/945096/items/8PB22K95"],"itemData": {"id":992,"type":"report","title":"La Puente Valley County Water District's Experience with ISEP", "collection-title":"Presentation of Carollo Engineers, Inc. and Association of California

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CSL CITATION {"citationID": "AbXppyV6", "properties": {"formattedCitation": "(J. Min et al.,
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"Brown", "given": "J." ], "issued": {"date-
parts":[["2003",4]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. Based on the high affinity of most resins for
perchlorate, direct competition from these co-contaminants would be expected to be low.
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### 2.1.3 Is the technology reliable enough to continuously meet a drinking water MCL?

Yes. Numerous full-scale drinking water facilities are using ion exchange to meet the State of California's MCL for perchlorate (see Question [ REF Ref525556890 \r \h ], below). In general,

ion exchange is an established, reliable technology that has been used successfully to meet other MCLs.

#### 2.1.4 Is additional research needed?

No. Additional research is not required.

### 2.2 History of Full-Scale Operation for Ion Exchange

### 2.2.1 Do existing studies include full-scale operations at drinking water treatment facilities?

Yes. The literature identifies 44 full-scale facilities applying ion exchange for perchlorate removal [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"Db8Iyu4N","properties": {"formattedCitation":"(USEPA, 2018)", "plainCitation":"(USEPA, 2018)", "noteIndex":0}, "citationItems":[{"id":246,"uris":["http://zotero.org/groups/945096/items/VUJUPN7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData": {"id":246, "type":"article", "title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters", "publisher":"EPA \*\*\*-\*-\*
\*\*\*\*\*", "author":[{"family":"USEPA", "given":""}], "issued": {"date-parts":[["2018"]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Many of these facilities are drinking water treatment facilities. With the majority of these facilities located in California, the full-scale use of the technology appears to be focused on compliance with that State's MCL for perchlorate.

The data on full-scale facilities demonstrate the increasing use of perchlorate-selective resins. Currently, more than half of the identified full-scale facilities (18 of 23 facilities where information on resin type is available) use perchlorate-selective resins. An additional two facilities are reportedly planning to switch to perchlorate-selective resin [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"8nzutk01","properties":{"formattedCitation":"(Blute, 2012; Wu & Blute, 2010)","plainCitation":"(Blute, 2012; Wu & Blute,

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Document","author":[{"family":"Blute","given":"N.K."}],"issued":{"date-parts":[["2012",12]]}}}, {"id":1058,"uris":["http://zotero.org/groups/945096/items/2QPEXW23"],"uri":["http://zotero.org/groups/945096/items/2QPEXW23"],"itemData":{"id":1058,"type":"spe ech","title":"Perchlorate Removal Using Single-Pass Ion Exchange Resin - Pilot Testing Purolite A532E at the San Gabriel B6 Plant","publisher-place":"Hollywood, CA","event":"2010 California-Nevada AWWA Spring Conference","event-place":"Hollywood, CA","author":[{"family":"Wu","given":"X."},{"family":"Blute","given":"N.K."}],"issued":{"dat e-parts":[["2010",3,31]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. Thus, perchlorate-selective resin appears to have become the technology of choice for perchlorate ion exchange facilities.

### 2.2.2 Are there studies of full-scale treatment of residuals that fully characterize residual waste streams and disposal options?

Yes. Almost 79 percent (30 of 38) of the full-scale perchlorate ion exchange facilities for which waste management data are available operate on a throwaway basis. This statistic includes all but one of the full-scale facilities using perchlorate-selective resin. An additional two facilities are reportedly planning to switch away from regeneration to disposal of spent resin [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"83c3WTYX","properties": {"formattedCitation":"(Blute, 2012; Wu & Blute, 2010)", "plainCitation":"(Blute, 2012; Wu & Blute,

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Document","author":[{"family":"Blute","given":"N.K."}],"issued":{"date-parts":[["2012",12]]}}},{"id":1058,"uris":["http://zotero.org/groups/945096/items/2QPEXW23"],"uri":["http://zotero.org/groups/945096/items/2QPEXW23"],"itemData":{"id":1058,"type":"speech","title":"Perchlorate Removal Using Single-Pass Ion Exchange Resin - Pilot Testing Purolite A532E at the San Gabriel B6 Plant","publisher-place":"Hollywood, CA","event":"2010 California-Nevada AWWA Spring Conference","event-place":"Hollywood, CA","author":[{"family":"Wu","given":"X."},{"family":"Blute","given":"N.K."}],"issued":{"date-parts":[["2010",3,31]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. These systems generate solid waste in the form of spent resin loaded with perchlorate and other anions. The facilities that operate with resin regeneration are older facilities. New facilities installed to comply with a perchlorate MCL would be unlikely to adopt this mode of operation.

The primary concern with spent resin is that hazardous co-contaminants (such as arsenic, uranium, and chromium) might accumulate on the resin. For example, Tripp et al. [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"LTmBSWYw","properties":{"formattedCitation":"(2003)","plainCitation":"(2003)","noteIndex":0},"citationItems":[{"id":1013,"uris":["http://zotero.org/groups/945096/items/J24BWNQH"],"uri":["http://zotero.org/groups/945096/items/J24BWNQH"],"itemData":{"id":1013, "type":"article","title":"Treatment of perchlorate in groundwater by ion exchange technology","publisher":"AWWA Research Foundation","call-number":"Publication No. 909","author":[{"family":"Tripp","given":"A.R."},{"family":"Clifford","given":"D."},{"family":"Roberts","given":"D.J."},{"family":"Cang","given":"Y."},{"family":"Aldridge","given":"L."},{"family":"Gillogly","given":"T."},{"family":"Boulos","given":"L."}],"issued":{"date-parts":[["2003"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] suggest that a perchlorate-selective resin would require frequent disposal to prevent arsenic and uranium build-up. Recent studies of various perchlorate-selective resins, however, have shown that build-up of metals results in concentrations that are below regulatory limits that would require disposal as a hazardous waste, both under federal requirements and California's more stringent limits [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"VVmZ1vIM","properties": {"formattedCitation":"(Blute et al., 2006; Russell et al., 2008; Wu & Blute, 2010)", "plainCitation":"(Blute et al., 2006; Russell et al., 2008; Wu &

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Blute.
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2010)", "noteIndex":0}, "citationItems":[{"id":987, "uris":["http://zotero.org/groups/945096/items/ 8Z7K9ZUJ"], "uri": ["http://zotero.org/groups/945096/items/8Z7K9ZUJ"], "itemData": {"id":987," type": "speech", "title": "Bench and Pilot Testing of High Capacity, Single-Pass Ion Exchange Resins for Perchlorate Removal", "publisher-place": "San Antonio, TX", "event": "2006 AWWA Annual Conference & Exposition", "event-place": "San Antonio, TX", "author": [{"family": "Blute", "given": "N.K."}, {"family": "Seidel", "given": "C.J."}, {"family": " McGuire", "given": "M.J." \}, \{ \[ \family \]: "Qin", \[ \given \]: "D." \}, \{ \[ \family \]: \[ \given \]: \[ \], \[ \] ssued": {"dateparts":[["2006",6]]}}},{"id":1097,"uris":["http://zotero.org/groups/945096/items/NLAFHBV2"], "uri":["http://zotero.org/groups/945096/items/NLAFHBV2"],"itemData":{"id":1097,"type":"spee ch", "title": "Pilot Testing of Single Pass Perchlorate-Selective Ion Exchange Resins at Three Utilities in the Main San Gabriel Basin", "publisher-place": "Cincinnati, OH", "event": "AWWA Water Quality Technology Conference & Exposition", "event-place": "Cincinnati, OH","author":[{"family":"Russell","given":"C.G."},{"family":"Qin","given":"G."},{"family":"B lute", "given": "N.K.", {"family": "McGuire", "given": "M.J.", {"family": "Williams", "given": "C."} ],"issued":{"date-

parts":[["2008",11]]}}},{"id":1058,"uris":["http://zotero.org/groups/945096/items/2QPEXW23"],"uri":["http://zotero.org/groups/945096/items/2QPEXW23"],"itemData":{"id":1058,"type":"spe ech","title":"Perchlorate Removal Using Single-Pass Ion Exchange Resin - Pilot Testing Purolite A532E at the San Gabriel B6 Plant","publisher-place":"Hollywood, CA","event":"2010 California-Nevada AWWA Spring Conference","event-place":"Hollywood, CA","author":[{"family":"Wu","given":"X."},{"family":"Blute","given":"N.K."}],"issued":{"dat e-parts":[["2010",3,31]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. The same studies found that uranium build-up might require special handling as a radioactive waste in only one of the 12 samples tested (total across all three studies).

Because of the shorter life of conventional (not perchlorate-selective) resins, metals accumulation in these resins likely would be even lower and, thus, the same result should hold true (although few full-scale systems would be expected to use these resins). A number of studies are also available characterizing spent regenerant [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"Hf5RObPs","properties": {"formattedCitation":"(Batista et al., 2003; Berlien, 2003; Case et al., 2004; Gu et al., 2002; Lutes et al., 2010; Montgomery Watson Harza (MWH) and University of Houston, 2003)", "plainCitation":"(Batista et al., 2003; Berlien, 2003; Case et al., 2004; Gu et al., 2002; Lutes et al., 2010; Montgomery Watson Harza (MWH) and University of Houston,

2003)" "potal play": () "citation Items": [ ("id": 003, "uris": ["lhttp://getara.org/graups/045006/items/

2003)","noteIndex":0},"citationItems":[{"id":993,"uris":["http://zotero.org/groups/945096/items/5MCR8IMA"],"uri":["http://zotero.org/groups/945096/items/5MCR8IMA"],"itemData":{"id":993,"type":"article-journal","title":"Combining Ion-Exchange (IX) Technology and Biological Reduction for Perchlorate Removal", "container-title":"Remediation", "page":"21-38", "volume":"13", "issue":"1", "author":[{"family":"Batista", "given":"J.R."}, {"family":"Gingras ","given":"T.M."}, {"family":"Vieira", "given":"A.R."}], "issued":{"date-parts":[["2003",1,2]]}}}, {"id":992, "uris":["http://zotero.org/groups/945096/items/8PB22K95"]," uri":["http://zotero.org/groups/945096/items/8PB22K95"], "itemData":{"id":992, "type":"report", "title":"La Puente Valley County Water District's Experience with ISEP", "collection-title":"Presentation of Carollo Engineers, Inc. and Association of California Water

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i":["http://zotero.org/groups/945096/items/WU4Z87VI"],"itemData":{"id":1116,"type":"report",
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Houston"}],"issued": {"date-parts":[["2003"]]}}}],"schema": "https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. Again though, few full-scale systems would
be expected to operate with resin regeneration.
```

# 2.2.3 Can the bench or pilot studies be scaled up to represent full-scale treatment, including residuals generation and handling?

Yes. As a mature and established technology, the scale-up of ion exchange, in general, from bench- to pilot- to full-scale is well understood.

#### 2.2.4 Is additional research needed?

No. Additional research is not required.

### 2.3 General Geographic Applicability for Ion Exchange

#### 2.3.1 What regions do the existing research areas represent?

Studies of ion exchange treatment of perchlorate have primarily been conducted in California and Nevada. For perchlorate-selective resin in particular, most recent studies have used water that is representative of those areas.

### 2.3.2 Is it known that regional water quality variations will limit treatment effectiveness or reliability in some areas?

No. Although most of the existing research is for a limited region, there are no data indicating that regional water quality variations will limit effectiveness or reliability. Given that the effect of source water quality parameters on perchlorate-selective resin is limited (see Question [ REF \_Ref525290149 \r \h ]), source water conditions in other regions are not likely to have a substantial impact.

### 2.3.3 Are there any regional issues with respect to residuals handling or water resource use?

There are regions where disposal of spent regenerant would be an issue. Few full-scale systems, however, would be expected to operate with resin regeneration. Regional barriers are not anticipated with respect to spent resin disposal unless co-occurring contaminants that accumulate on the resin are classified as hazardous or radioactive (see Question [ REF \_Ref525289868 \r \h ]).

#### 2.3.4 Is additional research needed?

No. Additional research is not required.

### 2.4 Compatibility of Ion Exchange with Other Treatment Processes

### 2.4.1 Have the effects (adverse or beneficial) of the treatment process on other processes likely to be present at existing plants been evaluated?

Yes. Ion exchange can have an adverse effect on treated water chemistry by increasing corrosivity (see Question [ REF\_Ref525290744 \r \h ]). The technology can also have a beneficial effect by removing other undesirable anions from the treated water (e.g., arsenic, uranium), even when using perchlorate-selective resin (see Questions [ REF\_Ref525290149 \r \h ] and [ REF\_Ref525289868 \r \h ]).

# 2.4.2 Will additional pre- or post-treatment be required for integration into an existing (or planned) treatment train?

Possibly. The treated water chemistry changes resulting from ion exchange might require post-treatment corrosion control or alter existing corrosion control or disinfection requirements.

#### 2.4.3 Is additional research needed?

No. Additional research is not required.

# 2.5 Ability of Ion Exchange to Bring all of the Water System into Compliance

### 2.5.1 Will the treatment process adversely affect the distribution system or water resource decisions?

Ion exchange treatment can increase the corrosivity of treated water [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "CZf30VJz", "properties": {"formattedCitation": "(Berlien, 2003; Betts, 1998; USEPA, 2005)", "plainCitation": "(Berlien, 2003; Betts, 1998; USEPA, 2005)", "noteIndex":0}, "citationItems":[{"id":992, "uris":["http://zotero.org/groups/945096/items/ 8PB22K95"], "uri": ["http://zotero.org/groups/945096/items/8PB22K95"], "itemData": {"id":992, "t ype":"report", "title": "La Puente Valley County Water District's Experience with ISEP", "collection-title": "Presentation of Carollo Engineers, Inc. and Association of California Water Agencies", "author": [{"family": "Berlien", "given": "M.J."}], "issued": {"dateparts":[["2003",4]]}}},{"id":991,"uris":["http://zotero.org/groups/945096/items/BNWD5VQP"], "uri":["http://zotero.org/groups/945096/items/BNWD5VQP"], "itemData": {"id":991, "type": "artic le-journal", "title": "Rotation ion-exchange system removes perchlorate", "page": "454A-455A", "volume": "32", "journal Abbreviation": "Environ. Sci. Technol.", "author": [{"family": "Betts", "given": "K.S."}], "issued": {"dateparts":[["1998"]]}}}, {"id":1046,"uris":["http://zotero.org/groups/945096/items/EWAQ4GEK"]," uri":["http://zotero.org/groups/945096/items/EWAQ4GEK"],"itemData":{"id":1046,"type":"artic le", "title": "Perchlorate Treatment Technology Update: Federal Facilities Forum Issue Paper","publisher": "Office of Solid Waste and Emergency Response. EPA 542-R-05-015", "author": [{"literal": "USEPA"}], "issued": {"date $parts": \hbox{\tt [["2005",5]]}\}\}], "schema": "https://github.com/citation-style-parts": \hbox{\tt [["2005",5]]}}\}], "schema": "https://github.com/citation-style-parts": \hbox{\tt [["2005",5]]}}], "schema": "https://github.com/citation-style-parts": "https$ language/schema/raw/master/csl-citation.json"}] because of the addition of chloride ions and/or removal of carbonates and bicarbonates. Berlien [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"v50fvsPm","properties": {"formattedCitation":"(2003)","plainCitation":"(2003)"," noteIndex":0}, "citationItems":[{"id":992, "uris":["http://zotero.org/groups/945096/items/8PB22K 95"], "uri": ["http://zotero.org/groups/945096/items/8PB22K95"], "itemData": {"id":992, "type": "re port", "title": "La Puente Valley County Water District's Experience with ISEP", "collectiontitle": "Presentation of Carollo Engineers, Inc. and Association of California Water Agencies", "author": [{"family": "Berlien", "given": "M.J."}], "issued": {"dateparts":[["2003",4]]}},"suppress-author":true}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ] reports this problem with a full-scale application of ion exchange for perchlorate treatment. Treated water had a pH of approximately 7 and created red water problems in older homes with galvanized steel pipe. The operators corrected this problem by adding sodium hydroxide to raise the pH to approximately 8.2 and adding polyphosphates as an additional protection measure. Thus, distribution system effects can be managed by adjusting corrosion control programs.

The potentially large volume of spent regenerant could be a barrier in regions with water resource issues. Few full-scale systems, however, would be expected to operate with resin regeneration.

### 2.5.2 Might the treatment process, residuals handling, or pre- or post-treatment requirements raise new environmental quality concerns?

The disposal of large volumes of spent regenerant could create an environmental quality concern. Few full-scale systems, however, would be expected to operate with resin regeneration.

#### 2.5.3 Is additional research needed?

No. Additional research is not required.

# 2.6 Reasonable Cost Basis for Ion Exchange for Large and Medium Systems

# 2.6.1 Is the technology currently used by medium and large systems (including uses for other treatment purposes)?

Yes. The 44 full-scale perchlorate ion exchange systems identified in the literature include a number of medium and large systems: 31 are larger than 1 million gallons per day (MGD) and six are larger than 10 MGD, with the largest being 14.4 MGD.

### 2.6.2 Do the treatment studies provide sufficient information for design assumptions to allow cost modeling?

Detailed data are available from the treatment studies for all of the relevant design parameters, including:

- Resin type
- Vessel configuration (i.e., number of vessels in series)
- Empty bed contact time (EBCT)
- Resin bed life
- Surface loading rate
- Regeneration parameters.

#### 2.6.3 Is additional research needed?

No. Additional research is not required.

### 3 Best Available Technology Evaluation for Biological Treatment

The State of California has identified biological treatment (along with ion exchange) as one of two BATs for achieving compliance with its standard for perchlorate in drinking water (CCR, Title 22, Chapter 15, Section 64447.2). Biological treatment of perchlorate is the process by which bacteria are used to reduce perchlorate to chlorate, chlorite, chloride, and oxygen. The process typically involves the addition of an oxidizable substrate (also referred to as the electron donor or "food"), such as acetate or ethanol. Biological treatment offers complete destruction of the perchlorate ion, eliminating the need for management of perchlorate-bearing waste streams.

The most promising designs for biological treatment of perchlorate at drinking water facilities are those that operate either in a fixed bed or a fluidized bed configuration. Both fixed bed and fluidized bed designs involve a media bed that provides a surface on which perchlorate-reducing bacteria grows. For fixed bed reactors, influent water is typically passed under pressure through a static media bed located in a vessel. An alternative fixed bed design uses a gravity-fed concrete basin to hold the biologically active media. Fluidized bed bioreactor designs use vessels where high influent rates in an up-flow design fluidize the media bed allowing for more surface area for biomass growth. California's BAT for perchlorate specifies fluidized bed biological treatment.

### 3.1 High Removal Efficiency for Biological Treatment

### 3.1.1 Have high removal efficiencies that achieve potential MCLs been documented?

Yes. [ REF \_Ref297622446 \h \\* MERGEFORMAT ] summarizes the removal efficiencies and resulting concentrations reported in the literature. It shows that fixed and fluidized bed reactors have consistently achieved removal efficiencies greater than 90 percent, reducing perchlorate to levels that are usually below detection limits of 4  $\mu$ g/L or lower, even given very high influent perchlorate concentrations. Most of the data in the exhibit are from laboratory-, pilot-, and field-scale tests of biological treatment. Also included, however, are data from several full-scale treatment systems.

# Exhibit [ SEQ Exhibit \\* ARABIC ]. Perchlorate Effectiveness Results for Biological Treatment

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
>99%	<4	Bench-scale fixed bed	None	Sand / Acetate	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"4MEIBxIB","pro perties":{"formattedCitation":" (Kim & Logan, 2000)","plainCitation":"(Kim & Logan, 2000)","noteIndex":0},"citatio nItems":[{"id":1135,"uris":["htt p://zotero.org/groups/945096/items/G9LNRV7T"],"uri":["htt p://zotero.org/groups/945096/items/G9LNRV7T"],"itemDat a":{"id":1135,"type":"article-journal","title":"Fixed-bed bioreactor treating perchlorate-contaminated waters","container-title":"Environmental Engineering Science","volume":"17","issu e":"5","author":[{"family":"Kim ","given":"K."},{"family":"Loga n","given":"B.E."}],"issued":{"date-parts":[["2000"]]}}}],"schema": "https://github.com/citation-style-language/schema/raw/maste r/csl-citation.json"}]
>99%	<4	Bench-scale fixed bed	Nitrate (0.02), sulfate (0.04)	Celite / Acetate	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citation D":"95JFe0tq","properties":{"formattedCitation":"(Losi, Giblin, Hosangadi, & Frankenberger, 2002)","plainCitation":"(Losi, Giblin, Hosangadi, & Frankenberger, 2002)","noteIndex":0},"citationItems":[{"id":1125,"uris":["http://zotero.org/groups/945096/items/PYWQUJ5P"],"uri":["http://zotero.org/groups/945096/items/PYWQUJ5P"],"itemData":{"id":1125,"type":"articlejournal","title":"Bioremediation of perchlorate-contaminated groundwater using a packed bed

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
					biological reactor", "container-title": "Bioremediation Journal", "page": "97- 104", "volume": "6", "issue": "2", "author": [{"family": "Losi", "give n": "M.E."}, ("family": "Giblin", "g iven": "T.", {"family": "Hosanga di", "given": "V.", {"family": "Fra nkenberger", "given": "W.T."}], "issued": {"date- parts": [["2002"]]}}}], "schema": "https://github.com/citation- style- language/schema/raw/maste r/csl-citation.json"}]
>98%	<3	Bench-scale fixed bed	Nitrate (13), sulfate (9.3 to16.8)	GAC / Acetic acid or proprietary carbohydrate solution	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"XsXEUMJv","pr operties":{"formattedCitation" :"(Upadhyaya, Kotlarz, Togna, & Raskin, 2015)","plainCitation":"(Upad hyaya, Kotlarz, Togna, & Raskin, 2015)","dontUpdate":true,"no telndex":0},"citationItems":[{"i d":1071,"uris":["http://zotero.or rg/groups/945096/items/KLW CLIE4"],"uri":["http://zotero.or g/groups/945096/items/KLW CLIE4"],"itemData":{"id":1071 ,"type":"article- journal","title":"Carbohydrate- Based Electron Donor for Biological Nitrate and Perchlorate Removal From Drinking Water","container- title":"Journal - American Water Works Association","page":"E674- E684","volume":"107","issue" :"12","source":"Wiley Online Library", "abstract":"This study evaluated the feasibility of replacing acetic acid with a commercial carbohydrate- based electron donor (CBED) for removal of nitrate and perchlorate (CIO4-) from drinking water. Bench-scale biologically active carbon fixed-bed and fluidized-bed reactors (FXBR and FLBR,

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
>94%	<4	Bench-scale fixed	Nitrate (4)	Sand, plastic media	respectively), with an initial empty bed contact time (EBCT) of 42.8 min, were fed simulated groundwater containing 15 mg/L nitrate as nitrogen and 200 µg/L CIO4—. EBCT in the FLBR after final expansion was 80.5 min. During the first 100 days using acetic acid at 125 mg/L chemical oxygen demand (COD), complete nitrate removal was achieved in both systems, whereas perchlorate in the FXBR and FLBR effluents remained below 3 and 6 µg/L CIO4—, respectively. For comparable removals, influent COD requirement was higher with the CBED. Biomass yields with acetic acid and the CBED were 0.54—0.58 and 0.59—0.74 mg CODsiomass/mg CODsubstrate, respectively. The higher yield with the CBED resulted in more frequent maintenance requirements.", "DOI": "10.594 2/jawwa.2015.107.0143", "IS SN": "1551-8833", "language": "en", "autho r": [{"family": "Upadhyaya", "giv en": "Giridhar"}, {"family": "Raskin", "given": "Ratily": "Kotl arz", "given": "Nadine"}, ("family": "Togna", "given": "Paul"}, ("family": "Raskin", "given": "Lutg arde"}], "issued": {"dateparts": [["2015", 12, 1]]}}], "sch ema": "https://github.com/citat ion-style-language/schema/raw/maste r/csl-citation.json"}]
/ <del>94</del> %	\ 4	bench-scale fixed	। ।।।।। ate (4)	Sand, plastic media / Acetic acid	CSL_CITATION  {"citationID":"qPM4HxKc","pr operties":{"formattedCitation" :"(Case et al., 2004; B. Min, Evans, Chu, & Logan, 2004)","plainCitation":"(Case et al., 2004; B. Min, Evans, Chu, & Logan,

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Removal	Resulting	Scale and	Other Analytes	Media / Electron	Data Source(s)
Efficiency	Concentration	Reactor Type	(mg/L)	Donor	
	(µg/L)				2004)","dontUpdate":true,"no
					teIndex":0},"citationItems":[{"i
					d":1173,"uris":["http://zotero.o
					rg/groups/945096/items/313Z
					VUFE"], "uri": ["http://zotero.or
					g/groups/945096/items/3I3Z VUFE"],"itemData":{"id":1173
					,"type":"report","title":"Perchlo
					rate Research Partnership -
					Overview and Technology
					Review","publisher":"America
					n Water Works Association
					Research
					Foundation", "publisher-
					place":"Denver, CO","event- place":"Denver,
					CO","author":[{"family":"Case
					","given":"T."},{"family":"Cann
					on","given":"F."},{"family":"Clif
					ford","given":"D."},{"family":"R
					ittman","given":"B."},{"family":
					"Logan", "given": "B."}, {"family"
					:"Evans", "given": "P."}, {"family
					":"Adham", "given": "S."}, {"fami
					ly":"Aldridge","given":"L."}],"is sued":{"date-
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					"uri":["http://zotero.org/groups
					/945096/items/VTV55N8F"],"i
					temData":{"id":1118,"type":"a
					rticle-
					journal","title":"Perchlorate removal in sand and plastic
					media
					bioreactors", "container-
					title":"Water
					Research","page":"47-
					60","volume":"38","issue":"1",
					"source":"PubMed","abstract"
					:"The treatment of
					perchlorate-contaminated groundwater was examined
					using two side-by-side pilot-
					scale fixed-bed bioreactors
					packed with sand or plastic
					media, and bioaugmented
					with the perchlorate-
					degrading bacterium
					Dechlorosoma sp. KJ.
					Groundwater containing
L		<u> </u>	<u> </u>		perchlorate (77microg/L),

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
					nitrate (4mg-NO(3)/L), and dissolved oxygen (7.5mg/L) was amended with a carbon source (acetic acid) and nutrients (ammonium phosphate). Perchlorate was completely removed (<4microg/L) in the sand medium bioreactor at flow rates of 0.063-0.126L/s (1-2gpm or hydraulic loading rate of 0.34-0.68L/m(2)s) and in the plastic medium reactor at flow rates of <0.063L/s. Acetate in the sand reactor was removed from 43+/-8 to 13+/-8mg/L (after day 100), and nitrate was completely removed in the reactor (except day 159). A regular (weekly) backwashing cycle was necessary to achieve consistent reactor performance and avoid short-circuiting in the reactors. For example, the sand reactor detention time was 18min (hydraulic loading rate of 0.68L/m(2)s) immediately after backwashing, but it decreased to only 10min 1 week later. In the plastic medium bioreactor, the relative changes in detention time due to backwashing were smaller, typically changing from 60min before backwashing. We found that detention times necessary for complete perchlorate removal were more typical of those expected for mixed cultures (10-18min) than those for the pure culture (<1min) reported in our previous laboratory studies. Analysis of intra-column perchlorate profiles revealed that there was simultaneous removal of dissolved oxygen, nitrate, and perchlorate, and that oxygen and nitrate

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
	(pg/L)				removal was always complete prior to complete perchlorate removal. This study demonstrated for the first time in a pilot-scale system, that with regular backwashing cycles, fixed- bed bioreactors could be used to remove perchlorate in groundwater to a suitable level for drinking water.","DOI":"10.1016/j.watr es.2003.09.019","ISSN":"004 3-1354","note":"PMID: 14630102","journalAbbreviati on":"Water Res.","language":"eng","auth or":[{"family":"Evans","given":"B ooki"},{"family":"Chu","given":"A llyson K."},{"family":"Logan","given": "Bruce E."}],"issued":("date- parts":[["2004",1]]}}}],"schem a":"https://github.com/citation -style- language/schema/raw/maste r/csl-citation.json"}]
>93%	<5	Full-scale fixed bed1	Nitrate	GAC / Acetic acid	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"MzwlaxyZ", "properties":{"formattedCitation":" (U.S. Department of Defense (U.S. DoD), 2008a)", "plainCitation":"(U.S. Department of Defense (U.S. DoD), 2008a)", "dontUpdate":true, "noteIndex":0}, "citationItems":[{"id":1074, "uris":["http://zotero.org/groups/945096/items/2Z CNIFHT"], "uri":["http://zotero.org/groups/945096/items/2Z CNIFHT"], "itemData":["id":1074, "type":"report", "title":"Direct Fixed-bed Biological Perchlorate Destruction Demonstration", "genre":"EST CP Final Report (ER-0544)", "author":["literal":"U.S. Department of Defense (U.S. DoD)"]], "issued":["date-

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
	., .				parts":[["2008",9,25]]}}], "sch ema":"https://github.com/citat ion-style- language/schema/raw/maste r/csl-citation.json"}]
>92%	<4	Bench-scale fixed bed	Sulfate (0 to 220)	GAC/Acetate or ethanol	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"rISAf9UG", "properties": ("formattedCitation":" (Brown, Snoeyink, Raskin, & Lin, 2003)", "plainCitation":" (Brown, Snoeyink, Raskin, & Lin, 2003)", "noteIndex":0}, "citationItems": [{"id":984, "uris": ["http://zotero.org/groups/945096/items/FFPYJC8D"], "uri": ["http://zotero.org/groups/945096/items/FFPYJC8D"], "itemData": {"id":984, "type": "articlejournal", "title": "The sensitivity of fixed-bed biological perchlorate removal to changes in operating conditions and water quality characteristics", "containertitle": "Water Research", "page": "206-214", "volume": "37", "issue": "1", "author": [{"family": "Brown", "given": "J.C."}, {"family": "Raskin", "given": "L."}, {"family": "Raskin", "given": "L."}, {"family": "Lin", "given": "R"}], "issued": {"date-parts": [["2003"]]}}}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]
92% to 99%	<4	Field-scale fixed bed <sup>4</sup>	Sulfate (140 to 250), Nitrate (6 to 29), DO (4 to 8)	GAC / Acetic acid	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"B5AvYc5U","pr operties":{"formattedCitation" :"(Brown et al., 2005; The Interstate Technology & Regulatory Council (ITRC) Team, 2008)","plainCitation":"(Brown et al., 2005; The Interstate Technology & Regulatory Council (ITRC) Team, 2008)","noteIndex":0},"citatio

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Removal	Resulting	Scale and	Other Analytes	Media / Electron	Data Source(s)
Efficiency	Concentration (µg/L)	Reactor Type	(mg/L)	Donor	, ,
					nltems":[{"id":983, "uris":["http://zotero.org/groups/945096/items/L8HMQ8VP"], "uri":["http://zotero.org/groups/945096/items/L8HMQ8VP"], "itemData": {"id":983, "type": "articlejournal", "title": "Fixed-bed biological treatment of perchlorate-contaminated drinking water", "containertitle": "Journal AWWA", "volume": "97", "issue ": "9", "author": [{"family": "Brown", "given": "J.C."}, {"family": "Anderson", "given": "R.D."}, {"family": "Boulos", "given": "L."}, {"family": "Boulos", "given": "C.J. G."}], "issued": {"dateparts": [["2005"]]}}, {"id":1083, "uris": ["http://zotero.org/groups/945096/items/5PV8GPIA"], "uri": ["http://zotero.org/groups/945096/items/5PV8GPIA"], "itemData": {"id":1083, "type": "article", "title": "Technical/Regulatory Guidance: Remediation Technologies for Perchlorate Contamination in Water and Soil", "URL": "http://www.eosremediation.com/download/Perchlorate/ITRC%20PERC-2.pdf", "author": [{"literal": "The Interstate Technology & Regulatory Council (ITRC) Team")], "issued": {"dateparts": [["2008", 3]]}, "accessed": {"dateparts": [["2008", 3]]}, "accessed": {"dateparts": [["2018", 10, 13]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}]
>99%	<0.5	Full-scale fluidized bed <sup>1</sup>	Various	GAC / Acetic acid	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"gMjfE7DX","pro perties":{"formattedCitation":" (U.S. Department of Defense (U.S. DoD), 2009; Webster & Crowley, 2010, 2016; Webster & Litchfield, 2017)","plainCitation":"(U.S.

Removal	Resulting	Scale and	Other Analytes	Media / Electron	Data Source(s)
Efficiency	Concentration (µg/L)	Reactor Type	(mg/L)	Donor	
					Department of Defense (U.S. DoD), 2009; Webster & Crowley, 2010, 2016; Webster & Litchfield, 2017)", "noteIndex":0}, "citatio nItems":[{"id":1081, "uris":["http://zotero.org/groups/945096/items/9FHLVTXY"], "uri":["http://zotero.org/groups/945096/items/9FHLVTXY"], "itemData":{"id":1081, "type":"report", "title":"Demonstration of a Full-Scale Fluidized Bed Bioreactor for the Treatment of Perchlorate at Low Concentrations in Groundwater", "genre":"Environmental Security Technology Certification Program (ESTCP) Final Report (ER-0543)", "author":[{"family":"U.S. Department of Defense (U.S. DoD)", "given":""}], "issued":["date-parts":[["2009"]]}}}, {"id":1019, "uris":["http://zotero.org/groups/945096/items/BI7SF8HW"], "uri":["http://zotero.org/groups/945096/items/BI7SF8HW"], "itemData":{"id":1019, "type":"speech", "title":"Full-Scale Implementation of a Biological Fluidized Bed Drinking Water Treatment Plant for Nitrate and Perchlorate Treatment", "publisher-place":"Ontario, CA", "event":"2010 Water Education Foundation Water Quality and Regulatory Conference", "event-place":"Ontario, CA", "author":[{"family":"Webster", "given":"T.D."}, {"family":"Crowley", "given":"T.J."}, "issued":{"date-parts":[["2010",11,3]]}}}, {"id":1018, "uris":["http://zotero.org/groups/945096/items/BI5LYMZP"], "uri":["http://zotero.org/groups/945096/items/BI5LYMZP"], "uris":["http://zotero.org/groups/945096/items/BI5LYMZP"], "uris":["http://zotero.org/groups/9450

Removal	Resulting	Scale and	Other Analytes	Media / Electron	Data Source(s)
Efficiency	Concentration (µg/L)	Reactor Type	(mg/L)	Donor	
					roups/945096/items/BI5LYM ZP"],"itemData":{"id":1018,"ty pe":"speech", "title":"Biologica I treatment of perchlorate in groundwater.","event":"AWW A Annual Conference and Exposition", "author":[{"family":"Webster", "given":"T.D."},{"family":"Crowley", "given":"T.J."}], "issued":{"date-parts":[["2016",6,21]]}}},{"id":1012, "uris":["http://zotero.org/groups/945096/items/64HZKA 2M"], "uri":["http://zotero.org/groups/945096/items/64HZKA 2M"], "itemData":{"id":1012, "ty pe":"article-journal", "title":"Full-scale biological treatment of nitrate and perchlorate for potable water production", "containertitle":"Journal AWWA", "page":"30-40", "volume":"109", "issue":"5", "author":[{"family":"Webster", "given":"T.D."}, {"family":"Litchfield", "given":"M.H."}], "issue d":{"date-parts":[["2017"]]}}}], "schema": "https://github.com/citationstyle-language/schema/raw/maste r/csl-citation.json"}]
>99%	<5	Bench-scale fluidized bed	Nitrate, metals, volatile organics	GAC / Acetic acid	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"BWN3iGiU","pr operties":{"formattedCitation":"(Polk et al., 2001)","plainCitation":"(Polk et al., 2001)","noteIndex":0},"citationItems":[{"id":1106,"uris":["htt p://zotero.org/groups/945096/items/82QHFD5E"],"uri":["htt p://zotero.org/groups/945096/items/82QHFD5E"],"itemDat a":{"id":1106,"type":"speech", "title":"Case Study of Ex-Situ Biological Treatment of Perchlorate-Contaminated Groundwater","publisher-place":"San Diego, CA", "event":"4th Tri-Services

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
	(Fg/ -)				Environmental Technology Symposium", "event-place": "San Diego, CA", "author": [{"family": "Polk", "given": "J."}, {"family": "Murray ", "given": "C."}, {"family": "One wokae", "given": "D.E."}, {"family": "Tolbert", "given": "D.E."}, {"family": "Togna", "given": "A.P." }, {"family": "Guarini", "given": "W.J."}, {"family": "Frisch", "given": "W.J."}, {"family": "Del Vecchio", "given": "M."}], "issue d": {"date-parts": [["2001", 6]]}}}], "schem a": "https://github.com/citation-style-language/schema/raw/maste r/csl-citation.json"}]
>99%	220 to 280	Bench-scale fluidized bed	Nitrate (15.4), sulfate (12.5)	GAC / Acetate or proprietary glycerol solution	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"sfS1st9B","properties":{"formattedCitation":"(Kotlarz, Upadhyaya, Togna, & Raskin, 2016)","plainCitation":"(Kotlarz, Upadhyaya, Togna, & Raskin, 2016)","noteIndex":0},"citationItems":[{"id":1132,"uris":["http://zotero.org/groups/945096/items/E5WRR4HD"],"itemData":{"id":1132,"type":"article-journal","title":"Evaluation of electron donors for biological perchlorate removal highlights the importance of diverse perchlorate-reducing populations","containertitle":"Environmental Science: Water Research & Technology","page":"1049-1063","volume":"2","author":[{"family":"Kotlarz","given":"N."},{"family":"Togna","given":"P."},{"family":"Togna","given":"P."},{"family":"Raskin","given":"P."},{"family:"Raskin","given":"P."},{"family:"Raskin","given":"P."},{"family:"Raskin","given":"P."},{"family:"Raskin","given":"Raskin","given":"P."},{"family:"Raskin","given":"Raski

Removal	Resulting	Scale and	Other Analytes	Media / Electron	Data Source(s)
Efficiency	Concentration (µg/L)	Reactor Type	(mg/L)	Donor	Data dourse(s)
					language/schema/raw/maste r/csl-citation.json"} ]
>99%	350 to <4	Full-scale fluidized bed <sup>2</sup>	Nitrate (1.9), sulfate (300)	GAC / Acetic acid, ethanol	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"hRCwdIbm","pr operties":{"formattedCitation":"(Polk et al., 2001)","plainCitation":"(Polk et al., 2001)","noteIndex":0},"citatio nItems":[{"id":1106,"uris":["htt p://zotero.org/groups/945096 /items/82QHFD5E"],"uri":["htt p://zotero.org/groups/945096 /items/82QHFD5E"],"itemDat a":{"id":1106,"type":"speech", "itille":"Case Study of Ex-Situ Biological Treatment of Perchlorate-Contaminated Groundwater","publisher-place":"San Diego, CA","event":"4th Tri-Services Environmental Technology Symposium","event-place":"San Diego, CA","author":[{"family":"Polk", "given":"J."},{"family":"One wokae","given":"D.E."},{"family":"Tolbert","given":"D.E."},{"family":"Togna","given":"A.P."},{"family":"Fisch","given":"W.J."},{"family":"Frisch","given":"S."},{"family":"Frisch","given":"W.J."},{"family":"Del Vecchio","given":"M."}],"issue d":{"date-parts":[["2001",6]]}}}],"schem a":"https://github.com/citation-style-language/schema/raw/maste r/csl-citation.json"}]
>99%	<2	Bench-scale fluidized bed	Sulfate (5 to 10)	GAC, sand / Ethanol, methanol, or mix	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"qSwnvSu3","pro perties":{"formattedCitation":" (Greene & Pitre, 2000)","plainCitation":"(Gree ne & Pitre, 2000)","dontUpdate":true,"no teIndex":0},"citationItems":[{"i d":1146,"uris":["http://zotero.org/groups/945096/items/LS4 6AD5H"],"uri":["http://zotero.o

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
	(1-3-1)				rg/groups/945096/items/LS4 6AD5H"],"itemData":{"id":114 6,"type":"chapter","title":"Trea tment of Groundwater Containing Perchlorate using Biological Fluidized Bed Reactors with GAC or Sand Media","container- title":"Perchlorate in the Environment","publisher":"KI uwer Academic/Plenum","publishe r-place":"New York, NY","event-place":"New York, NY","author":[{"family":"Gree ne","given":"M.R."},{"family":" Pitre","given":"M.P."}],"editor" :[{"family":"Urbansky","given": "E.T."}],"issued":{"date- parts":[["2000"]]}}}],"schema": "https://github.com/citation- style- language/schema/raw/maste r/csl-citation.json"}]
>99%	<4	Full-scale fluidized bed <sup>3</sup>	Not reported	GAC / Ethanol	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"sAW1jo38","pro perties":{"formattedCitation":" (Greene & Pitre, 2000)","plainCitation":"(Gree ne & Pitre, 2000)","dontUpdate":true,"no telndex":0},"citationItems":[{"i d":1146,"uris":["http://zotero.org/groups/945096/items/LS46AD5H"],"uri":["http://zotero.org/groups/945096/items/LS46AD5H"],"itemData":{"id":1146,"type":"chapter","title":"Treatment of Groundwater Containing Perchlorate using Biological Fluidized Bed Reactors with GAC or Sand Media", "containertitle":"Perchlorate in the Environment", "publisher":"Kluwer Academic/Plenum", "publisher r-place":"New York, NY", "event-place":"New York, NY", "author":[{"family":"Gree

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
					ne","given":"M.R."},{"family":" Pitre","given":"M.P."}],"editor" :[{"family":"Urbansky","given": "E.T."}],"issued":{"date- parts":[["2000"]]}}],"schema": "https://github.com/citation- style- language/schema/raw/maste r/csl-citation.json"}]
>97%	<6	Bench-scale fluidized bed	Nitrate (13), sulfate (9.3 to 16.8)	GAC / Acetic acid or proprietary carbohydrate solution	[ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"yEumWcAd","pr operties":{"formattedCitation" :"(Upadhyaya, Kotlarz, Togna, & Raskin, 2015)", "plainCitation":"(Upadhyaya, Kotlarz, Togna, & Raskin, 2015)", "dontUpdate":true, "no telndex":0}, "citationItems":[{"id":1071,"uris":["http://zotero.org/groups/945096/items/KLW CLIE4"], "uri":["http://zotero.org/groups/945096/items/KLW CLIE4"], "itemData":{"id":1071, "type":"article-journal", "title":"Carbohydrate-Based Electron Donor for Biological Nitrate and Perchlorate Removal From Drinking Water", "containertitle":"Journal - American Water Works Association", "page":"E674-E684", "volume":"107", "issue":"12", "source":"Wiley Online Library", "abstract":"This study evaluated the feasibility of replacing acetic acid with a commercial carbohydrate-based electron donor (CBED) for removal of nitrate and perchlorate (CIO4–) from drinking water. Bench-scale biologically active carbon fixed-bed and fluidized-bed reactors (FXBR and FLBR, respectively), with an initial empty bed contact time (EBCT) of 42.8 min, were fed simulated groundwater containing 15 mg/L nitrate as nitrogen and 200 μg/L CIO4–

Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
					. EBCT in the FLBR after final expansion was 80.5 min. During the first 100 days using acetic acid at 125 mg/L chemical oxygen demand (COD), complete nitrate removal was achieved in both systems, whereas perchlorate in the FXBR and FLBR effluents remained below 3 and 6 μg/L ClO4-, respectively. For comparable removals, influent COD requirement was higher with the CBED. Biomass yields with acetic acid and the CBED were 0.54–0.58 and 0.59–0.74 mg CODbiomass/mg CODsubstrate, respectively. The higher yield with the CBED resulted in more frequent maintenance requirements.", "DOI": "10.594 2/jawwa.2015.107.0143", "IS SN": "1551-8833", "language": "en", "autho r": [("family": "Upadhyaya", "giv en": "Giridhar"), {"family": "Kotl arz", "given": "Nadine"}, {"family": "Togna", "given": "Paul"}, {"family": "Raskin", "given": "Lutg arde"}, "issued": {"dateparts": [["2015", 12, 1]]}}], "sch ema": "https://github.com/citat ion-stylelanguage/schema/raw/maste r/csl-citation.json"}]
92 to 98%	<4	Field-scale fluidized bed⁵	Various	GAC / Ethanol	[ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"nj3wXeRQ","pro perties":{"formattedCitation":" (Gilbert, Clark, Kavanaugh, McCarty, & Trussell, 2001; Harding Engineering and Environmental Services (ESE), 2001)","plainCitation":"(Gilber t, Clark, Kavanaugh, McCarty, & Trussell, 2001; Harding Engineering and Environmental Services (ESE),

2001)*_"noteIndex"-0]*.claid   nltems*[\data"-1149, \undersigned \text{"interms/NHNCXT28"}, \undersigned \text{"interms/NHNCXT28"}, \undersigned \text{"interms/NHNCXT28"}, \undersigned \text{"interms/NHNCXT28"}, \undersigned \text{"interms/NHNCXT28"}, \undersigned \text{"intermotor org/groups/946966}   interms/NHNCXT28"], \undersigned \text{"intermotor org/groups/946966}   interms/NHNCXT28"], \undersigned \undersign	Removal Efficiency	Resulting Concentration (µg/L)	Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
style- language/schema/raw/maste r/csl-citation.json"}]						nltems":[{"id":1149, "uris":["htt p://zotero.org/groups/945096 /items/NHNGXTZ8"], "uri":["htt p://zotero.org/groups/945096 /items/NHNGXTZ8"], "itemDat a":["id":1149, "type":"report", "title":"Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California", "collectiontitle":"Expert Panel Final Report prepared for Aerojet", "author":[{"family":"Gilbert", "given":"J.B."}, {"family":"Clark", "given":"R."}, {"family":"Kavanaugh", "given":"M."}, {"family":"McCarty", "given":"P."}, {"family":"Trussell", "given":"R.R."}], "issued":{"dateparts":["2001"]]}}}, "id":1139, "uris":["http://zotero.org/groups/945096/items/ZPGXUZPL"], "uri":["http://zotero.org/groups/945096/items/ZPGXUZPL"], "itemData":{"id":1139, "type":"report", "title":"Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collectiontitle":"Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author":[{"family":"Harding Engineering and Environmental Services (ESE)", "given":""}], "issued":{"dateparts":["2001"]]}}}], "schema": "https://github.com/citationstylelanguage/schema/raw/maste

### Notes:

- 1. Rialto Well #2 site in Rialto, California
- 2. Longhorn Army Ammunition Plant in Karnak, Texas
- 3. Aerojet facility in Rancho Cordova, California
- 4. Six-month field test in Santa Clarita, California
- 5. Eight-month field test in Rancho Cordova, California, supplying water for potable use

## 3.1.2 Are the effects of water quality parameters on treatment effectiveness and reliability well-known?

Yes. As shown in [ REF Ref297622741 \h \\* MERGEFORMAT ], biological treatment remains effective even in the presence of certain co-occurring contaminants. Nitrate and sulfate were present in nearly all of the studies and did not appear to interfere with the removal efficiency of the process. Biological treatment also has been shown effective in the presence of metals, volatile organic compounds, and other contaminants including N-nitrosodimethlamine and 1,4-dioxane [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"LkCeiwr9", "properties": {"formattedCitation":"(Harding Engineering and Environmental Services (ESE), 2001; Polk et al., 2001; U.S. Department of Defense (U.S. DoD), 2000)", "plainCitation": "(Harding Engineering and Environmental Services (ESE), 2001; Polk et al., 2001; U.S. Department of Defense (U.S. DoD), 2000)", "noteIndex":0}, "citationItems":[{"id":1139, "uris":["http://zotero.org/groups/945096/item s/ZPGXUZPL"], "uri": ["http://zotero.org/groups/945096/items/ZPGXUZPL"], "itemData": {"id":1 139,"type":"report","title":"Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author": [{"family": "Harding Engineering and Environmental Services (ESE)", "given": ""}], "issued": {"dateparts":[["2001"]]}}},{"id":1106,"uris":["http://zotero.org/groups/945096/items/82QHFD5E"],"ur i":["http://zotero.org/groups/945096/items/82QHFD5E"],"itemData":{"id":1106,"type":"speech", "title": "Case Study of Ex-Situ Biological Treatment of Perchlorate-Contaminated Groundwater", "publisher-place": "San Diego, CA", "event": "4th Tri-Services Environmental Technology Symposium", "event-place": "San Diego, CA","author":[{"family":"Polk","given":"J."},{"family":"Murray","given":"C."},{"family":"One wokae", "given": "D.E." }, { "family": "Tolbert", "given": "D.E." }, { "family": "Togna", "given": "A.P." },{"family":"Guarini","given":"W.J."},{"family":"Frisch","given":"S."},{"family":"Del Vecchio", "given": "M," }], "issued": {"dateparts":[["2001",6]]}}}, {"id":1078,"uris":["http://zotero.org/groups/945096/items/U99PPGIM"]," uri":["http://zotero.org/groups/945096/items/U99PPGIM"],"itemData":{"id":1078,"type":"report ", "title": "Ammonium Perchlorate Biodegradation for Industrial Wastewater Treatment", "collection-title": "ESTCP Cost and Performance Report", "author": [{"literal": "U.S. Department of Defense (U.S. DoD)"}],"issued":{"dateparts":[["2000"]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"}].

Nevertheless, raw water quality plays a role in the design of a biological treatment system. In identifying design criteria for use in full-scale treatment plant designs, the Harding ESE [ ADDIN ZOTERO ITEM CSL CITATION

{"citationID":"zlliHyxF","properties": {"formattedCitation":"(2001)","plainCitation":"(2001)","n oteIndex":0},"citationItems":[{"id":1139,"uris":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"uri":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"itemData": {"id":1139,"type ":"report","title":"Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California","collection-title":"Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin,California","author":[{"family":"Harding Engineering and Environmental Services

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(ESE)", "given": ""}], "issued": {"date-parts": [["2001"]]}}, "suppress-
author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-
citation.json"} ] authors included expected raw water dissolved oxygen, nitrate, perchlorate, and
total phosphorous concentrations as necessary considerations, along with water temperature. In
particular, temperature plays an important role in determining the rate of biomass growth.
Electron donor dose requirements increase with decreasing temperature. At temperatures below
10 degrees C, biomass growth is inhibited and bioremediation becomes unfeasible [ ADDIN
ZOTERO ITEM CSL CITATION
{"citationID": "pNAKxFcY", "properties": {"formattedCitation": "(Dugan, 2010a, 2010b; Dugan et
al., 2009)", "plainCitation": "(Dugan, 2010a, 2010b; Dugan et al.,
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s/IIXUW45F"], "uri": ["http://zotero.org/groups/945096/items/IIXUW45F"], "itemData": {"id":115
9,"type":"article","title":"Supporting data for presentation: The Impact of Temperature on
Biological Perchlorate Removal and Downstream Effluent Polishing", "publisher": "U.S.
Environmental Protection Agency, Office of Research and Development, National Risk
Management Research
Laboratory", "author": [{"family": "Dugan", "given": "N.R."}], "issued": {"date-
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XS"],"uri":["http://zotero.org/groups/945096/items/X3WWHCXS"],"itemData":{"id":1160,"type
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Treatment", "publisher-place": "Seattle, WA", "event": "2009 AWWA Water Quality Technology
Conference & Exposition", "event-place": "Seattle,
WA", "author": [\{"family": "Dugan", "given": "N.R."\}, \{"family": "Williams", "given": "D.J."\}, \{"family": "Williams", "given": "D.J."], \{"family": "Williams", "given": "D.J."], ["family": "D.J."], ["family": "Williams", "given": "D.J."], ["family": "
ily":"Meyer", "given": "M."}, {"family": "Schneider", "given": "R.R."}, {"family": "Speth", "given": "
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language/schema/raw/master/csl-citation.json"}].
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## 3.1.3 Is the technology reliable enough to continuously meet a drinking water MCL?

Continuous destruction of perchlorate in a biological treatment system depends heavily on influent water temperature (see above under Question [ REF \_Ref525292787 \r \h ]). Thus, systems with seasonal variation in water temperature such that temperature drops below 10 degrees C in the winter months would not be able to rely on biological treatment year-round. Systems with a constant water temperature or one that remains warm enough year-round, on the other hand, should be able to continuously meet an MCL.

### 3.1.4 Is additional research needed?

No. Additional research is not required.

### 3.2 History of Full-Scale Operation for Biological Treatment

## 3.2.1 Do existing studies include full-scale operations at drinking water treatment facilities?

Yes. Although most of the full-scale systems are part of perchlorate remediation projects in which treated water is not used as drinking water, fluidized bed operations supplying drinking water do exist. For example, one remediation facility conducted an eight-month fluidized bed field test that supplied potable water to local water companies [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "S3DYMP8P", "properties": {"formattedCitation": "(Gilbert et al., 2001; Harding Engineering and Environmental Services (ESE), 2001)", "plainCitation": "(Gilbert et al., 2001; Harding Engineering and Environmental Services (ESE), 2001)", "noteIndex":0}, "citationItems":[{"id":1149, "uris":["http://zotero.org/groups/945096/item s/NHNGXTZ8"], "uri": ["http://zotero.org/groups/945096/items/NHNGXTZ8"], "itemData": {"id": 1149, "type": "report", "title": "Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California", "collection-title": "Expert Panel Final Report prepared for Aerojet", "author": [{"family": "Gilbert", "given": "J.B."}, {"family": "Clark", "given": "R."}, {"family": "R."}, {"family": "Clark", "given": "R."}, {"family": "Clark", "given": "R."}, {"family": "R."}, {"fam ":"Kavanaugh", "given": "M." }, { "family": "McCarty", "given": "P." }, { "family": "Trussell", "given": " R.R."}],"issued":{"dateparts":[["2001"]]}}},{"id":1139,"uris":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"u ri":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"itemData":{"id":1139,"type":"report" "title": "Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author": [{"family": "Harding Engineering and Environmental Services (ESE)", "given": ""}], "issued": {"date-parts": [["2001"]]}}}], "schema": "https://github.com/citationstyle-language/schema/raw/master/csl-citation.json"} ]. Furthermore, the success of several demonstration studies led to the design and installation of a full-scale fluidized bed system supplying drinking water to the West Valley Water District and the City of Rialto. This system completed construction in 2013 and the system underwent extensive testing before receiving its operating permit and beginning to produce drinking water in 2016 [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"gw7638Au","properties": {"formattedCitation":"(Webster & Crowley, 2010, 2016; Webster & Litchfield, 2017)", "plainCitation": "(Webster & Crowley, 2010, 2016; Webster & Litchfield, 2017)", "noteIndex":0}, "citationItems":[{"id":1019, "uris":["http://zotero.org/groups/945096/item s/BI7SF8HW"], "uri": ["http://zotero.org/groups/945096/items/BI7SF8HW"], "itemData": {"id":10 19,"type": "speech", "title": "Full-Scale Implementation of a Biological Fluidized Bed Drinking Water Treatment Plant for Nitrate and Perchlorate Treatment", "publisher-place": "Ontario, CA", "event": "2010 Water Education Foundation Water Quality and Regulatory Conference", "event-place": "Ontario, CA", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Crowley", "given": "T.J."}], "issu ed":{"dateparts":[["2010",11,3]]}}},{"id":1018,"uris":["http://zotero.org/groups/945096/items/BI5LYMZP "],"uri":["http://zotero.org/groups/945096/items/BI5LYMZP"],"itemData": {"id":1018,"type":"sp

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eech","title":"Biological treatment of perchlorate in groundwater.","event":"AWWA Annual Conference and Exposition","author":[{"family":"Webster","given":"T.D."},{"family":"Crowley","given":"T.J."}],"issued":{"date-parts":[["2016",6,21]]}}},{"id":1012,"uris":["http://zotero.org/groups/945096/items/64HZKA2 M"],"uri":["http://zotero.org/groups/945096/items/64HZKA2M"],"itemData":{"id":1012,"type": "article-journal","title":"Full-scale biological treatment of nitrate and perchlorate for potable water production","container-title":"Journal AWWA","page":"30-40","volume":"109","issue":"5","author":[{"family":"Webster","given":"T.D."},{"family":"Litch field","given":"M.H."}],"issued":{"date-parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].
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## 3.2.2 Are there studies of full-scale treatment of residuals that fully characterize residual waste streams and disposal options?

Yes. Because biological treatment offers complete destruction of the perchlorate ion, the technology does not generate a perchlorate-bearing waste stream. An active bioreactor, however, will have a continuous growth of biomass. Assuming the addition of a sufficient amount of electron donor substrate, the quantity of biomass generated will depend on the concentrations of dissolved oxygen, nitrate, and perchlorate available for consumption. In most bioreactor designs, excess biomass must be removed periodically, which results in one or more residual streams.

In fixed bed bioreactors, biomass removal typically is accomplished using a backwash process,

which generates spent backwash water containing the excess biosolids (and some lost media). This backwash water is non-toxic and can typically be discharged to a local sewer [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"3FB0USxA","properties": {"formattedCitation":"(U.S. Department of Defense (U.S. DoD), 2008a)","plainCitation":"(U.S. Department of Defense (U.S. DoD), 2008a)","noteIndex":0},"citationItems":[{"id":1074,"uris":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"uri":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"itemData":{"id":1074,"type":"report","title":"Direct Fixed-bed Biological Perchlorate Destruction Demonstration","genre":"ESTCP Final Report (ER-0544)","author":[{"literal":"U.S. Department of Defense (U.S. DoD)"}],"issued":{"date-parts":[["2008",9,25]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. For facilities without the option of sewer disposal, a clarification and recycle process would be needed.

For fluidized bed reactors, one case study describes the use of a continuously operated separation device that uses supplied air to remove media and biomass from the top of the bed and direct it to a separation chamber. This arrangement was used in combination with an in-bed eductor to intermittently remove biomass growth from deeper within the bed. After treatment through an adsorption clarifier and multimedia filter, the study reports that the remaining residuals were "dilute enough that no special handling or pretreatment requirements should be necessary for most/all POTWs to accept" [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"Frf46rxD","properties": {"formattedCitation":"(U.S. Department of Defense (U.S. DoD), 2009)", "plainCitation":"(U.S. Department of Defense (U.S. DoD),

2009)","noteIndex":0},"citationItems":[{"id":1081,"uris":["http://zotero.org/groups/945096/items/9FHLVTXY"],"uri":["http://zotero.org/groups/945096/items/9FHLVTXY"],"itemData":{"id": 1081,"type":"report","title":"Demonstration of a Full-Scale Fluidized Bed Bioreactor for the Treatment of Perchlorate at Low Concentrations in Groundwater","genre":"Environmental Security Technology Certification Program (ESTCP) Final Report (ER-0543)","author":[{"family":"U.S. Department of Defense (U.S. DoD)","given":""}],"issued":{"date-parts":[["2009"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Downstream polishing through filtration, when used as post-treatment (see Question [ REF Ref525293735 \r \h ]), can also generate residual wastes in the form of backwash water and separated solids. The authors of the Harding ESE [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "Mbll2ueB", "properties": {"formattedCitation": "(2001)", "plainCitation": "(2001)", " noteIndex":0}, "citationItems":[{"id":1139, "uris":["http://zotero.org/groups/945096/items/ZPGX UZPL"], "uri": ["http://zotero.org/groups/945096/items/ZPGXUZPL"], "itemData": {"id":1139, "ty pe":"report", "title": "Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author": [{"family": "Harding Engineering and Environmental Services (ESE)", "given": ""}], "issued": {"date-parts": [["2001"]]}}, "suppressauthor":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/cslcitation.json"} ] report suggest that clarifier solids could be discharged directly to sewer or filter pressed to reduce volume prior to ultimate disposal. The full-scale drinking water treatment facility in Rialto uses dissolved air floatation, followed by a sludge press, to treat backwash from post-treatment filtration [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"5Mxr8mDs","properties": {"formattedCitation":"(Webster & Litchfield, 2017)", "plainCitation": "(Webster & Litchfield, 2017)","noteIndex":0},"citationItems":[{"id":1012,"uris":["http://zotero.org/groups/945096/item s/64HZKA2M"], "uri": ["http://zotero.org/groups/945096/items/64HZKA2M"], "itemData": {"id": 1012, "type": "article-journal", "title": "Full-scale biological treatment of nitrate and perchlorate for potable water production", "container-title": "Journal AWWA", "page": "30-40","volume":"109","issue":"5","author":[{"family":"Webster","given":"T.D."},{"family":"Litch field", "given": "M.H." ], "issued": {"dateparts":[["2017"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Backwash water from downstream polishing would be expected to have characteristics similar to water from direct backwash of a fixed bed reactor.

## 3.2.3 Can the bench or pilot studies be scaled up to represent full-scale treatment, including residuals generation and handling?

Yes. Given the experience with full-scale remediation projects, the bench studies, pilot studies, and temporary field tests generally provide sufficient data to represent full-scale drinking water treatment.

#### 3.2.4 Is additional research needed?

No. Additional research is not required.

### 3.3 General Geographic Applicability for Biological Treatment

### 3.3.1 What regions do the existing research areas represent?

The studies of biological treatment of perchlorate have been conducted in California and Texas.

## 3.3.2 Is it known that regional water quality variations will limit treatment effectiveness or reliability in some areas?

As discussed above (see Question [ REF \_Ref525292787 \r \h ]), water temperature is a critical variable in the ability of biological treatment to continuously destroy perchlorate. Regions not studied may be more likely than California and Texas to have cold water and/or seasonally variable water temperature. Because the effect of temperature is well understood, however, it should be feasible to determine whether biological treatment will be effective for a given system based on a water temperature record.

In addition to external electron donors, bacteria in bioreactors require macro- and micro-nutrients in order to grow and effectively reduce perchlorate. Thus, concentrations of these nutrients in the raw water are a consideration in bioreactor effectiveness. Macro-nutrients include phosphorous and nitrogen, and necessary micro-nutrients include sulfur and iron. While source water typically contains sufficient micro-nutrients, it sometimes has insufficient amounts of phosphorous and nitrogen to allow for bacterial growth. As a result, some full-scale designs have required supplemental addition of one or both of these nutrients [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "NDoHjLOr", "properties": {"formattedCitation": "(Harding Engineering and Environmental Services (ESE), 2001; U.S. Department of Defense (U.S. DoD), 2008a, 2009)", "plainCitation": "(Harding Engineering and Environmental Services (ESE), 2001; U.S. Department of Defense (U.S. DoD), 2008a, 2009)","noteIndex":0},"citationItems":[{"id":1139,"uris":["http://zotero.org/groups/945096/item s/ZPGXUZPL"], "uri": ["http://zotero.org/groups/945096/items/ZPGXUZPL"], "itemData": {"id": 1 139, "type": "report", "title": "Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author": [{"family": "Harding Engineering and Environmental Services (ESE)","given":""}],"issued":{"dateparts":[["2001"]]}}},{"id":1074,"uris":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"ur i":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"itemData":{"id":1074,"type":"report"," title": "Direct Fixed-bed Biological Perchlorate Destruction Demonstration", "genre": "ESTCP Final Report (ER-0544)", "author": [{"literal": "U.S. Department of Defense (U.S. DoD)"}],"issued":{"dateparts":[["2008",9,25]]}}},{"id":1081,"uris":["http://zotero.org/groups/945096/items/9FHLVTX Y"],"uri":["http://zotero.org/groups/945096/items/9FHLVTXY"],"itemData":{"id":1081,"type":" report", "title": "Demonstration of a Full-Scale Fluidized Bed Bioreactor for the Treatment of Perchlorate at Low Concentrations in Groundwater", "genre": "Environmental Security Technology Certification Program (ESTCP) Final Report (ER-0543)", "author": [{"family": "U.S. Department of Defense (U.S. DoD)", "given": ""}], "issued": {"dateparts":[["2009"]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ].

## 3.3.3 Are there any regional issues with respect to residuals handling or water resource use?

No. Regional residual handling and water resource needs are not expected to affect technology feasibility.

### 3.3.4 Is additional research needed?

No. Additional research is not required.

# 3.4 Compatibility of Biological Treatment with Other Treatment Processes

# 3.4.1 Have the effects (adverse or beneficial) of the treatment process on other processes likely to be present at existing plants been evaluated?

Yes. Biological treatment results in the production of soluble microbial organic products that become part of the treated water. The additional microorganisms increase disinfection demand for the downstream treatment processes. The biological treatment process also depletes the levels of oxygen in the treated water and can add turbidity and sulfides, which can have adverse effects on downstream treatment processes if not managed through post-treatment. Beneficial effects of biological treatment, on the other hand, include the potential to remove nitrate and disinfection byproduct precursors.

## 3.4.2 Will additional pre- or post-treatment be required for integration into an existing (or planned) treatment train?

Yes. Post-treatment will be needed to control the effects on other treatment processes (and also on the distribution system; see Question [ REF Ref525293735 \r \h ]). In the field study of biological treatment for potable water [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "2lg17wWj", "properties": {"formattedCitation": "(Gilbert et al., 2001; Harding Engineering and Environmental Services (ESE), 2001)", "plainCitation": "(Gilbert et al., 2001); Harding Engineering and Environmental Services (ESE), 2001)","noteIndex":0},"citationItems":[{"id":1149,"uris":["http://zotero.org/groups/945096/item s/NHNGXTZ8"],"uri":["http://zotero.org/groups/945096/items/NHNGXTZ8"],"itemData":{"id": 1149, "type": "report", "title": "Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California", "collection-title": "Expert Panel Final Report prepared for Aerojet", "author": [{"family": "Gilbert", "given": "J.B."}, {"family": "Clark", "given": "R."}, {"family": "R."}, {"fam ":"Kavanaugh", "given": "M." }, { "family": "McCarty", "given": "P." }, { "family": "Trussell", "given": " R.R."}],"issued":{"dateparts":[["2001"]]}}}, {"id":1139,"uris":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"u ri":["http://zotero.org/groups/945096/items/ZPGXUZPL"],"itemData":{"id":1139,"type":"report" ,"title": "Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel Basin, California", "author": [{"family": "Harding Engineering and Environmental Services (ESE)", "given": ""}], "issued": {"date-parts": [["2001"]]}}}], "schema": "https://github.com/citationstyle-language/schema/raw/master/csl-citation.json"}], biological treatment was part of a train of seven different unit processes. The train included an aerator and multimedia filter serving as

post-treatment for the biological treatment step. Post-treatment was then followed by downstream processes to address other contaminants and water quality concerns. The downstream processes were an air stripper, advanced oxidation, granular activated carbon, and disinfection. Investigators concluded that each of the downstream treatment processes met desired removal efficiencies in a reliable manner [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"8ZXO6T2U", "properties": {"formattedCitation":"(Gilbert et al., 2001)", "plainCitation": "(Gilbert et al., 2001)", "noteIndex":0}, "citationItems":[{"id":1149, "uris":["http://zotero.org/groups/945096/item s/NHNGXTZ8"],"uri":["http://zotero.org/groups/945096/items/NHNGXTZ8"],"itemData":{"id": 1149,"type":"report","title":"Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California", "collection-title": "Expert Panel Final Report prepared for Aerojet", "author": [{"family": "Gilbert", "given": "J.B."}, {"family": "Clark", "given": "R."}, {"family": "R."}, {"fam ":"Kavanaugh", "given": "M." }, { "family": "McCarty", "given": "P." }, { "family": "Trussell", "given": " R.R." ], "issued": {"date-parts": [["2001"]]}}}], "schema": "https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. Thus, post-treatment appears able to manage the potential impacts of biological treatment on common downstream treatment processes.

### 3.4.3 Is additional research needed?

No. Additional research is not required.

# 3.5 Ability of Biological Treatment to Bring all of the Water System into Compliance

## 3.5.1 Will the treatment process adversely affect the distribution system or water resource decisions?

Yes, although distribution system impacts might be managed by post-treatment processes. As discussed under Question [ REF Ref525293762 \r \h ], biological treatment adds microorganisms, depletes oxygen, and can add turbidity and sulfides. Therefore, post-treatment will typically be required for production of drinking water. Typical post-treatment processes include [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "s298IrQb", "properties": {"formattedCitation": "(Dordelmann, 2009; Harding Engineering and Environmental Services (ESE), 2001; U.S. Department of Defense (U.S. DoD), 2008a; Webster & Crowley, 2016; Webster & Litchfield, 2017)", "plainCitation": "(Dordelmann, 2009; Harding Engineering and Environmental Services (ESE), 2001; U.S. Department of Defense (U.S. DoD), 2008a; Webster & Crowley, 2016; Webster & Litchfield, 2017)", "noteIndex":0}, "citationItems":[{"id":1164, "uris":["http://zotero.org/groups/945096/item s/Z7PC3BME"],"uri":["http://zotero.org/groups/945096/items/Z7PC3BME"],"itemData":{"id":1 164,"type": "speech", "title": "Full-Scale Biological Denitrification Plants in Germany, Austria and Poland", "publisher-place": "Seattle, WA", "event": "2009 AWWA Water Quality Technology Conference & Exposition", "event-place": "Seattle, WA", "author": [{"family": "Dordelmann", "given": "O."}], "issued": {"dateparts":[["2009",11]]}}},{"id":1139,"uris":["http://zotero.org/groups/945096/items/ZPGXUZPL"] "uri":["http://zotero.org/groups/945096/items/ZPGXUZPL"], "itemData": {"id":1139, "type": "rep ort", "title": "Final: Phase 2 Treatibility Study Report, Aerojet GET E/F Treatment Facility, Sacramento, California", "collection-title": "Prepared for U.S. Environmental Protection Agency Region IX and Baldwin Park Operable Unit Cooperating Respondents, San Gabriel

```
Basin, California", "author": [{"family": "Harding Engineering and Environmental Services
(ESE)", "given": ""}], "issued": {"date-
parts":[["2001"]]}}}, {"id":1074,"uris":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"ur
i":["http://zotero.org/groups/945096/items/2ZCNIFHT"],"itemData":{"id":1074,"type":"report","
title":"Direct Fixed-bed Biological Perchlorate Destruction Demonstration", "genre": "ESTCP
Final Report (ER-0544)", "author": [{"literal": "U.S. Department of Defense (U.S.
DoD)"}],"issued":{"date-
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"],"uri":["http://zotero.org/groups/945096/items/BI5LYMZP"],"itemData":{"id":1018,"type":"sp
eech", "title": "Biological treatment of perchlorate in groundwater.", "event": "AWWA Annual
Conference and
Exposition", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Crowley", "given": "T.J."
}],"issued":{"date-
parts":[["2016",6,21]]}}},{"id":1012,"uris":["http://zotero.org/groups/945096/items/64HZKA2
M"],"uri":["http://zotero.org/groups/945096/items/64HZKA2M"],"itemData": {"id":1012,"type":
"article-journal", "title": "Full-scale biological treatment of nitrate and perchlorate for potable
water production", "container-title": "Journal AWWA", "page": "30-
40","volume":"109","issue":"5","author":[{"family":"Webster","given":"T.D."},{"family":"Litch
field", "given": "M.H." ], "issued": {"date-
parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]:
```

- reoxygenation or aeration for saturation with oxygen, using hydrogen peroxide addition or an aeration tank
- a polishing filter (using GAC or mixed media) for removal of turbidity, sulfide, and/or dissolved organic content, possibly including coagulant addition before filtration
- disinfection via ultraviolet light or chlorination.

```
For the full-scale system supplying drinking water, the permit requirements also include
instrumentation and controls (chlorine, pH, nitrate, sulfide, total organic carbon, and turbidity) to
monitor performance [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID": "X2B7SNPa", "properties": {"formattedCitation": "(Webster & Crowley, 2016;
Webster & Litchfield, 2017)", "plainCitation": "(Webster & Crowley, 2016; Webster & Litchfield,
2017)", "noteIndex":0}, "citationItems":[{"id":1018, "uris":["http://zotero.org/groups/945096/item
s/BI5LYMZP"],"uri":["http://zotero.org/groups/945096/items/BI5LYMZP"],"itemData":{"id":10
18,"type":"speech", "title": "Biological treatment of perchlorate in
groundwater.", "event": "AWWA Annual Conference and
Exposition", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Crowley", "given": "T.J."
}],"issued":{"date-
parts":[["2016",6,21]]}}},{"id":1012,"uris":["http://zotero.org/groups/945096/items/64HZKA2
M"], "uri": ["http://zotero.org/groups/945096/items/64HZKA2M"], "itemData": {"id":1012, "type":
"article-journal", "title": "Full-scale biological treatment of nitrate and perchlorate for potable
water production", "container-title": "Journal AWWA", "page": "30-
40","volume":"109","issue":"5","author":[{"family":"Webster","given":"T.D."},{"family":"Litch
field", "given": "M.H." ], "issued": {"date-
parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"}].
```

In the field study of biological treatment for potable water, which included the post-treatment processes listed above, investigators concluded that bacterial re-growth in the water distribution system would not be significant [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"eBy7BuRM","properties":{"formattedCitation":"(Gilbert et al., 2001)","plainCitation":"(Gilbert et al., 2001)","noteIndex":0},"citationItems":[{"id":1149,"uris":["http://zotero.org/groups/945096/items/NHNGXTZ8"],"uri":["http://zotero.org/groups/945096/items/NHNGXTZ8"],"itemData":{"id": 1149,"type":"report","title":"Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California","collection-title":"Expert Panel Final Report prepared for Aerojet","author":[{"family":"Gilbert","given":"J.B."},{"family":"Clark","given":"R."},{"family":"Kavanaugh","given":"M."},{"family":"McCarty","given":"P."},{"family":"Trussell","given":"R.R."}],"issued":{"date-parts":[["2001"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}]. They did not, however, address the potential for other distribution systems impacts.

## 3.5.2 Might the treatment process, residuals handling, or pre- or post-treatment requirements raise new environmental quality concerns?

Yes. Any of the impacts discussed above under Questions [ REF \_Ref525293762 \r \h ] and [ REF \_Ref525293735 \r \h ], if not adequately managed through post-treatment, could create new environmental quality concerns.

### 3.5.3 Is additional research needed?

Although Gilbert et al. [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"olfre7L8","properties": {"formattedCitation":"(2001)","plainCitation":"(2001)","n oteIndex":0},"citationItems":[{"id":1149,"uris":["http://zotero.org/groups/945096/items/NHNG XTZ8"],"uri":["http://zotero.org/groups/945096/items/NHNGXTZ8"],"itemData": {"id":1149,"ty pe":"report","title":"Review of Phase 2 Treatibility Study: Aerojet Facility, Rancho Cordova, California","collection-title":"Expert Panel Final Report prepared for Aerojet","author":[{"family":"Gilbert","given":"J.B."},{"family":"Clark","given":"R."},{"family":"Kavanaugh","given":"M."},{"family":"McCarty","given":"P."},{"family":"Trussell","given":"R.R."}],"issued":{"date-parts":[["2001"]]}},"suppressauthor":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] concluded that bacterial re-growth in the water distribution system would not be significant after post-treatment, additional research might be needed on the effectiveness of post-treatment processes in mitigating other distribution system impacts.

# 3.6 Reasonable Cost Basis for Biological Treatment for Large and Medium Systems

## 3.6.1 Is the technology currently used by medium and large systems (including uses for other treatment purposes)?

Yes. The full-scale system suppling drinking water was initially designed to treat 3 million gallons per day (MGD), with an ultimate capacity of 6 MGD so that water from additional wells might be treated in the future [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"VIVeF6LF","properties": {"formattedCitation":"(Webster & Crowley, 2016; Webster & Litchfield, 2017)", "plainCitation":"(Webster & Crowley, 2016; Webster & Litchfield,

```
2017)", "noteIndex":0}, "citationItems":[{"id":1018, "uris":["http://zotero.org/groups/945096/item
s/BI5LYMZP"], "uri": ["http://zotero.org/groups/945096/items/BI5LYMZP"], "itemData": {"id":10
18,"type": "speech", "title": "Biological treatment of perchlorate in
groundwater.", "event": "AWWA Annual Conference and
Exposition", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Crowley", "given": "T.J."
}]."issued":{"date-
parts":[["2016",6,21]]}}}, {"id":1012,"uris":["http://zotero.org/groups/945096/items/64HZKA2
M"],"uri":["http://zotero.org/groups/945096/items/64HZKA2M"],"itemData": {"id":1012,"type":
"article-journal", "title": "Full-scale biological treatment of nitrate and perchlorate for potable
water production", "container-title": "Journal AWWA", "page": "30-
40", "volume": "109", "issue": "5", "author": [{"family": "Webster", "given": "T.D."}, {"family": "Litch
field", "given": "M.H." ], "issued": {"date-
parts":[["2017"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. Fluidized bed reactors also have been used in
remedial applications with design flows up to 10 MGD [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID":"sGC6KL1a","properties": {"formattedCitation":"(Greene &
Pitre, 2000)", "plainCitation": "(Greene & Pitre,
2000)", "noteIndex":0}, "citationItems":[{"id":1146, "uris":["http://zotero.org/groups/945096/item
s/LS46AD5H"],"uri":["http://zotero.org/groups/945096/items/LS46AD5H"],"itemData": {"id":11
46,"type":"chapter","title":"Treatment of Groundwater Containing Perchlorate using Biological
Fluidized Bed Reactors with GAC or Sand Media", "container-title": "Perchlorate in the
Environment", "publisher": "Kluwer Academic/Plenum", "publisher-place": "New York,
NY", "event-place": "New York,
NY", "author": [{"family": "Greene", "given": "M.R."}, {"family": "Pitre", "given": "M.P."}], "editor":
[{"family":"Urbansky", "given": "E.T."}], "issued": {"date-
parts":[["2000"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. This application re-injects treated water into
an underlying aquifer; water is not used for drinking water.
```

## 3.6.2 Do the treatment studies provide sufficient information on design assumptions to allow cost modeling?

Detailed data are available from the treatment studies for all of the relevant design parameters, including:

- Support media type
- EBCT
- Bed expansion (for fluidized bed reactors)
- Electron donor type and dosage
- Nutrient addition
- Backwash design (for fixed bed reactors)
- Recycle rate (for fluidized bed reactors).

### 3.6.3 Is additional research needed?

No. Additional research is not required.

# 4 Best Available Technology Evaluation for Reverse Osmosis

Membrane filtration processes physically remove perchlorate ions from drinking water. These processes separate a solute such as perchlorate ions from a solution by forcing the solvent to flow through a membrane at a pressure greater than the normal osmotic pressure. The membrane is semi-permeable, transporting different molecular species at different rates. Water and low-molecular weight solutes pass through the membrane and are removed as permeate, or filtrate. Dissolved and suspended solids are rejected by the membrane and are removed as concentrate, or reject. This technique does not destroy the perchlorate ion and, therefore, creates a subsequent need for disposal or treatment of perchlorate-contaminated waste (the concentrate).

Membranes may remove ions from feed water by a sieving action (called steric exclusion), or by electrostatic repulsion of ions from the charged membrane surface. Membrane filtration technologies evaluated for perchlorate treatment include RO, nanofiltration (NF), and ultrafiltration (UF). As discussed under Question [ REF \_Ref525557229 \r \h ], bench studies of NF and UF membranes show significant variability in these membranes' ability to remove perchlorate, depending on other constituents of the source water. Therefore, RO is the membrane process most suited for evaluation as BAT for perchlorate.

### 4.1 High Removal Efficiency for Reverse Osmosis

### 4.1.1 Have high removal efficiencies that achieve potential MCLs been documented?

Yes. Pilot-scale treatability work at the Metropolitan Water District of Southern California showed that NF and RO membranes consistently removed greater than 80 percent of the perchlorate (up to 98 percent for RO and 92 percent for NF) depending on influent concentration [ ADDIN ZOTERO\_ITEM CSL\_CITATION

{"citationID":"9aji4xxe","properties": {"formattedCitation":"(Liang, Scott, Palencia, & Bruno, 1998)","plainCitation":"(Liang, Scott, Palencia, & Bruno,

1998)","noteIndex":0},"citationItems":[{"id":1015,"uris":["http://zotero.org/groups/945096/item s/IQVVPD73"],"uri":["http://zotero.org/groups/945096/items/IQVVPD73"],"itemData":{"id":10 15,"type":"paper-conference","title":"Investigation of Treatment Options for Perchlorate Removal.","publisher":"La Verne, CA: Metropolitan Water District of Southern California","publisher-place":"San Diego, CA","event":"AWWA Water Quality Technology Conference","event-place":"San Diego,

CA","author":[{"family":"Liang","given":"S."},{"family":"Scott","given":"K.N."},{"family":"Palencia","given":"L.S."},{"family":"Bruno","given":"J."}],"issued":{"date-parts":[["1998"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. Recycling 50 percent of the reject had no effect on overall perchlorate rejection. [ REF\_Ref297714356 \h \\* MERGEFORMAT ] summarizes effectiveness results for this pilot-scale work, along with results from additional, smaller scale bench studies.

Bench-scale studies show the effects of steric/size exclusion and electrostatic exclusion on perchlorate transport through membranes to varying degrees. RO, while removing perchlorate,

also removes most other salts, requires high operating pressures, and is prone to significant flux decline. Membrane processes that operate at lower pressures, such as NF or UF, may be effective for perchlorate removal through selectivity based on size and/or charge. However, bench studies show significant variability in these membranes' ability to remove perchlorate, depending on other constituents of the source water. One bench study modified commercial NF membranes using layer-by-layer surface deposition of polyelectrolytes. This study showed that the modified NF membranes could achieve perchlorate removal nearly equal to that of RO membranes. The study, however, did not examine the effect of differing source water quality on the membranes and research on the modified membranes does not yet appear to have progressed beyond the lab [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"RxMxvldJ","properties":{"formattedCitation":"(Sanyal, Sommerfeld, & Lee, 2015)","plainCitation":"(Sanyal, Sommerfeld, & Lee, 2015)","noteIndex":0},"citationItems":[{"id":1014,"uris":["http://zotero.org/groups/945096/items/VTUDRPDL"],"itemData":{"id":1014,"type":"article-journal","title":"Design of ultrathin nanostructured polyelectrolyte-based

membranes with higher perchlorate rejection and high permeability", "container-title": "Separation and Purification Technology", "volume": "145", "issue": "113-

119", "author": [{"family": "Sanyal", "given": "O."}, {"family": "Sommerfeld", "given": "A.N."}, {"family": "Lee", "given": "I."}], "issued": {"date-

parts":[["2015"]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}].

Exhibit [ SEQ Exhibit \\* ARABIC ]. Perchlorate Effectiveness Results for Membranes

		Raw Water	Location and Source	
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
RO and NF [ ADDIN	RO up to 98%	20 to 2,000 µg/L	Metropolitan Water	Pilot study
ZOTERO_ITEM	NF up to 92%	(some trials used	District of Southern	(12 gpm)
CSL_CITATION		perchlorate-spiked	California, La Verne	
{"citationID":"Jkd1SU74		source water)	Treatment Plant, CA;	
","properties":{"formatte			Pretreated Colorado	
dCitation":"(Liang et al.,			River Water	
1998)","plainCitation":"(				
Liang et al.,				
1998)","noteIndex":0},"c				
itationItems":[{"id":1015,				
"uris":["http://zotero.org/				
groups/945096/items/IQ				
VVPD73"],"uri":["http://z				
otero.org/groups/94509				
6/items/IQVVPD73"], "ite				
mData":{"id":1015,"type				
":"paper-				
conference", "title": "Inve				
stigation of Treatment				
Options for Perchlorate				
Removal.","publisher":"				
La Verne, CA:				
Metropolitan Water				
District of Southern				
California", "publisher-				

Toohnology/Course	Domoval Efficiency	Raw Water	Location and Source Water	Study Soalo
Technology/Source place":"San Diego,	Removal Efficiency	Concentration	vvater	Study Scale
CA","event":"AWWA				
Water Quality				
Technology				
Conference", "event-				
place":"San Diego,				
CA","author":[{"family":"				
Liang", "given": "S."}, {"fa				
mily":"Scott","given":"K.				
N."},{"family":"Palencia",				
"given":"L.S."},{"family":				
"Bruno","given":"J."}],"is				
sued":{"date-				
parts":[["1998"]]}}],"sch				
ema":"https://github.co				
m/citation-style-				
language/schema/raw/				
master/csl-				
citation.json"} ]				
Surfactant modified UF	Up to 80%	100 μg/L	Synthetic water and a	Bench study
[ ADDIN		(perchlorate-spiked)	blend of Colorado	(225 milliliters per
ZOTERO_ITEM			River Water and State	minute)
CSL_CITATION			Project Water from	
{"citationID":"7Od8XAK			the Metropolitan	
N", "properties": {"formatt			Water District, CA	
edCitation":"(Jaekyung				
Yoon et al.,				
2003)","plainCitation":"(				
Jaekyung Yoon et al.,				
2003)","noteIndex":0},"c				
itationItems":[{"id":1085,				
"uris":["http://zotero.org/ groups/945096/items/P				
F2DJFVN"],"uri":["http://				
zotero.org/groups/9450				
96/items/PF2DJFVN"],"i				
temData":{"id":1085,"typ				
e":"article-				
journal","title":"Use of				
surfactant modified				
ultrafiltration for				
perchlorate (Cl(O)(4-))				
removal","container-				
title":"Water				
Research","page":"2001				
-				
2012","volume":"37","iss				
ue":"9","source":"PubM				
ed","abstract":"Determin				
ations of perchlorate				
anion (CIO(4)(-))				
transport and rejection				
were performed using a				
surfactant modified				

		Raw Water	Location and Source	
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
ultrafiltration (UF)				
membrane. Perchlorate				
anion (at a				
concentration of 100				
microg/L of CIO(4)(-),				
spiked with KClO(4))				
was introduced to the				
membrane as a pure				
component, in binary				
mixtures with other				
salts, cationic and				
anionic surfactants, and				
at various ionic strength				
conditions				
(conductivity). Also, a				
natural source water				
was spiked with				
perchlorate in the				
presence of cationic				
and anionic surfactants				
and used to determine				
the effects of a complex				
mixture (including				
natural organic matter				
(NOM)) on the				
observed rejection. All				
filtration measurements				
were performed at				
approximately the same				
permeate flow rate in				
order to minimize				
artifacts from mass				
transfer at the				
membrane interface.				
The objective of this study was to modify a				
negatively charged UF				
membrane in terms of the fundamental				
mechanisms, steric/size exclusion and				
electrostatic exclusion				
and to enhance				
perchlorate rejection,				
with synthetic water and				
a blend of Colorado River water and State				
Project water				
(CRW/SPW). Previous				
work suggested that				
perchlorate was				
dominantly rejected by				
electrostatic exclusion				
for charged				

		Dani Matan	1	1
Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
nanofiltration (NF) and	rtomo van Emolomo y		114101	Staay Stars
UF membranes				
(Rejection of				
perchlorate by reverse				
osmosis, nanofiltration				
and ultrafiltration (UF)				
membranes:				
mechanism and				
modeling. Ph.D.				
dissertation, University				
of Colorado, Boulder,				
USA, 2001). In that				
research, perchlorate				
rejection capability was				
quickly lost in the				
presence of a sufficient				
amount of other ions.				
However, this study				
showed that CIO(4)(-)				
was excluded from a				
(negatively) charged UF				
membrane with pores				
large with respect to the				
size of the ion. Although				
perchlorate rejection				
capability due to				
apparent electrostatic				
force was reduced in				
the presence of a				
cationic surfactant, a				
desired amount of the				
ClO(4)(-) was excluded				
by steric exclusion. The				
steric exclusion was				
due to decreasing				
membrane pore size				
caused by the				
adsorption of the				
cationic				
surfactant.","DOI":"10.1				
016/S0043-				
1354(02)00600-				
0","ISSN":"0043-				
1354","note":"PMID:				
12691884","journalAbbr				
eviation":"Water				
Res.","language":"eng",				
"author":[{"family":"Yoon				
","given":"Jaekyung"},{"f				
amily":"Yoon","given":"Y				
eomin"},{"family":"Amy",				
"given":"Gary"},{"family"				
:"Cho","given":"Jaeweo				
n"},{"family":"Foss","giv				
11 /, tailiny . 1 055 , 91V				

Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
en":"David"},{"family":"K	-			·
im","given":"Tae-				
Hyung"}],"issued":{"date				
parts":[["2003",5]]}}],"sc				
hema":"https://github.co				
m/citation-style-				
language/schema/raw/				
master/csl-				
citation.json"} ]				
NF and UF [ ADDIN	Up to 75%	100 μg/L	Synthetic water with	Bench study
ZOTERO_ITEM		(perchlorate-spiked)	pure component	(no flow given)
CSL_CITATION			perchlorate, also	
{"citationID":"GRtQHCC			combined with other	
e","properties":{"formatt			salts	
edCitation":"(Y. Yoon,				
Amy, Cho, Her, &				
Pellegrino,				
2002)","plainCitation":"(				
Y. Yoon, Amy, Cho,				
Her, & Pellegrino,				
2002)","noteIndex":0},"c				
itationItems":[{"id":1087,				
"uris":["http://zotero.org/				
groups/945096/items/D				
GUADSQZ"],"uri":["http:				
//zotero.org/groups/945				
096/items/DGUADSQZ"				
],"itemData":{"id":1087,"				
type":"article-				
journal","title":"Transpor				
t of perchlorate (ClO4-)				
through NF and UF				
membranes","container-				
title":"Desalination","pag				
e":"11-				
17","volume":"147","issu				
e":"1","source":"Science				
Direct","abstract":"Meas				
urements of the				
rejection of perchlorate				
anion (ClO4-) were				
performed using				
nanofiltration (NF) and				
ultrafiltration (UF)				
membranes. Aqueous				
solutions of perchlorate				
(at a concentration of				
100 μg/L of ClO4- by				
"spiking" with KClO4)				
were fed to the				
membrane test				
apparatus. Solutions				
contained only				

	·		T-2	<b>j</b>
T b l	DI Esti	Raw Water	Location and Source	Otrocks October
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
perchlorate, or an				
additional salt (KCl,				
K2SO4, or CaCl2) at				
overall ionic strengths				
of 30, 60, or 115				
mS/cm, and pH				
adjusted to 4, 6, 8, or				
10. The data were				
modeled by application				
of a non-equilibrium				
thermodynamic model.				
The model has five				
parameters: the				
molecular transport				
coefficient (ω), osmotic				
pressure gradient (ΔΠ),				
molecular reflection				
coefficient (σ), the				
average bulk fluid				
interfacial concentration				
between the feed and				
permeate side (Cavg),				
and the solvent flux				
(Jv). These parameters				
were determined by				
independent				
measurements (and				
calculation with				
minimum assumptions.)				
For example, the				
molecular transport				
coefficient (ω) was				
obtained by diffusion				
cell measurements				
under varying pH and				
conductivity conditions				
— generally it				
decreased with				
increasing pH and				
increased with				
conductivity for the				
membranes in our				
study. Measured and				
predicted perchlorate				
transport was in good				
agreement. Overall, the				
results indicate that, in				
a pure component				
system, target ions (in				
this case CIO4-) can be				
excluded from				
(negatively) charged				
membranes with pores				
large with respect to the				

Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
size of the ion, but this rejection capability is quickly lost in the presence of a sufficient amount of other ions that can screen the apparent electrostatic force field. As intuitively expected, the perchlorate flux is governed by convection in large pore membranes.","DOI":"10. 1016/S0011-9164(02)00564-7","ISSN":"0011-9164","journalAbbreviati on":"[{"family":"Yoon", "giv en":"Yeomin"},{"family":"Amy", "given":"Gary"},{"family":"Pellegrino", "given ":"John"}, "given":"Ja eweon"}, {"family":"Pellegrino", "given ":"John"}], "issued":{"dat e-parts":[["2002",9,10]]}}}], "schema":"https://github.com/citation-style-language/schema/raw/master/csl-	Removal Efficiency	Concentration	Water	Study Scale
citation.json"}]  NF and UF [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"j6jDrku0"," properties":{"formattedC itation":"(Y. Yoon, Amy, Cho, & Pellegrino, 2005)","plainCitation":"( Y. Yoon, Amy, Cho, & Pellegrino, 2005)","noteIndex":0},"c itationItems":[{"id":1086, "uris":["http://zotero.org/ groups/945096/items/C 6EW7ZPT"],"uri":["http:// /zotero.org/groups/9450 96/items/C6EW7ZPT"]," itemData":{"id":1086,"ty pe":"article- journal","title":"Systemat ic Bench-Scale	NF up to 80% (natural water) or 89% (synthetic water) UF up to 5% (natural water) or 66% (synthetic water)	100 μg/L (perchlorate-spiked)	Synthetic water and Colorado River Water from the Metropolitan Water District, CA, spiked with perchlorate	Bench study (100 to 225 milliliters per minute)

<b></b>	·		·	·
		Raw Water	Location and Source	
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
Assessment of				
Perchlorate (ClO4 -)				
Rejection Mechanisms				
by Nanofiltration and				
Ultrafiltration				
Membranes","container-				
title":"Separation				
Science and				
Technology","page":"21				
05-				
2135","volume":"39","iss				
ue":"9","source":"Taylor				
and				
Francis+NEJM","abstra				
ct":"Measurements of				
the rejection of				
perchĺorate anion (CIO4				
-) have been performed				
by using two thin-film				
composite nanofiltration				
(NF) membranes and				
four ultrafiltration (UF)				
membranes. The latter				
four membranes are all				
from the same				
manufacturer and,				
ostensibly, from the				
same material family.				
These were chosen to				
systematically change				
the membranes steric				
properties, while				
keeping the same				
material chemistry,				
thus, the enthalpic				
interactions should stay				
constant. The				
perchlorate anion (at a				
concentration of 100 g/L				
of ClO4 – by "spiking"				
with KClO4) was				
presented to the				
membrane as a pure				
component, in binary				
mixtures with other				
salts, and at varying pH				
and ionic strength				
(conductivity). Also, a				
natural source water				
was "spiked" with				
perchlorate anion and				
used to document the				
I				
effects of a complex				
mixture, including			L	

Technology/Source natural organic matter on the observed rejection. All filtration measurements were performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case CIO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to opyright in the United States "700"*10.1081/ SS. States "700"*10.1081/ SS. States "700"*10.1081/ SS. Tyffamily" "Yeon", "(given" "Yeonin ")", "family" "Yeon", "(given" "Jeony"), "(giv	<del></del>	·		<b>r</b>	<b></b>
Inatural organic matter, on the observed rejection. All filtration measurements were performed at approximately the same permanded from the total mass transfer at the membrane interface. In general, the results indicate that in a pure component system, target ions (in this case CIC4 -) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States,"DOI."10.1081/SS. 12003304."(ISSN."01.49. 6395.","author".[("family"."Yoon","given "Veomin ","("family"."Prom","given "Veomin ","("family"."Prom","given "Veomin ","("family"."Prom","given "Veomin ","("family"."Prom","given "Veomin ","("family"."Plellegrino"."given "Veomin ","("family"."hoo ","given "Veomin ","given "Veomin ","("family"."hoo ","given "Veomin ","("family"."hoo ","given "Veomin ","given "Veomi			Raw Water	Location and Source	
on the observed rejection. All filtration measurements were performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case CIO4 –) can be significantly excluded from like-charged membrane with pores large with respect to the size of the lon, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostate force field. #Contribution of the US Government, not subject to copyright in the United States", TOO!"*10.1081/ SS-12003304", "ISSN": "01 49-6395", *author", *["family", "Amy", "given", "aweem", "("family", "Amy", "given", "aweem"), "["family", "Amy", "given", "aweem", "("family", "Amy", "given", "aweem"), "["family", "Amy", "given", "aweem", "("family", "Amy", "given", "aweem", "("family", "Bellegrino", "given", "aweement, "aweement, "aweement, "aweement, "aweement, "aweement, "aweement, "aweement		Removal Efficiency	Concentration	Water	Study Scale
rejection All filtration measurements were performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface, in general, the results indicate that, in a pure component system, target lons (in this case CIO4 -) can be significantly excluded from like-changed membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"**10.1081/ SS- 120039304*, "ISSN":01 49- 6395*," "author"**[("family", "Yoor", "given", "Yoori", "given", "Yoori", "given", "Yoori", "given", "Yoori", "given", "Sewoon", "fa milly*" "Pelleginto", "given ""cary", "given ""sewoon", "fa milly*" "Polleginto", "given ""cary", "given ""sewoon", "fa milly*" "Polleginto", "given ""sewoon", "given "sewoon", "given ""sewoon", "given "sewoon", "given "sewoon", "gi					
measurements were performed at a approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target lons (in this case CIO4 –) can be significantly excluded from like-oharged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field.   #COntribution of the US Government, not subject to copyright in the United States", "DOI"-"10.1081/ SS-12009394", "ISSN"-"D1 49-6395", author" ("family" "Pon", "given". "Yeomin", "family" "Amy", "given "", "seweon"), "family", "Pelleginto", "given "", "seweon"), "family", "Pelleginto", "given "", "seweon"), "family", "Pelleginto", "given "", seweon"), "family", "Pelleginto", "given "", seweon", "family", "Pelleginto", "given ", seweon", "family", "Pelleginto", given ", seweon", "family", "Pelleginto", "given ", seweon", "family", "Pelleginto", "given ", seweon", "family", "Pelleginto"					
performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface, in general, the results indicate that, in a pure component system, target lons (in this case CIO4 -) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other lons that can screen the electrostatic force field. ##Contribution of the US Government, not subject to copyright in the United States', "POIC**10.1081/SS-12003304*,"ISSN****014_9- 6395*,"author*;["family"*-Yoon" given**","Yoon" given**","Yoon" given**","Yoon", given**","Sswed*;"("fat e- grafts*;["2005*,1.2]]]}]];", schema*","https://github. com/citation.style- language/schema/raw/ masterics!- citation.json*)]  RO and NF [ADDIN RO up to 95% NF up to 70% (perchlorate-spiked)  CSL_CITATION (CitationD***)CA7Zkc C**,"Properfies*;"(Tormat)  C', "properfies*;"(Tormat)  Properties*;"(Tormat)  RO and NF [ADDIN RO up to 95% (perchlorate-spiked)  The Castaic Lake Water Agency, CA (perchlorate-spiked)					
approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface, in general, the results indicate that, in a pure component system, target lons (in this case CIO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the lon, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI". "10.1081/SS-Government, not subject to copyright in the United States", "DOI". "10.1081/SS-Tauthor". "["family". "Yoon", "given". "Yeomin "], "["family". "Amy", "given ""."Cary); "["family". "Non", "given". "Seweon", "family". "Polongino", "given ""."John")], "issued": "["dat e-parts": "["2005*1,12]]])]); schema". "https://github.com/citation.style-language/schema/raw/masterics! citation.json".]  RO and NF [ADDIN RO up to 95% NF up to 70% (perchlorate-spiked)  For ground waters from the Castaic Lake Water Agency, CA (no flow given)					
permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case CIO4 -) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States" PODI"-10.1081/SS-120039304", "ISSNI"-101 49-6395", "author": ["family" "Yoon", "given"-"Yeomin "], "family", "Piellegrino", "given "", "Garry", "family", "Piellegrino", "given "", "onn", "given", "giv	1 '				
minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target lons (in this case CIO4 -) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. **Contribution of the US Government, not subject to copyright in the United States, "DOI!".10.1081/SS-120039304", "ISSN": "01 49-6395", "author". "("family" "Yoon," given". "Yeomin ",","family". "Any", "given """oary", "[family". "Cho", "given". "Jeaveon", "("at e-parts": "["2005", 1.2]])))], "schema: "https://github.com/citation-style language/schema/raw/master/cis-citation_json")]." RO and NF [ADDIN ZOTERO_ITEM CSL_CITATION ("Interface of the ion of the Castaic Lake Water Agency, CA ("on flow given) ("Interface of the castaic Lake Water Agency, CA ("on flow given) (")" "properties", "formits"."					
mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case CIO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"-*10.1081/SS-120039304", "ISSN":"01 49-6395", "author": [{"family" 'Chon", "given". "Yeomin", "family", "Thon", "given". "Yeomin", "family", "Pleligrino", "given". "John"), "issued", "dat eparts", "[[2005*1, 2]]]))], "schema", "https://github.com/citation-style-language/schema/raw/master/csl-citation_json")]  RO and NF [ADDIN ZOTERO_ITEM CSL_CITATION] *COTERO_ITEM CSL_CITATION ("citation!D": TOSA7ZKC C", "properties", "formatt	1 '				
membrane interface. In general, the results in dicate that, in a pure component system, target ions (in this case CIO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #COntribution of the US Government, not subject to copyright in the United States,"DOI:"10.1081/SS-120039304","ISSN":"01 49-6395", "author":["family":"Yoon," "given":"Yoomin ","Idamily":"Non," "given":"Jeomin ","Idamily":"Cho", "given":"Jeomin ","Idamily":"Cho", "given":"Jeomin ","Idamily":"Cho", "given":"Jeomin style language/schema/raw/ master/csl-citation.json"]:  RO and NF [ADDIN ZOTEM OITEM O					
general, the results indicate that, in a pure component system, target ions (in this case CIO4 ~) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI":*10.1081/SS-120039304", "ISSN":"01 49-6395", "author", ["family" "Yoon", "given":"Yeomin ","family":"Amy", "given ","Gary", "family":"Cho" "given":"Jaeweon", ["fa mily":"Pellegrino", "given ","John"], "issued", "[dat e-parts"]["2005*, 1.2]])}}])" scheme", "https://jaithub.com/citation-style-language/schema/raw/master/cisl-citation_json"]]  RO and NF [ADDIN NF up to 70% (perchlorate-spiked) the Castaic Lake Water Agency, CA ("), "properties", "formatt					
indicate that, in a pure component system, target ions (in this case CiO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN":"01 49-6395", "author": {"family" "Amy "given ""Yoon" "given" "John"], "family" "Amy "given """own" "javen" "Jaweom"], "family" "Thellegrino", given ""-John"], "isaued"; "fat engage/schema/raw/master/cisl-citation.json"]  RO and NF [ADDIN ZOTERO], TEM SP Up to 95% NF up to 70% (perchlorate-spiked)  NF up to 70% (perchlorate-spiked)  RO up to 95% NF up to 70% (perchlorate-spiked)  RO and NF [ADDIN ZOTERO], TEM SP Up to 70% (perchlorate-spiked)  CSL CITATION ("citation!D": "C55A7ZkC ("", "properties"; "formatt					
component system, target ions (in this case CICI4 -) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States' "POIL***IO1.0181/SS-120039304*, "ISSN***********************************					
larget ions (in this case CIO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN":"01 49-95%", "author": ["family":"Yoon", "given": "Yeomin ",", f'amily":"Amy", "given ""Gay", f'amily:""Amy", "given ""Gay", f'amily:""Amy", "given ""Gay", f'amily:""Cho" ", "given": "Jaeweon"), f'a mily:""Pellegrino", "given ""Jaeweon"), f'a mily:""Pellegrino", "given ""Jaeweon"), f'a mily:""Pellegrino", "given ""Jaeweon"), f'a mily:""Pellegrino", "given ""Jaeweon"), f'a mily:""Bellegrino", given ""Jaeweon", f'a mily:""Jaeweon", f'a mily:""					
CIOA —) can be significantly excluded from like—charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN"."01 49-6395", "author":["family"."Yoon", "given-"."Yeomin "), ("family"."Yoon", "given-"."Yeomin "), "Gary", "family"."Pellegrinod ", given "":"John"], "family"."Pellegrinod ", given "":"John"], "issued:", "dat e-parts":["2005", 1,2]]))]]," schema". "https://github.com/citation-style-language/schema/raw/master/cisl-citation_json"}]  RO and NF [ADDIN ZOTERO_ITEM NF up to 70% (perchlorate-spiked) the Castaic Lake Water Agency, CA ("or flow given) "CitationID"."505A7ZkC (", "properties";"formatt					
significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN"."01 49-6395, "author": ["family": "Yoon", "given": "Yeomin "], "[ramily": "Amy", "given ""Gary", "framily": "Amy", "given ""Gary", "framily": "Pellegrino", "given ""Jamewon"], "family": "Pellegrino", "given ""John"], "issued": "fdat e-parts": ["2005", 1,2]]))]." schema: "https://github.com/citation-style-language/schema/raw/master/cisal-citation_json".]  RO and NF [ADDIN ZOTERO   TEM SOTE   NF up to 70% (perchlorate-spiked) the Castaic Lake (no flow given) ("citation D": "205A7ZkC ("") properties"; "formatt					
from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN": "01 49-6395", "author": ["family": "Yoon", "given: "Yeomin ", "family": "Yoon", "given: "Yeomin ", "family": "Amy", "given """-Gary"}, "family": "Pelegino", "given """-John"]], "issued", "("dat e-parts": ["2005", 1.2]]])]], "schema: "Intps://github.com/citation-style-language/schema/raw/master/csl-citation_json"}]  RO and NF [ADDIN ZOTERO_ITEM NF up to 70% (perchlorate-spiked) (no flow given) ("citationID": "O5A7ZKC (") "properties": "fformatt					
membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304","ISSN":"01 49-6395", "author": [("family": "Feomin "), "("family": "Amy", "given ":"Gary"), ("family": "Cho" ", "given": "Jaeweon"), ["family": "Pellegrino", "given ":"John")], "issued": ("dat e-parts": [["2005", 1,2]]}}], "schema": "https://github.com/citation.style-language/schema/raw/master/csl-citation.json")] RO and NF [ADDIN ZOTERO_ITEM SON WITH DATE of the Castaic Lake Water Agency, CA ("citationID":"05A7ZKC (") "properties": "["formatt"]					
large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. ## Contribution of the US Government, not subject to copyright in the United States", "DOI": "10.1081/SS-120039304", "ISSN": "01 49-6395", "author". ["family": "Yoon", "given": "Yeomin ",","("amily": "Amy", "given "","Gamy", "family": "Cho", "given": "Jeaweon"), "["family": "John"], "["saweon", "["family": "John"], "["sisued": ["dat e-parts": ["2005", 1.2]]}}], "schema": "https://github.com/citation.style-language/schema/raw/master/csl-citation_json")   RO and NF [ADDIN ZOTENO_ITEM NF up to 70% (perchlorate-spiked)   Ground waters from the Castaic Lake Water Agency, CA (no flow given)   CitationID": "05A7ZkC ("", "properties": "fformatt	_				
size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI"."10.1081/SS-120039304", "ISSN":"01 49-6395", "author": [{"family" :"Yoon", "given": "Yeomin ",", ("family":"Amy", "given ":"Gary"), {"family":"Cho" ","given": "Javewoon"), ("family":"Pellegrino", "given ":"Javewoon"), ("family":"Pellegrino", "given ":"Johnn"), "issued": ("dat e-parts": [["2005", 1,2]]))}]," schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json")}] RO and NF [ADDIN ZOTERO_ITEM CSL_CITATION (perchlorate-spiked) (perchlorate-spiked) (no flow given) ("citationID":"OSA7ZkC ("properties": ("formatt					
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Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
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High-Pressure				
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Stream Treatment by				
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RO [ ADDIN	From 125–2,000 µg/L	125 to 2,000 µg/L	Unspecified	Bench study
ZOTERO_ITEM	to 5–80 µg/L		perchlorate-	(no flow given)
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Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
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RO and NF [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"v3oKqGrj", "properties":{"formatted Citation":"(J. Yoon, Yoon, Amy, & Her, 2005)", "plainCitation":"(J. Yoon, Yoon, Yoon, Amy, & Her, 2005)", "noteIndex":0},"c itationItems":[{"id":1011, "uris":["http://zotero.org/groups/945096/items/IIJ W6E8Q"], "uri":["http://zotero.org/groups/945096/items/IIJW6E8Q"], "item Data":{"id":1011, "type":" articlejournal", "title":"Determin ation of perchlorate rejection and associated inorganic fouling (scaling) for reverse osmosis and nanofiltration membranes under various operating conditions", "containertitle":"Journal of Environmental Engineering", "page":"72 6-733", "author":[{"family":"	RO up to 95% NF up to 55%	100 μg/L (perchlorate-spiked)	Blend of Colorado River Water and State Project Water from the Metropolitan Water District, CA, spiked with perchlorate	Bench study (20 milliliters per minute)

		Raw Water	Location and Source	
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
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Y."]],"issued":{"date- parts":[["2005"]]}}],"sch ema":"https://github.co m/citation-style- language/schema/raw/				

Tashmala (***/O · · · · ·	Damayal Feet stars	Raw Water	Location and Source	Chade Cond
Technology/Source	Removal Efficiency	Concentration	Water	Study Scale
master/csl-				
citation.json"} ]				
RO, NF, and surface	RO up to 95.8%	10,000 µg/L	Perchlorate-spiked	Bench study (0.26
modified NF [ ADDIN	NF up to 70.1%	(perchlorate-spiked)	deionized water	gpm)
ZOTERO_ITEM	Surface modified NF			
CSL_CITATION	up to 93%			
{"citationID":"Ta38LMgn				
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al.,				
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of ultrathin				
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membranes with higher				
perchlorate rejection				
and high				
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## 4.1.2 Are the effects of water quality parameters on treatment effectiveness and reliability well-known?

Yes. In general, water quality affects the design (e.g., concentrate volume, cleaning frequency, antiscalant selection) of an RO system, but not removal efficiency. The literature specifically for perchlorate removal by membranes supports this conclusion. Higher variability in perchlorate removal with water quality has been found for NF and UF membranes than for RO membranes.

High levels of alkaline earth cations (Ca<sup>2+</sup> or Mg<sup>2+</sup>) can cause membrane scaling (Yoon et al., 2003), leading to a decline in product water flux. One study showed that calcium carbonate scaling was also associated with a decline in perchlorate rejection, likely because the scale reduced the surface charge of the membrane [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"OHeHUKrz", "properties": {"formattedCitation":"(J. Yoon, Amy, et al., 2005)", "plainCitation": "(J. Yoon, Amy, et al., 2005)", "noteIndex":0}, "citationItems":[{"id":1010, "uris":["http://zotero.org/groups/945096/item s/HPHVBSWB"], "uri": ["http://zotero.org/groups/945096/items/HPHVBSWB"], "itemData": {"id ":1010,"type":"article-journal","title":"Transport of target anions, chromate (Cr (VI)), arsenate (As (V)), and perchlorate (ClO4), through RO, NF, and UF membranes.", "containertitle":"Water Science and Technology", "page": "327-334", "volume": "51", "issue": "6-7","author":[{"family":"Yoon","given":"J."},{"family":"Amy","given":"G."},{"family":"Yoon", "given":"Y."}], "issued": {"date-parts":[["2005"]]}}}], "schema": "https://github.com/citationstyle-language/schema/raw/master/csl-citation.json"} ]. Other substances, such as silica and microbial biomass, may also cause flux decline; however, there are no studies of the resulting effect on perchlorate rejection.

Membrane fouling can be reduced either by reducing the pH of the feed water or by adding an antiscalant chemical. However, for membranes that reject perchlorate electrostatically (primarily NF and UF membranes), studies of several synthetic waters show that a reduced feed pH reduces the rejection of perchlorate [ ADDIN ZOTERO ITEM CSL CITATION {"citationID": "wHL9VrcI", "properties": {"formattedCitation": "(J. Yoon, Amy, et al., 2005; J. Yoon, Yoon, et al., 2005; Y. Yoon et al., 2005)", "plainCitation": "(J. Yoon, Amy, et al., 2005; J. Yoon, Yoon, et al., 2005; Y. Yoon et al., 2005)", "noteIndex":0}, "citationItems":[{"id":1010, "uris":["http://zotero.org/groups/945096/item s/HPHVBSWB"], "uri": ["http://zotero.org/groups/945096/items/HPHVBSWB"], "itemData": {"id ":1010,"type": "article-journal", "title": "Transport of target anions, chromate (Cr (VI)), arsenate (As (V)), and perchlorate (ClO4), through RO, NF, and UF membranes.", "containertitle":"Water Science and Technology", "page": "327-334", "volume": "51", "issue": "6-7","author":[{"family":"Yoon","given":"J."},{"family":"Amy","given":"G."},{"family":"Yoon", "given":"Y."}],"issued":{"dateparts":[["2005"]]}}}, {"id":1011,"uris":["http://zotero.org/groups/945096/items/IIJW6E8Q"],"uri ":["http://zotero.org/groups/945096/items/IIJW6E8Q"],"itemData":{"id":1011,"type":"articlejournal", "title": "Determination of perchlorate rejection and associated inorganic fouling (scaling) for reverse osmosis and nanofiltration membranes under various operating conditions", "container-title": "Journal of Environmental Engineering", "page": "726-733", "author": [{"family": "Yoon", "given": "J."}, {"family": "Yoon", "given": "Y."}, {"family": "Am y", "given": "G." }, {"family": "Her", "given": "N." }], "issued": {"dateparts":[["2005",5]]}}},{"id":1086,"uris":["http://zotero.org/groups/945096/items/C6EW7ZPT"],

"uri":["http://zotero.org/groups/945096/items/C6EW7ZPT"],"itemData":{"id":1086,"type":"artic le-journal", "title": "Systematic Bench-Scale Assessment of Perchlorate (ClO4 –) Rejection Mechanisms by Nanofiltration and Ultrafiltration Membranes", "container-title": "Separation Science and Technology", "page": "2105-2135", "volume": "39", "issue": "9", "source": "Taylor and Francis+NEJM", "abstract": "Measurements of the rejection of perchlorate anion (ClO4 –) have been performed by using two thin-film composite nanofiltration (NF) membranes and four ultrafiltration (UF) membranes. The latter four membranes are all from the same manufacturer and, ostensibly, from the same material family. These were chosen to systematically change the membranes steric properties, while keeping the same material chemistry, thus, the enthalpic interactions should stay constant. The perchlorate anion (at a concentration of 100 g/L of ClO4 – by "spiking" with KClO4) was presented to the membrane as a pure component, in binary mixtures with other salts, and at varying pH and ionic strength (conductivity). Also, a natural source water was "spiked" with perchlorate anion and used to document the effects of a complex mixture, including natural organic matter, on the observed rejection. All filtration measurements were performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case ClO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI": "10.1081/SS-120039304", "ISSN": "0149- $6395", "author": [\{"family": "Yoon", "given": "Yeomin"\}, \{"family": "Amy", "given": "Gary"\}, \{"family": "Amy", "given": "Gary"\}, \{"family": "Amy", "given": "Gary", "given": "$ ily":"Cho", "given": "Jaeweon" }, {"family": "Pellegrino", "given": "John" }], "issued": {"dateparts":[["2005",1,2]]}}}],"schema":"https://github.com/citation-stylelanguage/schema/raw/master/csl-citation.json"} ]. The lower pH has been shown to diminish the negative surface charge of the membranes, inhibiting the electrostatic rejection mechanism. One study [ ADDIN ZOTERO ITEM CSL CITATION {"citationID":"m52HhJol", "properties": {"formattedCitation":"(J. Yoon, Amy, et al., 2005)", "plainCitation": "(J. Yoon, Amy, et al., 2005)","noteIndex":0},"citationItems":[{"id":1010,"uris":["http://zotero.org/groups/945096/item s/HPHVBSWB"], "uri": ["http://zotero.org/groups/945096/items/HPHVBSWB"], "itemData": {"id ":1010,"type": "article-journal", "title": "Transport of target anions, chromate (Cr (VI)), arsenate (As (V)), and perchlorate (ClO4), through RO, NF, and UF membranes.", "containertitle":"Water Science and Technology","page":"327-334","volume":"51","issue":"6-7","author":[{"family":"Yoon","given":"J."},{"family":"Amy","given":"G."},{"family":"Yoon", "given":"Y."}],"issued":{"date-parts":[["2005"]]}}}],"schema":"https://github.com/citationstyle-language/schema/raw/master/csl-citation.json"} ] demonstrated that a phosphonate-based antiscalant improved both product water flux and perchlorate rejection. In these studies, perchlorate rejection by RO membranes was much less sensitive to the feed water pH.

The same studies demonstrated that a high concentration of other ions, particularly divalent cations, in the membrane feed water can reduce perchlorate rejection. Again, the studies attributed the reduced rejection to a diminished membrane surface charge. One study that included one natural water and several synthetic waters [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"2RU601Fz","properties": {"formattedCitation":"(Y. Yoon et al., 2005)","plainCitation":"(Y. Yoon et al., 2005)","noteIndex":0}, "citationItems":[{"id":1086,"uris":["http://zotero.org/groups/945096/item

s/C6EW7ZPT"],"uri":["http://zotero.org/groups/945096/items/C6EW7ZPT"],"itemData": {"id":1 086,"type":"article-journal","title":"Systematic Bench-Scale Assessment of Perchlorate (ClO4 –) Rejection Mechanisms by Nanofiltration and Ultrafiltration Membranes","container-title":"Separation Science and Technology","page":"2105-2135","volume":"39","issue":"9","source":"Taylor and

Francis+NEJM", "abstract": "Measurements of the rejection of perchlorate anion (ClO4 –) have been performed by using two thin-film composite nanofiltration (NF) membranes and four ultrafiltration (UF) membranes. The latter four membranes are all from the same manufacturer and, ostensibly, from the same material family. These were chosen to systematically change the membranes steric properties, while keeping the same material chemistry, thus, the enthalpic interactions should stay constant. The perchlorate anion (at a concentration of 100 g/L of ClO4 – by "spiking" with KClO4) was presented to the membrane as a pure component, in binary mixtures with other salts, and at varying pH and ionic strength (conductivity). Also, a natural source water was "spiked" with perchlorate anion and used to document the effects of a complex mixture, including natural organic matter, on the observed rejection. All filtration measurements were performed at approximately the same permeate flow rate to minimize artifacts from mass transfer at the membrane interface. In general, the results indicate that, in a pure component system, target ions (in this case ClO4 –) can be significantly excluded from like-charged membranes with pores large with respect to the size of the ion, but this rejection capability decreases in the presence of a sufficient amount of other ions that can screen the electrostatic force field. #Contribution of the US Government, not subject to copyright in the United States", "DOI": "10.1081/SS-120039304", "ISSN": "0149-6395", "author": [{"family": "Yoon", "given": "Yeomin"}, {"family": "Amy", "given": "Gary"}, {"family": "Gary"}, {"fam ily":"Cho", "given": "Jaeweon" }, { "family": "Pellegrino", "given": "John" } ], "issued": { "date-

ily":"Cho","given":"Jaeweon"},{"family":"Yeomin"},{"family":"Amy","given":"Gary"},{"family":"Cho","given":"Jaeweon"},{"family":"Pellegrino","given":"John"}],"issued":{"date-parts":[["2005",1,2]]}}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] found that the natural water had worse perchlorate rejection than the most similar synthetic water for NF and UF membranes.

### 4.1.3 Is the technology reliable enough to continuously meet a drinking water MCL?

Yes. In general, RO is an established, reliable technology that has been used successfully to meet other MCLs. There is nothing unique about perchlorate removal by RO that suggests using it for this contaminant would reduce the technology's reliability.

#### 4.1.4 Is additional research needed?

No. Additional research is not required.

### 4.2 History of Full-Scale Operation for Reverse Osmosis

### 4.2.1 Do existing studies include full-scale operations at drinking water treatment facilities?

There are no known full-scale RO facilities specifically for the removal of perchlorate. There are, however, a large number of drinking water treatment facilities that use RO for other contaminants.

# 4.2.2 Are there studies of full-scale treatment of residuals that fully characterize residual waste streams and disposal options?

There are no known full-scale studies of residuals from RO facilities specifically for the removal of perchlorate. In general, however, the characteristics of RO residuals are predictable and handling and treatment options are well understood. RO produces a waste stream called the concentrate (or reject). This waste stream contains all removed dissolved solids. Membrane system designs generally set a recovery rate (i.e., the ratio of permeate to feed flow) based on the scaling potential of the feed water. The presence of a particular target contaminant has little or no effect on the selected recovery rate. Typical recovery rates are 70 to 85 percent, which means that concentrate flows can account for 15 to 30 percent of influent (i.e., 100 percent minus the recovery rate). There is nothing unique about perchlorate removal by RO that suggests recovery rates and concentrate flows would be different. Therefore, it is likely that the concentrate flow from a full-scale RO facility removing perchlorate would represent a substantial share of influent flows, implying a fairly large perchlorate-contaminated waste stream for subsequent treatment or disposal.

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For disposal of RO residuals, the majority of systems use surface water discharge or discharge to
sanitary sewer, with a small number using deep well injection, evaporation ponds, or spray
irrigation [ ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"j91VNm1d", "properties": {"formattedCitation":"(U.S. Department of Interior
(DoI), 2001)", "plainCitation": "(U.S. Department of Interior (DoI),
2001)", "noteIndex":0}, "citationItems":[{"id":1080, "uris":["http://zotero.org/groups/945096/item
s/42RS957X"],"uri":["http://zotero.org/groups/945096/items/42RS957X"],"itemData":{"id":108
0,"type":"report", "title": "Membrane Concentrate Disposal: Practices and
Regulation", "collection-title": "Desalination and Water Purification Research and Development
Program Report No. 69", "publisher": "Bureau of Reclamation, Technical Service Center, Water
Treatment Engineering and Research
Group","URL":"http://www.usbr.gov/pmts/water/media/pdfs/report069.pdf","author":[{"literal":
"U.S. Department of Interior (DoI)"}],"issued":{"date-
parts":[["2001",9]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ]. The large volume of residuals is a well-known
obstacle to adoption of RO technology, in general. In the case of perchlorate removal by
centralized treatment plants, the high perchlorate concentration in the residuals might limit the
disposal options or require additional treatment prior to disposal, depending on state and local
discharge regulations. Studies of treatment of perchlorate-bearing RO residuals are limited to a
few laboratory-scale studies. These include biological [ ADDIN ZOTERO ITEM
CSL CITATION {"citationID": "S96CtKCx", "properties": {"formattedCitation": "(Giblin, Losi,
Hosangadi, & Frankenberger, 2002)", "plainCitation": "(Giblin, Losi, Hosangadi, &
Frankenberger,
2002)", "noteIndex":0}, "citationItems":[{"id":1150, "uris":["http://zotero.org/groups/945096/item
s/QMU78XKX"], "uri": ["http://zotero.org/groups/945096/items/QMU78XKX"], "itemData": {"id"
:1150,"type":"article-journal","title":"Bacterial Perchlorate Reduction in Simulated Reverse
Osmosis Rejectate", "container-title": "Bioremediation Journal", "page": "105-
112", "volume": "6", "issue": "2", "author": [{"family": "Giblin", "given": "T."}, {"family": "Losi", "giv
en":"M.E."},{"family":"Hosangadi","given":"V."},{"family":"Frankenberger","given":"W.T."}],
"issued": {"date-parts": [["2002"]]}}}], "schema": "https://github.com/citation-style-
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language/schema/raw/master/csl-citation.json"} ] and thermal treatment [ ADDIN
ZOTERO ITEM CSL CITATION
{"citationID":"H9rR7CX7","properties": {"formattedCitation":"(Applied Research Associates
(ARA), 2000)", "plainCitation": "(Applied Research Associates (ARA),
2000)","noteIndex":0},"citationItems":[{"id":995,"uris":["http://zotero.org/groups/945096/items/
I87D72M4"], "uri": ["http://zotero.org/groups/945096/items/I87D72M4"], "itemData": {"id":995, "t
ype":"report", "title": "Final Report: Hydrothermal/Thermal Decomposition of
Perchlorate", "publisher": "Investigators: L. Li, E.N. EPA Small Business Innovation Research,
EPA Grant Number 68D99032", "author": [{"literal": "Applied Research Associates
(ARA)"}],"issued":{"date-parts":[["2000"]]}}}],"schema":"https://github.com/citation-style-
language/schema/raw/master/csl-citation.json"} ] of RO concentrate. Urbansky and Schock [
ADDIN ZOTERO ITEM CSL CITATION
{"citationID":"dkrfUKm7","properties": {"formattedCitation":"(1999)", "plainCitation":"(1999)",
"noteIndex":0}, "citationItems":[{"id":1069, "uris":["http://zotero.org/groups/945096/items/K8JC
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e":"article-journal", "title": "Issues in managing the risks associated with perchlorate in drinking
water", "container-title": "Journal of Environmental Management", "page": "79-
95", "volume": "56", "issue": "2", "source": "ScienceDirect", "abstract": "Perchlorate (ClO4–)
contamination of ground and surface waters has placed drinking water supplies at risk in
communities throughout the US, especially in the West. Several major assessment studies of that
risk in terms of health and environmental impact are expected to be released by the US
Environmental Protection Agency in early 1999, and preparations for how best to manage and
minimize that risk are underway. Perchlorate salts are used in rocket and missile propulsion;
therefore, it is believed that the pollution is derived primarily from defense and supporting
industry. Due to the perchlorate anion's fundamental physical and chemical nature, the
contamination is difficult to treat or remediate. The current work describes the evolution of the
unique team-based governmental response to the problem and the rapidity of its development.
Technologies under consideration that may prove feasible for treating contaminated water
supplies are discussed and evaluated. The impact of these treatment technologies on other
regulatory compliance matters and limitations of space, cost, and other resources are considered.
Practical guidelines for approaching the problem are outlined, and current research needs are
identified.","DOI":"10.1006/jema.1999.0274","ISSN":"0301-
4797", "journal Abbreviation": "Journal of Environmental
Management", "author": [{"family": "Urbansky", "given": "E. T."}, {"family": "Schock", "given": "M.
R."\],"issued":\{"date-parts":\[\["1999",6,1\]\}\,\"suppress-
author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-
citation.json"} ] note, however, that membrane filtration point-of-use devices can be practical
options for homeowners, or other small or remote users. Depending upon the permitted
perchlorate discharge levels, the concentrate can often be disposed in the sanitary sewer system,
where it will essentially recombine with the raw water in the sewage stream.
```

# 4.2.3 Can the bench or pilot studies be scaled up to represent full-scale treatment, including residuals generation and handling?

Yes. As a mature and established technology, the scale-up of RO, in general, from bench- to pilot- to full-scale is well understood.

#### 4.2.4 Is additional research needed?

In general, additional research is not required. In cases where regional or system-specific conditions associated with perchlorate-bearing residuals management present a significant barrier, however, additional research on residuals treatment prior to disposal would be useful.

### 4.3 General Geographic Applicability for Reverse Osmosis

#### 4.3.1 What regions do the existing research areas represent?

As shown in [ REF \_Ref297714356 \h \\* MERGEFORMAT ], most of the existing pilot- and bench-scale research on RO for perchlorate removal has used water from systems in California.

# 4.3.2 Is it known that regional water quality variations will limit treatment effectiveness or reliability in some areas?

No. Although most of the existing research is for a limited region, there are no data indicating that regional water quality variations will limit effectiveness or reliability. As discussed under Question [ REF \_Ref525296057 \r \h ], water quality affects the design (e.g., concentrate volume, cleaning frequency, antiscalant selection, temperature) of an RO system, but not its effectiveness or reliability.

### 4.3.3 Are there any regional issues with respect to residuals handling or water resource use?

The large volume of water "lost" as RO residuals can be an issue in regions where water scarcity is a concern. The Small Business Advocacy Review Panel [ ADDIN ZOTERO\_ITEM CSL CITATION

{"citationID":"gdtGpwWB","properties": {"formattedCitation":"(1999)","plainCitation":"(1999)", "noteIndex":0}, "citationItems":[{"id":1032,"uris":["http://zotero.org/groups/945096/items/UPX K6DUI"],"uri":["http://zotero.org/groups/945096/items/UPXK6DUI"],"itemData": {"id":1032,"ty pe":"report","title":"Report of the Small Business Advocacy Review Panel on EPA's Planned Proposal of the National Primary Drinking Water Regulation for

Arsenic", "author": [{"literal": "Small Business Advocacy Review Panel"}], "issued": {"date-parts": [["1999"]]}}, "suppress-author": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] pointed out that a water rejection rate of 20 to 25 percent can present a problem where water is scarce, such as in the western states. The availability of discharge options for residuals is also a region- and system-specific issue, depending on location, climate, and state and local regulations. The technology is more likely to be feasible when ocean discharge or evaporation ponds are an option.

#### 4.3.4 Is additional research needed?

No. Additional research is not required.

### 4.4 Compatibility of Reverse Osmosis with Other Treatment

#### **Processes**

# 4.4.1 Have the effects (adverse or beneficial) of the treatment process on other processes likely to be present at existing plants been evaluated?

Yes. Adverse effects are unlikely. RO might have some effect on treated water chemistry (see Question [ REF \_Ref297723336 \r \h ]), which might alter corrosion control or blending requirements. Generally, however, these effluent chemistry changes should not require significant adjustments to downstream treatment processes. With regard to beneficial effects, RO membranes can remove a wide range of contaminants, including inorganic ions, total dissolved solids, nitrate, radionuclides, total organic carbon, some disinfection byproduct precursors, and synthetic organic chemicals. Since RO permeate has a reduced chlorine demand, its finished water requires a low dose of disinfectant.

# 4.4.2 Will additional pre- or post-treatment be required for integration into an existing (or planned) treatment train?

Yes. Post-treatment is typically required to control corrosion impacts (see Question [ REF Ref297723336 \r \h ]).

#### 4.4.3 Is additional research needed?

No. Additional research is not required.

# 4.5 Ability of Reverse Osmosis to Bring all of the Water System into Compliance

### 4.5.1 Will perchlorate treatment affect the distribution system or water resource decisions?

Yes. The permeate from RO filtration is essentially deionized water, and generally requires post treatment for corrosion control before it enters a distribution system [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"ESfj3Lck","properties": {"formattedCitation":"(American Water Works Assocation and American Society of Civil Engineers (AWWA/ASCE), 2005)","plainCitation":"(American Water Works Assocation and American Society of Civil Engineers (AWWA/ASCE),

2005)","noteIndex":0},"citationItems":[{"id":998,"uris":["http://zotero.org/groups/945096/items/KMLLX76Z"],"uri":["http://zotero.org/groups/945096/items/KMLLX76Z"],"itemData":{"id":998,"type":"book","title":"Water Treatment Plant Design","publisher":"AWWA","publisher-place":"Denver, CO","edition":"4th","event-place":"Denver, CO","author":[{"literal":"American Water Works Assocation and American Society of Civil Engineers (AWWA/ASCE)"}],"issued":{"date-

parts":[["2005"]]}}}],"schema":"https://github.com/citation-style-

language/schema/raw/master/csl-citation.json"} ]. In other drinking water treatment applications, the permeate is often blended with untreated water to produce a less corrosive finished water. If the source water has a sufficiently low concentration of perchlorate and other contaminants, blending may reduce post-treatment requirements. Thus, distribution system effects can be managed by adjusting corrosion control programs or blending practices.

As discussed under Question [ REF \_Ref297720638 \r \h ], the large volume of RO residuals might have an impact on water resource decisions in regions where water scarcity is a concern.

# 4.5.2 Might the treatment process, residuals handling, or pre- or post-treatment requirements raise new environmental quality concerns?

Yes. The disposal of large volumes of RO residuals could create an environmental quality concern. As discussed under Question [ REF \_Ref297720638 \r \h ], discharge concerns are region- and system-specific.

#### 4.5.3 Is additional research needed?

No. Additional research is not required.

# 4.6 Reasonable Cost Basis for Reverse Osmosis for Large and Medium Systems

### 4.6.1 Is the technology currently used by medium and large systems (including uses for other treatment purposes)?

Yes. Although there are no known full-scale RO facilities specifically for the removal of perchlorate, there are a large number of medium and large systems that use RO for other contaminants.

# 4.6.2 Do the treatment studies provide sufficient information for design assumptions?

Relevant design parameters for RO include:

- Flux rate
- Membrane type
- Membrane array configuration
- Recovery rate
- Pretreatment requirements
- Cleaning procedures.

Assumptions about these parameters for RO, in general, are determined based on major water quality parameters, such as hardness parameters, chloride, sulfate, silica, pH, silt density index, and total dissolved solids. They typically are not affected by trace contaminant influent concentrations or removal requirements. There is nothing unique about perchlorate removal by RO that suggests a different relationship between the major water quality parameters and typical design requirements.

#### 4.6.3 Is additional research needed?

No. Additional research is not required.

### 5 Summary of Best Available Technology Evaluation

[ REF \_Ref330451823 \\* MERGEFORMAT ] provides a summary of the evaluation results for the three technologies against each of the criteria. Based on this evaluation, the overall conclusions are:

- Ion exchange is a potential BAT. It has been show to achieve high removal efficiency for perchlorate, particularly with the use of perchlorate-selective resin. It is a mature and established technology in general and has been used for full-scale treatment of perchlorate at a large number of facilities. The use of disposable perchlorate-selective resins eliminates concerns about large volumes of liquid residuals in the form of spent regenerant.
- Biological treatment is a potential BAT. It has been shown to achieve high removal
  efficiency for perchlorate, including at a full-scale fluidized bed drinking water treatment
  facility. The technology may, however, need to be used in conjunction with posttreatment processes to ensure finished water quality and mitigate distribution system
  impacts. Water temperatures also may restrict the technology's applicability on a regional
  or system-specific basis.
- RO is a potential BAT. It has been shown to achieve high removal efficiency for perchlorate at bench- and pilot-scale. Although no full-scale results are available specifically for perchlorate, RO is a mature and established technology in general. Scale-up of RO systems depends primarily on major water quality parameters and is not dependent on the characteristics of trace contaminants like perchlorate. Large volumes of residual concentrate, however, will likely restrict the technology's applicability on a system-specific basis. Additional research on treatment of perchlorate-bearing RO residuals could help mitigate this issue in some cases.

# Exhibit [ SEQ Exhibit \\* ARABIC ]. Perchlorate Technologies Evaluated Against BAT Criteria

Criterion	Ion Exchange	Biological Treatment	Reverse Osmosis
1. High Removal Efficiency	······································		
1.1. Have high removal efficiencies that achieve potential MCLs been documented?	Yes	Yes	Yes
1.2. Are the effects of water quality parameters on treatment effectiveness and reliability well-known?	Yes	Yes	Yes
1.3. Is the technology reliable enough to continuously meet a drinking water MCL?	Yes	Depends on temperature	Yes
1.4. Is additional research needed?	No	No	No
2. History of Full-Scale Operation			
2.1. Do existing studies include full-scale operations at drinking water treatment facilities?	Yes	Yes (using fluidized bed reactors)	Yes (for other treatment purposes)
2.2. Are there studies of full-scale treatment of residuals that fully characterize residual waste streams and disposal options?	Yes	Yes	Yes (for other treatment purposes)
2.3. Can the bench or pilot studies be scaled up to represent full-scale treatment, including residuals generation and handling?	Yes	Yes	Yes
2.4. Is additional research needed?	No	No	Maybe
3. General Geographic Applicability			
3.1. What regions do the existing research studies represent?	Primarily California	Primarily California	Primarily California
3.2. Is it known that regional water quality variations will limit treatment effectiveness or reliability in some areas?	No	Yes	No
3.3. Are there any regional issues with respect to residuals handling or water resource use?	Not likely	No	Yes
3.4. Is additional research needed?	No	No	No
4. Compatibility with Other Treatment Processes			
4.1. Have the effects (adverse or beneficial) of the treatment process on other processes likely to be present at existing plants been evaluated?	Yes	Yes	Yes
4.2. Will additional pre- or post-treatment be required for integration into an existing (or planned) treatment train?	Possibly	Yes	Yes
4.3. Is additional research needed?	No	No	No
5. Ability to Bring All of the Water System into Compliance	'		'
5.1. Will the treatment process adversely affect the distribution system or water resource decisions?	Possibly	Yes	Yes
5.2. Might the treatment process, residuals handling, or pre- or post-treatment requirements raise new environmental quality concerns?	Not likely	Yes	Yes
5.3. Is additional research needed?	No	Maybe	No
6. Reasonable Cost Basis for Large and Medium Systems		<b>, , , , , , , , , , , , , , , , , , , </b>	,
6.1. Is the technology currently used by medium and large systems (including uses for other treatment purposes)?	Yes	Yes	Yes (for other treatment purposes)
6.2. Do the treatment studies provide sufficient information on design assumptions to allow cost modeling?	Yes	Yes	Yes
6.3. Is additional research needed?	No	No	No
			1

### 6 Small System Compliance Technology Evaluation

### 6.1 SSCT Analysis Method

A technology must be both effective and affordable to be designated an SSCT. Technologies that meet the effectiveness criterion include those designated as BATs for the proposed rule: anion exchange, biological treatment, and centralized RO. For an MCL greater than or equal to 4 micrograms per liter ( $\mu$ g/L), certified POU RO devices also meet the effectiveness criterion.<sup>3</sup>

To evaluate affordability, EPA compared incremental costs per household for each perchlorate-removal technology against an *expenditure margin*. [ REF \_Ref326056084 \h \\* MERGEFORMAT ] shows the expenditure margins for each system size category. It also shows how EPA derived the expenditure margins, beginning with estimates of MHI, which vary by system size category. The annual affordability threshold for household expenditures on drinking water is 2.5 percent of MHI. EPA deducted estimates of baseline or current water bills from the affordability threshold to obtain the expenditure margin estimates.

# Exhibit [ SEQ Exhibit \\* ARABIC ]. Expenditure Margins for SSCT Affordability Analysis

System Size (Population Served)	Median Household Income <sup>1</sup> (a)	Affordability Threshold <sup>2</sup> (b) = 2.5% x a	Baseline Water Cost <sup>3</sup> (c)	Expenditure Margin (d) = b - c
25-500	\$52,791	\$1,320	\$341	\$979
501-3,300	\$51,093	\$1,277	\$395	\$883
3,301-10,000	\$55,975	\$1,399	\$412	\$987

#### Notes:

1. MHI based on U.S. Census 2010 American Community Survey (ACS) 5-year estimates [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"fjCG5GoK","properties":{"formattedCitation":"(U.S. Census Bureau, 2010)","plainCitation":"(U.S. Census Bureau, 2010)","plainCitation")

2010)", "noteIndex":0}, "citationItems":[{"id":1017, "uris":["http://zotero.org/groups/945096/items/WJ35QNBT"], "uri":["http://zotero.org/groups/945096/items/WJ35QNBT"], "itemData":{"id":1017, "type":"article", "title":"American Community Survey, 5-year Estimates (2006-2010)", "author":[{"family":"U.S. Census Bureau", "given":""}], "issued":{"date-

parts":[["2010"]]}}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] stated in 2010 dollars, adjusted to 2017 dollars using the CPI (for all items) for areas under 50,000 persons.

2. Affordability threshold equals 2.5 percent of MHI.

3. Household water costs derived from 2006 Community Water System Survey [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"X3nlQwhM", "properties": {"formattedCitation":"(USEPA, 2009)", "plainCitation":"(USEPA, 2009)", "noteIndex":0}, "citationItems": [{"id":262, "uris": ["http://zotero.org/groups/945096/items/DZNAAV6M"], "uri": ["http://zotero.org/groups/945096/items/DZNAAV6M"], "itemData": {"id":262, "type": "article", "title": "2006 Community Water System Survey - Volume II: Detailed Tables and Survey Methodology", "URL": "https://www.epa.gov/dwstandardsregulations/community-water-system-survey", "author": [{"iteral": "USEPA"}], "issued": {"date-parts": [["2009", 5]]}, "accessed": {"date-parts": [["2018", 8, 17]]}}}, "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}], based on residential revenue per connection within each size category, adjusted to 2017 dollars based on the CPI (for all items) for areas under 50,000 persons.

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 $<sup>^3</sup>$  POU RO devices that are certified as meeting NSF/ANSI Standard 58 have demonstrated an ability to reduce perchlorate concentrations from 130  $\mu$ g/L to 4  $\mu$ g/L or less ([ HYPERLINK

<sup>&</sup>quot;http://www.nsf.org/consumer/drinking\_water/perchlorate\_reduction.asp" ]). As of August 2018, there is no standard for POU anion exchange devices.

The cost per household varies by technology and by system size category. EPA used the following method to estimate per-household costs using EPA's work breakdown structure (WBS) technology cost models:

- Estimate system-level costs for capital expenditures and annual operating and maintenance (O&M) costs
- Estimate daily design flow and average flow based on median population
  - o Estimate capital cost using a technology-specific WBS cost curve and design flow
  - o Estimate O&M costs using a WBS cost curve and average flow
- Calculate annual total costs (annualized capital expenditures plus O&M costs)
- Divide total annual costs by the median number of households served.

The WBS models generate capital costs based on equipment that can handle peak production levels or design flows. Annual costs are based on average daily flows. [ REF \_Ref326097745 \h \\* MERGEFORMAT ] shows the design and average flow estimates for the median system in each system size category. It also shows the population served by the median system and the number of households served.

### Exhibit [ SEQ Exhibit \\* ARABIC ]. Design and Average Flow Estimates and Service Estimates for the 50<sup>th</sup> Percentile or Median System

System Size (Population Served)	System Population <sup>1</sup> (a)	System Households <sup>2</sup> (b) = a/2.65	Groundwater System Design Flow <sup>3</sup> (MGD)	Groundwater System Average Flow <sup>3</sup> (MGD)	Surface Water System Design Flow <sup>3</sup> (MGD)	Surface Water System Average Flow <sup>3</sup> (MGD)
25-500	110	42	0.049	0.012	0.050	0.015
501-3,300	1,143	431	0.46	0.15	0.46	0.16
3,301-10,000	5,422	2,046	2.0	0.77	2.0	0.74

#### Notes:

- 1. Median system populations are from SDWISFED, January 2004.
- 2. Median system household estimates equal median populations divided by 2.57 persons per household (based on the 2004 U.S. Census mean).
- 3. Flow estimates are based on regression equations that relate population and design or average flows.

EPA generated costs for each system size category for 38 treatment technology scenarios. There are 12 scenarios for anion exchange comprising all combinations of two source waters (ground and surface), two resin lives (250,000 and 170,000 bed volumes), and three cost levels (low, mid, and high). There are 18 scenarios for biological treatment that are combinations of two source waters, three reactor types (fixed bed pressure vessel, fixed bed gravity basin, and fluidized bed), and three cost levels. There are 6 scenarios for RO utilized as centralized treatment to account for two source waters, and three cost levels, and two design flow ranges. Finally, there are two scenarios for POU RO to account for two source waters. Costs for POU RO do not vary by cost level input (high, mid, low). USEPA [ ADDIN ZOTERO\_ITEM CSL\_CITATION {"citationID":"2Q4yNirn","properties": {"formattedCitation":"(2018)","plainCitation":"(2018)"," noteIndex":0}, "citationItems": [{"id":246,"uris":["http://zotero.org/groups/945096/items/VUJUP N7L"],"uri":["http://zotero.org/groups/945096/items/VUJUPN7L"],"itemData": {"id":246,"type": "article","title":"Technologies and Costs for Treating Perchlorate-Contaminated Waters", "publisher":"EPA \*\*\*-\*-\*-

\*\*\*\*","author":[{"family":"USEPA","given":""}],"issued":{"date-parts":[["2018"]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] contains the cost curve parameters for all of the treatment technology scenarios. There are separate parameter sets for capital costs and O&M costs and for small, medium, and large system sizes (corresponding to design flows ranges of < 1.0 MGD,  $\geq$  1.0 MGD to < 10 MGD, and  $\geq$  10 MGD).

For each scenario, EPA estimated capital and O&M costs for the three system size categories and then calculated total annual costs. For anion exchange, biological, and central RO treatment technologies, EPA annualized capital costs at 7 percent over the expected useful life of centralized treatment equipment and added the result to O&M costs. For POU RO devices, EPA annualized capital costs (i.e., for the devices and installation) over the estimated 10-year life of the POU device. Finally, EPA divided total annual costs by the number of households served to derive per-household incremental costs for perchlorate treatment. EPA assessed affordability by comparing these values with the expenditure margins.

#### 6.2 Results

[ REF \_Ref326111577 \h \\* MERGEFORMAT ] provides ranges of per-household costs for each technology and system size category. The ranges indicate minimum and maximum costs across the scenarios noted in the previous section. For each system size category, the per-household cost range for anion exchange is lower than the corresponding expenditure margin in [ REF \_Ref326056084 \h ]. POU RO devices meet the affordability criteria for the two smaller size categories. EPA's WBS model for POU treatment does not cover systems larger than 3,300 people (greater than 1 MGD design flow).

Exhibit [ SEQ Exhibit \\* ARABIC ]. Expenditure Margins for SSCT Affordability Analysis

System Size (Population Served)	Ion Exchange	Biological Treatment	Reverse Osmosis	Point-of-Use Reverse Osmosis
25-500	\$378 to \$610	\$2,146 to \$3,709	\$2,272 to \$2,671	\$265 to \$271
501-3,300	\$98 to \$148	\$324 to \$566	\$561 to \$688	\$250 to \$251
3,301-10,000	\$104 to \$153	\$211 to \$315	\$431 to \$493	Not applicable1

Note:

The results are mixed for biological treatment and RO. For both technologies, the cost range exceeds the expenditure margin for the smallest system size category. The cost range falls below the expenditure margin for the two larger system size categories. Therefore, biological treatment and centralized RO meet the SSCT criteria for the two larger systems size categories, but not the smallest one. [ REF \_Ref326100138 \h \\* MERGEFORMAT ] provides a summary of which technologies meet SSCT criteria for the three system size categories.

# Exhibit [ SEQ Exhibit \\* ARABIC ]. SSCT Affordability Analysis Results – Technologies that Meet Effectiveness and Affordability Criteria

System Size	lon	Biological	Reverse	Point-of-Use Reverse
(Population Served)	Exchange	Treatment	Osmosis	Osmosis

<sup>1.</sup> EPA's WBS model for POU treatment does not cover systems larger than 3,300 people (greater than 1 MGD design flow), because implementing and maintaining a large-scale POU program is likely to be impractical.

### Best Available Technologies and Small System Compliance Technologies for Perchlorate in Drinking Water

25-500	Yes	No	No	Yes
501-3,300	Yes	Yes	Yes	Yes
3,301-10,000	Yes	Yes	Yes	Not applicable1

#### Note:

<sup>1.</sup> EPA's WBS model for POU treatment does not cover systems larger than 3,300 people (greater than 1 MGD design flow), because implementing and maintaining a large-scale POU program is likely to be impractical.

### 7 References

[ ADDIN ZOTERO\_BIBL {"uncited":[],"omitted":[],"custom":[]} CSL\_BIBLIOGRAPHY ]



# **Technologies and Costs for Treating Perchlorate-Contaminated Waters**

Office of Water (4607M) EPA \*\*\*-\*-\*\*\* November 2018 www.epa.gov/safewater

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[ TOC \O "1-2" \H \Z \T "HEADING 3,3" ]

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### List of Exhibits

[ TOC h z t "Exh Title,1" c "Figure" ]

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#### List of Abbreviations and Symbols

ANSI American National Standards Institute

ARA Applied Research Associates
BAT best available technology
BPOU Baldwin Park Operable Unit

BV bed volumes

CDPH California Department of Public Health<sup>1</sup>

ClO<sub>4</sub> perchlorate anion
DO dissolved oxygen
EBCT empty bed contact time

EPA U.S. Environmental Protection Agency

GAC granular activated carbon gfd or gpd/ft<sup>2</sup> gallons per day per square foot

gfd/psi gallons per day per square foot per pound per square inch

gpm gallons per minute

gpm/ft<sup>2</sup> gallons per minute per square foot gpm/ft<sup>3</sup> gallons per minute per cubic foot

HRT hydraulic residence time
LSI Langelier saturation index
MCL maximum contaminant level

μg/L micrograms per liter
 mg/L milligrams per liter
 MGD million gallons per day
 MWH Montgomery Watson Harza
 NDMA N-nitrosodimethylamine

NF nanofiltration

NSF International, The Public Health and Safety Company

O&M operation and maintenance ORNL Oak Ridge National Laboratory

PNDM Perchlorate and Nitrate Destruction Module

POTW publicly-owned treatment works

POU point-of-use

PRB perchlorate-reducing bacteria

RO reverse osmosis SDI silt density index

SSCT small system compliance technology

T&C technology and costs
TDP Technology Design Panel
TOC total organic carbon
UF ultrafiltration

VOC volatile organic compound
WBS work breakdown structure

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<sup>&</sup>lt;sup>1</sup> Formerly, the California Department of Health Services

### 1 Introduction

### 1.1 Background

The perchlorate anion (ClO<sub>4</sub>-) is an inorganic ion that consists of a tetrahedral array of oxygen atoms around a central chlorine atom. Perchlorate is primarily an anthropogenic contaminant that generally occurs as a perchlorate salt. These salts are used in a wide range of applications, including pyrotechnics and fireworks, blasting agents, matches, lubricating oils, textile dye fixing and so on. Common salts of perchlorate ion are ammonium, potassium, and sodium perchlorate. Ammonium perchlorate, used in rocket and missile propellant, accounts for approximately 90 percent of perchlorate salts production (Xu et al., 2003). These salts are highly soluble in water, and dissociate completely to their cations and anions (perchlorate).

Perchlorate can persist in the environment for many decades under typical groundwater and surface water conditions because of its resistance to reaction with other mutually occurring compounds or elements. The physiochemical properties of perchlorate limit its treatment alternatives. For example, conventional treatment (coagulation and filtration) does not remove perchlorate because it is a poor complexing anion and does not form any complexes easily with other chelating ligands or cations, making it harder to remove perchlorate by chemical precipitation or complexation process.

The U.S. Environmental Protection Agency (EPA) is proposing to regulate perchlorate in drinking water distributed by certain public water systems. In 2011, EPA determined that a national primary drinking water regulation for perchlorate would result in a meaningful opportunity to reduce health risks (USEPA, 2011). Based on the best available scientific information on the health effects of perchlorate, EPA is proposing a maximum contaminant level goal of 56  $\mu$ g/L and an enforceable MCL of 56  $\mu$ g/L. EPA is also requesting comment on 18  $\mu$ g/L for the MCL.

To assist in this evaluating the national costs associated with removing perchlorate, this document describes treatment technologies that have the potential to remove or destroy perchlorate in drinking water. It also presents estimated costs associated with the installation and operation of these technologies. The technologies evaluated here can achieve very high perchlorate removal efficiencies (e.g., 95 percent or greater). Given the high efficiencies, EPA assumes systems will blend treated water and untreated water to meet the MCL. Accordingly, the costs presented here reflect systems designed and operated to take advantage of the technologies' high removal effectiveness and the cost curves should be applied to design and average flows adjusted for blending, as discussed in Chapter [ REF Ref326047741 \r \h ].

### 1.2 Organization and Overview

This report is organized as follows:

- Evaluation of technologies (or other options) for complying with potential perchlorate standards (Chapters [ REF \_Ref529892564 \r \h ] through [ REF \_Ref529892577 \r \h ])
- Costs for treatment technologies (Chapter [ REF Ref326047741 \r \h ]).

The technology evaluations in Chapter [ REF \_Ref529892618 \r \h ] through [ REF \_Ref529892638 \r \h ] describe treatment technologies with the potential to remove or destroy perchlorate in drinking water. Specifically, they address treatment effectiveness for the following:

- ion exchange (Chapter [ REF Ref529892648 \r \h ])
- biological treatment (Chapter [ REF Ref529892663 \r \h ])
- membrane technologies (Chapter [ REF Ref529892673 \r \h ])
- point-of-use (POU) treatment (Chapter [ REF Ref529892683 \r \h ]).

For each technology, the corresponding chapter provides an overview of how the technology operates and summarizes its effectiveness for removal or destruction of perchlorate. Each technology summary also incorporates available findings with respect to variability under different source water conditions. Information on process waste characterization and management is also provided. Each summary concludes with a compilation of the engineering design specifications available from the documents reviewed.

Chapter [ REF \_Ref529892703 \r \h ] discusses alternative, nontreatment options that might be used in lieu of treatment to comply with potential perchlorate standards. Chapter [ REF \_Ref326047741 \r \h ] (in combination with Appendices B and C) presents estimated costs for installing and operating each of the technologies or options discussed in Chapters [ REF \_Ref529892737 \r \h ] through [ REF \_Ref529892767 \r \h ]. Appendix A presents available information on the potential treatment of residuals from perchlorate removal, a topic that is relevant to several of the technologies. Appendix B provides complete cost equations for the technologies and nontreatment options evaluation. Appendix C presents example cost model outputs for selected flow rates, allowing review of individual cost line items.

#### 1.3 Information Sources

The information presented in this document is a summary of EPA's literature search to evaluate the state of science with respect to treatment alternatives for perchlorate-contaminated drinking source water. The objectives of the literature review were to:

- identify what technologies are being studied and tested
- summarize the evidence regarding effectiveness
- characterize other factors relevant for drinking water treatment (e.g., pre- and post-treatment requirements and waste characterization and management options)
- identify key research gaps.

### 2 Ion Exchange

### 2.1 Operating Principle

Ion exchange is a physical/chemical separation process in which an ion such as perchlorate in the feed water is exchanged for an ion (typically chloride) on a resin generally made of synthetic beads or gel. Feed water passes through a bed of resin in a vessel or column. The operation typically continues until the resin does not have sufficient exchange sites available for perchlorate. At this point, the resin may be disposed and replaced or regenerated. Regeneration occurs when the exhausted resin is rinsed with a concentrated chloride solution. Because of the overwhelming concentration, the chloride in the regenerant replaces the adsorbed ions on the resin, returning the resin to its original state.

The fate of perchlorate treated through ion exchange depends on how the spent resin is managed. If the resin is disposed after exhaustion, the perchlorate remains bound to the spent resin. If the resin is regenerated, the perchlorate becomes concentrated in the spent regenerant solution. Perchlorate will be destroyed only if the spent regenerant is further treated (e.g., through physical, chemical, or biological reduction).

Because it is a large, poorly hydrated, hydrophobic anion (see, for example, Gu et al., 2001 and Batista et al., 2002), perchlorate interacts readily with certain types of anion exchange resins, particularly those described as strong-base resins. Several types of resins have the potential to remove perchlorate effectively, at least initially. The key differences among the resins are in their long-term capacity, particularly in the presence of competing anions, and their ease of regeneration. These differences can be significant in terms of the type and quantity of waste generated from the treatment process.

The resin types studied for perchlorate removal include strong-base and weak-base anion resins with polystyrenic, polyacrylic, and polyvinylpyridine matrices, with the most extensive study of strong-base polystyrenic and polyacrylic resins (see, for example, Tripp et al., 2003 and Batista et al., 2000). The category of strong-base polystyrenic resins includes those typically used for removal of nitrate (i.e., "nitrate-selective" resins). In addition, in the early 2000s, researchers at the Department of Energy's Oak Ridge National Laboratory (ORNL) developed a specialized perchlorate-selective resin. Investigators describe this resin as "bifunctional," because it contains two functional groups, one to enhance selectivity and the other to aid kinetics (Batista et al., 2002; Gu et al., 2003). More recently, a number of new, competing perchlorate-selective resins have become available. These include an updated version of the original bifunctional resin, along with other single functional group (often tributylamine) resin formulations (Blute et al., 2006; Russell et al., 2008; Wu and Blute, 2010).

There are significant differences among the strong-base polyacrylic, strong-base polystyrenic, nitrate-selective, and perchlorate-selective resins in terms of their relative affinity for perchlorate (Tripp et al., 2003; Boodoo, 2003; Batista et al., 2002; Darracq et al., 2014). [ REF \_Ref286409769 \h \\* MERGEFORMAT ] categorizes these resins in order of their perchlorate affinity. Although certain resin types have higher relative affinity than others, all of the types shown in [ REF \_Ref286409769 \h \\* MERGEFORMAT ], along with weak-base resins, are able to remove perchlorate. The key differences among the resins are in their long-term capacity,

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particularly in the presence of competing anions, and their ease of regeneration. Because of these differences, the recent literature indicates a trend toward the increased use of perchlorate-selective resins, which are generally disposed of, rather than regenerated. Section [ REF \_Ref286409819 \r \h ] discusses the removal rates achieved by different resin types in more detail. Section [ REF \_Ref286409855 \r \h ] discusses resin capacity in light of raw water quality and Section [ REF \_Ref286409883 \r \h ] covers regeneration needs.

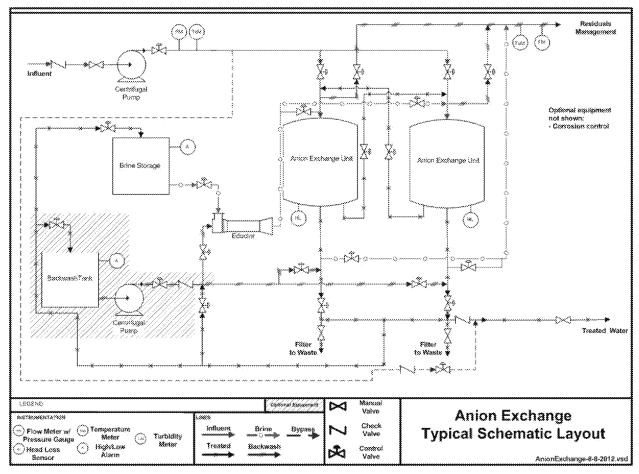
Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Perchlorate Affinity by Resin Type

Perchlorate Affinity	Resin Type	
Lowest affinity	Strong-base polyacrylic	
Lower affinity	Strong-base polystyrenic	
Higher affinity	Nitrate-selective	
Highest affinity	Perchlorate-selective	

Conventional ion exchange systems use a fixed resin bed where, after exhaustion of the resin, operators will take a vessel out of service temporarily to either remove and dispose of the spent resin or regenerate the resin.

[ REF \_Ref286408337 \h \\* MERGEFORMAT ] provides a schematic drawing for a conventional ion exchange system with brine regeneration. With resin disposal, instead of regeneration, as is common for perchlorate-selective resin, the schematic layout becomes simpler. Designs with disposable resin do not require brine storage, eductors, or brine piping. As discussed in Section [ REF \_Ref286408573 \r \h ], another possible option (instead of disposal) for selective resins is to use a novel procedure involving a tetrachloroferrate solution for regeneration. [ REF \_Ref286408793 \h \\* MERGEFORMAT ] provides a schematic drawing for an ion exchange system using disposable resin (i.e., without regeneration).

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Typical Schematic Layout for Perchlorate Removal by Ion Exchange with Brine Regeneration



# Optional equip not shown: - Corresion control Anion Exchange Unit Anion Exchange Unit Treated Wati Captural Espain Anion Exchange Check Valve O Flow Meter w/ O Temperature Typical Schematic Layout Turbidity

#### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Typical Schematic Layout for Perchlorate Removal by Ion Exchange with Resin Disposal

#### 2.2 Effectiveness for Perchlorate Removal

Treates

The State of California has identified ion exchange (along with fluidized bed biological treatment) as one of two BATs for achieving compliance with its standard for perchlorate in drinking water (California Code of Regulations, Title 22, Chapter 15, Section 64447.2). Researchers have demonstrated that ion exchange is capable of removing perchlorate to levels below 2 to 4 µg/L, even given very high influent perchlorate concentrations. This result corresponds, generally, to a removal efficiency in the 90 percent range, depending on the influent concentration. [ REF Ref286411328 \h \\* MERGEFORMAT ] summarizes the removal efficiencies reported in the literature. It includes results from studies conducted in the laboratory, in the field at pilot scale, and in full-scale application.

Pressure Gausse

Head Loss

High/Low Alarm

AnicoExchanga-no regen\_5-6-2012 ver

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Perchlorate Effectiveness Results for Ion Exchange

Resin Type <sup>1</sup>	Removal Efficiency	Resulting Concentration (µg/L)	Study Scale <sup>2</sup>	Data Source(s)
SB	>77% to >94%	<4	Р	Venkatesh et al., 2000; GWRTAC, 2001
	>95.7% to >97%	<4	F	Praskins, 2003; Berlien, 2003; GWRTAC, 2001
	>97.5% to >98.1%	<2,000	F	Praskins, 2003; GWRTAC, 2001; Wagner and Drewry, 2000
	>98%	<4	Р	ITRC, 2008
	>98% and >99.6%	<4	Р	Venkatesh et al., 2000; GWRTAC, 2001
SB-S, SB-A, WB-S, WB-A	>99.9%	<20	L	Batista et al., 2000; Batista et al., 2002
NS	>44%	<4	F	CA EPA, 2004
	>60%	<4	F	CA EPA, 2004
	>60%	<4	F	CA EPA, 2004
	>76%	<4	F	ITRC, 2008
	>85% and >96%	<4	Р	Burge and Halden, 1999
	>99.3%	<3	Р	Gu et al., 1999; Gu et al., 2003
PS	Not specified	<4	F	ITRC, 2008
	>60%	<4	F	ITRC, 2008
	>60% to >73%	<4	F	Hayward and Gillen, 2005; Siemens, 2009b
	>75% to >80%	<2	L, P	Blute et al., 2006
	>82%	<2	Р	Lutes et al., 2010
	>83% to >95%	<2	Р	Russell et al., 2008
	>84%	<4	Р	ITRC, 2008
	>92%	<4	F	ITRC, 2008
	>93.3% to >97.8%	<1	F	Siemens, 2009c; Membrane Technology, 2006
	>94%	<2	Р	Wu and Blute, 2010
	>97.5%	<0.35	F	ITRC, 2008
	>98%	<1	Р	ITRC, 2008
	>98.6%	<4	F	ITRC, 2008
	>97.6% to >99.2%	<0.5	F	Drago and Leserman, 2011
	>99.3%	<3	Р	Gu et al., 1999; Gu et al., 2003
	>99.7%	<3	L	Gu et al., 1999
WB-S	>98.5%	<0.1	Р	U.S. DoD, 2008a
	>99.7%	<4	Р	U.S. DoD, 2007
Not specified	>60%	<4	F	CA EPA, 2004
	>60% to >98%	<4	F	ITRC, 2008
	>71%	<4	F	ITRC, 2008
	>73%	<4	F	Fontana Water Company, 2010; ITRC, 2008
	>75%	<5	F	Santschi, 2010
	>90%	<2	F	ITRC, 2008
	>96% to >99.7%	<4	L	GWRTAC, 2001
	>99%	<4	F	Siemens, 2009a

#### Notes:

<sup>1.</sup> SB = strong-base; SB-S = strong-base polystyrenic; SB-A = strong-base polyacrylic; WB-S = weak-base polystyrenic; WB-A = weak-base polyacrylic; NS = nitrate-selective strong-base polystyrenic; PS = perchlorate selective

<sup>2.</sup> L = laboratory study; P = field pilot study; F = full-scale

[ REF \_Ref286411328 \h \\* MERGEFORMAT ] also shows the variety of resin types that have been tested for perchlorate removal. These resin types include strong-base polyacrylic, strong-base polystyrenic (including nitrate-selective), weak-base polyacrylic, weak-base polystyrenic, and perchlorate-selective. All of these resin types can attain very high perchlorate removals, at least initially. While Batista et al. (2000 and 2002) have suggested that weak-base resins may have certain advantages in terms of regenerant treatment (see Section [ REF \_Ref292709289 \r \h ] and Appendix A), tests of these resins have been limited, with only a few studies documented in the reviewed literature (Batista et al., 2000 and 2002; Boodoo, 2006; U.S. DoD, 2007 and 2008a). Furthermore, the use of weak-base resins could require pH adjustment.

Additional support for the effectiveness of ion exchange for perchlorate removal is evident from the number of full-scale facilities that are currently using the technology. As shown in [REF\_Ref286411968 \h \\* MERGEFORMAT], the literature identifies 44 full-scale facilities applying ion exchange for perchlorate removal. [REF\_Ref286411968 \h \\* MERGEFORMAT] also demonstrates the increasing use of perchlorate-selective resins. These installations include both remediation sites and facilities producing drinking water.

Currently, the majority of the identified full-scale facilities (18 of 23 facilities where information on resin type is available) currently use perchlorate-selective resins. An additional two facilities are reportedly planning to switch to perchlorate-selective resin (Wu and Blute, 2010; Blute, 2012). Thus, perchlorate-selective resin appears to have become the technology of choice for perchlorate ion exchange facilities.

<sup>&</sup>lt;sup>2</sup> While Tripp et al. (2003) also examined strong base polyvinylpyridine resins, comparable quantitative data on their removal efficiency are not available.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Full-scale Ion Exchange Installations for Perchlorate

Location	Flow rate (gallons per minute [gpm])	Resin Type <sup>1</sup>	Resin Fate <sup>2</sup>	Start Date	Data Source(s)
Kerr-McGee, Henderson, Nevada	300 to 600		D	November 1999 <sup>3</sup>	Min et al., 2003; Praskins, 2003; GWRTAC, 2001; CA EPA, 2003; CA EPA, 2004; Croft, 2004; ITRC, 2008; Roefer, 2013
LaPuente Valley County Water District, California	2,500	PS <sup>9</sup>	D 9	February 2000	Wagner and Drewry, 2000; GWRTAC, 2001; Min et al., 2003; Berlien, 2003; Praskins, 2003; CA EPA, 2003; Coppola, 2003; CA EPA, 2004; Bull et al., 2004; Pollack, 2004; ITRC, 2008; Russell et al., 2009; Faccini et al., 2016
Lawrence Livermore National Laboratory, Livermore, California	0.7 to 3.5	NS	R	November 2000	CA EPA, 2004; USEPA, 2005; ITRC, 2008
Kerr-McGee, Henderson, Nevada	200 to 560		R	March 2002 <sup>4</sup>	Min et al., 2003; Wagner and Drewry, 2000; CA EPA, 2003; CA EPA, 2004; Croft, 2004; USEPA, 2005; ITRC, 2008; Roefer, 2013; Faccini et al., 2016
California Domestic Water Company, Whittier, California	5,000	PS <sup>8</sup>	D	July 2002	Coppola, 2003; Min et al., 2003; Bull et al., 2004; CA EPA, 2004; Pollack, 2004; Blute et al., 2006; ITRC, 2008; Purolite, 2011; Russell et al., 2009
Gage 51-1, City of Riverside, California	2,000	SB-S	D	October 2002	Lu, 2003; CA EPA, 2004; Bull et al., 2004; Pollack, 2004
Tippecanoe, City of Riverside, California	5,000	SB-S	D	December 2002	Min et al., 2003: Lu, 2003; CA EPA, 2004; Bull et al., 2004; Pollack, 2004; USEPA, 2005; ITRC, 2008
R&H System, La Verne, California			R	2003 or earlier	Min et al., 2003
City of Pomona, California	10,000	NS <sup>10</sup>	R <sup>10</sup>	2003 or earlier	Coppola, 2003; Min et al., 2003; Bull et al., 2004; CA EPA, 2004; USEPA, 2005
Baldwin Park, California	7,000 and 7,500		R	2003 or earlier	Min et al., 2003

	Flow rate (gallons per minute				
Location	[gpm])	Resin Type <sup>1</sup>	Resin Fate <sup>2</sup>	Start Date	Data Source(s)
Edwards Air Force Base, Site 285, California	30	PS	R	2003	ITRC, 2008
West San Bernadino Water District, Rialto, California	2,000	PS	D	May 2003	Coppola, 2003; CA EPA, 2003; CA EPA, 2004; Bull et al., 2004; Pollack, 2004; ITRC, 2008
West Valley Water District, San Bernadino, California	2,000		D	June 2003	Coppola, 2003; CA EPA, 2004; USEPA, 2005; Siemens, 2009d;
Aerospace Manufacturer, Maryland	20		D 7	October 2003	Siemens, 2009a
Wells 15, 17, and 24, City of Colton, California	3,600	PS	D	August 2003	Bull et al., 2004; CA EPA, 2004; Xiong and Zhao, 2004; ITRC, 2008
Airport Treatment Plant, City of Rialto, California	2,000	PS	D	August 2003	Coppola, 2003; CA EPA, 2004; Bull et al., 2004; Purolite, 2011
Delta Treatment Plant, City of Monterey Park, California	4,050		D	July 2003	Coppola, 2003; CA EPA, 2003; CA EPA, 2004; Pollack, 2004
Santa Clara Valley Water District			D	Prior to December 2003	Coppola, 2003
Colony and County Wells, West San Martin Water Works, West San Martin, California	800	NS	D	2004 or earlier	CA EPA, 2004; Bull et al., 2004; USEPA, 2005; ITRC, 2008
Texas Street, City of Redlands, California	1,100	PS	D	2004	City of Redlands, 2004; Bull et al., 2004; ITRC, 2008
Fontana Union Water Co., Fontana, California	6,000	PS	D	January 2004	CA EPA, 2004; USEPA, 2005; ITRC, 2008; Fontana Water Company, 2010
Castaic Lake Water Agency, Whittaker, California	300	PS <sup>11</sup>	D 11	Prior to March 2004	Bull et al., 2004; USEPA, 2005; Blute, 2012
City of Morgan Hill, California	400 to 1,000		D	Prior to March 2004 <sup>5</sup>	Bull et al., 2004; CA EPA, 2004; USEPA, 2005; Russell et al., 2009
Big Dalton Well, San Gabriel Water Quality Association, Baldwin Park, California	3,000			Prior to March 2004	Bull et al., 2004
Camping World, San Martin County Water District, California	2,000			Prior to March 2004	Bull et al., 2004

Location	Flow rate (gallons per minute [gpm])	Resin Type <sup>1</sup>	Resin Fate <sup>2</sup>	Start Date	Data Source(s)
Southern California Water Co., South San Gabriel, California	750			Prior to March 2004	Bull et al., 2004
Fontana (F17 site), San Gabriel Valley Water Company, El Monte, California	5,000	PS	D	Prior to March 2004	Bull et al., 2004; Lutes et al., 2010 ; Purolite, 2011
B6 Well Site, San Gabriel Valley Water Company, El Monte, California	7,800	6	R <sup>6</sup>	May 2004	CA EPA, 2003; CA EPA, 2004; Pollack, 2004; Russell et al., 2009
Valley County Water District, Baldwin Park, California	7,800	6	R <sup>6</sup>	June 2004	Coppola, 2003; CA EPA, 2003; CA EPA, 2004; Pollack 2004; Russel et al., 2009
Jet Propulsion Laboratory, Pasadena, California	1,400		D	July 2004	USEPA, 2005
Aerojet, Sacramento, California	400 to 2,000	PS	D 7	August 2004	CA EPA, 2004; USEPA, 2005; ITRC, 2008; Purolite, 2011
Lincoln Avenue Water Co., Altadena, California	2,000	PS	D 7	August 2004	Hayward and Gillen, 2005; Siemens, 2009b; ITRC, 2008
Frank Perkins Road Treatment System, Massachusetts Military Reservation, Cape Cod, Massachusetts	300			September 2004	USEPA, 2005
B5 Well Site, San Gabriel Valley Water Company, El Monte, California	7,800	PS	D	December 2004	CA EPA, 2004; Pollack, 2004; Purolite, 2011
Phoenix Goodyear Airport North, City of Goodyear, Arizona	440		D	2005	ITRC, 2008
Aquarion Water Co., Millbury, Massachusetts	1,500	PS	D 7	June 2005	Siemens, 2009c; Membrane Technology, 2006
California Water Services Company, Porterville, California			D	April 2006	Siemens, 2006
Camp Edwards portion of the Massachusetts Military Reservation, Cape Cod, Massachusetts	1,000	PS	D	2007	ITRC, 2008
Naval Weapons Industrial Reserve Plant, McGregor, Texas			D	2008 or earlier <sup>3</sup>	ITRC, 2008

Location	Flow rate (gallons per minute [gpm])	Resin Type <sup>1</sup>	Resin Fate <sup>2</sup>	Start Date	Data Source(s)
Arrowhead Regional Medical Center, Colton, California	600		1	January 2010	Water & Wastes Digest, 2010
Saugus Perchlorate Treatment Facility, Castaic Lake Water Agency, Santa Clarita, California	2,200	PS	D	May 2010	Drago and Leserman, 2011
Richardson Treatment Plant, Loma Linda, California	1,200			October 2010	Santschi, 2010
Monk Hill Water Treatment Plant, Pasadena Water and Power, Pasadena, California	7,000	PS	D	July 2011	Erdman, 2011; Blute, 2012
Golden State Water Service	2,000	PS	D	2011 or earlier	Purolite, 2011

#### Notes:

#### -- = Not reported

- 1. SB-S = strong-base polystyrenic; SB-A = strong-base polyacrylic; NS = nitrate-selective strong-base polystyrenic; PS = perchlorate-selective
- 2. D = Disposed; R = Regenerated
- 3. No longer in operation (replaced with biological reactor).
- 4. Discontinued after 6 months due to operational issues.
- 5. Inactive as of March 2004.
- 6. Reportedly planning to switch to use of perchlorate-selective resin with disposal instead of regeneration (Wu and Blute, 2010).
- 7. Specifically, spent resin at this facility is incinerated.
- 8. Switched from strong-base polystyrenic resin to perchlorate-selective resin as of 2011 (Purolite, 2011; Wu and Blute, 2010).
- 9. Switched from strong-base polyacrylic resin to perchlorate-selective resin with disposal instead of regeneration in July 2010 (Blute, 2012).
- 10. Currently installing perchlorate-selective resin in addition for part of the treatment train (Blute, 2012).
- 11. Installed perchlorate-selective resin beginning in 2011 (Blute, 2012).

### 2.3 Raw Water Quality Considerations

The most significant raw water quality consideration in ion exchange perchlorate treatment is the concentration of competing anions (particularly sulfate, nitrate, bicarbonate, and chloride). The effect of these anions is to decrease a resin's longer-term capacity to adsorb perchlorate, as they compete with perchlorate for exchange sites. Resin capacity (also termed resin life or run length) typically is measured by the number of bed volumes (BV) of water that can be treated before breakthrough of perchlorate. Competing anions take up available exchange sites, reducing perchlorate capacity. In addition, these anions may break through or peak<sup>3</sup> before perchlorate, affecting finished water quality and limiting the practical life of the resin more than perchlorate capacity alone. For example, Case et al. (2004) reported that a strong-base polyacrylic resin could treat 750 BV before perchlorate breakthrough. Nitrate peaking, however, would limit the use of the resin to 425 BV. In practice, however, systems can limit the impact of peaking by using multiple treatment trains in parallel. Also, the full-scale facility studied in Drago and Leserman (2011) eliminated chloride peaking by converting the resin from a chloride to a bicarbonate form prior to installation.

There are significant differences among resin types in terms of the relative impact of competing anions. This impact is related to the relative affinity of the resin for each anion present. [ REF \_Ref286412546 \h \\* MERGEFORMAT ] shows quantitative data from the literature on BV to perchlorate breakthrough for different resin types in the presence of differing concentrations of the major competing anions. The data shown in [ REF \_Ref286412546 \h \\* MERGEFORMAT ] are for initial detection of perchlorate (at detection limits between 1 and 4 µg/L, depending on the specific study) using a single resin column. After perchlorate breakthrough, most resins still have the capacity to continue adsorbing perchlorate before the resin is completely saturated. In practice, using two columns in series (a "lead-lag" configuration) can capture this extra capacity (Boodoo, 2003). For example, Gu et al. (1999) found breakthrough in a lead column after 8,500 BV for a nitrate-selective resin and 40,000 BV for a perchlorate-selective resin. Using a second (lag or polishing) column increased the resins' capacities to approximately 22,000 BV and 104,000 BV, respectively.

Precise, quantitative comparisons of the data in [ REF\_Ref286412546 \h \\* MERGEFORMAT ] are difficult because of variations among studies (e.g., influent concentrations of perchlorate and other constituents, definition of breakthrough, and specific resin manufacturer). The data, however, when combined with general conclusions in the literature, do allow for some general observations about differences among resin types.

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<sup>&</sup>lt;sup>3</sup> Peaking occurs when competing anions adsorbed early in a resins life are displaced by perchlorate, resulting in an effluent concentration of the competing anions greater than the influent concentration. See Boodoo (2003) for an example of peaking behavior.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Perchlorate Capacity by Resin Type and Competing Anions

Resin Type <sup>1</sup>	Data Source(s)	Perchlorate Capacity (BV to breakthrough)	Competing Anions (mg/L)			
			Sulfate	Nitrate <sup>2</sup>	Bicarbonate	Chloride
SB-A	Batista et al., 2000 and 2002	1,800 to 2,100	None	None	None	None
	Case et al., 2004	750	Present	Present	Present	No data
	Tripp et al., 2003	700	50	6	122	30
	Min et al., 2003	700	55	27	155	15
	Boodoo, 2003	500	44	40	170	13
	Lehman et al., 2008	650	56	63	300	25
	Batista et al., 2000 and 2002	O <sup>3</sup>	>2,000	>40	No data	No data
SB-S	Batista et al., 2000 and 2002	3,750	None	None	None	None
	Tripp et al., 2003	6,000	50	6	122	30
	Boodoo, 2003	5,000	44	40	170	13
	Batista et al., 2000 and 2002	600 <sup>3</sup>	>2,000	>40	No data	No data
NS	Batista et al., 2000	1,300	None	None	None	None
	Gu et al., 2003	14,000	173	3.2	226	356
	Tripp et al., 2003	25,0004	50	6	122	30
	Gu et al., 1999	8,500	14.9	61.2	No data	7.0
PS-Old	Gu et al., 2003	40,000	173	3.2	226	356
	Tripp et al., 2003	35,000 <sup>3</sup>	50	6	122	30
	Min et al., 2003	20,700	266	14	229	40
	Gu et al., 1999	40,000	14.9	61.2	No data	7.0
	Gu et al., 2007	37,000	16	10	No data	10
	Lutes et al., 2010	97,000	14.8	32.8	192	10.7
	Blute et al., 2006	~75,000	46.4	1.6	No data	No data
PS-New	Blute et al., 2006	~160,000	46.4	1.6	No data	No data
	Russell et al., 2008	~130,000 to 170,000	37	25	No data	23
	Russell et al., 2008	~125,000 to 140,000	44	33	No data	27
	Russell et al., 2008	~105,000 to 130,000	53	61	No data	47
	Wu and Blute, 2010	~130,000	45	33	No data	No data
	Drago and Leserman, 2011	~30,000 to 60,000	130 to 220	18	No data	17 to 29
WB-S	Batista et al., 2000	500	None	None	None	None
	U.S. DoD, 2007	3,000 to 4,000	3	4	No data	4
	U.S. DoD, 2008a	9,700	14	31	150	11
	Boodoo, 2006	15,000	No data	No data	No data	No data
WB-A	Batista et al., 2000 and 2002	2,700 to 2,800	None	None	None	None
	Batista et al., 2000 and 2002	800 to 1,000	100	100	None	100

#### Notes:

<sup>1.</sup> SB-S = strong-base polystyrenic; SB-A = strong-base polyacrylic; WB-S = weak-base polystyrenic; WB-A = weak-base polyacrylic; WB = weak-base (type not specified); NS = nitrate-selective strong-base polystyrenic; PS-Old = original perchlorate-selective resin developed by ORNL; PS-New = more recently developed perchlorate-selective resins.

<sup>2.</sup> as NO3

<sup>3.</sup> The presence of humic substances caused reduced resin capacity.

<sup>4.</sup> Column fouling (due to experimental design) may have caused reduced resin capacity.

#### 2.3.1 Strong-base Polyacrylic Resins

The order of affinity of strong-base polyacrylic resins is as follows (Boodoo, 2003; Gu et al., 2001):

sulfate > perchlorate > nitrate > chloride > bicarbonate

Accordingly, Tripp et al. (2003) and Boodoo (2003) conclude that the capacity of strong-base polyacrylic resins is most significantly affected by sulfate concentration. Using computer modeling, Tripp et al. (2003) predicted an 88 percent decrease in strong-base polyacrylic capacity as influent sulfate increased from 1 to 250 milligrams per liter (mg/L). The decrease in capacity was rapid as sulfate increased (see Tripp et al., 2003, Figure 4.61). The cross-study data in [REF\_Ref286412546 \h \\* MERGEFORMAT] are consistent with this prediction. For this resin type, Batista et al. (2000 and 2002) found a perchlorate capacity of 1,800 to 2,100 BV for a laboratory solution with no sulfate. In comparison, sulfate concentrations in the range of 40 to 60 mg/L reduced capacity to 500 to 700 BV in other studies (Case et al., 2004; Tripp et al., 2003; Min et al., 2003; Boodoo, 2003; Lehman et al., 2008). Batista et al. (2000 and 2002) found that very high sulfate concentrations (greater than 2,000 mg/L) prevented a polyacrylic resin from removing any perchlorate (i.e., a capacity of 0 BV), although the investigators suggest fouling of the resin as a contributing factor to the decreased capacity.

Tripp et al. (2003) predicted that strong-base polyacrylic resins are not as sensitive to increasing nitrate concentrations (see Tripp et al., 2003, Figure 4.62). Capacity decreased linearly by 36 percent as influent nitrate increased from 0.1 to 20 mg/L as N (Tripp et al., 2003). Again, the cross-study data support this conclusion, finding similar capacities (500 to 700 BV) for nitrate concentrations from 6 to 63 mg/L as NO<sub>3</sub> (Case et al., 2004; Tripp et al., 2003; Min et al., 2003; Boodoo, 2003; Lehman et al., 2008).

#### 2.3.2 Strong-base Polystyrenic Resins

The order of affinity of strong-base polystyrenic resins is as follows (Boodoo, 2003; Gu et al., 2001):

Therefore, the capacity of strong-base polystyrenic resins should be affected by sulfate concentration, but to a lesser degree than that of polyacrylic resins. Using computer modeling, Tripp et al. (2003) predicted a 94 percent decrease in strong-base polystyrenic capacity as influent sulfate increased from 1 to 250 mg/L. The decrease in capacity, however, was not as rapid as for the polyacrylic resin. Interpolation of Figure 4.61 in Tripp et al. (2003) shows a capacity decrease of approximately 60 percent for the polystyrenic resin at 50 mg/L sulfate, compared to a decrease of nearly 80 percent for the polyacrylic resin. Similarly, the cross-study data in [REF\_Ref286412546 \h \\* MERGEFORMAT ] show a relatively high perchlorate capacity (5,000 to 6,000 BV) for strong-base polystyrenic resins at moderate sulfate concentrations (40 to 50 mg/L) (Tripp et al., 2003; Boodoo, 2003).

Tripp et al. (2003) predicted that strong-base polystyrenic resins, like polyacrylic resins, are not as sensitive to increasing nitrate concentrations (see Tripp et al., 2003, Figure 4.62). Capacity decreased linearly by 26 percent as influent nitrate increased from 0.1 to 20 mg/L (Tripp et al.,

2003). The cross-study data are consistent with this observation, finding similar capacities (5,000 to 6,000 BV) for nitrate concentrations from 6 to 40 mg/L (Tripp et al., 2003; Boodoo, 2003).

#### 2.3.3 Nitrate-Selective Resins

The order of affinity of nitrate-selective resins is as follows (Boodoo, 2003; Burge and Halden, 1999):

perchlorate > nitrate > sulfate > chloride > bicarbonate

Accordingly, computer modeling performed by Tripp et al. (2003) found these resins affected by both sulfate and nitrate, with the greater effect caused by nitrate. Nitrate-selective capacity for perchlorate decreased by 76 percent as nitrate increased from 0.1 to 20 mg/L, compared to 64 percent for a sulfate increase from 1 to 250 mg/L (Tripp et al., 2003). The cross-study data in [ REF\_Ref286412546 \h \\* MERGEFORMAT ] appear consistent, if one ignores an anomalous data point from Batista et al. (2000). Capacities shown are 25,000 BV for moderate sulfate and moderate nitrate (Tripp et al., 2003), 14,000 BV for high sulfate and low nitrate (Gu et al. 2003), and 8,500 BV for low sulfate and high nitrate (Gu et al. 1999).

#### 2.3.4 Perchlorate-Selective Resins

As discussed in Section [ REF \_Ref529882092 \r \h ], researchers at ORNL developed the first perchlorate-selective resin in the early 2000s. This original, bifunctional resin was known as "BiQuat" and licensed to Purolite for sale under the name Purolite A530E (Boodoo, 2003). More recently, a number of new, competing perchlorate-selective resins have become commercially available. These include (U.S. Filter, 2004; Russell et al., 2008; Blute et al., 2006; Wu and Blute, 2010; Drago and Leserman, 2011; Darracq et al., 2014):

- Purolite A532E (an updated version of the old A530E resin)
- Purolite MCG-P2
- Resin Tech SIR-110-HP
- Rohm & Haas PWA2
- Dow PSR2
- Calgon CalRes 2109.

The order of affinity of the original perchlorate-selective resin was the same as that for nitrate-selective resins (i.e., perchlorate > nitrate > sulfate > chloride), but the perchlorate affinity relative to nitrate affinity was nearly an order of magnitude greater (Boodoo, 2003). Boodoo (2003) suggested that this original resin would be negatively affected by high nitrate concentrations. The cross-study data in [ REF \_Ref286412546 \h \\* MERGEFORMAT ], however, suggest that the resin was not, in fact, very sensitive to competing anions. Capacity remained high (20,700 to 97,000 BV) for a wide range of nitrate and sulfate concentrations (1.6 to 61.2 mg/L and 14.8 to 266 mg/L, respectively) (Gu et al., 2003; Tripp et al., 2003; Min et al., 2003; Gu et al., 1999; Gu et al., 2007; Lutes et al., 2010; Blute et al., 2006).

For the more recently developed perchlorate-selective resins, the data in [ REF \_Ref286412546 \h \\* MERGEFORMAT ] generally show significantly greater perchlorate capacity than the original resin. In the presence of moderate levels of both sulfate and nitrate, the new resins showed capacities of approximately 105,000 to 170,000 BV (Blute et al., 2006; Russell et al.,

2008; Wu and Blute, 2010). Even at higher sulfate levels, at least one of the new resins showed equal or greater capacity than the older resin (Drago and Leserman, 2011). The differences in capacity among the individual new resins appear to be less significant than the difference between the new resins as a group and the old resin. In the studies shown in [ REF \_Ref286412546 \h \\* MERGEFORMAT ] (Blute et al., 2006; Russell et al., 2008; Wu and Blute, 2010; Drago and Leserman, 2011), different resins performed better depending on variations in specific competing anions and other site-specific conditions (i.e., no one of the new resins was consistently superior to the others in all of the studies).

#### 2.3.5 Weak-base Resins

As shown in [REF\_Ref286412546 \h \\* MERGEFORMAT ], researchers have conducted only limited study on the effect of competing anions on the perchlorate capacity of weak-base resins. Sulfate and nitrate reduce the capacity of weak-base polyacrylic resins when they are present at relatively high levels (100 mg/L each) (Batista et al., 2000; Batista et al., 2002). Data are not available, however, on the effect of more moderate levels of competing anions. Batista et al. (2000) reported that weak-base polystyrenic resins have a relatively low capacity (500 BV) even absent competing anions. Pilot tests at Redstone Arsenal in Alabama used a weak-base polystyrenic resin produced by Purolite. This resin achieved 15,000 BV (Boodoo, 2006), 3,000 to 4,000 BV (U.S. DoD, 2007), or 9,700 BV (U.S. DoD, 2008a) capacity, depending on the test conditions. U.S. DoD (2008a) suggested that competing ions such as nitrate would reduce the capacity of this resin, but data are not available on the magnitude of such competitive effects.

### 2.3.6 Raw Water Quality Considerations other than Sulfate and Nitrate Competition

Although most investigators identify bicarbonate and chloride as other major competing anions, the affinity of ion exchange resins for these anions is less than that for perchlorate, sulfate, and nitrate. Therefore, their impact on resin perchlorate capacity would be expected to be less than that of sulfate and nitrate. There are, however, no quantitative data in the literature on the effects of these major anions. Similarly, U.S. DoD (2002) indicates that raw water pH can strongly influence treatment effectiveness, but no quantitative data are available.

Other co-contaminants that may affect perchlorate capacity include arsenic (Tripp et al., 2003; Berlien, 2003), uranium (Tripp et al., 2003; Min et al., 2003), and chromium (Min et al., 2003). Based on the high affinity of most resins for perchlorate, direct competition from these co-contaminants would be expected to be low. Accumulation of these contaminants in high concentrations on the resin or in regenerant solution may affect disposal (see Section [ REF \_Ref286414613 \r \h ]), limiting the practical life of a resin. For example, Tripp et al. (2003) suggests that a perchlorate-selective resin would require regeneration every 10,000 BV to prevent arsenic and uranium build-up. Recent studies of various perchlorate-selective resins, however, have shown that build-up of metals results in concentrations that are below regulatory limits that would require disposal as a hazardous waste, both under federal requirements and California's more stringent limits (Russell et al., 2008; Blute et al., 2006; Wu and Blute, 2010). The same studies found that uranium build-up might require special handling as a radioactive waste in only one of the 12 samples tested (total across all three studies). In some cases, however, incineration facilities have facility-specific restrictions on uranium concentrations that are more stringent that the regulatory thresholds for radioactive waste.

#### 2.4 Pre- and Post-Treatment Needs

Suspended solids and organic substances in source water can cause clogging or fouling of ion exchange resins (U.S. DoD, 2002; Batista et al., 2002; Gu et al., 2003). For example, Batista et al. (2002) found that high concentrations of humic substances (as measured by total organic carbon [TOC]) caused fouling of both strong-base polyacrylic and polystyrenic resins, interfering with perchlorate removal. In spite of earlier conclusions that the perchlorate-selective bifunctional resin would require no pre-treatment (Gu et al., 1999; ORNL, 2002), in pilot testing of the resin, Gu et al. (2003) found that precipitation and/or deposition of iron oxyhydroxides and microbial biomass caused significant clogging and fouling of the resin columns. The addition of a fine in-line filter resolved the problem. Similarly, the full-scale system studied in Drago and Leserman (2011) included pre-treatment bag filters and sulfuric acid addition to minimize scaling. Therefore, the presence of suspended solids and organic matter may require the use of filtration and/or chemical addition as pretreatment. Although the literature reviewed for this report does not identify the specific conditions under which filtration is needed (e.g., concentration of total suspended solids), these conditions are expected to be similar to those documented in application of ion exchange treatment for other contaminants. For perchlorateselective resins, pre-filtration (bag or cartridge filters) may be required regardless of water quality because the long run lengths (see Section [ REF Ref350336683 \r \h ]) can result in greater solids accumulation.

Batista et al. (2000) indicates that weak-base resins require carbonation to treat perchlorate. Carbonation can be accomplished by adding carbon dioxide or bicarbonate (generated by feeding sodium bicarbonate through a strong-acid cationic resin) to the feed. Thus, carbonation may constitute a pre-treatment step required for the use of weak-base resins. More information is required on the need for and practicality of this step to evaluate the use of weak-base resins. The weak-base resin piloted in Alabama requires pH reduction with carbon dioxide or acid as a pre-treatment step, carbon dioxide removal as a post-treatment step, and pH adjustment, if needed, as a post-treatment step (Boodoo, 2006; U.S. DoD, 2007 and 2008a). An operational pH between 3 and 5, with a target of 4, is required for this resin to operate effectively (U.S. DoD, 2007 and 2008a).

Ion exchange treatment can increase the corrosivity of treated water (Betts, 1998; Berlien, 2003; U.S EPA, 2005) because of the addition of chloride ions and/or removal of carbonates and bicarbonates. Berlien (2003) reports this problem with a full-scale application of ion exchange for perchlorate treatment. Treated water had a pH of approximately 7 and created red water problems in older homes with galvanized steel pipe. The operators corrected this problem by adding sodium hydroxide to raise the pH to approximately 8.2 and adding polyphosphates as an additional protection measure. For applications of weak-base resins where pre-treatment pH adjustment would be required, increasing the pH after treatment would also be necessary for corrosion control.

Tripp et al. (2003) and Min et al. (2003) indicate that N-nitrosodimethylamine (NDMA) may form within certain polystyrenic resins and leach into the treated water. Recent studies of perchlorate-selective resins (Russell et al., 2008; Blute et al., 2006; Wu and Blute, 2010; Drago and Leserman, 2011) have shown leaching of various nitrosamines at first flush or soon after startup, with levels that decline to below detection over time (four hours to one week, depending

on the specific resin and nitrosamine). Thus, nitrosamine leaching appears to occur periodically and temporarily (e.g., after installation of new resin). Issues with nitrosamine leaching might be avoided if resins are sufficiently flushed prior to use. For example, the full-scale system studied in Drago and Leserman (2011) minimized nitrosamine leaching with changes to the manufacturer's pre-delivery rinsing and preparation procedures. For one resin, however, Wu and Blute (2010) suggest that rinsing longer than manufacturer recommendations may be required to eliminate leaching concerns. Blute et al. (2006) also hypothesize that nitrosamines could be eliminated by downstream processes such as ultraviolet light treatment, if present.

# 2.5 Waste Generation and Residuals Management Needs

After a resin reaches its perchlorate capacity (see Section [ REF \_Ref286414835 \r \h ]), the operator must either dispose and replace the resin or regenerate the resin using a chemical solution to remove the adsorbed anions. The former option, commonly termed "throwaway" operation, generates solid waste in the form of spent resin loaded with perchlorate and other anions. The latter option generates a liquid waste in the form of spent regenerant with concentrated perchlorate and other anions. Both options can also generate liquid waste in the form of spent wash water when initial flushing is required upon installation of new resin (e.g., to prevent nitrosamine leaching, see Section [ REF \_Ref350937116 \r \h ]). As shown in [ REF \_Ref286411968 \h \\* MERGEFORMAT ], almost 79 percent (30 of 38) of full-scale facilities for which waste management data are available dispose of spent resin. This statistic includes all but one of the full-scale facilities using perchlorate-selective resin. An additional two facilities are reportedly planning to switch away from regeneration to disposal of spent resin (Wu and Blute, 2010; Blute, 2012).

## 2.5.1 Disposal

In systems using resin disposal without regeneration, calculation of the quantity of solid waste generated would be straightforward, based on the quantity of resin used and the life of the resin. Spent resin characteristics depend on resin type, influent water quality, and the life of the resin. As discussed in Section [ REF \_Ref286658749 \r \h ], studies of metals build-up in perchlorate-selective resins have found that these resins are not likely to meet regulatory definitions of hazardous waste (Russell et al., 2008; Blute et al., 2006; Wu and Blute, 2010). Because of the shorter life of conventional resins, metals accumulation in these resins likely would be even lower and, thus, the same result should hold true. A typical destination for non-hazardous spent resin would be disposal in an off-site landfill. Based on the data in [ REF \_Ref286411968 \h \\* MERGEFORMAT ], however, at least four of the full-scale facilities appear to be sending their spent resin to incineration facilities.

#### 2.5.2 Regeneration

In systems using regeneration, the characteristics and quantity of spent regenerant depend on the type of regenerant solution used. The type of regenerant solution selected depends, in turn, on the type of resin used. In addition, the quantity of spent regenerant can be reduced if the regenerant is treated and reused. Appendix A discusses regenerant treatment.

In conventional ion exchange processes using strong-base resins (i.e., for removal of arsenic), operators regenerate spent resin using a brine solution of concentrated sodium chloride or potassium chloride. For resins loaded with perchlorate, however, regeneration with brine is more

difficult because of the high relative affinity of most resins for perchlorate. In general, the higher the perchlorate affinity of a resin (see [ REF \_Ref286409769 \h \\* MERGEFORMAT ]), the more difficult it is to regenerate using conventional brine solutions. For example, Tripp et al. (2003) found that regeneration required a significantly greater quantity of brine solution for strong-base polystyrenic resins than for strong-base polyacrylic resins. Similarly, Batista et al. (2000) were able to successfully regenerate a perchlorate-loaded strong-base polyacrylic resin with 12 percent sodium chloride, removing 96 percent of the loaded perchlorate. In comparison, they found regeneration of a nitrate-selective resin using the same solution very ineffective, removing only 17.3 percent of the loaded perchlorate. Investigators, therefore, suggest that highly perchlorate-selective resins cannot be regenerated at all using conventional regenerant solutions (Boodoo, 2003; Gu et al., 2001; Gu et al., 2003; Batista et al., 2000; Darracq et al., 2016).

# Regeneration of Non-selective Strong-Base Resins

It appears that, while regeneration of non-selective resins using brine solutions is feasible, large quantities of spent regenerant may result. Although polyacrylic resins regenerate easily, they have relatively short run lengths (see Section [ REF \_Ref286415007 \r \h ]) and, therefore, more frequent regeneration. On the other hand, while polystyrenic resins have longer run lengths, they require more regenerant. Based on computer modeling, Tripp et al. (2003) conclude that the polyacrylic resins are more efficient in terms of quantity of spent regenerant as a percentage of water treated. In practice, regeneration may be accomplished using partial exhaustion-partial regeneration. In this scenario, operators regenerate the resin well before perchlorate breakthrough (e.g., at the point of sulfate or nitrate breakthrough) and regenerate using smaller quantities of brine than would be required for complete perchlorate removal. This practice allows operation for a number of cycles until perchlorate builds up on the resin and complete regeneration is required, and may result in lower overall generation of spent regenerant.

Tripp et al. (2003) suggest partial exhaustion-partial regeneration operation for pilot testing of non-selective resins. Results of tests of this approach reported by Case et al. (2004) suggest that partial exhaustion-partial regeneration is effective, at least for the strong-base polyacrylic resin, with no change in performance for over 20 cycles.

Based on the designs reported in Montgomery Watson Harza (MWH) and University of Houston (2003) and Case et al. (2004), spent regenerant generation in the pilot tests would be 1.5 to 1.6 percent of treated water (6.4 BV of regenerant/400 to 425 BV of treated water) for the polyacrylic resin and 1.4 percent of treated water (9 BV of regenerant/625 BV of treated water) for the polystyrenic resin.<sup>4</sup> Data on spent regenerant generation are not available for the full-scale operations using conventional treatment configurations, although Betts (1998) reports that conventional processes (for contaminants other than perchlorate) typically generate 2 to 5 percent brine waste.<sup>5</sup>

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<sup>&</sup>lt;sup>4</sup> These estimates are for a conventional treatment configuration, as opposed to an ISEP design, and do not account for additional brine waste generated at the time of complete regeneration.

<sup>&</sup>lt;sup>5</sup> In comparison, a full-scale ISEP design, using its continuous regeneration approach, reportedly generates 1% brine waste in practice (Berlien, 2003).

One option for increasing the efficiency of regenerating strong-base polystyrenic resins may be heating the regenerant brine. Based on laboratory experiments, Tripp et al. (2003) found that perchlorate affinity decreases with increasing temperature for these resins. Based on these results and computer modeling, they concluded that heating the brine to 40 degrees centigrade would make the polystyrenic resins equivalent to the polyacrylic resins in terms of quantity of spent regenerant as a percentage of water treated. Heating to 60 degrees centigrade would make the polystyrenic resins more efficient than the polyacrylic resins.

Spent brine generated from the regeneration of perchlorate-loaded resins contains high concentrations of perchlorate. Spent brine might be expected to contain 1.5 to 3.0 mg/L of perchlorate (Case et al., 2004). The high perchlorate concentrations may mean that the waste must be treated prior to disposal. Regenerant treatment can have an additional advantage in that it may allow for reuse of the regenerant multiple times, reducing the quantity of waste generated. Appendix A discusses regenerant treatment in more detail.

Spent brine also contains high levels of the competing anions present in the influent water (nitrate, sulfate, and bicarbonate) (Batista et al., 2002; Berlien, 2003; MWH and University of Houston, 2003). The presence of bicarbonate, in particular, can have practical implications for waste management. A full-scale system in LaPuente, California, experienced scaling problems in the waste line due to elevated levels of carbonates and bicarbonates. The operator began adding hydrochloric acid to the line to lower pH from approximately 8.5 to approximately 7 and remedy the problem (Berlien, 2003). Case et al. (2004) also reported that regular maintenance to prevent the build-up of scale was needed in pilot tests.

# Regeneration of Selective Resins<sup>7</sup>

Selective resins are difficult to regenerate and are generally disposed of, rather than regenerated. As discussed above and shown in [REF\_Ref286411968 \h \\* MERGEFORMAT ], the majority of the full-scale ion exchange systems, including all but one of those using perchlorate-selective resin, operate on a throwaway basis. Because of the difficulty regenerating selective resins using conventional brine solutions, researchers at ORNL have developed a regeneration process for these resins. The process uses tetrachloroferrate anions (FeCl<sub>4</sub>-), formed in a ferric chloride solution in the presence of an excess amount of hydrochloric acid or chloride. Because it also has a strong relative affinity, the tetrachloroferrate anion readily displaces perchlorate from the resin. The tetrachloroferrate anion, however, also decomposes rapidly as the chloride concentration in solution decreases, converting to positively charged iron species. The positively charged iron species desorb from the resin by charge repulsion, leaving the resin in its original state with chloride as the counter anion. After tetrachloroferrate regeneration, the resin must be rinsed with dilute hydrochloride acid to wash sorbed ferric ions and excess regenerant off the bed (Gu et al., 2003; Gu et al., 2001). This rinse is necessary to ensure complete removal of the ferric

<sup>&</sup>lt;sup>6</sup> Data are for a conventional treatment system, Berlien (2003) estimates 2 to 4 mg/L of perchlorate for regenerant from an ISEP system.

<sup>&</sup>lt;sup>7</sup> Note that, because it is an emerging technology with as yet limited full-scale application, the novel tetrachloroferrate regeneration approach discussed in this section is not among the scenarios modeled for cost estimating purposes in Chapter [ REF Ref326047741 \r \h ].

ions to prevent precipitation of iron oxyhydroxides that may clog the bed when the resin is reused for treatment after regeneration (Gu et al., 2001).

Gu et al. (2003 and 2001) conducted laboratory-scale and small-scale field pilot tests of this regeneration technology for two types of resin: a commercial nitrate-selective resin and the perchlorate-selective bifunctional resin. They found that nearly complete regeneration could be achieved with as little as two BV of regenerant solution (Gu et al., 2003). They also found no significant deterioration in resin performance after repeated loading and regeneration cycles (Gu et al., 2001). On the basis that the perchlorate-selective bifunctional resin could treat 40,000 BV before breakthrough, they calculated spent regenerant generation at less than 0.005 percent of water treated (Gu et al., 2003). For the nitrate-selective resin, the comparable generation rate would be 0.014 percent (based on the 14,000 BV to breakthrough they reported for this resin). Note that these generation rates do not include the dilute acid rinse required following regeneration, which the investigators report requires 20 to 30 BV of less than 0.01 percent hydrochloric acid (Gu et al., 2001; Gu et al., 2003).

Like conventional brine solutions, spent tetrachloroferrate regenerant is expected to contain high concentrations of perchlorate. Gu et al. (2003) found that perchlorate concentration peaked at 6,000 mg/L in the regenerant from the nitrate-selective resin and 60,000 mg/L in the regenerant from the perchlorate-selective bifunctional resin. As for conventional brine solutions, these high concentrations indicate that treatment of the spent regenerant may be required before disposal or reuse. Batista et al. (2002) expect that the spent tetrachloroferrate regenerant would also contain other competing anions, low pH, and high concentrations of iron in the form of Fe<sup>+3</sup>, which could have implications for treatment. Appendix A discusses regenerant treatment. Gu et al. (2003) did not detect perchlorate in the dilute hydrochloric acid rinse solution. They suggest that this solution could be neutralized with dilute sodium hydroxide and readily mixed with the treated water or discharged to a publicly owned treatment works.

The ORNL researchers have examined this novel regeneration technology in small-scale field pilot tests (Gu et al., 2003 and 2005). Full-scale application reportedly began at Edwards Air Force Base in January 2003 (U.S. DoD, 2002), but no data are available on results at this installation. More recently, Lutes et al. (2010) reported on a field demonstration of the tetrachloroferrate regeneration approach using full-scale vessel. This study not only involved a larger scale application than the previous work by Gu et al. (2003 and 2005), but also slightly different parameters (e.g., more bed volumes of tetrachloroferrate, fewer bed volumes of dilute acid).

More recently, laboratory studies have examined the use of biological treatment to remove perchlorate from exhausted ion exchange resins. Although these studies suggest that bioregeneration has the potential to be effective for selective resins, the research has not yet progressed beyond batch experiments (Sharbat and Batista, 2013; Faccini et al., 2016).

#### Regeneration of Weak-base Resins

Batista et al. (2000; 2002) state that the primary advantage of weak-base resins is that they can potentially be regenerated using caustic solutions (sodium hydroxide or ammonium hydroxide), instead of conventional brine solutions. They suggest that this is an advantage because such solutions may be more amenable to biological treatment. In laboratory tests, they demonstrated

that, while a caustic solution was ineffective in regenerating a weak-base polystyrenic resin, weak-base polyacrylic resins could be regenerated easily using 1 percent sodium hydroxide, removing more than 76.5 percent of loaded perchlorate (Batista et al., 2000). Although Batista et al. (2002) suggest that caustic solutions used for regenerating weak-base resins would have high pH (greater than 11) and high ammonium concentration (if ammonium hydroxide is used), they have not published further data on the quantity or characteristics of these spent regenerant solutions.

Boodoo (2006) indicates that the weak-base resin piloted in Alabama could be regenerated using a caustic solution, followed by neutralization/protonation of the resin bed with an acid solution. The quantity of spent caustic solution was estimated to be less than 0.004 percent of water treated. Later tests showed that the weak-base resins were effectively regenerated using volumes of regenerant equal to less than 0.03 percent to less than 0.05 percent of water treated (U.S. DoD, 2007 and 2008a).

# 2.6 Critical Design Parameters

Critical design parameters that are specific to ion exchange systems removing perchlorate are:

- Resin type
- Vessel configuration (i.e., number of vessels in series)
- Empty bed contact time (EBCT)
- Resin bed life
- Surface loading rate
- Regeneration parameters.

[ REF \_Ref286650075 \h \\* MERGEFORMAT ] shows values for these parameters used in pilot- and full-scale systems removing perchlorate. Design data are available in the literature for only a few of the many full-scale conventional ion exchange applications. Therefore, much of the data presented in [ REF \_Ref286650075 \h \\* MERGEFORMAT ] are from pilot-scale tests or are for proposed scale-up units.<sup>8</sup>

The paragraphs below discuss each of the parameters listed [ REF \_Ref286650075 \h \\* MERGEFORMAT ] in more detail. Values for other ion exchange design parameters (e.g., resin density, resin expansion during backwash, resin loss during backwash and regeneration), while not specifically addressed in the literature reviewed here, are well documented for ion exchange treatment in general. EPA has no reason to expect a significant difference in these parameters for ion exchange systems treating perchlorate. This section provides a general discussion of the design parameters and the range of values reported in the literature for these parameters. Chapter [ REF \_Ref326047741 \r \h ] identifies the specific values for each parameter used in EPA's cost estimates.

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<sup>&</sup>lt;sup>8</sup> The data shown are for conventional (rather than ISEP) treatment configurations only. Because of its proprietary nature, available design data for ISEP are limited. Furthermore, facilities developed since 2004 are not using ISEP for perchlorate removal. Finally, data for ISEP systems would not be applicable in models that simulate conventional ion exchange configurations.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Critical Design Parameters for Ion Exchange Systems

Ion Exchange Critical Design Parameter	Value from the References				
Resin Type	See text.				
Vessels Configuration	Lead-lag configuration in all full-scale applications for which data are available (Lu, 2003; Siemens, 2009a; Siemens, 2009b; Fontana Water Company, 2010; Drago and Leserman, 2011).				
	Lead-lag configuration in pilot tests and field demonstrations (Boodoo, 2006; Lehman et al., 2008; U.S. DoD, 2008a; U.S. DoD, 2007; ITRC, 2008).				
	Gu et al. (1999) recommend lead-lag configuration in proposed scale-up unit.				
	Tripp et al. (2003) assume parallel operation in model plant designs for cost estimation, but recommend lead-lag design for nitrate-selective or perchlorate-selective resins.				
Empty Bed Contact	Conventional Resins:				
Time	6 minutes per vessel in full-scale application at the City of Riverside (Lu, 2003).				
	2.5 to 4.2 minutes per vessel in model plant designs used for cost estimation (Tripp et al., 2003).				
	1.5 minutes per column in field pilot tests (MWH and University of Houston, 2003; Lehman et al., 2008).				
	4 to 6 minutes per column in laboratory column tests (Batista et al., 2000).				
	Nitrate-selective and Perchlorate-selective Resins:				
	One resin manufacturer recommends 1.5 minutes per vessel for nitrate-selective resin in full-scale applications (Boodoo, F. Personal communication. March 2, 2006).				
	1 minute per column in small-scale field pilot tests of perchlorate-selective resins (Gu et al., 2003).				
	45 seconds per vessel in proposed scale-up unit for perchlorate-selective resins (Gu et al., 1999).				
	1.5 minutes per column in pilot tests of perchlorate-selective resins (Blute et al., 2006; Russell et al., 2008; Wu and Blute, 2010).				
Resin bed life	Dependent on resin type and competing anion concentrations. See text.				
Surface loading rate	Conventional Resins:				
	Maximum 9 to 15 gpm per square foot (gpm/ft²) in model plant designs used for cost estimation (Tripp et al., 2003).				
	19 gpm/ft² in field pilot tests (MWH and University of Houston, 2003).				
	9.7 gpm/ft² in pilot tests (U.S. DoD, 2008a).				
	Perchlorate-selective Resins:				
	40 to 50 gpm/ft² in proposed scale-up unit (Gu et al., 1999).				
	12 gpm/ft² in pilot tests (Blute et al., 2006; Russell et al., 2008; Wu and Blute, 2010).				

lon Exchange Critical Design Parameter	Value from the References		
Regenerant loading rate and application time	Strong-base Polyacrylic Resins:		
	3 to 6% sodium hydroxide at 25 lbs/ft³ for 45 minutes followed by 3 BV of rinse in field pilot tests (MWH and University of Houston, 2003).		
	30 lbs/ft³ in model plant designs used for cost estimation (Tripp et al., 2003).		
	6% sodium hydroxide at 25 lbs/ft³ in pilot tests (Lehman et al., 2008).		
	Strong-base Polystyrenic Resins:		
	6% sodium hydroxide at 35 lbs/ft³ for 63 minutes followed by 3 BV of rinse in field pilot tests (MWH and University of Houston, 2003).		
	36 lbs/ft³ for partial regeneration and 400 lbs/ft³ for full regeneration in model plant designs used for cost estimation (Tripp et al., 2003).		
	Perchlorate-selective Resins:		
	2 to 4 BV of tetrachloroferrate followed by 20 to 30 BV of dilute hydrochloric acid followed by rinse with water or dilute bicarbonate (no data on loading rates) in small-scale field pilot tests (Gu et al., 2003).		
	6 BV of tetrachloroferrate, 14 BV of dilute hydrochloric acid, 21 BV rinse with water/ dilute bicarbonate in field demonstration (Lutes et al., 2010).		

#### Resin Type

As discussed in Section [ REF \_Ref286657733 \r \h ], a variety of resin types have been tested for perchlorate removal. The selection of resin type will affect most other critical design parameter values. [ REF \_Ref286650075 \h \\* MERGEFORMAT ] and the paragraphs below present data for all major resin categories. As discussed in Section [ REF \_Ref286657611 \r \h ] and shown in [ REF \_Ref286411968 \h \\* MERGEFORMAT ], however, perchlorate-selective resin appears to have become the technology of choice for modern perchlorate ion exchange facilities when perchlorate is the only contaminant of concern. Thus, where possible, the discussion focuses on parameters specific to perchlorate-selective resins.

#### **Vessel Configuration**

Ion exchange vessels can be configured in series or in parallel. In a parallel configuration, one or more vessels are in use, while other vessels are being regenerated or are on standby (Clifford, 1999). Influent water to be treated is divided equally among the operational vessels. Systems set in parallel are generally used to increase throughput. For contaminants that are difficult to remove (such as perchlorate), however, a series configuration can be effective to achieve a greater resin bed life (Boodoo, 2003; Gu et al., 1999). A series configuration allows for operation of the first vessel to a later point on the breakthrough curve because the second vessel can capture the initial breakthrough concentrations from the first vessel, keeping the final treated water from the system below a specified target concentration. As discussed above, series configurations are also known as "lead-lag" designs. As shown in [ REF \_Ref286650075 \h \\* MERGEFORMAT ], series (lead-lag) operation is generally recommended for perchlorate removal.

#### **Empty Bed Contact Time**

EBCT is defined as the volume of resin, including voids, divided by the flow rate. The minimum EBCT required varies depending on the specific contaminant treated, the required contaminant

removal percentage, the type of resin used, and other influent water characteristics (e.g., the presence of competing chemical species). In general, the EBCT for ion exchange removal of conventional anions (e.g., sulfates, nitrates, arsenic) usually ranges between 1.5 and 7 minutes per vessel. As shown in [REF\_Ref286650075 \h \\* MERGEFORMAT], recommended EBCTs for perchlorate removal using conventional resins span this range, mostly falling in the middle to upper end of the range. Selective resins (particularly perchlorate-selective resins), however, remain effective at higher flow rates, which correspond to shorter EBCTs. For example, perchlorate-selective resins can be employed at flow rates of 0.5 to 4 BV per minute (Gu et al., 1999; Gu et al., 2001; Gu et al., 2003). Correspondingly, recommended EBCTs for perchlorate-selective resins shown in [REF\_Ref286650075 \h \\* MERGEFORMAT] are 1.5 minutes per vessel and less.

## **Surface Loading Rate**

Loading rate is the velocity of flow through the resin measured in units of flow rate per unit area (e.g., gpm/ft²). The surface area of the treatment pressure vessels must be selected to maintain loading rates within reasonable bounds. As shown in [ REF \_Ref286650075 \h \\* MERGEFORMAT ], perchlorate-selective resins may have the potential to remain effective at higher maximum loading rates than conventional resins, although the data remain somewhat uncertain.

#### Resin Bed Life

Section [ REF \_Ref286654638 \r \h ] provides a detailed discussion of resin capacity (or bed life), which varies depending on resin type and water quality. [ REF \_Ref286412546 \h \\* MERGEFORMAT ] in that section shows detailed data from the literature on resin life. The capacities presented in [ REF \_Ref286412546 \h \\* MERGEFORMAT ] can be extended by series (lead-lag) operation. On the other hand, these capacities can be limited by breakthrough or peaking of co-contaminants.

#### **Regeneration Parameters**

Regeneration parameters determine the concentration and quantity of chemicals (i.e., chloride brine or tetrachloroferrate solution) required to restore a resin's capacity to remove perchlorate. They also determine the quantity of waste regenerant generated. As discussed in Section [ REF \_Ref286653866 \r \h ], regeneration requirements depend on the type of resin used. [ REF \_Ref286650075 \h \\* MERGEFORMAT ] shows the available data on regeneration parameters that might be applicable if a system chose to regenerate its resin. As discussed in Section [ REF \_Ref286658589 \r \h ] and shown in [ REF \_Ref286411968 \h \\* MERGEFORMAT ], however, the majority of full-scale ion exchange systems operate on a throwaway basis.

# 3 Biological Treatment

# 3.1 Operating Principle

Biological treatment of perchlorate is the process by which bacteria are used to reduce perchlorate to chlorate, chloride, and oxygen. Biological treatment offers complete destruction of the perchlorate ion, eliminating the need for management of perchlorate-bearing waste streams. While there have been a wide variety of laboratory and pilot-scale tests exploring perchlorate treatment using bioreactors, the number of full-scale designs is still very limited.

The fundamental physical and chemical nature of perchlorate complicates the biological treatment process. Common reducing agents do not reduce perchlorate, and common cations do not precipitate it (Urbansky, 1998). Despite its strength as an oxidizing agent, the perchlorate ion is slow to react due to the presence of the highly-oxidized central halogen atom, chlorine (VII). This low reactivity, however, is a matter of kinetics rather that thermodynamics. Urbansky (1998) reports the standard half reactions for reductions to chloride (Eq 1) and chlorate (Eq 2) are favorable processes from a thermodynamic standpoint:

Eq 1 [EMBED Equation.3]
Eq 2 [EMBED Equation.3]

Therefore, the key to reducing perchlorate is finding the right catalyst. Coates et al. (2000) report that the scientific literature has evidence of the microbially-catalyzed reduction of chlorine oxyanions (a group to which perchlorate belongs) dating back over half a century. More recently, Coates et al. (2000), Logan (2001), and others have enumerated perchlorate-reducing bacteria (PRB) in a broad spectrum of environments nationwide, and demonstrated that the microbial reduction of perchlorate is a much more ubiquitous and diverse metabolism than previously considered.

According to Xu et al. (2003), researchers have isolated and characterized many PRB. These microorganisms are all facultative anaerobes and are capable of reducing both perchlorate and chlorate to chloride for energy and growth. Although reduction does not take place (or at least not very quickly) in the presence of a high concentration of dissolved oxygen (DO), most PRB isolates can use oxygen as a terminal electron acceptor. Many PRB partially or completely reduce nitrate. The presence of nitrate usually decreases the rate of perchlorate reduction (until the nitrate is depleted). Most PRB do not reduce sulfate, and none (thus far) use Fe(III), another common component of ground water, as an electron acceptor. Marqusee (2001) presented a summary of electron acceptor use in groundwater samples collected from several locations with varying levels of nitrate (<1.2 to 59 mg/L), perchlorate (9.8 to 666 mg/L), and sulfate (15 to 1,620 mg/L). Generally, the microbial consortia preferred these electron acceptors in the following order:

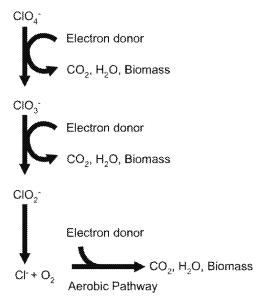
nitrate > perchlorate > sulfate

Researchers have isolated both heterotrophic and autotrophic PRB (Xu et al., 2003). Many studies have used acetate or ethanol as a single substrate (also referred to as the electron donor or "food") for heterotrophic perchlorate reduction; however, optimal substrate use is strain-

dependent. Many studies have also used supplemented nitrogen and phosphorus as necessary nutrients for the growth of PRB. There is no definitive information on what trace nutrients or metals are needed for growth. In one instance reported in Xu et al. (2003), researchers found iron, molybdenum, and selenium in purified perchlorate reductase. Chadhuri et al. (2002) established that the PRB *Dechlorosoma suillum* did not reduce perchlorate without the presence of molybdenum. Several field studies presented later in this chapter achieved perchlorate degradation simply through the addition of an oxidizable substrate (e.g. acetate or ethanol) and nitrogen and phosphorous.

[ REF \_Ref286664846 \h \\* MERGEFORMAT ] represents the three-step mechanism now widely accepted for bacterial respiration using perchlorate, which sequentially produces chlorate, chlorite, and chloride and oxygen (Xu et al., 2003). While researchers believe *perchlorate* reductase and chlorite dismutase are the central enzymes catalyzing the reactions, they are not sure if PRB use these enzymes exclusively or use a broader range of enzymes for perchlorate and chlorate reduction.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Biological Perchlorate Reduction Pathway



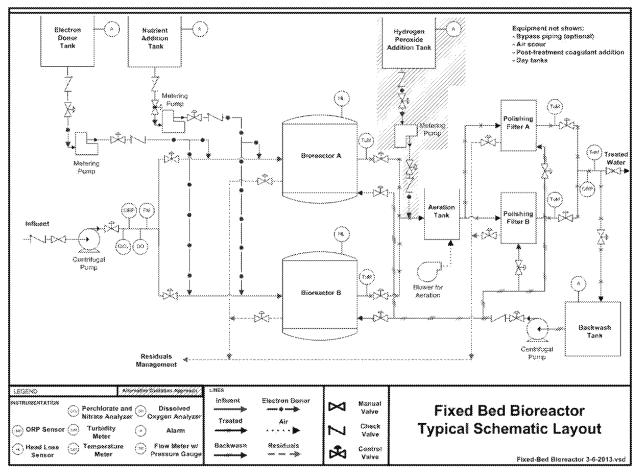
Biological treatment technologies that take advantage of the mechanism shown in [ REF Ref286664846 \h \\* MERGEFORMAT ] include:

- heterotrophic fixed bed (or packed bed) reactors
- fluidized bed reactors
- membrane biofilm reactors
- autotrophic hydrogen reactors
- continuously-stirred tank reactors
- *in situ* permeable biological barrier
- in situ electron donor delivery
- phytoremediation.

The most promising designs for biological treatment of perchlorate at drinking water facilities are those that operate either in a fixed bed or fluidized bed configuration. The California Department of Public Health (CDPH) also has identified membrane biofilm reactors as a "promising" technology for perchlorate treatment (Meyer, 2012). The first approved, full-scale membrane biofilm system for nitrate removal began operation in California in 2011 (Friese et al., 2013). Full-scale experience with the technology for perchlorate treatment, however, is limited. The literature indicates that *in situ* and other technologies are not currently used with the intent to create potable water supplies. Therefore, the remainder of this chapter focuses on the first two technologies listed above (fixed bed and fluidized bed reactors).

Both fixed bed and fluidized bed designs involve a media bed that provides a surface on which the PRB grows. The PRB can be initially introduced to the reactor with cell cultures, or the system can rely on natural populations of PRB. Drinking water systems typically rely on natural populations. Lab studies have used a variety of media in effectively reducing perchlorate, including granular activated carbon (GAC), anthracite, sand, and plastic media. Full-scale designs for perchlorate treatment have primarily used GAC. For fixed bed reactors, influent water is typically passed under pressure through a static media bed located in a vessel. An alternative fixed bed design is to use a gravity-fed concrete basin to hold the biologically active media. Fluidized bed bioreactor designs use vessels where flow, including a recycled portion, is pumped into the reactor at high rates in an up-flow design, fluidizing the media bed and allowing for more surface area for biomass growth. [ REF \_Ref286827225 \h \\* MERGEFORMAT ] and [ REF \_Ref329773247 \h \\* MERGEFORMAT ] provide schematic drawings for fixed bed and fluidized bed biological treatment, respectively.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Typical Schematic Layout for Fixed Bed Biological Treatment



#### Electron Nutrient - Bypass piping (optional) - Post-treatment coagulant addition Biomactor A mideilaG Fump Tank Bloreactor 8 Centrifusa .... Centrifugal LEGENO Recycle Valva Perchiorate and Dissolved Fluidized Bed Bioreactor Electron Dono Oxygen Analyze Biffuent Mitrata Anaiyxar Check Turbidity Typical Schematic Layout Air Alasm Treated Figur Meter w Residuais (%) Sackwast Pluidized-Sed Bioreactor Treatment 3-6-2013.vs

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Typical Schematic Layout for Fluidized Bed Biological Treatment

#### 3.2 Effectiveness for Perchlorate Removal

The State of California has identified fluidized bed biological treatment (along with ion exchange) as one of two BATs for achieving compliance with its standard for perchlorate in drinking water (California Code of Regulations, Title 22, Chapter 15, Section 64447.2). The literature contains substantial evidence of biologically-based technologies capable of reducing perchlorate to low levels in water. [REF\_Ref286666758 \h \\* MERGEFORMAT] summarizes the removal efficiencies reported in the literature. It shows that fixed and fluidized bed reactors have consistently achieved removal efficiencies greater than 90 percent, reducing perchlorate to levels that are usually below detection.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Perchlorate Effectiveness Results for Biological Treatment

Removal Efficiency	Resulting Concentration (µg/L)	Study Scale and Reactor Type	Other Analytes (mg/L)	Media / Electron Donor	Data Source(s)
>92%	<4	Bench-scale fixed bed	Sulfate (0 to 220)	GAC/ Acetate or ethanol	Brown et al., 2003
>94%	<4	Bench-scale fixed bed	Nitrate (4)	Sand, plastic media / Acetic acid	Case et al., 2004; Min et al., 2004
>99%	<4	Bench-scale fixed bed	None	Sand / Acetate	Kim and Logan, 2000
>99%	<4	Bench-scale fixed bed	Nitrate (0.02), sulfate (0.04)	Celite / Acetate	Losi et al., 2002
>98%	<3	Bench-scale fixed bed	Nitrate (13), sulfate (9.3 to16.8)	GAC / Acetic acid or proprietary carbohydrate solution	Upadhyaya et al., 2015
>93%	<5	Pilot-scale fixed bed <sup>1</sup>	Nitrate	GAC / Acetic acid	U.S. DoD, 2008b
92% to 99%	<4	Pilot-scale fixed bed <sup>2</sup>	Sulfate (140 to 250), Nitrate (6 to 29), DO (4 to 8)	GAC / Acetic acid	ITRC, 2008; Brown et al., 2005
>97%	<6	Bench-scale fluidized bed	Nitrate (13), sulfate (9.3 to 16.8)	GAC / Acetic acid or proprietary carbohydrate solution	Upadhyaya et al., 2015
>99%	220 to 280	Bench-scale fluidized bed	Nitrate (15.4), sulfate (12.5)	GAC / Acetate or proprietary glycerol solution	Kotlarz et al., 2016
>99%	<5	Bench-scale fluidized bed	Nitrate, metals, volatile organics	GAC / Acetic acid	Polk et al., 2001
>99%	<2	Bench-scale fluidized bed	Sulfate (5 to 10)	GAC, sand / Ethanol, methanol, or mix	Greene and Pitre, 2000
92 to 98%	<4	Pilot-scale fluidized bed <sup>3</sup>	Various	GAC / Ethanol	Harding ESE, 2001; Gilbert et al., 2001
>99%	<0.5	Pilot-scale fluidized bed <sup>4</sup>	Various	GAC / Acetic acid	U.S. DoD, 2009; Webster and Crowley, 2010; Webster and Crowley, 2016; Webster and Litchfield, 2017
>99%	350 to <4	Full-scale fluidized bed <sup>5</sup>	Nitrate (1.9), sulfate (300)	GAC / Acetic acid, ethanol	Polk et al., 2001
>99%	<4	Full-scale fluidized bed <sup>6</sup>	Not reported	GAC / Ethanol	Greene and Pitre, 2000

#### Notes:

- 1. Ten-month field test at Rialto Well #2 site in Rialto, California treating an average of 20.2 gpm
- 2. Six-month field test in Santa Clarita, California treating up to 4.5 gpm
- 3. Eight-month field test in Rancho Cordova, California, supplying water for potable use with a design flow of 1,800 gpm
- 4. Field test at Rialto Well #2 site in Rialto, California treating an average of 50 gpm
- 5. Facility at Longhorn Army Ammunition Plant in Karnak, Texas operating at 50 gpm
- 6. Aerojet facility in Rancho Cordova, California operating at 3,500 gpm

The two full-scale systems identified in [ REF \_Ref286666758 \h \\* MERGEFORMAT ] are perchlorate remediation projects. Treated water from these facilities is not used as drinking water. One of the pilot-scale studies in [ REF \_Ref286666758 \h \\* MERGEFORMAT ], however, was an eight-month field test that supplied potable water to local water companies (Harding ESE, 2001; Gilbert et al., 2001). Furthermore, the success of the demonstration studies in Rialto led to the design and installation of a full-scale fluidized bed system supplying drinking water to the West Valley Water District and the City of Rialto (Webster and Crowley, 2010; Webster and Crowley, 2016; Webster and Litchfield, 2017). Section [ REF \_Ref292953558 \r \h ] provides additional details regarding the two systems designed to produce municipal drinking water (the permanent full-scale Rialto and West Valley facility and the Aerojet demonstration). Section [ REF \_Ref292953621 \r \h ] discusses several other large-scale treatment systems.

## 3.2.1 Biological Treatment for Municipal Drinking Water Supply

Rialto Well #6 and West Valley Well #11, Rialto, California. As a result of the successful full-scale demonstration at Rialto Well #2 (see Section [ REF \_Ref292953621 \r \h ]), Envirogen installed a full-scale fluidized bed treatment system designed to supply drinking water from two other nearby wells to West Valley Water District and the City of Rialto. The system is sited on a former landfill and initially designed to treat 3 million gallons per day (MGD), with an ultimate capacity of 6 MGD so that water from additional wells might be treated in the future.

Envirogen completed construction in 2013 and the system underwent extensive testing before receiving its operating permit and beginning to produce drinking water in 2016. The system includes two 14-foot diameter, 24-foot tall bioreactor vessels followed by two 12-foot diameter, 24-foot tall aeration vessels. The bioreactors include a unique system of biomass separators at the surface of the vessels. The separators remove excess biomass that detaches from the media with air scour and agitation.

Additional post-treatment includes two multimedia filters with a dissolved air floatation system for removing solids from filter backwash, followed by chlorination. This additional post-treatment is designed to replicate the surface water treatment process that West Valley Water District operates at another location and satisfy the permit requirement that the biological treatment system meet the Enhanced Surface Water Treatment rule. The permit requirements also include instrumentation and controls (chlorine, pH, nitrate, sulfide, total organic carbon, and turbidity) to monitor performance (Webster and Crowley, 2016; Webster and Litchfield, 2017).

Aerojet Field Test, Rancho Cordova, California. Aerojet conducted an eight-month field test at its facility in Rancho Cordova, California (where the company also operates the full-scale fluidized bed-based system described in [ REF \_Ref292953621 \r \h ]) of a treatment system that included a fluidized bed biological reactor. The system removed perchlorate, nitrate, volatile organic compounds (VOCs), NDMA, and 1,4-dioxane from contaminated groundwater and supplied the treated water to local water companies for potable use (Harding ESE, 2001).

In addition to the fluidized bed reactor (designed to remove perchlorate and nitrate), the other components of the system were a multimedia filter (to remove biomass and GAC fines), an air stripper (for VOC removal), ultraviolet light/chemical oxidation (for other contaminant removal), liquid phase GAC adsorption (for other contaminant removal), disinfection (for potable water supply requirements), and clarification (for filter backwash) (Harding ESE, 2001).

The bioreactor component of the treatment system targeted perchlorate and nitrate. The total design flow rate of the fluidized bed reactor used during the field study was 1,800 gpm, which was the flow rate required to maintain the required fluidization of the GAC media. Bioreactor forward flow was variable depending on the desired recycle rate. For example, a forward flow rate of 1,200 gpm results in a recycle rate of 600 gpm (Harding ESE, 2001). To control fluidized bed level, a biofilm control system sheared excess biomass from the GAC and discharged cleaned GAC and biomass back into the reactor.

A panel of treatment experts convened by Aerojet to review results from the study concluded the following about the performance of the Aerojet Baldwin Park Operable Unit (BPOU) system (Gilbert et al., 2001):

- This combination of treatment processes removed all target chemicals below regulatory standards needed to meet potable water requirements.
- Each of the treatment processes met desired removal efficiencies in a reliable manner.
- The overall process was stable when the optimum ethanol dosage was maintained.

Investigators sampled water quality throughout the post-treatment process to evaluate the fate of excess biomass leaving the bioreactor. Specifically, they examined assimilable organic carbon, biodegradable dissolved organic carbon, and heterotrophic plate counts. Based on these analyses, they concluded that bacterial re-growth in the water distribution system would not be significant. The expert panel noted, however, that researchers did not solve the problem of cleaning excessive biomass from the GAC media. The episodic nature of the current cleaning process created problems of excessive biosolids loading on the subsequent filter that impacted water quality (Gilbert et al., 2001).

#### 3.2.2 Other Large-Scale Biological Treatment Systems

**Rialto Well #2, Rialto, California.** Basin Water Inc. and Carollo Engineers Inc. both installed bioreactors as part of a perchlorate treatability field study Rialto, California. These reactors treated water from Well #2, which had been abandoned because of perchlorate contamination (Webster and Litchfield, 2017). Carollo Engineers installed and operated two parallel fixed-bed bioreactors for nitrate and perchlorate removal in a study, conducted over a 10 month period starting in February 2007, with the goal of producing treated water that met all drinking water standards. The treatment process consisted of parallel packed bed bioreactors with a 2 foot diameter and 4.7 foot bed-depth, followed by hydrogen peroxide dosing and biofiltration. The final step of the treatment process included chlorine disinfection. The bioreactor was subjected to various perchlorate spiking tests and consistently was able to treat to below detection limits for perchlorate for spiking concentrations up to 930  $\mu$ g/L. The study showed that fixed-bed reactors, in combination with post-treatment processes, are a cost-effective way to produce potable water with perchlorate concentrations below detectable levels (U.S. DoD, 2008b).

Basin Water installed a full-size fluidized bed bioreactor at the same Rialto well location and tested the reactor under various operating conditions between 2007 and 2008. The purpose of the study was to test and validate the following items as they pertain to drinking water treatment (U.S. DoD, 2009; Webster and Crowley, 2010; Webster and Litchfield, 2017):

- ex situ bioremediation of nitrate and low concentrations of perchlorate though a fluidized bed reactor
- the short-term and long-term performance effects in allowing the system to be self-inoculated with incoming groundwater versus manually inoculating with a non-pathogenic microbial consortium
- short-term performance effects in the simulation of both a feed pump failure and an electrical shutdown
- the use of a post aeration vessel, multimedia filter, and GAC to produce to produce potablelike effluent water stream
- operational effectiveness of on-line nitrate and perchlorate analyzer systems
- long-term monitoring of system robustness and performance under steady-state and spiking perchlorate concentrations.

The Basin Water field study utilized acetic acid as the electron donor for a single fluidized bed reactor. The average flow rate into the reactor was 50 gpm. In steady-state operation, the system consistently treated perchlorate to less than 0.5 μg/L at varying influent concentrations and flow rates (Webster and Litchfield, 2017). Spiking studies showed that the maximum perchlorate concentration that could be consistently treated through the fluidized bed at a flow rate of 25 gpm was 4,000 μg/L of perchlorate, with 99.65 percent removal. The study also examined various system shut-down scenarios and proved that the fluidized bed reactor, in combination with other post-treatment processes, could treat perchlorate-contaminated groundwater to meet drinking water quality standards. The fluidized bed reactor system was able to effectively clean biosolids and maintain a consistent fluidized bed height, though the process was not described in detail (U.S. DoD, 2009). Operating costs were demonstrated to be \$125 to \$150 per acre foot treated. The success of the study led to the design and installation of the full-scale system producing drinking water from Rialto Well #6 and West Valley Well #11 (see Section [ REF Ref491163681 \r \h ]) (Webster and Crowley, 2010; Webster and Litchfield, 2017).

Aerojet Facility, Rancho Cordova, California. Aerojet installed four fluidized bed reactors with GAC media designed and supplied by Envirogen, Inc. Envirogen designed the system to treat up to 8 mg/L of perchlorate with a loading rate of 44 pounds per day per 1,000 cubic feet of reactor volume. Each reactor has a design capacity of 1,800 gpm. Since its installation in 1998, Aerojet has operated this system at about 3,500 gpm (less than 900 gpm per reactor), treating concentrations of about 3,500  $\mu$ g/L perchlorate to non-detect levels (less than 4  $\mu$ g/L). Aerojet uses ethanol as the electron donor, and they re-inject the treated water into an underlying aquifer. Envirogen selected GAC media (versus sand) on the basis of pilot-testing results (Greene and Pitre, 2000).

Longhorn Army Ammunitions Plant, Karnack, Texas. U.S. Filter/Envirex and Envirogen developed and supplied a full-scale 50 gpm fluidized bed reactor with GAC media and acetic acid/nutrients addition to treat perchlorate-contaminated groundwater. After start up and acclimation, the system treats perchlorate concentrations of up to 35 mg/L (16.5 mg/L on average), reducing them to at least the target goal of 350  $\mu$ g/L, and routinely to below the 5  $\mu$ g/L analytical reporting limit. The system discharges treated water to a nearby stream (Polk et al., 2001).

**Kerr-McGee and Pepcon Facilities, Henderson, Nevada.** Full-scale fluidized bed reactors were installed to remove perchlorate at these two nearby remediation sites. The system at the Kerr-McGee site began operation in 2004, replacing ion exchange systems, and was expanded in 2006 to remove approximately 3,000 pounds per day of perchlorate from groundwater. The system at the Pepcon site began operation in 2012, replacing an *in-situ* bioremediation system. It is designed to remove approximately 1,000 pounds per day of perchlorate (Roefer, 2013).

# 3.3 Raw Water Quality Considerations

As shown in [ REF\_Ref286666758 \h \\* MERGEFORMAT ], biological treatment remains effective even in the presence of certain co-occurring contaminants. Nitrate and sulfate were present in nearly all of the studies and did not appear to interfere with the removal efficiency of the process. Biological treatment also has been shown effective in the presence of metals, volatile organic compounds, and other contaminants including NDMA and 1,4-dioxane (Polk et al., 2001; U.S. DoD, 2000; Harding ESE, 2001).

Nevertheless, raw water quality plays a role in the design of a biological treatment system. In identifying design criteria for use in full-scale treatment plant designs, the Harding ESE (2001) authors included expected raw water dissolved oxygen, nitrate, perchlorate, and total phosphorous concentrations as necessary considerations, along with water temperature. In particular, temperature plays an important role in determining the rate of biomass growth. Electron donor dose requirements increase with decreasing temperature. At temperatures below 10 degrees C, biomass growth is inhibited and bioremediation becomes unfeasible (Dugan et al., 2009; Dugan, 2010a; Dugan, 2010b).

In addition, bacteria in bioreactors require macro- and micro-nutrients in order to grow and effectively reduce perchlorate. Thus, concentrations of these nutrients in the raw water are a consideration in bioreactor effectiveness. Macro-nutrients include phosphorous and nitrogen, and necessary micro-nutrients include sulfur and iron. While source water typically contains sufficient micro-nutrients, it often has insufficient amounts of phosphorous or nitrogen to allow for bacterial growth. As a result, some full-scale designs have required supplemental addition of one or both of these nutrients (Harding ESE, 2001; U.S. DoD, 2008b; U.S. DoD, 2009).

#### 3.4 Pre- and Post-Treatment Needs

Although the literature did not contain any studies that examine pre-treatment of source waters prior to biological treatment, certain groundwater conditions require pre-conditioning. For example, acidic ground water is not compatible with either robust microbial growth or common metallic system construction materials. In such cases, operators must raise the pH level prior to treatment. Consequently, some carbonates, sulfides, and oxides less soluble at neutral a pH might precipitate out and require filtration.

To produce drinking water from an impacted source, some form of multi-barrier system, beginning with biological treatment, may be necessary. For example, the treatment train used at the Aerojet BPOU to create potable water consists of seven different unit processes, as described in Section [ REF \_Ref286737077 \r \h ]. Furthermore, biological treatment itself results in the production of soluble microbial organic products that become part of the treated water. Some of this material is biodegradable, and the microorganisms (at least in the case of perchlorate and

nitrate reduction) tend to be the normal soil bacteria that are involved in the natural nitrogen cycle and common in all agricultural soils (Gilbert et al., 2001). In addition, the biological treatment process also depletes the levels of oxygen in the treated water. Therefore, post-treatment will typically be required for production of drinking water. Typical post-treatment processes include (Harding ESE, 2001; U.S. DoD, 2008b; Dordelmann, 2009; Webster and Crowley, 2016; Webster and Litchfield, 2017):

- reoxygenation or aeration for saturation with oxygen, using hydrogen peroxide addition or an aeration tank
- a polishing filter (using GAC or mixed media) for removal of turbidity, sulfide, and/or dissolved organic content, possibly including coagulant addition before filtration
- disinfection via ultraviolet light or chlorination.

# 3.5 Waste Generation and Residuals Management Needs

Because biological treatment offers complete destruction of the perchlorate ion, the technology does not generate a perchlorate-bearing waste stream. An active bioreactor, however, will have a continuous growth of biomass resulting from consumption of dissolved oxygen, nitrates, and perchlorate. In most bioreactor designs, excess biomass must be removed periodically. This removal results in one or more residual streams, the characteristics of which depend on the removal process used.

In fixed bed bioreactors, biomass removal typically is accomplished using a backwash process, which generates spent backwash water containing the excess biosolids (and some lost media). This backwash water is non-toxic and can typically be discharged to a local sewer. For facilities without the option of sewer disposal, a clarification and recycle process would be needed (U.S. DoD, 2008b). For fluidized bed reactors, one case study describes the use of a continuously operated separation device that uses supplied air to remove media and biomass from the top of the bed and direct it to a separation chamber. This arrangement was used in combination with an in-bed eductor to intermittently remove biomass growth from deeper in the bed. After treatment through an adsorption clarifier and multimedia filter, the study reports that the remaining residuals were "dilute enough that no special handling or pretreatment requirements should be necessary for most/all publicly-owned treatment works (POTWs) to accept" (U.S. DoD, 2009).

Downstream polishing through filtration (see Section [ REF \_Ref286739214 \r \h ]), when used as post-treatment, can also generate residual wastes in the form of backwash water and separated solids. The authors of the Harding ESE (2001) report suggest that clarifier solids could be discharged directly to sewer or filter pressed to reduce volume prior to ultimate disposal. The full-scale drinking water treatment facility in Rialto uses dissolved air floatation, followed by a sludge press, to treat backwash from post-treatment filtration (Webster and Litchfield, 2017). Backwash water from downstream polishing would be expected to have characteristics similar to water from direct backwash of a fixed bed reactor.

In addition, biological treatment can itself be a treatment technology for residuals from other perchlorate removal technologies, such as spent ion exchange regenerant or membrane reject. Appendix A discusses this use of biological treatment.

# 3.6 Critical Design Parameters

Critical design parameters for biological treatment systems removing perchlorate are:

- Support media type
- EBCT or hydraulic residence time (HRT)
- Bed expansion (for fluidized bed reactors)
- Electron donor type and dosage
- Nutrient addition
- Backwash and biomass separation design
- Recycle rate (for fluidized bed reactors)
- Post-treatment requirements.

The paragraphs below discuss each of these parameters in more detail. As noted in Section [ REF \_Ref286743169 \r \h ], fixed bed and fluidized bed reactors are the most promising approaches for drinking water treatment. Therefore, this section focuses on design parameters specific to these two types of biological treatment system. Three of the full-scale studies (Harding ESE, 2001; U.S. DoD, 2008b; U.S. DoD, 2009) identified critical design criteria for these types of reactors and were instrumental in guiding the discussion here. This section provides a general discussion of the design parameters and the range of values reported in the literature for these parameters. Chapter [ REF \_Ref326047741 \r \h ] identifies the specific values for each parameter used in EPA's cost estimates.

#### **Support Media Type**

As discussed in Section [ REF \_Ref286743169 \r \h ], in both fixed and fluidized bed reactors, PRB require a media surface on which to grow. As shown in [ REF \_Ref286666758 \h \\* MERGEFORMAT ], studies have used a variety of media in effectively reducing perchlorate, including GAC, anthracite, sand, and plastic media. Full-scale designs for perchlorate treatment, however, have primarily used GAC (Harding ESE, 2001; Greene and Pitre, 2000; Polk et al., 2001; U.S. DoD, 2008b; U.S. DoD, 2009).

#### **Empty Bed Contact Time or Hydraulic Residence Time**

EBCT is defined as the volume of support media divided by the flow rate. Minimum EBCT needed in order for perchlorate to be fully reduced is the primary design parameter in sizing fixed bed bioreactor vessels. For fluidized bed bioreactors, HRT is the more accurate term for the primary design parameter. HRT is the time required for treated water to move through the fluidized media bed. For larger flows, multiple bioreactors will be operated in parallel. Typical full-scale bioreactor designs have an EBCT or HRT in the range of 10 to 12 minutes for both fixed bed (U.S. DoD, 2008b) and fluidized bed reactors (Harding ESE, 2001).

#### **Bed Expansion**

For fluidized bed bioreactors, an additional important vessel sizing consideration is bed expansion. Target bed expansions are used to determine the height of the vessels. Peer review of EPA's biological treatment cost model (see Chapter [ REF \_Ref326047741 \r \h ]) provided information on typical bed expansion values. Typically, the vessel is filled with a fixed bed depth of media and initially fluidized to 40 to 50 percent. As the biomass grows, the fluidized media bed expands to 70 percent of the initial fixed bed depth. Biomass separation maintains expansion

at this target level. Additional space at the top of the vessel provides a safety factor to prevent fluidized media from exiting the reactor.

#### **Electron Donor Type and Dosage**

As discussed above, bioreactor designs require the presence of an electron donor (or substrate) for the reduction of perchlorate. For fixed bed bioreactors, electron donors are injected into the influent water prior to entering the bioreactor. In fluidized bed bioreactors, injection typically occurs in the recycle stream. As shown in [REF\_Ref286666758 \h \\* MERGEFORMAT ], a wide variety of electron donors have been tested, including acetate, acetic acid, lactate, ethanol, methanol, carbohydrate by-product, hydrogen, propane, and proprietary glycerol- or carbohydrate-based solutions. Full-scale designs for perchlorate treatment, however, have typically used acetic acid or ethanol (Harding ESE, 2001; U.S. DoD, 2008b; U.S. DoD, 2009; Greene and Pitre, 2000; Polk et al., 2001).

Determining the correct electron donor dose is critical in the effectiveness of perchlorate reduction in the bioreactor. Since oxygen and nitrates will be reduced prior to perchlorate reduction, the electron donor dose must be large enough to fully reduce all three. However, the dose cannot be too large or sulfides might form and be present in the effluent along with excess organic carbon, requiring additional post-treatment (Harding ESE, 2001).

Electron donor dose is dependent on site-specific conditions, including raw water characteristics. Therefore, determining the dose requirements for the electron donor typically requires pilot study tests, along with stoichiometric and thermodynamic calculations. The following site-specific relationships were developed for two full-scale treatment designs using ethanol and acetic acid:

$$C_e = 0.903 \text{ O}_2 + 2.229 \text{ NO}_s^-\text{N} + 0.581 \text{ ClO}_4^-$$
(Harding ESE, 2001)
$$C_{aa} = O_2 + 2.86 \text{ NO}_s^-\text{-N} + 0.64 \text{ ClO}_4^-$$
(U.S. DoD, 2008b)

where:

 $C_e$  = required ethanol concentration (mg/l)

 $C_{aa}$  = required acetic acid concentration (mg/l)

NO<sub>3</sub>-N = influent nitrate-nitrogen concentration (mg/l)

 $O_2$  = influent DO concentration (mg/l)

 $ClO_4$  = influent perchlorate concentration (mg/l)

As can be seen in these stoichiometric equations, influent concentrations of oxygen, nitratenitrogen, and perchlorate need to be available to determine the electron dose. Tests have proven that the electron donor dose is dependent on temperature and that decreasing the water temperature will result in an increase in necessary electron donor dose. However, these small changes are in the range of 3 percent change of dose for a 5 degrees C change (Harding ESE, 2001).

#### **Nutrient Addition**

As discussed in Section [ REF \_Ref286744617 \r \h ], bacteria in bioreactors require nutrients. Although these nutrients are sometimes present in source water, full-scale designs have required addition of macro-nutrient such as nitrogen and/or phosphorous. While there are a number of methods for adding nitrogen and phosphorous to the influent, the typical options are addition of ammonium chloride for supplemental nitrogen addition and/or addition of phosphoric acid for supplemental phosphorous (Harding ESE, 2001; U.S. DoD, 2008b).

## **Backwash and Biomass Separation Design**

As discussed in Section [ REF \_Ref286744911 \r \h \\* MERGEFORMAT ], bioreactors require periodic removal of accumulated biomass. Removal of the biomass is achieved in different ways depending on the bioreactor design type. For fixed bed bioreactors, an air scour is used followed by water flush to remove biomass from the media. A backwash basin and pump is needed to supply the water for backwashing. This backwash design is similar to that used in other treatment technologies, such as GAC and greensand filtration. Typical fixed bed bioreactors have a backwash interval in the range of 17 to 24 hours, determined based on target head across the media bed (U.S. DoD, 2008b).

For fluidized bed bioreactors, the expanded bed volume within the fluidized bed reactor will continue to expand as biomass accumulates. This expansion occurs because the specific density of the media plus biomass is less than that of the media alone. It becomes necessary to dislodge biomass growth from the media, thus increasing the specific density and decreasing the fluidized bed volume. Effective methods and equipment for biomass separation include mechanical separation of biomass using eductors or using air scour to sheer the biomass from the media. The amount of air required for biomass separation is around 0.3 cubic feet per hour per gpm of treated water (U.S. DoD, 2009).

# **Recycle Rate**

Designs of fluidized bed reactors use high pumping rates to keep the media in the bioreactor fluidized. A recycle stream is typically used to provide enough water to satisfy these pumping rates. Since feed pumping rates also determine the expansion height of the media bed, the recycle rate must be limited so that the feed pumping rate doesn't expand the fluidized media past the target height. A typical recycle rate, based on peer review comments on EPA's biological treatment cost model (see Chapter [ REF Ref326047741 \r \h ]), is 50 percent.

#### **Post-treatment Requirements**

Section [ REF\_Ref350938612 \r \h ] identifies post-treatment processes typically required for biological treatment, which can include reoxygenation, filtration, and disinfection.

# 4 Membrane Technologies

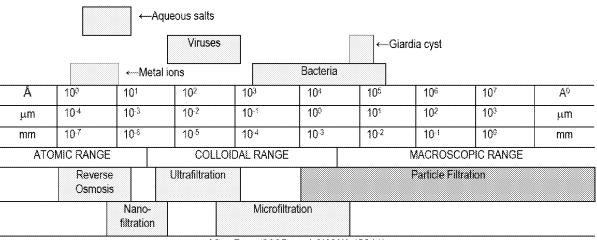
# 4.1 Operating Principle

Membrane filtration processes physically remove perchlorate ions from drinking water. This technique does not destroy the perchlorate ion and, therefore, creates a subsequent need for disposal or treatment of perchlorate-contaminated waste. Membrane filtration technologies evaluated for perchlorate treatment include reverse osmosis (RO), nanofiltration (NF), and ultrafiltration (UF).

These processes separate a solute such as perchlorate ions from a solution by forcing the solvent to flow through a membrane at pressure. RO depends on applying high pressures across the membrane in the range of roughly 100 to 1,000 pounds per square inch gauge (psig) in order to overcome the osmotic pressure differential between the saline feed and product waters. The NF process uses pressures in the range of 75 to 150 psig, while pressures for UF typically range from 3 to 40 psig (USEPA, 2003).

In all three processes, the membrane is semi-permeable, transporting different molecular species at different rates. Water and low-molecular weight solutes pass through the membrane and are removed as permeate, or filtrate. Dissolved and suspended solids are rejected by the membrane. Along with a portion of the feed water, these solids are removed as concentrate, or reject. The size range of the rejected solids varies by the type of membrane used, as shown in [REF Ref329764419 \h \\* MERGEFORMAT ].

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Particle Sizes and Membrane Process Ranges



After Dow (2005) and AWWA (2011)

Membranes may remove ions from feed water by a sieving action (called steric exclusion), or by electrostatic repulsion of ions from the charged membrane surface. RO membranes, which have an effective pore size of 0.001 microns or less, primarily remove perchlorate by the steric mechanism. UF membranes, with a pore size of roughly 0.01 to 0.1 microns, remove perchlorate primarily by electrostatic repulsion. NF membranes have effective pore sizes of roughly 0.001 to 0.01 microns. Various NF membranes may operate by both mechanisms to varying degrees

(Yoon et al., 2004; USEPA, 2005; Nam et al., 2005; Yoon et al., 2005a; Yoon et al., 2005b; Amy et al., 2006).

As discussed below in Section [ REF \_Ref329764874 \r \h ], of the membrane processes, RO has shown the most promise for removing perchlorate. Therefore, this chapter focuses primarily on the technical details of the RO treatment process. It discusses UF and NF mainly in terms of the available data on their effectiveness, with less discussion of their operating principle and relevant design parameters. Note, however, that practical operation of an NF system is very similar to that of an RO system.

For municipal drinking water treatment, RO membranes are most often used in a spiral-wound configuration. A spiral wound membrane consists of several membrane envelopes, layered with feed spacers and rolled together in a spiral around a central permeate collection tube. Each envelope consists of a flat membrane sheet folded in half over a porous membrane permeate carrier, and glued on the remaining three sides to completely enclose the carrier. The envelopes are connected to a central permeate collection tube. After the envelopes and feed spacers are rolled around the tube, the assembly is enclosed in a shell to form a membrane element. Different RO elements are manufactured for different scenarios, including seawater desalination (seawater RO), treatment of brackish water with dissolved solids roughly in the range of thousands of mg/L (brackish water RO), and treatment of less saline water (low-pressure RO). Elements that are intended for higher feed salinity have smaller effective pore sizes. They therefore offer higher rejection of dissolved ions, and require higher pressures for operation.

Multiple RO or NF elements are placed within a pressure vessel. To achieve the target removal efficiency and water recovery, these pressure vessels often are arranged in sequential stages, typically up to three depending on the recovery to be achieved (AWWA/ASCE, 2005; Dow, 2005). When multiple stages are used, the number of pressure vessels decreases from stage to stage. Permeate or finished water is collected from each pressure vessel. The concentrate from the first membrane stage serves as the feed to the second and the concentrate from the second stage serves as the feed to the third. Consequently, each successive stage of the process increases the total system recovery (Jacangelo et al., 1998). As the feed water travels through the membrane system and becomes more concentrated, its osmotic pressure increases. The feed pressure must overcome this osmotic pressure. The final concentration in the concentrate water therefore has a major effect on the required feed pressure and energy use.

The membrane stages in combination make up an RO treatment train. A treatment system may have multiple trains. [REF\_Ref286827747 \h \\* MERGEFORMAT] provides a schematic drawing for an RO treatment facility; each rectangular box within a train represents a pressure vessel that contains multiple membrane elements. An NF treatment facility would be nearly identical, with the primary difference being the type of membranes used and the operating pressures.

# Pretrestment Antificaciant System Trans Market From RO Saleds From RO Saled

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Typical Schematic Layout for a Reverse Osmosis (or Nanofiltration) Treatment Facility

#### 4.2 Effectiveness for Perchlorate Removal

Although the literature pertaining to perchlorate removal using membrane technologies is limited, pilot and bench-scale studies have demonstrated that perchlorate can be substantially removed by RO. The studies also demonstrate widely varying removal of perchlorate with NF and UF processes, and have investigated favorable conditions for its removal (Liang et al., 1998; Yoon et al., 2002; Yoon et al., 2003; Yoon et al., 2004; Nam et al., 2005; Yoon et al., 2005a; Yoon et al., 2015). There is no large-scale demonstration study of membrane use for perchlorate removal.

Pilot-scale treatability work at the Metropolitan Water District of Southern California showed that NF and RO membranes consistently removed greater than 80 percent of the perchlorate (up to 98 percent for RO and 92 percent for NF) depending on influent concentration (Liang et al., 1998). Recycling 50 percent of the concentrate had no effect on overall perchlorate rejection. [ REF\_Ref286757715 \h \\* MERGEFORMAT] summarizes effectiveness results for this pilot-scale work, along with results from additional, smaller scale bench studies.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Perchlorate Effectiveness Results for Membranes

Technology/Source	Removal Efficiency	Raw Water Concentration	Location and Source Water	Study Scale
RO and NF (Liang et al., 1998)	RO up to 98% NF up to 92%	20 to 2,000 µg/L (some trials used perchlorate-spiked source water)	Metropolitan Water District of Southern California, La Verne Treatment Plant, CA; Pretreated Colorado River Water	Pilot study (12 gpm)
Surfactant modified UF (Yoon et al., 2003)	Up to 80%	100 µg/L (perchlorate-spiked)	Synthetic water and a blend of Colorado River Water and State Project Water from the Metropolitan Water District, CA	Bench study (225 milliliters per minute)
NF and UF (Yoon et al., 2002)	Up to 75%	100 μg/L (perchlorate-spiked)	Synthetic water with pure component perchlorate, also combined with other salts	Bench study (no flow given)
NF and UF (Yoon et al., 2004)	NF up to 80% (natural water) or 89% (synthetic water) UF up to 5% (natural water) or 66% (synthetic water)	100 µg/L (perchlorate-spiked)	Synthetic water and Colorado River Water from the Metropolitan Water District, CA, spiked with perchlorate	Bench study (100 to 225 milliliters per minute)
RO and NF (Nam et al., 2005)	RO up to 95% NF up to 70%	100 µg/L (perchlorate-spiked)	Ground waters from the Castaic Lake Water Agency, CA	Bench study (no flow given)
RO (USEPA, 2005)	From 125–2,000 µg/L to 5–80 µg/L	125 to 2,000 µg/L	Unspecified perchlorate- contaminated ground water	Bench study (no flow given)
RO and NF (Yoon et al., 2005a)	RO up to 95% NF up to 55%	100 μg/L (perchlorate-spiked)	Blend of Colorado River Water and State Project Water from the Metropolitan Water District, CA, spiked with perchlorate	Bench study (20 milliliters per minute)
RO, NF, and UF (Yoon et al., 2005b)	RO up to 95% NF up to 78% UF up to 29%	100 µg/L (perchlorate-spiked)	Synthetic water	Bench study (no flow given)
RO, NF, and surface modified NF (Sanyal et al., 2015)	RO up to 95.8% NF up to 70.1% Surface modified NF up to 93%	10,000 µg/L (perchlorate-spiked)	Perchlorate-spiked deionized water	Bench study (0.26 gpm)

Bench-scale studies show the effects of steric/size exclusion and electrostatic exclusion on perchlorate transport through membranes to varying degrees. RO, while removing perchlorate,

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also removes most other salts, requires high operating pressures, and is prone to significant flux decline. Membrane processes that operate at lower pressures, such as NF or UF, may be effective for perchlorate removal through selectivity based on size and/or charge. However, bench studies show significant variability in these membranes' ability to remove perchlorate, depending on other constituents of the source water. See Section [ REF \_Ref292709479 \r \h ] for further discussion. One bench study modified commercial NF membranes using layer-by-layer surface deposition of polyelectrolytes. This study showed that the modified NF membranes could achieve perchlorate removal nearly equal to that of RO membranes. The study, however, did not examine the effect of differing source water quality on the membranes and research on the modified membranes does not yet appear to have progressed beyond the lab (Sanyal et al., 2015).

# 4.3 Raw Water Quality Considerations

High levels of alkaline earth cations ( $Ca^{2+}$  or  $Mg^{2+}$ ) can cause membrane scaling (Yoon et al., 2003), leading to a decline in product water flux. One study showed that calcium carbonate scaling was also associated with a decline in perchlorate rejection, likely because the scale reduced the surface charge of the membrane (Yoon et al., 2005b). Other substances, such as silica, may also cause flux decline; however, there are no studies of the resulting effect on perchlorate rejection.

Membrane fouling may be reduced either by reducing the pH of the feed water or by adding an antiscalant chemical. However, for membranes that reject perchlorate electrostatically (primarily NF and UF membranes), studies of several synthetic waters show that a reduced feed pH reduces the rejection of perchlorate (Yoon et al., 2004; Yoon et al., 2005a; Yoon et al., 2005b). The lower pH has been shown to diminish the negative surface charge of the membranes, inhibiting the electrostatic rejection mechanism. One study (Yoon et al., 2005b) demonstrated that a phosphonate-based antiscalant improved both product water flux and perchlorate rejection. In these studies, perchlorate rejection by RO membranes was much less sensitive to the feed water pH.

The same studies demonstrated that a high concentration of other ions, particularly divalent cations, in the membrane feed water can reduce perchlorate rejection. Again, the studies attributed the reduced rejection to a diminished membrane surface charge. One study that included one natural water and several synthetic waters (Yoon et al., 2004) found that the natural water had worse perchlorate rejection than the most similar synthetic water for NF and UF membranes.

#### 4.4 Pre- and Post-Treatment Needs

In general, pretreatment requirements for membrane technologies depend on influent water quality as well as the type of membrane used. RO and NF membranes are often used after media filtration, or more recently, after UF or microfiltration membranes. Membrane filtration processes often include a prescreen or cartridge filter to remove sediment that could damage the membranes. RO and NF membranes often require pH adjustment or antiscalant.

The pilot study of RO and NF membrane elements (Liang et al., 1998) included prechlorination and conventional filtration (rapid mix, flocculation, sedimentation, and filtration). Pretreatment requirements, however, typically are independent of the specific contaminant targeted for

removal. Calculations such as the silt density index (SDI), found in ASTM standard D3739-94, can provide insight into the fouling problems that are inherent in any membrane system. SDI measures the fouling potential of suspended solids. Manufacturers typically specify maximum SDIs of 3 to 5 for RO and NF elements. In addition, it is important to model and conduct pilot studies to assess the potential for fouling from substances such as calcium carbonate, silica, calcium fluoride, barium sulfate, calcium sulfate, strontium sulfate, and calcium phosphate. The Langelier saturation index (LSI), described in ASTM standard D4189-94, characterizes the potential for CaCO<sub>3</sub> scaling. The LSI is used to indicate the tendency of water to precipitate, dissolve, or be in equilibrium with calcium carbonate, and what pH change is required to bring the water back to equilibrium. The scaling potential of other substances may be determined from a saturation calculation.

Although the perchlorate literature does not address post-treatment requirements, the permeate from RO filtration is essentially deionized water, and generally requires post treatment for corrosion control before it enters a distribution system (AWWA/ASCE, 2005). In other drinking water treatment applications using groundwater, the permeate is often blended with untreated water to produce a less corrosive finished water. If the source water has a sufficiently low concentration of perchlorate and other contaminants, higher rates of blending will be possible, likely reducing post-treatment requirements.

# 4.5 Waste Generation and Residuals Management Needs

Membrane filtration produces a waste stream called the concentrate (or reject). This waste stream contains all removed dissolved and suspended solids, and must be further treated and/or disposed of. Membrane system designs generally set a recovery rate (i.e., the ratio of permeate to feed flow) based on the scaling potential of the resulting concentrate water. The presence of a particular target contaminant has little or no effect on the selected recovery rate. Therefore, it is likely that the concentrate flow would represent a substantial share of influent flows. In other applications, concentrate flows can account for 15 to 30 percent of influent, which implies a fairly large perchlorate-contaminated waste stream for subsequent treatment or disposal.

In general, full-scale RO systems handle concentrate using surface water discharge or discharge to sanitary sewer, with a small number using deep well injection, evaporation ponds, or spray irrigation (U.S. DoI, 2001). The large volume of residuals is a well-known obstacle to adoption of RO technology. In the case of perchlorate removal by centralized treatment plants, the high perchlorate concentration in the residuals might limit the disposal options or require additional treatment prior to disposal, depending on state and local discharge regulations. Studies of treatment of perchlorate-bearing RO residuals are limited to a few laboratory-scale studies. These include biological (Giblin et al., 2002) and thermal treatment (ARA, 2000) of RO concentrate, discussed in more detail in Appendix A.

In addition, periodic cleaning of the membrane system is necessary to recover productivity lost to fouling. This cleaning may include cycles of acid and caustic wash, depending on the nature of the fouling. Since the spent cleaning solution is generated infrequently and in small amounts, it is typically diluted by and handled with the concentrate.

# 4.6 Critical Design Parameters

As discussed in Section [ REF \_Ref329764874 \r \h ], pilot and bench-scale studies have demonstrated that perchlorate can be substantially removed by RO. The studies demonstrate widely varying removal of perchlorate with NF and UF processes. Therefore, this section focuses on critical design parameters for RO. For comparison, it notes available data for NF, but does not discuss UF.

Critical design parameters for RO systems removing perchlorate are:

- Feed water quality
- Membrane type and feed water pressure
- Recovery rate
- Flux rate
- Pretreatment requirements.

[ REF \_Ref286758013 \h \\* MERGEFORMAT ] shows design information from the pilot-scale work performed at the Metropolitan Water District of Southern California, La Verne Treatment Plant, which used RO and NF to remove perchlorate from pretreated Colorado River Water (Liang et al., 1998). The paragraphs below discuss each of the parameters listed above in more detail. Values for other RO design parameters (e.g., cleaning procedures, residuals discharge options), while not specifically addressed in the literature reviewed here, are well documented for RO treatment in general. EPA has no reason to expect a significant difference in these parameters for RO systems treating perchlorate. This section provides a general discussion of the design parameters and the range of values reported in the literature for these parameters. Chapter [ REF \_Ref326047741 \r \h ] identifies the specific values for each parameter used in EPA's cost estimates.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Critical Design Parameters for Reverse Osmosis (and Nanofiltration)

Membrane Filtration Critical Design Parameter	Value from the Reference (Liang et al., 1998)				
Feed water quality	TOC = 2.40 – 3.50 mg/L UV <sup>254</sup> = 0.024 – 0.032 /cm Conductivity = 969 – 1030 micro-ohm/cm Temperature = 26°C pH = 8.09 – 8.24 Turbidity = 0.12 – 0.18 NTU Particle count = 113 – 1590 /milliliter				
Membrane type	One membrane element, 4-inch diameter, 40 inches long, thin-film composite with negative surface charge  RO 72 square foot active area, specific flux 0.14 gallons per day per square foot (gpd/ft² or gfd) per pound per square inch (gfd/psi)  NF 82 square foot active area, specific flux 0.17 gfd/psi, molecular weight cutoff 300 Daltons				
Average feed water pressure	RO 106 psig NF 87 psig				
Recovery rate	No data is given.				
Flux rate	15 gfd for both RO and NF				
Pretreatment	The raw water was treated via prechlorination, rapid mix, flocculation, sedimentation, and filtration prior to passing through the membranes. The prechlorination dose was adjusted to maintain an effluent free-chlorine residual of 0.5 to 1.0 mg/L. The coagulation step used 5 mg/L alum and 1 mg/L poly-DAD/MAC polymer. Six column-type filters were used: four dual-media filters with 20 inches of anthracite coal and 8 inches of silica sand; and two tri-media filters with 8 inches of silica sand and 3 inches of ilmenite. The filter effluent was passed through a dechlorination unit.				

## **Feed Water Quality**

As discussed in Section [ REF \_Ref329766889 \r \h ], feed water quality can determine pretreatment and cleaning requirements. Furthermore, as discussed below, it affects the values achievable for other relevant design parameters, such as recovery and flux rate. Finally, higher levels of total dissolved solids correspond to higher osmotic pressure in the membrane concentrate, and thus increase energy requirements. Feed water quality parameters that are crucial include temperature, pH, SDI, total dissolved solids, and concentrations of ions that can lead to the oversaturation of scaling salts, such as those listed in Section [ REF \_Ref329766889 \r \h ].

## **Membrane Type and Feed Water Pressure**

Membrane elements from different manufacturers, and different elements from the same manufacturer, may have widely varying water and ion permeabilities. Effective pore size and maximum feed pressure determine whether a membrane is characterized as RO or NF. As discussed in Section [ REF \_Ref329767986 \r \h ], RO membranes have pore sizes of 0.001 microns or less and operate at pressures in the range of roughly 100 to 1,000 psig. NF membranes have pore sizes of roughly 0.001 to 0.01 microns and use pressures in the range of 75 to 150 psig. Membrane elements also are characterized by their diameter (usually 4 to 18 inches),

active area (in square feet), and specific flux rate (measured in gfd per psi of net driving pressure<sup>9</sup>). Other relevant operating specifications include maximum recovery in one element, minimum concentrate flow, maximum feed SDI, minimum operating pH, maximum operating temperature, and maximum pressure drop permitted in a single element and a complete pressure vessel.

[ REF \_Ref286758013 \h \\* MERGEFORMAT ] shows data on the specific membranes tested for perchlorate by Liang et al. (1998). The RO membranes (which showed higher perchlorate removal efficiency than the NF membranes) operated at the low end of the typical range for that technology (106 psig).

## **Recovery Rate**

As discussed above, RO produces a permeate flow (water with most dissolved solids removed) and a concentrate flow (residual water rejected by the membrane). The recovery rate is the percentage of the influent flow that is recovered as permeate. Increasing the recovery rate will increase the concentration of dissolved solids in the membrane reject water, and will thus increase the required feed pressure and the potential for membrane scaling. Thus, the achievable recovery rate depends on the quality of the source water as well as the pretreatment of the water (AWWA/ASCE, 2005), and systems with high levels of total dissolved solids in their feed water will typically operate at lower recovery rates than systems with lower levels.

For a given membrane and feed water, a higher recovery rate will require the use of more elements in series. The model accomplishes this by increasing the number of elements per pressure vessel and/or by increasing the number of stages in the system. For NF membrane elements, the target recovery will typically be between 80 and 90 percent. For RO elements, the target recovery will typically be between 50 and 85 percent. Although Liang et al. (1998) did not report recovery rates achieved in their pilot studies, recovery rate is driven by overall feed water quality, not the specific contaminant being targeted. Therefore, EPA has no reason to expect a significant difference in recovery rate for RO systems treating perchlorate from the values typically documented for RO systems in general.

#### Flux Rate

The flux of the system is the rate of permeate water per unit of membrane area, typically measured in gfd. While each stage of a membrane system will have a different flux, the average flux over all elements is a fundamental design parameter. In general, the higher the quality of the feed water, the higher the flux that may be achieved. Operating with excessively high flux, however, leads to fouling of the membrane elements. Depending on the nature of the fouling, it may be reversed by cleaning, or may require replacement of the elements.

For many ground waters, systems can operate successfully with fluxes between 16 and 20 gfd. Surface waters require lower fluxes, typically between 12 and 17 gfd, depending on the SDI of

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<sup>&</sup>lt;sup>9</sup> Net driving pressure is equal to the feed pressure at a particular point in the pressure vessel minus the osmotic pressure of the feed water.

the source water. Pretreatment will usually permit the use of a higher flux. The pilot studies by Liang et al. (1998) used a flux rate of 15 gfd, in the typical range for surface water.

#### **Pretreatment Requirements**

As discussed in Section [ REF \_Ref329766889 \r \h ], to reduce fouling of the membrane, some type of pretreatment is usually required. The pilot studies by Liang et al. (1998) used extensive pretreatment, the equivalent of conventional filtration. Conventional filtration, however, would likely already be present at surface water sources that require additional treatment to remove perchlorate. The types of pretreatment that would more likely need to be added to an existing treatment train for implementation of RO include those identified in Section [ REF \_Ref329766889 \r \h ], such as cartridge filtration and acid and/or antiscalant addition.

# 5 Point-of-Use Treatment

# 5.1 Operating Principle

A POU device uses a miniaturized version of a centralized treatment process to meet water quality standards for consumption at individual taps (e.g., a kitchen sink). When a system installs, controls (i.e., owns), and maintains POU devices at all customer locations where water is consumed (e.g., residences), it can forego centralized treatment (USEPA, 2006b). Because POU devices treat a small fraction of the water delivered by a system, a compliance program that relies on POU devices may be more cost-effective for smaller systems.

For perchlorate removal, the NSF Joint Committee on Drinking Water Treatment Units has added a protocol to NSF/American National Standards Institute (ANSI) Standard 58: Reverse Osmosis Drinking Water Treatment Systems that requires an RO unit to be able to reduce perchlorate from a challenge level of 130 µg/L to a target level of 4 µg/L (NSF, 2004). Several organizations (e.g., NSF International, Underwriters Laboratories, Water Quality Association) provide third-party testing and certification that POU devices meet drinking water treatment standards. There are no perchlorate certification standards for other types of POU devices such as those using ion exchange media. Therefore, the discussion in this section focuses on POU RO devices.

The operating principle for POU RO devices is the same as centralized RO: steric exclusion and electrostatic repulsion of ions from the charged membrane surface. In addition to an RO membrane for dissolved ion removal, POU RO devices often have a sediment pre-filter and a carbon filter in front of the RO membrane, a 3- to 5-gallon treated water storage tank, and a carbon filter between the tank and the tap.

To meet a perchlorate drinking water standard, a system would need to purchase, install, and maintain certified POU RO devices for all customers. Usually, a system would install a single POU RO device at the kitchen tap for each residential customer. Nonresidential customers might require multiple devices (e.g., for drinking fountains). Installation requires retrofitting the device into existing plumbing fixtures (e.g., tapping into the water supply line to insert a treated water line with a dedicated tap and adding a wastewater connection for the RO membrane concentrate or reject). Maintenance primarily consists of filter replacement, often on a fixed schedule that varies by filter type. Monitoring water quality at individual treated water taps will also be necessary to demonstrate compliance with a perchlorate drinking water standard.

#### 5.2 Effectiveness for Perchlorate Removal

There are no perchlorate removal case studies that use POU RO. Nevertheless, bench-scale and pilot testing indicates that RO membranes should be able to effectively remove perchlorate ions. Furthermore, devices certified under NSF/ANSI Standard 58 have a demonstrated 97 percent perchlorate reduction capability based on reducing perchlorate from challenge level of 130  $\mu$ g/L to a target level of 4  $\mu$ g/L.

Boodoo (2003) provides an assessment of possible configurations for POU or point-of-entry ion exchange devices that achieve high removal rates. There are, however, no certification standards for such devices.

# 5.3 Raw Water Quality Considerations

Because the POU RO devices will be installed at service taps that are downstream of a system's entry point to the distribution system, EPA assumes that the raw water entering a POU RO device will be water that is suitable for consumption except for an exceedance of the proposed perchlorate regulatory standard. As noted in the next section, POU RO devices include pre-filters to address potential interference of delivered water quality with RO performance.

#### 5.4 Pre- and Post-Treatment Needs

POU RO devices include various filters to address pre- and post-treatment concerns. Most devices include a sediment filter for solids removal to prevent membrane fouling and a pre-RO carbon filter to remove chlorine and organic compounds that could impair membrane function. They also include a carbon filter after the membrane and storage tank to remove any organics that may remain or bacterial growth that occurs during storage. Because the POU device is installed at the tap, there are no potential adverse impacts on the distribution system.

# 5.5 Waste Generation and Residuals Management Needs

The treatment process waste comprises wastewater and used filter cartridges. Waste disposal methods must comply with state and local requirements. The wastewater connection is generally plumbed to the household sewer system, which uses either an on-site septic system or a centralized wastewater collection system for disposal. Depending on state and local regulations, the used cartridge filters may be included in household solid waste (USEPA, 2006b).

# 5.6 Critical Design Parameters

In addition to the POU devices themselves, there are several components to the design of a POU program that are primary cost drivers. These include the following:

- POU RO device installation
- Public education program development
- POU device monitoring
- POU device maintenance.

Chapter [ REF\_Ref326047741 \r \h ] discusses each of these parameters in more detail and identifies the specific values for each used in EPA's cost estimates.

## 6 Nontreatment Alternatives

# 6.1 Application Principle

For small water utilities that lack the financial and/or technical capacity to implement a new treatment-based compliance strategy, nontreatment options may offer a more cost-effective path to compliance. Nontreatment options essentially replace the contaminated water source with water that meets drinking water standards, including a new standard for perchlorate.

Nontreatment solutions for drinking water compliance include the following: well rehabilitation; contaminant source elimination; new well construction; and interconnecting with another system to purchase water (USEPA, 2006c). The feasible nontreatment options will depend on site-specific circumstances such as system size, source water type, contaminant reduction needs, and proximity to alternative water sources. For small systems, neither well rehabilitation for contaminated ground water sources nor source elimination (e.g., remediation of perchlorate-contaminated sediments or ground water) is likely to be feasible and cost-effective solutions. Another option – blending water from existing wells – may be a feasible, low-cost option for systems with multiple wells including some for which perchlorate does not exceed the proposed perchlorate standard. For systems that cannot blend source water to comply with the proposed standard, two feasible nontreatment options include a new well to replace the contaminated source water and an interconnection to purchase water from a supplier. These two options (new wells and interconnection) are likely to have higher costs than the other options (well rehabilitation and source elimination) (USEPA, 2006c).

The costs associated with drilling a new well include the initial hydrological assessment, pilot hole drilling, developing the final well design, drilling the well bore, installing well casings, screens, and filters, development of the well, and installation of the pump and power source (Harter, 2003). A hydrological assessment identifies ground water sources of suitable quality and adequate long-term supply. When replacing an existing well, the costs will also include connecting the well to the existing water distribution system.

The interconnection option involves laying a pipeline to connect the affected system to the distribution network of a neighboring system that can provide adequate water that meets all applicable drinking water standards. Costs include the cost of purchased water as well as construction and maintenance of the interconnection pipeline. Pipeline costs will depend on proximity of the neighboring system, topography of the distance to be covered, and right-of-way requirements for pipes and booster pump stations.

# **6.2 Compliance Effectiveness**

Nontreatment options achieve compliance by replacing a perchlorate-contaminated water source with an alternative water source that meets a perchlorate standard. This strategy is inherently compliant as long as the new water source is not at risk for perchlorate contamination. If the wholesale supplier of purchased water has perchlorate contamination, it must implement an effective treatment process because the water it sells must comply with the perchlorate standard before it can be distributed to the purchasing system.

# 6.3 Raw Water Quality Considerations

A system will need to determine whether the change in source water may affect other existing treatment processes (e.g., chlorination), or if changes in water quality may affect the distribution system (e.g., purchased water has a different pH). Changes in delivered water chemistry that result in major process additions or changes could diminish the cost-effectiveness of nontreatment options.

#### 6.4 Pre- and Post-Treatment Needs

By definition, there are no pre-treatment needs to consider with a change in source water. All treatment adjustments to account for differences in source water quality would necessarily occur after the point of source water connection. If the alternative water source has chemical parameters that differ substantially from the original source water and may affect water quality elsewhere in the system, then there may be "post-treatment" needs to adjust water chemistry.

# 6.5 Waste Generation and Residuals Management Needs

An interconnection or new well should not have incremental wastes or residuals requiring management.

# 6.6 Critical Design Parameters

For new wells, key design parameters are the following:

- Total flow rate requirements and flow per well
- Well depth (and screened depth)
- Pump type
- Distance from well to distribution system.

For an interconnection option, key design parameters include:

- Flow rate requirements
- Distance to interconnection water supply
- Pressure at water supply source
- Cost of purchased water.

Chapter [ REF \_Ref326047741 \r \h ] discusses each of these parameters in more detail and identifies the specific values for each used in EPA's cost estimates.

## 7 Costs for Treatment Technologies and Nontreatment Options

Note: The technologies evaluated here can achieve very high perchlorate removal efficiencies (e.g., 95 percent or greater). Given the high efficiencies, EPA assumes systems will blend treated water and untreated water to meet the MCL. Accordingly, the costs presented here reflect systems designed and operated to take advantage of the technologies' high removal effectiveness and the cost curves should be applied to design and average flows adjusted to account for the blending rate, as discussed in the following paragraphs.

A blending rate is the proportion of influent water that has to be treated. For example, a blending rate of 0.6 means 60 percent of the water is treated and then blended with 40 percent untreated water. This rate depends on baseline perchlorate concentration, the treatment target concentration, and the removal efficiency of the treatment process (i.e., the percent of baseline perchlorate removed during treatment). For a treatment efficiency of 95 percent (or 0.95), the following equation defines the treatment target concentration of perchlorate ( $P_t$ ) as a weighted average of the baseline concentration ( $P_b$ ) and the treated water concentration [ $P_b$  x (1-0.95)] where the weights – based on the blending rate,  $P_b$  are (1- $P_b$ ) for the untreated water and  $P_b$  for the treated water:

$$P_t = (1 - B) \times P_b + B \times (P_b \times (1 - 0.95)).$$

Rearranging terms to solve for B (the blending rate) shows that the blending rate increases when the baseline concentration increases or the treatment target concentration decreases.

$$B = \frac{(P_b - P_t)}{P_b \times 0.95}$$

The cost curves presented here use the <u>treatment process flow</u> as the independent variable. Treatment process flow can be calculated from entry point flow by incorporating the blending rate (B) as follows:

Treatment Process Flow =  $B \times Entry\ Point\ Flow$ .

#### 7.1 Introduction

#### 7.1.1 Overview and Cost Modeling Approach

This chapter presents estimated costs for installing and operating the technologies and nontreatment options discussed in Chapters [ REF \_Ref529892942 \r \h ] through [ REF \_Ref529892955 \r \h ]. Based on the information in those chapters, particularly the data on engineering design specifications, EPA developed work breakdown structure (WBS) cost estimating models for each of the perchlorate treatment technologies. The WBS models are spreadsheet-based engineering models for individual treatment technologies, linked to a central database of component unit costs. EPA developed the WBS model approach as part of an effort to address recommendations made by the Technology Design Panel (TDP), which convened in 1997 to review the Agency's methods for estimating drinking water compliance costs (USEPA,

1997).<sup>10</sup> In general, the WBS approach involves breaking a process down into discrete components for the purpose of estimating unit costs. The WBS models represent improvements over past EPA cost estimating methods by increasing comprehensiveness, flexibility, and transparency. By adopting a WBS-based approach to identify the components that should be included in a cost analysis, the models produce a more comprehensive assessment of the capital and operating requirements for a treatment system. The documentation for the individual WBS models (USEPA, 2018, 2017a-c, 2007) provides complete details on the structure, content, and use of the models. EPA used the WBS models to develop the costs presented in this chapter.

The remainder of this section provides a brief overview of the common elements of all the WBS models and information on the anticipated accuracy of the resulting cost estimates. Subsequent sections describe how EPA used each individual technology-specific WBS model to estimate costs for perchlorate treatment and present the resulting cost estimates.

#### 7.1.2 Work Breakdown Structure Models

Each WBS model contains the work breakdown for a particular treatment process and preprogrammed engineering criteria and equations that estimate equipment requirements for user-specified design requirements (e.g., system size and influent water quality). Each model also provides unit and total cost information by component (e.g., individual items of capital equipment) and totals the individual component costs to obtain a direct capital cost. Additionally, the models estimate add-on costs (permits, pilot study, and land acquisition costs for each technology), indirect capital costs, and annual operation and maintenance (O&M) costs, thereby producing a complete compliance cost estimate.

Primary inputs common to all of the WBS models include design flow and average flow in MGD. Each WBS model has default designs (input sets) that correspond to the eight standard flow sizes in EPA's flow characterization paradigm for public water systems (see [ REF \_Ref256672992 \h \\* MERGEFORMAT ]), but the models can generate designs for many other combination of flows. To estimate costs for perchlorate compliance, EPA fit cost curves to the WBS estimates for up to 49 different flow rates. 

[ REF \_Ref326230172 \r \h ] through [ REF \_Ref329773612 \r \h ] and Appendix B are in the form of equations.

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<sup>&</sup>lt;sup>10</sup> The TDP consisted of nationally recognized drinking water experts from U.S. EPA, water treatment consulting companies, public and private water utilities and suppliers, equipment vendors, and Federal and State regulators in addition to cost estimating professionals.

<sup>&</sup>lt;sup>11</sup> Specifically, for each scenario modeled and separately for total capital and for O&M costs, EPA fit up to three curves: one covering small systems (less than 1 MGD design flow), one covering medium systems (1 MGD to less than 10 MGD design flow), and one covering large systems (10 MGD design flow and greater). For each curve fit, EPA chose from among several possible equation forms: linear, quadratic, cubic, power, exponential, and logarithmic. EPA chose the form that resulted in the best correlation coefficient (R²), subject to the requirement that the equation must be monotonically increasing over the appropriate range of flow rates (i.e., within the flow rate category, the equation must always result in higher estimated costs for higher flow systems than for lower flow systems).

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Model Size Categories Based on EPA's Flow Characterization Paradigm

Size Category	Population Served	Design Flow (MGD)	Average Flow (MGD)
1	25 to 100	0.030	0.007
2	101 to 500	0.124	0.035
3	501 to 1,000	0.305	0.094
4	1,001 to 3,300	0.740	0.251
5	3,301 to 10,000	2.152	0.819
6	10,001 to 50,000	7.365	3.200
7	50,001 to 100,000	22.614	11.087
8	Greater than 100,000	75.072	37.536

Another input common to all of the WBS models is "component level" or "cost level." This input drives the selection of materials for items of equipment that can be constructed of different materials. For example, a low cost system might include fiberglass pressure vessels and PVC piping. A high cost system might include stainless steel pressure vessels and stainless steel piping. The component level input also drives other model assumptions that can affect the total cost of the system, such as building quality and heating and cooling. The component level input has three possible values: low cost, mid cost, and high cost. To estimate costs for perchlorate compliance, EPA generated separate cost curves for each of the three component levels, thus creating a range of cost estimates for use in national compliance cost estimates.

The third input common at all of the WBS models is system automation, which allows the design of treatment systems that are operated manually or with varying degrees of automation (i.e., with control systems that reduce the need for operator intervention). The cost estimates in the technology-specific sections below are for systems that are fully automated, minimizing the need for operator intervention and reducing operator labor costs.

The WBS models generate cost estimates that include a consistent set of capital, add-on, indirect, and O&M costs. [REF\_Ref256677692 \h \\* MERGEFORMAT] identifies these cost elements, which are common to all of the WBS models and included in the cost estimates below. The exhibit also provides references for further information on the methods and assumptions used in the WBS models to estimate the costs for each of these cost elements.

#### 7.1.3 WBS Model Accuracy

Costs for a given system can vary depending on site-specific conditions (e.g., raw water quality, climate, local labor rates, and location relative to equipment suppliers). The costs presented here are based on national average assumptions and include a range (represented by low, mid, and high cost curves) intended to encompass the variation in costs that systems would incur to remove perchlorate. To validate the engineering design methods used by the WBS models and increase the accuracy of the resulting cost estimates, EPA has subjected the individual models to a process of external peer review by nationally recognized technology experts.

The anion exchange model underwent peer review in 2005, during an early stage of its development. One peer reviewer responded that resulting cost estimates were in the range of budget estimates (+30 to -15 percent). The other two reviewers thought anion exchange estimates were order of magnitude estimates (+50 to -30 percent), with an emphasis on the estimates being

high. The anion exchange model has since undergone extensive revision, both in response to the peer review and to adapt it for perchlorate treatment (see Section [ REF \_Ref326230172 \r \h ], below).

The RO/NF model underwent peer review in 2007. The majority of peer reviewers who evaluated the model expressed the opinion that resulting cost estimates would be in the range of budget estimates (+30 to -15 percent). The RO/NF model has since undergone substantial revision in response to the peer review comments.

The biological treatment model underwent peer review in early 2012. One reviewer thought the model underestimated O&M costs by 20 to 30 percent (which would be in the range of an order of magnitude estimate), but overestimated capital costs by about 25 percent (which would be in the range of a budget estimate). A second reviewer responded that direct capital costs were at the fringes of a budget estimate (+30 to -15 percent), while total capital costs were in the order of magnitude range (+50 to -30 percent) or possibly even better, in the budget estimate range. This reviewer's conclusions about total capital costs were based on comparison to preliminary costs for a plant currently under construction, for which the model underestimated costs. The final reviewer responded that costs were budget estimates (+30 to -15 percent). The biological treatment model has since undergone revision in response to the peer review comments.

The POU model underwent peer review in 2006. Reviewers felt that the default assumptions may tend to overstate "out-of-pocket" costs to systems because very small systems could use volunteers to perform some tasks. While this may be true, EPA's model is designed to estimate the opportunity costs for a successful POU program that is consistent with EPA POU Guidance, which does not include volunteers. The POU model has since been revised in response to the peer review comments.

EPA received peer review comments on the nontreatment model in May 2012. The first reviewer responded that cost estimates resulting from the nontreatment model were in the range of budget estimates (+30 to -15 percent). The second reviewer thought the cost estimates were order of magnitude estimates (+50 to -30 percent). The third reviewer felt the cost estimates were definitive (+15 to -5 percent), except for land costs, which were difficult to assess due to regional variations. Revision of the nontreatment model in response to the peer review comments was recently completed.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Cost Elements Included in All WBS Models

Cost Category and Components Included	For Further Information
Direct Capital Costs	
<ul> <li>Technology-specific equipment (e.g., vessels, basins, pumps, blowers, treatment media, piping, valves)</li> </ul>	Technology-specific sections below
Instrumentation and system controls	USEPA (2018, 2017a-c), Appendix A
Buildings	USEPA (2018, 2017a-c), Appendix B
Residuals management equipment	USEPA (2018, 2017a-c), Appendix C
Add-on Costs	
• Land	USEPA (2018, 2017a-c), Chapter 2
Permits	
Pilot testing	
Indirect Capital Costs	
Mobilization and demobilization	USEPA (2018, 2017a-c), Appendix D
Architectural fees for treatment building	
• Equipment delivery, equipment installation, and contractor's overhead an	nd profit
Sitework	
Yard piping	
Geotechnical	
Standby power	
Electrical infrastructure	
Process engineering	
Contingency	
Miscellaneous allowance	
Legal, fiscal, and administrative	
Sales tax	
Financing during construction	
Construction management	
O&M Costs	
<ul> <li>Operator labor for technology-specific tasks (e.g., managing regeneration backwash, or media replacement)</li> </ul>	n, Technology-specific sections below
<ul> <li>Materials for maintenance and operation of technology-specific equipmer</li> </ul>	nt
<ul> <li>Replacement of technology-specific equipment that occurs on an annual</li> </ul>	basis
(e.g., treatment media)	
<ul> <li>Energy for operation of technology-specific items of equipment (e.g., blow mixers)</li> </ul>	wers,
Operator labor for operation and maintenance of process equipment	USEPA (2018, 2017a-c), Appendix E
Operator labor for building maintenance	
Managerial and clerical labor	
Materials for maintenance of booster or influent pumps	
Materials for building maintenance	
Energy for operation of booster or influent pumps	
Energy for lighting, ventilation, cooling, and heating	
Residuals management operator labor, materials, and energy	USEPA (2018, 2017a-c), Appendix C
Residuals disposal and discharge costs	

#### 7.2 Costs for Ion Exchange

#### 7.2.1 Model Components and Assumptions

USEPA (2017a) provides a complete description of the engineering design process used by the WBS model for perchlorate ion exchange. The perchlorate ion exchange model can estimate costs for removing perchlorate using any of the following resin types described Chapter 2: strong-base polyacrylic, strong-base polystyrenic, nitrate-selective, and perchlorate-selective. In addition to the common WBS direct capital cost items listed in [REF\_Ref256681355 \h \\* MERGEFORMAT], the ion exchange model for perchlorate includes the following technology-specific equipment:

- Booster pumps for influent water
- Pressure vessels that contain the anion resin bed
- Tanks and pumps for backwashing the vessels
- Tanks, mixers, and eductors for delivering the solution used in regenerating the resin (if regeneration is used)
- Pre-treatment cartridge filters
- Tanks and pumps for post-treatment corrosion control (optional)
- Equipment for managing residuals (spent backwash, spent resin, and, potentially, spent regenerant)
- Associated piping, valves, and instrumentation.

The ion exchange model for perchlorate also adds the following technology-specific O&M elements:

- Operator labor for resin changeouts
- Operator labor for regeneration (if regeneration is used)
- Spent resin replacement and disposal
- Labor and replacement cartridges for pre-treatment filters
- Chemical usage (if corrosion control or regeneration is used).

For small systems (less than 1 MGD), the ion exchange model for perchlorate assumes the use of package treatment systems that are pre-assembled in a factory, mounted on a skid, and transported to the site. The model estimates costs for package systems by costing all individual equipment line items (e.g., vessels, interconnecting piping and valves, instrumentation, and system controls) in the same manner as custom-engineered systems. This approach is based on vendor practices of partially engineering these types of package plants for specific systems (e.g., selecting vessel size to meet flow and treatment criteria). The model applies a variant set of design inputs and assumptions that are intended to simulate the use of a package plant and that reduce the size and cost of the treatment system. USEPA (2017a) provides complete details on the variant design assumptions used for package plants.

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<sup>&</sup>lt;sup>12</sup> The input also allows for selection of an alternative resin type, by entering appropriate design assumptions for their resin and selecting "Alternative resin (user defined)" for this input.

The paragraphs below describe the specific inputs and assumptions that EPA used to generate the costs in Section [ REF \_Ref329333708 \r \h ]. These inputs assume treatment with perchlorate breakthrough defined so that the treatment process maintains a minimum of 95 percent removal. Other inputs and assumptions not discussed below (e.g., number of booster pumps, treated water corrosion control, bed expansion) remained as described in USEPA (2017a).

#### Resin Type

As noted above, the perchlorate ion exchange model can estimate costs for removing perchlorate using several different resin types. Based on the information in Section [ REF \_Ref286409819 \r \h ] and [ REF \_Ref286411968 \h \\* MERGEFORMAT ], however, EPA believes that any new ion exchange facilities removing perchlorate will use perchlorate-selective resin. Therefore, the cost estimates below assume the use of perchlorate-selective resin and, accordingly, the paragraphs below describe the inputs and assumptions associated with perchlorate-selective resins only.

#### Regeneration or Throwaway Operation

The perchlorate ion exchange model has an option to design and estimate costs for a system either with or without regeneration capability. As discussed in Section [ REF \_Ref292805680 \r \h ], nearly all of the full-scale facilities using perchlorate-selective resin rely on disposal instead of regeneration. Therefore, the cost estimates below assume throwaway operation.

#### Number of Bed Volumes before Regeneration/Throwaway

The perchlorate ion exchange model requires entry of the number of bed volumes before perchlorate breakthrough. System configuration (i.e., parallel or series operation) can have a significant effect on bed volumes to breakthrough. As discussed below, the model default assumption is series (lead-lag) operation. The data shown in [REF\_Ref286412546 \h \\* MERGEFORMAT] are for initial perchlorate breakthrough using a single resin column. Studies of the capacity of the older perchlorate-selective resin (Gu et al., 1999) found breakthrough in a lead column after 40,000 BV. Using a second (lag or polishing) column increased the resin's capacity to approximately 104,000 BV. Similar studies of the performance of the new resins in a lead-lag configuration are not available.

Given the lack of precise data on the performance of the new resins in a lead-lag configuration and given expected site-specific variations in water quality (e.g., concentrations of competing anions), the cost estimates below consider two scenarios for bed life. Although one of the scenarios assumes a longer bed life than the other (meaning better performance and lower costs), both scenarios are designed to be conservative (erring on the site of higher costs).

The first scenario assumes an increase in capacity of the new resins from lead-lag operation similar to that observed for the older resin. This scenario starts from the <u>lowest</u> single-column capacity (about 100,000 BV) observed for any of the new resins in several pilot studies (Blute et al., 2006; Russell et al., 2008; Wu and Blute, 2010). It multiplies this capacity by 2.5 (the approximate increase observed for the older resin in Gu et al., 1999) to account for lead-lag operation. This calculation results in 250,000 BV to breakthrough. The second scenario starts from the highest single-column capacity (about 170,000 BV) observed for any of the new resins

the pilot studies, but assumes <u>no increase in capacity from lead-lag operation</u>. Thus, the second scenario assumes 170,000 BV to breakthrough.

#### Number of Vessels in Series (parallel or series operation)

As discussed in Section [ REF \_Ref292959102 \n \h ], series (lead-lag) operation is generally recommended for perchlorate removal. Therefore, the cost estimates below assume two vessels in series.

#### **Theoretical Total Empty Bed Contact Time**

As discussed in Section [ REF \_Ref292959102 \n \h ], recommended EBCTs for perchlorate-selective resins are 1.5 minutes per vessel and less. Therefore, the cost estimates below assume a total EBCT of 3 minutes (which corresponds to 1.5 minutes per vessel, given two vessels in series).

#### **Backwash System Design**

The perchlorate ion exchange model assumes that periodic backwashing during operation is not required when throwaway operation is chosen. It does, however, assume an initial rinse is required during resin installation. Therefore, for systems of 1 MGD design flow and larger the cost estimates below include the cost of equipment (pumps and storage tanks) to accomplish this initial rinse. For small systems, the cost estimates assume the initial rinse can be accomplished using existing equipment.

#### **Residuals Management**

The perchlorate ion exchange model includes the option to dispose of spent resin by incineration. As discussed in Section [REF\_Ref292960035 \n \h], a number of full-scale facilities dispose of their spent perchlorate-selective resin by incineration. Although incineration is a more expensive option than landfill disposal, the overall impact on operating costs is small. Incineration increases model cost estimates by approximately \$0.01 per thousand gallons of treated water produced, which is less than 3 percent of the total cost of treatment even for the largest systems. Therefore, the cost estimates below assume disposal by incineration, although some systems might have the slightly cheaper option of landfill disposal available. They assume that spent rinse water from the initial resin rinse is discharged to a POTW.

#### **Surface Loading Rate**

As discussed in Section [ REF \_Ref292959102 \n \h ], maximum surface loading rates vary by resin type. Based on the data presented in Section [ REF \_Ref292959102 \n \h ] and comments from the experts who reviewed initial drafts of this document (Meyer, 2012; Blute, 2012; Drago,

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<sup>&</sup>lt;sup>13</sup> This option is activated by entering the cost of incineration under "Alternate media disposal cost (including transportation)" in the O&M assumptions module of the perchlorate version of the ion exchange model. When a value is present for this assumption, the model uses this unit cost in estimating total non-hazardous waste disposal cost. If this assumption is left blank, the model uses the default unit cost from the central WBS database, which reflects disposal in an off-site non-hazardous waste landfill.

2012), EPA chose a maximum surface loading rate of 12 gpm/ft<sup>2</sup> for perchlorate-selective resin. The cost estimates below incorporate this assumption.

#### Regeneration Solution Type and Assumptions

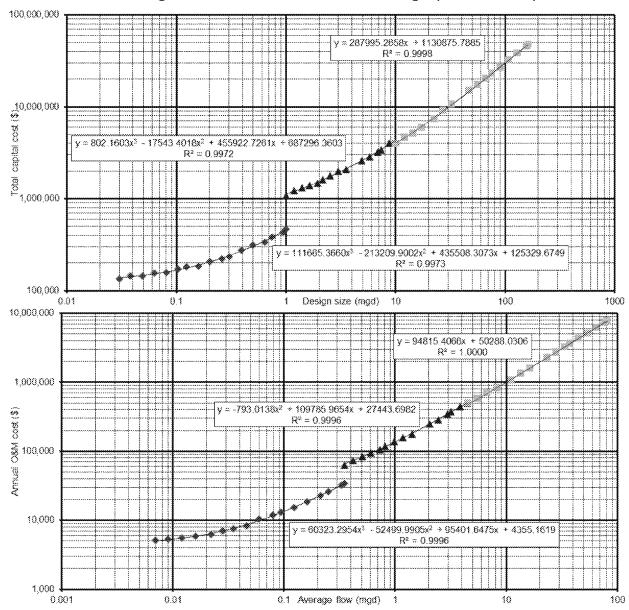
The perchlorate ion exchange model incorporates the option to regenerate selective resins using the novel tetrachloroferrate regeneration solution discussed in Section [ REF \_Ref292709289 \n \h ]. Specifically, when regeneration is chosen for nitrate- or perchlorate-selective resins, the model assumes the use of tetrachloroferrate solution. When regeneration is chosen for non-selective strong-base polyacrylic or polystyrenic resins, the model assumes the use of conventional brine solution. The selection of resin type also controls the values of a number of critical design assumptions relevant to each regeneration process (e.g., brine concentration, tetrachloroferrate solution strength, regeneration time). The cost estimates below, however, assume throwaway operation and, therefore, do not incorporate these assumptions.

#### 7.2.2 Cost Estimates

The graphs below plot WBS cost model results in 2017 dollars at the mid cost level for removal of perchlorate from groundwater using perchlorate-selective ion exchange, assuming 250,000 BV to breakthrough ([REF\_Ref292961817 \h \\* MERGEFORMAT]) and 170,000 BV to breakthrough ([REF\_Ref292961823 \h \\* MERGEFORMAT]). The costs assume treatment with perchlorate breakthrough defined so that the treatment process maintains a minimum of 95 percent removal. The flow rates shown on the x-axes and as independent variables in the equations are treatment process flows. To account for blending, treatment process flows should be calculated from entry point flows by incorporating a blending rate as discussed in the introduction to this chapter.

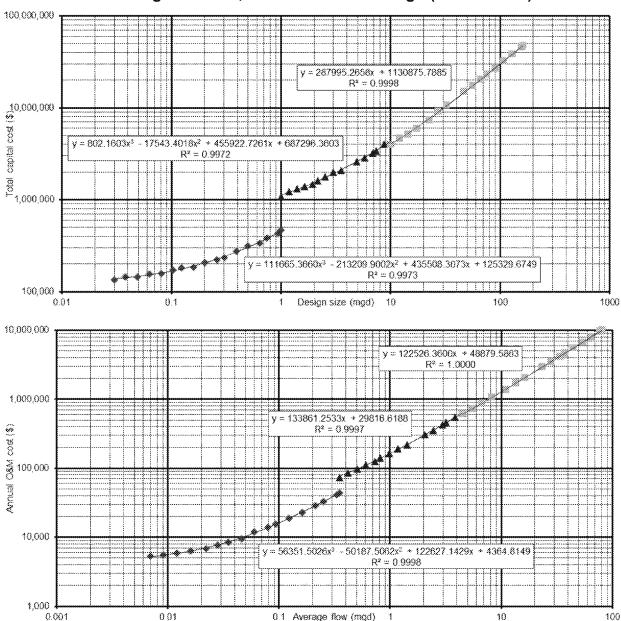
In these exhibits, note that costs increase at 1 MGD design flow (0.355 MGD average flow) because of the transition from package systems (used by small systems) to custom-engineered systems (used by large systems). Appendix B provides complete cost equations for both bed life scenarios, including the high, mid, and low cost levels and for treatment of groundwater and surface water. Appendix C presents example WBS model outputs at selected flow rates, allowing review of individual cost line items.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Perchlorate-Selective Ion Exchange with 250,000 BV to Breakthrough (2017 dollars)



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# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Perchlorate-Selective Ion Exchange with 170,000 BV to Breakthrough (2017 dollars)



#### 7.3 Costs for Biological Treatment

#### 7.3.1 Model Components and Assumptions

USEPA (2017b) provides a complete description of the engineering design process used by the WBS model for biological treatment. The biological treatment model can estimate costs for three types of bioreactor designs:

- Fixed media bed pressure vessels
- Fixed media bed gravity basin
- Fluidized bed pressure vessels.

In addition to the common WBS direct capital cost items listed in [ REF\_Ref256677692 \h \\* MERGEFORMAT], the biological treatment model includes the following technology-specific equipment:

- Booster pumps for influent water
- Equipment (e.g., tanks, pumps) for electron donor addition
- Equipment (e.g., tanks, pumps) for nutrient addition
- Bioreactors (either pressure vessels or concrete basins) that contain the GAC media bed
- Tanks, pumps, and blowers for backwashing the bioreactors (fixed bed designs only) and post-treatment filters (if used)
- Pumps for recycled water (fluidized designs only)
- Blowers for biomass separation (fluidized designs only)
- Equipment for post-treatment aeration or hydrogen peroxide addition (optional)
- Post-treatment coagulant addition and mixed media filtration (optional)
- Equipment for managing residuals (spent backwash)
- Associated piping, valves, and instrumentation (including online perchlorate and nitrate analysis instruments).

The biological treatment model also adds the following technology-specific O&M elements:

- Operator labor for managing backwashes
- Operator labor and materials for maintaining concrete basins (gravity designs only)
- Operator labor and materials for maintaining backwash pumps and air scour blowers (and biomass removal blowers for fluidized bed designs)
- Chemical usage for electron donor and nutrient addition (and post-treatment hydrogen peroxide addition, if used)
- Coagulant and polymer usage for spent backwash treatment (and post-treatment filtration, if used)
- Attrition loss replacement for bioreactor media (and post-treatment filter media, if used)
- Consumables used in online perchlorate and nitrate analysis.

For small systems (less than 1 MGD), the biological treatment model applies a set of design inputs and assumptions that reduce the size and cost of the treatment system relative to larger systems. Some of these small system assumptions are similar to those used for package plants in the ion exchange model (e.g., skid-mounted pressure vessels, reduced need for booster pumping). Because package plants are not currently available for biological treatment, however,

the biological treatment model assumptions do not differ as greatly between small and large systems. USEPA (2017b) provides complete details on the variant design assumptions used for small systems.

The paragraphs below describe specific inputs and assumptions that EPA used to generate the costs in Section [ REF \_Ref326051577 \r \h \\* MERGEFORMAT ]. Other inputs and assumptions not discussed below (e.g., number of booster pumps, bioreactor dimensions) were as described in USEPA (2017b).

#### **Electron Donor Type**

The biological treatment model allows the user to select between acetic acid and ethanol as the electron donor type. As discussed in Section [ REF \_Ref326052320 \r \h \\* MERGEFORMAT ], these are the most common electron donors used in full-scale biological systems treating perchlorate. The cost estimates below assume acetic acid as the electron donor.

#### **Electron Donor Dose**

As discussed in Section [ REF \_Ref326052320 \r \h \\* MERGEFORMAT ], electron donor dose is typically determined using pilot studies along with stoichiometric calculations. The biological treatment model requires the user to input the electron donor dose. For comparison purposes, the user can enter raw waster quantities of perchlorate, nitrate, and dissolved oxygen and the model will display the results of stoichiometric calculations using the site-specific equations discussed in Section [ REF \_Ref326052320 \r \h \\* MERGEFORMAT ]. The cost estimates below assume 10 mg/L of acetic acid.

#### **Nutrient Requirements**

The biological treatment model includes four options for nutrient addition:

- no additional nutrients required
- additional nitrogen required
- additional phosphorous required
- both nitrogen and phosphorous required.

For designs that require additional nutrients, the model prompts the user to input doses for ammonium chloride and/or phosphoric acid. Based on comments from the peer review of the biological treatment model, the cost estimates below assume additional phosphorous is required and achieved using 1 mg/L (measured as phosphorus) of phosphoric acid.

#### **Empty Bed Contact Time/Hydraulic Residence Time**

The biological treatment model uses EBCT in sizing fixed bed bioreactors and HRT in sizing fluidized bed bioreactors. As discussed in Section [ REF \_Ref326052320 \r \h \\* MERGEFORMAT ], typical full-scale bioreactor designs have an EBCT/HRT in the range of 10 to 12 minutes for both fixed bed (U.S. DoD, 2008b) and fluidized bed reactors (Harding ESE, 2001). The cost estimates below assume an EBCT/HRT value of 12 minutes.

#### Interval between Backwashes

For fixed bed bioreactors, the biological treatment model requires the user to input the interval between backwashes. <sup>14</sup> For designs that include post-treatment filters (see below), the model requires a separate backwash interval for these filters. The cost estimates below assume backwash intervals of 24 hours for fixed bed bioreactors and 36 hours for post-treatment filters.

#### **Post-Treatment Options**

Biological treatment results in the production of soluble microbial organic products that become part of the treated water. The biological treatment process also depletes the levels of oxygen in the treated water. Therefore, based on peer review comments, post-treatment will be required for production of drinking water. The biological treatment model allows users to choose whether to include post-treatment coagulant addition and filtration for removal of turbidity, sulfide, and/or dissolved organic content. It also allows users to choose from aeration or hydrogen peroxide addition for post-treatment oxidation. The cost estimates below include post-treatment filtration with coagulant addition and aeration for post-treatment oxidation.

Although post-treatment disinfection typically is needed following biological treatment, disinfection is required for most water systems regardless of whether perchlorate treatment is present. Therefore, the costs of disinfection are not attributable exclusively to perchlorate compliance. Accordingly, the cost estimates below do not include post-treatment disinfection; they assume that existing disinfection facilities are sufficient and located appropriately.

#### Fluidized Bed Expansion

Based on peer review comments on the biological treatment model, the cost estimates below assume bed expansion of 70 percent for fluidized beds. They also include freeboard above the expanded bed of 4 feet for small systems (less than 1 MGD design flow) and 7 feet for larger systems (1 MGD and greater).

#### Fluidized Bed Recycle Rate

Based on peer review comments on the biological treatment model, the cost estimates below assume a recycle rate of 50 percent for fluidized beds.

#### 7.3.2 Cost Estimates

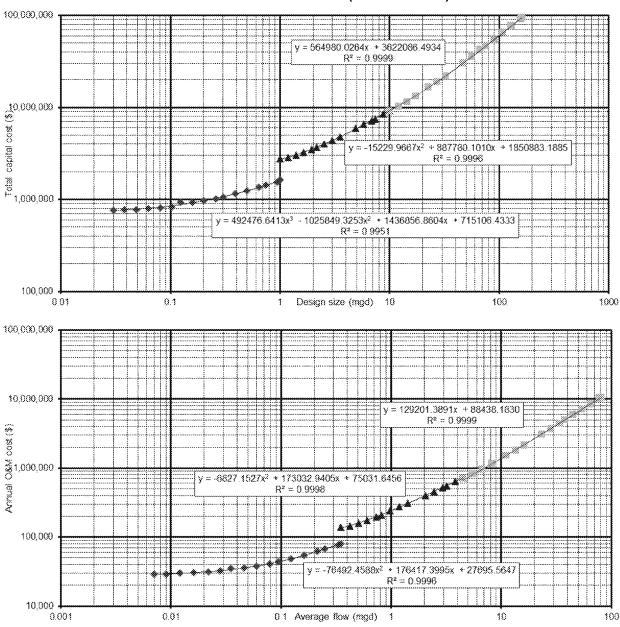
The graphs below plot WBS cost model results in 2017 dollars at the mid cost level for removal of perchlorate from groundwater using biological treatment with fixed bed pressure vessels ([REF\_Ref326062889 \h \\* MERGEFORMAT]), fixed bed gravity basins ([REF\_Ref326062900 \h \\* MERGEFORMAT]), and fluidized bed pressure vessels ([REF\_Ref329596213 \h \\* MERGEFORMAT]). In the exhibits, note that costs increase at 1 MGD design flow (0.355 MGD average flow) because of the change in assumptions used for small systems versus those for large systems. Appendix B provides complete cost equations for all three design types, including the high, mid, and low cost levels and for treatment of groundwater

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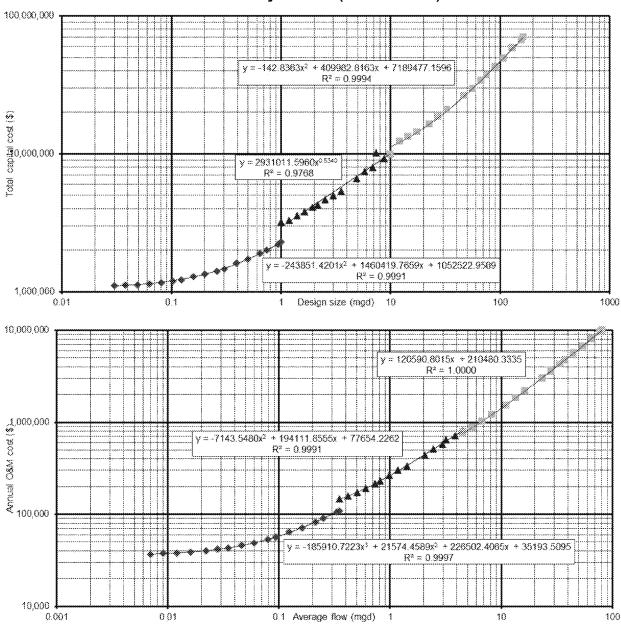
<sup>&</sup>lt;sup>14</sup> For fluidized bed bioreactors, the model uses a continuous biomass separation device consisting of a blower to agitate excess biomass and send it out of the bioreactor in the effluent, so no such interval is needed.

and surface water. Appendix C presents example WBS model outputs for selected flow rates, allowing review of individual cost line items.

# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Biological Treatment with Fixed Bed Pressure Vessels (2017 dollars)

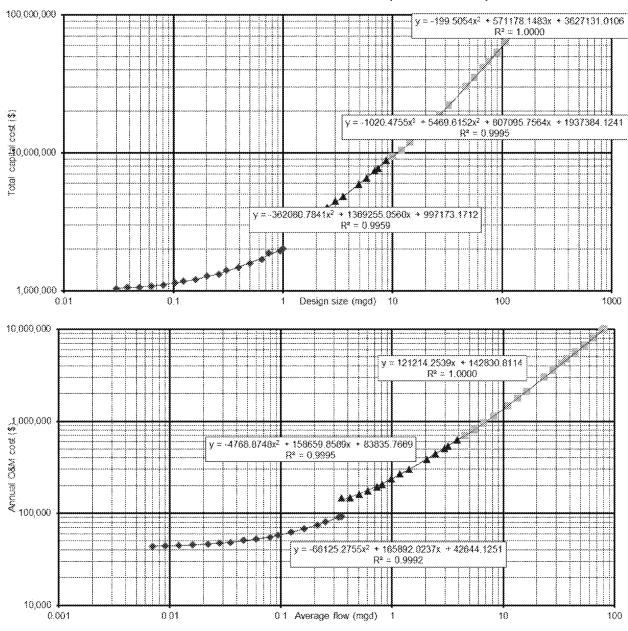


# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Biological Treatment with Fixed Bed Gravity Basins (2017 dollars)



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# Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Biological Treatment with Fluidized Bed Pressure Vessels (2017 dollars)



#### 7.4 Costs for Reverse Osmosis

#### 7.4.1 Model Components and Assumptions

USEPA (2018) provides a complete description of the engineering design process used by the WBS model for RO/NF. The model can estimate costs for a multistage RO installation, based on an input water quality analysis and treatment parameters. Although it is also capable of estimating costs for NF, the model was used here specifically for RO, because that membrane technology showed the most promise for removing perchlorate. As discussed in Chapter [ REF Ref329770602 \r \h ], NF (and UF) demonstrated widely varying removal of perchlorate.

In addition to the common WBS direct capital cost items listed in [ REF \_Ref256677692 \h \\* MERGEFORMAT ], the RO model includes the following technology-specific equipment:

- Pressure vessels, membrane elements, piping, valves, connectors, and steel structure for the membrane racks
- High-pressure pumps for influent water and (optionally) interstage pressure boost
- Valves for concentrate control and (optionally) per-stage throttle
- Cartridge filters for pretreatment
- Tanks, pumps, and mixers for pretreatment chemicals
- Tanks, pumps, screens, cartridge filters, and heaters for membrane cleaning
- Equipment for managing RO concentrate and spent cleaning chemicals
- Associated pipes, valves, and instrumentation.

The RO model also includes the following technology-specific O&M elements:

- Operator labor and replacement elements for pretreatment cartridge filters
- Chemical usage for pretreatment
- Labor and materials for routine operation and maintenance of membrane units
- Energy for high-pressure pumping
- Replacement of membrane elements
- Labor, materials, and chemical usage for membrane cleaning
- Disposal costs for spent cartridge filters and membrane elements
- Fee for disposal of concentrate at a POTW (if POTW disposal is selected).

The paragraphs below describe specific inputs and assumptions that EPA used to generate the costs in Section [ REF\_Ref329770828 \r \h \\* MERGEFORMAT ]. Other inputs and assumptions not discussed below (e.g., cleaning interval, permeate throttling and interstage boost, membrane life) were as described in USEPA (2018).

#### Water Type

As described in Section [ REF \_Ref329784338 \r \h ], the composition of the feed water affects pretreatment and cleaning requirements, the range of permissible RO train design parameters, and energy usage for pumping. The WBS model includes three default ground waters and three default surface waters, ranging from high to low quality (i.e., from low to high total dissolved solids and scaling potential). The particular water parameters are based on a survey of membrane feed water characteristics in the literature. The cost estimates below and in Appendix B are intended to reflect the incremental cost of removing perchlorate from otherwise deliverable water

using RO. Therefore, they use the default high quality water parameters built in to the WBS model. Total dissolved solids for the high quality surface water is approximately 360; for high quality ground water, total dissolved solids is approximately 500 mg/L. USEPA (2018) documents the other relevant characteristics of these default waters.

#### **Membrane Element**

The WBS model includes the option of NF, low-pressure RO, or brackish water RO membrane elements, with a diameter of 4 inches, 8 inches, or 16 to 18 inches. (Not all manufacturers use the same size for their largest diameter elements, but the model is independent of the exact diameter.) Since not all NF membranes are effective for perchlorate treatment, the cost estimates below assume use of low-pressure RO membrane elements, consistent with the type of elements shown to be effective by Liang et al. (1998). For very small systems, the cost estimates use 4-inch diameter elements; above that point they use 8-inch elements. The switch from 4-inch to 8-inch elements takes place at about 75,000 gallons per day.

#### **Recovery and Flux Rates**

The WBS model takes target recovery and flux rates, and designs the reverse osmosis train to come as close as possible to them. The flux rate, in combination with the system design flow, determines the total membrane area in the system, and therefore the total number of membrane elements to be used. The recovery rate affects the number of membrane elements in series. For instance, two stages of seven elements each would give fourteen elements in series, while three stages of six elements each would give eighteen. To maintain adequate crossflow, each individual element is limited in the recovery it can achieve. Therefore, a higher recovery rate requires more elements in series.

In general, the cost estimates use recovery rates of 80 to 85 percent. At small flows, the small number of membrane elements limits flexibility in the system design; therefore, estimates up to about 500,000 gallons per day may use recovery rates as low as 70 percent.

Flux rates are based on the recommendations of various manufacturers for waters of different challenge. For ground water, the estimates use flux rates of 19 gfd. For surface water, the rates are 15 to 16 gfd.

#### **Pretreatment**

The RO model always includes 5-micron cartridge filters to protect the membrane elements from suspended solids. The cost estimates also include the addition of sulfuric acid and scale inhibitor to prevent fouling of the membrane elements.

The cost estimates include the addition of enough sulfuric acid to ensure that at least 1.5 mol/L of inorganic carbon is present in the RO permeate, to be available for alkalinity recovery; the resulting dose ranges from 16 to 45 mg/L.

While there are many scale inhibitors available, the model includes two general classes, which it calls basic antiscalant (to address calcium carbonate scaling) and premium antiscalant (to address sulfate salts and silica scaling). The cost estimates below include the addition of basic antiscalant. The antiscalant doses are sufficient to provide a dose of 15 mg/L in the concentrate,

assuming that all antiscalant is retained on the feed/concentrate side of the membrane. This requirement corresponds to a feed water dose from 2.25 to 4.5 mg/L.

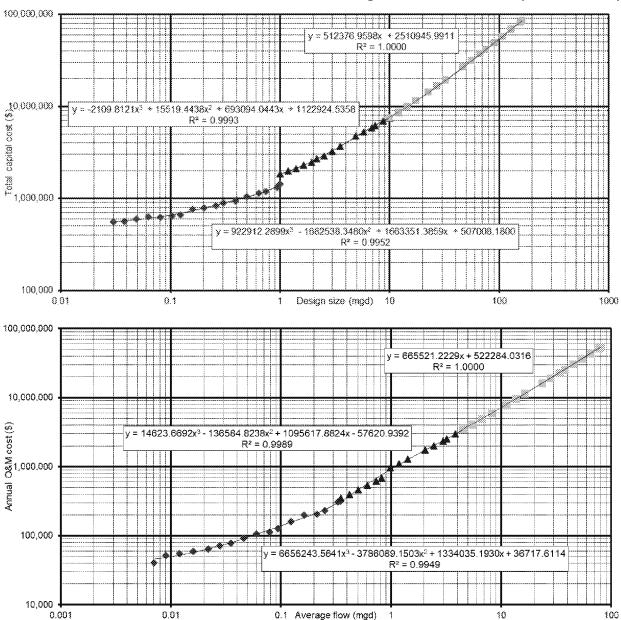
#### **Residuals Discharge**

Residuals discharge is usually a major contributor to the cost of an RO facility. The RO/NF process generates two residuals streams: the membrane concentrate and spent cleaning solution. Since the spent cleaning solution is generated infrequently and in small amounts, the model assumes that it will be diluted and discharged with membrane concentrate. The cost estimates below assume the combined residuals are sent to a POTW. Although it might be impractical for most POTWs to treat very large concentrate flows, this scenario results in more conservative estimates (i.e., erring on the side of higher costs) than surface water (ocean) discharge or deep well injection.

#### 7.4.2 Cost Estimates

The graphs below plot WBS cost model results in 2017 dollars at the mid cost level for removal of perchlorate from groundwater using RO ([REF\_Ref329598812 \h \\* MERGEFORMAT]). Appendix B provides complete cost equations, including the high, mid, and low cost levels and for treatment of groundwater and surface water. Appendix C presents example WBS model outputs for selected flow rates, allowing review of individual cost line items.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Removal of Perchlorate from Groundwater Using Reverse Osmosis (2017 dollars)



#### 7.5 Costs for Point-of-use Technologies

#### 7.5.1 Model Components and Assumptions

The document Cost Evaluation of Point-of-Use and Point-of-Entry Treatment Units for Small Systems: Cost Estimating Tool and User Guide (USEPA, 2007) provides a complete description of the WBS model for POU technologies. The POU model is capable of estimating equipment costs for a variety of POU devices, including POU RO devices and replacement filters. The POU model also includes the cost of the following other components of a complete POU program:

- POU RO device installation
- Public education program development
- POU device monitoring
- POU device maintenance.

Because only small systems would be expected to use POU programs, the POU model covers only the first four size categories shown in [ REF \_Ref256672992 \h \\* MERGEFORMAT ]. Also, the POU model does not include assumptions or materials of construction that vary based on a "component level" or "cost level" input. Therefore, unlike the other models, it does not generate separate estimates for low-, mid-, and high-cost scenarios.

To use the POU model in estimating costs for perchlorate, EPA selected a program using RO devices. [REF\_Ref293040823 \h \\* MERGEFORMAT] identifies the values used for parameters (other than equipment costs) that drive the costs of a POU RO program. EPA developed these assumptions based on EPA Guidance (USEPA, 2006b) and case study data, as discussed in detail in the paragraphs below.

### Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. POU Model Assumptions for Perchlorate

Parameter Category	Value	
Installation labor time	Plumber installation time: 2 hours per POU device (NSF, 2005)	
	Scheduling time: 0.5 hours per household (USEPA, 2006b)	
Public education program	Public meeting-related time: 20 hours	
	Other outreach time (e.g., program updates in a billing mailer): 4 hours	
Monitoring requirements	Initial monitoring for all units; annual monitoring for 1/3 of units (USEPA, 2006b)	
	Sampling time: 0.25 hours per sampling event (NSF, 2005)	
Filter replacement	Replacement schedule: RO element (3 years); post-RO carbon filter (1 year); pre-RO filters (9 months) (manufacturer recommendations)	
	Filter replacement time: 0.5 hour per change-out (NSF, 2005)	
	Scheduling time: 0.5 hours per household (USEPA, 2006b)	

#### **POU RO Device Installation**

Installation of the POU RO devices will be the responsibility of the water system. The utility can, however, hire a licensed plumber or representative of the product manufacturer to install the devices. Based on the variety of plumbing issues encountered among older housing units in a

rural community, NSF (2005) recommends using an experienced plumber to perform the installations.

The POU model contains a default estimate of two hours per household to install the POU RO. A variety of factors such as existing plumbing conditions and travel distance will affect installation times across sites. The estimate is consistent with case study data. In a Grimes, California, arsenic demonstration program (NSF, 2005), POU adsorptive filter installation times ranged from 15 minutes to 3 hours depending on the accessibility of piping and the need for additional lines (e.g., to provide treated water to ice-makers). The mean device installation time was one hour, but total plumber billing records indicated that twice as much time was spent on all installation-related activities (e.g., additional time to obtain special plumbing fittings and return visits to homes when residents missed their appointments).

Installation costs also include administrative time for system staff to contact homeowners to schedule an installation appointment. EPA assumed an average of 30 minutes (0.5 hours) per household to schedule an appointment. Scheduling effort is likely to vary across customers, with some being relatively easy to schedule while others may require multiple calls to identify and contact the correct homeowners or to handle situations such as homeowner reluctance to participate or language barriers (USEPA, 2006b).

#### **Public Education Program**

EPA Guidance (2006b) recommends that systems implement a public education program to obtain and maintain customer participation and long-term customer satisfaction with the POU program. The two main program elements recommended in USEPA (2006b) are: public meetings prior to installing any POU devices to educate customers about the regulatory compliance requirements and the role of the POU devices; and POU program updates in billing mailers and on information flyers posted in public locations such as a post office, a public library, or a website. The POU model includes labor costs for the following program elements:

- preparing information for one public meeting
- attending the meeting
- preparing an additional billing mailer with program updates.

Public education program costs are not available from POU case studies. USEPA (2007) provides a detailed breakdown of the assumptions used to generate the time estimates shown in [ REF\_Ref293040823 \h \\* MERGEFORMAT ]. It also describes the costs for materials such as information flyers for the public meeting, meeting announcements, and billing mailers.

#### **POU Device Monitoring**

A system that implements a POU compliance strategy will need to monitor the quality of water produced by the treatment devices to demonstrate compliance with a perchlorate standard. The system will need to work with the appropriate regulatory agency to establish an approved compliance-monitoring schedule (USEPA, 2006b). The resulting monitoring schedule may have sampling rates in initial year that differ from sampling rates in subsequent years. EPA Guidance (2006b) provides an example of a monitoring schedule in which samples are taken from every unit during the first year to confirm that the units are working properly, and then monitoring

frequency declines to one-third of units each subsequent year. EPA's cost estimates incorporate these monitoring frequencies.

Monitoring costs include sampling time, shipping fees, and laboratory analysis fees. The average sampling is 15 minutes (0.25 hours). To minimize the burden on households as well as system resources, EPA assumes that sampling occurs during installation or maintenance trips. The assumption is consistent with the Grimes case study cost analysis (NSF, 2005) used an estimate of 15 minutes per sampling event.

#### **POU Device Maintenance**

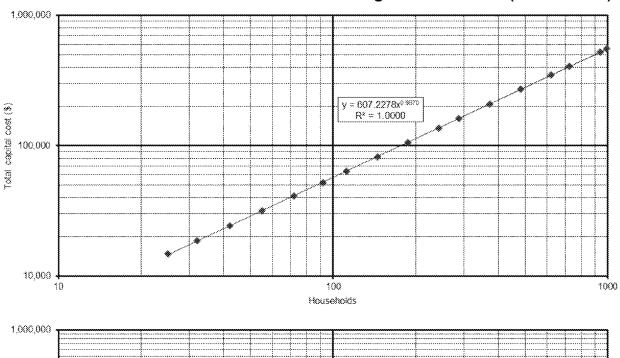
Maintenance for the POU RO device primarily includes replacing the four filters: RO membrane, two carbon filters, and the sediment filter. Replacement schedules reflect average useful lives based on vendor recommendations. On average, the RO membrane is replaced once every three years based on average replacement schedules across vendors, and the other filter cartridges are changed once per year.

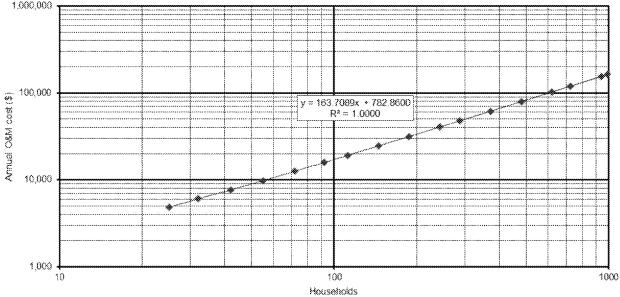
In addition to replacement filter costs, maintenance costs include scheduling time and time to change filters. The Grimes case study cost analysis (NSF, 2005) used an estimate of 15 minutes per filter change out. EPA assumed the average length of a maintenance call 30 minutes (0.5 hours) because the most frequent type of visit involves changing two filters. EPA used the same 30-minute scheduling time assumption that it used for initial installation.

#### 7.5.2 Cost Estimates

[ REF \_Ref326065349 \h \\* MERGEFORMAT ] plots WBS cost model results in 2017 dollars for removal of perchlorate from groundwater using POU treatment. Note that this exhibit is plotted based on the number of households served, rather than the system flow, because number of households is the more relevant parameter for POU treatment. EPA limits the POU model to a maximum of approximately 1,000 households served because implementing and maintaining a POU program for a greater number of households is likely to be impractical. Therefore, the exhibit does not extend beyond this maximum. As discussed above, the POU model also does not generate separate high, mid, and low cost estimates. Appendix B contains complete cost equations for POU treatment, including for groundwater and surface water. For use in national cost estimating, it also contains equations on the basis of design and average flow, in addition to those on the basis of households served. Appendix C presents example WBS model outputs for selected flow rates, allowing review of individual cost line items.

## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Cost Results for Removal of Perchlorate from Groundwater Using POU Treatment (2017 dollars)





#### 7.6 Costs for Nontreatment Options

#### 7.6.1 Overview

USEPA (2017c) provides a complete description of the engineering design process used by the WBS model for nontreatment options. The model can estimate costs for two nontreatment options: interconnection with another system and drilling a new well to replace a contaminated one. EPA based the model components, design parameters, and default user input values on information available for nontreatment case studies and cost analyses for prior regulations (USEPA, 1995; USEPA, 2006a; USEPA, 2006c; USEPA, 2008; AWWA, 2005; Lucey, 2008). These studies involved compliance via nontreatment options for contaminants such as arsenic, volatile organic contaminants, and radionuclides. The case studies are for systems that range in size from serving small communities with fewer than 100 connections to systems distributing 2.5 MGD. Although EPA does not have nontreatment case studies specific to perchlorate, the design and cost information contained in the case studies for other contaminants is transferable. Nontreatment options are less likely to be available for larger systems because of the quantity of water required. Therefore, EPA's WBS nontreatment cost model generates costs only for systems serving less than 10,000 people.

As discussed in Section [ REF \_Ref293044924 \r \h ], the two options covered by the WBS nontreatment model (new wells or interconnection) are likely to have higher costs than other nontreatment options available for perchlorate. The sections below identify the specific cost elements included under each option. They also describe the specific inputs and assumptions that EPA used to generate the costs for each option in Section [ REF \_Ref293045245 \r \h ]. For both options, the cost estimates assume that systems choosing a nontreatment option do so because they have an alternative source that will not require additional water treatment to address changes in raw water quality (i.e., no post-treatment). Because of this, they further assume no incremental waste or residuals management costs.

#### 7.6.2 Model Components and Assumptions for New Wells

In addition to the common WBS direct capital cost items listed in [ REF\_Ref256681355 \h \\* MERGEFORMAT ], the WBS nontreatment model includes the following direct capital cost items specific to the new well option:

- Well casing, screens, and plugs
- Well installation costs including drilling, development, gravel pack, and surface seals
- Well pumps
- Piping (buried) and valves to connect the new well to the system.

It includes the following option-specific O&M elements:

- Operator labor for operating and maintaining well pumps and valves
- Materials for maintaining well pumps
- Energy for operating well pumps.

The option includes a small shed or other low cost building at the well site along with materials and labor for maintenance of this building. In calculating land costs, it incorporates a 100 foot buffer on all sides of the new well building to allow for a sanitary control area around the well,

as recommended in the peer review comments on the model. For new wells, the model includes all of the indirect capital costs shown in [ REF \_Ref256681355 \h \\* MERGEFORMAT ], except for yard piping. The paragraphs below describe specific inputs and assumptions used to generate costs for perchlorate under the new well nontreatment option.

#### Total Flow Rate Requirements and Flow per Well

As with other WBS models, design and average flow are inputs to the nontreatment model. In the case of nontreatment approaches, however, "design" flow is actually the peak flow required by the system, rather than the design capacity of a treatment plant. In the new well nontreatment option, the flow rate requirements determine the number of new wells required. The cost estimates below assume one new well would be installed per 500 gpm of water production capacity required.

#### Well Depth

Well depth will vary for each site depending on the geological formations and aquifer depths. Geophysical studies prior to well installation will provide guidance on optimum well depths. The model has pumps available to serve wells up to 1,350 feet in depth. The cost estimates below assume a 250-foot well depth. The estimates assume 50 percent of this depth is screened, allowing for sections of casing both above and below the well screen.

#### **Pump Type**

The size of the well will depend on the diameter of the pump used to draw water from the aquifer. The model contains three sizes of submersible pumps: 4-, 6-, and 8-inch diameter. Each size can serve a range of flows and depths, so the default size varies across system flows. The cost estimates below assume 4-inch pumps for systems in the smallest size category (25 to 100 people) and 6-inch pumps for larger systems.

#### **Distance from Well to Distribution System**

The distance between a new well and the distribution system affects pipe installation costs. No case studies provided distance information. The cost estimates below assume a default value of 500 feet.

#### 7.6.3 Model Components and Assumptions for Interconnection

In addition to the common WBS direct capital cost items listed in [ REF \_Ref256681355 \h \\* MERGEFORMAT ], the WBS nontreatment model includes the following direct capital cost items specific to the interconnection option:

- Booster pumps or pressure reducing valves (depending on pressure at supply source)
- Concrete vaults (buried) for booster pumps or pressure reducing valves
- Interconnecting piping (buried) and valves.

It includes the following option-specific O&M elements:

- Cost of purchased water
- Operator labor for operating and maintaining booster pumps or pressure reducing valves (depending on pressure at supply source) and interconnecting valves

- Materials for maintaining booster pumps (if required by pressure at supply source)
- Energy for operating booster pumps (if required by pressure at supply source).

The option does not include any buildings. It includes all of the indirect capital costs shown in [ REF \_Ref256681355 \h \\* MERGEFORMAT ], except for yard piping, site work, and architectural fees. The paragraphs below describe specific inputs and assumptions used to generate costs for perchlorate under the interconnection nontreatment option.

#### Flow Rate Requirements

As with other WBS models, design and average flow are inputs to the nontreatment model. In the case of nontreatment approaches, however, "design" flow is actually the peak flow required by the system, rather than the design capacity of a treatment plant. In the interconnection nontreatment option, the flow rate requirements determine a number of system and equipment parameters, including pipeline and valve size and pump capacity and energy use (if required by pressure at the supply source).

#### **Distance to Interconnection Water Supply**

For utilities that have the ability to purchase water from a neighboring system, the capital cost of the interconnection project will depend on the distance between the two systems. If the systems are far apart geographically, the cost of installing a pipeline may be too high to make an interconnection project feasible. Also, a larger booster pump will be required to overcome friction losses along longer pipelines. The cost estimates below assume an average interconnection distance of 10,000 feet, based on comments from the peer review of the nontreatment model.

#### **Pressure at Supply Water Source**

The water pressure of purchased water may require adjustment prior to entering the purchasing system's distribution network (e.g., to account for elevation differences). If the wholesale supplier does not have enough pressure to meet the distribution needs of the interconnection project, then booster pumps are needed to move water from the supply source into the distribution system. The booster pump size is based on flow rate as well as distance and grade to the distribution system. If the supply source has more pressure than necessary then pressure reducing valves are needed. Based on comments from the peer review, the cost estimates below assume that differences in pressure between the supplier and the purchasing system are minimal, so that neither booster pumps nor pressure reducing valves are needed.

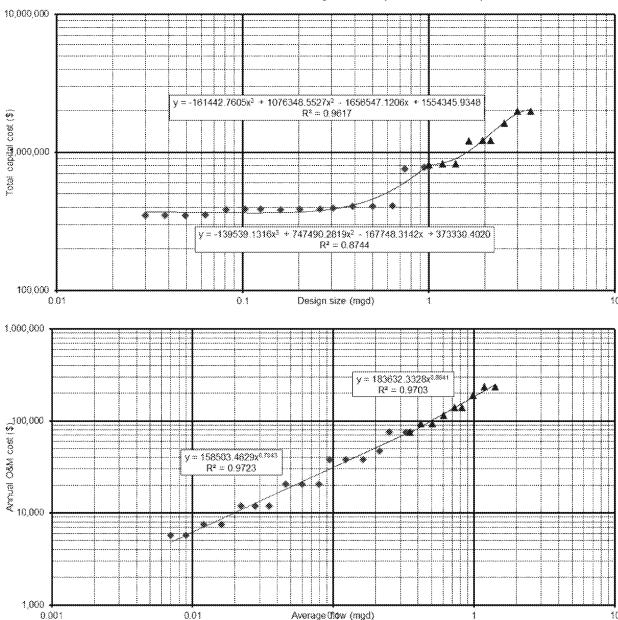
#### Cost of Purchased Water

An interconnection project will include one or more water rates paid to the wholesale system by the purchasing system. The model assumption is a single water rate for the average cost in dollars per thousand gallons of purchased water. Based on data in USEPA (2009), the mean revenue per thousand gallons among all wholesale systems is \$1.85. The cost estimates below assume a higher cost of purchased water of \$2.00 per thousand gallons, based on comments from the peer review.

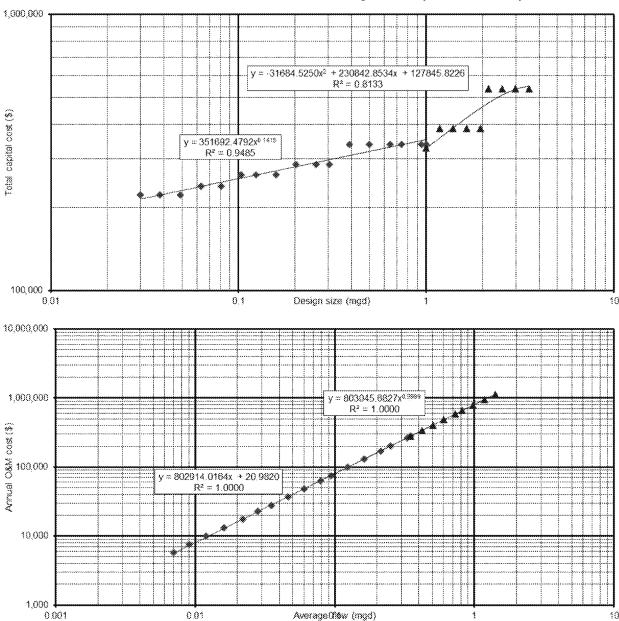
#### 7.6.4 Cost Estimates

The graphs below plot WBS cost model results in 2017 dollars at the mid cost level for the two nontreatment options for systems using groundwater: new wells ([REF\_Ref326066318 \h \\* MERGEFORMAT]) and interconnection ([REF\_Ref326066329 \h \\* MERGEFORMAT]). The exhibits do not extend beyond 3.536 MGD design flow, because the nontreatment model does not generate costs for larger systems. Appendix B provides complete cost equations for both nontreatment options, including the high, mid, and low cost levels and for interconnection of groundwater and surface water systems. Appendix C presents example WBS model outputs for selected flow rates, allowing review of individual cost line items.

Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for New Wells at Groundwater Systems (2017 dollars)



## Exhibit [ STYLEREF 1 \s ]-[ SEQ Exhibit \\* ARABIC \s 1 ]. Mid Cost Results for Interconnection of Groundwater Systems (2017 dollars)



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## **Appendix A: Residuals Treatment**

For certain residuals from technologies removing perchlorate from drinking water, additional treatment may be desirable or required. Specifically, residuals from both ion exchange (see Chapter 2) and membrane technologies (see Chapter 4) may be amenable to treatment prior to their disposal, discharge, or reuse. One form of residuals treatment that has received considerable research attention uses biological treatment (see Chapter 3). Thus, the topic of residuals treatment is relevant to several of the technologies discussed in this document. Because of its cross-technology applicability, residuals treatment research is summarized here in an appendix.

Treatment technologies for residuals from technologies removing perchlorate from drinking water include biological treatment and physical/chemical reduction. <sup>15</sup> This appendix provides an overview of the status of each residuals treatment technology.

## A.1 Biological Treatment of Residuals

Biological treatment of concentrated waste streams from ion exchange processes can be difficult due the microbial toxicity associated with the high salt content of the brine. In the case of an anion exchange process, the regeneration of the resin typically generates a 7 to 12 percent sodium hydroxide brine solution enriched in perchlorate. Gingras and Batista (2002) were unable to adapt a PRB culture to degrade perchlorate in an ion exchange brine. As little as 1 percent sodium hydroxide reduced perchlorate reduction rates by their perchlorate-degrading culture by half (Gingras and Batista, 2002). Batista et al. (2002) indicate that the use of halotolerant microbes that can reduce perchlorate in the presence of high levels of salinity would be required for effective biological brine treatment.

Pilot tests conducted by MWH and the University of Houston, however, evaluated biological treatment of spent brines with promising initial results. A biological system based on a marine sediment inoculum was shown to successfully and consistently remove perchlorate at concentrations of 1.5 to 3.0 mg/L from the brine to below 0.2 mg/L, the goal established for effective reuse of the brine. Investigators observed that as the treated brine was recycled to regenerate the exhausted ion exchange resin, the resin maintained its ability to remove perchlorate to less than 4  $\mu$ g/L. After 30 reuse cycles, various constituents accumulated within the brine, including bicarbonate, sulfate, uranium, arsenic, chromium, fluoride, barium, and silica. These accumulated chemicals are site specific and depend upon raw water quality. Control of this accumulation through additional treatment might be required depending on site-specific concentrations. The researchers identified mixing and maintaining an anoxic environment as key operational requirements. Regular maintenance to remove scale build-up prevents system clogging. The researchers noted that the potential for precipitation and clogging is high under standard operating conditions (Case et al., 2004).

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<sup>&</sup>lt;sup>15</sup> In addition, several sources (Boodoo, 2006; U.S. DoD, 2007; U.S. DoD, 2008a) present information on a potential treatment approach for spent caustic regenerant from an ion exchange system using weak-base resin. This approach uses a small, strong-base resin scavenger bed, with the spent resin from the scavenger bed ultimately disposed by incineration. Because study of this approach has been limited to weak-base resin regenerant, this appendix does not include further discussion of this residuals treatment method.

The literature does not include any detailed evaluation of the feasibility of biological treatment of spent tetrachloroferrate regenerant. Batista et al. (2002) suggest that such a system could be used if pH adjustment were applied to raise the pH of the regenerant. Recent studies of tetrachloroferrate regeneration (Gu et al., 2005; Lutes et al., 2010) do not include examination of biological treatment of spent regenerant.

Batista et al. (2000; 2002) have suggested that a primary advantage of weak-base resins is that caustic solutions used to regenerate these resins may be more amenable to biological treatment than conventional brine solutions. Instead of containing sodium chloride, which can be toxic to some microorganisms, these solutions could contain ammonium hydroxide, which can be used as a nutrient by microbes (Batista et al., 2000). On the other hand, the caustic solutions would require pH adjustment to lower the pH. Also, high levels of ammonium (greater than 0.3 percent) could be toxic to anaerobic biological systems. Ammonium might need to be air stripped or biologically oxidized to nitrate (Batista et al., 2002). Recent studies of regenerated weak-base resins (U.S. DoD, 2007 and 2008a) do not include examination of biological treatment of spent regenerant.

Xu et al. (2003) notes that the waste stream containing perchlorate produced from a RO process contains much lower amount of salts (less than 1 percent) than the sodium hydroxide brine generated using an ion exchange process. Giblin et al. (2002) inoculated a PBR with the pure culture perclace and tested its ability to remove perchlorate from a simulated RO rejectate. The researchers found that this system removed 98 percent of perchlorate from a twice-concentrated rejectate (total dissolved solids of 0.4 percent) with an influent perchlorate concentration of 8 mg/L and a residence time of 2 hours. The system removed nitrate simultaneously with perchlorate from an initial concentration as high as 900 mg/L to below 4 mg/L. Despite the efficiency of perchlorate removal, the system suffered from clogging due to precipitation of the high total dissolved solids of the twice-concentrated rejectate.

## A.2 Physical/Chemical Reduction of Residuals

Calgon Carbon has developed a proprietary physical/chemical brine treatment system called the Perchlorate and Nitrate Destruction Module (PNDM). The PNDM is a high-pressure and high-temperature catalytic process that uses ammonia as a chemical reductant to reduce the nitrate and perchlorate in spent brine (MWH and University of Houston, 2003). In pilot tests that incorporated a nanofiltration unit to remove sulfate from the brine after PNDM treatment, Venkatesh et al. (2000) found that the PNDM system was able to reduce perchlorate in the spent brine from 60,000 to 70,000  $\mu$ g/L to less than the detection limit of 125  $\mu$ g/L. This reduction allowed reuse of the brine and reduced overall waste generation from 1.75 percent of water treated to 0.17 percent of water treated. The remaining waste consisted of the sulfate-laden reject from the nanofiltration step. With 70 mg/L chloride and 50 mg/L sulfate in this reject, the study suggests that blending with treated water is a feasible option (Venkatesh et al., 2000).

The pilot tests conducted by MWH and the University of Houston evaluated the PNDM system for treatment of spent brine from a conventional ion exchange configuration. The biological treatment initial results were promising. PNDM was able to reduce perchlorate in brine to below 0.2 mg/L and allowed reuse of the brine for up to 30 cycles (Case et al., 2004).

The MWH and University of Houston pilot tests also evaluated an electrolytic treatment process to reduce perchlorate in brine. This process, developed by Ionex for the treatment of nitrate contaminated water, is called IXL (MWH and University of Houston, 2003). Initial results, however, demonstrated minimal perchlorate reduction with this system (Case et al., 2004).

Applied Research Associates (ARA) has proposed an Integrated Thermal Treatment Process for treating spent brine. This process would concentrate the spent brine using reverse osmosis, thereby rejecting sulfate and nitrate salts, and concentrating the perchlorate. The concentrated perchlorate would then be thermally destroyed. The treated brine would be suitable for reuse and ARA estimates that the quantity of brine waste for disposal would be reduced by 95 to 99 percent. This thermal treatment approach, however, has been tested only in the laboratory for synthetic spent brine (ARA, 2000).

The Oak Ridge National Laboratory researchers have developed a proprietary methodology for reducing perchlorate in spent tetrachloroferrate regenerant solution. Reportedly, in this methodology, perchlorate decomposes into chloride under certain catalytic conditions within a few hours to one day with an initial perchlorate concentration of about 7,000 mg/L. The process does not otherwise alter the properties of the regenerant solution and allows it to be reused (Gu et al., 2003). More recently, Gu et al. (2005) reported on a process in which tetrachloroferrate regenerant is reduced with ferrous chloride in a thermoreactor. In this process, they observed perchlorate destruction efficiency of 92 to 97 percent, which enabled recycling of the spent regenerant. Lutes et al. (2010) further tested this process in a field-scale reactor and found perchlorate destruction efficiency from 73.6 to greater than 99.7 percent, a median efficiency of greater than 99.2 percent. The process has been patented and licensed exclusively to Calgon Carbon (Lutes et al., 2010).

## **Appendix B: Cost Equations**

#### Notes:

• Cost equations presented here take one of the following forms, identified by which coefficients (C1 through C10) are nonzero:

Cost = 
$$C1 Q^{C2}$$
  
or =  $C3 \text{ Ln}(Q) + C4$   
or =  $C5 e^{(C6 Q)}$   
or =  $C7 Q^3 + C8 Q^2 + C9 Q + C10$ 

where Q is design flow in MGD for total capital costs, or average flow in MGD for annual O&M costs.

• Equations are designated as for small, medium, or large systems. These equations apply as follows:

Small system equations apply where design flow (Q) is less than 1 MGD Medium system equations apply where design flow (Q) is 1 MGD or greater, but less than 10 MGD

Large system equations apply where design flow (Q) is 10 MGD or greater

• For Point of Use/Point of Entry, alternative equations also are included where:

Cost = 
$$CI H^{C2}$$
  
or =  $C3 Ln(H) + C4$   
or =  $C5 e^{(C6 H)}$   
or =  $C7 H^3 + C8 H^2 + C9 H + C10$ 

where H is number of households served for both total capital costs and annual O&M costs.

- EPA developed each equation using the method described in Section [ REF \_Ref351026444 \r \h ].
- Equations are derived from the following data files, with columns rearranged for ease of reference:
  - o Results Summary GW 082817 for groundwater systems
  - o Results Summary SW 082817 for surface water systems.
- For Anion Exchange for Perchlorate, the Perchlorate-selective 170,000 BV scenario corresponds to the Alternative resin (user defined) option in the source data files.
- For Point of Use/Point of Entry, costs do not vary by component level input (high, mid, low); equations are not presented for medium and large systems.

- For Non-treatment, medium system size curves are valid only up to 3.536 MGD design flow (1.417 MGD groundwater average flow and 1.345 MGD surface water average flow); equations are not presented for systems of greater size.
- For Non-treatment, equations are not presented for New Wells for surface water systems, because this option is not likely to be available for surface water systems.

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## **B.1** Capital and O&M Cost Curve Parameters for Anion Exchange Treatment Scenarios

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Perchlorate-selective 250,000 BV	GW	Small	Low	Total Capital	0	0	0	0	0	0	165188.3006	-319172.535	472901.5875	81055.1941	17.74117647
Perchlorate-selective 250,000 BV	GW	Medium	Low	Total Capital	0	0	0	0	0	0	814.5718	-15685.9444	419749.9937	596924.4296	32.34
Perchlorate-selective 250,000 BV	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	281010.0818	993827.2688	33.97647059
Perchlorate-selective 250,000 BV	GW	Small	Mid	Total Capital	0	0	0	0	0	0	111665.366	-213209.9	435508.3073	125329.6749	17.69411765
Perchlorate-selective 250,000 BV	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	802.1603	-17543.4018	455922.7261	687296.3603	31.74
Perchlorate-selective 250,000 BV	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	287995.2658	1130875.789	33.95882353
Perchlorate-selective 250,000 BV	GW	Small	High	Total Capital	0	0	0	0	0	0	186840.2446	-375907.224	708990.975	171244.269	20.24705882
Perchlorate-selective 250,000 BV	GW	Medium	High	Total Capital	0	0	0	0	0	0	2804.3789	-52743.211	810631.6507	852108.8158	33.78
Perchlorate-selective 250,000 BV	GW	Large	High	Total Capital	0	0	0	0	0	0	0	0	444429.0529	1729726.784	34.71764706
Perchlorate-selective 250,000 BV	GW	Small	Low	Annual O&M	0	0	0	0	0	0	60323.2954	-52499.9905	95401.6475	4355.1619	17.74117647
Perchlorate-selective 250,000 BV	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-749.5103	107314.0049	25199.9683	32.34
Perchlorate-selective 250,000 BV	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	94348.8299	54918.7765	33.97647059
Perchlorate-selective 250,000 BV	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	60323.2954	-52499.9905	95401.6475	4355.1619	17.69411765
Perchlorate-selective 250,000 BV	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-793.0138	109785.9654	27443.6982	31.74
Perchlorate-selective 250,000 BV	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	94815.4066	50288.0306	33.95882353
Perchlorate-selective 250,000 BV	GW	Small	High	Annual O&M	0	0	0	0	0	0	0	-22144.6512	92551.7336	4510.6253	20.24705882
Perchlorate-selective 250,000 BV	GW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-864.5234	111139.9758	27196.8415	33.78
Perchlorate-selective 250,000 BV	GW	Large	High	Annual O&M	0	0	0	0	0	0	0	27.2244	93568.467	64877.4787	34.71764706
Perchlorate-selective 170,000 BV	GW	Small	Low	Total Capital	0	0	0	0	0	0	165188.3006	-319172.535	472901.5875	81055.1941	17.74117647
Perchlorate-selective 170,000 BV	GW	Medium	Low	Total Capital	0	0	0	0	0	0	814.5718	-15685.9444	419749.9937	596924.4296	32.34

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	C3	C4	C5	C6	C7	C8	C9	C10	Useful Life
Perchlorate-selective 170,000 BV	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	281010.0818	993827.2688	33.97647059
Perchlorate-selective 170,000 BV	GW	Small	Mid	Total Capital	0	0	0	0	0	0	111665.366	-213209.9	435508.3073	125329.6749	17.69411765
Perchlorate-selective 170,000 BV	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	802.1603	-17543.4018	455922.7261	687296.3603	31.74
Perchlorate-selective 170,000 BV	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	287995.2658	1130875.789	33.95882353
Perchlorate-selective 170,000 BV	GW	Small	High	Total Capital	0	0	0	0	0	0	186840.2446	-375907.224	708990.975	171244.269	20.24705882
Perchlorate-selective 170,000 BV	GW	Medium	High	Total Capital	0	0	0	0	0	0	2804.3789	-52743.211	810631.6507	852108.8158	33.78
Perchlorate-selective 170,000 BV	GW	Large	High	Total Capital	0	0	0	0	0	0	0	0	444429.0529	1729726.784	34.71764706
Perchlorate-selective 170,000 BV	GW	Small	Low	Annual O&M	0	0	0	0	0	0	56351.5026	-50187.5062	122627.1429	4364.8149	17.74117647
Perchlorate-selective 170,000 BV	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	0	131580.5513	27443.0764	32.34
Perchlorate-selective 170,000 BV	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	122034.7577	53803.0803	33.97647059
Perchlorate-selective 170,000 BV	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	56351.5026	-50187.5062	122627.1429	4364.8149	17.69411765
Perchlorate-selective 170,000 BV	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	0	133861.2533	29816.6188	31.74
Perchlorate-selective 170,000 BV	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	122526.36	48879.5863	33.95882353
Perchlorate-selective 170,000 BV	GW	Small	High	Annual O&M	0	0	0	0	0	0	57937.7322	-51428.4226	123700.2943	4441.0268	20.24705882
Perchlorate-selective 170,000 BV	GW	Medium	High	Annual O&M	0	0	0	0	0	0	-751.2236	4495.5752	128358.2051	31778.107	33.78
Perchlorate-selective 170,000 BV	GW	Large	High	Annual O&M	0	0	0	0	0	0	0	0	123508.7003	38942.7634	34.71764706
Perchlorate-selective 250,000 BV	SW	Small	Low	Total Capital	0	0	0	0	0	0	165274.688	-319329.202	472978.9565	81058.2965	17.74117647
Perchlorate-selective 250,000 BV	SW	Medium	Low	Total Capital	0	0	0	0	0	0	839.1878	-16075.29	421154.5885	595626.0307	32.34
Perchlorate-selective 250,000 BV	SW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	280265.6113	1007900.297	33.98823529
Perchlorate-selective 250,000 BV	SW	Small	Mid	Total Capital	0	0	0	0	0	0	111750.4602	-213363.524	435584.8953	125332.7918	17.69411765
Perchlorate-selective 250,000 BV	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	830.5377	-17999.3282	457555.6503	685792.943	31.74

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	C3	C4	C5	C6	C7	C8	C9	C10	Useful Life
Perchlorate-selective 250,000 BV	SW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	287250.061	1144962.062	33.97058824
Perchlorate-selective 250,000 BV	SW	Small	High	Total Capital	0	0	0	0	0	0	186942.4138	-376095.796	709087.0242	171247.8793	20.24705882
Perchlorate-selective 250,000 BV	SW	Medium	High	Total Capital	0	0	0	0	0	0	2838.7252	-53286.93	812529.4756	850397.7787	33.78
Perchlorate-selective 250,000 BV	SW	Large	High	Total Capital	0	0	0	0	0	0	0	0	443634.4238	1743834.45	34.72941176
Perchlorate-selective 250,000 BV	SW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-18741.0467	90566.6107	4395.0332	17.74117647
Perchlorate-selective 250,000 BV	SW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-598.7219	109065.4542	24276.2342	32.34
Perchlorate-selective 250,000 BV	SW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	95522.9118	71724.5663	33.98823529
Perchlorate-selective 250,000 BV	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-18741.0467	90566.6107	4395.0332	17.69411765
Perchlorate-selective 250,000 BV	SW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	0	109075.9181	28101.0967	31.74
Perchlorate-selective 250,000 BV	sw	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	96019.7175	67473.7023	33.97058824
Perchlorate-selective 250,000 BV	sw	Small	High	Annual O&M	0	0	0	0	0	0	0	-19036.6668	91488.2917	4470.8681	20.24705882
Perchlorate-selective 250,000 BV	sw	Medium	High	Annual O&M	0	0	0	0	0	0	-1039.9695	6031.209	101107.6527	31206.5711	33.78
Perchlorate-selective 250,000 BV	sw	Large	High	Annual O&M	0	0	0	0	0	0	0	0	97012.5603	58878.3927	34.72941176
Perchlorate-selective 170,000 BV	sw	Small	Low	Total Capital	0	0	0	0	0	0	165274.688	-319329.202	472978.9565	81058.2965	17.74117647
Perchlorate-selective 170,000 BV	sw	Medium	Low	Total Capital	0	0	0	0	0	0	839.1878	-16075.29	421154.5885	595626.0307	32.34
Perchlorate-selective 170,000 BV	SW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	280265.6113	1007900.297	33.98823529
Perchlorate-selective 170,000 BV	sw	Small	Mid	Total Capital	0	0	0	0	0	0	111750.4602	-213363.524	435584.8953	125332.7918	17.69411765
Perchlorate-selective 170,000 BV	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	830.5377	-17999.3282	457555.6503	685792.943	31.74
Perchlorate-selective 170,000 BV	SW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	287250.061	1144962.062	33.97058824
Perchlorate-selective 170,000 BV	SW	Small	High	Total Capital	0	0	0	0	0	0	186942.4138	-376095.796	709087.0242	171247.8793	20.24705882
Perchlorate-selective 170,000 BV	SW	Medium	High	Total Capital	0	0	0	0	0	0	2838.7252	-53286.93	812529.4756	850397.7787	33.78

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	С9	C10	Useful Life
Perchlorate-selective 170,000 BV	SW	Large	High	Total Capital	0	0	0	0	0	0	0	0	443634.4238	1743834.45	34.72941176
Perchlorate-selective 170,000 BV	SW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-18636.7959	118095.4167	4401.0983	17.74117647
Perchlorate-selective 170,000 BV	SW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-594.8003	136645.3404	24283.6711	32.34
Perchlorate-selective 170,000 BV	SW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	123198.8697	70820.6621	33.98823529
Perchlorate-selective 170,000 BV	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-18636.7959	118095.4167	4401.0983	17.69411765
Perchlorate-selective 170,000 BV	SW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	0	136673.7625	28097.7332	31.74
Perchlorate-selective 170,000 BV	SW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	123719.8582	66316.3849	33.97058824
Perchlorate-selective 170,000 BV	SW	Small	High	Annual O&M	0	0	0	0	0	0	0	-18919.6285	119009.5802	4477.345	20.24705882
Perchlorate-selective 170,000 BV	SW	Medium	High	Annual O&M	0	0	0	0	0	0	-1038.7155	6027.6327	128706.7892	31207.6465	33.78
Perchlorate-selective 170,000 BV	SW	Large	High	Annual O&M	0	0	0	0	0	0	0	0	124761.0668	57213.3622	34.72941176

Cost = C1 \* Q ^ C2 + C3 \* Ln(Q) + C4 + C5 \* Exp (C6 \* Q) + C7 \* Q^3 + C8 \* Q^2 + C9 \* Q + C10

Where Q is design flow in MGD for total capital costs and average flow in MGD for annual O&M costs

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## **B.2** Capital and O&M Cost Curve Parameters for Biological Treatment Scenarios

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	С7	C8	C9	C10	Useful Life
Fixed Bed Pressure Vessel	GW	Small	Low	Total Capital	0	0	0	0	0	0	360664.2489	-719515.71	1107604.628	643541.8543	24.46470588
Fixed Bed Pressure Vessel	GW	Medium	Low	Total Capital	0	0	0	0	0	0	-1230.8188	4236.7079	740451.7857	1652605.61	31.79333333
Fixed Bed Pressure Vessel	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	-33.3723	558954.1451	2437567.235	31.31764706
Fixed Bed Pressure Vessel	GW	Small	Mid	Total Capital	0	0	0	0	0	0	492476.6413	-1025849.33	1436856.86	715106.4333	23.12941176
Fixed Bed Pressure Vessel	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	0	-15229.9667	887780.101	1850883.189	31.34
Fixed Bed Pressure Vessel	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	564980.0264	3622086.493	31.35882353
Fixed Bed Pressure Vessel	GW	Small	High	Total Capital	0	0	0	0	0	0	770133.6573	-1683703.84	2371838.543	836853.9079	25.47647059
Fixed Bed Pressure Vessel	GW	Medium	High	Total Capital	0	0	0	0	0	0	-2465.7652	13844.2369	1241563.686	2576024.887	33.52
Fixed Bed Pressure Vessel	GW	Large	High	Total Capital	0	0	0	0	0	0	7.225	-1535.1083	1011253.121	3406280.863	33.12352941
Fixed Bed Pressure Vessel	GW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-45961.3314	158156.0739	27926.5161	24.46470588
Fixed Bed Pressure Vessel	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	1184.1123	-14333.9198	180043.52	62255.524	31.79333333
Fixed Bed Pressure Vessel	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	126538.9848	104543.6316	31.31764706
Fixed Bed Pressure Vessel	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-76492.4588	176417.3995	27695.5647	23.12941176

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	С3	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fixed Bed Pressure Vessel	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-6827.1527	173032.9405	75031.6456	31.34
Fixed Bed Pressure Vessel	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	129201.3891	88438.183	31.35882353
Fixed Bed Pressure Vessel	GW	Small	High	Annual O&M	0	0	0	0	0	0	91643.6805	-120401.753	185483.6022	28225.6213	25.47647059
Fixed Bed Pressure Vessel	GW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-7136.1131	175890.2434	75595.8595	33.52
Fixed Bed Pressure Vessel	GW	Large	High	Annual O&M	0	0	0	0	0	0	-2.6058	349.0829	122625.9002	143270.435	33.12352941
Fixed Bed Gravity Basin	GW	Small	Low	Total Capital	0	0	0	0	0	0	-199338.503	108033.2323	1102079.756	943520.528	28.67058824
Fixed Bed Gravity Basin	GW	Medium	Low	Total Capital	2610057.586	0.5365	0	0	0	0	0	0	0	0	32.84
Fixed Bed Gravity Basin	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	369737.988	7037405.771	31.62352941
Fixed Bed Gravity Basin	GW	Small	Mid	Total Capital	0	0	0	0	0	0	0	-243851.42	1460419.766	1052522.957	27.42941176
Fixed Bed Gravity Basin	GW	Medium	Mid	Total Capital	2931011.596	0.534	0	0	0	0	0	0	0	0	32.36666667
Fixed Bed Gravity Basin	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	-142.8363	409982.8163	7189477.16	31.61764706
Fixed Bed Gravity Basin	GW	Small	High	Total Capital	0	0	0	0	0	0	202978.9468	-743201.86	2218774.508	1232771.998	29.86470588
Fixed Bed Gravity Basin	GW	Medium	High	Total Capital	3735296.534	0.5134	0	0	0	0	0	0	0	0	33.99333333
Fixed Bed Gravity Basin	GW	Large	High	Total Capital	0	0	0	0	0	0	0	-127.9367	472526.5562	9035136.284	32.77058824

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	С3	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fixed Bed Gravity Basin	GW	Small	Low	Annual O&M	0	0	0	0	0	0	-244584.565	45607.9876	217316.197	33588.7142	28.67058824
Fixed Bed Gravity Basin	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-6559.4032	186687.2639	75555.8157	32.84
Fixed Bed Gravity Basin	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	118304.6266	225492.918	31.62352941
Fixed Bed Gravity Basin	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	-185910.722	21574.4589	226502.4065	35193.5095	27.42941176
Fixed Bed Gravity Basin	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-7143.548	194111.8555	77654.2262	32.36666667
Fixed Bed Gravity Basin	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	120590.8015	210480.3335	31.61764706
Fixed Bed Gravity Basin	GW	Small	High	Annual O&M	0	0	0	0	0	0	0	-76893.7765	246083.0756	36218.5012	29.86470588
Fixed Bed Gravity Basin	GW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-7101.1547	196805.7999	79557.3458	33.99333333
Fixed Bed Gravity Basin	GW	Large	High	Annual O&M	0	0	0	0	0	0	-1.6021	227.8999	115719.1797	264822.2223	32.77058824
Fluidized Bed Pressure Vessel	GW	Small	Low	Total Capital	0	0	0	0	0	0	-165867.923	-28166.6787	1025869.936	861046.6691	27.07647059
Fluidized Bed Pressure Vessel	GW	Medium	Low	Total Capital	0	0	0	0	0	0	-1758.0291	18962.1919	649400.0564	1694269.883	31.5
Fluidized Bed Pressure Vessel	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	-184.9186	540273.6829	2881509.567	30.98823529
Fluidized Bed	GW	Small	Mid	Total Capital	0	0	0	0	0	0	0	-362080.784	1369255.056	997173.1712	25.97058824

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	C3	C4	C5	C6	C7	C8	C9	C10	Useful Life
Pressure Vessel															
Fluidized Bed Pressure Vessel	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	-1020.4755	5469.6152	807095.7564	1937384.124	31
Fluidized Bed Pressure Vessel	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	-199.5054	571178.1483	3627131.011	30.94705882
Fluidized Bed Pressure Vessel	GW	Small	High	Total Capital	0	0	0	0	0	0	0	-575922.439	2258409.846	1191761.253	28.81764706
Fluidized Bed Pressure Vessel	GW	Medium	High	Total Capital	0	0	0	0	0	0	-2245.5247	23231.221	1173986.54	2689612.461	33.32
Fluidized Bed Pressure Vessel	GW	Large	High	Total Capital	0	0	0	0	0	0	2.0713	-750.9524	958369.9281	4502384.894	33.18823529
Fluidized Bed Pressure Vessel	GW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-61439.932	156677.6914	37218.503	27.07647059
Fluidized Bed Pressure Vessel	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-4273.302	151726.0086	77431.6855	31.5
Fluidized Bed Pressure Vessel	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	118931.5849	156492.319	30.98823529
Fluidized Bed Pressure Vessel	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-66125.2755	165892.0237	42644.1251	25.97058824
Fluidized Bed Pressure Vessel	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-4768.8748	158659.8589	83835.7669	31

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fluidized Bed Pressure Vessel	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	121214.2539	142830.8114	30.94705882
Fluidized Bed Pressure Vessel	GW	Small	High	Annual O&M	0	0	0	0	0	0	0	-69019.5329	173487.8271	46943.947	28.81764706
Fluidized Bed Pressure Vessel	GW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-5090.8845	161873.9793	83364.5317	33.32
Fluidized Bed Pressure Vessel	GW	Large	High	Annual O&M	0	0	0	0	0	0	0	0	125753.2694	116080.0404	33.18823529
Fixed Bed Pressure Vessel	SW	Small	Low	Total Capital	0	0	0	0	0	0	376530.23	-743505.873	1116042.748	643296.522	24.46470588
Fixed Bed Pressure Vessel	SW	Medium	Low	Total Capital	0	0	0	0	0	0	-1306.1553	5435.3388	734513.1972	1657773.704	31.79333333
Fixed Bed Pressure Vessel	SW	Large	Low	Total Capital	0	0	0	0	0	0	-1.3562	301.9828	536580.3031	2719357.571	31.31764706
Fixed Bed Pressure Vessel	SW	Small	Mid	Total Capital	0	0	0	0	0	0	376530.23	-743505.873	1116042.748	643296.522	24.46470588
Fixed Bed Pressure Vessel	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	-1306.1553	5435.3388	734513.1972	1657773.704	31.79333333
Fixed Bed Pressure Vessel	SW	Large	Mid	Total Capital	0	0	0	0	0	0	-1.3562	301.9828	536580.3031	2719357.571	31.31764706
Fixed Bed Pressure Vessel	SW	Small	High	Total Capital	0	0	0	0	0	0	787557.8629	-1711551.3	2382625.539	836532.0821	25.47647059
Fixed Bed Pressure Vessel	sw	Medium	High	Total Capital	0	0	0	0	0	0	-2509.418	14699.0305	1235969.914	2580751.403	33.52

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fixed Bed Pressure Vessel	sw	Large	High	Total Capital	0	0	0	0	0	0	7.2587	-1541.3548	1010264.132	3400477.674	33.11764706
Fixed Bed Pressure Vessel	SW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-30006.0913	151948.7891	27786.4856	24.46470588
Fixed Bed Pressure Vessel	SW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-6472.2083	171959.0723	65214.7013	31.79333333
Fixed Bed Pressure Vessel	SW	Large	Low	Annual O&M	0	0	0	0	0	0	0	-50.0737	133952.7487	116880.3955	31.31764706
Fixed Bed Pressure Vessel	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-47426.4264	167113.6865	27582.7544	23.11764706
Fixed Bed Pressure Vessel	SW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-7571.6831	182100.1207	70400.4121	31.33333333
Fixed Bed Pressure Vessel	SW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	-48.638	136680.3001	104187.003	31.35882353
Fixed Bed Pressure Vessel	SW	Small	High	Annual O&M	0	0	0	0	0	0	0	-54988.4534	172649.2074	28155.967	25.47647059
Fixed Bed Pressure Vessel	SW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-7944.4304	185272.3046	70814.7398	33.52
Fixed Bed Pressure Vessel	SW	Large	High	Annual O&M	0	0	0	0	0	0	0	-47.6434	142797.7639	76278.2943	33.11764706
Fixed Bed Gravity Basin	SW	Small	Low	Total Capital	0	0	0	0	0	0	-194555.784	103163.0231	1100879.375	944940.1123	28.66470588
Fixed Bed Gravity Basin	SW	Medium	Low	Total Capital	2611597.865	0.5352	0	0	0	0	0	0	0	0	32.84
Fixed Bed Gravity Basin	SW	Large	Low	Total Capital	0	0	0	0	0	0	0	-97.4647	385856.916	6442653.679	31.6
Fixed Bed Gravity Basin	sw	Small	Mid	Total Capital	0	0	0	0	0	0	0	-241076.609	1455779.615	1054413.219	27.42941176

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fixed Bed Gravity Basin	sw	Medium	Mid	Total Capital	2932533.852	0.5325	0	0	0	0	0	0	0	0	32.36666667
Fixed Bed Gravity Basin	SW	Large	Mid	Total Capital	0	0	0	0	0	0	1.4957	-485.1302	428334.3343	6915717.611	31.59411765
Fixed Bed Gravity Basin	SW	Small	High	Total Capital	0	0	0	0	0	0	0	-437897.61	2100725.869	1242829.494	29.86470588
Fixed Bed Gravity Basin	SW	Medium	High	Total Capital	3737297.463	0.5121	0	0	0	0	0	0	0	0	33.98666667
Fixed Bed Gravity Basin	SW	Large	High	Total Capital	0	0	0	0	0	0	-2.2055	429.0556	433942.163	9481048.78	32.74117647
Fixed Bed Gravity Basin	SW	Small	Low	Annual O&M	0	0	0	0	0	0	-295094.821	107422.9682	200339.6042	33528.7699	28.66470588
Fixed Bed Gravity Basin	SW	Medium	Low	Annual O&M	0	0	0	0	0	0	0	-7078.7184	196694.4742	70481.3509	32.84
Fixed Bed Gravity Basin	SW	Large	Low	Annual O&M	0	0	0	0	0	0	0	-53.1455	125332.4549	225578.8415	31.6
Fixed Bed Gravity Basin	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-39781.1639	225091.591	34770.0602	27.42941176
Fixed Bed Gravity Basin	SW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-7922.9458	205415.3033	71952.8472	32.36666667
Fixed Bed Gravity Basin	SW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	-49.281	127203.5984	221133.0458	31.59411765
Fixed Bed Gravity Basin	SW	Small	High	Annual O&M	0	0	0	0	0	0	-265504.07	96891.6626	214451.7462	36419.1032	29.86470588
Fixed Bed Gravity Basin	SW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-7678.6964	207592.0873	74095.0996	33.98666667
Fixed Bed Gravity Basin	sw	Large	High	Annual O&M	0	0	0	0	0	0	0	0	127770.6593	245595.9535	32.74117647

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Fluidized Bed Pressure Vessel	SW	Small	Low	Total Capital	0	0	0	0	0	0	-175859.574	-15604.5734	1022532.417	861605.1699	27.08235294
Fluidized Bed Pressure Vessel	SW	Medium	Low	Total Capital	0	0	0	0	0	0	-1853.4978	20390.6304	642914.6996	1699906.968	31.5
Fluidized Bed Pressure Vessel	SW	Large	Low	Total Capital	0	0	0	0	0	0	0	-184.3819	539381.2799	2877427.798	30.99411765
Fluidized Bed Pressure Vessel	SW	Small	Mid	Total Capital	0	0	0	0	0	0	97225.5085	-508869.031	1424987.278	993856.3133	25.97647059
Fluidized Bed Pressure Vessel	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	0	-11461.5583	883976.9293	1850411.667	30.99333333
Fluidized Bed Pressure Vessel	SW	Large	Mid	Total Capital	0	0	0	0	0	0	0	-197.1765	569947.5394	3626369.958	30.95294118
Fluidized Bed Pressure Vessel	SW	Small	High	Total Capital	0	0	0	0	0	0	0	-579364.841	2261637.972	1192128.15	28.81764706
Fluidized Bed Pressure Vessel	SW	Medium	High	Total Capital	0	0	0	0	0	0	-2497.6049	26750.0461	1159072.431	2703008.262	33.32
Fluidized Bed Pressure Vessel	SW	Large	High	Total Capital	0	0	0	0	0	0	0	-236.2879	924354.1605	4920097.242	33.19411765
Fluidized Bed Pressure Vessel	SW	Small	Low	Annual O&M	0	0	0	0	0	0	0	-48119.3527	151722.3833	37060.3885	27.08235294
Fluidized Bed	SW	Medium	Low	Annual O&M	0	0	0	0	0	0	683.1623	-8992.2086	166147.0816	70692.5368	31.5

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
Pressure Vessel															
Fluidized Bed Pressure Vessel	SW	Large	Low	Annual O&M	0	0	0	0	0	0	0	-77.7363	127503.5199	136708.2479	30.99411765
Fluidized Bed Pressure Vessel	sw	Small	Mid	Annual O&M	0	0	0	0	0	0	0	-51241.3715	160339.9151	42473.2257	25.97647059
Fluidized Bed Pressure Vessel	SW	Medium	Mid	Annual O&M	0	0	0	0	0	0	0	-5128.5574	165688.954	80177.5823	30.99333333
Fluidized Bed Pressure Vessel	sw	Large	Mid	Annual O&M	0	0	0	0	0	0	0	-76.4289	129834.5381	126303.1959	30.95294118
Fluidized Bed Pressure Vessel	sw	Small	High	Annual O&M	0	0	0	0	0	0	0	-52938.0151	167468.4034	46763.9136	28.81764706
Fluidized Bed Pressure Vessel	SW	Medium	High	Annual O&M	0	0	0	0	0	0	0	-5503.3764	169209.2828	79563.3081	33.32
Fluidized Bed Pressure Vessel	SW	Large	High	Annual O&M	0	0	0	0	0	0	0	-69.7757	134187.4415	107314.9692	33.19411765

Cost = C1 \* Q ^ C2 + C3 \* Ln(Q) + C4 + C5 \* Exp (C6 \* Q) + C7 \* Q^3 + C8 \* Q^2 + C9 \* Q + C10
Where Q is design flow in MGD for total capital costs and average flow in MGD for annual O&M costs

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## **B.3** Capital and O&M Cost Curve Parameters for Reverse Osmosis Treatment Scenarios

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
High Qual GW	GW	Small	Low	Total Capital	0	0	0	0	0	0	419730.7281	-788209.698	1051419.014	442510.3593	24.25882353
High Qual GW	GW	Medium	Low	Total Capital	0	0	0	0	0	0	-2786.1614	27165.9297	609019.2816	834956.3855	30.08666667
High Qual GW	GW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	501464.6241	2255690.684	29.34117647
High Qual GW	GW	Small	Mid	Total Capital	0	0	0	0	0	0	922912.2899	-1682538.35	1663351.386	507008.18	21.88823529
High Qual GW	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	-2109.8121	15519.4438	693094.0443	1122924.536	28.22
High Qual GW	GW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	512376.9598	2510945.991	29.54705882
High Qual GW	GW	Small	High	Total Capital	0	0	0	0	0	0	766189.679	-1380227.98	1525573.733	570809.2307	24.37058824
High Qual GW	GW	Medium	High	Total Capital	0	0	0	0	0	0	-2107.3138	14031.7866	729255.4764	1165183.835	29.69333333
High Qual GW	GW	Large	High	Total Capital	0	0	0	0	0	0	0	43.302	528875.1103	2776676.683	29.88823529
High Qual GW	GW	Small	Low	Annual O&M	0	0	0	0	0	0	6656243.564	-3786089.15	1334035.193	36717.6114	24.25882353
High Qual GW	GW	Medium	Low	Annual O&M	0	0	0	0	0	0	14623.6692	-136584.824	1095617.882	-57620.9392	30.08666667
High Qual GW	GW	Large	Low	Annual O&M	0	0	0	0	0	0	0	0	665521.2229	522284.0316	29.34117647
High Qual GW	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	6656243.564	-3786089.15	1334035.193	36717.6114	21.88823529
High Qual GW	GW	Medium	Mid	Annual O&M	0	0	0	0	0	0	14623.6692	-136584.824	1095617.882	-57620.9392	28.22
High Qual GW	GW	Large	Mid	Annual O&M	0	0	0	0	0	0	0	0	665521.2229	522284.0316	29.54705882
High Qual GW	GW	Small	High	Annual O&M	0	0	0	0	0	0	6656243.564	-3786089.15	1334035.193	36717.6114	24.37058824
High Qual GW	GW	Medium	High	Annual O&M	0	0	0	0	0	0	14623.6692	-136584.824	1095617.882	-57620.9392	29.69333333
High Qual GW	GW	Large	High	Annual O&M	0	0	0	0	0	0	0	0	665521.2229	522284.0316	29.88823529
High Qual SW	SW	Small	Low	Total Capital	0	0	0	0	0	0	149193.1749	-494226.186	1100191.755	425799.664	24.38235294
High Qual SW	sw	Medium	Low	Total Capital	0	0	0	0	0	0	0	-20031.7053	832358.7573	712738.9352	30.41333333

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
High Qual SW	SW	Large	Low	Total Capital	0	0	0	0	0	0	0	0	471323.5873	2660282.067	30.1
High Qual SW	SW	Small	Mid	Total Capital	0	0	0	0	0	0	710187.6349	-1479543.01	1773005.169	482742.038	22.09411765
High Qual SW	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	992.0306	-36548.5993	935963.6527	986641.0663	28.54666667
High Qual SW	SW	Large	Mid	Total Capital	0	0	0	0	0	0	0	0	482545.7284	2941963.972	30.31764706
High Qual SW	SW	Small	High	Total Capital	0	0	0	0	0	0	521778.1744	-1165763.6	1665430.871	535877.9142	24.41764706
High Qual SW	SW	Medium	High	Total Capital	0	0	0	0	0	0	0	-22177.6296	904905.0085	1106566.125	30.02666667
High Qual SW	sw	Large	High	Total Capital	0	0	0	0	0	0	0	0	506185.1909	3060324.078	30.66470588
High Qual SW	sw	Small	Low	Annual O&M	0	0	0	0	0	0	6589736.981	-3985129.69	1508433.457	36750.2247	24.38235294
High Qual SW	sw	Medium	Low	Annual O&M	0	0	0	0	0	0	57128.2822	-406859.935	1486200.795	-119209.519	30.41333333
High Qual SW	SW	Large	Low	Annual O&M	0	0	0	0	0	0	11.5056	-1617.309	643746.394	390169.1403	30.1
High Qual SW	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	6589736.981	-3985129.69	1508433.457	36750.2247	22.09411765
High Qual SW	sw	Medium	Mid	Annual O&M	0	0	0	0	0	0	57128.2822	-406859.935	1486200.795	-119209.519	28.54666667
High Qual SW	SW	Large	Mid	Annual O&M	0	0	0	0	0	0	11.5056	-1617.309	643746.394	390169.1403	30.31764706
High Qual SW	SW	Small	High	Annual O&M	0	0	0	0	0	0	6589736.981	-3985129.69	1508433.457	36750.2247	24.41764706
High Qual SW	SW	Medium	High	Annual O&M	0	0	0	0	0	0	57128.2822	-406859.935	1486200.795	-119209.519	30.02666667
High Qual SW	SW	Large	High	Annual O&M	0	0	0	0	0	0	11.5056	-1617.309	643746.394	390169.1403	30.66470588

Cost = C1 \* Q ^ C2 + C3 \* Ln(Q) + C4 + C5 \* Exp (C6 \* Q) + C7 \* Q^3 + C8 \* Q^2 + C9 \* Q + C10
Where Q is design flow in MGD for total capital costs and average flow in MGD for annual O&M costs

## B.4 Capital and O&M Cost Curve Parameters for Point-of-Use Treatment Scenarios (Flow Basis)

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	С7	C8	C9	C10	Useful Life
POU Reverse Osmosis	GW	Small	n/a	Total Capital	552638.6724	1.0352	0	0	0	0	0	0	0	0	10
POU Reverse Osmosis	GW	Small	n/a	Annual O&M	0	0	0	0	0	0	0	-99260.9973	496421.4345	1775.6587	10
POU Reverse Osmosis	sw	Small	n/a	Total Capital	555487.2438	1.0462	0	0	0	0	0	0	0	0	10
POU Reverse Osmosis	sw	Small	n/a	Annual O&M	0	0	0	0	0	0	0	0	460080.1825	688.6075	10

Cost = C1 \* Q ^ C2 + C3 \* Ln(Q) + C4 + C5 \* Exp (C6 \* Q) + C7 \* Q^3 + C8 \* Q^2 + C9 \* Q + C10

Where Q is design flow in MGD for total capital costs and average flow in MGD for annual O&M costs

# B.5 Capital and O&M Cost Curve Parameters for Point-of-Use Treatment Scenarios (Household Basis)

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	С7	C8	C9	C10	Useful Life
POU Reverse Osmosis	GW	Small	n/a	Total Capital	607.2278	0.987	0	0	0	0	0	0	0	0	10
POU Reverse Osmosis	GW	Small	n/a	Annual O&M	0	0	0	0	0	0	0	0	163.7089	782.86	10
POU Reverse Osmosis	SW	Small	n/a	Total Capital	608.0877	0.9868	0	0	0	0	0	0	0	0	10
POU Reverse Osmosis	sw	Small	n/a	Annual O&M	0	0	0	0	0	0	0	0	163.7627	788.7267	10

Cost = C1 \* H ^ C2 + C3 \* Ln(H) + C4 + C5 \* Exp (C6 \* H) + C7 \* H^3 + C8 \* H^2 + C9 \* H + C10

Where H is number of households served

## **B.6** Capital and O&M Cost Curve Parameters for Non-Treatment Scenarios

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	C3	C4	C5	C6	С7	C8	C9	C10	Useful Life
Interconnection	GW	Small	Low	Total Capital	338077.9484	0.1356	0	0	0	0	0	0	0	0	17
Interconnection	GW	Medium	Low	Total Capital	0	0	0	0	0	0	0	-31084.5197	227314.1808	118001.2476	21.98888889
Interconnection	GW	Small	Mid	Total Capital	351692.4792	0.1415	0	0	0	0	0	0	0	0	16.92941176
Interconnection	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	0	-31684.525	230842.8534	127845.8226	21.74444444
Interconnection	GW	Small	High	Total Capital	353452.9305	0.1363	0	0	0	0	0	0	0	0	17
Interconnection	GW	Medium	High	Total Capital	0	0	0	0	0	0	0	-32984.1112	240477.5383	121870.7095	21.9
Interconnection	GW	Small	Low	Annual O&M	0	0	0	0	0	0	0	0	802914.0164	20.982	17
Interconnection	GW	Medium	Low	Annual O&M	803045.6827	0.9999	0	0	0	0	0	0	0	0	21.98888889
Interconnection	GW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	0	802914.0164	20.982	16.92941176
Interconnection	GW	Medium	Mid	Annual O&M	803045.6827	0.9999	0	0	0	0	0	0	0	0	21.74444444
Interconnection	GW	Small	High	Annual O&M	0	0	0	0	0	0	0	0	802914.0164	20.982	17
Interconnection	GW	Medium	High	Annual O&M	803045.6827	0.9999	0	0	0	0	0	0	0	0	21.9
New Well Construction	GW	Small	Low	Total Capital	0	0	0	0	0	0	0	293739.0823	-56575.0209	187426.9077	17.1
New Well Construction	GW	Medium	Low	Total Capital	0	0	0	0	0	0	0	-7446.9958	329941.3709	82930.8255	22.41111111
New Well Construction	GW	Small	Mid	Total Capital	0	0	0	0	0	0	-139539.132	747490.2819	-167748.314	373330.402	33.86470588
New Well Construction	GW	Medium	Mid	Total Capital	0	0	0	0	0	0	-161442.761	1076348.553	-1656547.12	1554345.935	39.21111111
New Well Construction	GW	Small	High	Total Capital	0	0	0	0	0	0	-186326.987	850714.244	-198767.813	384792.9094	34.31764706
New Well Construction	GW	Medium	High	Total Capital	0	0	0	0	0	0	-170156.492	1136151.351	-1756631.22	1640223.951	39.38888889
New Well Construction	GW	Small	Low	Annual O&M	158503.4629	0.7043	0	0	0	0	0	0	0	0	17.1
New Well Construction	GW	Medium	Low	Annual O&M	183632.3328	0.8641	0	0	0	0	0	0	0	0	22.41111111

Design	GW/ SW	Size Category	Comp Level	Cost Type	C1	C2	СЗ	C4	C5	C6	C7	C8	C9	C10	Useful Life
New Well Construction	GW	Small	Mid	Annual O&M	158503.4629	0.7043	0	0	0	0	0	0	0	0	33.86470588
New Well Construction	GW	Medium	Mid	Annual O&M	183632.3328	0.8641	0	0	0	0	0	0	0	0	39.21111111
New Well Construction	GW	Small	High	Annual O&M	158503.4629	0.7043	0	0	0	0	0	0	0	0	34.31764706
New Well Construction	GW	Medium	High	Annual O&M	183632.3328	0.8641	0	0	0	0	0	0	0	0	39.38888889
Interconnection	SW	Small	Low	Total Capital	338077.9484	0.1356	0	0	0	0	0	0	0	0	17
Interconnection	SW	Medium	Low	Total Capital	0	0	0	0	0	0	0	-31084.5197	227314.1808	118001.2476	21.98888889
Interconnection	SW	Small	Mid	Total Capital	351692.4792	0.1415	0	0	0	0	0	0	0	0	16.92941176
Interconnection	SW	Medium	Mid	Total Capital	0	0	0	0	0	0	0	-31684.525	230842.8534	127845.8226	21.74444444
Interconnection	SW	Small	High	Total Capital	353452.9305	0.1363	0	0	0	0	0	0	0	0	17
Interconnection	SW	Medium	High	Total Capital	0	0	0	0	0	0	0	-32984.1112	240477.5383	121870.7095	21.9
Interconnection	SW	Small	Low	Annual O&M	0	0	0	0	0	0	0	0	802695.4	116.2102	17
Interconnection	SW	Medium	Low	Annual O&M	803045.4022	0.9999	0	0	0	0	0	0	0	0	21.98888889
Interconnection	SW	Small	Mid	Annual O&M	0	0	0	0	0	0	0	0	802695.4	116.2102	16.92941176
Interconnection	SW	Medium	Mid	Annual O&M	803045.4022	0.9999	0	0	0	0	0	0	0	0	21.74444444
Interconnection	SW	Small	High	Annual O&M	0	0	0	0	0	0	0	0	802695.4	116.2102	17
Interconnection	SW	Medium	High	Annual O&M	803045.4022	0.9999	0	0	0	0	0	0	0	0	21.9

Cost = C1 \* Q ^ C2 + C3 \* Ln(Q) + C4 + C5 \* Exp (C6 \* Q) + C7 \* Q^3 + C8 \* Q^2 + C9 \* Q + C10

Where Q is design flow in MGD for total capital costs and average flow in MGD for annual O&M costs

## **Appendix C: Example WBS Model Outputs**

#### Notes:

- Example outputs presented here correspond to treatment of groundwater.
- To show the variations among both system size and cost level, the examples chosen for each scenario modeled typically include a low cost small system, a mid cost medium system, and a high cost large system.
- Each of the examples is among the individual flow rate-specific estimates used to generate the cost equations presented in Appendix B (see Section [ REF \_Ref351114655 \r \h ] for details on the method used to develop the equation).

# Anion Exchange for Perchlorate, design 0.500 mgd, average 0.162 mgd, Low-Cost Components, Resin type: Perchlorate-selective 250,000 BV

WBS #	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Pressure Vessels - Carbon Steel - Plastic Internals	35,590	30
2.1	Ion Exchange Resin - Perchlorate-selective	40,858	N/A
3.4.1	Caustic Storage Tanks - Plastic/XLPE	844	7
3.4.2	Caustic Storage Tanks - Heat Tracing	1,454	7
3.4.3	Caustic Storage Tanks - Insulation	163	7
4.1	Cartridge Filters - Cartridge Filters	24,784	30
5.1.1	Backwash Piping - PVC	205	17
5.3.1	Process Piping - PVC	311	17
5.5.1	Inlet and Outlet Piping - PVC	311	17
5.7.1	Caustic Piping - PVC	60	17
5.8.1	Residuals Piping - PVC	198	17
5.8.2	Residuals Piping - Excavation	1,064	17
5.8.3	Residuals Piping - Bedding	61	17
5.8.4	Residuals Piping - Backfill and Compaction	493	17
5.8.5	Residuals Piping - Thrust Blocks	96	17
	Valves and Fittings - Motor/Air Operated (on/off) - Process -		
6.1.1	Polypropylene/PVC	9,580	20
	Valves and Fittings - Motor/Air Operated (on/off) - Backwash -		
6.1.2	Polypropylene/PVC	4,218	20
6.2.1	Valves and Fittings - Manual - Inlet and outlet - Polypropylene/PVC	1,282	20
6.2.2	Valves and Fittings - Manual - Process - Polypropylene/PVC	1,282	20
6.3.2	Valves and Fittings - Check Valves - Inlet and Outlet - Polypropylene/PVC	2,705	20
6.3.7	Valves and Fittings - Check Valves - Residuals - Polypropylene/PVC	243	20
8.2.1	Mixers for Caustic Storage Tanks - Mounted	1,656	22
11.1.1	Instrumentation and Controls - Flow Meters - Inlet and Outlet - Propeller	4,141	14
11.3.1	Instrumentation and Controls - Flow Meters - Backwash - Propeller	3,227	14
11.4.1	Instrumentation and Controls - Flow Meters - Residuals - Propeller	2,625	14
11.9	Instrumentation and Controls - High/Low Alarm (for caustic tanks)	593	14
11.12	Instrumentation and Controls - Head loss sensors	4,242	14
11.13.1	Instrumentation and Controls - Sampling Ports - Stainless Steel	250	30
12.1.1	System Controls - PLC Units - PLC racks/power supplies	340	8
12.1.2	System Controls - PLC Units - CPUs	1,256	8
12.1.3	System Controls - PLC Units - I/O discrete input modules	307	8
12.1.4	System Controls - PLC Units - I/O discrete output modules	375	8
12.1.5	System Controls - PLC Units - I/O combination analog modules	1,958	8
12.1.6	System Controls - PLC Units - Ethernet modules	1,730	8
12.1.9	System Controls - PLC Units - UPSs	563	8
12.2.1	System Controls - Operator Equipment - Drive controllers	1,072	14
12.2.2	System Controls - Operator Equipment - Operator interface units	3,911	8
13.1.1	Building Structures and HVAC - Building 1 - Small Low Cost Shed	11,961	20
13.3	Building Structures and HVAC - Concrete Pad	2,592	37
Indirect	Indirect and Add-On Costs (contingency from model)	96,284	20
	Process Cost	168,600	
	System Cost	264,884	
	O&M Cost	18,351	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Construction Management	4,102
Process Engineering	33,720
Site Work	3,332
Yard Piping	2,810
Geotechnical	0
Standby Power	0
Electrical (including yard wiring)	15,405
Mobilization and Demobilization	0
Architectural Fees for Treatment Building	0
Permits	24
Pilot Study	15,573
Land Cost	1,086
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Contingency (0.0%)	0
Miscellaneous Allowance (10.0%)	16,860
Legal, Fiscal, and Administrative (2.0%)	3,372
Sales Tax (0.0%)	0
Financing during Construction (0.0%)	0

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (8 hrs/yr @ \$45.2396/hr)	371
Clerical (8 hrs/yr @ \$30.4776/hr)	250
Operator (82 hrs/yr @ \$31.9149/hr)	2,616
Cartridge filter replacement (7 filters/yr @ \$191.6015/sf/yr)	1,380
Facility maintenance (materials and labor) (260 sf @ \$5.7866/sf/yr)	1,505
Sodium Hydroxide - Small Qty (5268 lbs/yr @ \$0.302/lb)	1,591
Perchlorate-selective (32 cf/yr @ \$256.5165/cf)	8,111
Energy for backwash/rinse pumps (0 Mwh/yr @ \$0.1212/kwh)	0
Energy for lighting (0 Mwh/yr @ \$0.1212/kwh)	3
Energy for ventilation (0 Mwh/yr @ \$0.1212/kwh)	7
POTW discharge fees (953 gal/yr @ \$0.3881/gal)	370
Spent resin disposal (1 ton/yr @ \$697.6744/ton)	474
Spent cartridge filter disposal (0 ton/yr @ \$74.9152/ton)	6
Miscellaneous Allowance (0 @ \$)	1,668

# Anion Exchange for Perchlorate, design 5.809 mgd, average 2.455 mgd, Mid-Cost Components, Resin type: Perchlorate-selective 250,000 BV

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Pressure Vessels - Carbon Steel - Plastic Internals	264,172	35
2.1	Ion Exchange Resin - Perchlorate-selective	487,390	N/A
3.1.1	Backwash/Rinse Tanks - Fiberglass	11,822	25
3.4.1	Caustic Storage Tanks - Fiberglass	5,578	10
3.4.2	Caustic Storage Tanks - Heat Tracing	1,573	10
3.4.3	Caustic Storage Tanks - Insulation	947	10
3.5.1	Caustic Day Tanks - Fiberglass	4,749	10
3.5.2	Caustic Day Tanks - Heat Tracing	1,455	10
3.5.3	Caustic Day Tanks - Insulation	182	10
4.1	Cartridge Filters - Cartridge Filters	166,640	35
5.1.1	Backwash Piping - CPVC	7,502	22
5.3.1	Process Piping - CPVC	9,549	22
5.5.1	Inlet and Outlet Piping - CPVC	19,446	22
5.7.1	Caustic Piping - CPVC	1,119	22
5.8.1	Residuals Piping - CPVC	2,590	22
5.8.2	Residuals Piping - Excavation	1,187	22
5.8.3	Residuals Piping - Bedding	70	22
5.8.4	Residuals Piping - Backfill and Compaction	550	22
5.8.5	Residuals Piping - Thrust Blocks	409	22
6.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Process - Cast Iron	53,747	25
6.1.2	Valves and Fittings - Motor/Air Operated (on/off) - Backwash - Cast Iron	44,639	25
6.1.6	Valves and Fittings - Motor/Air Operated (on/off) - Caustic - Stainless Steel	2,072	25
6.2.1	Valves and Fittings - Manual - Inlet and outlet - Cast Iron	5,215	25
6.2.2	Valves and Fittings - Manual - Process - Cast Iron	11,357	25
6.3.1	Valves and Fittings - Check Valves - Backwash - Cast Iron	3,857	25
6.3.2	Valves and Fittings - Check Valves - Inlet and Outlet - Cast Iron	10,467	25
6.3.7	Valves and Fittings - Check Valves - Residuals - Cast Iron	1,082	25
7.1	Pumps - Booster	69,732	20
7.2	Pumps - Backwash/Rinse	23,885	20
8.2.1	Mixers for Caustic Storage Tanks - Mounted	1,802	25
8.3.1	Mixers for Caustic Day Tanks - Mounted	1,658	25
11.1.1	Instrumentation and Controls - Flow Meters - Inlet and Outlet - Venturi	17,176	15
11.3.1	Instrumentation and Controls - Flow Meters - Backwash - Venturi	12,496	15
11.4.1	Instrumentation and Controls - Flow Meters - Residuals - Venturi	9,686	15
11.6	Instrumentation and Controls - High/Low Alarm (for backwash tanks)	593	15
11.9	Instrumentation and Controls - High/Low Alarm (for caustic tanks)	1,185	15
11.11	Instrumentation and Controls - Temperature meters	593	15
11.12	Instrumentation and Controls - Head loss sensors	23,332	15
11.13.1	Instrumentation and Controls - Sampling Ports - Carbon Steel	500	25
11.17	Instrumentation and Controls - Turbidity meters	5,223	15
12.1.1	System Controls - PLC Units - PLC racks/power supplies	680	10
12.1.2	System Controls - PLC Units - CPUs	1,256	10
12.1.3	System Controls - PLC Units - I/O discrete input modules	614	10
12.1.4	System Controls - PLC Units - I/O discrete output modules	375	10
12.1.5	System Controls - PLC Units - I/O combination analog modules	5,221	10
12.1.6	System Controls - PLC Units - Ethernet modules	1,730	10
12.1.7	System Controls - PLC Units - Base expansion modules	118	10
12.1.8	System Controls - PLC Units - Base expansion controller modules	86	10
12.1.9	System Controls - PLC Units - UPSs	563	10
12.2.1	System Controls - Operator Equipment - Drive controllers	6,435	15
12.2.2	System Controls - Operator Equipment - Operator interface units	920	10
12.2.3	System Controls - Operator Equipment - PC Workstations	1,006	10
12.2.4	System Controls - Operator Equipment - Printers - laser jet	627	10

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
12.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10
12.3.1	System Controls - Controls Software - Operator interface software	366	10
12.3.2	System Controls - Controls Software - PLC programming software	474	10
12.3.3	System Controls - Controls Software - PLC data collection software	690	10
12.3.4	System Controls - Controls Software - Plant intelligence software	11,480	10
13.1.1	Building Structures and HVAC - Building 1 - Medium Quality	390,703	40
	Building Structures and HVAC - Building 1 - Heating and Cooling System -		
13.1.3.1	Heat pump	5,957	25
13.3	Building Structures and HVAC - Concrete Pad	53,134	40
Indirect	Indirect and Add-On Costs (contingency from model)	1,073,196	40
	Process Cost	1,770,308	
	System Cost	2,843,504	
	O&M Cost	284,473	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Construction Management	115,738
Process Engineering	212,437
Site Work	60,232
Yard Piping	26,287
Geotechnical	0
Standby Power	21,078
Electrical (including yard wiring)	132,051
Mobilization and Demobilization	87,480
Architectural Fees for Treatment Building	35,984
Permits	24
Pilot Study	69,717
Land Cost	11,216
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Contingency (0.0%)	0
Miscellaneous Allowance (10.0%)	177,031
Legal, Fiscal, and Administrative (2.0%)	35,406
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	88,515

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (47 hrs/yr @ \$57.6375/hr)	2,733
Clerical (47 hrs/yr @ \$39.3563/hr)	1,866
Operator (474 hrs/yr @ \$35.9445/hr)	17,043
Materials for booster pumps (calculated as a percentage of capital)	697
Materials for backwash/rinse pumps (calculated as a percentage of capital)	239
Cartridge filter replacement (58 filters/yr @ \$197.9583/sf/yr)	11,402
Facility maintenance (materials and labor) (4700 sf @ \$5.9929/sf/yr)	28,166
Sodium Hydroxide - Small Qty (79132 lbs/yr @ \$0.302/lb)	23,895
Perchlorate-selective (479 cf/yr @ \$256.5165/cf)	122,919
Energy for booster pumps (216 Mwh/yr @ \$0.1212/kwh)	26,226
Energy for backwash/rinse pumps (0 Mwh/yr @ \$0.1212/kwh)	0
Energy for lighting (4 Mwh/yr @ \$0.1212/kwh)	540
Energy for ventilation (4 Mwh/yr @ \$0.1212/kwh)	444
Heat pump (cooling mode) (39 Mwh/yr @ \$0.1212/kwh)	4,732
Heat pump (62 Mwh/yr @ \$0.1212/kwh)	7,516
POTW discharge fees (16757 gal/yr @ \$0.1762/gal)	2,953
Spent resin disposal (10 ton/yr @ \$697.6744/ton)	7 <i>,</i> 188
Spent cartridge filter disposal (1 ton/yr @ \$74.9152/ton)	52
Miscellaneous Allowance (0 @ \$)	25,861

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# Anion Exchange for Perchlorate, design 56.271 mgd, average 28.136 mgd, High-Cost Components, Resin type: Perchlorate-selective 250,000 BV

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Pressure Vessels - Stainless Steel	6,247,920	35
2.1	Ion Exchange Resin - Perchlorate-selective	4,672,704	N/A
3.4.1	Caustic Storage Tanks - Stainless Steel	20,040	35
3.4.2	Caustic Storage Tanks - Heat Tracing	2,341	10
3.4.3	Caustic Storage Tanks - Insulation	4,613	10
3.5.1	Caustic Day Tanks - Stainless Steel	5,461	35
3.5.2	Caustic Day Tanks - Heat Tracing	1,544	10
3.5.3	Caustic Day Tanks - Insulation	794	10
4.1	Cartridge Filters - Cartridge Filters	1,280,742	35
5.1.1	Backwash Piping - Stainless Steel	68,948	45
5.3.1	Process Piping - Stainless Steel	65,361	45
5.5.1	Inlet and Outlet Piping - Stainless Steel	127,977	45
5.7.1	Caustic Piping - Stainless Steel	27,579	45
5.8.1	Residuals Piping - Stainless Steel	29,507	45
5.8.2	Residuals Piping - Excavation	1,334	45
5.8.3	Residuals Piping - Bedding	79	45
5.8.4	Residuals Piping - Beckfill and Compaction	618	45
5.8.5	Residuals Piping - Thrust Blocks	1,070	45
6.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Process - Stainless Steel	1,247,412	25
0.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Process - Stainless Steel  Valves and Fittings - Motor/Air Operated (on/off) - Backwash - Stainless	1,247,412	
6.1.2	Steel	405,104	25
6.1.6	Valves and Fittings - Motor/Air Operated (on/off) - Caustic - Stainless Steel	8,267	25
6.2.1	Valves and Fittings - Motor/All Operated (01/011) - Caustic - Stanliess Steel  Valves and Fittings - Manual - Inlet and outlet - Cast Iron	22,996	25
6.2.2	Valves and Fittings - Manual - Process - Stainless Steel	337,430	25
6.3.1	Valves and Fittings - Check Valves - Backwash - Stainless Steel	9,128	25
6.3.2	Valves and Fittings - Check Valves - Inlet and Outlet - Stainless Steel  Valves and Fittings - Check Valves - Inlet and Outlet - Stainless Steel	75,474	25
6.3.7			
7.1	Valves and Fittings - Check Valves - Residuals - Stainless Steel  Pumps - Booster	3,068 412,163	25 20
7.1	Pumps - Backwash/Rinse	42,669	20
8.2.1	Mixers for Caustic Storage Tanks - Mounted	3,221	25
8.3.1	Mixers for Caustic Storage Fails - Mounted  Mixers for Caustic Day Tanks - Mounted	1,765	25
11.1.1	Instrumentation and Controls - Flow Meters - Inlet and Outlet - Orifice Plate	12,366	15
11.3.1	Instrumentation and Controls - Flow Meters - Backwash - Magnetic	8,824	15
11.4.1	Instrumentation and Controls - Flow Meters - Residuals - Magnetic	6,832	15
11.4.1	Instrumentation and Controls - Flow Meters - Residuals - Magnetic  Instrumentation and Controls - High/Low Alarm (for caustic tanks)	1,185	15
11.11	Instrumentation and Controls - Temperature meters	593	15 15
11.11	Instrumentation and Controls - Temperature meters  Instrumentation and Controls - Head loss sensors		15
11.12	Instrumentation and Controls - need loss sensors  Instrumentation and Controls - Sampling Ports - Carbon Steel	95,449 1,300	25
11.13.1	Instrumentation and Controls - Sampling Ports - Carbon Steel	5,223	15
12.1.1	System Controls - PLC Units - PLC racks/power supplies	1,361	10
12.1.2	System Controls - PLC Units - CPUs	•	10
12.1.2	System Controls - PLC Units - CPUS  System Controls - PLC Units - I/O discrete input modules	1,256 1,227	10
12.1.4	System Controls - PLC Units - I/O discrete output modules	375	10
12.1.4	System Controls - PLC Units - I/O combination analog modules	14,359	10
12.1.6 12.1.7	System Controls - PLC Units - Ethernet modules System Controls - PLC Units - Base expansion modules	1,730 355	10 10
12.1.7	System Controls - PLC Units - Base expansion modules  System Controls - PLC Units - Base expansion controller modules	257	10
12.1.8	System Controls - PLC Units - Base expansion controller modules  System Controls - PLC Units - UPSs	563	10
12.1.9	System Controls - PEC Office - OPSS  System Controls - Operator Equipment - Drive controllers	9,652	15
12.2.1	System Controls - Operator Equipment - Drive controllers  System Controls - Operator Equipment - Operator interface units	9,632	10
	System Controls - Operator Equipment - Operator Interface units  System Controls - Operator Equipment - PC Workstations		10
12.2.3	System Controls - Operator Equipment - PC Workstations  System Controls - Operator Equipment - Printers - laser jet	1,006 627	10
12.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
12.3.1	System Controls - Controls Software - Operator interface software	366	10
12.3.2	System Controls - Controls Software - PLC programming software	474	10
12.3.3	System Controls - Controls Software - PLC data collection software	690	10
12.3.4	System Controls - Controls Software - Plant intelligence software	11,480	10
13.1.1	Building Structures and HVAC - Building 1 - High Quality	1,555,399	40
	Building Structures and HVAC - Building 1 - Heating System - Natural gas		
13.1.2.1	condensing furnace	64,314	25
13.1.3.1	Building Structures and HVAC - Building 1 - Cooling System - Air conditioner	25,745	25
13.2.1	Building Structures and HVAC - Building 2 - High Quality	412,924	40
	Building Structures and HVAC - Building 2 - Heating and Cooling System -		
13.2.3.1	Heat pump	34,649	25
13.3	Building Structures and HVAC - Concrete Pad	236,511	40
Indirect	Indirect and Add-On Costs (contingency from model)	9,150,973	40
	Process Cost	17,635,986	
	System Cost	26,786,959	
	O&M Cost	2,713,642	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Construction Management	749,880
Process Engineering	1,410,879
Site Work	262,459
Yard Piping	195,489
Geotechnical	32,795
Standby Power	147,274
Electrical (including yard wiring)	1,530,645
Mobilization and Demobilization	431,365
Architectural Fees for Treatment Building	144,432
Permits	1,731
Pilot Study	69,717
Land Cost	91,577
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Contingency (6.2%)	1,084,613
Miscellaneous Allowance (10.0%)	1,763,599
Legal, Fiscal, and Administrative (2.0%)	352,720
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	881,799

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Breakdown of O&M costs	Annual Cost (\$/year)
Manager (195 hrs/yr @ \$71.8488/hr)	14,034
Clerical (195 hrs/yr @ \$39.3563/hr)	7,688
Operator (1953 hrs/yr @ \$43.8427/hr)	85 <i>,</i> 639
Materials for booster pumps (calculated as a percentage of capital)	4,122
Materials for backwash/rinse pumps (calculated as a percentage of capital)	427
Cartridge filter replacement (960 filters/yr @ \$207.2477/sf/yr)	198,958
Facility maintenance (materials and labor) (20480 sf @ \$5.9929/sf/yr)	122,734
Sodium Hydroxide - Large Qty (906405 lbs/yr @ \$0.1254/lb)	113,672
Perchlorate-selective (5492 cf/yr @ \$256.5165/cf)	1,408,734
Energy for booster pumps (2480 Mwh/yr @ \$0.1212/kwh)	300,573
Energy for backwash/rinse pumps (0 Mwh/yr @ \$0.1212/kwh)	3
Energy for lighting (160 Mwh/yr @ \$0.1212/kwh)	19,397
Energy for ventilation (28 Mwh/yr @ \$0.1212/kwh)	3,376
Air conditioning (67 Mwh/yr @ \$0.1212/kwh)	8,167
Heat pump (cooling mode) (283 Mwh/yr @ \$0.1212/kwh)	34,319
Heat pump (82 Mwh/yr @ \$0.1212/kwh)	9,947
Natural gas condensing furnace (25251 therms/yr @ \$0.7941/therm)	20,051
POTW discharge fees (189467 gal/yr @ \$0.1682/gal)	31,868
Spent resin disposal (118 ton/yr @ \$697.6744/ton)	82,377
Spent cartridge filter disposal (12 ton/yr @ \$74.9152/ton)	863
Miscellaneous Allowance (0 @ \$)	246,695

## Biological Treatment, design 0.500 mgd, average 0.162 mgd, Low-Cost Components, Design Type: Fixed Bed Pressure Vessel

Item	Total Cost (\$)	Useful Life (yrs)
		30
		N/A
		37
	121,148	37
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	2,478	20
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	905	20
	2.262	20
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		20
	283	20
	424	20
	141	20
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		15
		15
Pumps and Blowers - Blowers - Air Scour - Positive Displacement	23,134	20
	Item Bioreactors - Pressure Vessels - Carbon Steel - Plastic Internals Bioreactors - Media - GAC Post-Treatment Filters - Filter Basins - Concrete Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash System) Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash System) Post-Treatment Filters - Filter Basins - Aluminum Railing Post-Treatment Filters - Filter Basins - Backfill and Compaction Post-Treatment Filters - Filter Basins - Backfill and Compaction Post-Treatment Filters - Media - Sand Post-Treatment Filters - Media - Sand Tanks - Electron Donor Storage Tanks - Plastic/XLPE Tanks - Electron Donor Day Tanks - Plastic/XLPE Tanks - Electron Donor Day Tanks - Plastic/XLPE Tanks - Residuals Holding Tanks/Basins - Plastic/XLPE Tanks - Residuals Holding Tanks/Basins - Plastic/XLPE Tanks - Aeration Tanks - Plastic/XLPE Tanks - Aeration Tanks - Plastic/XLPE Tanks - Polymer Storage Tanks - Plastic/XLPE Tanks - Polymer Storage Tanks - Plastic/XLPE Tanks - Coagulant Mix Tank - Plastic/XLPE Tanks - Coagulant Mix Tank - Plastic/XLPE Tanks Piping - Backwash Piping - PVC Piping - Process Piping - PVC Piping - Residuals Piping - PVC Piping - Residuals Piping - PVC Piping - Residuals Piping - PVC Piping - Polymer Addition Piping - PVC Valves and Fittings - Motor/Air Operated (on/off) - Process - Polypropylene/PVC Valves and Fittings - Motor/Air Operated (on/off) - Phosphoric Acid - Polypropylene/PVC Valves and Fittings - Motor/Air Operated (on/off) - Polymer-Polypropylene/PVC Valves and Fittings - Motor/Air Operated (on/off) - Polymer-Polypropylene/PVC Valves and Fittings - Motor/Air Operated (on/off) - Polymer-Polypropylene/PVC Valves and Fittings - Motor/Air Operated (on/off) - Polymer-Polypropylene/PVC Valves and Fittings - Check Valves - Backwash - Polypropylene/PVC Valves and Fittings - Check Valves - Backwash - Polypropylene/PVC Valves and Fittings - Check Valves	Bioreactors - Pressure Vessels - Carbon Steel - Plastic Internals Bioreactors - Media - GAC 30,950 Post-Treatment Filters - Filter Basins - Concrete 14,675 Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash System) 121,148 Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash System) Post-Treatment Filters - Filter Basins - Aluminum Railing 1,795 Post-Treatment Filters - Filter Basins - Aluminum Stairs 8,843 Post-Treatment Filters - Filter Basins - Excavation 8,987 Post-Treatment Filters - Filter Basins - Excavation 8,987 Post-Treatment Filters - Filter Basins - Backfill and Compaction 3,334 Post-Treatment Filters - Media - Anthracite 7,920 Post-Treatment Filters - Media - Anthracite 7,920 Post-Treatment Filters - Media - Sand 1,360 Post-Treatment Filters - Media - Sand 1,360 Post-Treatment Filters - Media - Sand 1,371 Tanks - Residuals Holding Tanks - Plastic/XLPE 1,171 Tanks - Electron Donor Storage Tanks - Plastic/XLPE 1,396 Tanks - Acration Tanks - Plastic/XLPE 1,296 Tanks - Acration Tanks - Plastic/XLPE 1,596 Tanks - Acration Tanks - Plastic/XLPE 1,045 Tanks - Coagulant Mix Tank - Plastic/XLPE Tanks - Coagulant Mix Tank - Plastic/XLPE 1,045 Tanks - Coagulant Mix Tank - Plastic/XLPE Tanks 1,571 Piping - Backwash Piping - PVC 1,134 Piping - Phocsphoric Acid Piping - PVC 1,134 Piping - Phocsphoric Acid Piping - PVC 1,134 Piping - Process Piping - PVC 1,134 Piping - Process Piping - PVC 1,297 Piping - Residuals Piping - PVC 2,198 Piping - Residuals Piping - PVC 2,198 Piping - Residuals Piping - PVC 2,198 Polypropylene/PVC 2,1987 Valves and Fittings - Motor/Air Operated (on/off) - Backwash - Polypropylene/PVC 2,287 Valves and Fittings - Motor/Air Operated (on/off) - Residuals - Polypropylene/PVC 2,287 Valves and Fittings - Motor/Air Operated (on/off) - Residuals - Polypropylene/PVC 2,287 Valves and Fittings - Motor/Air Operated (on/off) - Residuals - Polypropylene/PVC 2,287 Valves and Fittings - Motor/Air Operated (on/off) - Polymer-Polypropylene/PVC 2,288 Valves and Fitti

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WBS#	Item	Total Cost (\$)	Useful Life (yrs)
6.8.3	Pumps and Blowers - Blowers - Aeration - Positive Displacement	20,390	20
6.10.1	Pumps and Blowers - Polymer Metering - PVC - Electric	5,165	15
7.1.1	Mixers - Mixers for Electron Donor Storage Tanks - Mounted	1,720	22
7.2.1	Mixers - Mixers for Electron Donor Day Tanks - Mounted	1,669	22
7.5.1	Mixers - Mixers for Phosphoric Acid Storage Tanks - Mounted	1,650	22
7.11.1	Mixers - Mixers for Polymer Storage Tanks - Mounted	1,695	22
7.12.1	Mixers - Coagulant Mix Tank Mixers - Mounted	1,752	22
8.1	Solids Transfer - Eductors for Holding Tanks	1,690	40
	Solids Transfer - Dry Feeders for Filter Coagulant Addition - Volumetric		
8.2.1	Feeder	15,177	20
9.1.1	Instrumentation - Flow Meters - Inlet and Outlet - Propeller	4,141	14
9.3.1	Instrumentation - Flow Meters - Backwash - Propeller	4,544	14
9.4.1	Instrumentation - Flow Meters - Residuals - Propeller	1,786	14
9.7	Instrumentation - High/Low Alarm (for holding tanks)	593	14
9.8	Instrumentation - Temperature meters	2,373	14
9.9	Instrumentation - Head loss sensors	4,242	14
9.10.1	Instrumentation - Sampling Ports - Stainless Steel	350	30
9.12	Instrumentation - ORP sensor	5,265	14
9.14	Instrumentation - Turbidity meters	20,893	14
9.15	Instrumentation - Perchlorate/Nitrate Analyzer	24,574	14
9.17	Instrumentation - Dissolved Oxygen Analyzer	2,941	14
10.1.1	System Controls - PLC Unit(s) - PLC racks/power supplies	680	8
10.1.2	System Controls - PLC Unit(s) - CPUs	1,256	8
10.1.3	System Controls - PLC Unit(s) - I/O discrete input modules	614	8
10.1.4	System Controls - PLC Unit(s) - I/O discrete output modules	375	8
10.1.5	System Controls - PLC Unit(s) - I/O combination analog modules	5,874	8
10.1.6	System Controls - PLC Unit(s) - Ethernet modules	1,730	8
10.1.7	System Controls - PLC Unit(s) - Base expansion modules	118	8
10.1.8	System Controls - PLC Unit(s) - Base expansion controller modules	86	8
10.1.9	System Controls - PLC Unit(s) - UPSs	563	8
10.2.1	System Controls - Operator Equipment - Drive controllers	15,014	14
10.2.2	System Controls - Operator Equipment - Operator interface units	3,911	8
11.1.1	Building Structures and HVAC - Building 1 - Low Quality	86,721	37
11.3	Building Structures and HVAC - Concrete Pad	9,072	37
14.1	Solids drying pad	1,296	37
Indirect	Indirect and Add-On Costs (contingency from model)	425,303	37
	Process Cost	627,852	
	System Cost	1,053,155	
	O&M Cost	52,275	
	Totals are computed before component costs are rounded		

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Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	0
Architectural Fees for Treatment Building	0
Site Work	15,507
Yard Piping	3,375
Geotechnical	4,736
Standby Power	0
Electrical (including yard wiring)	53,206
Contingency	0
Process Engineering	125,570
Construction Management and GC Overhead	11,913
Permits	903
Pilot Study	132,669
Land Cost	2,082
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	62,785
Legal, Fiscal, and Administrative (2.0%)	12,557
Sales Tax (0.0%)	0
Financing during Construction (0.0%)	0

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (45 hrs/yr @ \$45.2396/hr)	2,049
Administrative (45 hrs/yr @ \$31.9149/hr)	1,446
Operator (321 hrs/yr @ \$30.4776/hr)	9,776
Materials for residuals pumps (calculated as a percentage of capital)	140
Materials for backwash air scour blowers (calculated as a percentage of capital)	231
Materials for aeration blowers (calculated as a percentage of capital)	204
Materials for filter basins (calculated as a percentage of capital)	1,404
Facility maintenance (materials and labor) (1100 sf @ \$5.7866/sf/yr)	6,365
Acetic Acid (24657 lbs/yr @ \$0.0978/lb)	2,412
Phosphoric Acid - Small Qty (2086 lbs/yr @ \$0.5492/lb)	1,145
Ferric Chloride - Small Qty (4931 lbs/yr @ \$0.9961/lb)	4,912
Polymers - Large Qty (2521 lbs/yr @ \$0.8113/lb)	2,045
GAC annual attrition replacement - Bioreactor (1701 lbs/yr @ \$1.9176/lb)	3,262
Sand annual attrition replacement- Filter (15 cuft/yr @ \$23.5561/cuft)	361
Anthracite annual attrition replacement- Filter (1555 lbs/yr @ \$0.5093/lb)	792
Consumables for online perchlorate analysis (NA)	10,117
Energy for backwash pumps (2 Mwh/yr @ \$0.1212/kwh)	230
Energy for residuals pumps (0 Mwh/yr @ \$0.1212/kwh)	15
Energy for blowers (1 Mwh/yr @ \$0.1212/kwh)	100
Energy for lighting (0 Mwh/yr @ \$0.1212/kwh)	60
Energy for ventilation (1 Mwh/yr @ \$0.1212/kwh)	74
Holding basins solids disposal (5 ton/yr @ \$74.9152/ton)	381
Miscellaneous Allowance (0 @ \$)	4,752

## Biological Treatment, design 5.809 mgd, average 2.455 mgd, Mid-Cost Components, Design Type: Fixed Bed Gravity Basin

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.2.1	Bioreactors - Gravity Basins - Concrete	106,224	40
1.2.2	Bioreactors - Gravity Basins - Internals (Underdrain/Backwash System)	492,141	40
1.2.3	Bioreactors - Gravity Basins - Aluminum Railing	8,526	40
1.2.4	Bioreactors - Gravity Basins - Aluminum Stairs	8,435	40
1.2.5	Bioreactors - Gravity Basins - Excavation	48,763	40
1.2.6	Bioreactors - Gravity Basins - Backfill and Compaction	11,850	40
1.3.1	Bioreactors - Media - GAC	406,254	N/A
2.1.1	Post-Treatment Filters - Filter Basins - Concrete	96,486	40
	Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash	<u>'</u>	
2.1.2	System)	413,581	40
2.1.3	Post-Treatment Filters - Filter Basins - Aluminum Railing	7,068	40
2.1.4	Post-Treatment Filters - Filter Basins - Aluminum Stairs	8,435	40
2.1.5	Post-Treatment Filters - Filter Basins - Excavation	41,825	40
2.1.6	Post-Treatment Filters - Filter Basins - Backfill and Compaction	10,826	40
2.3.1	Post-Treatment Filters - Media - Anthracite	110,417	10
2.3.2	Post-Treatment Filters - Media - Sand	50,308	20
3.1.1	Tanks - Backwash Tanks - Fiberglass	57,627	25
3.2.1	Tanks - Electron Donor Storage Tanks - Fiberglass	10,837	10
3.4.1	Tanks - Phosphoric Acid Storage Tanks - Fiberglass	4,705	10
3.5.1		5,874	
3.7.1	Tanks - Electron Donor Day Tanks - Fiberglass  Tanks - Phosphoric Acid Day Tanks - Fiberglass	· · ·	10
3.7.1	·	4,705	10
201	Tanks - Residuals Holding Tanks/Basins - Concrete Basins (includes	150,286	40
3.8.1	Excavation, Backfill, and Compaction)		40
3.11.1	Tanks - Aeration Tanks - Fiberglass	12,202	25
3.11.2	Tanks - Aeration Tanks - Diffusers	413	10
3.12.1	Tanks - Polymer Storage Tanks - Fiberglass	9,932	10
3.13.1	Tanks - Coagulant Mix Tank - Fiberglass Tanks	11,769	25
4.1.1	Piping - Backwash Piping - CPVC	32,516	22
4.2.1	Piping - Electron Donor Piping - CPVC	352	22
4.4.1	Piping - Phosphoric Acid Piping - CPVC	352	22
4.5.1	Piping - Process Piping - CPVC	17,733	22
4.7.1	Piping - Inlet and Outlet Piping - CPVC	36,114	22
4.8.1	Piping - Residuals Piping - CPVC	4,003	22
4.11.1	Piping - Polymer Addition Piping - CPVC	480	22
5.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Process - Cast Iron	231,488	25
5.1.2	Valves and Fittings - Motor/Air Operated (on/off) - Backwash - Cast Iron	114,008	25
	Valves and Fittings - Motor/Air Operated (on/off) - Electron Donor -		
5.1.3	Stainless Steel	4,163	25
	Valves and Fittings - Motor/Air Operated (on/off) - Phosphoric Acid -		
5.1.5	Stainless Steel	4,163	25
5.1.7	Valves and Fittings - Motor/Air Operated (on/off) - Residuals - Cast Iron	5,311	25
5.1.8	Valves and Fittings - Motor/Air Operated (on/off) - Air Scour- Cast Iron	6,822	25
5.1.12	Valves and Fittings - Motor/Air Operated (on/off) - Aeration- Cast Iron	7,440	25
	Valves and Fittings - Motor/Air Operated (on/off) - Polymer-		
5.1.13	Polypropylene/PVC	2,614	25
5.2.1	Valves and Fittings - Manual - Inlet and outlet - Cast Iron	5,215	25
5.3.1	Valves and Fittings - Check Valves - Backwash - Cast Iron	7,911	25
5.3.2	Valves and Fittings - Check Valves - Residuals - Cast Iron	2,164	25
5.3.3	Valves and Fittings - Check Valves - Inlet and outlet - Cast Iron	5,234	25
5.3.5	Valves and Fittings - Check Valves - Electron Donor - Stainless Steel	1,032	25
5.3.7	Valves and Fittings - Check Valves - Phosphoric Acid - Stainless Steel	1,547	25
5.3.8	Valves and Fittings - Check Valves - Polymer - Polypropylene/PVC	180	25
6.1	Pumps and Blowers - Booster Pump	69,732	20

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
6.3	Pumps and Blowers - Backwash Pump	83,511	20
6.4	Pumps and Blowers - Residuals Pump	20,314	20
6.5.1	Pumps and Blowers - Electron Donor Metering - Stainless Steel - Electric	5,993	20
6.7.1	Pumps and Blowers - Phosphoric Acid Metering - PVC - Electric	1,612	20
6.8.1	Pumps and Blowers - Blowers - Air Scour - Positive Displacement	58,928	25
6.8.3	Pumps and Blowers - Blowers - Aeration - Positive Displacement	34,152	25
6.10.1	Pumps and Blowers - Polymer Metering - Stainless Steel - Motor Driven	14,614	20
7.1.1	Mixers - Mixers for Electron Donor Storage Tanks - Mounted	2,681	25
7.2.1	Mixers - Mixers for Electron Donor Day Tanks - Mounted	1,853	25
7.5.1	Mixers - Mixers for Phosphoric Acid Storage Tanks - Mounted	1,706	25
7.6.1	Mixers - Mixers for Phosphoric Acid Day Tanks - Mounted	1,651	25
7.11.1	Mixers - Mixers for Polymer Storage Tanks - Mounted	2,534	25
7.11.1	Mixers - Coagulant Mix Tank Mixers - Mounted	2,781	25
8.1	Solids Transfer - Eductors for Holding Tanks	9,084	45
0.1	Solids Transfer - Eductors for Filter Coagulant Addition - Volumetric	9,004	43
8.2.1	Feeder	15,180	25
	Instrumentation - Flow Meters - Inlet and Outlet - Venturi	-	
9.1.1	Instrumentation - Flow Meters - Met and Oddiet - Venturi	17,176 15,837	15 15
9.3.1	Instrumentation - Flow Meters - Backwash - Venturi Instrumentation - Flow Meters - Residuals - Venturi	<i>'</i>	
		9,686	15
9.5	Instrumentation - Level Switch/Alarm (for vessels)	2,963	15
9.6	Instrumentation - High/Low Alarm (for backwash tanks)	593	15
9.7	Instrumentation - High/Low Alarm (for holding tanks)	593	15
9.8	Instrumentation - Temperature meters	5,933	15
9.9	Instrumentation - Head loss sensors	10,605	15
9.10.1	Instrumentation - Sampling Ports - Carbon Steel	700	25
9.12	Instrumentation - ORP sensor	5,265	15
9.14	Instrumentation - Turbidity meters	57,455	15
9.15	Instrumentation - Perchlorate/Nitrate Analyzer	24,574	15
9.17	Instrumentation - Dissolved Oxygen Analyzer	2,941	15
10.1.1	System Controls - PLC Unit(s) - PLC racks/power supplies	1,361	10
10.1.2	System Controls - PLC Unit(s) - CPUs	1,256	10
10.1.3	System Controls - PLC Unit(s) - I/O discrete input modules	921	10
10.1.4	System Controls - PLC Unit(s) - I/O discrete output modules	375	10
10.1.5	System Controls - PLC Unit(s) - I/O combination analog modules	11,748	10
10.1.6	System Controls - PLC Unit(s) - Ethernet modules	1,730	10
10.1.7	System Controls - PLC Unit(s) - Base expansion modules	355	10
10.1.8	System Controls - PLC Unit(s) - Base expansion controller modules	257	10
10.1.9	System Controls - PLC Unit(s) - UPSs	563	10
10.2.1	System Controls - Operator Equipment - Drive controllers	20,376	15
10.2.2	System Controls - Operator Equipment - Operator interface units	920	10
10.2.3	System Controls - Operator Equipment - PC Workstations	1,006	10
10.2.4	System Controls - Operator Equipment - Printers - laser jet	627	10
10.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10
10.3.1	System Controls - Controls Software - Operator interface software	366	10
10.3.2	System Controls - Controls Software - PLC programming software	474	10
10.3.3	System Controls - Controls Software - PLC data collection software	690	10
10.3.4	System Controls - Controls Software - Plant intelligence software	11,480	10
11.1.1	Building Structures and HVAC - Building 1 - Medium Quality	479,622	40
	Building Structures and HVAC - Building 1 - Heating and Cooling System -	, -	
11.1.3.1	Heat pump	3,812	25
11.2.1	Building Structures and HVAC - Building 2 - Medium Quality	748,150	40
	Building Structures and HVAC - Building 2 - Heating and Cooling System -	<u> </u>	<del>-</del>
11.2.3.1	Heat pump	13,213	25
11.3	Building Structures and HVAC - Concrete Pad	183,377	40
14.1	Solids drying pad	7,128	40
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WBS#	Item	Total Cost (\$)	Useful Life (yrs)
Indirect	Indirect and Add-On Costs (contingency from model)	2,835,055	40
	Process Cost	4,551,427	
	System Cost	7,386,482	
	O&M Cost	511,935	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	226,694
Architectural Fees for Treatment Building	88,547
Site Work	206,712
Yard Piping	51,461
Geotechnical	62,238
Standby Power	28,054
Electrical (including yard wiring)	312,325
Contingency	0
Process Engineering	546,171
Construction Management and GC Overhead	286,932
Permits	7,622
Pilot Study	212,082
Land Cost	32,474
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	455,143
Legal, Fiscal, and Administrative (2.0%)	91,029
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	227,571

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (199 hrs/yr @ \$57.6375/hr)	11,493
Administrative (199 hrs/yr @ \$35.9445/hr)	7,167
Operator (1365 hrs/yr @ \$39.3563/hr)	53,734
Materials for booster pumps (calculated as a percentage of capital)	697
Materials for backwash pumps (calculated as a percentage of capital)	835
Materials for residuals pumps (calculated as a percentage of capital)	203
Materials for backwash air scour blowers (calculated as a percentage of capital)	589
Materials for aeration blowers (calculated as a percentage of capital)	342
Materials forbioreactor basins (calculated as a percentage of capital)	6,153
Materials for filter basins (calculated as a percentage of capital)	4,709
Facility maintenance (materials and labor) (15560 sf @ \$5.9929/sf/yr)	93,249
Acetic Acid (373663 lbs/yr @ \$0.0978/lb)	36,551
Phosphoric Acid - Small Qty (31607 lbs/yr @ \$0.5492/lb)	17,359
Ferric Chloride - Small Qty (74733 lbs/yr @ \$0.9961/lb)	74,442
Polymers - Large Qty (48210 lbs/yr @ \$0.8113/lb)	39,113
GAC annual attrition replacement - Bioreactor (19440 lbs/yr @ \$1.8038/lb)	35,065
Sand annual attrition replacement- Filter (178 cuft/yr @ \$23.5561/cuft)	4,192
Anthracite annual attrition replacement- Filter (18067 lbs/yr @ \$0.5093/lb)	9,201
Consumables for online perchlorate analysis (NA)	10,117
Energy for booster pumps (216 Mwh/yr @ \$0.1212/kwh)	26,226
Energy for backwash pumps (14 Mwh/yr @ \$0.1212/kwh)	1 <i>,</i> 754
Energy for residuals pumps (9 Mwh/yr @ \$0.1212/kwh)	1,108
Energy for blowers (10 Mwh/yr @ \$0.1212/kwh)	1,188
Energy for lighting (62 Mwh/yr @ \$0.1212/kwh)	7,522
Energy for ventilation (6 Mwh/yr @ \$0.1212/kwh)	698
Heat pump (cooling mode) (54 Mwh/yr @ \$0.1212/kwh)	6,509
Heat pump (99 Mwh/yr @ \$0.1212/kwh)	12,003
Holding basins solids disposal (42 ton/yr @ \$74.9152/ton)	3,175
Miscellaneous Allowance (0 @ \$)	46,540

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## Biological Treatment, design 56.271 mgd, average 28.136 mgd, High-Cost Components, Design Type: Fluidized Bed Pressure Vessel

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1.1	Bioreactors - Pressure Vessels - Stainless Steel	16,981,071	35
1.3.1	Bioreactors - Media - GAC	2,776,373	N/A
2.1.1	Post-Treatment Filters - Filter Basins - Concrete	509,517	40
	Post-Treatment Filters - Filter Basins - Internals (Underdrain/Backwash		
2.1.2	System)	1,692,172	40
2.1.3	Post-Treatment Filters - Filter Basins - Aluminum Railing	25,802	40
2.1.4	Post-Treatment Filters - Filter Basins - Aluminum Stairs	8,435	40
2.1.5	Post-Treatment Filters - Filter Basins - Excavation	236,337	40
2.1.6	Post-Treatment Filters - Filter Basins - Backfill and Compaction	41,320	40
2.3.1	Post-Treatment Filters - Media - Anthracite	1,039,884	10
2.3.2	Post-Treatment Filters - Media - Sand	473,789	20
3.2.1	Tanks - Electron Donor Storage Tanks - Stainless Steel	115,623	35
3.4.1	Tanks - Phosphoric Acid Storage Tanks - Fiberglass	4,944	10
3.5.1	Tanks - Electron Donor Day Tanks - Stainless Steel	19,303	35
3.7.1	Tanks - Phosphoric Acid Day Tanks - Fiberglass	4,944	10
3.8.1	Tanks - Residuals Holding Tanks/Basins - Steel Tanks	139,334	35
3.11.1	Tanks - Aeration Tanks - Steel	87,651	35
3.11.2	Tanks - Aeration Tanks - Diffusers	3,636	10
3.12.1	Tanks - Polymer Storage Tanks - Stainless Steel	25,447	35
3.13.1	Tanks - Coagulant Mix Tank - Steel Tanks	84,632	35
4.1.1	Piping - Backwash Piping - Stainless Steel	141,470	45
4.2.1	Piping - Electron Donor Piping - Stainless Steel	3,657	45
4.4.1	Piping - Phosphoric Acid Piping - Stainless Steel	3,657	45
4.5.1	Piping - Process Piping - Stainless Steel	88,253	45
4.7.1	Piping - Inlet and Outlet Piping - Stainless Steel	204,762	45
4.8.1	Piping - Residuals Piping - Stainless Steel	9,102	45
4.9.1	Piping - Fluidized Bed Recycle Piping - Stainless Steel	70,544	45
4.11.1	Piping - Polymer Addition Piping - Stainless Steel	7,801	45
5.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Process - Stainless Steel	1,355,857	25
J.1.1	Valves and Fittings - Motor/Air Operated (on/off) - Fracess - Stainless Steel	1,555,657	
5.1.2	Steel	785,358	25
J.1.Z	Valves and Fittings - Motor/Air Operated (on/off) - Electron Donor -	763,336	
5.1.3	Stainless Steel	27,895	25
3.1.3	Valves and Fittings - Motor/Air Operated (on/off) - Phosphoric Acid -	27,033	
5.1.5	Stainless Steel	27,895	25
J.1.J	Valves and Fittings - Motor/Air Operated (on/off) - Residuals - Stainless	27,033	
5.1.7	Steel	2,012	25
3.1.7	Valves and Fittings - Motor/Air Operated (on/off) - Air Scour- Stainless	2,012	
5.1.8	Steel	16,535	25
3.1.0	Valves and Fittings - Motor/Air Operated (on/off) - Fluidized Bed Recycle -	10,333	
5.1.9	Stainless Steel	269,670	25
3.1.3	Valves and Fittings - Motor/Air Operated (on/off) - Air Biomass Removal-	203,070	
5.1.11	Stainless Steel	6,007	25
5.1.12	Valves and Fittings - Motor/Air Operated (on/off) - Aeration- Stainless Steel	70,204	25
5.1.13	Valves and Fittings - Motor/Air Operated (on/off) - Polymer- Stainless Steel	3,353	25
5.2.1	Valves and Fittings - Manual - Inlet and outlet - Cast Iron	22,996	25
5.2.2	Valves and Fittings - Manual - Process - Stainless Steel	459,469	25
5.3.1	Valves and Fittings - Check Valves - Backwash - Stainless Steel	26,164	25
5.3.2	Valves and Fittings - Check Valves - Residuals - Stainless Steel  Valves and Fittings - Check Valves - Residuals - Stainless Steel	868	25
5.3.3	Valves and Fittings - Check Valves - Inlet and outlet - Stainless Steel	37,737	25
5.3.5	Valves and Fittings - Check Valves - Electron Donor - Stainless Steel	1,032	25
5.3.7	Valves and Fittings - Check Valves - Phosphoric Acid - Stainless Steel  Valves and Fittings - Check Valves - Phosphoric Acid - Stainless Steel	1,547	25
5.3.8	Valves and Fittings - Check Valves - Prosprioric Acid - Starriess Steel  Valves and Fittings - Check Valves - Polymer - Stainless Steel	868	25
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WBS#	Item	Total Cost (\$)	Useful Life (yrs)
6.1	Pumps and Blowers - Booster Pump	412,163	20
6.2	Pumps and Blowers - Recycle Pump	302,373	20
6.3	Pumps and Blowers - Backwash Pump	158,883	20
6.4	Pumps and Blowers - Residuals Pump	30,540	20
	Pumps and Blowers - Electron Donor Metering - Stainless Steel - Motor	·	
6.5.1	Driven	11,689	20
6.7.1	Pumps and Blowers - Phosphoric Acid Metering - PVC - Motor Driven	5,938	20
6.8.1	Pumps and Blowers - Blowers - Air Scour - Positive Displacement	59,261	25
6.8.2	Pumps and Blowers - Blowers - Biomass Removal - Positive Displacement	15,178	25
6.8.3	Pumps and Blowers - Blowers - Aeration - Positive Displacement	143,920	25
6.10.1	Pumps and Blowers - Polymer Metering - Stainless Steel - Motor Driven	15,347	20
7.1.1	Mixers - Mixers for Electron Donor Storage Tanks - Impeller	24,687	25
7.2.1	Mixers - Mixers for Electron Donor Day Tanks - Mounted	3,156	25
7.5.1	Mixers - Mixers for Phosphoric Acid Storage Tanks - Mounted	2,309	25
7.6.1	Mixers - Mixers for Phosphoric Acid Day Tanks - Mounted	1,692	25
7.11.1	Mixers - Mixers for Polymer Storage Tanks - Mounted	3,668	25
7.12.1	Mixers - Coagulant Mix Tank Mixers - Impeller	18,263	25
8.1	Solids Transfer - Eductors for Holding Tanks	15,000	45
	Solids Transfer - Dry Feeders for Filter Coagulant Addition - Volumetric	,	
8.2.1	Feeder	15,205	25
9.1.1	Instrumentation - Flow Meters - Inlet and Outlet - Orifice Plate	12,366	15
9.3.1	Instrumentation - Flow Meters - Backwash - Magnetic	21,398	15
9.4.1	Instrumentation - Flow Meters - Residuals - Magnetic	3,868	15
9.7	Instrumentation - High/Low Alarm (for holding tanks)	593	15
9.8	Instrumentation - Temperature meters	73,569	15
9.9	Instrumentation - Head loss sensors	131,508	15
9.10.1	Instrumentation - Sampling Ports - Carbon Steel	3,950	25
9.12	Instrumentation - ORP sensor	5,265	15
9.14	Instrumentation - Turbidity meters	396,960	15
9.15	Instrumentation - Perchlorate/Nitrate Analyzer	24,574	15
9.17	Instrumentation - Dissolved Oxygen Analyzer	2,941	15
10.1.1	System Controls - PLC Unit(s) - PLC racks/power supplies	4,082	10
10.1.2	System Controls - PLC Unit(s) - CPUs	1,256	10
10.1.3	System Controls - PLC Unit(s) - I/O discrete input modules	2,455	10
10.1.4	System Controls - PLC Unit(s) - I/O discrete output modules	750	10
10.1.5	System Controls - PLC Unit(s) - I/O combination analog modules	49,604	10
10.1.6	System Controls - PLC Unit(s) - Ethernet modules	1,730	10
	System Controls - PLC Unit(s) - Base expansion modules	1,302	10
10.1.8	System Controls - PLC Unit(s) - Base expansion controller modules	942	10
10.1.9	System Controls - PLC Unit(s) - UPSs	563	10
10.2.1	System Controls - Operator Equipment - Drive controllers	30,028	15
10.2.2	System Controls - Operator Equipment - Operator interface units	920	10
10.2.3	System Controls - Operator Equipment - PC Workstations	1,006	10
10.2.4	System Controls - Operator Equipment - Printers - laser jet	627	10
10.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10
10.3.1	System Controls - Controls Software - Operator interface software	366	10
10.3.2	System Controls - Controls Software - PLC programming software	474	10
10.3.3	System Controls - Controls Software - PLC data collection software	690	10
10.3.4	System Controls - Controls Software - Plant intelligence software	11,480	10
11.1.1	Building Structures and HVAC - Building 1 - High Quality	3,785,520	40
	Building Structures and HVAC - Building 1 - Heating System - Natural gas	3,7 33,323	70
11.1.2.1	condensing furnace	186,719	25
11.1.3.1	Building Structures and HVAC - Building 1 - Cooling System - Air conditioner	127,120	25
11.2.1	Building Structures and HVAC - Building 2 - High Quality	927,051	40
			10

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
	Building Structures and HVAC - Building 2 - Heating System - Natural gas		
11.2.2.1	condensing furnace	97,694	25
11.2.3.1	Building Structures and HVAC - Building 2 - Cooling System - Air conditioner	280,408	25
11.3	Building Structures and HVAC - Concrete Pad	664,821	40
14.1	Solids drying pad	9,720	40
Indirect	Indirect and Add-On Costs (contingency from model)	19,972,073	40
	Process Cost	36,019,163	
	System Cost	55,991,236	
	O&M Cost	3,636,058	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	880,869
Architectural Fees for Treatment Building	321,675
Site Work	730,092
Yard Piping	255,323
Geotechnical	203,370
Standby Power	238,619
Electrical (including yard wiring)	2,994,983
Contingency	3,243,958
Process Engineering	2,881,533
Construction Management and GC Overhead	1,510,943
Permits	32,593
Pilot Study	347,913
Land Cost	206,944
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	3,601,916
Legal, Fiscal, and Administrative (2.0%)	720,383
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	1,800,958

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Breakdown of O&M costs	Annual Cost (\$/year)
Manager (540 hrs/yr @ \$71.8488/hr)	38,763
Administrative (540 hrs/yr @ \$43.8427/hr)	23,653
Operator (4578 hrs/yr @ \$39.3563/hr)	180,188
Materials for booster pumps (calculated as a percentage of capital)	4,122
Materials for backwash pumps (calculated as a percentage of capital)	1,589
Materials for recycle pumps (calculated as a percentage of capital)	3,024
Materials for residuals pumps (calculated as a percentage of capital)	305
Materials for backwash air scour blowers (calculated as a percentage of capital)	744
Materials for aeration blowers (calculated as a percentage of capital)	1,439
Materials for filter basins (calculated as a percentage of capital)	19,627
Facility maintenance (materials and labor) (56180 sf @ \$5.9929/sf/yr)	336,679
Acetic Acid (4282440 lbs/yr @ \$0.0978/lb)	418,894
Phosphoric Acid - Small Qty (362238 lbs/yr @ \$0.5492/lb)	198,943
Ferric Chloride - Small Qty (856488 lbs/yr @ \$0.9961/lb)	853,160
Polymers - Large Qty (123761 lbs/yr @ \$0.8113/lb)	100,407
GAC annual attrition replacement - Bioreactor (83034 lbs/yr @ \$1.6718/lb)	138,819
Sand annual attrition replacement- Filter (1724 cuft/yr @ \$23.5561/cuft)	40,610
Anthracite annual attrition replacement- Filter (175011 lbs/yr @ \$0.5093/lb)	89,133
Consumables for online perchlorate analysis (NA)	10,117
Energy for booster pumps (2480 Mwh/yr @ \$0.1212/kwh)	300,573
Energy for recycle pumps (1240 Mwh/yr @ \$0.1212/kwh)	150,286
Energy for backwash pumps (36 Mwh/yr @ \$0.1212/kwh)	4,343
Energy for blowers (150 Mwh/yr @ \$0.1212/kwh)	18,183
Energy for lighting (1212 Mwh/yr @ \$0.1212/kwh)	146,962
Energy for ventilation (117 Mwh/yr @ \$0.1212/kwh)	14,167
Air conditioning (685 Mwh/yr @ \$0.1212/kwh)	82,991
Natural gas condensing furnace (142420 therms/yr @ \$0.7941/therm)	113,090
Holding basins solids disposal (196 ton/yr @ \$74.9152/ton)	14,695
Miscellaneous Allowance (0 @ \$)	330,551

## Reverse Osmosis / Nanofiltration, design 0.500 mgd, average 0.162 mgd, Low-Cost Components, Feed Water: High Qual GW

WBS #	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Membrane Process - Membrane Elements	42,359	N/A
1.2	Membrane Process - RO Pressure Vessels	15,339	17
1.3.1	Membrane Process - Feed Line Connectors - Victaulic, Painted	1,341	20
1.5.1	Membrane Process - Piping On Rack - Feed - Stainless Steel	7,454	40
1.5.2	Membrane Process - Piping On Rack - Permeate - PVC	245	40
1.5.3	Membrane Process - Piping On Rack - Concentrate - Stainless Steel	5,573	40
1.6	Membrane Process - Vessel Support Rack - Steel Beams	13,663	20
1.7	Membrane Process - Markup for Rack Assembly	30,343	22
2.1.1	Pretreatment Acid Tanks - Plastic (HXLPE)	1,015	7
2.2.1	Pretreatment Antiscalant Tanks - Plastic (XLPE)	823	7
2.3.1	Cleaning Solution Makeup Tanks - Plastic (XLPE)	2,274	7
2.4.1	Cleaning Chemical Storage Tanks - Acid storage - Plastic (XLPE)	799	7
2.4.2	Cleaning Chemical Storage Tanks - High pH storage - Plastic (XLPE)	799	7
2.8.1	Acid Day Tanks - Plastic/XLPE	814	7
2.10.1	Mixers for Antiscalant Storage Tanks - Mounted	1,652	22
3.1.1	Inlet and Outlet Piping - PVC	622	17
3.2.1	Cleaning System Piping - PVC	204	17
3.3.1	Residuals Piping - PVC	612	17
3.3.2	Residuals Piping - Excavation	1,187	17
3.3.3	Residuals Piping - Bedding	70	17
3.3.4	Residuals Piping - Backfill and Compaction	550	17
3.3.5	Residuals Piping - Thrust Blocks	409	17
	Motor/Air Operated (on/off) Valves - Pretreatment acid -		······
4.1.1	Polypropylene/PVC	1,983	20
4.1.2	Motor/Air Operated (on/off) Valves - Antiscalant - Polypropylene/PVC	1,983	20
4.1.3	Motor/Air Operated (on/off) Valves - Feed line - Polypropylene/PVC	4,174	20
4.1.4	Motor/Air Operated (on/off) Valves - Concentrate control - Cast Iron	2,770	20
4.1.10	Motor/Air Operated (on/off) Valves - Cleaning - Polypropylene/PVC	25,042	20
4.2.1	Manual Valves - Inlet and outlet - Polypropylene/PVC	1,282	20
4.3.1	Check Valves - Residuals - Polypropylene/PVC	680	20
4.3.2	Check Valves - Inlet - Polypropylene/PVC	1,353	20
4.3.4	Check Valves - Feed pumps - Polypropylene/PVC	1,360	20
4.3.5	Check Valves - Cleaning - Polypropylene/PVC	2,040	20
5.1.1	Acid Metering Pumps for Pretreatment - PVC - Electric	1,582	15
5.2.1	Antiscalant Metering Pumps for Pretreatment - PVC - Electric	887	15
5.4	Pumps - Feed Water	34,730	17
5.7	Pumps - Cleaning Pumps (separate for acid and caustic)	2,901	17
	Screens and Filters - Cartridge Filters for Feed	32,089	30
6.2.1	Screens and Filters - Security Screens for Cleaning - Simplex Basket Screens	11,080	30
6.3	Screens and Filters - Cartridge Filters for Cleaning	16,481	30
8.1	Teflon Immersion Heaters for Cleaning Tanks	3,321	14
9.1.1	Instrumentation - Flow Meters - Inlet and Outlet - Propeller	8,283	14
9.2.1	Instrumentation - Flow Meters - Membrane Trains - Feed Line - Propeller	7,289	14
9.2.1	Instrumentation - Flow Meters - Membrane Trains - Permeate Line -	7,289	74
9.3.1	Propeller	7,289	14
3.3.1	Instrumentation - Flow Meters - Membrane Trains - Concentrate Line -	7,283	
9.3.1	Propeller	5,905	14
9.4.1	Instrumentation - Flow Meters - Cleaning - Propeller	12,424	14
9.4.1	Instrumentation - Propeller	3,645	14
9.6	Instrumentation - Properler Instrumentation - Level Switches/Alarms (for cleaning tanks)	1,185	14
9.6	Instrumentation - Level Switches/Alarms (for cleaning tanks) Instrumentation - High/Low Alarms (for pretreatment chemical tanks)	<del> </del>	
		1,185	14
9.8	Instrumentation - High/Low Alarms (for cleaning chemical storage tanks)	1,185	14
9.1	Instrumentation - pH meters	10,530	14

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
9.11	Instrumentation - Temperature meters	1,780	14
9.12	Instrumentation - Conductivity meters	12,927	14
9.13	Instrumentation - Head loss sensors	8,484	14
9.14.1	Instrumentation - Sampling ports - Carbon Steel	900	22
10.1.1	System Controls - PLC Units - PLC racks/power supplies	680	8
10.1.2	System Controls - PLC Units - CPUs	1,256	8
10.1.3	System Controls - PLC Units - I/O discrete input modules	614	8
10.1.4	System Controls - PLC Units - I/O discrete output modules	375	8
10.1.5	System Controls - PLC Units - I/O combination analog modules	5,874	8
10.1.6	System Controls - PLC Units - Ethernet modules	1,730	8
10.1.7	System Controls - PLC Units - Base expansion modules	118	8
10.1.8	System Controls - PLC Units - Base expansion controller modules	86	8
10.1.9	System Controls - PLC Units - UPSs	563	8
10.2.1	System Controls - Operator Equipment - Drive controllers	9,652	14
10.2.2	System Controls - Operator Equipment - Operator interface units	3,911	8
11.1.1	Building Structures and HVAC - Building 1 - Low Quality	118,890	37
11.4	Building Structures and HVAC - Concrete Pad	18,143	37
Indirect	Indirect and Add-On Costs (contingency from model)	305,971	37
	Process Cost	518,789	
	System Cost	824,760	
	O&M Cost	198,657	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	31,613
Construction Management and GC Overhead	41,579
Contingency	0
Process Engineering	103,758
Site Work	19,992
Yard Piping	3,418
Geotechnical	0
Standby Power	0
Electrical (including yard wiring)	38,176
Architectural Fees for Treatment Building	0
Pilot Study	1,330
Land Cost	2,444
Permits	1,407
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	51,879
Legal, Fiscal, and Administrative (2.0%)	10,376
Sales Tax (0.0%)	0
Financing during Construction (0.0%)	0

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (116 hrs/yr @ \$45.2396/hr)	5,269
Administrative (116 hrs/yr @ \$30.4776/hr)	3,550
Operator (1165 hrs/yr @ \$31.9149/hr)	37 <i>,</i> 173
Materials for pretreatment (calculated as a percentage of capital)	346
Cartridge filter replacement (19 filters/yr @ \$173.693/filter)	3,275
Materials for membrane process (calculated as a percentage of capital)	424
Membrane replacement (10 element/yr @ \$564.7907/element)	5,809
Materials for cleaning (calculated as a percentage of capital)	305
Materials for feed water and booster pumps (calculated as a percentage of capital)	347

Breakdown of O&M costs	Annual Cost (\$/year)
Facility maintenance (materials and labor) (1560 sf @ \$5.7866/sf/yr)	9,027
Sulfuric Acid - Small Qty (23565 lbs/yr @ \$0.3087/lb)	7,274
Antiscalant - Basic (2468 lbs/yr @ \$1.8447/lb)	4,552
Membrane Cleaner - Low pH Sulfate Control (13 gal/yr @ \$27.5474/gal)	349
Membrane Cleaner - High pH Detergent (13 gal/yr @ \$31.6028/gal)	400
Energy for feed water and booster pumps (119 Mwh/yr @ \$0.1212/kwh)	14,385
Energy for lighting (2 Mwh/yr @ \$0.1212/kwh)	220
Energy for ventilation (3 Mwh/yr @ \$0.1212/kwh)	363
POTW discharge fees (19757898 gal/yr @ \$0.0044/gal)	87,493
Spent cartridge filter disposal (0 ton/yr @ \$74.9152/ton)	23
Spent membrane element disposal (0 ton/yr @ \$74.9152/ton)	14
Miscellaneous Allowance (0 @ \$)	18,060

## Reverse Osmosis / Nanofiltration, design 5.809 mgd, average 2.455 mgd, Mid-Cost Components, Feed Water: High Qual GW

WBS #	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Membrane Process - Membrane Elements	679,514	N/A
1.2	Membrane Process - RO Pressure Vessels	224,291	22
1.3.1	Membrane Process - Feed Line Connectors - Victaulic, Galvanized	23,514	25
1.5.1	Membrane Process - Piping On Rack - Feed - Stainless Steel	121,348	45
1.5.2	Membrane Process - Piping On Rack - Permeate - PVC	5,132	45
1.5.3	Membrane Process - Piping On Rack - Concentrate - Stainless Steel	102,499	45
1.6	Membrane Process - Vessel Support Rack - Steel Beams	56,866	25
1.7	Membrane Process - Markup for Rack Assembly	299,601	29
2.1.1	Pretreatment Acid Tanks - Fiberglass	7,338	10
2.2.1	Pretreatment Antiscalant Tanks - Fiberglass	5,085	10
2.3.1	Cleaning Solution Makeup Tanks - Fiberglass	17,704	10
2.4.1	Cleaning Chemical Storage Tanks - Acid storage - Fiberglass	4,785	10
2.4.2	Cleaning Chemical Storage Tanks - High pH storage - Fiberglass	4,785	10
2.8.1	Acid Day Tanks - Fiberglass	4,891	10
2.9.1	Antiscalant Day Tanks - Fiberglass	4,710	10
2.10.1	Mixers for Antiscalant Storage Tanks - Mounted	1,717	25
2.11.1	Mixers for Antiscalant Day Tanks - Mounted	1,651	25
3.1.1	Inlet and Outlet Piping - CPVC	29,641	22
3.2.1	Cleaning System Piping - CPVC	3,430	22
3.3.1	Residuals Piping - CPVC	13,641	22
3.3.2	Residuals Piping - Excavation	1,508	22
3.3.3	Residuals Piping - Bedding	89	22
3.3.4	Residuals Piping - Backfill and Compaction	699	22
3.3.5	Residuals Piping - Thrust Blocks	2,153	22
4.1.1	Motor/Air Operated (on/off) Valves - Pretreatment acid - Stainless Steel	1,665	25
4.1.2	Motor/Air Operated (on/off) Valves - Pretreatment acid - Stainless Steel  Motor/Air Operated (on/off) Valves - Antiscalant - Stainless Steel	1,665	25
4.1.3	Motor/Air Operated (on/off) Valves - Feed line - Cast Iron	36,483	25
4.1.4	Motor/Air Operated (on/off) Valves - Concentrate control - Stainless Steel	i	25
4.1.4		15,983	25
4.2.1	Motor/Air Operated (on/off) Valves - Cleaning - Stainless Steel  Manual Valves - Inlet and outlet - Cast Iron	202,457 6,852	25
4.2.1	Check Valves - Residuals - Cast Iron		25
		2,870	
4.3.2	Check Valves - Inlet - Cast Iron	6,754	25
	Check Valves - Feed pumps - Cast Iron	15,822	25
4.3.5	Check Valves - Cleaning - Cast Iron	5,786	25
5.1.1	Acid Metering Pumps for Pretreatment - PVC - Electric	2,963	20
5.2.1	Antiscalant Metering Pumps for Pretreatment - Stainless Steel - Electric	3,304	20
5.4	Pumps - Feed Water	213,062	20
	Pumps - Cleaning Pumps (separate for acid and caustic)	8,294	20
6.1	Screens and Filters - Cartridge Filters for Feed	203,680	35
	Screens and Filters - Security Screens for Cleaning - Simplex Basket	50 740	
6.2.1	Screens	53,718	35
6.3	Screens and Filters - Cartridge Filters for Cleaning	123,757	35
8.1	Teflon Immersion Heaters for Cleaning Tanks	23,304	15
9.1.1	Instrumentation - Flow Meters - Inlet and Outlet - Venturi	37,666	15
9.2.1	Instrumentation - Flow Meters - Membrane Trains - Feed Line - Venturi	47,512	15
	Instrumentation - Flow Meters - Membrane Trains - Permeate Line -		
9.3.1	Venturi	43,235	15
	Instrumentation - Flow Meters - Membrane Trains - Concentrate Line -		
9.3.1	Venturi	37,488	15
9.4.1	Instrumentation - Flow Meters - Cleaning - Venturi	56,499	15
9.5.1	Instrumentation - Venturi	14,412	15
9.6	Instrumentation - Level Switches/Alarms (for cleaning tanks)	1,185	15
9.7	Instrumentation - High/Low Alarms (for pretreatment chemical tanks)	1,185	15

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
9.8	Instrumentation - High/Low Alarms (for cleaning chemical storage tanks)	1,185	15
9.1	Instrumentation - pH meters	10,530	15
9.11	Instrumentation - Temperature meters	1,780	15
9.12	Instrumentation - Conductivity meters	17,775	15
9.13	Instrumentation - Head loss sensors	14,848	15
9.14.1	Instrumentation - Sampling ports - Stainless Steel	8,700	35
10.1.1	System Controls - PLC Units - PLC racks/power supplies	1,020	10
10.1.2	System Controls - PLC Units - CPUs	1,256	10
10.1.3	System Controls - PLC Units - I/O discrete input modules	2,455	10
10.1.4	System Controls - PLC Units - I/O discrete output modules	375	10
10.1.5	System Controls - PLC Units - I/O combination analog modules	7,832	10
10.1.6	System Controls - PLC Units - Ethernet modules	1,730	10
10.1.7	System Controls - PLC Units - Base expansion modules	237	10
10.1.8	System Controls - PLC Units - Base expansion controller modules	171	10
10.1.9	System Controls - PLC Units - UPSs	563	10
10.2.1	System Controls - Operator Equipment - Drive controllers	16,087	15
10.2.2	System Controls - Operator Equipment - Operator interface units	920	10
10.2.3	System Controls - Operator Equipment - PC Workstations	2,013	10
10.2.4	System Controls - Operator Equipment - Printers - laser jet	627	10
10.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10
10.3.1	System Controls - Controls Software - Operator interface software	366	10
10.3.2	System Controls - Controls Software - PLC programming software	947	10
10.3.3	System Controls - Controls Software - PLC data collection software	1,380	10
10.3.4	System Controls - Controls Software - Plant intelligence software	22,960	10
11.1.1	Building Structures and HVAC - Building 1 - Medium Quality	419,822	40
	Building Structures and HVAC - Building 1 - Heating System - Natural gas		
11.1.2.1	condensing furnace	61,640	25
11.4	Building Structures and HVAC - Concrete Pad	73,869	40
Indirect	Indirect and Add-On Costs (contingency from model)	1,857,320	40
	Process Cost	3,455,924	
	System Cost	5,313,244	
	O&M Cost	2,017,479	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	150,269
Construction Management and GC Overhead	192,949
Contingency	0
Process Engineering	360,983
Site Work	65,230
Yard Piping	13,803
Geotechnical	0
Standby Power	115,996
Electrical (including yard wiring)	252,677
Architectural Fees for Treatment Building	38,513
Pilot Study	64,004
Land Cost	11,857
Permits	3,530
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	345,592
Legal, Fiscal, and Administrative (2.0%)	69,118
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	172,796

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Breakdown of O&M costs	Annual Cost (\$)
Manager (634 hrs/yr @ \$57.6375/hr)	36,534
Administrative (634 hrs/yr @ \$39.3563/hr)	24,946
Operator (6338 hrs/yr @ \$35.9445/hr)	227,834
Materials for pretreatment (calculated as a percentage of capital)	2,083
Cartridge filter replacement (130 filters/yr @ \$203.4792/filter)	26,510
Materials for membrane process (calculated as a percentage of capital)	6,795
Membrane replacement (165 element/yr @ \$529.4913/element)	87,366
Materials for cleaning (calculated as a percentage of capital)	1,858
Materials for feed water and booster pumps (calculated as a percentage of capital)	2,131
Facility maintenance (materials and labor) (5090 sf @ \$5.9193/sf/yr)	30,129
Sulfuric Acid - Small Qty (253899 lbs/yr @ \$0.3087/lb)	78,367
Antiscalant - Basic (28052 lbs/yr @ \$1.8447/lb)	51,749
Membrane Cleaner - Low pH Sulfate Control (203 gal/yr @ \$27.5474/gal)	5,592
Membrane Cleaner - High pH Detergent (203 gal/yr @ \$31.6028/gal)	6,415
Energy for feed water and booster pumps (1872 Mwh/yr @ \$0.1212/kwh)	226,892
Energy for lighting (32 Mwh/yr @ \$0.1212/kwh)	3,911
Energy for ventilation (13 Mwh/yr @ \$0.1212/kwh)	1,547
Natural gas condensing furnace (23911 therms/yr @ \$0.7941/therm)	18,987
POTW discharge fees (224932839 gal/yr @ \$0.0044/gal)	994,106
Spent cartridge filter disposal (2 ton/yr @ \$74.9152/ton)	154
Spent membrane element disposal (2 ton/yr @ \$74.9152/ton)	165
Miscellaneous Allowance (0 @ \$)	183,407

# Reverse Osmosis / Nanofiltration, design 56.271 mgd, average 28.136 mgd, High-Cost Components, Feed Water: High Qual GW

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1	Membrane Process - Membrane Elements	4,612,457	N/A
1.2	Membrane Process - RO Pressure Vessels	1,522,459	22
1.3.1	Membrane Process - Feed Line Connectors - Victaulic, Galvanized	159,611	25
1.5.1	Membrane Process - Piping On Rack - Feed - Stainless Steel	823,697	45
1.5.2	Membrane Process - Piping On Rack - Permeate - PVC	34,835	45
1.5.3	Membrane Process - Piping On Rack - Concentrate - Stainless Steel	695,752	45
1.6	Membrane Process - Vessel Support Rack - Steel Beams	379,103	25
1.7	Membrane Process - Markup for Rack Assembly	2,026,759	29
2.1.1	Pretreatment Acid Tanks - Fiberglass	28,524	10
2.2.1	Pretreatment Antiscalant Tanks - Stainless Steel	11,510	35
2.3.1	Cleaning Solution Makeup Tanks - Stainless Steel	22,264	35
2.4.1	Cleaning Chemical Storage Tanks - Acid storage - Fiberglass	4,878	10
2.4.2	Cleaning Chemical Storage Tanks - High pH storage - Stainless Steel	4,710	35
2.8.1	Acid Day Tanks - Fiberglass	6,716	10
2.9.1	Antiscalant Day Tanks - Stainless Steel	4,880	35
2.10.1	Mixers for Antiscalant Storage Tanks - Mounted	2,412	25
2.11.1	Mixers for Antiscalant Day Tanks - Mounted	1,700	25
3.1.1	Inlet and Outlet Piping - Stainless Steel	136,508	45
3.2.1	Cleaning System Piping - Stainless Steel	24,848	45
3.3.1	Residuals Piping - Steel	74,713	35
3.3.2	Residuals Piping - Excavation	2,871	22
3.3.3	Residuals Piping - Bedding	144	22
3.3.4	Residuals Piping - Backfill and Compaction	1,331	22
3.3.5	Residuals Piping - Thrust Blocks	15,674	22
4.1.1	Motor/Air Operated (on/off) Valves - Pretreatment acid - Stainless Steel	1,665	25
4.1.2	Motor/Air Operated (on/off) Valves - Antiscalant - Stainless Steel	1,665	25
4.1.3	Motor/Air Operated (on/off) Valves - Feed line - Stainless Steel	685,694	25
4.1.4	Motor/Air Operated (on/off) Valves - Concentrate control - Stainless Steel	106,556	25
4.1.10	Motor/Air Operated (on/off) Valves - Cleaning - Stainless Steel	745,892	25
4.2.1	Manual Valves - Inlet and outlet - Cast Iron	22,996	25
4.3.1	Check Valves - Residuals - Stainless Steel	15,845	25
4.3.2	Check Valves - Inlet - Stainless Steel	37,737	25
4.3.4	Check Valves - Feed pumps - Stainless Steel	195,726	25
4.3.5	Check Valves - Cleaning - Stainless Steel	9,203	25
5.1.1	Acid Metering Pumps for Pretreatment - PVC - Motor Driven	8,099	20
5.2.1	Antiscalant Metering Pumps for Pretreatment - PVC - Motor Driven	6,078	20
5.4	Pumps - Feed Water	2,900,895	20
5.7	Pumps - Cleaning Pumps (separate for acid and caustic)	8,352	20
6.1	Screens and Filters - Cartridge Filters for Feed	1,569,824	35
0.1	Screens and Filters - Security Screens for Cleaning - Simplex Basket	2,303,024	
6.2.1	Screens	54,857	35
6.3	Screens and Filters - Cartridge Filters for Cleaning	126,670	35
8.1	Teflon Immersion Heaters for Cleaning Tanks	23,855	15
9.1.1	Instrumentation - Flow Meters - Inlet and Outlet - Orifice Plate	24,731	15
9.2.1	Instrumentation - Flow Meters - Membrane Trains - Feed Line - Magnetic	225,534	15
5.2.12	Instrumentation - Flow Meters - Membrane Trains - Permeate Line -	223,33	
9.3.1	Magnetic	225,534	15
3.3.2	Instrumentation - Flow Meters - Membrane Trains - Concentrate Line -	220,001	
9.3.1	Magnetic Magnetic	136,649	15
9.4.1	Instrumentation - Flow Meters - Cleaning - Orifice Plate	13,941	15
9.5.1	Instrumentation - Magnetic	25,692	15
9.6	Instrumentation - Level Switches/Alarms (for cleaning tanks)	1,185	15
9.7	Instrumentation - High/Low Alarms (for pretreatment chemical tanks)	1,185	15

WBS#	ltem	Total Cost (\$)	Useful Life (yrs)
9.8	Instrumentation - High/Low Alarms (for cleaning chemical storage tanks)	1,185	15
9.1	Instrumentation - pH meters	10,530	15
9.11	Instrumentation - Temperature meters	1,780	15
9.12	Instrumentation - Conductivity meters	100,186	15
9.13	Instrumentation - Head loss sensors	99,691	15
9.14.1	Instrumentation - Sampling ports - Stainless Steel	58,450	35
10.1.1	System Controls - PLC Units - PLC racks/power supplies	4,422	10
10.1.2	System Controls - PLC Units - CPUs	1,256	10
10.1.3	System Controls - PLC Units - I/O discrete input modules	13,195	10
10.1.4	System Controls - PLC Units - I/O discrete output modules	750	10
10.1.5	System Controls - PLC Units - I/O combination analog modules	35,897	10
10.1.6	System Controls - PLC Units - Ethernet modules	1,730	10
10.1.7	System Controls - PLC Units - Base expansion modules	1,420	10
10.1.8	System Controls - PLC Units - Base expansion controller modules	1,028	10
10.1.9	System Controls - PLC Units - UPSs	563	10
10.2.1	System Controls - Operator Equipment - Drive controllers	45,042	15
10.2.2	System Controls - Operator Equipment - Operator interface units	920	10
10.2.3	System Controls - Operator Equipment - PC Workstations	4,026	10
10.2.4	System Controls - Operator Equipment - Printers - laser jet	627	10
10.2.5	System Controls - Operator Equipment - Printers - dot matrix	642	10
10.3.1	System Controls - Controls Software - Operator interface software	366	10
10.3.2	System Controls - Controls Software - PLC programming software	1,895	10
10.3.3	System Controls - Controls Software - PLC data collection software	2,761	10
10.3.4	System Controls - Controls Software - Plant intelligence software	45,920	10
10.3.5	System Controls - Controls Software - Early warning software	13,395	10
11.1.1	Building Structures and HVAC - Building 1 - High Quality	640,174	40
	Building Structures and HVAC - Building 1 - Heating System - Natural gas		
11.1.2.1	condensing furnace	170,293	25
	Building Structures and HVAC - Building 1 - Heating and Cooling System -		
11.1.3.1	Air conditioner	1,104,140	25
11.2.1	Building Structures and HVAC - Building 2 - High Quality	362,372	40
	Building Structures and HVAC - Building 2 - Heating and Cooling System -		
11.2.3.1	Heat pump	4,053	25
11.3.1	Building Structures and HVAC - Building 3 - High Quality	1,286,225	40
	Building Structures and HVAC - Building 3 - Heating System - Natural gas		
11.3.2.1	condensing furnace	27,051	25
	Building Structures and HVAC - Building 3 - Cooling System - Air		
11.3.3.1	conditioner	7,126	25
11.4	Building Structures and HVAC - Concrete Pad	384,249	40
Indirect	Indirect and Add-On Costs (contingency from model)	10,347,182	40
	Process Cost	22,207,784	
	System Cost	32,554,965	
	O&M Cost	19,273,680	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Mobilization and Demobilization	508,239
Construction Management and GC Overhead	858,411
Contingency	0
Process Engineering	1,620,601
Site Work	291,037
Yard Piping	88,455
Geotechnical	0
Standby Power	1,039,493
Electrical (including yard wiring)	1,709,688
Architectural Fees for Treatment Building	195,959
Pilot Study	132,875
Land Cost	110,762
Permits	16,337
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	2,220,778
Legal, Fiscal, and Administrative (2.0%)	444,156
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	1,110,389

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (2379 hrs/yr @ \$71.8488/hr)	170,939
Administrative (2379 hrs/yr @ \$39.3563/hr)	93,634
Operator (23791 hrs/yr @ \$43.8427/hr)	1,043,082
Materials for pretreatment (calculated as a percentage of capital)	15 <i>,</i> 790
Cartridge filter replacement (885 filters/yr @ \$207.2116/filter)	183,293
Materials for membrane process (calculated as a percentage of capital)	46,125
Membrane replacement (1120 element/yr @ \$529.4913/element)	593,030
Materials for cleaning (calculated as a percentage of capital)	1,899
Materials for feed water and booster pumps (calculated as a percentage of capital)	29,009
Facility maintenance (materials and labor) (22710 sf @ \$5.9929/sf/yr)	136,098
Sulfuric Acid - Large Qty (2902863 lbs/yr @ \$0.1107/lb)	321 <i>,</i> 406
Antiscalant - Basic (320728 lbs/yr @ \$1.8447/lb)	591 <i>,</i> 654
Membrane Cleaner - Low pH Sulfate Control (1378 gal/yr @ \$27.5474/gal)	37 <i>,</i> 958
Membrane Cleaner - High pH Detergent (1378 gal/yr @ \$31.6028/gal)	43,546
Energy for feed water and booster pumps (21420 Mwh/yr @ \$0.1212/kwh)	2,596,501
Energy for lighting (199 Mwh/yr @ \$0.1212/kwh)	24,115
Energy for ventilation (90 Mwh/yr @ \$0.1212/kwh)	10,902
Air conditioning (1878 Mwh/yr @ \$0.1212/kwh)	227,680
Heat pump (15 Mwh/yr @ \$0.1212/kwh)	1,760
Heat pump (43 Mwh/yr @ \$0.1212/kwh)	5,236
Natural gas condensing furnace (80713 therms/yr @ \$0.7941/therm)	64,091
POTW discharge fees (2553033230 gal/yr @ \$0.0044/gal)	11,281,363
Spent cartridge filter disposal (11 ton/yr @ \$74.9152/ton)	832
Spent membrane element disposal (21 ton/yr @ \$74.9152/ton)	1,585
Miscellaneous Allowance (0 @ \$)	1,752,153

## Point of Use/Point of Entry, design 0.500 mgd, average 0.162 mgd, Contaminant: Perchlorate, Treatment Technology: POU Reverse Osmosis

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
23.1.1	Installed Treatment Equipment - POU/POE Unit Purchase	116,781	10
23.1.2	Installed Treatment Equipment - POU/POE Installation	41,270	10
23.1.3	Installed Treatment Equipment - Scheduling Time	7,330	N/A
23.2.1.1	Public Education - Technical Labor - Develop materials	319	N/A
23.2.1.3	Public Education - Technical Labor - Meetings	64	N/A
23.2.1.4	Public Education - Technical Labor - Post-meeting	64	N/A
23.2.2.1	Public Education - Clerical Labor - Develop materials	183	N/A
23.2.2.3	Public Education - Clerical Labor - Meetings	61	N/A
23.2.2.4	Public Education - Clerical Labor - Post-meeting	61	N/A
23.2.3.1	Public Education - Printed Material - Meeting flyers	8	N/A
23.2.3.2	Public Education - Printed Material - Meeting ads	60	N/A
23.2.3.4	Public Education - Printed Material - Meeting handouts	173	N/A
23.2.3.5	Public Education - Printed Material - Billing mailers	115	N/A
23.3.1	Initial Year Monitoring 1 - Sampling time	2,561	N/A
23.3.3	Initial Year Monitoring 1 - Analysis	35,074	N/A
23.3.4	Initial Year Monitoring 1 - Analysis (total coliform)	8,780	N/A
23.3.5	Initial Year Monitoring 1 - Shipping	598	N/A
Indirect	Indirect and Add-On Costs (contingency from model)	56,229	10
	Process Cost	213,501	
***************************************	System Cost	269,730	
***************************************	O&M Cost	79,513	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Permitting	4,961
Pilot Testing	4,961
Legal	4,961
Engineering	24,807
Contingency	16,538

Breakdown of O&M costs	Annual Cost (\$/year)
POU/POE Maintenance	9,978
Information updates	383
Maintenance Scheduling	9,529
Information updates	366
Sediment Pre-Filter	4,541
Pre-GAC Filter Cartridge	11,742
Post-GAC Filter Cartridge	6,785
RO Membrane	12,577
Billing mailers	173
Sampling time	1,277
Analysis	17,483
Shipping	304

## Non-Treatment, design 0.500 mgd, average 0.162 mgd, Low-Cost Components, Design Type: Interconnection

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
3.1.1	Piping - Interconnect - PVC	59,800	17
3.1.2	Piping - Interconnect - Excavation	91,114	17
3.1.3	Piping - Interconnect - Backfill and Compaction	42,241	17
3.1.4	Piping - Interconnect - Asphalt Patch	12,713	17
4.1.1	Valves - Isolation (PRV) and Street - Ductile Iron	8,217	20
6.1.1	Instrumentation - Flow Meters - Interconnect - Propeller	4,141	14
Indirect	Indirect and Add-On Costs (contingency from model)	106,490	17
	Process Cost	218,225	
	System Cost	324,716	
	O&M Cost	130,125	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	0
Process Engineering	43,645
Construction Management and GC Overhead	24,656
Site Work	0
Yard Piping	0
Geotechnical	0
Standby Power	0
Electrical (including yard wiring)	0
Mobilization and Demobilization	12,002
Architectural Fees for Treatment Building	0
Permits	0
Pilot Study	0
Land Cost	0
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	21,823
Legal, Fiscal, and Administrative (2.0%)	4,365
Sales Tax (0.0%)	0
Financing during Construction (0.0%)	0

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (0 hrs/yr @ \$45.2396/hr)	4
Administrative (0 hrs/yr @ \$31.9149/hr)	3
Operator (1 hrs/yr @ \$30.4776/hr)	28
Purchased Water (59130 K gal @ \$2K gal)	118,260
Miscellaneous Allowance (0 @ \$)	11,830

## Non-Treatment, design 0.500 mgd, average 0.162 mgd, Low-Cost Components, Design Type: New Well Construction

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1.1	Well Items - Well Casing - PVC	9,383	17
1.2.1	Well Items - Well screen - PVC Schedule 40	8,354	17
1.3.1	Well Items - Plugs - PVC	172	17
1.4	Well Items - Well Drilling	22,988	N/A
1.5	Well Items - Gravel Pack	14,275	N/A
1.6	Well Items - Well Development	875	N/A
1.7	Well Items - Surface seal well, concrete fill	5,125	N/A
3.2.1	Piping - Well - PVC	615	17
3.2.2	Piping - Well - Excavation	4,556	17
3.2.3	Piping - Well - Backfill and Compaction	2,112	17
4.2.2	Valves - Motor/Air Operated (on/off) - Well Pump - Polypropylene/PVC	1,282	20
5.2	Pumps - Well Pump	14,343	17
6.2.1	Instrumentation - Flow Meters - Well - Propeller	4,141	14
8.1.1	Building Structures - Building 1 - Small Low Cost Shed	9,200	20
Indirect	Indirect and Add-On Costs (contingency from model)	110,111	17
	Process Cost	97,419	
	System Cost	207,530	
	O&M Cost	38,061	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	0
Process Engineering	19,484
Construction Management and GC Overhead	12,509
Site Work	2,563
Yard Piping	0
Geotechnical	27,008
Standby Power	0
Electrical (including yard wiring)	8,822
Mobilization and Demobilization	7,278
Architectural Fees for Treatment Building	0
Permits	0
Pilot Study	0
Land Cost	20,757
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	9,742
Legal, Fiscal, and Administrative (2.0%)	1,948
Sales Tax (0.0%)	0
Financing during Construction (0.0%)	0

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (4 hrs/yr @ \$45.2396/hr)	193
Administrative (4 hrs/yr @ \$31.9149/hr)	136
Operator (43 hrs/yr @ \$30.4776/hr)	1,298
Facility maintenance (materials and labor) (200 sf @ \$5.7866/sf/yr)	1,157
Well pump (calculated as a percentage of capital)	143
Energy for well pumps (261 Mwh/yr @ \$0.1212/kwh)	31,674
Miscellaneous Allowance (0 @ \$)	3,460

## Non-Treatment, design 1.000 mgd, average 0.350 mgd, Mid-Cost Components, Design Type: Interconnection

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
3.1.1	Piping - Interconnect - PVC	59,800	22
3.1.2	Piping - Interconnect - Excavation	91,114	22
3.1.3	Piping - Interconnect - Backfill and Compaction	42,241	22
3.1.4	Piping - Interconnect - Asphalt Patch	12,713	22
4.1.1	Valves - Isolation (PRV) and Street - Ductile Iron	8,217	25
6.1.1	Instrumentation - Flow Meters - Interconnect - Venturi	12,496	15
Indirect	Indirect and Add-On Costs (contingency from model)	101,239	22
	Process Cost	226,580	
	System Cost	327,819	
	O&M Cost	281,089	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	0
Process Engineering	27,190
Construction Management and GC Overhead	25,561
Site Work	0
Yard Piping	0
Geotechnical	0
Standby Power	0
Electrical (including yard wiring)	0
Mobilization and Demobilization	9,970
Architectural Fees for Treatment Building	0
Permits	0
Pilot Study	0
Land Cost	0
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	22,658
Legal, Fiscal, and Administrative (2.0%)	4,532
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	11,329

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (0 hrs/yr @ \$51.7408/hr)	5
Administrative (0 hrs/yr @ \$34.0506/hr)	3
Operator (1 hrs/yr @ \$30.4776/hr)	28
Purchased Water (127750 K gal @ \$2K gal)	255,500
Miscellaneous Allowance (0 @ \$)	25,554

## Non-Treatment, design 1.000 mgd, average 0.350 mgd, Mid-Cost Components, Design Type: New Well Construction

	1 a . (d)	
Item	Total Cost (\$)	Useful Life (yrs)
Well Items - Well Casing - Stainless Steel	246,420	45
Well Items - Well screen - PVC Schedule 40	16,708	22
Well Items - Plugs - PVC	343	22
Well Items - Well Drilling	45,975	N/A
Well Items - Gravel Pack	28,550	N/A
Well Items - Well Development	1,750	N/A
Well Items - Surface seal well, concrete fill	10,250	N/A
Piping - Well - PVC	1,230	22
Piping - Well - Excavation	9,111	22
Piping - Well - Backfill and Compaction	4,224	22
Valves - Motor/Air Operated (on/off) - Well Pump - Cast Iron	9,920	25
Pumps - Well Pump	28,686	20
Instrumentation - Flow Meters - Well - Venturi	12,496	15
Building Structures - Building 1 - Small Low Cost Shed	18,401	25
Indirect and Add-On Costs (contingency from model)	380,036	45
Process Cost	434,064	
System Cost	814,100	
O&M Cost	76,202	
Totals are computed before component costs are rounded		
	Well Items - Well screen - PVC Schedule 40  Well Items - Plugs - PVC  Well Items - Well Drilling  Well Items - Gravel Pack  Well Items - Well Development  Well Items - Surface seal well, concrete fill  Piping - Well - PVC  Piping - Well - Excavation  Piping - Well - Backfill and Compaction  Valves - Motor/Air Operated (on/off) - Well Pump - Cast Iron  Pumps - Well Pump  Instrumentation - Flow Meters - Well - Venturi  Building Structures - Building 1 - Small Low Cost Shed  Indirect and Add-On Costs (contingency from model)  Process Cost  System Cost  O&M Cost	Well Items - Well Casing - Stainless Steel246,420Well Items - Well screen - PVC Schedule 4016,708Well Items - Plugs - PVC343Well Items - Well Drilling45,975Well Items - Gravel Pack28,550Well Items - Well Development1,750Well Items - Surface seal well, concrete fill10,250Piping - Well - PVC1,230Piping - Well - Excavation9,111Piping - Well - Backfill and Compaction4,224Valves - Motor/Air Operated (on/off) - Well Pump - Cast Iron9,920Pumps - Well Pump28,686Instrumentation - Flow Meters - Well - Venturi12,496Building Structures - Building 1 - Small Low Cost Shed18,401Indirect and Add-On Costs (contingency from model)380,036Process Cost434,064System Cost814,100O&M Cost76,202

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	0
Process Engineering	52,088
Construction Management and GC Overhead	35,031
Site Work	5,126
Yard Piping	0
Geotechnical	54,017
Standby Power	51,369
Electrical (including yard wiring)	41,566
Mobilization and Demobilization	25,182
Architectural Fees for Treatment Building	1,656
Permits	0
Pilot Study	0
Land Cost	40,210
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	43,406
Legal, Fiscal, and Administrative (2.0%)	8,681
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	21,703

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (9 hrs/yr @ \$51.7408/hr)	441
Administrative (9 hrs/yr @ \$34.0506/hr)	290
Operator (85 hrs/yr @ \$30.4776/hr)	2,596
Facility maintenance (materials and labor) (400 sf @ \$5.7866/sf/yr)	2,315
Well pump (calculated as a percentage of capital)	287
Energy for well pumps (523 Mwh/yr @ \$0.1212/kwh)	63,347
Miscellaneous Allowance (0 @ \$)	6,927

## Non-Treatment, design 3.536 mgd, average 1.417 mgd, High-Cost Components, Design Type: Interconnection

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
3.1.1	Piping - Interconnect - PVC	177,500	22
3.1.2	Piping - Interconnect - Excavation	103,262	22
3.1.3	Piping - Interconnect - Backfill and Compaction	47,873	22
3.1.4	Piping - Interconnect - Asphalt Patch	14,408	22
4.1.1	Valves - Isolation (PRV) and Street - Ductile Iron	20,394	25
6.1.1	Instrumentation - Flow Meters - Interconnect - Magnetic	11,277	15
Indirect	Indirect and Add-On Costs (contingency from model)	173,295	22
	Process Cost	374,714	
	System Cost	548,009	
	O&M Cost	1,137,900	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	17,763
Process Engineering	44,966
Construction Management and GC Overhead	30,378
Site Work	0
Yard Piping	0
Geotechnical	0
Standby Power	0
Electrical (including yard wiring)	0
Mobilization and Demobilization	16,487
Architectural Fees for Treatment Building	0
Permits	0
Pilot Study	0
Land Cost	0
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	37,471
Legal, Fiscal, and Administrative (2.0%)	7,494
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	18,736

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (0 hrs/yr @ \$57.6375/hr)	5
Administrative (0 hrs/yr @ \$35.9445/hr)	3
Operator (1 hrs/yr @ \$39.3563/hr)	36
Purchased Water (517205 K gal @ \$2K gal)	1,034,410
Miscellaneous Allowance (0 @ \$)	103,445

## Non-Treatment, design 3.536 mgd, average 1.417 mgd, High-Cost Components, Design Type: New Well Construction

WBS#	Item	Total Cost (\$)	Useful Life (yrs)
1.1.1	Well Items - Well Casing - Stainless Steel	616,050	45
1.2.1	Well Items - Well screen - PVC Schedule 40	41,769	22
1.3.1	Well Items - Plugs - PVC	858	22
1.4	Well Items - Well Drilling	114,938	N/A
1.5	Well Items - Gravel Pack	71,375	N/A
1.6	Well Items - Well Development	4,375	N/A
1.7	Well Items - Surface seal well, concrete fill	25,625	N/A
3.2.1	Piping - Well - PVC	3,075	22
3.2.2	Piping - Well - Excavation	22,778	22
3.2.3	Piping - Well - Backfill and Compaction	10,560	22
4.2.2	Valves - Motor/Air Operated (on/off) - Well Pump - Stainless Steel	53,278	25
5.2	Pumps - Well Pump	85,420	20
6.2.1	Instrumentation - Flow Meters - Well - Magnetic	6,832	15
8.1.1	Building Structures - Building 1 - Low Quality	79,569	40
Indirect	Indirect and Add-On Costs (contingency from model)	948,179	45
	Process Cost	1,136,501	
	System Cost	2,084,680	
	O&M Cost	236,470	
	Totals are computed before component costs are rounded		

Breakdown of indirect and add-on costs	Total Cost (\$)
Contingency	58,752
Process Engineering	136,380
Construction Management and GC Overhead	75,554
Site Work	12,815
Yard Piping	0
Geotechnical	135,042
Standby Power	101,216
Electrical (including yard wiring)	105,693
Mobilization and Demobilization	64,197
Architectural Fees for Treatment Building	7,161
Permits	854
Pilot Study	0
Land Cost	57,309
Installation, Transportation, and O&P (0.0%)	0
Instrumentation and Control (0.0%)	0
Miscellaneous Allowance (10.0%)	113,650
Legal, Fiscal, and Administrative (2.0%)	22,730
Sales Tax (0.0%)	0
Financing during Construction (5.0%)	56,825

Breakdown of O&M costs	Annual Cost (\$/year)
Manager (21 hrs/yr @ \$57.6375/hr)	1,227
Administrative (21 hrs/yr @ \$35.9445/hr)	765
Operator (213 hrs/yr @ \$39.3563/hr)	8,380
Facility maintenance (materials and labor) (1000 sf @ \$5.7866/sf/yr)	5,787
Well pump (calculated as a percentage of capital)	854
Energy for well pumps (1633 Mwh/yr @ \$0.1212/kwh)	197,960
Miscellaneous Allowance (0 @ \$)	21,497





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Office of Water EPA-HQ-OW-2018-0780 April 2019 [ HYPERLINK "http://water.epa.gov/drink" ]

#### **Executive Summary**

On February 11, 2011 (76 FR 7762; USEPA, 2011a), the United States Environmental Protection Agency (EPA) announced its decision to regulate perchlorate under the Safe Drinking Water Act (SDWA) based on its finding that perchlorate meets the SDWA's three criteria for regulating a contaminant: 1) the contaminant may have an adverse effect on the health of persons, 2) the contaminant is known to occur or there is a substantial likelihood that the contaminant will occur in public water systems (PWSs) with a frequency and at levels of public health concern, and 3) in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by PWSs.

EPA is proposing a Maximum Contaminant Level Goal (MCLG) and a National Primary Drinking Water Regulation (NPDWR) for control of perchlorate in PWSs. In accordance with Section 1412(b)(3)(c) of SDWA, EPA must prepare a Health Risk Reduction and Cost Analysis (HRRCA) of the proposed Maximum Contaminant Level (MCL) and alternative MCLs. The HRRCA must assess the quantifiable and non-quantifiable costs that are likely to occur as a result of compliance with the MCL. These costs could be for new treatment processes as well as incremental monitoring and administrative costs. EPA must also provide an estimate of the health risk reduction benefits likely to occur as a result of the treatment to comply with each perchlorate concentration level assessed. System customers will benefit from health risk reductions associated with lower contaminant exposure levels.

EPA evaluated the available peer-reviewed science and supporting studies, as well as data collected by accepted methods on the national occurrence of perchlorate in drinking water. EPA determined that the data from the Unregulated Contaminant Monitoring Rule 1 (UCMR 1) are the best available nationally representative data for characterizing the frequency and levels of perchlorate occurrence in public drinking water systems. The UCMR 1 perchlorate monitoring of drinking water – a census of large PWSs (serving more than 10,000 people) and a nationally representative statistical sample of small systems (serving 10,000 people or fewer) – represents the most extensive, nationally representative monitoring program for perchlorate in public drinking water systems. PWSs conducted UCMR 1 monitoring for perchlorate in drinking water between 2001 and 2005. EPA used UCMR 1 data to estimate the national occurrence of perchlorate in public drinking water systems. To support this analysis, EPA also reviewed state-sponsored studies (including Arizona, California, Iowa, Maryland, Massachusetts, Nevada, New Jersey and Texas) and other drinking water occurrence data (including American Water Works Association Research Foundation, American Water System Survey, Consumer Confidence Reports and the Environmental Working Group).

Occurrence analyses based on this UCMR 1 Perchlorate Dataset identify analytical detections in PWSs located in 26 states and 2 territories. Analytical detections of perchlorate at or above 4  $\mu$ g/L were identified in 1.58% (540) of the 34,132 UCMR 1 perchlorate samples. An estimated 4.60% of large PWSs (141 large systems) serving approximately 16.2 million people reported at least one detection of perchlorate at or above 4  $\mu$ g/L and 1.0% of small PWSs (8 small systems) serving approximately 13,000 people reported perchlorate detections at or above 4  $\mu$ g/L. The percentage of ground water and surface water PWSs reporting perchlorate detections is about the same. While perchlorate analytical detections are fairly numerous and widespread

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geographically, the UCMR 1 findings indicate that perchlorate occurs at relatively low levels. Fifteen systems had detections of perchlorate greater than 18  $\mu$ g/L; two systems had detections of perchlorate greater than 56  $\mu$ g/L.

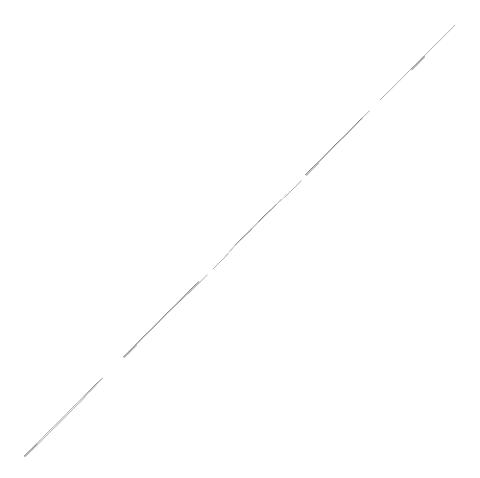
Exhibit ES-1. Systems and Populations with at least One Detection Relative to Select Thresholds Based on the updated UCMR 1 Perchlorate Dataset

System Type	Total Number of Systems in UCMR 1	Total Pop. Served by Systems in UCMR 1	Systems with at least one detection > threshold					
			≥ 4 µg/L		> 18 µg/L		> 56 µg/L	
			Number of Systems	Pop. Served by Systems	Number of Systems	Pop. Served by Systems	Number of Systems	Pop. Served by Systems
Small Systems (serving ≤10,000)	797	2,760,570	8	13,483	1	4,309	0	0
Large Systems (serving >10,000)	3,068	222,853,101	141	16,159,082	14	696,871	2	64,733
All Systems	3,865	225,613,671	149	16,172,565	15	701,180	2	64,733

<sup>&</sup>lt;sup>1</sup> The UCMR 1 minimum reporting level (MRL) for perchlorate was equal to 4 μg/L. Thus, assessments relative to the threshold of greater than or equal to 4 μg/L served to identify all perchlorate sample detections in UCMR 1.

#### **Contents**

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#### **Exhibits**

[ TOC \h \z \t "Caption,1,EXHIBIT (Caption),2,Caption-doc,3,Caption-doc + Centered,1" \c "Exhibit" ]



#### **Appendices**

- Appendix A: Occurrence in Ambient Water and in Drinking Water (from Sources Other than UCMR 1)
- Appendix B: A Review of Perchlorate Occurrence in California Public Drinking Water Systems: 1994-2011
- Appendix C: Data Quality Considerations from the Chamber of Commerce
- Appendix D: UCMR 1 Perchlorate Detections with Source Water Detection Categories Identified
- Appendix E: UCMR 1 (June 2013 Version) State-Level Occurrence Measures

#### **Abbreviations**

ADEQ Arizona Department of Environmental Quality

AMS Alfred Merritt Smith Treatment Plant

ASTSWMO Association of State and Territorial Solid Waste Management Officials

ATSDR Agency for Toxic Substances and Disease Registry

AWS American Water System

AWWA American Water Works Association

AwwaRF American Water Works Association Research Foundation

CA EPA California Environmental Protection Agency

CA OEHHA California's Office of Environmental Health Hazard Assessment

CA UCRM California's Unregulated Chemicals Requiring Monitoring

CAP Central Arizona Project

CASRN Chemical Abstract Services Registry Number

CCL Contaminant Candidate List

CCR Consumer Confidence Report

CDHS California Department of Health Services
CDPH California Department of Public Health

CWS Community Water System

DL Detection Limit

DQO Data Quality Objective
DWP Drinking Water Program

EP Entry Point

EPA Environmental Protection Agency

EWG / Environmental Working Group

HRRCA Health Risk Reduction and Cost Analysis
IA SWRL Iowa Statewide Rural Well Water Survey

IC Ion Chromatography

LC Liquid Chromatography

LCMRL Lowest Concentration Minimum Reporting Level

LIWC Long Island Water Conference

Massachusetts Department of Environmental Protection

[ PAGE \\* MERGEFORMAT ]

MCL Maximum Contaminant Level

MCLG Maximum Contaminant Level Goal

MDL Method Detection Limit

MEG Maximum Exposure Guidelines

MRL Minimum Reporting Level

MS Mass Spectrometry

MWD Metropolitan Water District

NAWQA National Water-Quality Assessment

NDEP Nevada Division of Environmental Protection

NHANES National Health and Nutrition Examination Survey

NJDEP New Jersey Department of Environmental Protection

NJDWQI New Jersey Drinking Water Quality Institute

NMED New Mexico Environment Department

NPDWR National Primary Drinking Water Regulation

NRC National Research Council

NTNCWS Non-Transient Non-Community Water System

NWIS National Water Information System

OGWDW Office of Ground Water and Drinking Water

PHG Public Health Goal

PPM Parts Per Million

PSV Preliminary Screening Values

PWS Public Water System

PWSID Public Water System Identification Number

QA Quality Assurance

QAPP Quality Assurance Project Plan

QC Quality Control RL Reporting Level

RM River Mountains Treatment Plant

RSD Relative Standard Deviation

SCDHS Suffolk County Department of Health Service

SCWA Suffolk County Water Authority

SDWA Safe Drinking Water Act

[ PAGE \\* MERGEFORMAT ]

SIM Selective Ion Monitoring

SNWS Southern Nevada Water System

SR Source Water

TCEQ Texas Commission on Environmental Quality

TDS Total Dissolved Solids

TNCWS Transient Non-Community Water System

TRI Toxics Release Inventory

UCMR 1 Unregulated Contaminant Monitoring Rule 1

USGAO United States Government Accountability Office

USGS United States Geological Survey

WaterRF Water Research Foundation

WSSC Washington Suburban Sanitary Commission

#### 1 Introduction

This occurrence and monitoring report describes the data and analyses used by the U.S. Environmental Protection Agency (EPA) to develop national estimates of perchlorate occurrence in public drinking water systems. Additional supplemental perchlorate occurrence background information and data were also reviewed. The information, data, and analyses described in this report are organized into six chapters and five appendices with a brief description of each chapter and appendix presented below.

- Chapter 1: The Introduction provides the regulatory history of perchlorate in the context of public drinking water and a summary of state drinking water standards and guidance levels.
- Chapter 2: Perchlorate Background provides information on perchlorate chemical and physical properties, sources of perchlorate, environmental fate, and laboratory analytical methods.
- Chapter 3: Perchlorate Occurrence Data from the Unregulated Contaminant Monitoring Rule 1 (UCMR 1) presents background information on the UCMR 1 Perchlorate Dataset, as well as the quality assurance / quality control (QA/QC) review of those data.
- Chapter 4: Analysis of the UCMR 1 Occurrence Data discusses the non-parametric approach for analyzing the occurrence data and presents results based on that analysis.
- Chapter 5: Laboratory Analytical Methods discusses the analytical methods used in the identification and quantification of perchlorate in drinking water.
- Chapter 6: References is a list of the cited and supporting scientific literature used in development of the document.
- **Appendix A** presents occurrence data compiled by states and other organizations for perchlorate from a variety of ambient water and non-UCMR 1 drinking water sources.
- Appendix B presents a detailed description and characterization of the perchlorate occurrence data provided by the State of California.
- **Appendix** C provides a summary of considerations for additional data quality measures related to comments from the Chamber of Commerce, as well as a comparison of current perchlorate data from the State of California with California UCMR 1 perchlorate data.
- **Appendix D** presents the public water system inventory and other background details regarding UCMR 1 perchlorate detections from source water samples and samples from entry points to the distribution systems as discussed in Section 3.4.2.
- **Appendix E** provides a summary of monitoring results and occurrence analyses by state based on the UCMR 1 Perchlorate Dataset (June 2013 Version).

# 1.1 SDWA Statutory Requirements and Rulemaking Process for Perchlorate in Drinking Water

The Safe Drinking Water Act (SDWA) Sections 1412(b)(1)-(6) and (15) describe requirements for regulating drinking water contaminants. The United States Environmental Protection Agency (EPA) included perchlorate on the first, second, and third Contaminant Candidate Lists (CCLs) that were published in the *Federal Register* on March 2, 1998, February 24, 2005, and October 8, 2009, respectively (USEPA, 1998; USEPA, 2005a; USEPA, 2009a). On May 1, 2007, EPA published an update on the agency's evaluation of perchlorate as part of the preliminary regulatory determination for 11 other CCL 2 contaminants (72 FR 24016; USEPA, 2007a). The agency did not make a preliminary determination for perchlorate as part of this notice. However, the agency requested public comment on the information included in the notice and on the options that the agency was evaluating, it also requested information that could assist the agency in its decision-making process.

On October 10, 2008, EPA published a preliminary regulatory determination for perchlorate (73 FR 60262; USEPA, 2008a), requesting public comment on its determination that development of a National Primary Drinking Water Regulation (NPDWR) for perchlorate would not present a meaningful opportunity for health risk reduction for persons served by public water systems (PWSs). The October 2008 notice describes in detail EPA's basis for its preliminary regulatory determination.

On January 8, 2009, EPA (USEPA, 2009b) announced that it was seeking advice from the National Academy of Sciences (NAS) before making a final determination on whether to issue a national regulation for perchlorate in drinking water. In conjunction with that announcement, EPA also issued an interim health advisory of 15 parts per billion (ppb or  $\mu g/L$ ) to assist state and local officials in addressing local contamination of perchlorate in drinking water and making a corresponding change to the factors considered in cleaning up Superfund sites.

On August 19, 2009 (74 FR 41883; USEPA, 2009c), EPA published the *Perchlorate Supplemental Request for Comments* and stated that it was not, at that time, planning to request additional National Research Council (NRC) review of issues related to perchlorate. Instead, EPA requested comment on additional approaches to analyzing data related to EPA's perchlorate regulatory determination. These additional comments were sought in an effort to ensure consideration of all potential options for evaluating whether there is a meaningful opportunity for human health risk reduction of perchlorate through an NPDWR. EPA stated that the alternative analyses (e.g., alternative health reference levels based on body weight and water consumption of 12 life stages ranging from birth to less than 21 years old) presented in the notice could lead the agency to make a determination to regulate perchlorate.

On February 11, 2011 (76 FR 7762; USEPA, 2011a), EPA announced its decision to regulate perchlorate under the SDWA based on its finding that perchlorate meets the SDWA's three criteria for regulating a contaminant: 1) the contaminant may have an adverse effect on the health of persons, 2) the contaminant is known to occur or there is a substantial likelihood that

the contaminant will occur in PWSs with a frequency and at levels of public health concern, and 3) in the sole judgment of the Administrator, regulation of such contaminant presents a meaningful opportunity for health risk reduction for persons served by PWSs.

On May 29, 2013, the EPA Science Advisory Board (SAB) recommended that EPA, as part of its development of a regulation for perchlorate, derive a perchlorate Maximum Contaminant Level Goal (MCLG) through the use of a physiologically-based pharmacokinetic/pharmacodynamic model based upon perchlorate's mode of action regarding human health. The findings of this modeling are to be applied toward the development of a Maximum Contaminant Level (MCL). EPA collaborated with the Food and Drug Administration to implement the SAB recommendations and conduct the modeling.

To address the SAB recommendations, EPA created a BBDR models that predicts changes in thyroid hormone levels as a result of nutritional iodine intake and perchlorate exposure in women prior to pregnancy and early gestation. These models were peer reviewed in January 2017<sup>4</sup>. Reviewers stressed the importance of developing an early pregnancy model when considering adverse neurodevelopmental impacts. EPA responded to this peer review by developing an early pregnancy model and updated key parameters for that model.

EPA carried out a subsequent peer review January 2018 to evaluate updates to the BBDR model and presented the alternative approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects. The January 2018 peer review was largely supportive of the efforts described in EPA's report titled "Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water"

When proposing an MCL, EPA must publish, and seek comment on, the Health Risk Reduction and Cost Analysis (HRRCA) of each alternative MCL considered (SDWA Section 1412(b)(3)(C)(i)). This includes the quantifiable and non-quantifiable benefits from reductions in health risk, including those from removing co-occurring contaminants (not counting the benefits resulting from compliance with other proposed or final regulations); costs of compliance (not counting costs resulting from other regulations); any increased health risks (including those from co-occurring contaminants) that may result from compliance; and incremental costs and benefits of each alternative MCL considered.

Once EPA makes a determination to regulate a contaminant in drinking water, SDWA Section 1412(b)(1)(E) requires that EPA issue a proposed NPDWR within 24 months and a final NPDWR within 18 months after the proposal (the statute allows a nine-month extension of this promulgation date).

Biologically Based Dose Response Models for the Effect of Perchlorate on Thyroid Hormones in the Infant, Breast Feeding Mother, Pregnant Mother, and Fetus: Model Development, Revision, and Preliminary Dose-Response Analysis. The report is available through the docket at [HYPERLINK "http://www.regulations.gov"] (Docket ID No. EPA-HQ-OW-2016-0439).

### 1.2 State Standards and Guidelines for Perchlorate in Drinking Water

There also have been state actions on perchlorate standards, guidelines, and advisories. In 2006, Massachusetts adopted a drinking water standard for perchlorate of 2  $\mu$ g/L, and in 2007, California promulgated a drinking water standard of 6  $\mu$ g/L. Other states have established non-enforceable guidance levels, action or advisory levels. [ REF \_Ref366605281 \\* MERGEFORMAT ] presents a summary of the state standards and guidance levels for perchlorate in drinking water and [ REF \_Ref366605289 \\* MERGEFORMAT ] presents a summary of the state guidance levels for perchlorate in ground water. Depending on the state, a particular level may require a PWS to notify the public, serve as a screening tool for further action, or guide clean-up actions.

Exhibit [ SEQ Exhibit \\* ARABIC ]: Summary of State Standards and Guidance Levels for Perchlorate in Drinking Water

State	Level (µg/L)	Description							
		State Drinking Water Standards							
California	California 6 Existing MCL								
Massachusetts	2	Existing MCL							
	State Dr	inking Water Guidance, Action of Advisory Levels <sup>1,2,3</sup>							
Arizona	14	Health-based guidance level (USEPA Region 9, 2016)							
Hawaii	26	Action level for drinking water (HI DOH, 2011)							
Maine	8.0	Maximum Exposure Guidelines (MEG) for Drinking Water (ME DHHS, 2016)							
Maryland	1	Advisory Level (ASTSWMO, 2011; Harford County Government, 2007)							
Nevada	18	Provisional action level based upon EPA guidance (NDEP, 2012)							
New Jersey	5	Interim Ground Water Quality Criteria (NJ DEP, 2016)							
New Mexico	1	Drinking Water Guidance Level (ASTSWMO, 2011)							
New Mexico	13.8	Tap water screening level (NMED, 2014)							
New York	5	Drinking Water Planning Level (ASTSWMO, 2011)							
INEW FOIR	18	State Guidance Level (NY DOH, 2010)							
Oregon	A	Recommended Action Level (OR DHS, 2004)							
Vermont	4	Interim enforcement standard (ASTSWMO, 2011)							

Adapted from the United States Government Accountability Office (USGAO) (2010) and Association of State and Territorial Solid Waste Management Officials (ASTSWMO) (2011).

 $<sup>^1</sup>$  A "Preliminary Screening Value" (PSV) of 24.5  $\mu$ g/L in Alabama had been cited in ASTSWMO (2011). However, that value can no longer be verified.

 $<sup>^2</sup>$  A "drinking water threshold level" and "interim action level" of 4  $\mu$ g/L in Kansas and Texas, respectively, had been cited in USGAO (2010). However, these values can no longer be verified.

<sup>&</sup>lt;sup>5</sup> In January 2011, the State of California proposed a Public Health Goal (PHG) of 1 μg/L for perchlorate. The existing California MCL is based on an earlier California 2004 PHG of 6 μg/L (CA EPA, 2011). In December 2012, CA's Office of Environmental Health Hazard Assessment (CA OEHHA) released a revised draft technical support document for the [ HYPERLINK "http://www.oehha.ca.gov/water/phg/120712Perchlorate.html" ] (CA EPA, 2012).

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Summary of State Guidance Levels for Perchlorate in Ground Water

State	Level (µg/L)	Description					
	State Gu	idance, Action or Advisory Levels for Perchlorate in Ground Water <sup>1</sup>					
Alaska	14	Groundwater clean-up level for ammonium perchlorate (AK DEC, 2017)					
Alaska	26	Groundwater clean-up level for perchlorate (AK DEC, 2009)					
Florida	4	Clean-up target level for potable water ( <i>This level, established in regulation, is not a standard but serves as a default level for contaminated site clean-ups. Alternative levels may be used where there is sufficient site-specific information.</i> ) (FL Department of the State, 2005)					
Florida	40	Clean-up target level for ground water of low yield or poor quality ( <i>This level, established in regulation, is not a standard but serves as a default level for contaminated site clean-ups. Alternative levels may be used where there is sufficient site-specific information.</i> ) (FL Department of the State, 2005)					
Hawaii	26	Action levels for ground water that could be a source of drinking water (HI DOH, 2011)					
i iawaii	600	Action levels for ground water not used as a source of drinking water (HI DOH, 2011)					
lowa	15	Statewide Standards for a Protected Groundwater Source (IA DNR, 2016)					
IOWa	75	Statewide Standards for an Unprotected Groundwater Source (IA DNR, 2016)					
Kansas	10.9	Default risk-based clean-up level for residential or drinking water pathway (based on established equations and current EPA reference dose) (KDHE, 2010)					
Kalisas	70.9	Default risk-based clean-up level for nonresidential pathway (based on established equations and current EPA reference dose) (KDHE, 2010)					
Maryland	2.6	Groundwater clean-up standards for Type I and II Aquifers (MDE, 2008)					
Missouri	10.9	Groundwater Target Concentrations at Point of Exposure (MO DNR, 2006)					
Nebraska	6.4	Voluntary Clean-up Program Remediation Goals (NDEQ, 2012)					
Nevada	18	Provisional action level used as default clean-up level for all ground water (NDEP, 2012)					
New Mexico	4-18	Ground water clean-up level (NMED, 2004)					
Texas	17 /	Protective clean-up level for residential land use (TCEQ, 2016)					
Техаъ	51	Protective clean-up level for industrial/commercial land use (TCEQ, 2016)					
	້ 2	Interim preventive action level (VT DEC, 2015)					
Vermont 4		Interim enforcement standard; interim groundwater quality standard. ( <i>This level is considered guidance, despite its being termed a "standard."</i> ) (VT DEC, 2015)					
Wisconsin	0.1	Public health ground water quality standard preventive action limit (WI Administrative Code, 2017)					
Wisconsin	1	Public health ground water quality standard enforcement standard (WI Administrative Code, 2017)					

Adapted from USGAO (2010) and ASTSWMO (2011).

 $<sup>^3</sup>$  In a self-published, non-peer-reviewed study by Integral Consulting (2016), additional and/or updated state advisories for perchlorate were noted, including: lowa, with an advisory level of 4.9  $\mu$ g/L; New Hampshire, with a public health goal of 1  $\mu$ g/L and; Vermont, with an advisory level of 2.2  $\mu$ g/L. These advisories are not included on the drinking water-related websites of these three states.

 $<sup>^1</sup>$  A "Preliminary Screening Value" (PSV) of 24.5  $\mu$ g/L in Alabama had been cited in ASTSWMO (2011). However, that value can no longer be verified.

# 2 Perchlorate Background

This section summarizes background on perchlorate including its chemical and physical properties; sources; environmental fate and transport of the perchlorate ion; and laboratory analytical methods.

#### 2.1 Chemical and Physical Properties

Perchlorate is an inorganic chemical containing one chlorine atom bound to four oxygen atoms in a tetrahedral configuration (see [ REF \_Ref366605356 ]). As such, perchlorates (ClO<sub>4</sub>-) are a group of anions that form salts with most cations.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Chemical Structure of Perchlorate

Figure based on information from the Hazardou's Substances Data Bank (HSDB), 2012.

Commonly used perchlorate salts include ammonium perchlorate and potassium perchlorate. Perchlorate is also used as a component of sodium perchlorate, aluminum perchlorate, hydrazine perchlorate, hydrogen perchlorate, hydroxylammonium perchlorate, lithium perchlorate, magnesium perchlorate, nitronium perchlorate, and as perchloric acid. As an anion, there is no single Chemical Abstracts Service Registry Number (CASRN) for perchlorate, as each salt has its own properties. Registry numbers for the most common forms of perchlorate and their chemical and physical properties are presented in [ REF \_Ref353686474 \\* MERGEFORMAT ].

Exhibit [ SEQ Exhibit \\* ARABIC ]: Physical and Chemical Properties

		Data								
Property	Perchlorate	Ammonium perchlorate	Potassium perchlorate	Sodium perchlorate						
CASRN	14797-73-0	7790-98-9	7778-74-7	7601-89-0						
Chemical Formula	CIO4-	NH <sub>4</sub> ClO <sub>4</sub>	KCIO <sub>4</sub>	NaClO <sub>4</sub>						
Molecular Weight	99.45 g/mol (ChemIDPlus, 2011)	117.49 g/mol (Merck, 1983)	138.55 g/mol (Merck, 1983)	122.44 g/mol (Merck, 1983)						

		Da	ta		
Property	Perchlorate	Ammonium perchlorate	Potassium perchlorate	Sodium perchlorate	
Color/ Physical State	1	Colorless or white orthorhombic crystals (Hazardous Substances Data Bank [HSDB], 2011)	Colorless orthorhombic crystals or white crystalline powder (HSDB, 2011; Agency for Toxic Substances and Disease Registry [ATSDR], 2008; Merck, 1983)	White orthorhombic, deliquescent crystals (HSDB, 2011; CRC Press, 1981; Merck, 1983)	
Boiling Point			400 deg C dec. (CRC Press, 1981)		
Melting Point		130 deg C (ATSDR, 2008)	400 deg C dec. (ATSDR, 2008); 525 deg C (HSDB, 2011)	471 deg C dec. (ATSDR, 2008) 482 deg C dec (CRC Press 1981); 480 deg C dec. (HSDB, 2011)	
Density		1.95 g/mL (CRC Press, 1981; Merck, 1983)	2.52 g/mL at 10 deg C (CRC Press, 1981; Merck, 1983)	2.52 g/cm3 (HSDB, 2011); 2.02 g/mL (Merck, 1983; ATSDR, 2008)	
Vapor Pressure		Very low (ATSDR, 2008)	Very low (ATSDR, 2008)	Very low (ATSDR, 2008)	
Solubility in Water		200 g/L at 25 deg C (HSDB, 2011), 249 g/L at 25 deg & (ATSDR, 2008)	20.6 g/L at 25 deg C (ATSDR, 2008); 15 g/L at 25 deg C (HSDB, 2011)	2100 g/L at 25 deg C (HSDB, 2011; ATSDR, 2008)	
Other Solvents	- /	Acetic acid, slightly soluble alcohol (CRC Press, 1981); methanol, slightly soluble ethanol and acetone, very slightly soluble ethyl acetate and ether (HSDB, 2011)	Very slightly soluble alcohol (CRC Press, 1981); insoluble in alcohol and ether (HSDB, 2011)	Alcohol (CRC Press, 1981)	

g/mol = grams per mole; "--" indicates that no information was found; deg C = degrees Celsius; dec. = decomposes;  $g/mL = grams per milliliter; g/cm^3 = grams per cubic centimeter; g/L = grams per liter$ 

#### 2.2 Sources of Perchlorate

Perchlorate is both a naturally occurring and man-made chemical. The following section presents information on perchlorate's natural sources; its production and anthropogenic sources; and environmental fate and transport when naturally occurring or man-made perchlorate is present in soil and/or water.

#### 2.2.1 Natural Sources

Caliche ores, found in Chile, rich in sodium nitrate (NaNO<sub>3</sub>), are a natural source of perchlorate (USEPA, 2001a). This sodium nitrate (known as Chilean saltpeter) has been mined and refined to produce commercial fertilizers used in the United States. Perchlorate has also been found in other geologic materials. Orris *et al.* (2003) measured perchlorate at levels exceeding 1,000 parts per million (ppm or mg/kg) in several samples of natural minerals, including potash ore from New Mexico and Saskatchewan, Canada; playa crust from Bolivia; and hanksite from California.

Rao *et al.* (2007) found widespread, naturally-occurring perchlorate in diverse unsaturated zones (just below the root zone) in arid and semi-arid regions of Nevada, New Mexico, Texas and Utah. It was postulated that the amount of naturally-occurring perchlorate in aggregate is sufficiently large to affect groundwater when recharge percolates through these unsaturated zones. The perchlorates and other salts (e.g., chlorides) in this region are present due to atmospheric deposition and concentration through evaporation over thousands of years. Walvoord *et al.* (2003) as cited in Trumpolt *et al.* (2005) presents the theory that chlorides from land and sea are blown into the atmosphere where they react photochemically with ozone to produce perchlorate in a process similar to that which produces nitrate. Plummer *et al.* (2005) as cited in Rao *et al.* (2007) cite meteoric events as potential natural sources of perchlorate.

Dasgupta *et al.* (2005) detected perchlorate in many rain and snow samples and demonstrated that perchlorate is formed by a variety of simulated atmospheric processes suggesting that natural, atmospherically derived perchlorate exists in the environment. Barron *et al.* (2006) developed a method for the rapid determination of perchlorate in rainwater samples, with a detection limit (DL) between 70 and 80 ng/L. Of 10 rainwater samples collected in Ireland in 2005, perchlorate was detected in 4 samples at concentrations between 0.075  $\mu$ g/L and 0.113  $\mu$ g/L, and in 1 other sample at 2.8  $\mu$ g/L. Kang *et al.* (2006) conducted seven-day experiments to determine if it was possible to produce perchlorate by exposing various chlorine intermediates to UV radiation in the form of high intensity UV lamps and/or ambient solar radiation. Perchlorate formation was demonstrated in aqueous salt solutions with initial concentrations of hypochlorite, chlorite, or chlorate between 100 and 10,000 mg/L.

#### 2.2.2 Production and Use

While perchlorate has a wide variety of industrial uses, it is primarily used in the form of ammonium perchlorate as an oxidizer in solid fuels used to power rockets, missiles, and fireworks. Perchlorate can also be present in road flares, lubricating oils, matches, aluminum refining, rubber manufacturing, paint and enamel manufacturing, leather tanning, and as a dye mordant

As noted above, Chilean saltpeter has been mined and refined to produce commercial fertilizers. Before 2001, these accounted for about 0.14% of fertilizer application in the United States (USEPA, 2001a). USEPA (2001a) conducted a broad survey of fertilizers and other raw materials and found that all products surveyed were devoid of perchlorate except for those known to contain or to be derived from mined Chilean saltpeter.

Historically, American Pacific Corporation and Kerr-McGee were the major producers of ammonium perchlorate. Their perchlorate production took place at facilities in Nevada and Utah. Smaller manufacturers located in New York, Oregon, Mississippi, and California ceased production between 1948 and 1975. Kerr-McGee ceased ammonium perchlorate production in July 1998 (Nevada Division of Environmental Protection [NDEP], 2011).

No production data on this contaminant are available from EPA's Inventory Update Reporting program, and no industrial release data are available from EPA's Toxics Release Inventory (TRI). The list of chemicals for which TRI reporting is required has never included perchlorate (USEPA, 2011b).

The Agency for Toxic Substances and Disease Registry (ATSDR) (2008) reports that recent production data for ammonium perchlorate as well as the other forms of perchlorate listed in [REF\_Ref353686474 \\* MERGEFORMAT] are lacking. In 1994, U.S. production of ammonium perchlorate was estimated at 22 million pounds (Mendiratta *et al.*, 1996).

Sodium hypochlorite (NaOCl) is effectively used for water disinfection. However, perchlorate has been detected in hypochlorite solutions. A study by the American Water Works Association and Water Research Foundation (AWWA/WaterRF, 2009) found that perchlorate can be present in hypochlorite solutions and can continue to form with the rate of formation depending on storage conditions. The study found that to minimize perchlorate formation sodium hypochlorite should be stored in dark and cool conditions, diluted if possible, and used within a few weeks of manufacture.

In an earlier limited study, the Massachusetts Department of Environmental Protection (MassDEP) found that perchlorate may be present in sodium hypochlorite solutions used in water and wastewater treatment plants, and that the level of perchlorate occurrence depends upon storage conditions and the initial purity of the stock solution (MassDEP, 2006a). The Town of Tewksbury, Massachusetts conducted a small study to evaluate the impact of storage conditions (temperature and light) on a new shipment of sodium hypochlorite stock solution. Perchlorate concentrations in the new stock solution were found to increase from 0.2  $\mu$ g/L to levels ranging from 995  $\mu$ g/L to 6,750  $\mu$ g/L, depending on the storage conditions. Accounting for the large dilution factor (e.g., 20,090 to 1 ratio) used in chlorination processes at drinking water treatment plants, MassDEP (2006a) concluded that "absent additional efforts to minimize breakdown of hypochlorite solutions, it would appear that low levels of the perchlorate ion (0.2  $\mu$ g/L to 0.4  $\mu$ g/L) detected in a drinking water supply disinfected with sodium hypochlorite solutions could be attributable to the chlorination process."

It is not clear at this time what proportion of perchlorate found in public water supplies or entering the food chain comes from these various natural and anthropogenic sources. The significance of different sources likely varies regionally. A study by Dasgupta *et al.* (2006) analyzes the three principal sources of perchlorate and their relative contributions to the food chain. These include use as an oxidizer including rocket propellants, Chilean saltpeter used principally as fertilizer, and perchlorate produced by natural atmospheric processes. Dasgupta *et al.* (2006) concluded that while there may be some localized exceptions, fertilizer with Chilean saltpeter likely has an equal or greater contribution to the food chain than oxidizer contributions

(not including fireworks). Contributions of perchlorate by natural processes are generally even less.

## 2.3 Environmental Fate and Transport of the Perchlorate Ion

Perchlorate salts are highly soluble in water, and because perchlorate sorbs poorly to mineral surfaces and organic material, perchlorate is mobile in soil and aqueous environments (ATSDR, 2008; USEPA, 2002).

Most perchlorate salts exposed to water will readily dissolve and dissociate into the cation and perchlorate anion (Gullick *et al.*, 2001). The perchlorate ion is unlikely to form insoluble metal complexes in water or be removed from water in this manner (Cotton and Wilkinson, 1980). The perchlorate ion is very stable and inert to reduction despite the high oxidation state of chlorine (Urbansky, 2000). Ionized salts will not undergo hydrolysis and perchlorates are not expected to undergo direct photolysis or to volatilize in water (ATSDR, 2008; HSDB, 2011). Therefore, the ion may persist for decades under normal environmental conditions in ground water and surface water (Gullick *et al.*, 2001).

Biological removal and uptake of perchlorate has been observed in aquatic conditions. Anaerobic microbial biodegradation of perchlorate occurs in anoxic ground water and sediments (ATSDR, 2008). In the absence of nitrate and with influent perchlorate concentrations of up to 32 mg/L, wetlands have been shown to reduce perchlorate concentrations to less than 4 µg/L (Tan *et al.*, 2004). In a study of willow trees in sand bioreactors designed to remove perchlorate from contaminated water, 11% of the original perchlorate concentration measured throughout the whole tree was lost to degradation (to chloride) in the leaves (Nzengung *et al.*, 1999). Aquatic organisms have also demonstrated uptake of perchlorate. One lake study found an indication of food transfer of perchlorate from ingestion of contaminated periphyton, detritus, or invertebrates (Theodorakis *et al.*, 2006). However, another study found that, in aquatic organisms, bioconcentration of perchlorates is low (Dean *et al.*, 2004). For a detailed discussion on the biological treatment of perchlorate see EPA's report entitled, "Technologies and Costs for Treating Perchlorate-Contaminated Waters."

The perchlorate ion is unlikely to adsorb to soil since studies show that in solutions of moderate ionic strength, the perchlorate ion is only weakly adsorbed to mineral surfaces (Logan, 2001; Urbansky and Brown, 2003; Urbansky and Collette, 2001). Therefore, perchlorate will be highly mobile in soil, traveling over the soil with surface water runoff or migrating through the soil to ground water systems (ATSDR, 2008; Gullick *et al.*, 2001). This has been corroborated by studies that have measured perchlorate in surface water and ground water far from release sites (ATSDR, 2008). While perchlorate is not subject to volatilization (partitioning to the gaseous or vapor phase) to any significant extent, it may be transported to the atmosphere by mechanical means: e.g., wind-borne erosion of perchlorate aerosols (fine particles in the solid phase) and perchlorate-contaminated soil particles (ATSDR, 2008).

Removal of perchlorate from the soil by living organisms has been shown to occur through microbial degradation as well as plant uptake and degradation (ATSDR, 2008; Tipton *et al.*, 2003). Anaerobic microbial degradation in soil has been seen in both laboratory and in situ studies; however, soil type and the presence of nitrate will influence the rate of degradation

(Herman and Frankenberger, 1998; Logan, 1998; Tan et al., 2004). In one experimental study, perchlorate in Yolo loam fully degraded in 30 days while no degradation occurred in Columbia loam (Tipton et al., 2003). Both soil samples were three years old. The authors inferred that carbon-rich Yolo loam had preserved colonies of perchlorate-digesting bacteria better than the carbon-poor Columbia loam, and viewed their results as confirming the importance of biodegradation (and the relative unimportance of adsorption) as a fate process for perchlorate in soil. A study of sediment and soil from two Texas sites contaminated by perchlorate showed that degradation in the soil and sediments did not readily occur until most of the nitrate had degraded (Tan et al., 2004). A study of perchlorate uptake by tobacco showed that through a wide range of soil perchlorate concentrations, uptake and accumulation occurred but there were not enough data available to show whether or not the plant degraded the perchlorate (Ellington et al., 2001). Perchlorate uptake in plants has also been observed in salt cedar, cucumber, lettuce, and soybean, though nutrients such as nitrate may hinder the uptake of perchlorate (Urbansky et al., 2000; Yu et al., 2004).

Perchlorate salts released to the atmosphere are expected to exist adsorbed to particulate matter or as solid aerosols (ATSDR, 2008). Removal is likely to occur through deposition not degradation since reaction with gas-phase oxidants and photolysis is unlikely (ATSDR, 2008). A study investigating perchloric acid photolysis in the atmosphere found photolysis to be negligible (Jaegle *et al.*, 1996).

# 3 Perchlorate Occurrence Data in Unregulated Contaminant Monitoring Rule 1

This section describes the UCMR 1 perchlorate occurrence data set used by EPA to generate national estimates of perchlorate occurrence in drinking water. The agency determined that the best available source of perchlorate occurrence data in drinking water is the EPA administered UCMR 1. UCMR 1 is the only nationally representative drinking water monitoring program that has sampled for perchlorate. This section provides an overview of the UCMR program and the first UCMR monitoring cycle, UCMR 1. Details are provided to characterize the UCMR 1 data including sampling schedule, frequency, locations, system source water types and size, and the perchlorate laboratory analytical method used. See Section 4 for a more detailed discussion of the perchlorate occurrence analysis conducted using the UCMR 1 data.

In addition to the UCMR 1 data, EPA evaluated numerous national-scale studies and surveys, and state-sponsored studies of perchlorate occurrence in drinking water and ambient water. Appendix A presents a summary of these datasets, with the exception of the State of California. Data from the State of California are described separately and in detail in Appendix B.

### 3.1 UCMR 1 Program Overview

In 1999, EPA developed the UCMR program in coordination with the Contaminant Candidate List (CCL) and the National Drinking Water Contaminant Occurrence Database to provide national occurrence information on unregulated contaminants (64 FR 50556, USEPA, 1999d; 65 FR 11372, USEPA, 2000; and 66 FR 2273, USEPA, 2001b). UCMR 1 established a three-tiered approach for monitoring contaminants based on the availability of analytical methods and information on contaminant properties. The first tier and highest priority of the three-tier ranking system, designated as List 1, included perchlorate and other unregulated contaminants for which suitable laboratory methods are available; these were scheduled to undergo full "Assessment Monitoring."

The UCMR operates on a five-year cycle, with the first cycle extending from 2001 through 2005, although most monitoring was conducted from 2001 to 2003. For UCMR 1, EPA required all PWSs serving more than 10,000 people ("large" systems), plus a statistically representative national sample of 800 PWSs serving 10,000 people or fewer ("small" systems), to conduct Assessment Monitoring of List 1 contaminants (USEPA, 2001c). Approximately one-third of the participating small systems were scheduled to monitor List 1 contaminants during each calendar year from 2001 through 2003. Large systems could conduct one year of monitoring for List 1 contaminants anytime during the 2001-2003 UCMR 1 period.

The UCMR 1 program design, including system selection and overall monitoring approach, was peer reviewed (USEPA, 1999e). The program was designed expressly to provide nationally representative occurrence data for unregulated contaminants in public drinking water

systems.<sup>6</sup> There is no nationally representative alternative to the UCMR 1 data. EPA selected PWSs to conduct monitoring for the UCMR 1 program based on the number of people they serve, the source of their water, and whether they serve the same customers year-round. All large community water systems (CWSs) and non-transient non-community water systems (NTNCWSs), as well as a national sample of 800 small CWSs and NTNCWSs were required to conduct monitoring under the UCMR 1. Two categories of PWSs were exempt from UCMR 1 monitoring: PWSs that purchase their entire water supply from another PWS and transient non-community water systems (TNCWSs). For more details on the classification of public drinking water systems, go to: [HYPERLINK

"http://water.epa.gov/infrastructure/drinkingwater/pws/factoids.cfm" ].

The objective of the UCMR 1 sampling approach for small systems was to collect contaminant occurrence data from a statistically selected, nationally representative sample of systems. Small system sampling was stratified and population-weighted, and included some sampling adjustments such as selection of at least two systems from each state. As previously stated, with contaminant monitoring data from all large PWSs and a nationally representative sample of small PWSs, the UCMR 1, List 1 Assessment Monitoring program is nationally representative of contaminant occurrence in public drinking water systems and therefore is suitable for national estimates of perchlorate occurrence in public drinking water systems.

Perchlorate occurrence is known to have decreased in some important source waters (such as the Lower Colorado River) and in the two states (Massachusetts and California) that now regulate perchlorate in drinking water. (These important cases are discussed in Appendix A and Appendix B of this report.) However, because perchlorate remains unregulated and largely unmonitored across most of the United States, other locations, areas and regions have no newer comparable data (and across these other regions perchlorate occurrence could have stayed about the same, increased or decreased). Therefore, to conduct a consistent and statistically sound assessment of national perchlorate occurrence in public drinking water systems, EPA is using a modified UCMR 1 dataset in this report.

## 3.2 Monitoring Frequency and Location

UCMR 1 required surface water systems to sample four times at each entry point to the distribution system (or entry point) over a one-year period, while ground water systems were required to sample twice per entry point over a one-year period. One of the quarterly (surface water systems) or semi-annual (ground water systems) sampling events had to occur in the defined vulnerable period of May through July, or an alternate vulnerable period designated by the state, to ensure monitoring of potentially higher contaminant concentrations. Surface water systems had to select either the first, second, or third month of a quarter and then take the remaining samples at three-month intervals for the following three quarters of the monitoring year. Ground water systems were required to designate a sample collection during one of the

<sup>&</sup>lt;sup>6</sup> See USEPA (2005e), USEPA (2007b), and USEPA (2008b) for more information on all aspects of UCMR 1, including study design, completeness, and quality assurance/quality control procedures.

<sup>&</sup>lt;sup>7</sup> For UCMR 1 a vulnerable period is the season of greatest likelihood of contaminant occurrence, generally the months of late spring and early summer which are characterized by high volumes of surface water runoff and ground water recharge.

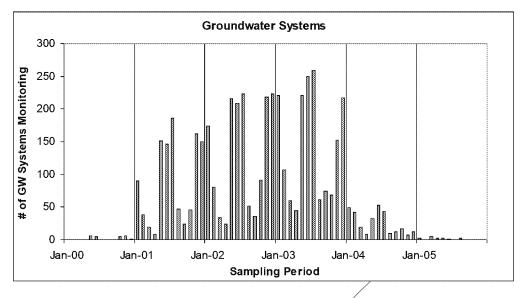
most vulnerable months, and then schedule another sample collection within five to seven months.

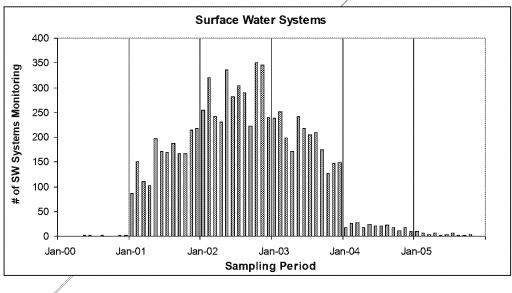
PWSs sampled at entry points after treatment. These entry points were meant to be representative of each principal non-emergency source of water in use over the 12-month monitoring period. UCMR 1 explicitly allowed source water monitoring, under certain conditions. As EPA stated in the Federal Register on September 17, 1999, "In response to public comment, EPA modified the rule [from the proposal] to allow alternative sampling points to be used: sampling points identified by the State for compliance monitoring under 40 CFR 141.24(f)(1), (2), and (3), and/or source (raw) water sampling points, if the State uses source water monitoring as a more stringent monitoring requirement" (64 FR 50570; USEPA, 1999d). EPA also stated that, "If monitoring at source (raw) water sampling points indicates detection of any of the contaminants on the monitoring list, then the system in most cases will be required to shift its unregulated contaminant monitoring to the entry point to the distribution system. These flexibilities in the sampling location should enable systems and States to coordinate compliance and unregulated contaminant monitoring more extensively."

EPA designed the UCMR 1 program with geographic, geologic, and temporal variability in mind. A census of all large systems and a primarily population-weighted statistical sample of small systems ensured monitoring across the varied geography and geology of the country. An additional program requirement was the selection for monitoring of at least two small systems in each state. The study design addressed temporal variability in contaminant occurrence by defining a vulnerable period and requiring at least one UCMR 1 sample at each system during that period. Large systems could conduct their one year of monitoring anytime during the UCMR 1 period from 2001 to 2003. Like small systems, their monitoring schedules were spread throughout the year and were to include one sample during what was considered the most vulnerable season. In this way, the national UCMR 1 results reflect multiple seasons and multiple years across the country and therefore are not biased by weather conditions of a single season, year, or geographic region.

[REF\_Ref526571210 h] illustrates the monthly distribution of ground water and surface water perchlorate sampling events. Ground water sampling events, which were conducted biennially, were concentrated in the summer months (May, June, July) and the winter months (November, December, January). No distinct seasonal pattern is evident in the surface water sampling, as those systems sampled on a quarterly schedule.

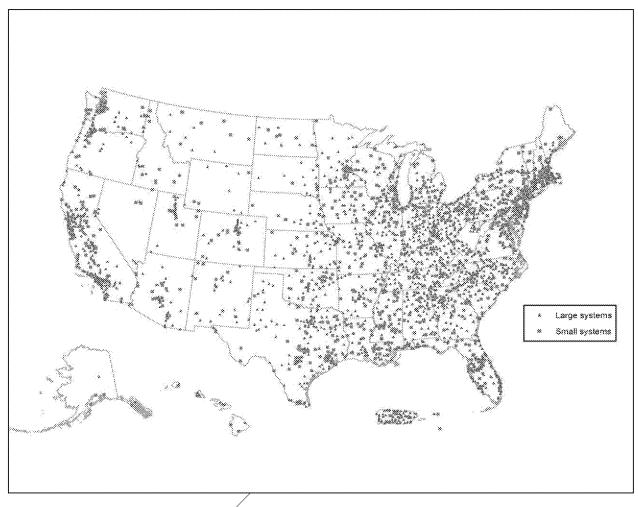
Exhibit [ SEQ Exhibit \\* ARABIC ]: Number of PWSs Collecting UCMR 1 Ground Water (GW) and Surface Water (SW) Samples for Perchlorate Analysis by Month During the Sampling Period (Top: GW Systems; Bottom: SW Systems)





[ REF \_Ref358310547 ] is a map of all large and small systems that submitted UCMR 1 data to EPA. At least two small systems were sampled in every state and most territories. One large system and two small systems from American Samoa were originally included in the sampling plan, but none of these three systems provided data.

### Exhibit [ SEQ Exhibit \\* ARABIC ]: PWSs with UCMR 1 Monitoring Results



# 3.3 Completeness of UCMR 1 Perchlorate Dataset

To ensure that occurrence estimates based on UCMR 1 data dependably reflect national conditions, EPA assessed the completeness and representativeness of the UCMR 1 contaminant sample data. Background discussions of data quality issues can be found in the UCMR 1 statistical design (USEPA, 2001d) and the quality assurance project plan (QAPP) (USEPA, 2003). The QAPP specified quantitative data quality objectives (DQOs) for the completeness and representativeness of small system data collected under UCMR 1. The small system data in the final UCMR 1 data set satisfy those DQOs, indicating the small system data are complete and representative. Although no formal DQOs were established for large systems, the large system census had a very high participation rate and a very large portion of the submitted data passed the general data quality criteria checks. These and other quality assessments suggest the large system contaminant occurrence data are dependable for national contaminant occurrence analyses.

Nearly all PWSs (99.2%) reported the results of their perchlorate assessment results to EPA (3,865 large and small PWSs responded out of a total of 3,897 PWSs). Of the 800 representative small systems required to collect samples, EPA received monitoring results from 797 systems (99.6%). Of the 3,097 large systems identified for inclusion in the census, EPA received monitoring results from 3,068 systems (99.1%). About two-thirds of the non-responsive systems were served by ground water (note that perchlorate detections were more frequent in surface water systems).

Texas had 37.5% (12 of the 32) of the non-responsive large PWSs. Of the 184 large PWSs in Texas that did report perchlorate results, about 2.2% had perchlorate detections. (Nationally, an estimated 4.6% of large PWSs reported at least one detection of perchlorate). These 32 non-responsive large systems serve approximately 0.7% of the total population. If any of these non-responsive systems had detectable levels of perchlorate, then the UCMR 1 results would underestimate actual occurrence.

# 3.4 Summary of UCMR 1 Perchlorate Monitoring Data and QA/QC Review

This section provides a description of the two phases of QA/QC review of the UCMR 1 Perchlorate Dataset. Section 3.4.1 describes EPA's initial quality review and data check of the UCMR 1 data. Section 3.4.2 describes EPA's consideration of the "Information Quality Guidelines (IQG) Request for Correction" received from the Chamber of Commerce in September 2012 and the resulting QA/QC review of the UCMR 1 perchlorate data. Section 3.4.3 summarizes the final version of the UCMR 1 perchlorate data set (post-QA/QC reviews) that was used as the basis of the occurrence analyses presented in this report.

#### 3.4.1 QA/QC Review - Phase 1

UCMR 1 data were collected from all 50 states, plus Washington, D.C., Tribal Nations, Puerto Rico, the American Virgin Islands, Guam, and the Commonwealth of the Northern Mariana Islands. A total of 34,728 perchlorate samples were collected with 647 detections reported. Samples were collected between May 1, 2000 and October 25, 2005, with almost 94% of samples collected between January 2001 and December 2003. EPA reviewed the 34,728 perchlorate samples and identified 397 sample records (10 of which were detections) that did not meet quality approval requirements for the following reasons:

- (1) records from non-approved perchlorate labs;
- (2) records identified as duplicates (i.e., having the same PWSID, Facility ID, Sample Point ID, and sample collection date)
  - a. if there were duplicate detections, the lesser of the two analytical results was deleted;
  - b. if there was a mix of non-detect and detect duplicates, the non-detect record was deleted;
  - c. if there were duplicate non-detections, all but one of the duplicate records was deleted;

- (3) records from CA4810015 were deleted because it was determined the system uses the same water source as CA4810003, and including data from both systems would provide duplicate results; and
- (4) records from four systems were excluded (MA4261024, PR0005226, PR0005246, and PR0005617) because the size of the populations they served had changed and they were no longer officially considered large systems.

In all, 397 samples and 10 detections were identified from the QA/QC review described above; see USEPA (2008b) for more information on the QA/QC review. Excluding those 397 samples and 10 detections resulted in a UCMR 1 perchlorate data set with 34,331 samples from 3,865 PWSs with a reported 637 detections. A second phase of data quality checks, described below in Section 3.4.2, resulted in additional revisions to the data set.

#### 3.4.2 QA/QC Review – Phase 2

In September of 2012, EPA received a "Request for Correction" letter from the United States Chamber of Commerce regarding information and data (regarding the occurrence of perchlorate in drinking water) used by EPA in its determination to regulate perchlorate. This letter stated that EPA relied upon data that did not comply with data quality guidelines. In response to the letter, EPA reassessed some components of the UCMR 1 data and some more recent studies that considered local perchlorate occurrence. A summary of EPA's consideration of the Chamber of Commerce comments is included below. Additional details regarding these considerations are included in Appendix C of this report.

#### UCMR 1 Perchlorate Source Water Sampling Data Review

The Chamber of Commerce letter stated that some UCMR 1 raw source water sample analytical detections of perchlorate did "not comply with data quality guidelines because it was not collected by accepted methods." Of the 34,331 total samples in the UCMR 1 Perchlorate Dataset, 69% (23,731) were collected at the entry points to the distribution system while the remaining 31% (10,600) were collected from untreated (but UCMR 1 eligible) source water sample locations. Of the 637 samples that detected perchlorate, 56% (355) were collected at the entry points to the distribution system while the remaining 44% (282) were collected from source water sample locations. All source water samples with perchlorate detections were collected from PWSs serving more than 10,000 people.

In a response to the Chamber of Commerce in February of 2013,<sup>8</sup> EPA explained that UCMR 1 allowed source water sampling points in a particular State if that State uses source water monitoring as a more stringent compliance monitoring requirement. EPA did, however, conduct a more detailed assessment of the source water sample detections; a detailed description of this evaluation is presented in Appendix C. Ultimately, EPA determined that, as is consistent with State compliance monitoring programs that enable source water sampling, it was most appropriate to exclude the source water sample detections from the UCMR 1 Perchlorate Dataset

<sup>&</sup>lt;sup>8</sup> [ HYPERLINK "https://www.epa.gov/sites/production/files/2015-06/documents/12004-response.pdf" ]

when those samples had appropriate follow-up entry point samples that were included in the UCMR 1 Perchlorate Dataset. In contrast, any source water sample perchlorate detections for which no follow-up entry point sampling was conducted by PWSs were retained in the UCMR 1 Perchlorate Dataset. Following this convention, the resulting UCMR 1 data set contains 34,132 perchlorate samples from 3,865 systems with a total of 540 detections<sup>9</sup> (1.58% of all samples) from 149 PWSs (3.86% of all PWSs).

#### Follow-up Information Relevant to UCMR 1 Perchlorate Data

The Chamber of Commerce letter also stated that some UCMR 1 data did "not comply with data quality guidelines because it is not representative of current conditions." The Chamber of Commerce provided information on follow-up sampling, as well as additional research into some of the UCMR 1 perchlorate sampling and detections that indicated some of the detections included in the UCMR 1 data set did not fully reflect conditions at some PWSs at the time of the UCMR 1 sampling. Information such as state drinking water annual compliance reports, and studies by Brandhuber *et al.* (2009) and AWWA (2008) were noted by the Chamber and were reviewed and considered by EPA.

In response to the Chamber's request, the EPA evaluated publicly available compliance data from the States of California and Massachusetts. The UCMR 1 dataset was designed to be nationally representative of perchlorate occurrence at the time of sampling (2001 through 2005). EPA acknowledges that conditions may have changed regarding perchlorate occurrence after the time of UCMR 1 sampling. EPA believes that it is important to consider the effect that the State mandated regulations for perchlorate in drinking water has had on the contaminant occurrence in public water systems.

To update the occurrence data for systems sampled during UCMR1 from the States of California and Massachusetts, EPA identified all systems and corresponding entry points which had reported perchlorate detections in UCMR 1. Once the systems and entry points with detections were appropriately identified, EPA then used a combination of available data from Consumer Confidence Reports (CCRs) and perchlorate compliance monitoring data from California ([HYPERLINK "https://sdwis.waterboards.ca.gov/PDWW/"]) and Massachusetts ([HYPERLINK "https://www.mass.gov/service-details/public-water-supplier-document-search"]) to match current compliance monitoring data (where available) to the corresponding water systems and entry points sampled during UCMR1.

Out of the 540 detections resulting from the previous QA/QC step, EPA updated data for 321 detections (320 from California systems & 1 from a Massachusetts system). The convention used by EPA to accomplish the substitution of data was to match entry points with compliance data for active entry points based on most recently reported monitoring compliance data, if more than one data point was reported for an entry point, the assigned value is an average of the annual monitoring results at the entry point. In cases were EPA could not find updated entry point data, then the original data from UCMR1 for such entry point was kept.

<sup>&</sup>lt;sup>9</sup> The 540 detections are the result of the original 637 detections in the UCMR 1 minus 97 source water detections.

In addition, EPA used CCRs information for its data update effort. The process for data substitution using CCR information was accomplished by assigning a default value of 6 ppb to all entry points in a given system if perchlorate was reported in the CCR as detected and within compliance of the California MCL. For other systems, if perchlorate was reported as not detected in the CCR then a default value of 4 ppb (equal to the MRL) was assigned to all entry points in a given system. Appendix C includes copies of the State compliance records utilized by EPA for the data substitution effort.

#### 3.4.3 UCMR 1 Perchlorate Data Subsequent to Phase 1 and 2 QA/QC Review

Subsequent to the two phases of QA/QC review of the UCMR 1 perchlorate data, a total of 34,132 samples from 3,865 systems remained, including a total of 540 detections. This data set, referred to as the "UCMR 1 Perchlorate Dataset (2018 Version)," serves as the basis for all occurrence analyses, graphs, and maps presented in Section 4 of this report. [ REF \_Ref526856259 \h ] summarizes the count of records removed from the two phases of QA/QC review.

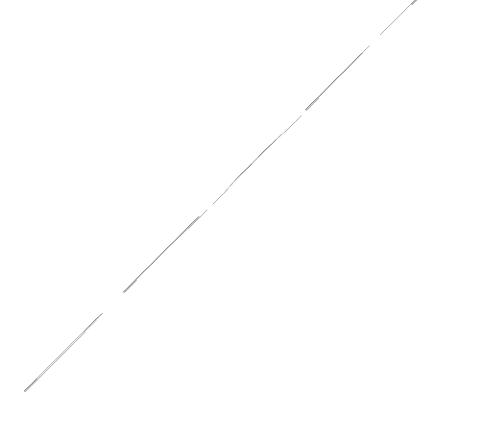
Exhibit [ SEQ Exhibit \\* ARABIC ]: Counts of the Number of Records Removed from the UCMR 1 Perchlorate Data set from Phase 1 and Phase 2 of QA/QC Review

Step		CMR 1 lorate Data
	Included	d Excluded
Original Records	(	34,728
Phase 1: Removal of duplicates, data from non-approximate perchlorate labs, and data from systems no longer corlarge systems. See Section 3.4.1 for details.		397
Phase 2: Removal of source water detections that had entry point samples that were included in the UCMR 1 Dataset. See Section 3.4.2 and Appendix C for details	Perchlorate 34,13	199
Final Records	;	34,132
Percent Included		98.3%

Note that Appendix D presents all 637 perchlorate detections from UCMR 1 sampling included in the data set prior to the Phase 2 QA/QC review (i.e., removal of source water detections with follow-up entry point sample). Specifically, the table in Appendix D presents (for each of the 637 original UCMR 1 individual detections) the state, public water system identification number (PWSID), Facility ID, sample point ID, sample point type (EP = entry point; SR = source water), the system's source water type, system size, the sample date, the result value (in  $\mu g/L$ ), and the source water sample category (if the detection was from source water).

### 3.5 UCMR 1 Analytical Method 314.0

EPA developed Method 314.0, revision 1.0 for the analysis of perchlorate under the UCMR 1. The Minimum Reporting Level (MRL) was 4  $\mu$ g/L and was based on EPA's experience and detailed assessments during method development. The ability to reliably quantify a concentration less than 4  $\mu$ g/L was not evaluated at the time because this MRL was adequate given the general levels of health concern during development of the UCMR 1 in the late 1990s. This analytical method was the only method used to analyze UCMR 1 water samples for perchlorate analysis. All perchlorate laboratory analytical results generated under UCMR 1 have undergone QA/QC review according to the UCMR 1 program. For a detailed discussion of laboratory analytical methods for perchlorate see Section 5.



#### 4 Analysis of the UCMR 1 Perchlorate Dataset

EPA has evaluated the available peer-reviewed science, perchlorate occurrence studies, and data collected by accepted methods on the national occurrence of perchlorate in drinking water. EPA has determined that UCMR 1 is the best available nationally representative data on the frequency and degree (concentration) of perchlorate in drinking water. As previously discussed, the UCMR 1 perchlorate monitoring of drinking water – conducted by a census of large PWSs (serving more than 10,000 people) and a nationally representative statistical sample of small systems (serving 10,000 people or fewer) – represents the most extensive, nationally representative monitoring program for perchlorate in public drinking water systems.

Section 4.1 describes the analytical approach used and provides the resulting tabulated and graphical findings that characterize the national occurrence of perchlorate in drinking water, based on the UCMR 1 data. The perchlorate occurrence findings are presented relative to a range of concentration thresholds to enable a characterization of the frequency and degree of perchlorate occurrence. The results of the analysis are presented in Sections 4.2, 4.3, and 4.4.

### 4.1 Analytical Approach and Resulting Occurrence Estimates

EPA analyzed the updated UCMR 1 perchlorate occurrence data described in Section 3.4.3 above (34,132 samples with 540 detections from 149 systems) to estimate the frequency and degree of perchlorate in public drinking water systems and to estimate the population-served by those systems. To estimate analytical monitoring results, EPA used a non-parametric approach which includes the number and percent of UCMR 1 samples with an analytical detection (a measured sample concentration of perchlorate) that exceeds a concentration threshold of interest. These counts are done at the system level and sample point level to characterize national perchlorate occurrence. Perchlorate occurrence was assessed relative to a range of concentration thresholds including 4 μg/L, 18 μg/L, and 56 μg/L. Additionally, assessments relative to the threshold of greater than or equal to 4 µg/L served to identify all perchlorate sample detections. The results of the analyses are presented in terms of number or percent of PWSs and/or samples with perchlorate concentrations greater than a particular threshold. Additionally, the number and percent of the population served by systems or by sample points with detections is also presented to provide an estimate of the population consuming public drinking water with various frequencies and degrees of perchlorate occurrence.

A limitation of the non-parametric approach is that it can only be used for threshold concentrations that are at or above the UCMR 1 MRL of 4  $\mu g/L$  for perchlorate. There are also no measures of uncertainty or estimates of error. The national estimates of perchlorate occurrence based on the non-parametric analyses of the UCMR 1 data are found in Sections 4.2.1 and 4.2.2 with additional state-level details presented in Appendix E. [ REF \_Ref358311810 ] presents some summary statistics of the UCMR 1 data used for all national occurrence analyses in this section.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Perchlorate Detection Rates and Summary of Detected Concentrations Based on the Updated UCMR 1 Perchlorate Dataset

Source water Type	Total Number of	Samples with	Concentration Value of Detections (µg/L)									
	Samples	Detections	Minimum	Median	Maximum							
Small Systems (serving ≤ 10,000 people)												
Ground Water	2,355	6	4.3	5.3	19.6							
Surface Water	940	9	4.5	5.9	6.8							
All Small Systems	3,295	15	4.3	5.8	19.6							
	Large System	s (serving > 10	,000 people	)								
Ground Water	16,121	166	4	6.1	70							
Surface Water	14,716	359	4	6.7	420							
All Large Systems	30,837	525	4	6.6	420							
All Systems												
All Systems	34,132	540	4	6.5	420							

As a supplement to the non-parametric occurrence analyses presented in this report, EPA also conducted parametric probabilistic modeling of national occurrence of perchlorate using the UCMR 1 Perchlorate Dataset (June 2013 Version). These model analyses estimate national perchlorate occurrence (including national extrapolations of the sample of small systems) provide projections of occurrence below the MRL of 4  $\mu$ g/L, and generate estimates of uncertainty and model error.

# 4.2 System-Level Analyses

EPA analyzed the updated UCMR 1 Perchlorate Dataset to determine system-level occurrence for the number of systems (and population served by those systems) with one or more analytical detections, and the number of systems with two or more analytical detections. EPA also evaluated the number of systems that had at least one detection at two of its sample points. This measure addresses the distribution of perchlorates occurrence throughout a system. For analyses relative to a particular concentration threshold, if a system is identified with two or more detections at a sample point, EPA used the maximum detected concentration in the analysis to estimate potential exposure for the population served by that system.

Perchlorate system-level occurrence analyses are presented in [ REF \_Ref358312523 ] through [ REF \_Ref368308823 \h ]. These analyses characterize the frequency and degree of perchlorate occurrence at concentrations (or thresholds) of 4, 18, and 56  $\mu g/L$ .

[ REF \_Ref358312523 \\* MERGEFORMAT ] presents the number of PWSs and associated populations that had at least one analytical detection of perchlorate relative to the various thresholds. A total of 149 systems serving approximately 16.2 million people were served water from PWSs that had at least one detection of perchlorate. As the detection threshold increases, the number of affected systems and associated populations decrease. At 18 µg/L, the

number of systems reporting perchlorate detections drops to 15 systems serving approximately 700,000 people. Two systems serving approximately 65,000 people had detections greater than  $56 \mu g/L$ .

[ REF \_Ref358312598 ] presents the number of PWSs and associated populations that had two or more analytical detections of perchlorate relative to the various thresholds. This estimate provides an assessment of perchlorate persistent or recurring occurrence. A total of 73 systems serving approximately 12 million people had two or more detections of perchlorate. As the detection threshold increases, the number of affected systems and associated populations decrease. At 18  $\mu$ g/L, the number of systems reporting perchlorate detections drops to 1 system serving approximately 40,000 people.

[ REF \_Ref368308823 ] presents the number of systems and associated populations that had at least one analytical detection at two or more sample points above the various thresholds. This assessment attempts to address the distribution of perchlorate occurrence throughout a system. Note that roughly half of all UCMR 1 systems sampled only at one sample point and, therefore, were not included in this analysis. A total of 52 systems (all large systems) serving about 10.8 million people had at least one detection of perchlorate at two or more sampling points. As the thresholds increase, the number of systems and associated proportional populations decrease.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Systems and Populations with a <u>SINGLE</u> Detection Relative to Various Thresholds Based on the Updated UCMR 1 Perchlorate Dataset

		Total			Concentration Thresholds						
System	Source Water	Number of	Total Pop. Served by Systems in UCMR 1	≥ 4 µg/L		> 18 µg/L		> 56 µg/L			
Туре	Туре	Systems in UCMR 1		Number of Systems	Pop. Served	Number of Systems	Pop. Served	Number of Systems	Pop. Served		
Small	Ground Water	590	1,939,815	5	7,360	1	4,309	0	0		
Systems (serving	Surface Water	207	820,755	3	6,123	0	0	0	0		
≤10,000)	All Systems	797	2,760,570	8	13,483	1 /	4,309	0	0		
Lawra	Ground Water	1,379	53,765,152	62	4,474,125	6	173,480	1	38,761		
Large Systems (serving	Surface Water	1,689	169,087,949	79	11,684,957	8	523,391	1	25,972		
>10,000)	All Systems	3,068	222,853,101	141	16,159,082	14	696,871	2	64,733		
	Ground Water	1,969	55,704,967	67	4,481,485	7	177,789	1	38,761		
All Systems	Surface Water	1,896	169,908,704	82	11,691,080	8	523,391	1	25,972		
	All Systems	3,865	225,613,671	149	16,172,565	15	701,180	2	64,733		

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Systems and Populations with <u>TWO OR MORE</u> Detections Relative to Various Thresholds Based on the updated UCMR 1 Perchlorate Dataset

		Total		Concentration Thresholds						
System	Source Water	Number of	Total Pop. Served by Systems in UCMR 1	≥ 4 µg/L		> 18	> 18 µg/L		56 μg/L	
Type	Type	Systems in UCMR 1		Number of Systems	Pop. Served	Number of Systems	Pop. Served	Number of Systems	Pop. Served	
Small	Ground Water	590	1,939,815	1	56	0	0	0	0	
Systems (serving	Surface Water	207	820,755	3	6,123	0 /	0	0	0	
≤10,000)	All Systems	797	2,760,570	4	6,179	0 /	0	0	0	
Laura	Ground Water	1,379	53,765,152	32	3,105,652	0	0	0	0	
Large Systems (serving	Surface Water	1,689	169,087,949	37	9,044,290	0	0	0	0	
>10,000)	All Systems	3,068	222,853,101	69	12,149,942	0	0	0	0	
	Ground Water	1,969	55,704,967	33	3,105,708	0	0	0	0	
All Systems	Surface Water	1,896	169,908,704	40	9,050,413	0	0	0	0	
	All Systems	3,865	225,613,671	73	12,156,121	0	0	0	0	

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Systems and Populations with At Least One Detection At Two or More Sample Points Based on the updated UCMR 1 Perchlorate Dataset

		Total		Concentration Thresholds						
System	Source Water	Number of	Total Pop. Served by	≥ 4 µg/L		> 18 µg/L		> 5	56 μg/L	
Type	Type	Systems in UCMR 1	Systems in UCMR 1	Number of Systems	Pop. Served	Number of Systems	Pop. Served	Number of Systems	Pop. Served	
Small	Ground Water	590	1,939,815	0	0	0	0	0	0	
Systems (serving	Surface Water	207	820,755	0	0	0 /	0	0	0	
≤10,000)	All Systems	797	2,760,570	0	0	ø	0	0	0	
	Ground Water	1,379	53,765,152	24	2,890,719	1	38,761	0	0	
Large Systems (serving	Surface Water	1,689	169,087,949	28	7,951,355	0	0	0	0	
>10,000)	All Systems	3,068	222,853,101	52	10,842,074	1	38,761	0	0	
	Ground Water	1,969	55,704,967	24	2,890,719	1	38,761	0	0	
All Systems	Surface Water	1,896	169,908,704	28	7,951,355	0	0	0	0	
	All Systems	3,865	225,613,671	52	10,842,074	1	38,761	0	0	

### 4.3 Sample Point-Level Analyses

The system-level analysis in Section 4.2.1, while valuable for illustrative purposes, reflects the conservative assumption that if a perchlorate detection is found in a single entry (or sampling) point in a system, even in a system with multiple entry/sampling points, then the entire population served by the system is potentially served water containing perchlorate. However, there are 2,004 UCMR 1 PWSs that have more than 1 sample point and some get water from multiple water sources (such as a mix of purchased and non-purchased water, ground water and surface water, etc.). In systems with multiple water sources or water intakes, contaminant occurrence in one source or entry point does not necessarily mean occurrence in all sources or entry points that distribute water to consumers. Given the detailed sample point information in the UCMR 1 data, EPA conducted analyses at the sample point level to provide additional details of contaminant occurrence by sample point. These occurrence measures include a sample point "proportional population" occurrence assessment that estimates potential exposure (population served with water containing perchlorate) based on perchlorate occurrence at the sample point (instead of the entire PWS) level. These sample point level analyses were conducted for sample points with at least one perchlorate analytical detection, as well as sample points with two or more perchlorate analytical detections.

#### Sample Point "Proportional Populations"

The sample point proportional population occurrence measure is a less conservative estimate of the population served by a system with a perchlorate detection. The actual population served by the different distribution systems within a PWS with multiple distribution systems is not known; these data are not required for reporting. Therefore, to derive this sample point-level measure, EPA assumed that all distribution systems at PWSs with multiple distribution systems evenly serve the system's population. For example, if a PWS has two distribution systems, serves a population of 30,000, and has an analytical detection of perchlorate in one of its two sample locations, then a population of 15,000 (one half of 30,000) would be estimated to be potentially exposed to perchlorate. How well this assumption reflects actual populations exposed to perchlorate occurrence depends on the distribution system and service population configurations of individual systems.

#### Two Detections at One Sample Point

The "two detections at one sample point" occurrence measure identifies which PWSs have at least two or more analytical detections at any single sample point in the system. By counting individual sample points with at least two separate detections, the analysis provides an indication of persistent or recurring perchlorate occurrence over time at the particular sampling point location within the system. For analyses relative to a particular concentration threshold, if a system is identified with two or more detections at a sample point, EPA used the maximum detected concentration in the analysis to estimate potential exposure for the population served by that system.

These occurrence analyses are also presented relative to perchlorate concentrations thresholds of 4, 18, and 56  $\mu g/L$ . At all thresholds, detection rates were higher in large systems than in small systems.

[ REF \_Ref358346528 ] presents the number of sample points in UCMR 1 that had at least one perchlorate detection. A total of 336 sample points serving about 4.1 million people had at least one perchlorate detection. As the perchlorate concentration thresholds increase, the number of sample points and associated proportional populations decrease, ending at the 18  $\mu$ g/L threshold with 50 sample points, serving about 1.1 million people.

[ REF \_Ref358313783 ] presents the number of sample points in UCMR 1 that had at least two perchlorate detections above the perchlorate concentration thresholds. As described earlier, this analysis provides an indication of persistent, or recurring, perchlorate occurrence over time at the particular sampling location within the system. A total of 126 sample points serving about 1.2 million people had at least two detections of perchlorate). As the thresholds increase, the number of systems and associated proportional populations decrease, ending at the 18 µg/L threshold with 11 sample points, serving about 50,700 people.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Sample Points With at Least One Detection and Their Proportional Populations Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)

				Concentration Thresholds						
System	Source Water	Total UCMR 1	Total UCMR	≥ 4	µg/L	>1	8 μg/L	>56 μg/L		
Туре	Туре	Sample Points	Population	Number of Sample Points	Proportional Population	Number of Sample Points	Proportional Population	Number of Sample Points	Proportional Population	
Om all	Ground Water	1,211	1,939,815	5	3,361	5	3,361	1	2,155	
Small Systems (serving ≤10,000)	Surface Water	243	820,755	3	6,123	3	6,123	0	0	
<u> </u>	All Systems	1,454	2,760,570	8	9,484	8	9,484	1	2,155	
	Ground Water	8,212	53,765,152	123	628,539	116	591,693	34	242,251	
Large Systems (serving >10,000)	Surface Water	5,270	169,087,949	205	3,434,702	198	3,373,780	69	1,272,029	
~10,000)	All Systems	13,482	222,853,101	328	4,063,241	314	3,965,473	103	1,514,280	
	Ground Water	9,423	55,704,967	128	631,900	121	595,054	35	244,406	
All Systems	Surface Water	5,513	169,908,704	208	3,440,825	201	3,379,903	69	1,272,029	
	All Systems	14,936	225,613,671	336	4,072,725	322	3,974,957	104	1,516,435	

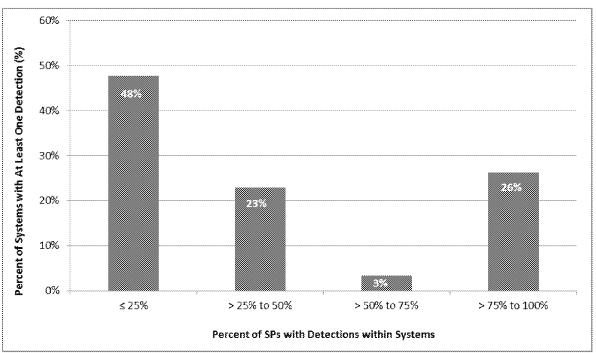
Note: Due to trailing decimal places (not shown in the table) for the population served by ground water and surface water sample points, the counts of total population served may appear to be slightly off.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Sample Points With at Least Two Detections and Their Proportional Populations Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)

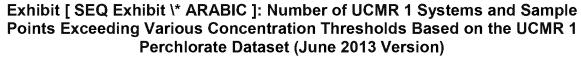
			Total UCMR 1	Concentration Thresholds						
System Type	Source	Total UCMR 1		≥ 4 µg/L		> 18 µg/L		> 56 µg/L		
	Water Type	Sample Points	Population	Number of Sample Points	Proportional Population	Number of Sample Points	Proportional Population	Number of Sample Points	Proportional Population	
	Ground Water	1,211	1,939,815	1	56	1	56	0	0	
Small Systems (serving ≤10,000)	Surface Water	243	820,755	3	6,123	3	6,123	0	0	
310,000)	All Systems	1,454	2,760,570	4	6,179	4	6,179	0	0	
	Ground Water	8,212	53,765,152	31	114,287	31	114,287	4	10,998	
Large Systems (serving >10,000)	Surface Water	5,270	169,087,949	91	1,123,105	84	1,075,939	25	468,004	
~10,000)	All Systems	13,482	222,853,101	122	1,237,392	115	1,190,226	29	479,002	
	Ground Water	9,423	55,704,967	32	114,343	32	114,343	4	10,998	
All Systems	Surface Water	5,513	169,908,704	94	1,129,228	87	1,082,062	25	468,004	
	All Systems	14,936	225,613,671	126	1,243,571	119	1,196,405	29	479,002	

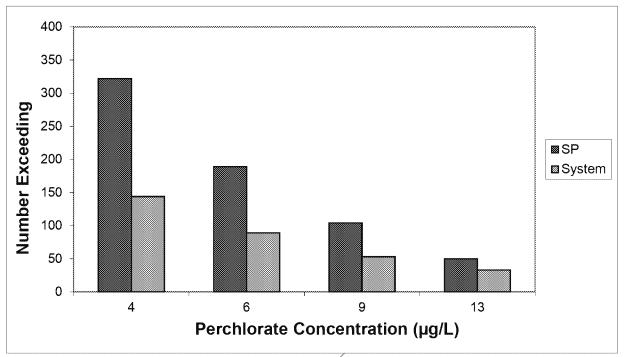
[ REF \_Ref353883366 ] shows the proportion of systems detecting perchlorate in various percentages of their sample points. Of the 149 PWSs with at least one detection of perchlorate, nearly half (48%) of the systems had detections in 25% or fewer of their sample points. A little more than half of the systems (52%) had detections in more than one-quarter of their sample points and 29% had detections in more than half of their sample points. (About 20% of all UCMR 1 systems with detections of perchlorate had only 1 sample point.)

Exhibit [ SEQ Exhibit \\* ARABIC ]: Portion of Systems with Perchlorate Detections in Various Percentages of System Sampling Points (Among Systems with at Least One Detection) Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)



[ REF \_Ref316583205 ] displays the number of PWSs and sample points with at least one detection that exceeded the perchlorate concentration thresholds of 4, 18, and 56  $\mu$ g/L. A total of 322 sample points at 144 systems had at least one detection greater than 4  $\mu$ g/L. Fifty sample points at 33 systems had at least one detection greater than 18  $\mu$ g/L. Note that systems and sample points with exceedances of the higher thresholds are counted multiple times in [ REF \_Ref316583205 ] (i.e., systems counted as exceeding 56  $\mu$ g/L are also counted as exceeding 4, and 18  $\mu$ g/L).

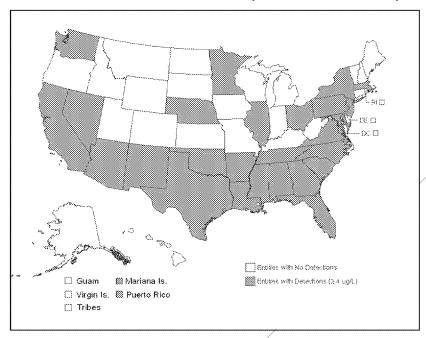




# 4.4 Spatial and Graphical Assessments

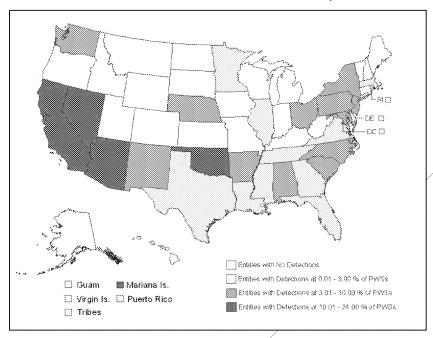
Spatial and graphical assessments of UCMR 1 perchlorate monitoring data are provided in this section. Perchlorate was detected at PWSs in 26 states, Puerto Rico, and the Commonwealth of the Northern Mariana Islands (see [ REF \_Ref316583235 ]). EPA identified perchlorate analytical detections in PWSs across states in the southern portion of the country, several states in the Northeast, Washington, Nebraska, Minnesota, Illinois, and Ohio. States with the highest percentage of systems with perchlorate analytical detections were California, Arizona, Nevada, and Oklahoma. The Northern Mariana Islands also had a high proportion of systems with analytical detections (33%), but since only three systems were sampled, this distinction is based on a relatively small sample size. Breakdowns of sampling efforts by state can be found in Appendix E, including state-level (non-parametric) counts of systems and population served by systems; systems (and population served by systems) with detections; and summaries of detected concentrations.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: Geographic Distribution of Perchlorate – States with At Least One Detection Equal to or Above the MRL (≥ 4 µg/L) Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)



This map depicts UCMR 1 results from both small systems and large systems. (Note: Small and large system data were collected in all states.) As discussed in this report, the statistical selection of UCMR 1 small systems was designed to be representative at the national-level, but not at the state level. Therefore, this map should only be considered an approximation of state-level patterns of contaminant occurrence.

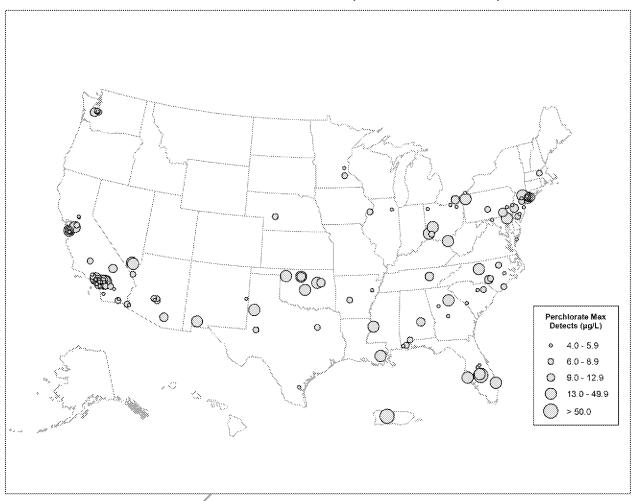
Exhibit [ SEQ Exhibit \\* ARABIC ]: Geographic Distribution of Perchlorate – State Percentage of PWSs with At Least One Detection Equal to or Above the MRL (≥ 4 µg/L) Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)



This map depicts UCMR 1 results from both small systems and large systems. (Note: Small and large system data were collected in all states.) As discussed in this report, the statistical selection of UCMR 1 small systems was designed to be representative at the national-level, but not at the state level. Therefore, this map should only be considered an approximation of state-level patterns of contaminant occurrence.

[REF\_Ref316583297] provides a map of the locations and concentrations of perchlorate detections in PWSs across the US. The map shows widespread detections of perchlorate across large portions of the United States and also indicates several specific areas of higher numbers, or clusters, of PWSs with detections. Southern California has many PWSs with detections as well as some PWSs with higher concentrations of perchlorate. The San Francisco Bay area in California and the mid-Atlantic region from the Washington, D.C. area to the greater metropolitan New York City area also had many systems with detections. Other areas with somewhat higher numbers of PWSs and/or higher concentrations are in Florida, Oklahoma, and Puerto Rico.

# Exhibit [ SEQ Exhibit \\* ARABIC ]: System-level Geographic Distribution of Perchlorate – Maximum Concentration of Detections per System Based on the UCMR 1 Perchlorate Dataset (June 2013 Version)



This map depicts UCMR 1 results from both small systems and large systems. The statistical selection of UCMR 1 small systems was designed to be representative at the national level, but not at the state level. Therefore, this map should only be considered an approximation of state-level patterns of contaminant occurrence.

#### 5 Laboratory Analytical Methods

#### 5.1 EPA Methods

EPA has developed five analytical methods for the identification and quantification of perchlorate in drinking water. These methods - 314.0, 314.1, 314.2, 331.0, and 332.0 - have different characteristics, performance, and costs. A description of these methods is presented below.

EPA Method 314.0, "Determination of Perchlorate in Drinking Water Using Ion Chromatography" (Revision 1.0, USEPA, 1999a) reports a method detection limit (MDL) of 0.53 μg/L. Single-laboratory mean percent recoveries in various aqueous matrices range from 86% to 113% with relative standard deviations (RSDs) of 1.0% to 12.8%. A minimum reporting level (MRL) is not specified in the method; however, a range of 3.0 to 5.0 μg/L is cited as a benchmark range for quality assurance/quality control (QA/QC) procedures. The MRL is typically established as either a concentration that is greater than three times the laboratory MDL or at a concentration that yields a response greater than a signal to noise ratio of five. In either case, the MRL must not be below the lowest instrument calibration standard (USEPA, 1999a). Method 314.0 was widely adopted as the standard perchlorate method.

Method 314.0 has the potential for interferences in the determination of perchlorate. EPA developed options for minimizing interferences to mitigate the potential perchlorate misidentification. EPA has identified three types of potential interferences:

- direct chromatographic coelution an analyte response is observed at very nearly the same retention time (i.e., the "time window" when an analyte emerges from the ion chromatography (IC) column and is "seen" by the detector) as perchlorate;
- concentration dependent coelution observed when the response of higher than typical concentrations of a neighboring peak (i.e., another compound in the sample) overlap into the retention time window of perchlorate; and
- ionic character displacement retention times may significantly shift due to the influence of high ionic strength matrices (high mineral content or hardness) overloading the exchange sites in the ion chromatography column and significantly shortening the retention time of perchlorate.

The possibility of interferences and perchlorate misidentification may become greater if reporting levels (RLs) are reduced from EPA's original MRL of 4  $\mu$ g/L associated with the use of EPA Method 314.0 for perchlorate monitoring under UCMR 1, to 1  $\mu$ g/L. Sensitivity can be improved by either increasing the sample injection volume or incorporating a smaller diameter analytical column. In both cases, however, column capacity will limit the volume of sample that can be injected on-column without degradation of chromatographic resolution. EPA understood the potential for reduced resolution when setting the UCMR 1 perchlorate MRL at 4  $\mu$ g/L rather than at a lower concentration.

Massachusetts Department of Environmental Protection reduced the reporting level to 1 µg/L by using lower concentration spiking solutions and standards for laboratory QC (MassDEP, 2004a) along with a series of initial and ongoing quality control requirements and limits (MassDEP, 2006a). The physical modifications made by MassDEP constitute such a significant modification that the modified method can no longer be considered EPA Method 314.0. An increase in sample injection volume and a smaller-bore chromatographic column would reduce the chromatographic resolution of EPA Method 314.0. As a result, the interferences that are identified in EPA Method 314.0 can be exacerbated for higher ionic strength samples with elevated total dissolved solids (TDS) and are more likely to have retention times that can result in falsely assigning these interferences as detections of perchlorate.

EPA believes that the injection and chromatographic conditions specified in EPA's original publication of EPA Method 314.0, along with mitigative steps that are included to reduce interferences, are critical to proper resolution and identification of perchlorate. Therefore, EPA has confidence in the analytical results and the detections found in UCMR 1 monitoring that resulted from the use of EPA Method 314.0 as originally published.

After EPA published Method 314.0, the agency adopted additional method development goals for the analysis of perchlorate in drinking water including: 1) reducing MRL to less than 1  $\mu$ g/L through the application of sample concentration techniques, microbore analytical columns, and advanced detection systems (i.e., mass spectrometry), 2) further increasing the tolerance for high ionic strength matrices, and 3) enhancing measurement selectivity.

EPA Method 314.1, "Determination of Perchlorate in Drinking Water Using Inline Column Concentration/Matrix Elimination Ion Chromatography with Suppressed Conductivity Detection" (Revision 1.0, USEPA, 2005b) documents EPA single-laboratory lowest concentration minimum reporting levels (LCMRLs) of less than 0.2  $\mu$ g/L (detection limit [DL] = 0.03  $\mu$ g/L) using online sample pre-concentration. The method uses matrix diversion to handle high ionic strength matrices (up to 1,000 mg/L total dissolved solids [TDS]) and added confirmation analysis using a second analytical column (USEPA, 2005b).

EPA Method 314.2, "Determination of Perchlorate in Drinking Water Using Two-Dimensional Ion Chromatography with Suppressed Conductivity Detection" (USEPA, 2008c) documents EPA single-laboratory LCMRLs of less than 0.1  $\mu$ g/L (DLs < 0.02  $\mu$ g/L) using large volume injection. The method uses 2-D chromatography to handle high ionic strength matrices (up to 1,000 mg/L TDS) by isolating perchlorate in the first dimension and measuring it in the second dimension (USEPA, 2008c).

EPA Method 331.0, "Determination of Perchlorate in Drinking Water by Liquid Chromatography Electrospray Ionization Mass Spectrometry" (Revision 1.0, USEPA, 2005c) documents EPA single-laboratory LCMRLs of less than 0.1  $\mu$ g/L (DLs < 0.01  $\mu$ g/L), applied multiple analytical advancements to a liquid chromatography (LC) analysis including a perchlorate selective LC column (IonPak AS-21), mass spectrometry (MS) or tandem mass spectrometry (MS/MS) detection for selectivity and sensitivity, and a custom O-18 isotopically labeled internal standard (Cl<sup>18</sup>O<sub>4</sub>-) (USEPA, 2005c).

EPA Method 332.0, "Determination of Perchlorate in Drinking Water by Ion Chromatography with Suppressed Conductivity and Electrospray Ionization Mass Spectrometry" (USEPA, Revision 1.0, 2005d) documents an EPA single-laboratory LCMRL of 0.1  $\mu$ g/L (DL = 0.02  $\mu$ g/L), applied multiple analytical advancements in an IC analysis including suppressed conductivity IC, MS or MS/MS detection for selectivity and sensitivity, and a custom O-18 isotopically labeled internal standard (Cl<sup>18</sup>O<sub>4</sub>-) (USEPA, 2005d).

In May 2012, the California Department of Public Health's Drinking Water and Radiation Laboratory published recommendations regarding the potential for perchlorate degradation in water samples. According to the guidance/recommendation, EPA Methods 314.1, 314.2, 331.0 and 332.0 require that water samples be filtered in the field and kept cold during shipment to the laboratory. EPA Method 314.0 does not include this sample preservation requirement. Since microbial degradation of perchlorate requires anaerobic conditions, field filtration is not necessary when using EPA Method 314.0 provided aerobic conditions are maintained for the samples until analysis. Aerobic conditions can be effected by half-filling sample bottles in the field, agitating them to dissolve air in the samples and then chilling them on ice for delivery to the laboratory (CA EPA, 2016).

It is anticipated that most drinking water samples will be aerobic in nature and will not become anaerobic prior to analysis within the 28-day holding time specified in EPA Method 314.0. Any air in the headspace of a sample to be analyzed by EPA Method 314.0 will support aeration of the sample due to agitation during shipping. EPA Method 314.0 indicates that perchlorate has been shown to be stable for at least 28 days.

[ REF \_Ref366605474 ] compares the five EPA analytical methods for the analysis of perchlorate in drinking water.

Exhibit [ SEQ Exhibit \\* ARABIC ]: Comparison of EPA Analytical Methods for the Analysis of Perchlorate in Drinking Water

Method	LCMRL or MRL <sup>1</sup> (µg/L)	MDL/DL (µg/L)	Perchlorate Retention Time (minutes)	Demonstrates Acceptable Performance in 1,000 mg/L TDS	Confirmation	Complexity
314.0	4.0 (MRL)	0.53	11	No <sup>10</sup>	Matrix Spike Assessment	Moderate
314.1	0.13 - 0.14	0.03	30-35	Yes	Second Column Analysis	Moderate
314.2	0.038 - 0.06	0.012- 0.018	37	Yes	2-D	High
331.0	0.022 (MS/MS) 0.056 (Selected Ion Monitoring [SIM])	0.005 (MS/MS) 0.008 (SIM)	9	Yes	MS/MS or MS	Moderate

 $<sup>^{10}</sup>$  EPA Method 314.0 was demonstrated to provide acceptable performance for samples up to 600 - 700 TDS using the AG16/AS16 analytical column

Method	LCMRL or MRL <sup>1</sup> (µg/L)	MDL/DL (µg/L)	Perchlorate Retention Time (minutes)	Demonstrates Acceptable Performance in 1,000 mg/L TDS	Confirmation	Complexity
332.0	0.10 (SIM)	0.02	8	Yes	MS	Moderate

<sup>1</sup>Value for EPA Method 314.0 is a "national" MRL for perchlorate monitoring conducted under UCMR 1. Remaining values are for EPA's single-laboratory LCMRLs generated during method development. MRLs have not been established for EPA Methods 314.1, 314.2, 331.0, and 332.0.

#### 5.2 Methods Used by States

EPA Methods 314.0, 314.1, 314.2, 331.0, and 332.0 underwent EPA's analytical method development and validation processes. The validation process includes a protocol for modifications to any existing EPA-approved analytical methods and a protocol for new determinative techniques. Both validation protocols are rigorous and consider many technical aspects of analytical method performance, (USEPA, 1999b; USEPA, 1999c) including:

- Detection limits:
- Instrument calibration;
- Precision and analyte recovery;
- Analyte retention times;
- Contamination in blanks;
- Development of Quality Control Acceptance Criteria;
- Analysis of field samples; and
- Other technical aspects of sample analysis and data reporting.

UCMR 1 required PWSs to conduct assessment monitoring for perchlorate using method 314.0. In 2004, MassDEP began to evaluate laboratory performance for the analysis of perchlorate in drinking water (MassDEP, 2004a; 2004b). MassDEP indicated use of a modified EPA Method 314.0 to achieve a reporting level of 1.0 µg/L (MassDEP wanted reliable laboratory performance to extend below what was to become their drinking water standard of 2.0 µg/L in 2006). The modifications include the use of lower spiking, lower concentration standard solutions, and an initial and on-going quality control protocol (MassDEP, 2004a; 2006a). In 2006, MassDEP published a data set of confirmed perchlorate detections above 1.0 µg/L (MassDEP, 2006b). These data were generated in 2004, 2005 and early 2006. Although the analytical method is not specified in MassDEP (2006b), presumably a modified version of EPA Method 314.0 was used.

By 2007, MassDEP had included EPA Methods 314.1, 331.0 and 332.0 as acceptable methods for the analysis of perchlorate in DW (MassDEP, 2007), indicating that these methods could achieve reliable quantitation at or below 1.0  $\mu$ g/L. The single-laboratory LCMRLs for these three methods are listed in [ REF Ref366605474 ].

Note that EPA Method 314.2 was not published until 2008. The single-laboratory LCMRLs for EPA Method 314.2 are 0.038-0.060  $\mu$ g/L. These LCMRLs are specific to EPA's method development laboratory. In consideration of natural variability in laboratory performance [PAGE \\* MERGEFORMAT]

nationwide, MRLs for EPA Methods 314.1, 314.2, 331.0 and 332.0 would possibly be less than  $1.0~\mu g/L$ . MRLs have not been determined for these methods because these methods have not been selected for use in a national regulatory program. Thus, reliable quantitation in the sub- $\mu g/L$  range should be attainable using EPA Methods 314.1, 314.2, 331.0, and/or 332.0 while reliable quantitation using EPA Method 314.0 is possible only at or above approximately 4  $\mu g/L$  to minimize the potential for false positives.

A review of available state data for perchlorate (i.e., California, Illinois, Massachusetts and Ohio) indicates the predominant use of EPA Method 314.0, although limited use of 314.1 and 331.0 is indicated by the State of Massachusetts. California perchlorate monitoring data indicates the use of methods "314" and "331."

#### 5.3 Laboratory Analysis Cost Estimates

EPA compiled cost estimates for perchlorate analytical methods by asking laboratories to provide a low-to-high price range that they might charge for running perchlorate methods. In April of 2018, three laboratories were contacted. The estimates, summarized in [ REF \_Ref526572158 \h ], vary depending on the quantity of samples submitted for analysis, target reporting levels and other analytical complexities or contractual factors.

Exhibit [ SEQ Exhibit \\* ARABIC ]: Cost Estimates of Laboratory Analytical Methods for Perchlorate

EPA Perchlorate Method	Low Estimate Per Analysis	High Estimate Per Analysis	Notes *
EPA METHOD 314.0	\$55	\$65	Range is based on information from two of the three laboratories contacted.
EPA METHOD 314.1		/-	None of the labs contacted run this method.
EPA METHOD 314.2		/	None of the labs contacted run this method.
EPA METHOD 331.0	\$90 /	\$175	Range is based on information from two of the three laboratories contacted.
EPA METHOD 332.0	\$175	\$175	Estimate is based on information from one of the three laboratories contacted.

<sup>\*</sup> Sample analysis cost may depend on the number of samples submitted and the turnaround time requested.

#### 6 References

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# Proposed Approaches to Inform the Derivation of a Maximum Contaminant Level Goal for Perchlorate in Drinking Water

Volume I

November 2018

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## **Abbreviations and Acronyms**

ADHD Attention deficit/hyperactivity disorder

ANOVA Analysis of variance

ANT Amsterdam Neuropsychological Test

ASD Autism spectrum disorder ATA American Thyroid Association

ATSDR Agency for Toxic Substances and Disease Registry

BBDR Biologically based dose-response

BMI Body mass index

BSID Bayley Scales of Infant Development

bw Body weight

CBCL Child Behavior Checklist CCL Contaminant Candidate List

CI Confidence interval

CMD Coefficient of mental development
EPA U.S. Environmental Protection Agency
FDA U.S. Food and Drug Administration
FiPS-S Finnish Prenatal Study of Schizophrenia

FMC Finnish Maternity Cohort

FSIQ Full-scale IQ

fT3 Free triiodothyronine fT4 Free thyroxine GW Gestational week(s)

hCG Human chorionic gonadotropin HPT Hypothalamic-pituitary-thyroid

HRL Health reference level
IQ Intelligence quotient
IQR Interquartile range
LDL Low-density lipoprotein
MCL Maximum Contaminant Level
MCLG Maximum Contaminant Level Goal
MDI Mental Development Index (of the BSID)

MRI Magnetic resonance imaging MRL Method reporting limit MS Mass spectrometry

NEPSY Developmental Neuropsychological Assessment NHANES National Health and Nutrition Examination Survey

NIS Sodium-iodide symporter

NOAEL No observed adverse effects level

NRC National Research Council

NSAID Non-steroidal anti-inflammatory drug
OHAT Office of Health Assessment and Translation

OR Odds ratio

PBDE Polybrominated diphenyl ether

PBPK/PD Physiologically based pharmacokinetic/pharmacodynamic

PC Partition coefficient
PCB Polychlorinated biphenyl

PDI Psychomotor Development Index (of the BSID)
PDP Pervasive Developmental Problems Scale

PFA Perfluorinated acid

PIQ Performance IQ

PPVT Peabody Picture and Vocabulary Test

RBC Red blood cell RfD Reference dose

ROB Rest-of-body or risk-of-bias

RR Relative risk

RSC Relative source contribution SAB Science Advisory Board SD Standard deviation

SDQ Strengths and Difficulties Questionnaire

SDWA Safe Drinking Water Act

SE Standard error

SRS Social Responsiveness Scale

T3 Triiodothyronine T4 Thyroxine

TBG Thyroxine-binding globulin

Tg Thyroglobulin

TgAb Thyroglobulin antibody

THoP Transient hypothyroxinemia of prematurity

TPO Ab Thyroid peroxidase antibody TRH Thyrotropin-releasing hormone

TSH Thyroid-stimulating hormone (or thyrotropin)

tT3 Total triiodothyronine tT4 Total thyroxine TTR Transthyretin VIQ Verbal IQ

VRM Visual Recognition Memory

WISC Wechsler Intelligence Scale for Children

WRAVMA Wide Range Assessment of Visual Motor Ability

# Acknowledgements

This document was prepared by the Environmental Protection Agency, Office of Ground Water and Drinking Water. Abt Associates Inc., with support from the Partnership in Pediatric and Environmental Health, provided assistance in performing the analyses detailed in this document.

### **Executive Summary**

The EPA is developing approaches to inform the derivation of a Maximum Contaminant Level Goal (MCLG) for perchlorate in accordance with recommendations made by the Agency's Science Advisory Board (SAB). The EPA has developed and peer reviewed both a biologically based doseresponse (BBDR; also known as a physiologically based pharmacokinetic/pharmacodynamic (PBPK/PD)) model that predicts the relationship between perchlorate exposure and thyroid hormone levels in sensitive life stages and the approach to connect the output of the BBDR model to neurodevelopmental outcomes using the epidemiological literature.

The purpose of this document is to present the revisions to the BBDR model and alternative approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects in response to the prior peer reviews. The document does not derive an MCLG; rather, it presents alternative approaches that might be used to inform future decisions including the derivation of an MCLG in accordance with the Safe Drinking Water Act (SDWA).

MCLGs are non-enforceable, health-based goals that the EPA sets for each regulated drinking water contaminant. In accordance with SDWA, they are set at a level at which no known or anticipated adverse human health effect would occur and to allow for an adequate margin of safety. MCLGs consider only public health, and not limits of analytical measurement and treatment technology effectiveness. SDWA requires that the EPA establish the enforceable Maximum Contaminant Level (MCL) as close as feasible to the MCLG taking costs and benefits into consideration.

In 2012, as a part of the National Primary Drinking Water Regulation development process for perchlorate and in accordance with the requirements of SDWA, the EPA sought recommendations from the SAB on approaches to inform the derivation of a perchlorate MCLG.

In 2013, the SAB recommended the following:

- Derive a perchlorate MCLG that addresses sensitive life stages through PBPK/PD modeling;
- "Expand the modeling approach to account for thyroid hormone perturbations and potential adverse neurodevelopmental outcomes from perchlorate exposure;
- Utilize a mode-of-action framework for developing the MCLG that links the steps in the proposed mechanism leading from perchlorate exposure through iodide uptake inhibition to thyroid hormone changes and finally neurodevelopmental impacts; [and]
- Extend the [BBDR] model expeditiously to...provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (SAB, 2013, p. 19).

The SAB stated that this data-driven approach represents a more rigorous way to address differences in biology and exposure between adults and sensitive life stages than is possible with the default approach for deriving an MCLG, and that the EPA should also consider available data on potential adverse health effects (neurodevelopmental outcomes) due to perturbations in thyroid hormone levels regardless of the cause of those perturbations (SAB, 2013).

Based on the SAB's recommendations, the EPA, with contributions from the Food and Drug Administration, developed a BBDR model to predict the effect of perchlorate on the thyroid gland in formula-fed and breastfed infants, and in lactating women. This draft model was integrated with a previously published model that similarly predicted the effects of perchlorate on serum thyroid hormone concentrations in the third-trimester pregnant woman and her fetus. The model was subjected to external peer review in January 2017. The final peer review report titled *External Peer Review for EPA's Draft Biologically Based Dose-Response (BBDR) Model and Draft BBDR Model Report for Perchlorate in Drinking Water* is available through the docket at [ HYPERLINK "https://www.regulations.gov/document?D=EPA-HQ-OW-2016-0439-0006" ].

The EPA considered all of the peer reviewers' recommendations from the January 2017 peer review and focused on those that were anticipated to be most important for increasing the scientific rigor of the model and modeling results. These revisions are summarized in Section [ REF \_Ref482972651 \r \h ] of this report; additional detail is available in Appendix A. Model revisions focused on the following key recommendations:

- Extending the model to early pregnancy;
- Incorporating biological feedback control of hormone production via thyroid-stimulating hormone or thyrotropin (TSH) signaling, such that the model can describe lower levels of iodide nutrition;
- Calibrating the model and evaluating its behavior for upper and lower percentiles of the population, as well as the population median; and
- Conducting an uncertainty analysis for key parameters.

The EPA has developed a two-stage approach linking the revised BBDR model results ("stage 1") with quantitative information on neurodevelopmental outcomes from epidemiological studies ("stage 2"). EPA has also developed an alternative population-based approach that uses the revised BBDR model to evaluate a shift in the population of pregnant women who could be hypothyroxinemic.<sup>1</sup>

The first stage of the two-stage approach is the development of the revised BBDR model that describes thyroidal hormone production in women of childbearing age with low/adequate iodine intake and predicts the relationship between perchlorate exposure and changes in thyroid hormone levels in early pregnancy. The available data for the second stage of the analysis comes from epidemiological studies that evaluated maternal thyroid hormone levels in early pregnancy and neurodevelopmental outcomes (these are not studies evaluating perchlorate exposure). The EPA presents and characterizes all of these studies in Section [ REF Ref488740188 \r \h ] of this report.

Using the output from the revised BBDR model and the quantitative relationships between thyroid hormone levels and neurodevelopmental effects from the published epidemiological studies, the EPA characterized the relationship between perchlorate exposure on free T4 (fT4) levels in pregnant

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Serum free thyroxine (fT4) value at the lower end of the normal range with normal levels of serum TSH, as opposed to hypothyroidism, which consists of low levels of serum fT4 levels and high levels of serum TSH. For the purposes of this report, fT4 and TSH values will be assumed to be measured via the serum unless otherwise stated.

mothers during early gestation and the potential for changes in neurodevelopmental outcomes in their offspring.

The EPA's alternative population-based approach estimates the shift in the population of hypothyroxinemic, pregnant women that would result from perchlorate exposure. The EPA used the BBDR model predictions to estimate the proportion of hypothyroxinemic pregnant mothers in the population, assuming a distribution of fT4 levels with a consistent iodine intake.

These approaches were peer reviewed in January 2018. Based on recommendations from the peer review panel, the following updates to the first draft of this report have been made:

- Reevaluated the derivation of the distribution of fT4 values output by the BBDR model.
- Updated the literature search to include studies between the last literature review and the peer review.
- Reevaluated the exclusion of identified literature from dose-response analysis for the relationship between fT4 and neurodevelopment.
- Evaluated the quality of studies with information on the dose-response relationship between thyroid hormones and neurodevelopment.
- Reanalyzed the dose-response information relating maternal thyroid hormone levels to IQ from the data provided by Dr. Korevaar.

The updates were made to increase the rigor of the approaches presented. The overall impression from the peer review was supportive of the methods presented in the first draft of this document as evidenced by the following quote:

"The panel commends EPA for the substantial amount of work done in creating the new modeling and preparing the report under review. It was highly responsive to the review comments from a year ago. Overall, the panel agreed that the EPA and its collaborators have prepared a highly innovative state-of-the-science set of quantitative tools to evaluate neurodevelopmental effects that could arise from drinking water exposure to perchlorate. While there is always room for improvement of the models, with limited additional work to address the committee's comments [in the peer review report], the current models are fit-for-purpose to determine an MCLG" [ ADDIN EN.CITE <EndNote><Cite><Author>External Peer Reviewers for U.S.

EPA</Author><Year>2018</Year><RecNum>1974</RecNum><Pages>2</Pages><DisplayText>(E PA, 2018, p. 2)</DisplayText><record><rec-number>1974</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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EPA</author></authors><secondary-authors><author>Versar Inc. for U.S.

EPA</author></secondary-authors></contributors><title>External peer review for U.S.

EPA's proposed approaches to inform the derivation of a maximum contaminant level goal for perchlorate in drinking

water</title></title></title></dates></urls></record></Cite></EndNote>

#### 1. Introduction

The EPA is developing approaches to inform the derivation of a Maximum Contaminant Level Goal (MCLG) for perchlorate. The EPA has developed and peer reviewed a biologically based doseresponse (BBDR; also known as a PBPK/PD) model that predicts the relationship between perchlorate exposure and thyroid hormone levels in sensitive life stages. This document presents revisions made to the BBDR model and approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects in response to the prior peer reviews.

#### 1.1 Background on Perchlorate

Perchlorate has both natural and manmade sources. It is formed naturally via atmospheric processes and can be found within mineral deposits in certain geographical areas, particularly those with arid environments [ ADDIN EN.CITE

<EndNote><Cite><Author>Trumpolt</Author><Year>2005</Year><RecNum>114</RecNum><Di splayText>(Trumpolt et al., 2005)</DisplayText><record><rec-number>114</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1465913666">114</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Trumpolt, Clayton W</author><author>Crain,

Michael</author><author>Cullison, Geoffrey D</author><author>Flanagan, Susan

JP</author><author>Lathrop,

Stephen</author></contributors></title>Perchlorate: sources, uses, and occurrences in the environment</title><secondary-title>Remediation: The Journal of Environmental Cleanup Costs, Technologies, & Devironmental Cleanup Costs, Devironmental Cleanup C

89</pages><volume>16</volume><number>1</number><dates><year>2005</year></dates><urls></urls></record></Cite></EndNote>]. In the past, nitrate fertilizers from Chilean deposits that contained perchlorate were widely used in the United States, which resulted in perchlorate contamination of soil and plants (Agency for Toxic Substances and Disease Registry (ATSDR), 2008). An additional source of perchlorate currently in the environment is from the production and usage of rocket fuel, by both the defense and aerospace industries (National Research Council, 2005). Perchlorate is also generated during the manufacture and use of pyrotechnic devices such as fireworks [ADDIN EN.CITE]

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><titles><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></EndNote>]. It is difficult to determine the annual production of perchlorate in these industries, as perchlorate is not included on the EPA's Toxics Release Inventory list of chemicals that must be reported. In 1994, annual production of perchlorate was estimated to be approximately 22 million pounds [ ADDIN EN.CITE <EndNote><Cite><Author>ATSDR</author><Year>2008</a>

playText>(ATSDR, 2008)</br>
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type><contributors><author>ATSDR</author></authors></contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>].

#### 1.1.1 Chemical Identity of Perchlorate

Perchlorate is a negatively charged ion that is composed of one chlorine atom and four oxygen atoms and is released to the environment in the form of a salt (a type of ionic compound) [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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type><contributors><author>NRC</author></contributors><title>Healt

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location>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-

date>National Research Council</modified-date></record></Cite></EndNote>]. The most common perchlorate salt produced in the United States is ammonium perchlorate, with magnesium, potassium, sodium, and lithium perchlorate also generated in substantial amounts [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-

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mobile in the aqueous environment, since the perchlorate salts are very soluble in water and do not adsorb to soil or sediment [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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Press</publisher><label>627612</label><urls><related-

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date>National Research Council</modified-date></record></Cite></EndNote>]. Because of their

low vapor pressure, perchlorate salts do not volatilize into the air from water [ ADDIN EN.CITE <EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><Dis playText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

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#### 1.1.2 Overview of Human Exposure and Health Effects

Humans are primarily exposed to perchlorate from ingestion of contaminated food and drinking water supplies [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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type><contributors><author>NRC</author></authors></contributors><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><pub-

location>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-

date>National Research Council</modified-date></record></Cite></EndNote>]. The EPA's analysis of nationally representative data from the first Unregulated Contaminant Monitoring Rule found that 4.1% of 3,865 public water systems tested for perchlorate had concentrations above the method reporting limit (MRL) of 4 µg/L. [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2008</Year><RecNum>121</RecNum><DisplayText>(U.S. EPA,

2008)</br>
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size="100%">https://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=P1004X7Q.TXT</style></url></relat ed-urls></url>></record></Cite></EndNote>]. The average concentration of perchlorate in samples above the MRL was 9.85  $\mu$ g/L, with a maximum concentration of 420  $\mu$ g/L [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2008</Year><RecNum>121</RecNum><DisplayText>(U.S. EPA, 2008)</DisplayText><record><rec-number>121</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465916938">121</key></foreign-keys><ref-type name="Government Document">46</ref-

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size="100%">https://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=P1004X7Q.TXT</style></url></relat
ed-urls></urls></record></Cite></EndNote>]. In an analysis of foods from the U.S. Food and Drug
Administration's (FDA's) Total Diet Study, approximately three-quarters of the foods tested had at
least one sample with detectable levels of perchlorate [ ADDIN EN.CITE
EN.CITE.DATA ]. Dietary sources with detectable levels of perchlorate include, but are not limited
to, cow's milk, breast milk, and lettuce [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Ingestion
exposures are of concern because perchlorate is easily and almost completely absorbed from the
gastrointestinal tract [ ADDIN EN.CITE
<EndNote><Cite><Author>Srinivasan</Author><Year>2009</Year><RecNum>118</RecNum><D
isplayText>(Srinivasan & Diramp; Viraraghavan, 2009)</br>
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type><contributors><author>>Srinivasan, Asha</author><author>Viraraghavan,
Thiruvenkatachari</author></authors></contributors></title>Perchlorate: Health effects and
technologies for its removal from water resources</title><secondary-title>International Journal of
Environmental Research and Public Health</secondary-title></title></fitles><periodical><full-
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title></periodical><pages>1418-
1442</pages><volume>6</volume><number>4</number><dates><year>2009</year></dates><url>
></urls></record></Cite></EndNote>]. In a 2004-2005 perchlorate dietary exposure study, FDA
estimated the average dietary perchlorate exposure for all persons aged 2 and above at 0.053 µg/kg
bw/day [ ADDIN EN.CITE
<EndNote><Cite><Author>FDA</Author><Year>2015</Year><RecNum>1916</RecNum><Displ
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Administration (FDA),</author></contributors></title>Preliminary estimation of
perchlorate dietary exposure based on FDA 2004/2005 exploratory
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size="100%">https://www.fda.gov/Food/FoodborneIIlnessContaminants/ChemicalContaminants/ucm
077653.htm</style></url></related-urls></urls></record></Cite></EndNote>]. In 2016, FDA
estimated that average perchlorate dietary intake for the total U.S. population had increased from its
previous assessment and ranged between 0.13 and 0.15 μg/kg bw/day, with estimated average intakes
ranging between 0.09 and 0.48 µg/kg bw/day based on age [ ADDIN EN.CITE
<EndNote><Cite><Author>Abt</Author><Year>2016</Year><RecNum>1917</RecNum><Displa
yText>(Abt, Spungen, Pouillot, Gamalo-Siebers, & DisplayText><record><rec-
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E.</author><author>Spungen, J.</author><author>Pouillot, R.</author><author>Gamalo-Siebers,
M.</author><author>Wirtz, M.</author></contributors><titles><title>Update on dietary
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intake of perchlorate and iodine from U.S. Food and Drug Administration's Total Diet Study: 2008–2012</title><secondary-title>Journal of Exposure Science and Environmental Epidemiology</secondary-title></title>-(title>)

title></periodical><dates><year>2016</year></dates><urls></urls></record></Cite></EndNote>]. Inhalation and dermal exposures to perchlorate in the general population are considered to be negligible due to perchlorate's low potential for volatilization and for absorption from the skin [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmcdxc5vxfpkax2vzp0ftv29"

timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-type><contributors><author>NRC</author></author></contributors><title>><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><publication>Washington, DC</publisher>National Academies
Press</publisher><label>627612</label><url></rr>

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-date>National Research Council</modified-date></record></Cite></EndNote>]. However, occupational exposures to perchlorate via inhalation of perchlorate dust or deposition into the mouth may occur [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></authors></contributors><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>].

Blount et al. (2006) measured perchlorate in urine samples collected from a sample of U.S. residents as part of the 2001-2002 National Health and Nutrition Examination Survey (NHANES). Blount et al. (2006) detected perchlorate at concentrations greater than 0.05  $\mu$ g/L in all urine samples tested, with a median concentration of 3.6  $\mu$ g/L (95th percentile at 14  $\mu$ g/L). Women of reproductive age (15-44 years) had a median urinary perchlorate concentration of 2.9  $\mu$ g/L (95th percentile at 13  $\mu$ g/L). The demographic with the highest concentration of urinary perchlorate was children (6-11 years), who had a median urinary perchlorate concentration of 5.2  $\mu$ g/L. Blount et al. (2006) estimated a total daily perchlorate dose for the NHANES participants aged 20 and older and found a median dose of 0.066  $\mu$ g/kg/day and estimated a 95th percentile dose of 0.234  $\mu$ g/kg/day. A study analyzing perchlorate exposure trends using NHANES cohorts from 2005-2014 found that urinary perchlorate concentrations ranged from 2.63 ng/mL to 3.86 ng/mL, with urinary perchlorate decreasing over this time period [ ADDIN EN.CITE

<EndNote><Cite><Author>Corey</Author><Year>2017</Year><RecNum>1918</RecNum><Disp layText>(Corey, Bell, & DisplayText)
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red-type name="Journal Article">17
ref-type><contributors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><aut

G.P.</author><author>Pleus, R.C.</author></authors></contributors><titles><title>Exposure of the U.S. population to nitrate, thiocyanate, perchlorate, and iodine based on NHANES 2005— 2014</title><secondary-title>Bulletin of Environmental Contamination and Toxicology</secondarytitle></titles><periodical><full-title>Bulletin of Environmental Contamination and Toxicology</fulltitle></periodical><pages>83-88</pages><volume>99</volume><dates><year>2017</year></dates><urls></urls></record></Cite ></EndNote>1. The main target organ for perchlorate's toxicity is the thyroid gland [ ADDIN EN.CITE <EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><Dis playText>(ATSDR, 2008)</br/>DisplayText><record><rec-number>115</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government" Document">46</reftype><contributors><authors><author>ATSDR</author></author></contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year></dates><publocation>Atlanta, GA</pub-location><publisher>U.S. Department of Health and Human Services</publisher><urls></urls></record></Cite></EndNote>]. Specifically, perchlorate competes with iodide<sup>2</sup> for transport through the sodium-iodide symporter (NIS) into the thyroid gland, which is a necessary step in the production of thyroid hormones Triiodothyronine (T3) and Thyroxine (T4) ADDIN EN.CITE <EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><Displ ayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</reftype><contributors><author>NRC</author></contributors><title>><title>Healt h implications of perchlorate ingestion</title></title></dates><year>2005</year></dates><publocation>Washington, DC</pub-location><publisher>National Academies Press</publisher><label>627612</label><urls><relatedurls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modifieddate>National Research Council</modified-date></record></Cite></EndNote>]. Therefore, perchlorate may lead to decreases in levels of these hormones [ ADDIN EN.CITE EN.CITE.DATA | 1 by decreasing iodide uptake. Thyroid hormones are essential to the growth and development of fetuses, infants, and young children, as well as to metabolism and energy regulation throughout the lifespan [ ADDIN EN.CITE <EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><Displ ayText>(NRC, 2005)</br>

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For the purposes of this report, "iodine" will be used to refer to dietary intake before entering the body. Once in the body, "iodide" will be used to refer to the ionic form.

Press</publisher><label>627612</label><urls><related-

1-6

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-date>National Research Council</modified-date></record></Cite></EndNote>]. Disturbances in thyroid levels can adversely affect systems throughout the body, including the cardiovascular, pulmonary, and reproductive systems [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008</Year><RecNum>115</RecNum><DisplayText>(ATSDR, 2008)</DisplayText><record><rec-number>115</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>ATSDR</author></author>></contributors><titles><title>To xicological profile for perchlorates</title></title><dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></EndNote>]. Due to their critical role in the development and maintenance of body systems, thyroid hormones are well regulated in order to keep levels within a narrow range.

Reductions in thyroid hormone synthesis due to perchlorate exposures are also expected to lead to increases in thyroid-stimulating hormone or thyrotropin (TSH), a biomarker of thyroid hormone insufficiency as a result of diminished negative feedback [ ADDIN EN.CITE <EndNote><Cite><Author>Mattie</Author><Year>2006</Year><RecNum>120</RecNum><Displ ayText>(Mattie, Strawson, &amp; Zhao, 2006)</DisplayText><record><rec-number>120</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465916467">120</key></foreign-keys><ref-type name="Book Section">5</ref-type><contributors><author>Mattie, David R</author><author>Strawson, Joan</author><author>Zhao, Jay</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></author></author></a></author></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor>

196</pages><dates><year>2006</per></dates><publisher>Springer</publisher><isbn>038731114
9</isbn><urls></record></EndNote>]. Several short-term (2-week exposure duration) studies have reported that exposures to higher doses of perchlorate in the general population were associated with decreases in iodide uptake, but not with changes in levels of thyroid hormones in the blood [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. However, it must be noted that, because the adult euthyroid human thyroid contains several months of T4 stored in the colloid, it is not expected that a 2-week study would result in a change in thyroid status (Dunn & Dunn, 2000, and Brabant et al., 1992, both as cited in Greer et al., 2002).

One small (n = 13), longer-term clinical trial with six-months of perchlorate ingestion at rates of 0.5 or 3 mg of perchlorate ingestion per day did not see any changes in thyroid function [ADDIN EN.CITE

<EndNote><Cite><Author>Braverman</Author><Year>2006</Year><RecNum>225</RecNum>
DisplayText>(L Braverman et al., 2006)
DisplayText><record><rec-number>225</rec-number><25</p>
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timestamp="1468246163">225
/key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Braverman, L</author><author>Pearce,
E.N.</author><author>Braverman, L</author><author>Seeley,
M.</author><author>Beck, B.</author><author>Magnani, B.</author><author>Bleunt,
B.C.</author><author>Firek, A.</author></author></contributors><title>Effects of six

months of daily low-dose perchlorate exposure on thyroid function in healthy volunteers</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>2721-

94</pages><volume>91</volume><number>7</number><dates><year>2006</year></dates><urls></urls></record></Cite></EndNote>]. However, urinary iodide levels among participants in this study were considerably higher than the national average (mean 257.8 µg/total volume before 0.5 mg dose of perchlorate; mean 311.5 µg/total volume before 3 mg dose of perchlorate), which could be protective. This is consistent with evidence that healthy adults may be protected against the thyroid hormone changes associated with perchlorate exposures given adequate dietary iodine intake [ ADDIN EN.CITE

<EndNote><Cite><Author>ATSDR</Author><Year>2008PlayText>(ATSDR, 2008)DisplayText><record><rec-number>115/rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
timestamp="1465914059">115</key></foreign-keys><ref-type name="Government Document">46</ref-</pre>

type><contributors><author>ATSDR</author></author>>/contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year>2008</year></dates><publication>Atlanta, GA</publication><publisher>U.S. Department of Health and Human Services</publisher><urls></record></Cite></EndNote>].

Somewhat contradictory to these clinical exposure studies are several more recent epidemiological studies that have found associations between low levels of urinary perchlorate and TSH increases and T4 decreases [ ADDIN EN.CITE | ADDIN EN.CITE.DATA | ]. Specifically, Blount et al. [ ADDIN ZOTERO | ITEM CSL | CITATION

{"citationID":"t37noWxV","properties": {"formattedCitation":"(2006)", "plainCitation":"(2006)"}, "cit ationItems": [{"id":2261,"uris": ["http://zotero.org/groups/945096/items/UW4TFPNI"], "uri": ["http://zotero.org/groups/945096/items/UW4TFPNI"], "itemData": {"id":2261, "type": "article-journal", "title": "Urinary Perchlorate and Thyroid Hormone Levels in Adolescent and Adult Men and Women Living in the United States", "container-title": "Environmental Health Perspectives", "source": "CrossRef", "URL": "http://www.ncbi.nlm.nih.gov/pmc/articles/PMC1764147/", "DOI": "10.1289/ehp.9466", "ISSN": "0091-

6765","language":"en","author":[{"family":"Blount","given":"Benjamin C."},{"family":"Pirkle","given":"James L."},{"family":"Osterloh","given":"John D."},{"family":"Valentin-Blasini","given":"Liza"},{"family":"Caldwell","given":"Kathleen L."}],"issued":{"date-parts":[["2006"]]},"accessed":{"date-parts":[["2017",4,12]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] found women with low urinary iodide to be particularly susceptible to altered thyroid hormone levels with perchlorate exposures. Additionally, Steinmaus et al. [ ADDIN ZOTERO\_ITEM CSL CITATION

{"citationID":"Df2fZpGA","properties":{"formattedCitation":"(2013)","plainCitation":"(2013)"},"cit ationItems":[{"id":2292,"uris":["http://zotero.org/groups/945096/items/35VPNIKR"],"uri":["http://zotero.org/groups/945096/items/35VPNIKR"],"itemData":{"id":2292,"type":"article-journal","title":"Combined Effects of Perchlorate, Thiocyanate, and Iodine on Thyroid Function in the National Health and Nutrition Examination Survey 2007-8","container-title":"Environmental research","volume":"123","source":"www.ncbi.nlm.nih.gov","abstract":"Perchlorate, thiocyanate, and low iodine intake can all decrease iodide intake into the thyroid gland. This can reduce thyroid

hormone production since iodide is a key component of thyroid hormone. Previous research has suggested that each of these factors

...","URL":"https://www.ncbi.nlm.nih.gov/pmc/articles/PMC3857960/","DOI":"10.1016/j.envres.201 3.01.005","note":"PMID:

23473920","language":"en","author":[{"family":"Steinmaus","given":"Craig"},{"family":"Miller","given":"Mark D."},{"family":"Cushing","given":"Lara"},{"family":"Blount","given":"Benjamin C."},{"family":"Smith","given":"Allan H."}],"issued":{"date-parts":[["2013",5]]},"accessed":{"date-parts":[["2017",5,5]]}},"suppress-author":true}],"schema":"https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] found an inverse relationship between thyroid hormone levels and perchlorate, and determined populations with concomitant exposures to other goitrogens (i.e., smokers) and low urinary iodide had an increased risk of altered thyroid hormone levels with perchlorate exposure. Furthermore, Steinmaus et al. (2016) evaluated a large pregnant population with early pregnancy exposure to perchlorate and found exposure to perchlorate to be associated with decreased fT4 and T4 and increased TSH.

Exposures to perchlorate and resulting changes in thyroid hormone levels may lead to adverse health effects, particularly in sensitive populations such as pregnant women, fetuses, and infants [ ADDIN EN.CITE

<EndNote><Cite><Author>Taylor</Author><Year>2014</Year><RecNum>119</RecNum><DisplayText>(Taylor et al., 2014)</DisplayText><record><rec-number>119</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1465916379">119</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Taylor, P</author><author>Okosieme, Onyebuchi E</author><author>Hales, C</author><author>Chiusano,

Elisabetta</author><author>Maina, Aldo</author><author>Joomun,

Mohamed</author><author>Bestwick, Jonathan P</author><author>Smyth,

Peter</author><author>Paradice, Ruth</author></author></contributors><titles><title>Maternal perchlorate levels in women with borderline thyroid function during pregnancy and the cognitive development of their offspring: data from the controlled antenatal thyroid study</title><secondary-title>The Journal of Clinical Endocrinology & Metabolism</secondary-

title></title></energy & title></energy & title></energy

Metabolism</full-title></periodical><pages>4291-

4298</pages><volume>99</volume><number>11</number><dates><year>2014</year></dates><is bn>0021-972X</isbn><urls></record></Cite></EndNote>]. Thyroid hormones are critical to neurodevelopment of the fetus, and requirements for these hormones increase during pregnancy [ ADDIN EN.CITE

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ExcludeAuth="1"><Author>Taylor</Author><Year>2014</Year><ReeNum>119</ReeNum><Displ ayText>(2014)</DisplayText><record><rec-number>119</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1465916379">119</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Taylor, P</author><author>Okosieme, Onyebuchi E</author><author>Murphy, Rhian</author><author>Hales, C</author><author>Chiusano,

Elisabetta</author><author>Maina, Aldo</author><author>Joomun,

Mohamed</author><author>Bestwick, Jonathan P</author><author>Smyth,

Peter</author><author>Paradice, Ruth</author></author></contributors><titles><title>Maternal perchlorate levels in women with borderline thyroid function during pregnancy and the cognitive development of their offspring: data from the controlled antenatal thyroid study</title><secondary-title>The Journal of Clinical Endocrinology & Metabolism</secondary-

title > < / title s > < periodical > < full-title > The Journal of Clinical Endocrinology & amp;

Metabolism</full-title></periodical><pages>4291-

4298</pages><volume>99</volume><number>11</number><dates><year>2014</year></dates><is bn>0021-972X</isbn><urls></record></Cite></EndNote>] found that mothers in the upper 10<sup>th</sup> percentile of perchlorate exposure during pregnancy had significantly increased odds of having offspring with intelligence quotient (IQ) scores in the lowest 10% of the population.

In addition, breastfed infants are completely dependent upon breast milk to provide iodine. Perchlorate exposures can lead to decreases in thyroid hormone levels in infants indirectly by decreasing iodine levels in breast milk or directly by decreasing hormone synthesis in infants, thereby leading to deficits in intellectual function [ ADDIN EN.CITE

<EndNote><Cite><Author>Leung</Author><Year>2010</Year><RecNum>117</RecNum><DisplayText>(Leung et al., 2010)</DisplayText><rec-number>117</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1465915921">117</key></foreign-keys>< ref-type name="Journal Article">17</ref-type>< contributors>< author> Leung, A. M.</author> < author> Pearce, E

N</author><author>Braverman, L</author></authors></contributors><titles><title>Perchlorate, iodine and the thyroid</title><secondary-title>Best Practice & Practice &

141</pages><volume>24</volume><number>1</number><dates><year>2010</year></dates><isbn >1521-690X</isbn><urls></record></Cite></EndNote>]. Individuals with thyroid conditions or iodine deficiencies are also expected to be at increased risk for adverse effects from perchlorate exposures [ ADDIN EN.CITE

 $<\!EndNote><\!Cite><\!Author>NRC<\!/Author><\!Year><\!2005<\!/Year><\!RecNum>1332<\!/RecNum><\!DisplayText><\!(NRC, 2005)<\!/DisplayText><\!rec-number>1332<\!/rec-number><\!foreign-keys><\!keyapp="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"$ 

app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-

type><contributors><author>NRC</author></contributors><title><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><pub-

location>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-date>National Research Council</modified-date></record></Cite></EndNote>].

#### 1.2 Evaluation of Perchlorate under the Safe Drinking Water Act

The EPA included perchlorate on the first three iterations of the Contaminant Candidate List (CCL), starting in 1998, after data collection and requests for comments. The CCL is a list of drinking water contaminants that are known or anticipated to occur in public water systems and are not currently subject to EPA drinking water regulations. Contaminants listed on the CCL may require future regulation under SDWA. The EPA uses the CCL to identify priority contaminants for regulatory decision making and information collection.

In 2005, at the request of the EPA and other federal agencies, the NRC evaluated the health implications of perchlorate ingestion. The NRC concluded that perchlorate exposure could inhibit the transport of iodide into the thyroid, leading to thyroid hormone deficiency [ADDIN EN.CITE <EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><Displ ayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</reftype><contributors><author>NRC</author></contributors><title>><title>Healt h implications of perchlorate ingestion</title></title></dates><year>2005</year></dates><publocation>Washington, DC</pub-location><publisher>National Academies Press</publisher><label>627612</label><urls><relatedurls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modifieddate>National Research Council</modified-date></record></Cite></EndNote>]. Significant inhibition of iodide uptake results in intra-thyroid iodide deficiency, decreased synthesis of key thyroid hormones (T3 and T4), and increased TSH. The NRC also concluded that a prolonged decrease of thyroid hormone is potentially more likely to have adverse effects in sensitive populations (e.g., people with thyroid disorders, pregnant women, fetuses, and infants). Based on the NRC's recommendations, the EPA adopted a reference dose (RfD) of 0.7 μg/kg/day in 2005 [ ADDIN EN.CITE <EndNote><Cite><Author>U.S. EPA</Author><Year>2005</Year><RecNum>295</RecNum><DisplayText>(U.S. EPA, 2005)</DisplayText><record><rec-number>295</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470935048">295</key></foreignkeys><ref-type name="Government Document">46</reftype><contributors><authors><author>U.S. EPA, </author></authors><secondaryauthors><author>National Center for Environmental Assessment</author></secondaryauthors></contributors><title>Integrated Risk Information System chemical assessment summary: Perchlorate and perchlorate salts</title></title></dates><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>><url>>< face="underline" font="default" size="100%">https://cfpub.epa.gov/ncea/iris/iris documents/documents/subst/1007 summary.pdf</st yle><style face="normal" font="default" size="100%"> </style></url></relatedurls></urls></record></Cite></EndNote>]. This value was based on a no observed effect level of 7 μg/kg/day identified by a study [ ADDIN EN.CITE <EndNote><Cite><Author>Greer</Author><Year>2002</Year><RecNum>204</RecNum><Displa yText>(Greer et al., 2002)</DisplayText><record><rec-number>204</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1467812686">204</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Greer, Monte A</author>Goodman,

Gay</author><author>Pleus, Richard C</author><author>Greer, Susan E</author></authors></contributors><titles><title>Health effects assessment for environmental perchlorate contamination: the dose response for inhibition of thyroidal radioiodine uptake in humans</title><secondary-title>Environmental Health Perspectives</secondary-title></title><periodical><full-title>Environmental Health Perspectives</full-title></periodical><pages>927</pages><volume>110
/volume><number>9</number><dates><year><2002</pre>// Addes><urls>// Cite>// EndNote>] in healthy adults for inhibition of radioactive iodide uptake and the application of an uncertainty factor of 10 for intra-species variability.

In October 2008, the EPA published a preliminary determination not to regulate perchlorate in drinking water using a health reference level (HRL) of 15  $\mu$ g/L, which was derived from the RfD of 0.7  $\mu$ g/kg/day, using a default body weight (70 kg), a default drinking water consumption rate (2 L/day), and a perchlorate-specific relative source contribution (RSC) of 62% for a pregnant woman [ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2008</Year><RecNum>121</RecNum>CDisplayText>(U.S. EPA, 2008)</DisplayText><record><rec-number>121</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465916938">121</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>U.S.

EPA,</author></authors></contributors></titles><tittle>Interim drinking water health advisory for perchlorate</title></title></dates></pe></dates></pub-location>Washington,
D.C.</pub-location><isbn>EPA 822-R-08-025</isbn><urls><related-urls><url><ti>style face="underline" font="default"

size="100%">https://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=P1004X7Q.TXT</style></url></relat ed-urls></urls></relat/cfite></EndNote>]. The RSC is the percentage of the RfD remaining for drinking water after the other sources of exposure to perchlorate (e.g., food) have been considered. In December 2008, the EPA issued an interim health advisory (15 μg/L perchlorate in drinking water) to provide guidance to state and local officials in their efforts to address this issue. In August 2009, the EPA published a supplemental request for comment with a new analysis that derived potential alternative HRLs for 14 life stages, including infants and children. The analysis used the RfD of 0.7 μg/kg/day and life stage-specific body weight and exposure information (i.e., drinking water intake, RSC) [ ADDIN EN.CITE <EndNote><Cite><Author>U.S.

EPA</Author><Year>2008</Year><RecNum>121</RecNum><DisplayText>(U.S. EPA, 2008)</DisplayText><record><rec-number>121</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465916938">121</key></foreign-keys><ref-type name="Government Document">46</ref-

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EPA,</author></authors></contributors><titles><title>Interim drinking water health advisory for perchlorate</title></title></title></dates><year><2008</year></dates><pub-location>Washington, D.C.</pub-location><isbn>EPA 822-R-08-025</isbn><urls><related-urls><url>styleface="underline" font="default"

size="100%">https://nepis.epa.gov/Exe/ZyPURL.egi?Dockey=P1004X7Q.TXT</style></url></relat ed-urls></urls></record></Cite></EndNote>]. The HRLs ranged from 1 µg/L to 47 µg/L.

After considering comments on the October 2008 and August 2009 notices, the EPA made a final determination to regulate perchlorate in drinking water in February 2011 [ ADDIN EN.CITE

<EndNote><Cite><Author>U.S.

EPA</Author><Year>2011</Year><RecNum>69</RecNum><DisplayText>(U.S. EPA, 2011)</DisplayText><record><rec-number>69</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1438181138">69</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>U.S.

EPA,</author></authors></contributors><titles><title>Drinking water: regulatory determination on perchlorate. Federal Register Notice. 76 FR No. 29. Pages 7762-7767. (February 11, 2011) (to be codified at 40 C.F.R pt. 141).</title></title></tole>
volume>2015
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year>2011
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size="100%">https://www.federalregister.gov/articles/2011/02/11/2011-2603/drinking-water-regulatory-determination-on-perchlorate
/style></url>
/url></related-urls>
/record></Cite></EndNote>]. As a result of the determination, the EPA initiated the process to develop an MCLG and National Primary Drinking Water Regulation for perchlorate under SDWA.

In accordance with SDWA, the Agency requested the EPA's Science Advisory Board (SAB) to provide recommendations for how to consider available data in deriving an MCLG for use in developing a National Primary Drinking Water Regulation for perchlorate. The EPA presented the SAB with a description of the general approach used by the Agency to derive an MCLG for non-carcinogenic chemicals. This approach relies on the following equation:

$$MCLG\left(\frac{\mu g}{L}\right) = \frac{RfD \times BW}{DWI} \times RSC$$

Where:

RfD = RfD of the contaminant ( $\mu g/kg/day$ ); in the initial assessment for perchlorate, the EPA used NRC's recommended RfD of 0.7  $\mu g/kg/day$ 

BW = body weight (kg)

DWI = drinking water ingestion rate (L/day)

RSC = relative source contribution

The EPA also presented information to the SAB about life stage considerations, PBPK modeling, and epidemiologic and biomonitoring studies. The SAB released its final report on May 29, 2013, and recommended that the EPA "derive a perchlorate MCLG that addresses sensitive life stages through physiologically based pharmacokinetic/pharmacodynamic (PBPK/PD) modeling based upon its mode of action rather than the default MCLG approach using the RfD and specific chemical exposure parameters" (SAB, 2013, p. 2). The SAB stated that this approach should follow perchlorate exposure through NIS inhibition to thyroid hormone change and finally to neurodevelopmental effects [ADDIN EN.CITE

<EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(SAB, 2013)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government"

Document">46</ref-type><contributors><author>SAB,</author></author>><author>>SAB,</author></author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><a

In addition, the SAB stated that the EPA should more directly consider thyroid hormone changes as relevant to sensitive life stages; specifically, the fetus of hypothyroxinemic pregnant women and infants and neonates exposed to perchlorate through either water-based formula preparations or the breast milk of lactating women [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(SAB, 2013)</br>

Text><record><rec-number><50</rec-number><foreign-keys><key</td> app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>>SAB,</author></author>><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributors><author>>contributo authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></date>><year></date>><publocation>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>]. This is because thyroid hormone deficiency is known to produce adverse effects on human neurodevelopment, an effect to which the above-mentioned sensitive life stages are especially vulnerable. This direction is different from the conclusions of the NRC report, which based the RfD on the non-adverse effect of reduced iodide uptake and suggested examining pregnant women with hypothyroidism or iodide deficiency [ ADDIN EN.CITE <EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><Displ ayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</reftype><contributors><author>NRC</author></authors></contributors><title>Healt h implications of perchlorate ingestion</title></title></dates><year>2005</year></dates><publocation>Washington, DC</pub-location><publisher>National Academies Press</publisher><label>627612</label><urls><relatedurls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modifieddate>National Research Council</modified-date></record></Cite></EndNote>]. The SAB approach focuses on the subtler changes in thyroid hormones (specifically fT4) associated with maternal hypothyroxinemia rather than broader changes in thyroid hormones (both fT4 and TSH) associated with hypothyroidism[]. Furthermore, the SAB recommended that the EPA consider available data on potential adverse health effects (i.e., neurodevelopmental outcomes) due to thyroid hormone level perturbations, regardless of the cause of those perturbations [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(SAB, 2013)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB,</author></author></author>

authors><author>U.S. Environmental Protection Agency,</author></secondary-authors></contributors><titles><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><year>2013</year></dates><publication>Washington, DC</publication><urls></urls></record></Cite></EndNote>].

To address these recommendations, the EPA revised the existing biologically based dose-response (BBDR) and PBPK/PD models [ ADDIN EN.CITE

<EndNote><Cite><Author>Lumen</Author><Year>2013</Year><RecNum>107</RecNum><Displ ayText>(Lumen, Mattie, & DisplayText><record><recnumber>107</rec-number><foreign-keys><key app="EN" db-"

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1450367396">107</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Lumen, A.</author><author>Mattie, D.R.</author><author>Fisher,

J.W.</author></authors></contributors><title>><title>Evaluation of perturbations in scrum thyroid hormones during human pregnancy due to dietary iodide and perchlorate exposure using a biologically based dose-response model</title><secondary-title>Toxicological Sciences</secondary-title></title></periodical><pages>320-341</pages><volume>133
133</pr>
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Author>U.S.

EPA</Author><Year>2009</Year><RecNum>205</RecNum><record><rec-number>205</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1467814701">205</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>U.S.

EPA,</author></authors></contributors><titles><title>Inhibition of the sodium-iodide symporter by perchlorate: an evaluation of lifestage sensitivity using physiologically based pharmacokinetic (PBPK) modeling (Final Report)</title></title></dates><year>2009</year></dates><publication>Washington, D.C. </publication><isbn>EPA/600/R-

08/106A</isbn><urls></record></Cite></EndNote>] to create a BBDR model that predicts changes in thyroid hormone (i.e., T4, fT4, and T3) levels as a result of nutritional iodine intake and perchlorate exposure in women prior to pregnancy and early gestation. The EPA originally developed a set of BBDR models that included all sensitive life stages identified by the SAB (i.e., fetuses (by modeling a pregnant mother at 40 gestational weeks), neonates, and infants (SAB, 2013)) with the pregnancy model representing the third trimester. These models were peer reviewed in January 2017.³ Reviewers stressed the importance of developing an early pregnancy model when considering adverse neurodevelopmental impacts. This was confirmed by the results of the EPA's literature review (summarized in Section [ REF \_Ref488161860 \n \h ]), which found limited to no data regarding altered thyroid hormone levels and subsequent neurodevelopment effects on life stages other than early pregnancy. Therefore, the EPA responded to this peer review by developing an early pregnancy

For a more detailed discussion on the development of the draft model, its calibration, and dose-response evaluations, please see the draft BBDR model report titled *Biologically Based Dose Response Models for the Effect of Perchlorate on Thyroid Hormones in the Infant, Breast Feeding Mother, Pregnant Mother, and Fetus: Model Development, Revision, and Preliminary Dose-Response Analysis.* The report is available through the docket at [ HYPERLINK "http://www.regulations.gov" ] (Docket ID No. EPA-HQ-OW-2016-0439-0006).

model and updated key parameters for that model. The models for the other sensitive life stages were not revised given the importance of early pregnancy thyroid hormones to offspring neurodevelopment [ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Consequently, this report concentrates on fetuses of hypothyroxinemic women early in pregnancy as the sensitive life stage. The EPA carried out a subsequent peer review in January 2018 to evaluate updates to the BBDR model and presented the approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects. The January 2018 peer review was largely supportive of the efforts described in the first draft of this document as evidenced by the following from the peer review final report:

The panel commends EPA for the substantial amount of work done in creating the new modeling and preparing the report under review. It was highly responsive to the review comments from a year ago. Overall, the panel agreed that the EPA and its collaborators have prepared a highly innovative state-ofthe-science set of quantitative tools to evaluate neurodevelopmental effects that could arise from drinking water exposure to perchlorate. While there is always room for improvement of the models, with limited additional work to address the committee's comments [in the peer review report], the current models are fit-for-purpose to determine an MCLG [ ADDIN EN.CITE <EndNote><Cite><Author>External Peer Reviewers for U.S. EPA</Author><Year>2018</Year><RecNum>1974</RecNum><Pages>2</Pages><DisplayText>(E PA, 2018, p. 2)</br>

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#### 1.3 Objective of the Report

The purpose of this document is to present revisions made to the BBDR model and approaches that link the revised perchlorate BBDR model predictions to neurodevelopmental effects in response to the prior peer reviews. The document does not derive an MCLG; rather, it presents alternative approaches that might be used to inform future decisions including the derivation of an MCLG in accordance with SDWA (see [ REF Ref417634154 \h ]).

# Figure [ SEQ Figure \\* ARABIC ]. Summary of Modeling Approach for Estimating Measurable Adverse Neurodevelopmental Impacts in Pregnant Women Exposed to Perchlorate

[EMBED Visio.Drawing.15]

Section [ REF \_Ref516911407 \r \h ] of the report describes how perchlorate can disrupt thyroid hormone homeostasis along with the common causes and consequences of such disruptions, including neurodevelopmental alterations following the proposed mode of action put forward by the NRC and modified by the SAB. Section [ REF \_Ref482972651 \r \h ] of the report provides an overview of the EPA's BBDR model, and Section [ REF \_Ref482274870 \r \h ] of the report describes how the BBDR model output relates to a distribution of thyroid hormone levels. The remainder of the report is focused on connecting the predicted changes in thyroid hormone levels from the BBDR model to downstream impacts, indicated as Step 2 in [ REF Ref417634154 \h \\* MERGEFORMAT ].

Section [ REF \_Ref488161860 \n \h ] provides details on the literature search and review process that led to the identification of studies to link the BBDR model outputs to quantitative alterations in neurodevelopmental endpoints. Section [ REF \_Ref456208917 \n \h \\* MERGEFORMAT ] outlines an approach to utilize the identified studies to link the revised perchlorate BBDR model predictions to neurodevelopmental effects. Section [ REF \_Ref481160968 \r \h ] includes a discussion on an alternative approach based on hypothyroxinemia. Section [ REF \_Ref488662930 \n \h ] provides a short description of other potential adverse health impacts besides neurodevelopment that may be related to perchlorate exposure. Lastly, Section [ REF \_Ref456170011 \r \h ] presents the main conclusions for this report.

## 2. Background on Thyroid Hormone Physiology and Connection to Neurodevelopment

In order to understand the approaches the EPA is considering to inform future decisions including the derivation of an MCLG in accordance with SDWA, it is first necessary to have a basic understanding of thyroid physiology (Section [ REF \_Ref427059872 \r \h ]), including during pregnancy (Section [ REF \_Ref456164565 \r \h ]), and pathophysiology (Section [ REF \_Ref486250699 \r \h ]). It is also necessary to understand the mode of action by which perchlorate impacts thyroid hormone levels as put forward by the NRC [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-

type > < contributors > < author > NRC < / author > < / contributors > < title > Healt

h implications of perchlorate ingestion</title></title></dates><year>2005</year></dates><pub-

location>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-

date>National Research Council</modified-date></record></Cite></EndNote>] and modified by the SAB [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(2013)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government Document">46</ref-

type><contributors><author>SAB,</author></author></author></author></author></author>U.S. Environmental Protection Agency,</author></secondary-

authors></contributors><titles><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><quer>2013</quer></dates><publication>Washington, DC</publication><urls></urls></record></Cite></EndNote>] (Section [ REF \_Ref427059909 \r \h ]), how perchlorate is distributed and eliminated from the body (Section [ REF \_Ref459122801 \n \h ]), and how thyroid hormones impact neurodevelopment (Section [ REF \_Ref427059920 \r \h ]). Having this basic understanding of the mode of action through which perchlorate could impact thyroid hormone levels, and how thyroid hormones could in turn impact neurodevelopment, will provide a basis for understanding the approaches presented in this document.

#### 2.1 Overview of Thyroid Physiology

The thyroid gland is part of a "self-regulating" loop referred to as the hypothalamic-pituitary-thyroid (HPT) axis ([ REF \_Ref454999548 \h ]), which controls metabolism, growth, and brain development and function [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller</Author><Year>2007</Year><RecNum>186</RecNum><Prefix>for review see </Prefix><DisplayText>(for review see Zoeller, Tan, & DisplayText),

2007)</DisplayTcxt><record><rec-number>186</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202457">186</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Zoeller, R

Thomas</author><author>Tan, Shirlee W</author><author>Tyl, Rochelle

W</author></authors></contributors></title>General background on the hypothalamic-

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pituitary-thyroid (HPT) axis</title><secondary-title>Critical Reviews in Toxicology</secondary-title></title></full-title>Critical reviews in toxicology</full-title></periodical><pages>11-53</pages><volume>37</volume><number>1-2</number><dates><year>2007</year></dates><isbn>1040-

8444</isbn><urls></record></Cite></EndNote>]. Low circulating levels of thyroid hormones stimulate hypothalamic secretion of thyrotropin-releasing hormone (TRH), which diffuses into the anterior pituitary gland and stimulates the endocrine cells within the anterior pituitary (thyrotropes) to release TSH into the circulatory system. TSH acts on the thyroid gland to stimulate iodine uptake to result in thyroid hormone production. Iodide is taken up from circulation by the NIS; serum iodide is transported into thyroid gland follicles for synthesis into the thyroid hormones T4 and to a lesser extent T3. Endocrine disruptors, such as perchlorate, could interfere with this process by directly inhibiting the uptake of iodide, thereby reducing the production of thyroid hormone.

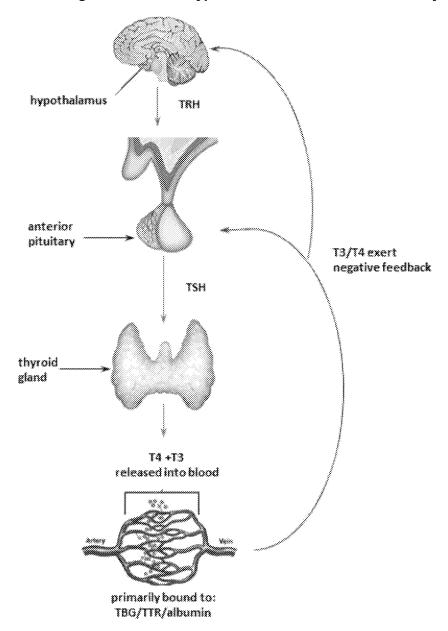
T3 and T4 are formed when thyroglobulin (Tg; the precursor protein of T3 and T4) produced by the thyroid follicular cells is iodinated by the enzyme thyroperoxidase, and the iodinated protein is cleaved. T4 is generally thought to act as the prohormone for the more biologically active T3, which is required for normal development of the central nervous system in fetuses and in infants, as well as for ultimate skeletal development and growth. However, it should be noted that the regulatory factors that modulate the delivery of T4 to the fetus are not fully understood. Both T3 and T4 are critical determinants of metabolic function in humans of all ages. T4 and T3 are released into circulation, where they are primarily bound to the carrier proteins thyroxine-binding globulin (TBG), transthyretin (TTR), albumin, and lipoproteins [ ADDIN EN.CITE ADDIN EN.CITE.DATA ].

T3 and T4 in the blood that are bound to these proteins act largely as reserves because they are less readily metabolized in the liver and cannot enter cells unless in free form [ ADDIN EN.CITE <EndNote><Cite><Author>Zoeller</Author><Year>2007</Year><RecNum>186</RecNum><Displ ayText>(Zoeller et al., 2007)</DisplayText><record><rec-number>186</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202457">186</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Zoeller, R Thomas</author><author>Tan, Shirlee W</author><author>Tyl, Rochelle W</author></authors></contributors><title>Seneral background on the hypothalamic-pituitary-thyroid (HPT) axis</title><secondary-title>Critical Reviews in Toxicology</secondary-title></title><periodical><full-title>Critical reviews in toxicology</full-title></periodical><pages>11-53</pages><volume>37</volume><number>1-2</number><dates><year>2007</year></dates><isbn>1040-8444</isbn><urls></urls></record></Cite></EndNote>]. However, in their free (or unbound) states (the free or unbound states of T4 and T3 are denoted as fT4 and fT3), both are available to be transported actively into specific cells. T4 is transported through the cellular membrane by specific transporters and transformed by deiodinases into T3, which is then transferred to the nucleus to cause transcriptional changes. Circulating T3 is mostly derived from peripheral monodeionization of T4 [

<EndNote><Cite><Author>Chanoine</Author><Year>1993</Year><RecNum>343</RecNum><Di splayText>(Chanoine, Braverman, & Di splayText>(Chanoine, Braverman, & Di splayText><record><rec-number>343</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495028650">343</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Chanoine, J.P. </author><author>Farwell, A.P.

</author></authors></contributors></title>The thyroid gland is a major source of circulating T3 in the rat</title><secondary-title>Journal of Clinical Investigation</secondarytitle></title></title></ent/title>Journal of Clinical Investigation</fulltitle></periodical><pages>2709-2713</pages><volume>91</volume><number>6</number><dates><year>1993</year></dates><url s></urls></record></Cite><Cite><Author>Peeters</Author><Year>2017</Year><RecNum>344</R ecNum><record><rec-number>344</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495029973">344</key></foreignkeys><ref-type name="Electronic Book">44</ref-type><contributors><authors><author>Peeters, R.</author><author>Visser, T.</author></authors><secondary-authors><author>De Groot, LJ</author><author>Chrousos, G</author><author>Dungan, K</author><author>et al.</author></secondary-authors></contributors></title>Metabolism of thyroid hormone</title></title></title>><dates><year>2017</year></dates><publisher>Endotext (Internet)</publisher><urls><related-urls><url><style face="underline" font="default" size="100%">https://www.ncbi.nlm.nih.gov/books/NBK285545/</style></url></relatedurls></urls></record></Cite></EndNote>]. Circulating T3 and T4 levels in an individual are maintained within a narrow range by a negative feedback loop with TSH from the pituitary and TRH from the hypothalamus (see [ REF Ref454999548 \h ]) that operates around a "set point." This set point is different from individual to individual, which generates a population variance in blood levels of thyroid hormone that is considerably broader than the individual variance [ ADDIN EN.CITE <EndNote><Cite><Author>Andersen</Author><Year>2002</Year><RecNum>333</RecNum><Di splayText>(S. Andersen, Pedersen, Bruun, & DisplayText><record><recnumber>333</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1494274249">333</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Andersen, S.</author><author>Pedersen, K.M.</author><author>Bruun, N.H.</author><author>Laurberg, P.</author></authors></contributors></title>Narrow individual variations in serum T(4) and T(3) in normal subjects: a clue to the understanding of subclinical thyroid disease</title><secondarytitle>Journal of Clinical Endocrinology and Metabolism</secondary-title></title></fitles><periodical><fulltitle>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1068-1072</pages><volume>87</volume><number>3</number><dates><year>2002</year></dates><url s></urls></record></Cite></EndNote>]. Therefore, in euthyroid individuals, serum T4 and T3 fluctuate within a fairly narrow range (about 10% of the population variance), maintained by the negative feedback relationship with serum TSH from the pituitary gland. This normal variation creates a situation where single measures of free or total T4 and TSH are a somewhat imprecise measure of an individual's average T4 and TSH concentrations [ ADDIN EN.CITE <EndNote><Cite><Author>Andersen</Author><Year>2002</Year><RecNum>333</RecNum><Di splayText>(S. Andersen et al., 2002)</br/>/DisplayText><record><rec-number>333</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1494274249">333</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Andersen, S.</author><author>Pedersen, K.M.</author><author>Bruun, N.H.</author><author>Laurberg, P.</author></authors></contributors></title>Narrow individual variations in serum T(4) and T(3) in normal subjects: a clue to the understanding of subclinical thyroid disease</title><secondarytitle>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><fulltitle>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>10681072 < pages > < volume > 87 < / volume > 3 < / number > 3 < / volume > < dates > < year > 2002 < / year > < / dates > < url s > < / crecord > < / Cite > < / EndNote > ].

### Figure [ SEQ Figure \\* ARABIC ]. Summary of HPT Axis Showing Negative Feedback of Increased Circulating T4 and T3 on Hypothalamus and Anterior Pituitary



It is generally thought that changes in thyroid hormone levels that occur during pregnancy operate independently of changes in the set point. For example, as the first trimester progresses, increases in

human chorionic gonadotropin (hCG),<sup>4</sup> estrogen, and thyroxine-binding protein result in a higher blood concentration of total T4, making more fT4 available to the developing fetus [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de

Escobar</br>
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1570</pages><volume>10</volume><number>12A</number><dates><year>2007</year></dates><isbn>1475-2727</isbn><urls></record></Cite></EndNote>]. Serum TSH concentrations are also mildly suppressed due to the negative feedback of this elevated serum T4 on TSH.

#### 2.2 Thyroid Physiology in Pregnancy

In general, pregnancy requires the maternal thyroid gland to increase thyroid hormone production by around 50% [ ADDIN EN.CITE

<EndNote><Cite><Author>Skeaff</Author><Year>2011</Year><RecNum>180</RecNum><Prefix >for review see </Prefix><DisplayText>(for review see Skeaff, 2011)</DisplayText><record><recnumber>180</rec-number><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466200929">180</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>>Skeaff, Sheila A</author></author>></contributors><title>Iodine deficiency in pregnancy: the effect on neurodevelopment in the child</title><secondary-title>Nutrients</secondary-

title></title></periodical><full-title>Nutrients</full-title></periodical><pages>265-

273</pages><volume>3</volume><number>2</number><dates><year>2011</year></dates><urls></urls></record></Cite></EndNote>] in order to provide sufficient fT4 for fetal brain development. In the first trimester of pregnancy, this increased output of thyroid hormones is, in part, due to estrogen increases during the first trimester, which increases the serum concentration of TBG, one of the main proteins to which T4 binds. The combination of increased TBG and hCG results in an increase in total T4 of about 50% over the pre-pregnancy level [ ADDIN EN.CITE

<EndNote><Cite><Author>Alexander</Author><Year>2017</Year><RecNum>1895</RecNum><

levels/</style></url></related-urls></urls></record></Cite></EndNote>].

hCG is a hormone produced during pregnancy by cells in the placenta [ADDIN EN.CITE <EndNote><Cite><Author>American Pregnancy
Association</Author><Year>2016</Year><RecNum>206</RecNum><DisplayText>(American Pregnancy Association, 2016)</DisplayText><record><rec-number>206</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1467819649">206</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><author>American Pregnancy Association,</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></a></arthor></author></a></author></a></arthor></a></arthor></author></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></arthor></a></artho

DisplayText>(Alexander et al., 2017)</DisplayText><record><rec-number>1895</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497970921">1895</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Alexander, E. K.</author><author>Pearce, E. N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan, S.</author></authors></contributors><title>>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondary-title>Thyroid</secondary-title></title></title><periodical><fulltitle>Thyroid</full-title></periodical><pages>315-389</pages><volume>27</volume><number>3</number><dates><year>2017</year></dates><urls ></urls></record></Cite></EndNote>]. This increase in serum total T4 also increases serum fT4 during the first trimester, although fT4 levels fall over the next two trimesters [ ADDIN EN.CITE <EndNote><Cite><Author>Alexander</Author><Year>2017</Year><RecNum>1895</RecNum>< DisplayText>(Alexander et al., 2017)</DisplayText><record><rec-number>1895</recnumber><foreign-keys><key app="EN" db-id="z9t0ayxyzdfermedxe5yxfpkax2yzp0ftv29" timestamp="1497970921">1895</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Alexander, E. K.</author><author>Pearce, E. N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen, H.</author><author>Dosiou, C., </author><author>Sullivan, S.</author></authors></contributors><title>>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondary-title>Thyroid</secondary-title></title><periodical><fulltitle>Thyroid</full-title></periodical><pages>315-389</pages><volume>27</volume><number>3</number><dates><year>2017</year></dates><urls ></urls></record></Cite></EndNote>]. The first-trimester increase in fT4 can lead in turn to a lower TSH concentration compared to the pre-pregnancy state.

Given the increased demand on the maternal thyroid in pregnancy, there is an increased demand for iodine to support additional hormone synthesis. If the maternal thyroid is fully functional, then sufficient dietary iodine intake will enable this to occur. According to the SAB, maternal hypothyroxinemia has been defined by a "variety of cutoffs...ranging from fT4 below the 10<sup>th</sup> or 5<sup>th</sup> percentiles to below the 2.5th percentile" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Pages>1 0</Pages><DisplayText>(SAB, 2013, p. 10)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>>SAB,</author></author>><secondaryauthors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></date>><year>2013</year></dates><publocation>Washington, DC</pub-location><urls></urls></record></Cite></EndNote>] in the population. Although a fully functional system (i.e., in euthyroid individuals) can compensate with additional iodine consumption, hypothyroxinemic individuals are already compromised, and iodine insufficiency can exacerbate this situation, putting the fetus at risk of neurological deficits [ ADDIN ADDIN EN.CITE.DATA | Maternal fT4 is particularly important in the first trimester of pregnancy because the fetal brain is solely dependent upon T4 of maternal origin during this period

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<EndNote><Cite><Author>Zoeller</Author><Year>2004</Year><RecNum>194</RecNum><Displ
ayText>(Zoeller & DisplayText><record><rec-number>194</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
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#### 2.3 Thyroid Pathophysiology

Historically, the most commonly observed thyroid conditions have been goiter (i.e., enlarged thyroid gland) and congenital hypothyroidism. Hypothyroidism is defined as serum fT4 below and serum TSH above the reference range. Hypothyroidism may result from iodine insufficiency, from genetic disorders, autoimmune disease, cancer, pituitary insufficiency, or environmental factors. The most common form of adult hypothyroidism results from an autoimmune thyroiditis (Hashimoto's disease) whereby a variety of cellular and antibody-related immune processes reduce thyroid hormone production within the gland [ ADDIN EN.CITE

<EndNote><Cite><Author>Pyzik</Author><Year>2015</Year><RecNum>198</RecNum><Prefix >for review see </Prefix><DisplayText>(for review see Pyzik, Grywalska, Matyjaszek-Matuszek, &amp; Roliński, 2015)</DisplayText><record><rec-number>198</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</td>

timestamp="1466427626">198</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pyzik, Aleksandra</author><author>Grywalska, Ewelina</author><author>Matyjaszek-Matuszek, Beata</author><author>Roliński, Jacek</author></authors></contributors><titles><title>Immune disorders in Hashimoto's Thyroiditis: What do we know so far?</title><secondary-title>Journal of Immunology Research</secondary-title></title><periodical><full-title>Journal of immunology research</full-title></periodical><volume>2015
/volume><dates><year>2015</pr>
//gear></dates><isbn>2314-8861</pr>
//isbn><urls></record></Cite></EndNote>]. Both post-partum thyroiditis and subacute lymphocytic (or silent) thyroiditis are considered subtypes of this disorder. Post-partum thyroiditis can occur in up to 5% of women and can involve periods of hyper- and hypo-thyroid activity [ADDIN EN.CITE

<EndNote><Cite><Author>Akamizu</Author><Year>2015
Year><RecNum>169
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/Prefix><DisplayText>(for review see Akamizu & DisplayText)
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/key>
/foreign-keys><ref-type name="Book Section">5</ref-type><contributors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><au

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L.J. </author > Chrouos, G. </author > Dungan, K. </author > cauthor > Feingold,
K.R.</author><author>Grossman, A.</author><author>Hershman, J.M.</author><author>Koch, C.
</author><author>Korbonits, M.</author><author>McLachlan, R.</author><author>New, M.
</author><author>Purnell, J.</author><author>Rebar, R. </author><author>Singer, F.
</author><author>Vinik, A. </author></secondary-
authors></contributors></title>Hashimoto's Thyroiditis</title><secondary-title>Endotext
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Inc.</publisher><urls></urls></record></Cite><Cite><Author>Pearce</Author><Year>2015</Year
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type><contributors><author>Pearce,
E.N.</author></authors></contributors><titles><title>Thyroid disorders during pregnancy and
postpartum</title><secondary-title>Best Practice & Dinical Obstetrics & postpartum
Gynaecology</secondary-title></title>>cperiodical><full-title>Best Practice & Description (Application of the Computation of th
Clinical Obstetrics & Samp; Gynaecology </ full-title > </periodical > <pages > 700-
706</pages><volume>29</volume><number>5</number><dates><year>2015</year></dates><isbn
>1521-6934</isbn><urls></urls></fcord></Cite></EndNote>]. Generally, the condition will
resolve itself; however, as many as 20% of cases may lead to permanent hypothyroidism.
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Hashimoto's disease increases the risk of hypothyroidism, including both subclinical and overt hypothyroidism. Subclinical hypothyroidism is defined by only elevated (or outside the reference range) TSH concentrations, with serum T3 and T4 concentrations that remain within the population reference range, while overt hypothyroidism is defined by elevated serum TSH concentrations concurrent with decreased (or outside the reference range) serum T3 and T4 concentrations. Subclinical hypothyroidism has the potential to progress to overt hypothyroidism over time; thus, it requires serum thyroid function test monitoring. Other related autoimmune hypothyroid conditions include primary thyroid atrophy (Ord's disease), which is characterized by thyroid atrophy through invasion by T-lymphocytes, and is likely part of the continuum of hypothyroid conditions, rather than a separate disorder from Hashimoto's disease [ ADDIN EN.CITE <EndNote><Cite><Author>Carlé</Author><Year>2009</Year><RecNum>172</RecNum><Displa yText>(Carlé et al., 2009)</DisplayText><record><rec-number>172</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466200413">172</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Carlé, Allan</author><author>Pedersen, Inge Bulow</author><author>Knudsen, Nils</author><author>Perrild, Hans</author><author>Ovesen, Lars</author><author>Jørgensen, Torben</author><author>Laurberg, Peter</author></authors></contributors><title>Thyroid volume in hypothyroidism due to autoimmune disease follows a unimodal distribution; evidence against primary thyroid atrophy and autoimmune thyroiditis being distinct diseases</title><secondary-title>The Journal of Clinical Endocrinology & December 2015 Endocrinology & Clinical Endocrinology & Detabolism</full-title></periodical><pages>833-839</pages><volume>94</volume><number>3</number><dates><year>2009</year></dates><isbn >0021-972X</isbn><urls></urls></record></Cite></EndNote>]. Viral or other infections can result in the painful condition of subacute granulomatous thyroiditis. This condition, like post-partum thyroiditis, is generally transient and requires only beta-blocker and non-steroidal anti-inflammatory

drugs (NSAIDs) or steroid treatment to resolve. Although not common, pituitary disorders or tumors can also result in reduced thyroid function through reduction in TSH output.

Thyroid peroxidase antibodies (also referred to as anti-thyroid peroxidase antibodies or TPO Abs) are a marker for thyroid autoimmunity, a condition in which a person's immune system antibodies mistakenly target parts of the thyroid gland or thyroid proteins. Thyroid autoimmunity leads to chronic inflammation of the thyroid, can cause tissue damage, and may disrupt thyroid function. The TPO Abs work against thyroid peroxidase, which is an enzyme involved in the synthesis of thyroid hormones [ ADDIN EN.CITE

<EndNote><Cite><Author>Braverman
Author><Year>2004</Year><RecNum>1991</pr>
DisplayText>(L Braverman & Eamp; Utiger, 2004)</pr>
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timestamp="1525112324">1991</key></foreign-keys><ref-type name="Book Section">5</ref-type><contributors><author>Braverman, L </author><author>Utiger,
R.S.</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a

#### 2.4 Perchlorate's Impact on Thyroid Physiology

The NRC (2005) proposed a mode of action for perchlorate. [ REF \_Ref455000569 \h ] is a reproduction of this mode of action modified to include hypothyroxinemia. Evaluating hypothyroxinemia as a result of perchlorate exposure is the mode of action that the SAB recommended the EPA utilize in evaluating the MCLG for perchlorate. [ REF \_Ref455000569 \h ] begins with exposure to perchlorate, which then leads to perchlorate in the bloodstream, followed by the inhibition of iodide uptake in the thyroid. The NRC concluded that these steps have all been observed in humans, and any subsequent steps or results are biologically possible, but have not been clearly demonstrated in humans. However, since the 2005 NRC report was published, several epidemiological studies have shown perchlorate exposure to be associated with changes in serum thyroid hormone levels [ ADDIN EN.CITE | ADDIN EN.CITE.DATA | ]. Furthermore, Taylor et al. (2014) demonstrated an association between high maternal perchlorate exposure and risk of low IQ in offspring.

In their review, NRC deemed the inhibition of iodide uptake in the thyroid to be a non-adverse effect, and concluded that the body's compensatory mechanisms would keep thyroid hormone levels within the population reference range [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-type><contributors><author>NRC</author></author></contributors><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><publication>Washington, DC</pub-location><publisher>National Academies

Press</publisher><label>627612</label><urls><related-

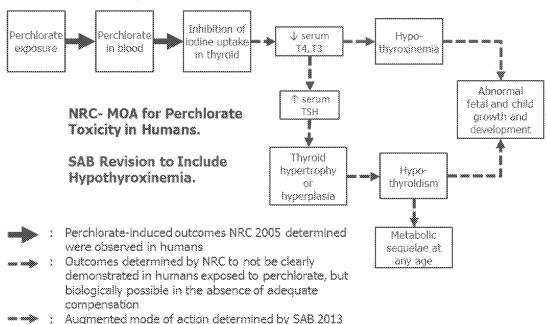
urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-

date>National Research Council</modified-date></record></Cite></EndNote>]. However, concerns have been raised regarding this conclusion given the lack of empirical evidence provided to support it [ ADDIN EN.CITE <EndNote><Cite><Author>Ginsberg</Author><Year>2007</Year><RecNum>345</RecNum><Dis playText>(Ginsberg, Hattis, Zoeller, & Samp; Rice, 2007)</DisplayText><record><recnumber>345</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495122121">345</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Ginsberg, G.</author><author>Hattis, R.</author><author>Zoeller, R Thomas</author><author>Rice, D.C.</author></authors></contributors><title>Evaluation of the U.S. EPA/OSWER Preliminary Remediation Goal for Perchlorate in Groundwater: Focus on Exposure to Nursing Infants</title><secondary-title>Environmental Health Perspectives</secondarytitle></title>><periodical><full-title>Environmental Health Perspectives</fulltitle></periodical><volume>115</volume><number>3</number><section>361</section><dates><y ear>2007</year></dates></urls></record></Cite></EndNote>]. The NRC did list subsequent biologically plausible steps that include the following: altered thyroid hormone levels (e.g., decreased T3 and T4, increased TSH; which has now been demonstrated in the epidemiologic literature), growth of the thyroid gland (e.g., thyroid hypertrophy and hyperplasia), and hypothyroidism (defined above). Hypothyroidism, deemed by the NRC to be the first adverse effect, could lead to abnormal fetal and child growth development, as well as metabolic disorders at all ages. Given the importance of a properly functioning thyroid for fetal and child growth [ ADDIN EN.CITE <EndNote><Cite><Author>Forhead</Author><Year>2014</Year><RecNum>1890</RecNum><Di splayText>(Forhead & DisplayText)</ri> number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497623515">1890</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Forhead, A J</author><author>Fowden, A L</author></authors></contributors></title>Thyroid hormones in fetal growth and prepartum maturation</title><secondary-title>Journal of Endocrinology</secondarytitle></title></periodical><full-title>Journal of Endocrinology</full-title></periodical><pages>R87-R103</pages><volume>221</volume><number>3</number><dates><year>2014</year><pubdates><date>June 1, 2014</date></pub-dates></dates><urls><relatedurls><url>http://joe.endocrinology-journals.org/content/221/3/R87.abstract</url></relatedurls></urls><electronic-resource-num>10.1530/joc-14-0025</electronic-resourcenum></record></Cite></EndNote>], both hypothyroidism and hypothyroxinemia may also be

In the SAB report on approaches to derive an MCLG for perchlorate, the SAB indicates that "the mode of action of perchlorate toxicity is well understood" (SAB, 2013, p. 2). The SAB finds that "hypothyroxinemia (i.e., low levels of thyroid hormone) is a more appropriate indicator of the potential adverse health effects than the more pronounced decreases in thyroid hormone associated with hypothyroidism" (SAB, 2013, p. 2). Moreover, the "sensitive populations EPA should consider for exposure to perchlorate are the fetuses of hypothyroxinemic pregnant women" (SAB, 2013, p. 2). Thus, this report is following the SAB's recommendations.

associated with abnormal fetal and child growth.

Figure [ SEQ Figure \\* ARABIC ]. Modified Representation of NRC's Suggested Mode of Action for Perchlorate Toxicity in Humans Indicating First Adverse Effect in the Continuum of Perchlorate Exposure to Effect, Revised to Include Hypothyroxinemia as recommended by the SAB



 $Source: Adapted \ from \ NRC\ [\ ADDIN\ EN.CITE\ < EndNote > < Cite\ Exclude Auth="1"> < Author> National\ Research$ 

Council</Author><Year>2005</Year><RecNum>116</RecNum><Suffix>`, Figure 5-2 (modified)</Suffix><DisplayText>(2005, Figure 5-2 (modified))</DisplayText><record><rec-number>116</rec-number><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914312">116</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>National Research Council

(NRC),</author></authors></contributors><titles><title>Health implications of perchlorate

exposure</title></title></dates><year>2005</year></dates><pub-location>Washington, D.C.</pub-location><publisher>The National Academies Press</publisher><urls></urls></record></Cite></EndNote>].

Notes: Solid arrows represent outcomes that the NRC (2005) determined were observed in humans during perchlorate exposure. Black dashed arrows represent outcomes determined by the NRC (2005) to have not been clearly demonstrated in humans exposed to perchlorate, but that are biologically possible in the absence of adequate compensation. Blue dashed arrows represent the augmented mode of action from the SAB.

In addition to the NRC's proposed mode of action for perchlorate, it is also important to note that the interruption of iodide uptake by perchlorate creates a deficit that can accentuate the effect of other goitrogenic (i.e., goiter-causing) anions on the symporter. For example, [ HYPERLINK \l "\_ENREF\_107" \o "Steinmaus, 2007 #199" ] [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Steinmaus</Author><Year>2007</Year><RecNum>199</RecNum>< DisplayText>(2007)</DisplayText><record><rec-number>199</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466428909">199</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Steinmaus, C</author><author>Miller, Mark D</author><author>Howd, Robert</author></author></authors></authors><tittle>Timpact of smoking and thiocyanate on perchlorate and thyroid hormone associations in the 2001-2002 National Health and Nutrition Examination Survey</title><secondary-title>Environmental Health Perspectives</secondary-title></title><periodical><full-title>Environmental Health

Perspectives</full-title></periodical><pages>1333-1338</pages><dates><vear>2007</vear></dates><isbn>0091-6765</isbn><urls></urls></record></Cite><Cite><Author>Steinmaus</Author><Year>2007</Year ><RecNum>199</RecNum><record><rec-number>199</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466428909">199</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>>Steinmaus, C</author><author>Miller, Mark D</author><author>Howd, Robert</author></authors></contributors><title>Impact of smoking and thiocyanate on perchlorate and thyroid hormone associations in the 2001-2002 National Health and Nutrition Examination Survey</title><secondary-title>Environmental Health Perspectives</secondary-title></title><periodical><full-title>Environmental Health Perspectives</full-title></periodical><pages>1333-1338</pages><dates><year>2007</year></dates><isbn>0091-6765</isbn><urls></record></Cite></EndNote>] demonstrated that the combined effect of smoking and perchlorate is much greater than the effect of perchlorate or smoking on its own. Smoking cigarettes greatly increases an individual's exposure to thiocyanate, a known goitrogen. Thus, this upstream effect creates a vulnerability that increases the potential for chemical-chemical interaction (e.g., between perchlorate and other goitrogenic anions; or, between perchlorate and other anti-thyroid agents, such as polychlorinated biphenyls (PCBs)). It also increases the potential for

chemical-nutritive status interaction (e.g., between perchlorate and low iodine intake), and

disease status) in which the functional reserve of the HPT system is limited.

perchlorate-physiological status interaction (e.g., between pregnancy, neonatal adaptation, aging, and

While perchlorate is expected to impair the production of T4 by the thyroid gland by competitively inhibiting the transfer of iodide into the thyroid gland, other environmental contaminants have been implicated as endocrine disruptors of other components of the HPT axis or plasma transport proteins. Multiple other chemicals, such as PCBs, polybrominated diphenyl ethers (PBDEs), bisphenol A, and perfluorinated acids (PFAs), also disrupt thyroid homeostasis [ ADDIN EN.CITE EN.CITE.DATA ]. These disruptions occur through various mechanisms of action including directly affecting the thyroid's responsiveness to TSH [ ADDIN EN.CITE <EndNote><Cite><Author>Bansal</Author><Year>2008</Year><RecNum>170</RecNum><Displ ayText>(Bansal & DisplayText><record><rec-number>170</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466200263">170</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Bansal, Ruby</author><author>Zoeller, R Thomas</author></authors></contributors></title>Polychlorinated biphenyls (Aroclor 1254) do not uniformly produce agonist actions on thyroid hormone responses in the developing rat brain</title><secondary-title>Endocrinology</secondary-title></title>><periodical><fulltitle>Endocrinology</full-title></periodical><pages>4001-4008</pages><volume>149</volume><number>8</number><dates><vear>2008</vear></dates><is bn>0013-7227</isbn><urls></record></Cite></EndNote>], affecting binding proteins [ ADDIN EN.CITE.DATA ], preventing NIS-mediated iodide uptake, and ADDIN EN.CITE affecting expression of thyroid hormone-producing genes [ ADDIN EN.CITE <EndNote><Cite><Author>Wu</Author><Year>2016</Year><RecNum>183</RecNum><Display Text>(Wu, Beland, & DisplayText><record><rec-number>183</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466201102">183</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><authors><author>Wu, Yuanfeng</author><author>Beland, Frederick A</author><author>Fang, Jia-Long</author></author><author>Fang, Jia-Long</author></author></author>><author>Fang, Jia-Long</author></author></author></author>><author>><author>Fang, Jia-Long</author></author></author></author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>><author>

#### 2.5 Distribution and Excretion of Perchlorate

Perchlorate salts such as ammonium, potassium, or sodium are highly water soluble and readily ionize in water. Once ingested, perchlorate is rapidly absorbed by the gastrointestinal tract and passed into circulation. In single-dose studies perchlorate has a short half-life of only 5-12 hours, but may be measured in urine within as little as 10 minutes to 3 hours after ingestion [ADDIN EN.CITE <EndNote><Cite><Author>Anbar</Author><Year>1959</Year><RecNum>264</RecNum><Displ ayText>(Anbar, Guttmann, & Dewitus, 1959)</DisplayText><record><rec-number>264</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468519973">264</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author><author>Anbar, M.</author><author>Guttmann, S.</author><author>Lewitus, Z.</author></contributors><title>The mode of action of perchlorate ions on the iodine uptake of the thyroid gland</title><secondarytitle>International Journal of Applied Radiation and Isotopes</secondarytitle></titles><periodical><full-title>International Journal of Applied Radiation and Isotopes</fulltitle></periodical><pages>87-96</pages><volume>7</volume>edition>1959/12/01</edition><keywords><keyword>Iodine/\*met abolism</keyword><keyword>Perchloric Acid/\*pharmacology</keyword><keyword>Thyroid Gland/\*pharmacology</keyword></keywords><dates><year>1959</year><pubdates><date>Dec</date></pub-dates></dates><isbn>0020-708X (Print)&#xD:0020-708X (Linking)</isbn><accession-num>13793274</accession-num><urls><related-urls><url> face="underline" font="default" size="100%">http://www.ncbi.nlm.nih.gov/pubmed/13793274</style></url></relatedurls></urls><research-notes><style face="bold" font="default" size="100%">Abt to Include</style></research-notes><language>eng</language></record></Cite></EndNote>]. Shortduration (2-week exposure) studies indicate a half-life of approximately 8 hours [ADDIN EN.CITE <EndNote><Cite><Author>Greer</Author><Year>2002</Year><RecNum>204</RecNum><Displa yText>(Greer et al., 2002)</DisplayText><record><rec-number>204</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1467812686">204</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Greer, Monte A</author>Goodman,

Gay</author><author>Pleus, Richard C</author><author>Greer, Susan E</author></authors></contributors></title>Health effects assessment for environmental perchlorate contamination: the dose response for inhibition of thyroidal radioiodine uptake in humans</title><secondary-title>Environmental Health Perspectives</secondarytitle></title></title><periodical><full-title>Environmental Health Perspectives</fulltitle></periodical><pages>927</pages><volume>110</volume><number>9</number><dates><year >2002</year></dates><urls></urls></record></Cite></EndNote>], with as much as 77% of the perchlorate being excreted per day with consumption of 0.14 mg/kg/day [ ADDIN EN.CITE <EndNote><Cite><Author>Lawrence</Author><Year>2000</Year><RecNum>228</RecNum><Di splayText>(Lawrence et al., 2000)</DisplayText><record><rec-number>228</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468246590">228</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Lawrence, J.E.</author><author>S.H. Lamm</author><author>S. Pino</author><author>K. Richman</author><author>Braverman, L</author></authors></contributors></title>The effect of short-term low-dose perchlorate on various aspects of thyroid function</title><secondary-title>Thyroid</secondarytitle></title></periodical><full-title>Thyroid</full-title></periodical><pages>659-663</pages><volume>10</volume><number>8</number><dates><year>2000</year></dates><url> ></urls></record></Cite></EndNote>]. A six-month low-dose study of healthy volunteers at 0.5 mg or 3.0 mg per day indicated daily excretion rates of  $0.3 \pm 0.07$  mg/day and  $2.1 \pm 0.43$  mg/day, respectively [ ADDIN EN.CITE <EndNote><Cite><Author>Braverman</Author><Year>2006</Year><RecNum>225</RecNum>< DisplayText>(L Braverman et al., 2006)</DisplayText><record><rec-number>225</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468246163">225</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>>Braverman, L</author><author>Pearce, E.N.</author><author>He, X.</author><author>Pino, S.</author><author>Seeley, M.</author><author>Beck, B.</author><author>Magnani, B.</author><author>Bleunt, B.C.</author><author>Firek, A.</author></contributors><title>Effects of six months of daily low-dose perchlorate exposure on thyroid function in healthy volunteers</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondarytitle></titles><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</fulltitle></periodical><pages>2721-94</pages><volume>91</volume><number>7</number><dates><year>2006</year></dates><urls> </urls></record></Cite></EndNote>], representing approximately 60% and 70% excretion of perchlorate per day. There is little to no metabolic conversion or catabolism of perchlorate ions prior to excretion.

The remaining perchlorate is transported by the NIS in the thyroid, kidney, gastric mucosa, salivary glands, placenta, mammary glands, and choroid plexus [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Placental NIS plays a role in ensuring adequate transfer of maternal iodide to the fetus by actively transferring iodide from the maternal blood circulation to the fetal blood circulation. Disruption of the supply processes, either due to low maternal T3/T4 levels, or by inhibition of the placental NIS by perchlorate, may further reduce fetal T3/T4 levels. Competitive inhibition of placental NIS by perchlorate could reduce uptake of iodide from the maternal circulation and reduce thyroid hormone production within the fetal thyroid gland. Further, once the fetus has a

functional thyroid, inhibition of the fetal thyroidal NIS, as well as iodide deficiency from the blockade of the placental NIS, could result in additional reductions in fetal T3/T4 levels.

There is evidence from *in vivo* studies, confirmed by cell culture work, showing that perchlorate is actively transported into the thyroid gland and this transport may be enhanced by increasing TSH and reduced in a dose-dependent manner by iodide [ ADDIN EN.CITE ADDIN EN.CITE.DATA 1. This active transport is potentially due to the upregulation of NIS [ ADDIN EN.CITE <EndNote><Cite><Author>Hussein</Author><Year>2012</Year><RecNum>303</RecNum><Dis playText>(Hussein, Abbas, El Wakil, Elsamanoudy, & Dariz, 2012)</DisplayText><record><rec-number>303</rec-number><foreign-keys><key app="EN" dbid="z9t0ayxyzdfermedxe5yxfpkax2yzp0fty29" timestamp="1471880522">303</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>>thor>Hussein, Abd El-Aziz M.</author><author>Abbas, Amr M.</author><author>El Wakil, Gehan A.</author><author>Elsamanoudy, Ayman Z.</author><author>El Aziz, Azza Abd</author></authors></contributors><title>Effect of chronic excess iodine intake on thyroid function and oxidative stress in hypothyroid rats</title><secondary-title>Canadian Journal of Physiology and Pharmacology</secondary-title></title><periodical><full-title>Canadian Journal of Physiology and Pharmacology</full-title></periodical><pages>617-625</pages><volume>90</volume><number>5</number><dates><year>2012</year><pubdates><date>2012/05/01</date></pub-dates></dates><publisher>NRC Research Press</publisher><isbn>0008-4212</isbn><urls><relatedurls><url>http://dx.doi.org/10.1139/y2012-046</url></related-urls></urls><electronic-resourcenum>10.1139/y2012-046</lectronic-resource-num><access-date>2016/08/22</accessdate></record></Cite></EndNote>]. These observations have implications for the placental transfer of iodide and perchlorate and subsequent fetal thyroid functions. As yet there is no direct evidence to indicate that the quantity of placental NIS is upregulated in response to increased TSH in the same way, or to the same extent, as seen in Tran et al. (2008). However, if this is the case, additional stimulation by maternal TSH in the presence of perchlorate may result in an absolute increase in NIS and subsequently increase perchlorate transfer to the placenta. The immature fetal HPT axis has very limited capacity to increase output of thyroid hormones [ ADDIN EN.CITE EN.CITE.DATA ], so the fetal HPT may not be able to adjust output in the face of reduced maternal fT4 supply and increased perchlorate transfer.

Perchlorate, like thiocyanate, selenium cyanate, and nitrate, competitively binds, in a dose-dependent manner, to the NIS and reduces iodide uptake by the thyroid gland. Recent work using a non-linear regression model shows that the Michaelis-Menten constant ( $K_m$ ), which characterizes the binding affinity of perchlorate for the NIS, is approximately 0.59  $\mu$ M, which is about 60% of that originally predicted [ ADDIN EN.CITE

<EndNote><Cite><Author>Schlosser
/Author><Year>2016
/Year><RecNum>293
/RecNum><DisplayText>(Schlosser, 2016)
/DisplayText><record><rec-number>293</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</p>
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/authors>
/contributors><author>Schlosser, P. M.</author>
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/contributors><author>Schlosser, P. M.
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/sessment, Washington, DC, USA.
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Journal of Applied Toxicology
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title></periodical><edition>2016/05/14</edition><dates><year>2016</year><published ates><date>May 13</date></published ates></dates><isbn>1099-1263 (Electronic)&#xD;0260-437X (Linking)</isbn><accession-num>27177048</accession-num><urls></urls><electronic-resource-num>10.1002/jat.3337</electronic-resource-num><remote-database-provider>NLM</remote-database-provider><language>Eng</language></record></Cite></EndNote>]. This means that approximately half as much perchlorate is required to out-compete iodide than suggested by earlier models. Since both iodide and perchlorate would be consumed during the day, changes in dietary intake, uptake, and competition between the two ions will vary depending on the foods consumed and water sources, and the rate of clearance of each ion from circulation.

### 2.6 Physiologic Connection between Thyroid Hormone Levels and Neurodevelopment

The profound effects of thyroid deficiencies during development were first documented in children with congenital hypothyroidism by Paracelsus [ ADDIN EN.CITE

<EndNote><Cite><Author>Cranefield</Author><Year>1963</Year><RecNum>1892</RecNum>
DisplayText>(Cranefield & DisplayText><record><rec-number>1892</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497969366">1892</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Cranefield, P.</author><author>Federn,

W.</author></contributors></title>Paracelsus on goiter and cretinism: A translation and discussion of &quot;De Struma, Vulgo Der Kropf&quot;</title><secondary-title>Bulletin of the History of Medicine</secondary-title></title><periodical><full-title>Bulletin of the History of Medicine</periodical><pages>463-

471</pages><volume>37</volume>dates><year>1963</year></dates><urls></record></Cit e></EndNote>]. Modern studies demonstrate that unless treated, children with congenital hypothyroidism will suffer from problems associated with somatic growth and an array of visuomotor and neurocognitive deficits. Aberrant patterns of growth, development, and organization of the cortex, cerebellum, hippocampus, basal ganglia, and brain stem can all result from reduced thyroid hormone exposure during development [ ADDIN EN.CITE

<EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(G. Williams, 2008; Zoeller & DisplayText><record><recnumber>193</rec-number><foreign-keys><key app="EN" db-"

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><uthors></authors></authors></title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title></periodical><full-title>Journal of neuroendocrinology</full-title></periodical><pages>784-

794 < / pages > < volume > 20 < / volume > < number > 6 < / number > < dates > < year > 2008 < / year > < / dates > < isbn > 1365 -

2826</isbn><urls></record></cite><Cite><Author>Zoeller</Author><Year>2004</Year>< RecNum>194</RecNum><record><rec-number>194</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Zoeller, R Thomas</author><author>Rovet, J</author></author></contributors><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental

findings</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title></title></periodical><full-title>Journal of neuroendocrinology</full-title></periodical><pages>809-

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>1365-2826</isbn><urls></record></Cite></EndNote>].

Thyroid hormones are essential for the development and differentiation of the developing brain. The brain and spinal cord begin development in the first half of the first trimester. fT4 passes through the blood-brain barrier via multiple, specific transporter proteins. Next, T4 is converted to T3 by the developing glial cells and then transported to neurons. T3 then interacts with nuclear receptors to tightly regulate gene expression so that neurogenesis, synaptogenesis, neuronal migration, cell differentiation, and myelination are developmentally appropriate. Deficiencies in thyroid hormones through iodine deficiency, congenital hypothyroidism, or maternal hypothyroidism/hypothyroxinemia can result in neurological impairments and intellectual deficits [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de

Escobar</br>
Escobar
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timestamp="1496433675"
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Morreale de Escobar
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3987</pages><volume>85</volume><number>11</number><dates><year>2000</year></dates><urls></urls></record></Cite></EndNote>].

The fetal thyroid gland first begins to concentrate iodide at about 11–12 weeks and begins significant contribution to fetal development around 16 weeks. However, the developing fetus still relies on some supply of maternal thyroid hormones, primarily fT4, until birth [ ADDIN EN.CITE <EndNote><Cite><Author>Morreale de

Escobar
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G Morreale de
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U37</pages><volume>151</volume><number>Suppl 3</number><dates><year>2004</year><pubdates><date>November 1, 2004</date></pub-dates></dates><urls><related-urls><url>http://www.eje-online.org/content/151/Suppl\_3/U25.abstract</url></related-urls></urls><electronic-resource-num>10.1530/eje.0.151U025</electronic-resource-num></record></Cite></EndNote>]. The fetal thyroid is not fully mature until birth, so premature birth can also be associated with low thyroid hormone concentrations. Thus, alterations in thyroid

hormone levels can impact brain development in a variety of ways depending on when the alteration occurs and for how long.

For example, early in brain development, fT4 deficiency can affect neuronal cell proliferation and cell migration in the hippocampus (essential for learning and memory, cognitive function), the cerebral cortex (essential for executive function, cognitive function), and the medial ganglionic eminence (transitory development structure responsible for guiding axonal migration) [ ADDIN EN.CITE <EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(G. Williams, 2008)</DisplayText><record><rec-number>193</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Williams,

GR</author></authors></contributors><titles><title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondary-title></title></periodical><full-title>Journal of neuroendocrinology</full-title></periodical><pages>784-

794</pages><volume>20</volume>6</number>6</number><dates><year>2008</year></dates><isbn>1365-2826</isbn><urls></record></Cite></EndNote>]. Maternal hypothyroidism or hypothyroxinemia at these early stages may result in problems with gross motor skills and visual attention and processing [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller
Author><Year><2004</pre>
/Year><RecNum>194
ayText>(Zoeller & amp; Rovet, 2004)
/DisplayText><record><rec-number>194
number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</pre>
timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Zoeller, R Thomas</author><author>Rovet, J</a>
J
/author>
/author>
/contributors><titles><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings
/title><secondary-title>
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/periodical><full-title>
Journal of neuroendocrinology
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818 < pages > < volume > 16 < / volume > 10 < / mumber > 10 < / mumber > < dates > < year > 2004 < / year > < / dates > < isb n > 1365-2826 < / isbn > < / urls > < / record > < / Cite > < / EndNote > ]. As the first trimester progresses, increases in hCG result in a higher production of T4 [ ADDIN EN.CITE

<EndNote><Cite><Author>Morreale de

Escobar</ali>
Escobar</ali>
Author><Year>2007</Year><RecNum>201</r>
RecNum><DisplayText>(G Morreale de Escobar et al., 2007)
DisplayText><record><rec-number>201</r>
rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</li>

timestamp="1466438813">201</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Morreale de Escobar, G</author><author>Obregón, María Jesús</author><author>Escobar del Rey, F. </author></authors></contributors><titles><title>Iodine deficiency and brain development in the first half of pregnancy</title><secondary-title>Public Health Nutrition</secondary-title></title><periodical><full-title>Public health nutrition</full-title></periodical><pages>1554-

1570</pages><volume>10</volume><number>12A</number><dates><year>2007</year></dates>< isbn>1475-2727</isbn><urls></record></Cite></EndNote>]. In addition, serum TBG increases during the first trimester, remaining elevated throughout pregnancy because estrogen increases the glycosylation of TBG, increasing its serum half-life. In the fetus this period also corresponds to neurogenesis in the midbrain and cortex, neuronal migration into the cortex,

deiodinase activity for conversion of T4 to T3, and presence of nuclear thyroid receptors in the brain [ ADDIN EN.CITE < EndNote > < Cite > < Author > Morreale de

Escobar</Author><Year>2004</Year><RecNum>49</RecNum><DisplayText>(G Morreale de Escobar et al., 2004)</br/>
/DisplayText><record><rec-number>49</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1437077734">49</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><authors><author>Morreale de Escobar, G</author><author>Obregón, M J</author><author>Escobar del Rey, F</author></authors></contributors><title>Role of thyroid hormone during early brain development</title><secondary-title>European Journal of Endocrinology</secondary-title></title>>epriodical><full-title>European Journal of Endocrinology</full-title></periodical><pages>U25-

U37</pages><volume>151</volume><number>Suppl 3</number><dates><year>2004</year><pubdates><date>November 1, 2004</date></pub-dates></dates><urls><relatedurls><url>http://www.eje-online.org/content/151/Suppl 3/U25.abstract</url></relatedurls></urls><electronic-resource-num>10.1530/eje.0.151U025</electronic-resourcenum></record></Cite></EndNote>1.

Higher levels of maternal fT4 throughout pregnancy allow for normal neuron and glial cell migration and differentiation, axonal growth, dendritic branching, neurogenesis, and synaptogenesis to occur [ ADDIN EN.CITE

<EndNote><Cite><Author>Williams</Author><Year>2008</Year><RecNum>193</RecNum><Dis playText>(G. Williams, 2008)</br/>
/DisplayText><record><rec-number>193</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1466202836">193</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Williams,

GR</author></authors></contributors></title>Neurodevelopmental and neurophysiological actions of thyroid hormone</title><secondary-title>Journal of Neuroendocrinology</secondarytitle></titles><periodical><full-title>Journal of neuroendocrinology</fulltitle></periodical><pages>784-

794</pages><volume>20</volume><number>6</number><dates><year>2008</year></dates><isbn >1365-2826</isbn><urls></urls></record></Cite></EndNote>1. The complex interconnectedness of brain regions required for normal function could be impaired by effects on all of these processes simultaneously or in sequence, and across multiple brain regions. Maternal thyroid deficiencies during these processes further affect visual processing, development of fine motor skills, IQ, and selective learning problems [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller</Author><Year>2004</Year><RecNum>194</RecNum><Displ ayText>(Zoeller & Dover, 2004; Zoeller et al., 2007)</br/>/DisplayText><record><recnumber>194</rec-number><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202917">194</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>> R Thomas</author><author>Rovet, J</author></authors></contributors><title>Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings</title><secondary-title>Journal of Neuroendocrinology</secondarytitle></titles><periodical><full-title>Journal of neuroendocrinology</full-

title></periodical><pages>809-

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>13652826</isbn><urls></record></Cite><Cite><Author>Zoeller</Author><Year>2007</Year>< RecNum>186</RecNum><record><rec-number>186</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1466202457">186</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Zoeller, R Thomas</author><author>Tan, Shirlee W</author><author>Tyl, Rochelle W</author></author></author></contributors><titles><title>General background on the hypothalamic-pituitary-thyroid (HPT) axis</title><secondary-title>Critical Reviews in Toxicology</secondary-title></title></periodical><pages>11-53</pages><volume>37
/volume><number>1-2
2
/number><dates><year>2007
/year></dates><isbn>1040-8444
8444
</ir>
/sibn><urls></record>
</fr>
/Cite></EndNote>]. Additionally, concern is emerging that there may be adverse effects associated with fT4 levels above the normal range (Korevaar et al., 2016).

Additionally, humans continue to undergo considerable post-natal brain development well through puberty and into early adulthood. A properly functioning HPT is necessary for proper neurologic development at this stage [ ADDIN EN.CITE

<EndNote><Cite><Author>Zoeller
Author><Year><2004</pre>
/Year><RecNum>194
ayText>(Zoeller & amp; Rovet, 2004)
/DisplayText><record><rec-number>194
number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</pre>
timestamp="1466202917">194</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><au

818</pages><volume>16</volume><number>10</number><dates><year>2004</year></dates><isb n>1365-2826</isbn><urls></record></Cite></EndNote>]. This is particularly true in the hippocampus, which is considered to have a relatively high degree of neuroplasticity in humans and undergoes considerable adult neurogenesis. How the behavioral and cognitive deficits associated with subclinical hypothyroidism and hypothyroxinemia are manifested within the brain is a research area where further evaluation is needed.

## 3. Overview of the EPA's Biologically Based Dose-Response Model for Perchlorate's Impact on Thyroid Hormone Levels

The EPA intends to inform the derivation of the MCLG for perchlorate by utilizing a BBDR model for early pregnancy to predict the impact that perchlorate will have on serum thyroid hormone levels at a given iodine intake (Step 1 of [ REF \_Ref417634154 \h ]). This section summarizes the history of the development of the BBDR model along with its current structure (Section [ REF \_Ref482890532 \r \h ]). Further details are provided on key model parameters and attributes (Section [ REF \_Ref482890596 \r \h ]) and Section [ REF \_Ref484089893 \r \h ]), which is followed by a summary of results relevant to this analysis (Section [ REF \_Ref483494050 \r \h ]). Lastly, there is a discussion of potential uncertainties associated with the model (Section [ REF \_Ref482972894 \r \h ]). The BBDR model overview provided here is supported by detailed descriptions in Appendix A.

The BBDR model can be described as containing two main components: (1) a pharmacokinetic model for perchlorate and iodide, which describes chemical absorption, distribution, metabolism, and excretion of these two anions; and (2) a pharmacodynamic model, which describes the joint effect of varying perchlorate and iodide blood concentrations on thyroidal uptake of iodide and subsequent production of thyroid hormones, most significantly T4. The pharmacokinetic portion contains a physiological description (organ volumes, blood flows) and chemical-specific information (partition coefficients, volume of distribution, rate constants for transport, metabolism, and elimination) that enable a prediction of perchlorate and iodide internal concentration at the critical target (i.e., thyroidal NIS) in association with a particular exposure scenario (route of exposure, age, dose level). This portion of the model is similar to other physiologically based pharmacokinetic models and for perchlorate is simplified by the absence of metabolism. The pharmacodynamic portion of the model uses this internal concentration to simulate how the chemical will act within a known mode of action to perturb host systems and lead to a toxic effect. Thus, BBDR modeling attempts to predict the internal dose of a chemical associated with a particular exposure scenario and the perturbation this internal dose can have on host systems.

#### 3.1 Background and Structure of BBDR Model Development

In 2012, the EPA sought advice from the SAB on how best to consider and interpret life stage information, epidemiologic and biomonitoring data, PBPK analyses, and the totality of perchlorate health information to derive an MCLG for perchlorate. In 2013, the SAB recommended the following ADDIN EN.CITE

<EndNote><Cite><Author>SAB</Author><Year>2013
Year><2013</p>
Year

Derive a perchlorate MCLG that addresses sensitive life stages through PBPK/PD modeling;

- "Expand the modeling approach to account for thyroid hormone perturbations and potential adverse neurodevelopmental outcomes from perchlorate exposure;
- Utilize a mode-of-action framework for developing the MCLG that links the steps in the proposed mechanism leading from perchlorate exposure through iodide uptake inhibition to thyroid hormone changes and finally neurodevelopmental impacts; [and]
- Extend the [BBDR] model expeditiously to...provide a key tool for linking early events with subsequent events as reported in the scientific and clinical literature on iodide deficiency, changes in thyroid hormone levels, and their relationship to neurodevelopmental outcomes during sensitive early life stages" (SAB, 2013, p. 19).

The SAB stated that this data-driven approach represents a more rigorous way to address differences in biology and exposure between adults and sensitive life stages than is possible with the default approach for deriving an MCLG, and that the EPA should also consider available data on potential adverse health effects (neurodevelopmental outcomes) due to perturbations in thyroid hormone levels regardless of the cause of those perturbations (SAB, 2013).

Based on the SAB's recommendations, the EPA, with contributions from FDA, first developed and integrated PBPK models for perchlorate and iodide with BBDR models for thyroid hormones to predict the effect of perchlorate on the thyroid gland in formula-fed and breastfed infants for the post-natal period from days 7 to 90, as well as lactating women and third-trimester pregnant women and her fetus. This draft version of the developed model was intended to be valid for fT4 levels ranging from euthyroid to as low as hypothyroxinemia, in which fT4 levels are decreased to the lower end of the normal (i.e., within the population reference) range and TSH levels remain within the normal range [ ADDIN EN.CITE

<EndNote><Cite><Author>Silva</Author><Year>1981</Year><RecNum>261</RecNum><Displa yText>(Silva & amp; Silva, 1981)
/DisplayText><record><rec-number>261
/rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"</p>
timestamp="1468519119">261
/key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Silva, J. E.</author><author>Silva,
S.</author></author>
/author>
/contributors>
/contributors>
/title>Interrelationships among serum thyroxine,
triiodothyronine, reverse triiodothyronine, and thyroid-stimulating hormone in iodine-deficient

triiodothyronine, reverse triiodothyronine, and thyroid-stimulating hormone in iodine-deficient pregnant women and their offspring: effects of iodine supplementation</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title>>cperiodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>671-7</pages><volume>52</volume><number>4</number><edition>1981/04/01</edition><keywords><keyword>Female

Blood/analysis</keyword><keyword>Humans</keyword><keyword>Infant,

Newborn</keyword><keyword>Iodides/urine</keyword><keyword>Iodine/ deficiency/therapeutic use</keyword><keyword><keyword>Pregnancy</keyword>Pregnancy Complications/

blood</keyword><keyword>Thyrotropin/blood</keyword><keyword>Thyroxine/

blood</keyword><keyword>Triiodothyronine/blood</keyword><keyword>Triiodothyronine,

Reverse/blood</keyword></keywords><dates><year>1981</year><pub-

 $dates > < date > Apr < / date > < / pub-dates > < / dates > < isbn > 0021-972X \ (Print) \& \#xD; 0021$ 

(Linking)</isbn><accession-num>7204539</accession-num><urls></urls><electronic-resource-

num>10.1210/jcem-52-4-671</electronic-resource-num><remote-database-provider>NLM</remote-database-provider><language>eng</language></record></Cite></EndNote>]. The model was limited to the range of perchlorate doses and dietary iodine intakes for which thyroid hormone levels are assumed to be maintained within a reference range without action of the TSH feedback mechanism.<sup>5</sup>

The model was subjected to external peer review in January 2017. The final peer review report titled External Peer Review for EPA's Draft Biologically Based Dose-Response (BBDR) Model and Draft BBDR Model Report for Perchlorate in Drinking Water is available through the docket at [ HYPERLINK "https://www.regulations.gov/document?D=EPA-HQ-OW-2016-0439-0006"].

The EPA considered all of the peer reviewers' recommendations and focused on those that were anticipated to be most important for increasing the scientific rigor of the model and modeling results. These revisions are summarized in this section; additional detail is available in Appendix A. Model revisions focused on the following key recommendations:

- Extending the model to early pregnancy;
- Incorporating biological feedback control of hormone production via TSH signaling, such that the model can describe lower levels of iodide nutrition;
- Calibrating the model and evaluating its behavior for upper and lower percentiles of the distribution of thyroid hormone levels in the population, as well as the population median thyroid hormone levels; and
- Conducting an uncertainty analysis for key parameters.

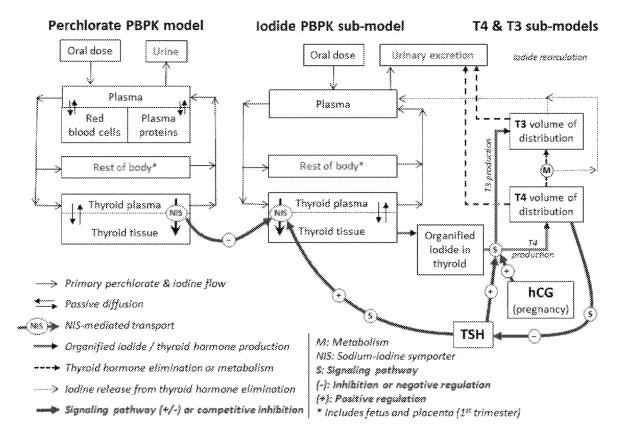
The BBDR model for which an overview is provided in this section and in [ REF \_Ref482868704 \h ] is described in complete detail in Appendix A. In this model, the EPA simulated the uptake and disposition of iodide and perchlorate and their competitive interaction with the thyroidal NIS within the thyroid and elsewhere in the body. Based upon the availability of iodide, thyroid hormone synthesis, distribution, and metabolic transformation are modeled to yield predictions of blood levels of T4 and T3 (i.e., total T4, fT4, free T3 (fT3), and total T3 (tT3)). Further, a TSH feedback loop has been incorporated in which TSH levels are simulated based upon the predicted level of fT4 according to the relationship between those hormones as observed in humans. In particular, a decrease in fT4 is predicted to increase blood TSH, the response to which is an increase in the thyroidal NIS expression, resulting in higher NIS activity. Increased blood TSH is also assumed to increase the production rate constants for T4 and T3. The BBDR model was first developed for non-pregnant women and then extended to pregnancy by incorporating time-dependent parameters based on changes observed in early pregnancy through gestational week (GW) 16. The BBDR model is designed to assess long-

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For a more detailed discussion on the development of the draft model, its calibration, and dose-response evaluations, please see the draft BBDR model report titled *Biologically Based Dose Response Models for the Effect of Perchlorate on Thyroid Hormones in the Infant, Breast Feeding Mother, Pregnant Mother, and Fetus: Model Development, Revision, and Preliminary Dose-Response Analysis.* The report is available through the docket at [ HYPERLINK "http://www.regulations.gov" ] (Docket ID No. EPA-HQ-OW-2016-0439-0006).

term, steady-state conditions in the non-pregnant woman and week-to-week variation in pregnancy, rather than short-term (hour-to-hour or day-to-day) fluctuations. Appendix A includes sensitivity and uncertainty analyses of the updated model.

Figure [ SEQ Figure \\* ARABIC ]. Overview of the Structure of the Early Pregnancy BBDR Model



As described in Section [ REF \_Ref482958608 \r \h ], the decision to focus on early pregnancy was based on the fact that the epidemiology supporting the relationship between maternal thyroid hormone levels and subsequent neurodevelopmental outcomes is stronger for deficits in fT4 during the first trimester as opposed to later in pregnancy (e.g., Pop et al., 1999, 2002). Given this focus on early pregnancy, the structure of the BBDR model is simplified from that of the previously described late-gestation model (previously reviewed in January 2017, based on Lumen et al., 2013), based upon the understanding that the fetal thyroid gland does not begin functioning until the second half of pregnancy. During early pregnancy, thyroid hormone-dependent neurodevelopment is driven by the supply of maternal thyroid hormone (in particular, serum concentrations of fT4) (Morreale de Escobar et al., 2004). Further, the epidemiology supporting this effect is stronger for first-trimester deficits in fT4 as opposed to later in pregnancy (Pop et al., 1999, 2002). In particular, the present model has been refined to focus on predicting maternal thyroid hormone levels during early pregnancy.

This model structure ([ REF Ref482868704 \h ]) is adapted from the third-trimester model of Lumen et al. (2013)<sup>6</sup> and the lactation iodide/thyroid hormone model of Fisher et al. (2016) with simplifications of some aspects and elaboration and adjustments of others. The model is comprised of submodels for perchlorate, iodide, T4, and T3. The six compartments used for perchlorate and iodide in the third-trimester BBDR model of Lumen et al. (2013) were condensed to essentially three compartments: plasma, thyroid, and "rest of body" (ROB). This choice is justified because the fetus and its constituent compartments are much smaller in the first trimester, and therefore it was deemed unnecessary to explicitly consider them. For example, at the end of the first trimester the fetus is expected to weigh less than 0.15 kg. To be clear, the placenta and fetus mass are included in the maternal ROB, and condensation of other maternal compartments into one compartment is not expected to significantly affect the predicted concentrations of the key anions and hormones given that they are described for long-term, continuous ingestion of iodine and perchlorate. Partitioning of perchlorate into red blood cells and binding to plasma protein are included in this model, unlike that of Lumen et al. (2013). Passive diffusion of perchlorate between plasma and red blood cells was included in the original human perchlorate PBPK model of Merrill et al. (2005) based on observations that iodide likewise distributed into red blood cells [ ADDIN EN.CITE <EndNote><Cite><Author>Rall</Author><Year>1950</Year><RecNum>2210</RecNum><Displa yText>(Rall, Power, & DisplayText><record><rec-number>2210</rec-

number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

38</pages><volume>314</volume><dates><year>2017</year></dates><urls></urls></record></Cite></ EndNote>] was published that extended the Lumen et al. (2013) model to include more severe iodine deficiency conditions, incorporate the effects of homeostatic mechanisms, and develop the extended model into a population-based model. Another study by Lumen & George [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Lumen</Author><Year>2017</Year><RecNum>1913</RecNum><DisplayT ext>(2017b)</DisplayText><record><rec-number>1913</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1503406958">1913</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Lumen, A.</author><author>George, N.I.</author></authors></contributors><titles><title>Evaluation of the risk of perchlorate exposure in a population of late-gestation pregnant women in the United States: Application of probabilistic biologically-based dose response modeling</title><secondary-title>Toxicology and Applied Pharmacology</secondary-title></title></error/title></error/title>Toxicology and applied pharmacology</full-title></periodical><pages>9-

14</pages><volume>322</volume>dates><year>2017</year></dates><urls></urls></record></Cite></ EndNote>] integrated the calibrated perchlorate exposure model from Lumen et al. (2013) into the population-based model in Lumen and George (2017a). Lumen and George (2017b) used Monte Carlo sampling and probabilistic simulation to quantify the distribution of urinary perchlorate for late-gestation pregnant women and to conduct an uncertainty analysis of influential perchlorate-specific parameters.

During the development of this report, a study by Lumen & George [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Lumen</Author><Year>2017</Year><RecNum>1912</RecNum><DisplayT ext>(2017a)</DisplayText><record><rec-number>1912</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1503406753">1912</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>Lumen, A.</author><author>George. N.I.</author></contributors><title><title>Estimation of iodine nutrition and thyroid function status in late-gestation pregnant women in the United States: Development and application of a population-based pregnancy model</title><secondary-title>Toxicology and Applied Pharmacology</secondary-title></titles><periodical><full-title>Toxicology and applied pharmacology</full-title></periodical><pages>24-

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timestamp="1542316851">2210</key></foreign-keys></ref-type name="Journal Article">17</ref-
type><contributors><author>Rall, J.E.</author><author>Power,
M.H.</author><author>Albert, A.</author></contributors></title>Distribution of
radioiodine in erythrocytes and plasma of man</title><secondary-title>Proceedings of the Society for
Experimental Biology and Medicin</secondary-title></title>>eriodical><full-title>Proceedings of
the Society for Experimental Biology and Medicin</full-title></periodical><pages>460-
461</pages><volume>74</volume><number>2</number><dates><year>1950</year></dates><urls
></urls></record></Cite></EndNote>]. Hays and Green [ ADDIN EN.CITE <EndNote><Cite
ExcludeAuth="1"><Author>Hays</Author><Year>1973</Year><RecNum>2209</RecNum><Displ
ayText>(1973)</DisplayText><record><rec-number>2209</rec-number><foreign-keys><key
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type><contributors><author>Hays, M.T.</author><author>Green,
F.A.</author></authors></contributors><title></title>In vitro studies of 99mTc-pertechnetate
binding by human serum and tissues</title><secondary-title>Journal of Nuclear
Medicine</secondary-title></title><periodical><full-title>Journal of Nuclear Medicine</full-
title><abbr-1>J Nucl Med</abbr-1></periodical><pages>149-
158</pages><volume>14</volume><number>3</number><dates><year>1973</year></dates><urls
></urls></record></Cite></EndNote>] and Scatchard and Black [ ADDIN EN.CITE
<EndNote><Cite
ExcludeAuth="1"><Author>Scatchard</Author><Year>1949</Year><RecNum>2211</RecNum><
DisplayText>(1949)</DisplayText><rec-number>2211</rec-number><foreign-keys><key
app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
timestamp="1542316918">2211</key></foreign-keys><ref-type name="Journal Article">17</ref-
type><contributors><author>>Scatchard, G.</author><author>Black, E.S.
</author></authors></contributors></title>The effect of salts on the isoionic and isoelectric
points of proteins</title><secondary-title>The Journal of Physical Chemistry</secondary-
title></title></title> <periodical><full-title>The Journal of Physical Chemistry</full-
title></periodical><pages>88-
99</pages><volume>53</volume><dates><year>1949</year></dates><urls></urls></record></Cite
></EndNote>] demonstrated binding of perchlorate to human plasma proteins.
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The BBDR model is dependent on the known physiological mechanisms that occur in the thyroid as they relate to iodide and perchlorate exposure as summarized in Section [ REF \_Ref482959130 \r \h ] [ ADDIN EN.CITE

is consumed and enters the plasma. From the plasma, iodide can be transported to the thyroid. In the thyroid, T3 and T4 are formed when Tg produced by the thyroid follicular cells is combined with iodide; this combined Tg-iodide is identified as "organified" or "bound" iodide in the model. Release of T4 and T3 into circulation is assumed to draw from this common thyroidal pool.

Because the model time-scale for changes in hormone levels is in weeks, while variation in TSH is known to occur on a much faster time-scale, it was deemed unnecessary to explicitly describe the production, distribution, and clearance of TSH. Instead, TSH is assumed to respond immediately to any changes in fT4, with TSH being a decreasing function of fT4: the higher the serum fT4 concentration, the lower TSH. The activity (Vmax) for NIS-mediated uptake of iodide (and perchlorate) by the thyroid and rate constants for production of T4 and T3 are then assumed to increase as TSH increases, leading to a negative feedback loop. However, the response of fT4 to this feedback is not instantaneous, since it takes time for an increased production rate to lead to increased circulating levels of fT4.

While any decrease in fT4 leads to an increase in TSH, the relationship is nonlinear with less of a change in TSH per unit change in fT4 when fT4 is in the euthyroid range than outside of it, based on observational data (Hadlow et al., 2013). Likewise, increases in fT4 lead to decreases in TSH.

In addition to regulation by TSH, hCG levels rise in early pregnancy, and this increases T4 production. (The Vmax for NIS-mediated uptake is also increased during pregnancy based on independent data [ ADDIN EN.CITE < EndNote > < Cite > < Author > Aboul-Khair</Author><Year>1964</Year><RecNum>361</RecNum><Suffix>'; see Appendix A', 1.4.11</Suffix><DisplayText>(Aboul-Khair, Crooks, Turnbull, & DisplayText), see Appendix A, 1.4.11)</br>

A, 1.4.11)
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record
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EISSN 1470-

8736</isbn><label>2140787</label><urls></urls><language>English</language></record></Cite></EndNote>], but that change is not described as an explicit function of hCG.) The stimulatory and feedback pathways are indicated by the thick, solid, red arrows (identified as "signaling pathways") in [REF\_Ref482868704 \h]. The perchlorate submodel is analogous to that of the iodide model and describes the assumed mode of action for perchlorate. That is, once perchlorate is consumed it is distributed throughout the body, including to the thyroid plasma, it binds to an anion binding site on NIS, where it competitively inhibits the binding and translocation of iodide to the cytosol. This inhibition reduces the thyroid's stores of organified iodide, and hence decreases the rate of formation and release of T3 and T4. As fT4 levels decline, the predicted concentration of TSH increases, thereby stimulating iodide uptake and thyroid hormone production, allowing for a partial compensation for the inhibition: the resulting decline in fT4 is less than it would be in the absence of this mechanism.

Subsequent sections describe key parameters in the model. Details regarding model calibration and the underlying mathematical functions of the submodels are provided in Appendix A.

#### 3.2 Summary of Key BBDR Model Parameters

The current BBDR model is built upon previous models of the thyroid for pregnancy [ ADDIN EN.CITE

<EndNote><Cite><Author>Merrill</Author><Year>2005</Year><RecNum>339</RecNum><DisplayText>(Merrill, 2005)</DisplayText><record><rec-number>339</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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43</pages><volume>83</volume><dates><year>2005</year></dates><urls><related-urls><url><style face="underline" font="default"

size="100%">http://dx.doi.org/10.1093/toxsci/kfi017</style></url></related-

urls></urls></record></Cite></EndNote>] [ ADDIN EN.CITE

<EndNote><Cite><Author>Lumen</Author><Year>2013</Year><RecNum>107</RecNum><DisplayText>(Lumen et al., 2013)</DisplayText><rec-number>107</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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timestamp="1450367396">107</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Lumen, A.</author><author>Mattie,

D.R.</author><author>Fisher, J.W.</author></authors></contributors><titles><title>Evaluation of perturbations in serum thyroid hormones during human pregnancy due to dietary iodide and perchlorate exposure using a biologically based dose-response model</title><secondary-title>Toxicological Sciences</secondary-title></title><periodical><full-title>Toxicological Sciences

341</pages><volume>133</volume><number>2</number><section>320</section><dates><year>2 013</year></dates><urls></urls><electronic-resource-num>10.1093/toxsci/kft078</electronic-resource-num></record></Cite></EndNote>] and lactation [ ADDIN EN.CITE

<EndNote><Cite><Author>Fisher</Author><Year>2016</Year><RecNum>250</RecNum><DisplayText>(Fisher, Wang, George, Gearhart, & DisplayText><record><recnumber>250</rec-number><foreign-keys><key app="EN" db-"

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468416471">250</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Fisher, J</author><author>George, Nysia I.</author>Gearhart, Jeffery M.</author><author>McLanahan, Eva

D.</author></authors></contributors><titles><title>Dietary iodine sufficiency and moderate insufficiency in the lactating mother and nursing infant: A computational perspective</title><secondary-title>PLoS ONE</secondary-title></title>ctitle>PLoS ONE</full-title>PLoS ONE</full-title>PLoS ONE

title></periodical><pages>e0149300</pages><volume>11</volume><number>3</number><dates><year>2016</year></dates><publisher>Public Library of Science</publisher><urls><related-urls><url><style face="underline" font="default"</p>

size="100%">http://dx.doi.org/10.1371%2Fjournal.pone.0149300</br>
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Table [ SEQ Table \\* ARABIC ]. Summary of Major Features of Current BBDR Modeling

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
Partition coefficient (PC) ROB: blood = (Prob_I) 0.243	The current model uses a weighted average of tissue-specific PCs for the new ROB compartment that is adjusted for NIS-mediated uptake in skin [ ADDIN EN.CITE <endnote><cite><author>Slominski</author><year>2002</year><recnum>2212</recnum><displa ytext="">(Slominski et al., 2002)<record><rec-number>2212</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1542317105">2212</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Slominski, A.</author><author>Wortsman, J.</author><author>Venkataraman, G.M.</author><author>Venkataraman, G.M.</author><author>Fisarchik, A.</author><author>Chung, J.H.</author><author>Giuliani, C.</author><author>Thornton, M.</author><author>Slugocki, G.</author><author>Tobin, D.J.</author><foreign-keys></foreign-keys>17</contributors></record></displa></cite></endnote>
Plasma protein binding of iodide = N/A	Not modeled in current BBDR model or January 2017.
Uptake into thyroid via NIS Thyroid lodide Uptake (VmaxNISF_t	The current model uses the same saturable Michaelis-Menten approach specifying a scaled Vmax (VmaxNISF_thy_I) and K <sub>m</sub> (KmNIS_I) for iodide, which is inhibited by perchlorate. Further, it uses the same K <sub>m</sub> as used in the Lumen et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Lumen</author><year>2013</year><recnum>107</recnum><displayte xt="">(2013)<record><rec-number>107</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1450367396">107</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>&gt;cauthor&gt;Lumen,</author></contributors></record></displayte></cite></endnote>

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
hy_I) = 690 nmol/h/kg <sup>0.75</sup> (non- pregnant value; this is scaled for pregnancy); NIS affinity (KmNIS_I) = 31.5 µM	A. <author>Mattie, D.R.</author> <author>Fisher, J.W.</author> <titles><title>Evaluation of perturbations in serum thyroid hormones during human pregnancy due to dietary iodide and perchlorate exposure using a biologically based dose-response model</title><secondary-title>Toxicological Sciences</secondary-title><pages>320-341</pages><volume>133 341<volume>133 341<ull> 341&lt;</ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></ull></volume></volume></titles>
lodide from T4 and T3 deiodination based upon degradation rate of T4 (KdegT4F) and T3 (KdegT3F)  KdegT4F = 1.9 x 10-5 L/hr/kg <sup>0.75</sup>	The current model decreases KdegT4F and KdegT3F by a factor of 10 compared to the Lumen et al. (2013) values of 1.9 x 10 <sup>-4</sup> and 1.7x10 <sup>-3</sup> L/hr/kg <sup>0.75</sup> , respectively. The decrease enables fitting of the model to the range of NHANES data, given uncertainties in other model parameters, with the procedure used by the EPA (see Appendix A, Section 2.2.3, for details).  The current model uses the assumption from Lumen et al. (2013) that 50% of T4 degradation is due to T3 releasing one iodide ion, while the other 50% is completely degraded, yielding four iodide ions.  The formula is: RdeiodT4=0.5(RdegT4 + 4*RdegT4) = 2.5*RdegT4 nmol/hr
KdegT3F = 1.7 x 10-4 L/hr/kg <sup>0.75</sup> Elimination to urine (CLFUI) = 0.0653 L/hr/kg <sup>0.75</sup> (non- pregnant value; this is scaled for pregnancy)	The current model begins with non-pregnant clearance of CLFUI = 0.0653 L/hr/kg <sup>0.75</sup> and increases over pregnancy by an empirical relationship between gestational age and iodide clearance (see Appendix A). Lumen et al. (2013) assumed that iodine clearance was a function of iodine intake, with the relationship fitted to data from Dunn & Delange [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Dunn</author><year>2005</year><recnum>711</recnum>CDisplayText&gt;(2005)<record><rec-number>711</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206430">7711</key></foreign-keys><ref-type name="Edited Book">28</ref-type><contributors><authors><author>Dunn, J. T.</author><author>Delange, F. M.</author></authors><secondary-authors><author>Braverman, L. E.</author><author>Utiger, R. D.</author></secondary-authors></contributors><tititle>lodine deficiencyWerner &amp; amp; Ingbar &amp; apos; s the thyroid: a fundamental and clinical text Yitile&gt; Secondary-title&gt; Secondary-title&gt; Secondary-authors&gt; Contributors&gt; Contributors Con</tititle></record></cite></endnote>

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
	Perchlorate Submodel
PC ROB:serum = (Prob_P) 0.558	The current model uses a weighted average of tissue-specific PCs for the new ROB compartment and includes NIS uptake in skin and mammary glands. The inclusion of the skin and mammary NIS increased the PC by 80%. Lumen et al. (2013) reported PCs for perchlorate for individual tissues that ranged from 0.13 to 0.56.
Plasma protein binding (Km_Bp) = 0.181 μM	Included in the current model; not included in Lumen et al. (2013).
Distribution into red blood cells (RBCs), RBC:serum PRBC_P = 0.8	Included in the current model; not included in Lumen et al. (2013).
Uptake into thyroid via NIS (VmaxNISF_t hy_P) = 650 nmol/h/kg <sup>0.75</sup> (non-pregnant value; this is scaled for pregnancy); (KmNIS_P) = 0.489 µM	The current model uses a K <sub>m</sub> for perchlorate binding to the NIS (KmNIS_P) that is 3-fold lower than Lumen et al. (2013) (i.e., a 3-fold higher affinity). Specifically, K <sub>m</sub> is the 2.5 <sup>th</sup> percentile lower confidence limit of the population median based upon EPA reanalysis of Greer et al. (2002). The value (50 <sup>th</sup> percentile = 0.73 μM) is similar to that obtained from re-analysis of <i>in vitro</i> binding data, 0.59 μM [ ADDIN EN.CITE <endnote><cite><author>Schlosser</author><year>2016</year><recnum>293</recnum><display text="">(Schlosser, 2016)<record><rec-number>293</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470167034">293</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Schlosser, P. M.</author><author></author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></contributors></record></display></cite></endnote>

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
Elimination to urine (CLFUP) = 0.105 L/hr/kg <sup>0.75</sup> (non-pregnant value; this is scaled for pregnancy)	The current model uses a higher urinary clearance compared to the clearance used in the Lumen et al. (2013) model (i.e., 0.05 L/hr/kg <sup>0.75</sup> ). The current model's clearance rate, CLFUP = 0.105 L/hr/kg <sup>0.75</sup> , is for non-pregnant women and based on EPA reanalysis of Greer et al. (2002); it is the 2.5th percentile of the sampling distribution for the population median (see Appendix A), with the 50th percentile being 0.122 L/hr/kg <sup>0.75</sup> . Urinary clearance increases over pregnancy by an empirical relationship between gestational age and glomerular filtration (see Appendix A). Lumen et al. (2013) fit a perchlorate clearance to Chilean [ ADDIN EN.CITE <endnote><cite><author>Téllez           Téllez         Téllez<!--</td--></author></cite></endnote>
Production of T4 (KProdT4F) = 5.83*10-7 L/hr/kg <sup>.75</sup> (non- pregnant value; this is scaled for pregnancy)	The current model uses a baseline first-order constant calibrated to NHANES 2007-2012 median, 10 <sup>th</sup> , or 90 <sup>th</sup> percentile non-pregnant data; the KProdT4F value for the median NHANES calibration is 5.83*10 <sup>-7</sup> L/hr/kg <sup>0.75</sup> , which is lower than 2.45 x 10 <sup>-6</sup> used by Lumen et al. (2013) based on fitting to data of Nicoloff, Low, Dussault, & Fisher [ ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;Ex</cite></endnote>

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
	num> <language>English</language> ]. This value is scaled in pregnancy with increasing KProdT4 through GW 16 (peak occurring ~ GW 9) based upon placental hCG increase over this time according to the linear relationship from Glinoer [ ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;ExcludeAuth="1"</cite></endnote>
TSH Feedback Loop	The current model incorporates a TSH feedback loop that adjusts the Vmax for uptake into the thyroid and production rate constants for T4 (KProdT4F) and T3 as TSH falls below or rises above its calibration point (i.e., the population percentile of TSH assumed to be consistent with the population percentiles of other thyroid hormones used for a particular calibration), in inverse relationship to corresponding increases or decreases in fT4. For example, when the model is calibrated to match median T4, T3, fT4, and fT3 levels, the median TSH level is selected as the calibration point, and regulation occurs as TSH goes above or below this level, in response to variation in fT4. The relationship between TSH and fT4 is explained by data in Hadlow et al. [ADDIN EN.CITE \text{EndNote} \text{Cite} \text{EndNote} \text{Cite} \text{ExcludeAuth="1"} \text{Author} \text{Hadlow} \text{Author}

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
	T4 Distribution
T4 Volume of Distribution (VDFT40) = 0.162 L/kg (non- pregnant value; this is scaled for pregnancy)	The current model has a baseline value (0.162 L/kg) for non-pregnant women that is scaled to decrease linearly with gestation time through GW 16 based upon a study that measured the distribution of radio-labeled T4 in pregnant and non-pregnant women. See Appendix A for additional details. The absolute volume of distribution is VDT4 = VDFT40*BW, so the decrease in the relative volume is somewhat offset by increasing BW. Lumen et al. (2013) used an estimate of 0.11 L/kg for the third trimester.
Free (fT4) to tT4 Ratio (FRT40) = 1.03 x 10 <sup>-4</sup> (non- pregnant value; this is scaled for pregnancy)	The current model uses a baseline value of 1.03 x 10 <sup>-4</sup> for median non-pregnant women and then estimates the ratio during pregnancy based on an empirical relationship derived from the ratio by gestational week. This relationship is based upon six studies as shown in Appendix A. Lumen et al. (2013) estimated that the fraction of total T4 converted to fT4 was approximately 9.0 x 10 <sup>-5</sup> .
KdegT4F = 1.9*10 <sup>-5</sup> L/hr/kg <sup>0.75</sup>	T4 Degradation  T4 degradation in the current model was selected to allow model calibration to the range of thyroid hormone values in NHANES as described above in the iodide submodel section.
Removal of placental and fetal compartment s and condensing of maternal system to three compartment s: serum, thyroid, and ROB	Additional Adjustments for Early Pregnancy  Simplification of BBDR model structure from Lumen et al. (2013); see [ REF _Ref482868704 \h ] and related adjustments to PCs, compartment volume, and flows.
fT4: total T4 median serum concentration ratio,	A polynomial regression for fT4 fit to five datasets describing initial increase and then decrease of fT4 during early pregnancy was divided by a Hill equation regression fit to five datasets describing the increase in total T4 during the first trimester. The resulting time-dependent fT4:T4 ratio (change vs. non-pregnant), which occurs due to increases in serum binding proteins, results in a decrease of approximately 30% in the fT4:T4 ratio during early pregnancy.

Key Elements and Parameter Values for Current Model	Sources, Derivation, and Differences from January 2017 (3 <sup>rd</sup> trimester) Model (based on Lumen et al. (2013))
modeled to vary during early pregnancy	
Note: This table provided in Appe	does not summarize all the BBDR model parameters. A more complete accounting of all model parameters is ndix A.

## 3.3 Overview of Key Model Attributes That May Impact Thyroid Hormone Production

As described in [ REF \_Ref483397647 \h ], the baseline non-pregnancy rate for T4 synthesis was developed by fitting NHANES 2007-2012 median, 10<sup>th</sup>, and 90<sup>th</sup> percentile data for non-pregnant women, with adjustment for increasing T4 production and varying fT4:T4 ratios during early pregnancy as seen in a number of studies (see Appendix A and related references (e.g., La'ulu & Roberts, 2011; Medici et al., 2012; Soldin et al., 2004; Stricker et al., 2007; Yan et al., 2011; Zhang et al., 2016)). There are several regulatory controls in the BBDR model that can impact T4 and T3 production, such as the inclusion of hCG in the model, varying iodine intake, and also the TSH feedback loop. The impact of hCG on T4 production is discussed in [ REF \_Ref483397647 \h ]. Further details regarding how low iodine intake and the TSH feedback loop were considered in calibrating the model are presented below.

#### 3.3.1 Low lodine Intake

Plots of NHANES 2007-2012 data for non-pregnant women demonstrated little relationship between iodine intake and fT4, except possibly for iodine intake levels below 75 μg/day. Given limited data in this range, the EPA has used the relationship between thyroidal iodide stores (mg) and iodine intake, which shows depletion of fT4 at iodine intake levels below 100 µg/day [ ADDIN EN.CITE <EndNote><Cite><Author>Delange</Author><Year>2000</Year><RecNum>682</RecNum><Dis playText>(Delange, 2000)</DisplayText><record><rec-number>682</rec-number><foreignkeys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206429">682</key></foreign-keys><ref-type name="Edited Book">28</reftype><contributors><author>Delange, F. M.</author></author></secondaryauthors><author>Braverman, L. E.</author><author>Utiger, R. D.</author></scondaryauthors></contributors><title>Iodine deficiency</title></title></pages>295-316</pages><edition>8th</edition><dates><year>2000</year></dates><pub-location>Philadelphia, PA</pub-location><publisher>Lippincott, Williams, and Wilkins</publisher><label>1247276</label><urls></urls><language>English</language></record> </Cite></EndNote>]. This relationship was used as part of a parameter calibration procedure to simulate population percentiles of fT4, T4, T3, and TSH identified from NHANES data (see Appendix A, Figures 54-56).

#### 3.3.2 TSH Feedback Loop

TSH increase in response to decreases in T4 has been captured in numerous studies that document the inverse relationship between these hormones. A compilation of data describing this relationship (Hadlow et al., 2013) has been analyzed by the EPA in Appendix A to enable TSH feedback loop incorporation into the BBDR model. The following equation elevates TSH in response to a lowered concentration of fT4, as might arise from goitrogen exposure or lowered iodine intake:

$$TSH(CfT4) = \frac{TSHtar \times TSHhad(CfT4)}{TSHhad(CfT4tar)}$$

In this equation, TSHtar is the "target" level of TSH in a population (percentile) or an individual corresponding to a given fT4 concentration (e.g., the median) in that population percentile or individual, and TSHhad(CfT4) is the empirical function fit to the collected fT4 and TSH data by Hadlow et al. (2013). For example, the median TSH concentration from NHANES data is associated with the median fT4 for the same population; so these two would be CfT4tar and TSHtar for the presumed median individual. Using that pair of values in the TSH equation above scales TSHhad(CfT4) such that when CfT4 = CfT4tar, TSH(CfT4) = TSHtar. This combination of target hormone levels is the set point where both are stable and TSH will not increase or decrease thyroid activity beyond the corresponding fT4 value. When fT4 concentration decreases TSH(CfT4) increases. The ratio, TSH(CfT4)/TSHtar, determines the degree to which the BBDR model will upregulate iodide uptake and T4 production (i.e., if TSHhad(CfT4) > TSHhad(CfT4tar)). The result is that the change in fT4 from the set point (CfT4tar) will be less than would occur in the absence of TSH feedback, but there will still be some change, hence a dose-response due to perchlorate.

The BBDR model achieves this feedback regulation by tuning three parameters, NIS Vmax (VmaxNISF thy I; i.e., modulation of number of symporter sites), KProdT4 (modulation of T4 synthesis rate), and KProdT3 (modulation of T3 synthesis rate). These three parameters are assumed to vary as (TSH(CfT4)/TSHtar)pTSH, where the coefficient pTSH determines the strength of TSH stimulation. However, to estimate the value of pTSH one would need measurements of TSH and the thyroid-specific parameters that TSH is presumed to regulate—the Vmax for iodine uptake and the T4 and T3 production rate constants (the relationship between T4 and T3 production and thyroid iodide stores). Such data are not available for humans, so a nominal value of pTSH = 1 was tested. Using this value, model predictions appear to be consistent with median NHANES data for nonpregnant women (see Section 3.2 in Appendix A). Model runs with pTSH = 1 were used in evaluating the shift in the hypothyroxinemic population (Section [ REF Ref517191523 \r \h ]). A lower bound of pTSH = 0.398, which is the ratio of a median value for TSH from NHANES (non-pregnant women) to the 97.5th percentile value from NHANES (non-pregnant women; details in Appendix A) was used for the dose-response analysis linking changes in fT4 to neurodevelopmental outcomes. This is based on an assumption that sensitive individuals with high TSH and average fT4 exist because the stimulus strength of TSH is proportionally weaker. TSH increases can drive corresponding increases in iodide uptake and T4 and T3 production as long as the iodide supply is sufficient. With decreasing iodine intake, colloidal iodide stores are modeled to decline, which ultimately impairs the thyroid's response to elevations in TSH. Using pTSH = 0.398 for the neurodevelopmental analysis ensures that the EPA is protecting a potentially sensitive population from adverse neurodevelopmental impacts. Using pTSH = 1 is more appropriate for an analysis of the shift in the hypothyroxinemic population given that the analysis is based on the full distribution of

fT4 levels and the approach aims to avoid the precursor effect to neurodevelopment (i.e., hypothyroxinemia).

#### 3.4 Results of the BBDR Model

The BBDR model can predict serum thyroid hormone levels at specific gestational weeks, given levels of iodine intake and doses of perchlorate. [ REF Ref482950735 \h ] displays plots that illustrate specific outputs of the model. The upper- and lower-bound simulations in this figure were developed using corresponding parameter sets for the median, upper, and lower bounds of fT4, T4, fT3, T3, and TSH. They are not true bounds, but are instead estimates for fT4 or TSH given NHANES data with median, 2.5th, and 97.5th percentiles for fT4, TSH, etc.

[ REF Ref482951228 \h ] and [ REF Ref483557363 \h ] summarize output of the BBDR model for various doses of perchlorate and gestational weeks and resulting fT4 and TSH levels at the median iodine intake (170 µg/day) and a low iodine intake, respectively. In both cases the model is calibrated to match median thyroid hormone levels given an ingestion of 90 µg/day of iodine—the median population as described in Appendix A. However, for [ REF Ref482951228 \h ] and [ REF Ref483557363 \h ], a weak level of TSH signaling strength, pTSH = 0.398, was used to illustrate the more sensitive dose-response from using that estimated lower bound. [ REF Ref512496933 \h ] and REF Ref512496934 \h ] report BBDR results using a median level of TSH signaling strength, or pTSH = 1. For analysis of a low-iodine population an iodine intake level of 75 μg/day was selected. This intake level represents between the 15th and 20th percentile of the population distribution of estimated iodine intake from NHANES. The EPA considered conducting the analysis at the lower iodine intake level of 50 µg/day, which represents approximately the 5th percentile of the NHANES distribution. However, at 50 µg/day of iodide, the BBDR model predicts TSH levels that would be elevated to within the clinically hypothyroid range before exposure to any perchlorate (TSH ranges

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For purposes of this analysis the EPA evaluated the American Thyroid Association's (ATA) 2017 recommendations for defining hypothyroidism [ ADDIN EN.CITE

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N.</author><author>Brent, G. A.</author><author>Brown, R. S.</author><author>Chen.

H.</author>Cauthor>Dosiou, C., </author>Cauthor>Sullivan,

S.</author></authors></contributors></title>>2017 Guidelines of the American Thyroid Association for the diagnosis and management of thyroid disease during pregnancy and the postpartum</title><secondary-title>Thyroid</secondary-title></title><periodical><fulltitle>Thyroid</full-title></periodical><pages>315-

<sup>389&</sup>lt;/pages><volume>27</volume><number>3</number><dates></ear>2017<//ear></dates></url s></record></Cite></EndNote>]. Specifically, the ATA recommends: "in the pregnancy setting, maternal hypothyroidism is defined as a TSH concentration elevated beyond the upper limit of the pregnancyspecific reference range." ATA goes on to state, in the absence of population- and trimester-specific references ranges defined by a provider's institute or laboratory, TSH reference ranges should be obtained from similar patient populations. From their recommended studies with trimester-specific data on a U.S. population, Lambert-Meserlian et al. [ ADDIN EN.CITE < EndNote > < Cite

between 4.51 and 5.41 mIU/L at zero dose of perchlorate when evaluating GW 12 or GW 13). In contrast, at 75  $\mu$ g/day iodine the BBDR-modeled concentrations of scrum fT4 and TSH are significantly reduced from the population median but are still within the euthyroid range. This comports with the goal of evaluating perchlorate effects on neurodevelopment when starting with a population of pregnant women who are borderline hypothyroxinemic.

As described in Sections [ REF \_Ref488740188 \n \h ] and [ REF \_Ref482958740 \r \h \\* MERGEFORMAT ], the identified epidemiological literature led the EPA to select GW 12 and GW 13 for analysis. Results from the BBDR model output are subsequently summarized for these gestational weeks.

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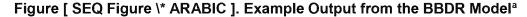
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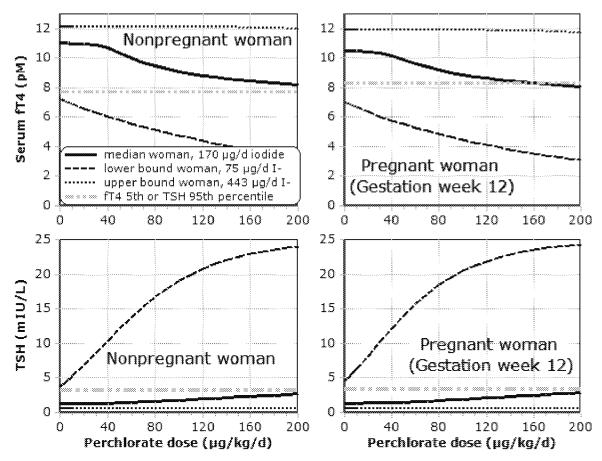
Goldman, Jane</author><author>Malone, Fergal D</author><author>Porter, T

Flint</author><author>Nyberg, David A</author><author>Bernstein,

Peter</author></authors></contributors><titles><title>First-and second-trimester thyroid hormone reference data in pregnant women: a FaSTER (First-and Second-Trimester Evaluation of Risk for aneuploidy) Research Consortium study</title><secondary-title>American Journal of Obstetrics and Gynecology</secondary-title></title><periodical><full-title>American journal of obstetrics and gynecology</full-title></periodical><pages>62-

e1</pages><volume>199</volume><number>1</number><dates><year>2008</year></dates><publisher >Elsevier</publisher><isbn>0002-9378</isbn><urls></urls></record></Cite></EndNote>] is the largest U.S.-based population with a reference range upper bound of 3.37 mIU/L for the first trimester (and 3.35 mIU/L for the second trimester). Therefore, these values were used to compare to BBDR output TSH values in the first trimester (or second trimester in cases of GW 15 and GW 16) to determine the presence of hypothyroidism.





<sup>a</sup> BBDR model simulations using parameter sets calibrated for median (black, solid curve), lower-bound (black, dashed curve), or upper-bound (black, dotted curve) fT4 levels, as described in Appendix A. TSH levels used for calibration were assumed to be inversely correlated with fT4; hence, the lower-bound level of TSH is associated with upper-bound fT4 and vice versa. Median and 90 μg/day percentile estimated iodine (I-) intake levels (170 and 443 μg/day, respectively) were used for simulations of median and upper-bound women. An iodine intake of 75 μg/day was used for the lower-bound woman, as described in the preceding text. Grey dash-dot lines show the 5<sup>th</sup> percentile of the estimated fT4 and upper bound for TSH, defining those boundaries of the euthyroid range.

Table [ SEQ Table \\* ARABIC ]. Predicted fT4 and TSH Concentrations at Various Doses of Perchlorate ( $\mu$ g/kg/day) at Specified Gestational Weeks for 170  $\mu$ g/day lodine Intake with pTSH = 0.398<sup>a</sup>

Perchlorate Dose	fT4 (pmol/L) (% Change from 0 Dose)		TSH (mIU/L) (% Change from 0 Dose)		
(µg/kg/day)	GW 12	GW 13	GW 12	GW 13	
0	10.67	10.64	1.35	1.35	
1	10.66	10.64	1.35	1.35	
	(-0.04%)	(-0.05%)	(0.04%)	(0.04%)	
2	10.66	10.63	1.35	1.35	
	(-0.09%)	(-0.10%)	(0.07%)	(0.08%)	
3	10.65	10.63	1.35	1.35	
	(-0.13%)	(-0.14%)	(0.11%)	(0.12%)	
4	10.65	10.62	1.35	1.35	
	(-0.17%)	(-0.19%)	(0.15%)	(0.17%)	
5	10.64	10.62	1.35	1.35	
	(-0.22%)	(-0.24%)	(0.18%)	(0.21%)	
10	10.62	10.59	1.35	1.36	
	(-0.46%)	(-0.51%)	(0.39%)	(0.45%)	

<sup>&</sup>lt;sup>a</sup> Additional details can be found in Appendix A.

Percentages in the table may not mirror independent calculations due to rounding.

Table [ SEQ Table \\* ARABIC ]. Predicted fT4 and TSH Concentrations at Various Doses of Perchlorate ( $\mu g/kg/day$ ) at Specified Gestational Weeks for 75  $\mu g/day$  lodine Intake with pTSH = 0.398<sup>a</sup>

Perchlorate Dose	fT4 (pmol/L) GW 12	(% Change from 0 Dose) GW 13	TSH (mIU/L) GW 12	(% Change from 0 Dose) GW 13
(µg/kg/day)	***			***************************************
0	8.85	8.84	1.95	1.95
1	8.78	8.77	2.00	2.01
	(-0.80%)	(-0.79%)	(2.94%)	(2.92%)
2	8.71	8.71	2.06	2.07
	(-1.56%)	(-1.54%)	(5.96%)	(5.92%)
3	8.65	8.64	2.12	2.13
	(-2.29%)	(-2.26%)	(9.07%)	(9.01%)
4	8.59	8.58	2.18	2.19
	(-2.98%)	(-2.95%)	(12.26%)	(12.18%)
5	8.53	8.52	2.25	2.25
	(-3.64%)	(-3.60%)	(15.53%)	(15.44%)
10	8.27	8.27	2.59	2.60
	(-6.56%)	(-6.52%)	(33.06%)	(32.94%)

<sup>&</sup>lt;sup>a</sup> Additional details can be found in Appendix A.

Percentages presented in the table may not mirror independent calculations due to rounding.

Table [ SEQ Table \\* ARABIC ]. Predicted fT4 and TSH Concentrations at Various Doses of Perchlorate ( $\mu$ g/kg/day) at Specified Gestational Weeks for 170  $\mu$ g/day lodine Intake with pTSH = 1<sup>a</sup>

Perchlorate Dose	fT4 (pmol/L) (% Change from 0 Dose)		TSH (mIU/L) (% Change from 0 Dose)		
(µg/kg/day)	GW 12	GW 13	GW 12	GW 13	
0	10.67	10.64	1.35	1.35	
1	10.66	10.64	1.35	1.35	
	(-0.02%)	(-0.03%)	(0.02%)	(0.02%)	
2	10.66	10.64	1.35	1.35	
	(-0.05%)	(-0.05%)	(0.04%)	(0.05%)	
3	10.66	10.63	1.35	1.35	
	(-0.07%)	(-0.08%)	(0.06%)	(0.07%)	
4	10.66	10.63	1.35	1.35	
	(-0.10%)	(-0.11%)	(0.08%)	(0.09%)	
5	10.65	10.63	1.35	1.35	
	(-0.12%)	(-0.14%)	(0.10%)	(0.12%)	
10	10.64	10.61	1.35	1.35	
	(-0.25%)	(-0.28%)	(0.21%)	(0.24%)	

<sup>&</sup>lt;sup>a</sup> Additional details can be found in Appendix A.

Percentages in the table may not mirror independent calculations due to rounding.

Table [ SEQ Table \\* ARABIC ]. Predicted fT4 and TSH Concentrations at Various Doses of Perchlorate ( $\mu g/kg/day$ ) at Specified Gestational Weeks for 75  $\mu g/day$  lodine Intake, pTSH = 1<sup>a</sup>

Perchlorate Dose	fT4 (pmol/L)	(% Change from 0 Dose)	TSH (mIU/L)	(% Change from 0 Dose)
(µg/kg/day)	GW 12	GW 13	GW 12	GW 13
0	8.85	8.84	1.95	1.95
1	8.81	8.80	1.98	1.99
	(-0.47%)	(-0.46%)	(1.68%)	(1.67%)
2	8.77	8.76	2.01	2.02
	(-0.92%)	(-0.91%)	(3.38%)	(3.36%)
3	8.73	8.72	2.05	2.05
	(-1.36%)	(-1.34%)	(5.12%)	(5.08%)
4	8.69	8.69	2.08	2.09
	(-1.78%)	(-1.76%)	(6.88%)	(6.83%)
5	8.66	8.65	2.11	2.12
	(-2.20%)	(-2.17%)	(8.67%)	(8.61%)
10	8.49	8.48	2.30	2.30
	(-4.11%)	(-4.08%)	(18.03%)	(17.93%)

<sup>&</sup>lt;sup>a</sup> Additional details can be found in Appendix A.

Percentages in the table may not mirror independent calculations due to rounding.

[ REF \_Ref482950735 \h ] shows that the BBDR model predicts very little difference in non-pregnant and first-trimester response to perchlorate. This likely occurs because the half-life of (organified)

iodine in the adult thyroid is several months (Dunn & Dunn, 2000, and Brabant et al., 1992, both as cited in Greer et al., 2002); hence the availability of thyroidal iodine in the first-trimester pregnant woman is determined to a very large extent by her nutrition and perchlorate exposure several years preceding pregnancy. Increased urinary clearance of both iodide and perchlorate during pregnancy, described by the model, have the conflicting effect of both reducing iodide availability in the mother and reducing perchlorate blood concentration, so these two components of pregnancy-related changes will cancel each other out to some extent. The similarity in dose-response also reflects the assumption that the small placental and fetal compartments, with the fetal thyroid only just becoming active at the end of the first trimester, will have very little effect on maternal anion and thyroid hormone concentrations, which led to the choice of a simplified model structure ([ REF Ref482868704 \h ]). Further, there is little predicted change in the percentage drop in fT4 or percentage increase in TSH between gestational weeks due to perchlorate exposure, although baseline TSH and total T4 levels do vary significantly with gestation. However, there is a significant drop in fT4 and increase in TSH with decreased iodine intake in the absence of perchlorate exposure. Comparing the predicted fT4 and TSH levels at zero dose of perchlorate in [ REF Ref482951228 \h \\* MERGEFORMAT ] (170 μg/day iodine intake) and [ REF Ref483557363 \h \\* MERGEFORMAT ] (75 μg/day iodine intake) or [ REF Ref512496933 \h ] and [ REF Ref512496934 \h ] (a 56% drop in iodine intake from the median level), shows approximately a 16% to 17% drop in fT4 and approximately a 44% to 58% increase in TSH, depending on the gestational week of interest.

On the contrary, the BBDR model predictions for perchlorate's effect on fT4 are much less than the impact of decreased intake of iodine from 170 to 75  $\mu$ g/day on fT4. For example, in the low-iodine intake individuals there is predicted to be about a 6% to 6.5% drop in fT4 with 10  $\mu$ g/kg/day exposure to perchlorate. This result is consistent with the underlying Greer et al. (2002) data in human test subjects, which showed a NOAEL for perchlorate inhibition of radioiodide uptake by the thyroidal NIS of 7  $\mu$ g/kg/day.

#### 3.5 Uncertainties Associated with the BBDR Model

The BBDR model allows for the evaluation of perchlorate effects on circulating levels of fT4 based upon interaction with the thyroidal NIS before and during early pregnancy at various iodine intakes. Being biologically based, the model aims to simulate a complex physiological process that has been relatively well studied, the HPT axis. Interest in how perchlorate may perturb this system has spurred development of this modeling framework to predict such influences. Given the complexity of the model there can be uncertainties in many different aspects, ranging from structural and functional relationships to specific parameter values for early pregnancy. McLanahan et al. [ ADDIN EN.CITE <EndNote><Cite

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E.</author><author>Campbell, J. L.</author><author>Fisher, J.

W.</author></authors></contributors><auth-address>Department of Environmental Health Sciences, University of Georgia, Athens, Georgia 30602-2102, USA.</auth-address><title>Competitive inhibition of thyroidal uptake of dietary iodide by perchlorate does not describe perturbations in rat serum total T4 and TSH</title><secondary-title>Environmental Health Perspectives</secondary-

effort.

title></title>><periodical><full-title>Environmental Health Perspectives</fulltitle></periodical><pages>731-8</pages><volume>117</volume><number>5</number><edition>2009/05/30</edition><keywords ><keyword>Animals</keyword><keyword>Biological Transport/ drug effects</keyword><keyword>Iodides/ metabolism</keyword><keyword>Male</keyword><keyword>Perchlorates/ toxicity</keyword><keyword>Rats</keyword><keyword>Thyroid Gland/ drug effects/ metabolism</keyword><keyword>Thyrotropin/ blood</keyword><keyword>Thyroxine/ blood</keyword><keyword>Water Pollutants, Chemical/ toxicity</keyword></keywords><dates><year>2009</year><pub-dates><date>May</date></pubdates></dates><isbn>1552-9924 (Electronic)&#xD:0091-6765 (Linking)</isbn><accessionnum>19479014</accession-num><urls></urls><custom2>2685834</custom2><electronic-resourcenum>10.1289/ehp.0800111</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>] examined an earlier version of the model against data in rats exposed to high-dose perchlorate with the conclusion that the model under-predicted the degree of T4 decline and TSH rise. Given differences across species in thyroid storage of iodide and set points, and the effects of high versus lower dose perchlorate, the previous validation effort does not necessarily reflect on the predictive capacity of the current model.

On the pharmacokinetic side of the model, the submodels for iodide and perchlorate have been extensively developed and revised in the past (Merrill et al., 2005; Clewell et al., [ ADDIN EN.CITE <EndNote><Cite

However, it does highlight the lack of useful calibration and validation data for this BBDR modeling

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A.</author><author>Gearhart, J. M.</author><author>Robinson, P. J.</author><author>Sterner, T. R.</author><author>Mattie, D. R.</author><author>Clewell, H.

J.</author></contributors></title>Perchlorate and radioiodide kinetics across life stages in the human: using PBPK models to predict dosimetry and thyroid inhibition and sensitive subpopulations based on developmental stage</title><secondary-title>Journal of Toxicology and Environmental Health, Part A: Current Issues</secondary-title>dalt-title>J Toxicol Environ Health A</alt-title><short-title>Journal of Toxicology and Environmental Health, Part A: Current Issues</short-title></title></periodical><full-title>Journal of Toxicology and Environmental Health, Part A: Current Issues
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428</pages><volume>70</volume><number>5</number><dates><year>2007</year></dates><isbn >ISSN 1528-7394</isbn><label>756734</label><urls><related-

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num><language>English</language></record></Cite></EndNote>]; Lumen et al., 2013), although, as pointed out in McLanahan, White, Flowers, & Schlosser [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>McLanahan</Author><Year>2014</Year><RecNum>255</RecNum>< DisplayText>(2014)</DisplayText><record><rec-number>255</rec-number><foreign-keys><key

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hormone levels.

app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468516804">255</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>McLanahan, E</author><author>White, Paul</author><author>Flowers, Lynn</author><author>Schlosser, Paul M.</author></authors></contributors><title>>The use of PBPK models to inform human health risk assessment: Case study on perchlorate and radioiodide human lifestage models</title><secondary-title>Risk Analysis</secondary-title></title><periodical><full-title>Risk Analysis</full-title></periodical><pages>356-366</pages><volume>34</volume><number>2</number><keywords><keyword>Human</keyword ><keyword>PBPK model</keyword><keyword>riadioiodide</keyword>ri sk assessment</keyword></keywords><dates><year>2014</year></dates><isbn>1539-6924</isbn><urls><related-urls><url><style face="underline" font="default" size="100%">http://dx.doi.org/10.1111/risa.12101</style></url></related-urls></url></rel resource-num>10.1111/risa.12101</electronic-resource-num></record></Cite></EndNote>], there are very few calibration data for perchlorate kinetics in humans, particularly at the life stage of interest. This may be of particular importance given that available biomarker data to relate perchlorate external dose to internal serum concentration are highly variable and uncertain. For example, the Lumen et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Year>2013</Year><RecNum>107</RecNum><DisplayText>(2013)</DisplayT ext><record><rec-number>107</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1450367396">107</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Lumen, A.</author><author>Mattie, D.R.</author><author>Fisher, J.W.</author></authors></contributors><title>Evaluation of perturbations in serum thyroid hormones during human pregnancy due to dietary iodide and perchlorate exposure using a biologically based dose-response model</title><secondary-title>Toxicological Sciences</secondarytitle></titles><periodical><full-title>Toxicological Sciences</full-title></periodical><pages>320-341</pages><volume>133</volume><number>2</number><section>320</section><dates><year>2 013</year></dates><urls></urls><electronic-resource-num>10.1093/toxsci/kft078</electronicresource-num></record></Cite></EndNote>] model was based on pregnant women living in a Chilean city with high levels of perchlorate in drinking water (Telléz Telléz et al., 2005). In that study, the standard deviation of the concentration of perchlorate in urine was similar to, or in one instance, greater than the mean urinary perchlorate concentration, indicating extreme variability. Furthermore, data used in the calibration from the Chilean city in Telléz Telléz et al. (2005) did not reflect a low-iodine intake population. Urinary iodide for the 37 subjects at the second pre-natal visit was, on average, 217 μg/L, with a standard deviation of 109 μg/L. Other toxicokinetic factors that may affect the concentration of perchlorate at the thyroidal NIS include the extent of protein binding, the perchlorate urinary excretion rate, and the partitioning into non-thyroidal compartments (i.e., the ROB compartment). However, of these three, only perchlorate urinary clearance affects the long-term average concentration of perchlorate at the thyroidal NIS predicted by the BBDR model, and given the half-life of iodine in the thyroid, it is this long-term average that determines the effect on thyroid

At steady state, or under long-term continuous exposure conditions, the (average) free concentration of perchlorate in plasma is independent of the fraction bound to protein. An increase in the concentration of binding proteins will result in a temporary decrease in the free perchlorate

concentration, resulting in a lower clearance rate to urine. Due to the lower clearance, there will then be an increase in total blood perchlorate concentration over a relatively short time (since exposure is ongoing) with urine clearance increasing in proportion to the free concentration and the amount bound to proteins also increasing, until steady state is again achieved with the free concentration returning to the same level as before the change in plasma proteins. The amount bound to proteins will then be increased in proportion to the increase in binding protein concentration. Likewise, a drop in binding proteins will cause a short-term increase in free perchlorate concentration, but this concentration will quickly return to the same steady-state level as before the protein level change as the amount bound to protein declines. Since such short-term fluctuations have minimal impact on the organified iodide in the thyroid, they are not significant to the overall dose-response.

There is some uncertainty due to the assumption that the concentration (and accumulation) of iodide and thyroid hormones in the placenta and fetus, and fetal metabolism of thyroid hormones, is not significantly different from the rest of the maternal body. For example, if the fetus and placenta are 2% of total maternal BW at a given point in gestation, and the amount of iodide accumulation or T4 metabolism in them is actually twice the average for the rest of the mother's body, the impact of neglecting that difference would only be a change of 2% in predicted maternal concentrations. The total rate of iodide accumulation or T4 metabolism in the fetus (not the rate per kg fetal BW) would have to be a significant fraction of the total maternal ingestion rate for iodine, or total maternal T4 metabolism and clearance, in order for the differences in the fetus to directly impact the maternal levels. (The model already accounts for the effects of hCG and changes in maternal physiology.) Hence the uncertainty from not including explicit placental and fetal compartments is determined by the uncertainty in the assumption that iodide accumulation and thyroid hormone metabolism are limited in early gestation.

The effect of the interaction between perchlorate and thiocyanate on thyroid hormone status seen in two epidemiology studies (Steinmaus et al., 2007; 2013) has been assumed to be due to a combined inhibitory effect on the NIS, but thiocyanate is also a competitive substrate for the thyroid peroxidase, a mechanism independent of perchlorate. While the BBDR model does not explicitly consider other goitrogens (e.g., thiocyanate, nitrate), simulating results for a mother with low iodine intake and low baseline fT4 may help to address the sensitivity of individuals with high exposures to other goitrogens.

The composite partition coefficient for perchlorate in ROB of 0.558 is considerably higher than that for iodide (0.243). Both perchlorate and iodide are subject to NIS-mediated uptake into mammary glands and skin, which is accounted for in the relevant partition coefficients. The higher PC for perchlorate (80% increase in perchlorate ROB coefficient) reflects the higher ratio of Vmax/K<sub>m</sub> for NIS-mediated transport into mammary glands and skin for perchlorate versus iodide. The BBDR model assumes this type of distributional impact of NIS uptake is plausible; however, it has not been evaluated using empirical data. But as noted above, this distribution will not impact the long-term average free serum concentration of perchlorate, which determines the effect on thyroid hormones.

A key pharmacokinetic factor is the perchlorate renal elimination rate, which is an influential parameter in the fitting procedure of Lumen et al. (2013), who attempted to match BBDR model predictions to biomarker measurements in pregnant Chilean women. However, that estimation assumed that perchlorate exposure was only in drinking water, with none coming from diet. When exposure is significantly under-estimated, a low level of urinary clearance would be required to match an observed serum concentration since the rate of absorption into the body is under-estimated. The

current modeling effort for early pregnancy uses an excretion rate that is more than twice that of the third-trimester Lumen et al. (2013) model. Merrill et al. (2005) estimated a urinary excretion rate (0.11 L/hr/kg<sup>0.75</sup>) that is similar to what is used presently to represent the non-pregnant state, and this was based upon mean urinary perchlorate concentrations measured in the Greer et al. (2002) study. Evaluation of the variability in perchlorate urinary excretion in available human datasets is presented in Appendix A.

Toxicodynamic aspects such as competitive inhibition at the NIS, depletion of iodide stores under different iodine intake levels and physiological states, and the ability of the TSH feedback loop to compensate for perturbations in thyroid function each have their own uncertain features. A recent version of the BBDR model recalibrated for greater perchlorate affinity of the NIS (Schlosser, 2016) was able to simulate the dose-response for perchlorate's effect on radioiodide uptake seen in Greer et al. (2002). That correspondence builds confidence in the BBDR model's ability to simulate current data informing the user of perchlorate toxicokinetics and toxicodynamics in humans. However, the Greer et al. (2002) study was short-term, and did not evaluate the iodine intake status, smoking status, or intake of other goitrogens in volunteers, which may affect the ability to detect low-level perchlorate effects in population studies. Greer et al. (2002) used subjects' radioiodide uptake data as their own control, implicitly assuming that the impact of these other factors did not change over the weeks during which the study was conducted. The degree to which there is inter-individual variation in NIS structure and/or regulation contributing to variance across the population in the impact of perchlorate exposure on thyroid function is currently unknown.

Data for perchlorate effects on circulating hormone levels that might be used for model evaluation are very limited. Clinical toxicology and occupational studies in high-dose subjects showing minimal changes in thyroid hormone levels (Braverman, 2007) are discordant with general population epidemiology studies at much lower exposures that show associations between urinary perchlorate, TSH increases, and T4 (or fT4) decreases [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Given the limited data available for model evaluation, perchlorate toxicokinetics and toxicodynamics in pregnancy remain a source of uncertainty. This could impact the resulting fT4 predictions of neurodevelopmental effects. As an example, a comparison analysis of the BBDR model to predicted fT4 and TSH changes using the epidemiology literature (i.e., Steinmaus et al., 2016) is provided in Appendix B. This analysis used Steinmaus et al. (2016) to evaluate the relationship between perchlorate exposure and fT4 and TSH. When considering the median iodine intake, the changes predicted by the BBDR model for TSH are smaller than that predicted by Steinmaus et al. (2016), even when compared to the predictions based on the lower beta estimate from the 95% confidence interval (CI). However, when comparing the BBDR model's low-iodine TSH changes to those predicted by Steinmaus et al. (2016), they are closer in magnitude to the lower effect estimates and steadily increase to greater predicted changes as doses increase. In evaluating the results for fT4, in all instances the BBDR model predicts a smaller change than that projected by the Steinmaus et al. (2016) slope. The closest match, especially at lower perchlorate doses, is between the low-iodine runs of the BBDR model and the low estimate of the Steinmaus et al. (2016) slope. In addition, an approach to assess changes in IQ as a result of changes in perchlorate exposure using Steinmaus et al. (2016) is provided in Appendix C.

Incorporation of the TSH feedback loop into the model introduces other uncertainties. This is because, as discussed in Fitzgerald and Bean [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Year>2016</Year><RecNum>1876</RecNum><DisplayText>(2016)</Display

Text><record><rec-number>1876</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermcdxc5vxfpkax2vzp0ftv29" timestamp="1495213070">1876</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Fitzgerald, S. P.</author><author>Bean, N. G.</author></authors></contributors><auth-address>Department of Internal Medicine and Department of Endocrinology, The Royal Adelaide Hospital, Adelaide, SA 5000, Australia; School of Medicine, The University of Adelaide, Adelaide, SA 5005, Australia. & #xD; School of Mathematical Sciences, The University of Adelaide, Adelaide, SA 5005, Australia; ARC Centre of Excellence for Mathematical and Statistical Frontiers, The University of Adelaide, Adelaide, SA 5005, Australia. <a href="mailto:/auth-address"><title>><title>The relationship between population T4/TSH set point data and T4/TSH physiology</title><secondary-title>Journal of Thyroid Research</secondary-title></title><periodical><full-title>Journal of Thyroid Research</fulltitle></periodical><pages>6351473</pages><volume>2016</volume><edition>2016/04/29</edition ><dates><year>2016</year></dates><isbn>2090-8067 (Print)</isbn><accessionnum>27123359</accession-num><urls></urls><custom2>4830732</custom2><electronic-resourcenum>10.1155/2016/6351473</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>], increasing TSH levels do not have a consistently predictable impact on T4 levels. Fitzgerald and Bean (2016, p. 1) conclude that their mathematically derived curves describing the T4 response to TSH imply "greater interindividual variation in the positive thyroid T4 response to TSH than in the central inhibitory TSH response to T4." Additionally, Hadlow et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Hadlow</Author><Year>2013</Year><RecNum>893</RecNum><Dis playText>(2013)</br> app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206433">893</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><authors><author>Hadlow, N. C.</author><author>Rothacker, K. M.</author><author>Wardrop, R.</author><author>Brown, S. J.</author><author>Lim, E. M.</author><author>Walsh, J. P.</author></authors></contributors><titles><title>The relationship between TSH and free T<sub>4</sub> in a large population is complex and nonlinear and differs by age and sex</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondarytitle><alt-title>J Clin Endocrinol Metab</alt-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><alt-periodical><full-title>J Clin Endocrinol Metab</full-title></alt-periodical><pages>2936-2943</pages><volume>98</volume><number>7</number><dates><year>2013</year></dates><isb n>ISSN 0021-972X
EISSN 1945-7197</isbn><label>3645386</label><urls><relatedurls><url>http://dx.doi.org/10.1210/jc.2012-4223</url></related-urls></urls><electronic-resourcenum>10.1210/jc.2012-4223</electronic-resourcenum><language>English</language></record></Cite></EndNote>] demonstrated that the relationship between TSH and T4, while a consistently decreasing function, is variable across the population with factors such as genetics, subclinical thyroid disease, age, and gender contributing to this variability. Thus, the ability of the thyroid to respond after perturbation by perchlorate varies across the population. Further, the association between perchlorate and diminished T4 in the face of elevated TSH (Steinmaus et al., 2013; Steinmaus et al., 2007; Blount et al., 2005) suggests that certain segments of the population will have a limited ability to respond to perturbations through the TSH feedback mechanism. The reason is that some segments of the population have increased NIS inhibition due to other exposures (e.g., smokers) or increased perchlorate inhibition of the NIS because of low iodine intake. The strength of the TSH stimulatory effect is varied in model

predictions to account for such variability. The results in [ REF \_Ref482951228 \h ] and [ REF \_Ref483557363 \h ], for example, are from assuming pTSH = 0.398 (i.e., a stimulatory effect  $\sim 60\%$  lower than pTSH = 1, results for which are presented in [ REF \_Ref512496933 \h ] and [ REF \_Ref512496934 \h ]) or direct proportionality, which otherwise appears to fit population median data reasonably well. Using either the weak or a stronger stimulatory term, the feedback does not maintain fT4 concentration exactly at the set point or target. Any exposure to perchlorate is predicted to reduce fT4 to some extent. The TSH feedback only reduces the dose-response compared to what is predicted without the term.

As mentioned above, the potential for perchlorate to interact with other goitrogens has been documented with respect to thiocyanate and tobacco smoking [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. As in the case with the low-iodide analysis of Blount et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Blount</Author><Year>2006</Year><RecNum>224</RecNum><Disp layText>(2006)</DisplayText><record><rec-number>224</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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L.</author><author>Osterloh, J. D.</author><author>Valentin-Blasini,

L.</author><author>Caldwell, K. L.</author></authors></contributors><title>Urinary perchlorate and thyroid hormone levels in adolescent and adult men and women living in the United States</title><secondary-title>Environmental Health Perspectives</secondary-title></title></periodical><full-title>Environmental Health Perspectives</full-title></periodical><pages>1865-

71</pages><volume>114</volume><number>12</number><dates><year>2006</year></dates><url s></urls></record></Cite></EndNote>], the interaction with thiocyanate was most evident in a nested design in which the vulnerable population of smokers could be evaluated separately from the remainder of the population [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. Since data used in the calibration of the BBDR model were taken from NHANES 2007-12, with exclusion only for frank thyroid disease or medical treatment, they are presumed to be representative of the U.S. population as a whole. Since the NHANES data are also used to estimate overall exposure of the U.S. population to other goitrogens, the degree to which these may not reflect actual population exposure would point to an underlying problem in the NHANES sampling. The use of these data in the calibration of baseline thyroid hormone status effectively means that the predicted perchlorate dose-response is additive to the effect of existing exposure to other goitrogens in the NHANES 2007-12 population. Currently, the vulnerability introduced by other goitrogens' interactions with perchlorate at the NIS or on fT4 population distributions remains a source of uncertainty that could impact the resulting fT4 predictions of neurodevelopmental effects. The current BBDR model and analysis assume that exposures to perchlorate and other goitrogens vary independently in the population and that the effects are additive.

Investigation of the effect of perchlorate on iodine transport by the NIS *in vitro* [ ADDIN EN.CITE <EndNote><Cite><Author>Kosugi</Author><Year>1996</Year><RecNum>280</RecNum><Displ ayText>(Kosugi et al., 1996)</DisplayText><record><rec-number>280</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468526353">280</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Kosugi, S.</author><author>Sasaki,

N.</author><author>Hai, N.</author><author>Sugawa, H.</author><author>Aoki, N.</author><author>Shigemasa, C.</author><author>Mori, T.</author><author>Yoshida, A.</author></authors></contributors><auth-address>Department of Laboratory Medicine, Kyoto University School of Medicine, Japan. kosugi@kuhp.kyto.u.ac.jp</authaddress><title>><title>Establishment and characterization of a Chinese hamster ovary cell line, CHO-4J, stably expressing a number of Na+/I- symporters</title><secondary-title>Biochemical and Biophysical Research Communications</secondary-title></title><periodical><fulltitle>Biochemical and Biophysical Research Communications</full-title></periodical><pages>94-101</pages><volume>227</volume><number>1</number><edition>1996/10/03</edition><keywor ds><keyword>Animals</keyword><keyword>CHO Cells</keyword><keyword>Carrier Proteins/ genetics/metabolism</keyword><keyword>Cell Line</keyword><keyword>Cloning. Molecular</keyword><keyword>Cricetinae</keyword><keyword>Ion Transport</keyword><keyword>Kinetics</keyword><keyword>Membrane Potentials</keyword><keyword>Membrane Proteins/ genetics/metabolism</keyword><keyword>Keyword><keyword>Symporters</keyword></key words><dates><year>1996</year><pub-dates><date>Oct 3</date></pubdates></dates></isbn>0006-291X (Print)&#xD:0006-291X (Linking)</isbn><accessionnum>8858109</accession-num><urls></urls><electronic-resourcenum>10.1006/bbrc.1996.1473</electronic-resource-num><remote-database-provider>NLM</remotedatabase-provider><language>eng</language></record></Cite></EndNote>] and revised mathematical analysis by Schlosser (2016) indicate that this effect is consistent with classical competitive inhibition of the transporter. An equation describing this competitive inhibition is provided in Appendix A, Section 1.2.2. Based on this mechanism, and the simple pharmacokinetics of perchlorate (no metabolism, ~ 100% urinary excretion), it is assumed that there is minimal additional uncertainty in extrapolating the effect of perchlorate on iodide uptake downward from the middle perchlorate exposure levels used by Greer et al. (2002) (i.e., 20 and 100 μg/kg/day) to lower exposure levels relevant to MCLG derivation and/or human environmental exposures. The various uncertainties discussed above and in Appendix A do apply to low-dose extrapolation, but based on this mechanistic understanding there should be no magnification of uncertainty at low doses. The model is likely to become more uncertain when extrapolating to higher exposure levels, where mechanisms not included in the model may become active.

Lastly, the potential for perchlorate to interact with pre-existing thyroidal diseases and conditions is a source of uncertainty. The BBDR model presented here is intended to describe an otherwise healthy (euthyroid) woman during the period extending from several years before pregnancy through gestational week 16. Because it assumes euthyroid status, it does not, for example, address thyroiditis, which is caused by the production of thyroid autoantibodies (e.g., Hashimoto's disease) and is a common condition that may involve impaired iodide uptake across the NIS and in some individuals may result in reduced production of T4 [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. The ATSDR Toxicological Profile for Perchlorate [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Year>2008</Year><RecNum>115</RecNum><DisplayText>(2008)</DisplayT ext><record><rec-number>115</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1465914059">115</key></foreignkeys><ref-type name="Government Document">46</reftype><contributors><authors><author>ATSDR</author></author></contributors><titles><title>To xicological profile for perchlorates</title></title></dates><year>2008</year></dates><publocation>Atlanta, GA</pub-location><publisher>U.S. Department of Health and Human

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Services</publisher><urls></record></Cite></EndNote>] states that individuals with Hashimoto's disease and exposure to perchlorate have relatively poor thyroid function. To the extent that such conditions or pre-conditions cause additional interference with iodide uptake and deplete colloidal stores of iodide, this may set the stage for a toxicant-disease interaction with perchlorate that would constitute an added vulnerability and uncertainty consideration.

## 4. Deriving a Distribution of Thyroid Hormone Levels from the BBDR Model Output

While the BBDR model can be calibrated to match specific percentiles of the population, the distributions in the underlying parameters required to predict the distribution of thyroid hormone levels have not been determined. Therefore, in order to predict the impact of perchlorate exposure on the population distribution of fT4 (i.e., on the population fraction of hypothyroxinemic individuals; those with fT4 below the reference range and with TSH in the normal range), the EPA estimated a distribution for fT4 plasma concentrations around the median modeled values while accounting for the effects of perchlorate and iodine. This section describes the process of estimating these distributions.

## 4.1 Evaluation of Distributional Shape and Derivation of the Geometric Standard Deviation

Results from the literature review (see Section [ REF \_ Ref488161860 \n \h ]) suggest that it would be most useful to estimate distributions around the BBDR model output for GWs 12 and 13. Furthermore, this review suggested that it would be most useful to concentrate on fT4 rather than other thyroid hormones given maternal fT4 has the strongest relationship with neurodevelopmental outcomes (Korevaar et al., 2016; Pop et al., 1999, 2003; Finken et al., 2013). To develop population distributions for fT4 concentrations around the BBDR-estimated median fT4 values, the EPA evaluated the fT4 data from the studies that were used to calibrate the BBDR model (see Figure A-33 in Appendix A). Only one study provided distributional data for GW 13 fT4 [ ADDIN EN.CITE <EndNote><Cite><Author>Männistö</Author><Year>2011</Year><RecNum>1206</RecNum><D isplayText>(Männistö et al., 2011)</DisplayText><record><rec-number>1206</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495206438">1206</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><author>Männistö, T.</author><author>Surcel, H. M.</author><author>Ruokonen, A.</author><author>Vääräsmäki, M.</author><author>Pouta, A.</author><author>Hartikainen, A. L.</author><author>Suvanto, E.</author></contributors><title><title>Early pregnancy reference intervals of thyroid hormone concentrations in a thyroid antibody-negative pregnant population</title><secondary-title>Thyroid</secondary-title><alt-title>Thyroid</alttitle></title></periodical><full-title>Thyroid</full-title></periodical><alt-periodical><fulltitle>Thyroid</full-title></alt-periodical><pages>291-298</pages><volume>21</volume><number>3</number><dates><year>2011</year></dates><isbn >ISSN 1050-7256&#xD:EISSN 1557-9077</isbn><accession-num>21254924</accessionnum><label>3590036</label><urls><relatedurls><url>http://dx.doi.org/10.1089/thy.2010.0337</url></related-urls></urls><electronic-resourcenum>10.1089/thy.2010.0337</electronic-resourcenum><language>English</language></record></Cite></EndNote>], so GW 12 data were used to estimate distributional parameters for both gestational weeks. Specifically, Li et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Li</Author><Year>2014</Year><RccNum>342</RccNum><DisplayT

ext>(2014)</br>
/DisplayText><record><rec-number>342</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1495025557">342</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Li, C.

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</author><author>Shan, Z. </author><author>Mao, J.</author><author>et
al.,</author></authors></contributors></title>><title>Assessment of thyroid function during first-
trimester pregnancy: what is the rational upper limit of serum TSH during the first trimester in
Chinese pregnant women?</title><secondary-title>Journal of Clinical Endocrinology and
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79</pages><volume>99</volume><number>1</number><dates><year>2014</year></dates><urls>
</urls></record></Cite></EndNote>], Männistö et al. [ ADDIN ZOTERO ITEM CSL CITATION
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num><language>English</language></record></Cite></EndNote>] were evaluated for GW 12 and
GW 13 ([ REF _Ref521509844 \h ]).
All of the examined study data demonstrated a positive skew, and overall the lognormal function
demonstrated a better fit than a normal distribution. Despite this, the available study data only
```

accounted for variation due to gestational week, and did not account for variation in perchlorate and

iodine intake in the measured populations. Because perchlorate and iodine can affect fT4 levels, and this relationship produced the estimated median BBDR values, the distribution around values estimated by the model from perchlorate and iodine intake should account for a small reduction in variation due to the effect of perchlorate and iodine intake. Additionally, as iodine has a demonstrated lognormal distribution with strong right skew (e.g., Blount et al., 2007), and is predicted to have a stronger effect on fT4 than perchlorate (see Section 3), the EPA assumed the error around predicted fT4 would likely be closer to normal than lognormal after accounting for perchlorate and iodine intake.

Standard deviations (SDs) were estimated accounting for gestational week using quantile data from each relevant study with the *get.norm.par()* function from package rriskDistributions in R [ ADDIN EN.CITE

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/author><author>Schueller,

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Liisa"}, ("family": "Suvanto", "given": "Eila"}], "issued": {"date-parts": [["2011"]]}}, "suppress-author": true}], "schema": "https://github.com/citation-style-language/schema/raw/master/csl-citation.json"}] that also contained GW 12 data. This introduced unnecessary complications into the estimation of a combined SD using data from both gestational weeks, as the two different weeks had different sample sizes, and the SD of the GW 13 data fell within the range of SDs estimated from other studies with GW 12 fT4 data. As such, the EPA opted to only estimate a combined SD using data from GW 12.

The EPA used the *combinevar()* function from the *fishmethods* package (Nelson, 2017) to estimate combined variance weighted by sample size (Appendix D). This yielded a representative SD for GW 12 and GW 13 based on the relevant data from each study. These SDs were assumed to represent the median-iodine intake distribution, as this is what these studies were used for in model calibration. After estimating the combined SD, the EPA assumed a small reduction in variability would occur when further accounting for known iodide and perchlorate intakes. The EPA based the estimate of this reduction on results from Steinmaus et al. (2016). Steinmaus et al. (2016) incorporated both iodide and perchlorate measurements into a regression model describing thyroid hormone levels. The overall model described a large portion of the variance. While both iodide and perchlorate significantly affected fT4 values, the authors did not provide simple measures to allow an assessment of how much each of these contributed to the overall fit. As such, the EPA estimated that the combination of these intakes would be analogous to an  $R^2$  of roughly 0.1, which would indicate a small to medium effect size. The EPA adjusted the standard deviation for this rough estimate as follows:

Adjusted SD = 
$$\sqrt[2]{(1-R^2)}$$
 \* Combined SD

While this method does not account for non-linearity in the effects of perchlorate and iodine intake, it does help to incorporate the model assumption that both iodine and perchlorate affect fT4 concentrations.

More information on these methods can be found in Appendix D.

Table [ SEQ Table \\* ARABIC ]. Estimated SDs, Sample Sizes, and Combined SDs Based on Data from Relevant Studies with fT4 Data from GW 12 and GW 13

	CombinedS	SampleSiz	C o m bi n e d S D R a w M e di a n lo di n e ln ta k	A dj us te d S D fo r M ed ia n lo di ne and N o P erch lo ra	A djusted S D for L o S lo di e and N o P etchlora
Study - Gestational Week Li et al. (2014) – Week 12	2. 1	1 2	е	te	te
	3	8			

Study - Gestational Week  Männistö et al. [ADDIN ZOTERO_ITEM CSL_CITATION ("citationID":"a2[q6ckhvho", "properties";("formattedCitation":"(2011)","plainCitation":"(2011)")," citationItems":[("ii"):20_"uris","[http://zotero.org/groups/945096/items/UADDN6QU"],"uri":["http://zotero.org/groups/945096/items/UADDN6QU"],"uri":["http://zotero.org/groups/945096/items/UADDN6QU"],"itemData":("ii"-920, "type":"articlejournal","itile":"Early pregnancy reference intervals of thyroid hormone concentrations in a thyroid antibody-negative pregnant population", "container-title":"Thyroid", "page":"291-298", "volume":"21", "issue":"3", "author":[("family":"Mānnistō", "given":"Tuija"), "family":"Surcel", "given":"Heljä.  Marja"), {"family":"Ruokonen", "given":"Aini"), {"family":"Vääräsmäki" "given":"Marja"), {"family":"Pouta", "given":"Aini"), {"family":"Järvelin", "given":"Marjo-Riitta"), {"family":"Bisued":"Aini", {"family":"Suvanto" given":"Liia"), "issued":[date-parts"[["2011"]])}, "suppressauthor":true)], "schema":"https://github.com/citation.style-language/schema/raw/master/csl-citation.json") ]  — Week 13	C o m bi n e d S D	SampleSize 325	Combined SDR aw Median lodine lntake	A djuste d S D fo r M ed ia n lo di ne an d N o P er ch lo ra te	A djuste d S D for L o w lo di ne an d N o P er ch lo ra te
Zhang et al. [ ADDIN ZOTERO_ITEM CSL_CITATION {"citationID":"a2bolpmks7v","properties":{"formattedCitation":"(2016)","plainCitation":"(2016)"}, "citationItems":[["id":918,"uris":["http://zotero.org/groups/945096/items/HTDB3EVS"],"uri":["htt p://zotero.org/groups/945096/items/HTDB3EVS"],"itemData":{"id":918,"type":"article-journal","title":"Reference Intervals of Thyroid Function During Pregnancy: Self-Sequential Longitudinal Study Versus Cross-Sectional Study","container-title":"Thyroid","page":"1786-1793","volume":"26","issue":"12","source":"CrossRef","DOI":"10.1089/thy.2016.0002","ISSN":"1050-7256, 1557-9077","shortTitle":"Reference Intervals of Thyroid Function During	1. 6 7	8 7			

Study - Gestational Week  Pregnancy", "language". "en", "author". [("family". "Zhang", "given". "Xiaomei"], ("family". "Yao", "given". "Baoting"), ("family". "Li", "given". "Chenyang", ("family". "Mao", "given". "Jinyuang"), ("family". "Weiwei"), ("family". "Xiaochung"), ("family". "Teng", "given". "Xiaochung"), ("family". "Li", "given". "Chenyang"), ("family". "Zhou", "given". "Weiwei"), ("family". "Li", "given". "Chenyang"), ("family". "Zhou", "given". "Shaowei"), ("family". "Gao", "given". "Zhang", "given". "Shaowei"), ("family". "Gao", "given". "Zhengnang", ("family". "Given". "Liu"), ("family". "Fan", "given". "Chenling"), ("family". "Teng", "given". "Jingnan"), ("family". "Gao", "given". "Zhengnang", ("family". "Gao", "given". "Liu"), ("family". "Fan", "given". "Chenling"), ("family". "Teng", "given". "Jingnan", "given", "Jingnan", "given", "Jingnan", "given", "Jingnan", "given", "Jingnan", "given", "Jingnan, "given", "Jingnan, "given", "Jingnan,	C o m bi n e d S D	S a m p l e S i z e	C o m bi n e d S D R a w M e di a n lo di n e ln ta k e	A djuste d S D for M ed ia n lo di ne and N o P er ch lo ra te	A dj us te d S D fo r L o w lo di ne an d N o P er ch lo ra te
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### 4.2 Results of Estimating a Distribution around BBDR Median Output

The EPA used estimated SDs to predict distributions of fT4 around BBDR median fT4 estimates, based on particular levels of iodine sufficiency and doses of perchlorate. The *qnorm()* function from the *stats* package in R (R Core Team, 2018), which is analogous to the Microsoft® Excel norm.inv function described in Ginsberg et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1">Exc

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P.</author></authors></contributors><titles><title>Methylmercury-induced inhibition of paraoxonase-1 (PON1) - Implications for cardiovascular risk</title><secondary-title>Journal of Toxicology and Environmental Health, Part A: Current Issues</secondary-

title></titles><periodical><full-title>Journal of Toxicology and Environmental Health, Part A: Current Issues</full-title></periodical><pages>1004-

1023</pages><volume>77</volume><number>17</number><section>1004</section><dates><year>2014</year></dates><urls></urls><electronic-resource-

num>10.1080/15287394.2014.919837</electronic-resource-num></record></Cite></EndNote>], was used for these predictions. [ REF \_Ref512860801 \h ] and [ REF \_Ref512860806 \h ] demonstrate fT4 distributions for various doses of perchlorate based on the EPA's calculated SDs for GW 12 and GW 13, respectively, for median iodine intake and low iodine intake assuming pTSH = 0.398. [ REF \_Ref512860812 \h ] and [ REF \_Ref512860816 \h ] demonstrate these results assuming pTSH = 1.

Table [ SEQ Table \\* ARABIC ]. Summary of BBDR Model Results for fT4 for GW 12 at 75 and 170  $\mu$ g/day lodine Intake at pTSH = 0.398 []

Perchlorate	Percentile fT4 (pmol/L)					
Dose (µg/kg/day)	2.5 <sup>th</sup>	5 <sup>th</sup>	10 <sup>th</sup>	50 <sup>th</sup>	95 <sup>th</sup>	97.5 <sup>th</sup>
		lodin	e Intake – 17	9.63		
0	6.72	7.36	8.09	10.67	13.97	14.61
1	6.72	7.35	8.08	10.66	13.97	14.60
2	6.71	7.35	8.08	10.66	13.97	14.60
3	6.71	7.34	8.07	10.65	13.96	14.60
4	6.70	7.34	8.07	10.65	13.96	14.59
5	6.70	7.33	8.06	10.64	13.95	14.59
10	6.67	7.31	8.04	10.62	13.93	14.56
		[915]]		r day <sup>a</sup>		
0	5.58	6.11	6.71	8.85	11.60	12.12
1	5.51	6.04	6.64	8.78	11.52	12.05
2	5.44	5.97	6.57	8.71	11.46	11.98
3	5.38	5.90	6.51	8.65	11.39	11.92
4	5.32	5.84	6.45	8.59	11.33	11.86
5	5.26	5.78	6.39	8.53	11.27	11.80
10	5.00	5.53	6.13	8.27	11.01	11.54

<sup>&</sup>lt;sup>a</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **2.01**.

<sup>&</sup>lt;sup>b</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **1.67**.

## Table [ SEQ Table \\* ARABIC ]. Summary of BBDR Model Results for fT4 for GW 13 at 75 and 170 $\mu g/day$ lodine Intake at pTSH = 0.398

Perchlorate	Percentile fT4 (pmol/L)					
Dose (µg/kg/day)	2.5 <sup>th</sup>	5 <sup>th</sup>	10 <sup>th</sup>	50 <sup>th</sup>	95 <sup>th</sup>	97.5 <sup>th</sup>
		facility	anielo-ivini	(des)		
0	6.70	7.33	8.07	10.64	13.95	14.59
1	6.70	7.33	8.06	10.64	13.95	14.58
2	6.69	7.32	8.06	10.63	13.94	14.58
3	6.69	7.32	8.05	10.63	13.94	14.57
4	6.68	7.31	8.04	10.62	13.93	14.57
5	6.67	7.31	8.04	10.62	13.93	14.56
10	6.65	7.28	8.01	10.59	13.90	14.53
		10.01				
0	5.57	6.09	6.70	8.84	11.59	12.12
1	5.50	6.02	6.63	8.77	11.52	12.05
2	5.43	5.96	6.56	8.71	11.45	11.98
3	5.37	5.89	6.50	8.64	11.39	11.92
4	5.31	5.83	6.44	8.58	11.33	11.86
5	5.25	5.77	6.38	8.52	11.27	11.80
10	4.99	5.52	6.12	8.27	11.01	11.54

<sup>&</sup>lt;sup>a</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **2.01**.

<sup>&</sup>lt;sup>b</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **1.67**.

## Table [ SEQ Table \\* ARABIC ]. Summary of BBDR Model Results for fT4 for GW 12 at 75 and 170 $\mu g/day$ lodine Intake at pTSH = 1

Perchlorate	Percentile fT4 (pmol/L)					
Dose (μg/kg/day)	2.5 <sup>th</sup>	5 <sup>th</sup>	10 <sup>th</sup>	50 <sup>th</sup>	95 <sup>th</sup>	97.5 <sup>th</sup>
		facility	almetro = 170 pc	Maye		
0	6.72	7.36	8.09	10.67	13.97	14.61
1	6.72	7.35	8.09	10.66	13.97	14.61
2	6.72	7.35	8.08	10.66	13.97	14.60
3	6.72	7.35	8.08	10.66	13.97	14.60
4	6.71	7.35	8.08	10.66	13.96	14.60
5	6.71	7.34	8.07	10.65	13.96	14.60
10	6.70	7.33	8.06	10.64	13.95	14.58
		letelle	e Intake = 75 un	s ay		
0	5.58	6.11	6.71	8.85	11.60	12.12
1	5.54	6.06	6.67	8.81	11.55	12.08
2	5.50	6.02	6.63	8.77	11.51	12.04
3	5.46	5.99	6.59	8.73	11.48	12.00
4	5.42	5.95	6.55	8.69	11.44	11.96
5	5.39	5.91	6.52	8.66	11.40	11.93
10	5.22	5.74	6.35	8.49	11.23	11.76

<sup>&</sup>lt;sup>a</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **2.01**.

<sup>&</sup>lt;sup>b</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **1.67**.

Table [ SEQ Table \\* ARABIC ]. Summary of BBDR Model Results for fT4 for GW 13 at 75 and 170  $\mu$ g/day lodine Intake at pTSH = 1

Perchlorate	Percentile fT4 (pmol/L)					
Dose (μg/kg/day)	2.5 <sup>th</sup>	5 <sup>th</sup>	10 <sup>th</sup>	50 <sup>th</sup>	95 <sup>th</sup>	97.5 <sup>th</sup>
		lodin	e installe = 1/1 is	rdeye		
0	6.70	7.33	8.07	10.64	13.95	14.59
1	6.70	7.33	8.06	10.64	13.95	14.58
2	6.69	7.33	8.06	10.64	13.95	14.58
3	6.69	7.33	8.06	10.63	13.94	14.58
4	6.69	7.32	8.05	10.63	13.94	14.58
5	6.69	7.32	8.05	10.63	13.94	14.57
10	6.67	7.30	8.04	10.61	13.92	14.56
		[ess])	(3)			
0	5.57	6.09	6.70	8.84	11.59	12.12
1	5.53	6.05	6.66	8.80	11.55	12.08
2	5.49	6.01	6.62	8.76	11.51	12.04
3	5.45	5.97	6.58	8.72	11.47	12.00
4	5.41	5.94	6.54	8.69	11.44	11.96
5	5.37	5.90	6.51	8.65	11.40	11.93
10	5.21	5.73	6.34	8.48	11.23	11.76

<sup>&</sup>lt;sup>a</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **2.01**.

#### 4.3 Limitations and Uncertainties of Method

Although the lognormal distribution better describes the reported quantiles of fT4 in the manuscripts reviewed when accounting for gestational week, the EPA opted to assume a normal distribution. Departures from normality were small in the published quantile data (Appendix D, Figure D-1), and it was assumed that the distribution of error around values of fT4 accounting for perchlorate and iodine would even more closely approach a normal distribution. The primary limitation of this process is that very few published studies allowed for the estimation of the strength of the effects of perchlorate and iodine on fT4 levels, or the shape of the relationships. A rough estimate of the distribution is provided based on an assumption that the combination of accounting for iodide and perchlorate would provide a reduction in error analogous to an  $R^2$  of 0.1. While the effects of perchlorate and iodide are non-linear, the EPA has no additional information that would allow for the further improvement of the estimation of the error distribution. The current method for estimating the error distribution avoids producing centile intervals that are likely to be wider than the true intervals.

<sup>&</sup>lt;sup>b</sup> 50<sup>th</sup> percentile is direct output from the BBDR model, additional percentile estimated by assuming a normal distribution with a SD = **1.67**.

# 5. Connecting Alterations in Thyroid Hormone Levels to Alterations in Neurodevelopment: Identification of Studies for Further Analysis

This section addresses the SAB recommendation to "identify literature and conduct analyses to support the model outputs for the downstream steps" [ ADDIN EN.CITE <EndNote><Cite><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Pages>2 7-28</Pages><DisplayText>(SAB, 2013, pp. 27-28)</DisplayText><record><rec-number>50</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-keys><ref-type name="Government" Document">46</ref-type><contributors><author>SAB.</author></author></author> authors><author>U.S. Environmental Protection Agency,</author></secondaryauthors></contributors><title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><year>2013</year></dates><publocation>Washington, DC</pub-location><urls></urls></record></EndNote>] from the BBDR's predicted changes in thyroid hormones. Specifically, this section presents the process by which literature was identified to complete the linkage between altered maternal T4 and fT4 (as predicted by the BBDR model) and adverse neurodevelopmental effects in offspring. Section [ REF Ref482275327 \r \h | presents the process used to identify the literature to support this approach. Sections [ REF Ref491075421 \n \h ] and [ REF Ref482275344 \r \h ] describe the literature search approach and results of the process used to identify relevant studies that can link the predictions of changes in fT4 from the BBDR model to incremental changes in neurodevelopment. Section [ REF Ref482275380 \r \h ] presents results from the literature that are used to support an alternative proposed approach presented in Section [ REF Ref481160968 \r \h ]: evaluating a shift in the proportion of the population that will fall below a hypothyroxinemic cut point, given exposure to perchlorate.

### 5.1 Searching the Published Literature

The available data for the second stage of the analysis come from epidemiological studies that evaluated maternal thyroid hormone levels in pregnancy and neurodevelopmental outcomes (these are not studies evaluating perchlorate exposure). Based on the recommendations of previous peer review panels, the EPA assumed that changes in thyroid hormone levels would be expected to lead to neurodevelopmental outcomes. For this reason, the EPA did not conduct a complete, systematic review of the body of literature on this topic. However, the EPA conducted a focused review of the published literature and identified epidemiological studies that examined thyroid hormone levels and neurodevelopmental outcomes.

The EPA conducted the literature searches in PubMed and Google Scholar focusing on epidemiologic studies that explicitly evaluated the association between maternal thyroid hormone T4 or fT4 levels and offspring neurodevelopment. A protocol was developed to identify and evaluate the studies located. Studies were categorized based on whether the analysis was based on a categorical or a continuous measure of thyroid hormones as they related to neurodevelopment, and then studies identified as potentially useful for further dose-response analysis went through a risk-of-bias (ROB) analysis. Search strings and categories are presented in [ REF \_Ref424292244 \h \\* MERGEFORMAT ]. As depicted in this table, the endpoints included neurodevelopmental measures, autism spectrum disorder (ASD), and attention deficit/hyperactivity disorder (ADHD). These

categories were chosen as outcomes associated with alterations in thyroid hormone levels based on the 2013 SAB recommendations, CalEPA's *Public Health Goal for Perchlorate in Drinking Water* [ ADDIN EN.CITE

<EndNote><Cite><Author>CalEPA</Author><Year>2011</Year><RecNum>296</RecNum><DisplayText>(CalEPA, 2011)</DisplayText><record><rec-number>296</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1470937233">296</key></foreign-keys><ref-type name="Computer Program">9</ref-type><contributors><author>>CalEPA</author></authors></contributors><title>Dr aft: Public health goal for perchlorate in drinking

water</title></title></title></title></title></title></title></url>
face="underline" font="default" size="100%">https://oehha.ca.gov/media/downloads/water/public-health-goal/010711perchloratephg.pdf</style></url></related-

urls></urls></record></Cite></EndNote>] and the National Research Council's *Health Implications* of *Perchlorate Ingestion* [ ADDIN EN.CITE

<EndNote><Cite><Author>NRC</Author><Year>2005</Year><RecNum>1332</RecNum><DisplayText>(NRC, 2005)</DisplayText><record><rec-number>1332</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1495206440">1332</key></foreign-keys><ref-type name="Book">6</ref-type><contributors><author>NRC</author></author></contributors><title>Healt h implications of perchlorate ingestion</title></title><dates><year>2005</year></dates><publication>Washington, DC</publication><publisher>National Academies Press

urls><url>http://www.nap.edu/catalog/11202.html</url></related-urls></urls><modified-date>National Research Council</modified-date></record></Cite></EndNote>].

Searches were restricted to studies published since 2000 to limit the results to the most recent science. Limiting studies to the past 17 years was supported by Pop et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayText>(1999)</DisplayText><record><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062208">40</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>Vauthor><author>Valsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Valer, H L</author></author></author></author></author><author>Valer, H L</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></arthor></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author><

155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><url s></urls></record></Cite></EndNote>] (this study was identified by SAB, 2013), who noted there were no studies evaluating the relationship between maternal thyroid hormone levels and infant neurodevelopment from iodine-sufficient populations prior to 1999. In addition, the SAB document [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>SAB</Author><Year>2013</Year><RecNum>50</RecNum><Display Text>(2013)</DisplayText><record><rec-number>50</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437138201">50</key></foreign-

keys><ref-type name="Government Document">46</ref-type><contributors><author>SAB,</author></author>><author>U.S.

Environmental Protection Agency,</author></secondary-authors></contributors></title>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></title></dates>

goal for perchlorate. EPA-SAB-13-004</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title><a href="mailto:color:black black bla

 $type > < contributors > < author > SAB, < / author > < / author > < author > U.S. \\ Environmental Protection Agency, < / author > < / secondary-author > < < < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < > < < < > < < < > < < > < < > < < < > < < < > < < < > < < < > < < < > < < < > < < < > < < < < > < < <$ 

authors></contributors><tittles>Advice on approaches to derive a maximum contaminant level goal for perchlorate. EPA-SAB-13-004</title></title></dates><quer>2013</quer></dates><publication>Washington, DC</publication><urls></urls></record></Cite></EndNote>], including some that were published prior to 2000. Titles and abstracts of all studies were reviewed to determine if the study had pertinent information relating thyroid hormone levels, particularly T4 or fT4, and adverse neurodevelopmental outcomes. Further, to ensure all relevant papers were captured, recent review papers, including three systematic reviews (Fetene et al., 2018; Moog et al., 2017, Thompson et al., 2018) pertaining to altered maternal thyroid hormone levels and offspring neurodevelopment were also evaluated.<sup>8</sup>

Reviewing the Google Scholar and PubMed literature searches resulted in the identification of 59 studies that assessed the relationship between maternal thyroid hormone levels and neurodevelopmental outcomes in the offspring. Review of the SAB document identified six additional papers<sup>9</sup> to evaluate, and review of recent systematic reviews identified six additional papers<sup>10</sup> to evaluate. Therefore, a total of 71 studies that pertained to altered maternal thyroid hormone levels and offspring neurodevelopment were identified for evaluation.

5-3

Summary tables of the articles were evaluated. If a study was identified in these tables as containing information on hypothyroxinemia or fT4 and an assessment of neurodevelopment, it was further reviewed for potential relevance.

<sup>&</sup>lt;sup>9</sup> Haddow et al. (1999); Man, Brown, and Serunian (1991); Pharoah, Connolly, Ekins, and Harding (1984); Pop et al. (1999); Velasco et al. (2009); Willoughby (2011).

Brown et al. (2015); Ghassabian et al. (2011); Ghassabian et al. (2012); Kasatkina et al. (2006); Lazarus et al. (2012); Muñoz, Figueras, & Puig (2009).

### Table [ SEQ Table \\* ARABIC ]. Summary of Search Strings Used

Search Category	Search Strings <sup>a</sup>
Endpoint: Neurodevelopment	Neurodevelopment AND hypothyroxinemiab
	"Neurodevelopmental effects" AND hypothyroxinemiab
	Hypothyroxinemia AND behavior
	Hypothyroxinemia AND behavior AND infant
	Hypothyroxinemia AND thyroxine AND behavior AND infant
	(Thyroxine OR T4 OR hypothyroxinemia) AND (neurodevelopment OR "mental ability" OR intelligence OR cognition OR cognitive OR motor OR language)
Endpoint: Attention Deficit/ Hyperactivity Disorder	"Maternal hypothyroxinemia" AND thyroxine AND ADHD AND infant
Endpoint: Autism	Hypothyroxinemia AND autism <sup>b</sup>
·	Hypothyroxinemia AND thyroxine AND autism AND infant

<sup>&</sup>lt;sup>b</sup> This search string resulted in more articles than the EPA could reasonably evaluate. Therefore, the search was constrained so that databases searched for those terms only in the title.

# 5.2 Approach to Locate Epidemiological Studies That Evaluated Maternal Thyroid Hormone Levels in Early Pregnancy and Neurodevelopmental Outcomes

To identify studies that linked incremental changes in maternal T4 or fT4 to incremental changes in offspring neurodevelopment (Step 2 in [ REF \_Ref417634154 \h ]), the EPA assessed each of the 71 studies identified using a four-step approach. First, studies were identified that were not compatible with BBDR model results (see Section [ REF \_Ref517273217 \r \h ]). Studies were then categorized depending on whether the analysis was based on a categorical or a continuous measure of thyroid hormones as they related to neurodevelopment (see Section [ REF \_Ref517273232 \r \h ]). Studies with potential for further dose-response analysis were then evaluated using the National Toxicology Program's Office of Health Assessment and Translation (OHAT) ROB tool (see Section [ REF \_Ref512802830 \r \h ]). Lastly, the remaining papers were evaluated to determine if it would be feasible to use data from the paper in dose-response analysis (see Section [ REF \_Ref457207854 \r \h ]).

Each step is described below.

#### 5.2.1 Step 1. Identify Studies That Are Not Compatible with BBDR Model Results

First, studies were identified that are not compatible with BBDR model results to inform MCLG development. Specifically, studies were identified with the following features:

The study design is not directly compatible with the BBDR model results. For example, if the study evaluated the impact of only neonatal thyroid hormones (i.e., at a potentially sensitive life stage), it cannot be used because the BBDR model is specific to early pregnancy. Further, if the study evaluates a population with an existing disease (i.e., hypothyroidism) that may have a different response to perchlorate compared to the euthyroid population, it was not considered compatible with BBDR model results. Additionally, if the study does not include

information on T4 or fT4, it does not assist in understanding the implications of the BBDR modeling results.

• The study does not have a population with an exposure window (i.e., when the thyroid hormone measurements are taken) that overlaps with the outputs for the BBDR model. Specifically, the study should evaluate thyroid hormone levels in pregnant mothers between conception and gestational week 16. The neurodevelopmental outcomes could be measured at any life stage.

Any studies identified in this step were categorized as Group 3.

### 5.2.2 Step 2. Categorize the Remaining Studies

The remaining studies were evaluated based on the analysis approach presented in the study. Studies were categorized based on the presence of an analysis that could inform how incremental changes in fT4 related to incremental changes in a neurodevelopmental outcome. Specifically:

- Studies that present any analyses based on a continuous measure of maternal thyroid hormone levels and subsequent offspring neurodevelopmental outcomes were placed into Group 1.
   These studies present data to inform an analysis that allows for an understanding of the impact of incremental changes in thyroid hormones and subsequent neurodevelopmental impacts.
- Studies that present analyses based only on categorical analyses of thyroid hormones (e.g., yes/no hypothyroxinemic) and subsequent neurodevelopmental effects were placed into Group 2.

Grouping of studies in this manner does not pertain to the quality of the data presented. Instead, it directly relates to the utility of the paper in informing a dose-response analysis as it pertains to evaluating the impact incremental changes in thyroid hormones and thus, potential incremental changes in neurodevelopment. Subsequently, very high quality papers that contributed significantly to the understanding of maternal thyroid hormones' impact on offspring neurodevelopment may be considered Group 2 or Group 3. This is simply because these papers do not allow for a quantitative understanding of the relationship between maternal thyroid hormone levels in early pregnancy and offspring neurodevelopment.

### 5.2.3 Step 3. Conduct Risk Of Bias (ROB) Evaluation of Group 1 Papers

A data quality analysis was conducted on each of the Group 1 papers as outlined in more detail in Appendix E. In summary, the ROB evaluation was based on the National Toxicology Program's 2015 *OHAT Risk of Bias Rating Tool for Human and Animal Studies* [ ADDIN EN.CITE <EndNote><Cite><Author>National Toxicology Program</Author><Year>2015</Year><RecNum>1987</RecNum><DisplayText>(National Toxicology Program, 2015b)</DisplayText><record><rec-number>1987</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1525098556">1987</key></foreign-keys><ref-type name="Government Document">46</ref-type><contributors><author>National Toxicology Program, </author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></a></author></a></author></a></author></a></arthor></a></author></a></arthor></a></author></a></author></a></al>

studies</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title>

- 1. Did the study design or analysis account for important confounding and modifying variables?
- 2. Were outcome data complete without attrition or exclusion from analysis?
- 3. Can we be confident in the exposure characterization?
- 4. Can we be confident in the outcome assessment?
- 5. Were all measured outcomes reported?

Criteria for ranking studies on each question were adapted from the OHAT ROB rating tool and are displayed in full in Appendix E. Possible ROB ratings for each question were: definitely low, probably low, probably high, or definitely high ROB. For example, under outcome assessment, studies were rated as definitely low ROB if they used well-established methods to assess outcomes, followed all subjects for the same length of time, and provided evidence of blinding of outcome assessors.

Based on the five ROB rankings, each Group 1 study was assigned a further rating of overall study quality ranging from Tier 1 (high quality) to Tier 3 (low quality). Specifically, these definitions are as follows:

- Tier 1: A study is rated as "definitely low" or "probably low" ROB for the majority of questions and is not rated as "definitely high" ROB for any question.
- Tier 2: Study meets criteria for neither the first nor third tiers.
- Tier 3: A study is rated as "definitely high" or "probably high" ROB for the majority of questions.

Any study ranked Tier 3 was not evaluated further. All Tier 1 and Tier 2 papers were further considered for feasibility in conducting additional analysis.

For the purposes of this document, the EPA is not selecting a key paper (or papers) on which to rely to inform the derivation of an MCLG for perchlorate (though any study ranked as Tier 3 was not evaluated further). Instead, this report presents evaluations of all identified papers with relevant and necessary data and approaches for how they may inform the derivation of an MCLG.

### 5.2.4 Step 4. Identify Group 1 Studies That Present Data to Inform a Dose-Response Relationship

After conducting the ROB evaluation, the EPA reviewed all Group 1 studies to identify those for which additional quantitative analysis could be conducted; that is, the studies that were identified as evaluating a continuous measure of fT4 and neurodevelopment and had all the requisite data to perform additional analysis to potentially inform the derivation of an MCLG.

## 5.3 Results of Evaluation to Identify Studies That Directly Relate Thyroid Hormone Levels with Neurodevelopment

As presented in [ REF\_Ref456203910 \h ], the literature review identified 71 studies that link thyroid hormone levels to neurodevelopmental effects. These studies were divided into three groups to facilitate evaluation:

- Group 1: Studies that may be able to quantitatively describe a relationship between incremental alterations in maternal thyroid hormone levels and alterations in offspring neurodevelopment. (These studies were then further categorized into tiers based on the ROB evaluation.)
- Group 2: Studies that do not have data from which to derive a quantitative relationship between maternal thyroid hormones and offspring neurodevelopment, but instead present only categorical analysis with thyroid hormones below and above a defined cut point and adverse neurodevelopmental outcomes.
- Group 3: Studies that present an analysis that is not directly compatible with BBDR output. For example, if the study evaluated data in pre-term infants or late-term pregnant women it would not inform use of the BBDR output.

## Figure [ SEQ Figure \\* ARABIC ]. Overview of Process and Results to Assess Studies Relating Thyroid Hormone Levels to Neurodevelopmental Outcomes

[EMBED Visio.Drawing.15]

Following the process outlined above, 16 studies were ultimately identified as Group 1 and subsequently evaluated to determine if relevant and necessary data were available for further analysis. Five of the Group 1 studies conducted analyses or presented data that would allow the EPA to quantitatively connect the results of the BBDR model to incremental changes in adverse neurodevelopmental effects. The following sections provide more detail about the process involved in each step of the evaluation and the results.

### 5.3.1 Results of Step 1

Out of the 71 papers identified in the literature review, the EPA identified 40 that are not compatible with BBDR model results. These Group 3 studies are listed in [ REF \_Ref425459365 \h ] along with the reason the study is not compatible with BBDR model results. [ REF \_Ref425459365 \h ] also lists the life stage of thyroid hormone measurement to aid in comparison across studies. Summaries of each study in Group 3 are presented in Appendix F.

Table [ SEQ Table \\* ARABIC ]. List of Studies Identified in Step 1 and Categorized as Group 3 with Rationale for Categorization

		Life Stage
		Life Stage of Thyroid Hormone Measurem
		Hormone
		Measurem
Study	Rationale	ent <sup>a</sup>

Andersen et al. [ ADDIN EN.CITE < EndNote > < Cite	The study	Pregnant
ExcludeAuth="1"> <author>Andersen</author> <year>2017</year> <recnum>2023</recnum> <displaytext>(2017)</displaytext> <record><rec-< th=""><th>design is not</th><th>woman</th></rec-<></record>	design is not	woman
number>2023 <foreign-keys><key <="" app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" th=""><th>directly</th><th></th></key></foreign-keys>	directly	
timestamp="1533923738">2023 <ref-type name="Journal Article">17</ref-type> <contributors><authors><author>Andersen,</author></authors></contributors>	relevant to the	
S.L. <author>Andersen, S.</author> <author>Liew, Z.</author> Vestergaard, P. <author>Olsen,</author>	BBDR model	
J. <title>&gt;&lt;title&gt;Maternal thyroid function in early pregnancy and neuropsychological performance of the child at 5&lt;/td&gt;&lt;td&gt;results. This&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;/tr&gt;&lt;tr&gt;&lt;td&gt;years of age</title> <secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title> <periodical><full-title>Journal of Clinical</full-title></periodical>	study	
Endocrinology and Metabolism <pages>660-</pages>	compared	
670 <volume>103</volume> <number>2</number> <dates><year>2017</year></dates> <urls></urls> ]	women with	
Maternal Thyroid Function in Early Pregnancy and Neuropsychological Performance of the Child at 5 Years of Age	thyroid	
material rayiona randien in Early rreginally and recare poyone logical resistance of the entitle at a rear of rige	dysfunction to	
	those without	
	dysfunction.	
	While there are	
	data on fT4	
	concentrations	
	and	
	neurodevelopm	
	ental	
	outcomes,	
	these	
	evaluations	
	included	
	women with	
	overt	
	hypothyroidism	
	and therefore	
	cannot be	
	evaluated with	
	respect only to	
	hypothyroxine	
	mia. For	
	several	
	endpoints (i.e.,	
	motor function	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
	and behavioral	
	problems) they	
	did note that	
	higher	
	associations	
	related to	
	thyroid	
	dysfunction	
	were likely due	
	to	
	hypothyroxine	
	mia, but no	
	quantitative	
	analysis was	
	provided.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Casey et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]	The study	Pregnant
Treatment of Subclinical Hypothyroidism or Hypothyroxinemia in Pregnancy	design is not directly relevant to the BBDR model results. This study examined the impact of treatment of maternal hypothyroxine mia on infant neurodevelopm ental outcome, not the relationship between maternal hypothyroxine mia and infant neurodevelopm	woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Chevrier et al. [ ADDIN EN.CITE <endnote><cite< td=""><td>The study</td><td>Pregnant</td></cite<></endnote>	The study	Pregnant
ExcludeAuth="1"> <author>Chevrier</author> <year>2011</year> <recnum>5</recnum> 50/RecNum>	population's exposure window (i.e., when the thyroid hormone measurements are taken) does not overlap with the outputs for the BBDR model. This study measured maternal thyroid hormones after 16 GW.	woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Craig et al. [ ADDIN EN.CITE <endnote><cite< td=""><td>The study</td><td>Pregnant</td></cite<></endnote>	The study	Pregnant
ExcludeAuth="1"> <author>Craig</author> <year>2012</year> <recnum>8</recnum> <displaytext>(2012)</displaytext> <record><rec-< td=""><td>population's</td><td>woman</td></rec-<></record>	population's	woman
number>8 <foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047630">8</key><td>exposure</td><td></td></foreign-keys>	exposure	
keys> <ref-type name="Journal Article">17</ref-type> <contributors><author>Craig, W Y</author><author>Allan, W C</author><author>Kloza,</author></contributors>	window does	
E M <author>Pulkkinen, A J</author> <author>Waisbren, S</author> Spratt, D I <author>Palomaki, G</author>	not overlap	
E <author>Neveaux, L M</author> <author>Haddow, J E</author> Mid-gestational maternal free thyroxine	with the	
concentration and offspring neurocognitive development at age two years <secondary-title>Journal of Clinical Endocrinology and</secondary-title>	outputs for the	
Metabolism <full-title>Journal of Clinical Endocrinology and Metabolism</full-title> <pages>E22-</pages>	BBDR model.	
E28 <volume>97</volume> <number>1</number> <section>E22</section> <dates><year>2012</year></dates> <urls></urls> <electronic-< td=""><td>This study</td><td></td></electronic-<>	This study	
resource-num>10.1210/jc.2011-1772]	measured	
Mid-Gestational Maternal Free Thyroxine Concentration and Offspring Neurocognitive Development at Age Two Years	maternal	
	thyroid	
	hormones after	
	16 GW.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Cusick and Georgieff [ ADDIN EN.CITE < EndNote > < Cite	The study	N/A -
ExcludeAuth="1"> <author>Cusick</author> <year>2016</year> <recnum>306</recnum> <displaytext>(2016)</displaytext> <record><rec< td=""><td>design is not</td><td>Thyroid</td></rec<></record>	design is not	Thyroid
number>306 <foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832047">306</key><key< td=""><td>directly</td><td>measurem</td></key<></foreign-keys>	directly	measurem
app="ENWeb" db-id="">0 <ref-type name="Journal Article">17</ref-type> <contributors><authors><author>Cusick, S.</author></authors></contributors>	relevant to the	ents were
E. <author>Georgieff, M. K.</author> <auth-address>Department of Pediatrics, University of Minnesota School of</auth-address>	BBDR model	not taken
Medicine, Minneapolis, MN. Department of Pediatrics, University of Minnesota School of Medicine, Minneapolis, MN. Electronic address:	results. This is	
georg001@umn.edu. <titles><title>The role of nutrition in brain development: The golden opportunity of the "First 1000&lt;/td&gt;&lt;td&gt;a review paper&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;/tr&gt;&lt;tr&gt;&lt;td&gt;Days"</title><secondary-title>The Journal of Pediatrics</secondary-title></titles> <periodical><full-title>The Journal of Pediatrics<td>that provides a</td><td></td></full-title></periodical>	that provides a	
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0022-3476 (Linking)</isbn> <accession-num>27266965</accession-num> <urls><related-< td=""><td>discussion and</td><td></td></related-<></urls>	discussion and	
ırls> <url><style face="underline" font="default" size="100%">http://www.ncbi.nlm.nih.gov/pubmed/27266965</style></url> <td>does not</td> <td></td>	does not	
urls> <custom2>PMC4981537</custom2> <electronic-resource-num>10.1016/j.jpeds.2016.05.013<td>provide</td><td></td></electronic-resource-num>	provide	
num>]	quantitative	
The Role of Nutrition in Brain Development: The Golden Opportunity of the "First 1000 Days"	information to	
	measure	
	neurodevelopm	
	ental	
	outcomes.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Dosiou and Medici [ ADDIN EN.CITE <endnote><cite excludeauth="1">cAuthor&gt;Dosiou/Author&gt;<pre>Year&gt;2017</pre><pre>/Year&gt;</pre><pre>RecNum&gt;325</pre><pre>/RecNum&gt;</pre><pre>DisplayText&gt;<pre>(2017)</pre><pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/DisplayText&gt;<pre>/Di</pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></cite></endnote>	The study design is not directly relevant to the BBDR model results. This is a review paper that does not provide quantitative information on the relationship between fT4 and outcomes.	Pregnant woman
Fetene et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Fetene</author><year>2017</year><recnum>1951</recnum><displaytext>(2017)</displaytext><record><rec- number="">1951<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1516819484">1951</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Fetene, Dagnachew Muluye</author><author>Betts, Kim S</author><author>Alati, Rosa</author>&gt;&gt;</contributors><title>Mechanisms in endocrinology: Maternal thyroid dysfunction during pregnancy and behavioural and psychiatric disorders of children: a systematic review</title><secondary-title>European Journal of Endocrinology</secondary-title><periodical><full-title>European Journal of Endocrinology</full-title></periodical><periodical><periodical><full-title>European Journal of Endocrinology R273 /pages&gt;<volume>177 /volume&gt;<number>5</number><dates><year>2017 //year&gt;</year></dates><isbn>0804- 4643 //isbn&gt;<url>&gt; //record&gt;</url></isbn></volume></full-title></periodical></periodical></rec-></record></cite> /EndNote&gt;]  Maternal Thyroid Dysfunction during Pregnancy and Behavioural and Psychiatric Disorders of Children: A Systematic Review</endnote>	The study design is not directly relevant to the BBDR model results. This is a review article that does not provide a quantitative meta-analysis.	Pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Furnica et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]  Update on a New Controversy in Endocrinology: Isolated Maternal Hypothyroxinemia	The study design is not directly relevant to the BBDR model results. This is a review paper that does not provide quantitative information on the relationship between fT4 and outcomes.	Pregnant woman
Getahun et al. [ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;ExcludeAuth=1"&gt;Excl</cite></endnote>	The study design is not directly relevant to the BBDR model results. This paper evaluated the relationship between maternal hypothyroidism and autism spectrum disorders.	Pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Ghassabian et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><a 1"="" href="ADDIN EN.CITE &lt;EndNote&gt;&lt;Cite ExcludeAuth="></a></cite></endnote>		

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Gutleb, Cambier, and Serchi [ ADDIN EN.CITE <endnote><cite< td=""><td>The study</td><td>N/A -</td></cite<></endnote>	The study	N/A -
ExcludeAuth="1"> <author>Gutleb</author> <year>2016</year> <recnum>309</recnum> <displaytext>(2016)</displaytext> <record><rec-< td=""><td>design is not</td><td>Thyroid</td></rec-<></record>	design is not	Thyroid
number>309 <foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832134">309</key><key< td=""><td>directly</td><td>measurem</td></key<></foreign-keys>	directly	measurem
app="ENWeb" db-id="">0 <ref-type name="Journal Article">17</ref-type> <contributors><author>&gt;<author>Cauthor&gt;</author></author></contributors>	relevant to the	ents were
A. <author>Cambier, S.</author> <author>Serchi, T.</author> <titles><title>Impact of endocrine distruptors on the&lt;/td&gt;&lt;td&gt;BBDR model&lt;/td&gt;&lt;td&gt;not taken&lt;/td&gt;&lt;/tr&gt;&lt;tr&gt;&lt;td&gt;thyroid system</title><secondary-title>Hormone Research in Paediatrics</secondary-title>&gt;<periodical><full-title>Hormone Research in</full-title></periodical></titles>	results. This	
Paediatrics <pages>271-</pages>	study reviewed	
278 <volume>86</volume> <dates><year>2016</year></dates> <urls>]</urls>	the effects that	
Impact of Endocrine Disruptors on the Thyroid Hormone System	endocrine	
	disruptors may	
	have on the	
	thyroid and did	
	not evaluate	
	the impact of	
	thyroid	
	hormones on	
	neurodevelopm	
	ent in offspring.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
ExcludeAuth="1"> <a erc.number="" href="https://erc.number&gt;&lt;a href=" https:=""><a a="" erc.number]<="" erc.number]<a="" href="https://erc.number]&lt;/a&gt;&lt;a href=" https:=""><a< td=""><td>The study design is not directly relevant to the BBDR model results. The study concentrated on the impact of maternal hypothyroidism and offspring neurodevelopm ent. Data were not presented to allow for an understanding of the relationship between incremental changes in thyroid hormones, or maternal hypothyroxine mia, and neurodevelopm ental changes.</td><td>Pregnant</td></a<></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a></a>	The study design is not directly relevant to the BBDR model results. The study concentrated on the impact of maternal hypothyroidism and offspring neurodevelopm ent. Data were not presented to allow for an understanding of the relationship between incremental changes in thyroid hormones, or maternal hypothyroxine mia, and neurodevelopm ental changes.	Pregnant

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Hamza, Hewedi, and Sallam [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Hamza</author>Year&gt;2013<recnum>14</recnum><displaytext>(2013)</displaytext><record><rec-number>14</rec-number>141414141415/rec-number&gt;15/rec-number&gt;16/reign-keys&gt;<key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047634">14</key><ref-type name="Journal Article">17</ref-type><contributors><author>&gt;author&gt;&gt;autho</author></contributors></record></cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. The study measured thyroid hormones in autistic children.	Child
Hollanders et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ]  No Association Between Transient Hypothyroxinemia of Prematurity and Neurodevelopmental Outcome in Young Adulthood	The study design is not directly relevant to the BBDR model results. The study assesses the impact of transient hypothyroxine mia of prematurity in pre-term or very low birthweight infants.	Neonate, child

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Hollanders et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Hollanders</author><pear>2016</pear><recnum>312</recnum><displaytext>(2016)</displaytext><record><rec-number>312</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832203">312</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>&gt;<author>&gt;<author>&gt;<author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></contributors></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The study assessed the impact of transient hypothyroxine mia of prematurity in pre-term infants or very low birthweight	Neonate, child
urls> <url>http://www.ncbi.nlm.nih.gov/pubmed/27343725</url> <electronic-resource-num>10.1016/j.psyneuen.2016.06.008</electronic-resource-num> ]  Transient Hypothyroxinemia of Prematurity and Problem Behavior in Young Adulthood	infants.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Hoshiko, Grether, Windham, Smith, and Fessel [ ADDIN EN.CITE < EndNote < Cite ExcludeAuth="1" > (Author > Hoshiko < Author > Year > 2011 < / Year > (RecNum > 17 < / RecNum > 10 > (Display Text > (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011) < (2011)	The study population's exposure window does not overlap with the outputs for the BBDR model. Thyroid hormone samples were measured in neonates within 24 hours of birth.	Neonate
Korzeniewski et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Korzeniewski</author><year>2013</year><recnum>20</recnum><displaytext>(2013)</displaytext><record><rec- number="">20<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047638">20</key><ref-type name="Journal Article">17</ref-type><contributors><author>Korzeniewski, S J</author><author>Pinto-Martin, J  A</author><author>Whitaker, A H</author><author>Feldman, J F</author><author>Lorenz, J M</author><author>Levy, S E</author><author>Movsas, T Z</author><author>Pappas, A</author><author>Pappas, A<td>The study design is not directly relevant to the BBDR model results. The study only examined low-birthweight/preterm infants.</td><td>Neonate</td></author></contributors></foreign-keys></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The study only examined low-birthweight/preterm infants.	Neonate

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Lazarus et al. [ ADDIN EN.CITE <endnote><cite< td=""><td>The study</td><td>Pregnant</td></cite<></endnote>	The study	Pregnant
ExcludeAuth="1"> <author>Lazarus</author> <year>2012</year> <recnum>2025</recnum> <displaytext>(2012)</displaytext> <record><rec-< td=""><td>design is not</td><td>woman,</td></rec-<></record>	design is not	woman,
number>2025 <foreign-keys><key <="" app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" td=""><td>directly</td><td>infant</td></key></foreign-keys>	directly	infant
timestamp="1533926600">2025 <ref-type name="Journal Article">17</ref-type> <contributors><author>Lazarus, J.,</author></contributors>	relevant to the	
<author>Bestwick, J. P., </author> <author>Channon, S., </author> <author>Paradice, R., </author> <author>Maina, A.,</author>	BBDR model	
<title>&gt;Antenatal thyroid screening and childhood cognitive function</title> <secondary-title>New England Journal of Medicine</secondary-title> > <periodical><full-title>New England Journal of Medicine</full-title></periodical>	results. This study	
title>]	examined the	
Antenatal Thyroid Screening and Childhood Cognitive Function	impact of	
Antenatal Phytola delecting and Orlination Cognitive Function	treatment of	
	maternal	
	hypothyroxine	
	mia on infant	
	neurodevelopm	
	ental outcome,	
	not the	
	relationship	
	between	
	maternal	
	hypothyroxine	
	mia and infant	
	neurodevelopm ental outcome.	
	ental outcome.	

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Leung [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Leung</author><year>2015</year><recnum>315</recnum><displaytext>(2015)</displaytext><record><rec- number="">315<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832294">315</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><authors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author< td=""><td>The study design is not directly relevant to the BBDR model results. This is a narrative discussion of Noten et al. (2015).</td><td>Pregnant woman</td></author<></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></authors></authors></authors></contributors></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This is a narrative discussion of Noten et al. (2015).	Pregnant woman
Lyall et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Lyall</author><year>2016</year><recnum>316</recnum><displaytext>(2016)</displaytext><record><rec- number="">316<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832313">316</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Lyall, K.<author><author>Anderson, M.<author>Kauthor&gt;Kharrazi, M.<author><author>Windham, G. C.<author><authors><authors><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><a< td=""><td>The study population's exposure window does not overlap with the outputs for the BBDR model. The study only measured thyroid hormones (T4 and TSH) in bloodspots of newborn infants.</td><td>Neonate</td></a<></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></authors></authors></author></author></author></author></author></author></author></authors></contributors></rec-></record></cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. The study only measured thyroid hormones (T4 and TSH) in bloodspots of newborn infants.	Neonate

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Man, Brown, and Serunian [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Man</author><year>1991</year><recnum>38</recnum><displaytext>(1991)</displaytext><record><rec- number="">38<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062207">38</key><ref-type name="Journal Article">17</ref-type><contributors><author>Man, E B</author><author>Brown, J F</author><author>Brown, J F</author><setion>Serunian, S A</setion></contributors><titles><title>Maternal hypothyroxinemia: Psychoneurological deficits of progeny</title><secondary-title>Annals of Clinical and Laboratory Science</secondary-title>Annals of Clinical and Laboratory Science<pages>227- 239</pages><volume>21 21 230 1901-7370/91/0700-0227 29 29 20 <p< td=""><td>The study design is not directly relevant to the BBDR model results. The study measured butanolextractable iodine and did not include data on T4 or fT4.</td><td>Pregnant woman</td></p<></volume></titles></foreign-keys></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The study measured butanolextractable iodine and did not include data on T4 or fT4.	Pregnant woman
Min et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA]  Maternal Hypothyroxinemia-Induced Neurodevelopmental Impairments in the Progeny	The study design is not directly relevant to the BBDR model results. This is a review paper that does not provide quantitative information to measure neurodevelopm ental outcomes.	Pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Moog et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Moog</author><year>2017</year><recnum>1952</recnum><displaytext>(2017)</displaytext><record><rec- number="">1952<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1516819484">1952</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>&gt;<author>&gt;<author>Moog, Nora K</author><author>Entringer, Sonja</author><author>Heim, Christine</author><author>Wadhwa, Pathik D</author><author>Kathmann, Norbert</author><author>Buss, Claudia</author></author></author>&gt;<ititle>&gt;<title>Influence of maternal thyroid hormones during gestation on fetal brain development</title><econdary-title>Neuroscience <li>Neuroscience<pages>68-100</pages><volume>342</volume></li> <li>Influence of Maternal Thyroid Hormones During Gestation on Fetal Brain Development</li> </econdary-title></ititle></contributors></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This is a review article that does not provide a quantitative meta-analysis.	Pregnant woman
Muñoz, Figueras, & Puig [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Muñoz</author><year>2009</year><recnum>2207</recnum><displaytext>(2009)</displaytext><record><recnumber>2207<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1542054961">2207</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>&gt;author&gt;&gt;author&gt;Muñoz, M.</author><author>Figueras, F.</author><author>Puig, M</author>&gt;<atitle>&gt;title&gt;La hipotiroxinemia gestacional se asocia al síndrome de déficit de atención e hiperactividad<secondary-title>Progresos de Obstetricia y Ginecología</secondary-title>//full-sitle&gt;//full-sitle&gt;//full-sitle&gt;//periodical&gt;/</atitle></contributors></recnumber></record></cite></endnote>	The study population's exposure window does not overlap with the output of the BBDR model. This study measures thyroid hormones in the third trimester. <sup>b</sup>	Pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Pearce [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Pearce</author><year>2015</year><recnum>324</recnum><displaytext>(2015a)</displaytext><record><rec- number="">324<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832512">324</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors>&lt;</authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></contributors></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This is a narrative discussion of Modesto et al. (2015).	Pregnant woman
Pérez-Lobato et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;ExcludeAuth=="1"&gt;ExcludeAuth====================================</cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. Thyroid hormones were measured in children (mean age was 9.8 years).	Child

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Pharoah, Connolly, Ekins, and Harding [ ADDIN EN.CITE < EndNote > Cite ExcludeAuth="1" > Author> Pharoah/Author> Year> 1984 < RecNum> 39 DisplayText> (1984) In the control of the control of the Children of the Children and Motor Performance of the Children  Pharoah, Connolly, Ekins, and Harding [ ADDIN EN.CITE < EndNote > Cite ExcludeAuth="1" > (1984)/DisplayText> < (1984)	The study design is not directly relevant to the BBDR model results. The study population was severely iodine- deficient.	Pregnant woman
Scratch et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Scratch</author><year>2014</year><recnum>27</recnum><displaytext>(2014)</displaytext><record><rec- number="">27<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047642">27</key><ref-type name="Journal Article">17</ref-type><contributors><author>Scratch, S E</author><author>Hunt, R W</author><author>Thompson, D K</author><author>Ahmadzai, Z M</author><author>Doyle, L W</author><author>Inder, T E</author>Anderson, P J&lt;</contributors></foreign-keys></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The study only examined very pre-term births.	Neonate

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Soldin et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Soldin</author><year>2002</year><recnum>28</recnum>CDisplayText&gt;(2002)<record><rec-number>28</rec-number><foreign-keys><key app="EN" db-="" id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047642">28</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Soldin, O.P.</author><author>Nandedkar, A.K.N.</author><author>Japal, K.M.</author><author>Stein, M</author><author>Mosee, S</author><author>Magrab, P</author><author>Lai, S</author><author>Laim, S.H.</author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><author><aut< td=""><td>The study population's exposure window does not overlap with outputs for the BBDR model. The study only measured thyroid hormones in infants.</td><td>Neonate</td></aut<></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></contributors></record></cite></endnote>	The study population's exposure window does not overlap with outputs for the BBDR model. The study only measured thyroid hormones in infants.	Neonate
Soldin, Lai, Lamm, and Mosee [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Soldin</author><year>2003</year><recnum>29</recnum><displaytext>(2003)<record><rec-number>29</rec-number><foreign-keys><key app="EN" db-="" id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047643">29</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Soldin, O.P.</author><author>Lai, S</author><author>Laimm, S H</author><author>Mosee, S</author></contributors><tittle>&gt;<tittle>Lack of a relation between human neonatal thyroxine and pediatric neurobehavioral disorders<secondary-title>Thyroid<periodical><full-title>Thyroid</full-title></periodical><pages>193- 204</pages><volume>13 volume&gt;<number>2 <pre>/number&gt;<section>193 <pre>/section&gt;<dates><year>2003 <pre>/year&gt; <pre>/dates&gt;<urls></urls><electronic-resource-num>10.1089/105072503321319503 <pre>/electronic-resource-num&gt; <pre>/record&gt; <pre>/Cite&gt;</pre></pre> <pre>/EndNote&gt;]</pre> <pre>Lack of a Relation Between Human Neonatal Thyroxine and Pediatric Neurobehavioral Disorders</pre></pre></electronic-resource-num></pre></pre></year></dates></pre></section></pre></number></volume></secondary-title></tittle></tittle></record></displaytext></cite></endnote>	The study population's exposure window does not overlap with outputs for the BBDR model. The study only measured thyroid hormones in infants.	Neonate

Study	Rationale	Life Stage of Thyroid Hormone Measurem enta
Suárez-Rodríguez, Azcona-San Julián, and Alzina de Aguilar [ ADDIN EN.CITE <endnote><cite auth="1" exclude="">&lt; Author&gt;Suarez-Rodríguez<year>2012</year><recnum>1888</recnum><displaytext>(2012)</displaytext><record><rec-number>1888</rec-number>1888<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1497462591">1888</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Suárez-Rodríguez, Marta</author>&gt;author&gt;Azcona-San Julián, Cristina<author>Alzina de Aguilar, Valentin</author></contributors><title>Hypothyroxinemia during pregnancy: the effect on neurodevelopment in the child</title><secondary-title>International Journal of Developmental Neuroscience</secondary-title><pages>435-438 438</pages><volume>30 volume&gt;<number>6 number&gt;6 humber&gt;<dates> year&gt;2012 year&gt;</dates><urls> urls&gt; record&gt; Cite &gt;<endnote>] Hypothyroxinemia during Pregnancy: The Effect on Neurodevelopment in the Child</endnote></urls></number></volume></record></cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. This study measured maternal thyroid hormones after 16 GW.	Pregnant woman
Tiemeier and Korevaar [ADDIN EN.CITE <endnote><cite excludeauth="1">Author&gt;TiemeierYear&gt;2016<recnum>326</recnum><displaytext>(2016)</displaytext><record><rec- number="">326<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832556">326</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors<<authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><autho< td=""><td>The study design is not directly relevant to the BBDR model results. This is a commentary article that does not provide quantitative information to measure neurodevelopm ental outcomes.</td><td>N/A - Thyroid measurem ents were not taken</td></autho<></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors<<authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></authors></contributors></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This is a commentary article that does not provide quantitative information to measure neurodevelopm ental outcomes.	N/A - Thyroid measurem ents were not taken

Study	Rationale	Life Stage of Thyroid Hormone Measurem enta
Velasco et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="2009/Pear&gt; 2009  /Year&gt; 2009 /Years of Life</cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. This study measured maternal thyroid hormones after 16 GW.	Pregnant woman
Vermiglio et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Vermiglio</author><year>2004</year><recnum>33</recnum>CisplayText&gt;(2004)<record><rec-number>33</rec-number>&lt;50reign-keys&gt;<key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047646">33</key><ref-type name="Journal Article">17</ref-type><contributors><author>Vermiglio, F</author><author>Castagna, M</author><author>Moleti, M</author><author>Scaffidi, G</author><author>Castagna, M G</author><author>Mattina, F</author><author>Violi, M A</author><author>Crisà, A</author><author>Artemisia, A</author><author>Trimarchi, F</author><author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;<author>Citles&gt;&lt;</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></contributors></record></cite></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. This study measured maternal thyroid hormones after 16 GW.	Pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
F. Williams et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Williams</author><year>2012</year><recnum>34</recnum><displaytext>(2012)</displaytext><record><rec- number="">34<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047646">34</key><ref-type name="Journal Article">17</ref-type><contributors><author>Williams, F</author><author>Watson, J</author><author>Ogston, S</author><author>Hume, R</author><author>Willatts, P</author><author>Visser, T</author><author>Scottish Preterm Thyroid Group,</author></contributors></foreign-keys></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This study evaluated the effect of maternal thyroid dysfunction on pre-term infants.	Neonate, pregnant woman
F. Williams et al. [ ADDIN EN.CITE <endnote> Cite ExcludeAuth="1"&gt; Author&gt; Williams  Vear&gt; 2013  <recnum> 35 </recnum> Display Text&gt; (2013)  <record> <rec-number> 35 </rec-number> &lt;5 /rec-number&gt; &lt;6 /rec-number&gt; &lt;5 /rec-number&gt; &lt;5 /rec-number&gt; &lt;6 /rec-number &lt;6 /rec-number&gt; &lt;6 /rec-number &lt;6 /</record></endnote>	The study population's exposure window does not overlap with the outputs for the BBDR model. Maternal fT4 was not evaluated in association with the neurodevelopm ental outcome.	Fetus, pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
F. L. R. Williams et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ] Supplemental lodide for Preterm Infants and Developmental Outcomes at 2 Years: An RCT	The study population's exposure window does not overlap with the output of the BBDR model. The study evaluated the impact of iodine supplementatio n in pre-term infants on subsequent neurodevelopm ent.	Infant
Willoughby [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Willoughby</author><year>2011</year><recnum>42</recnum><displaytext>(2011)</displaytext><record><rec- number="">42<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062209">42</key><ref-type name="Thesis">32</ref-type><contributors><author>Willoughby, K</author></contributors><tittle>Effects of early thyroid hormone deficiency on autobiographical memory and hippocampal structure and function during late childhood and early adolescence<secondary-title>Graduate Department of Psychology</secondary-title><pages>228</pages><volume>Doctorate of Philosophy //olume&gt;<dates><year>2011 //year&gt;</year></dates><publisher>University of Toronto</publisher><urls> // EndNote&gt;  Effects of Early Thyroid Hormone Deficiency on Autobiographical Memory and Hippocampal Structure During Late Childhood and Early Adolescence</urls></volume></tittle></foreign-keys></rec-></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The only results for T4 are related to cases of congenital hypothyroidism .	Child, neonate, pregnant woman

Study	Rationale	Life Stage of Thyroid Hormone Measurem ent <sup>a</sup>
Willoughby, McAndrews, and Rovet [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Willoughby</author><year>2014</year><recnum>36</recnum><displaytext>(2014)</displaytext><record><rec-number>36</rec-number>&gt;6<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047648">36</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Willoughby, K</author><author>McAndrews, M P</author><author>Rovet, JF</author><author>Rovet, JF</author><author>Hypothyroidism on offspring hippocampus and memory<author>HypothyroidHypothyroid</author></author></contributors></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. The results of the study pertain to the offspring of hypothyroid women.	Pregnant woman
Wong et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Wong</author><year>2016</year><recnum>330</recnum><displaytext>(2016)</displaytext><record><recnumber>330</recnumber>330530530530530530530540 imestamp="1491838588"&gt;330<ref-type name="Journal Article">17</ref-type><contributors><author>&gt;author&gt;&gt;au</author></contributors></record></cite></endnote>	The study design is not directly relevant to the BBDR model results. This study evaluated the risk of pre-term birth from thyroid disease.	N/A - Thyroid measurem ents were not taken

	Life Stage of Thyroid Hormone Measurem
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<sup>&</sup>lt;sup>a</sup> Pregnant woman: thyroid hormone measurement taken anytime during pregnancy; Fetus: thyroid hormone levels were measured from cord blood at birth. Neonate: thyroid hormone levels were measured from a source other than cord blood between birth and 28 days of age. Child: thyroid hormone levels were measured after 1 year of age.

b Assessment based upon presented results from Thompson et al. (2018). The original study is written in Spanish, and the full text was not reviewed at the time of report publication.

#### 5.3.2 Results of Step 2

After identifying 40 studies to categorize as Group 3 in Step 1, 31 studies remained for categorization in Step 2. It was determined whether or not each study presented an analysis that was based on a continuous measure of thyroid hormones. Fifteen studies were placed into Group 2 given that they only presented categorical analyses, and 16 were placed into Group 1 because they presented a continuous regression analysis. Group 1 studies may also have categorical results, but they were categorized based on the presence of any analysis based on a continuous measure of thyroid hormone levels. Summaries of the studies in each group are presented in subsequent sections.

### **Summary of Group 2 Studies**

Of the 31 studies evaluated in Step 2, 15 studies provide categorical data that assist in understanding the implications of altered maternal thyroid hormone levels and potential adverse neurodevelopmental outcomes in their offspring. However, these studies do not present data to inform an analysis that allows for an understanding of the impact of incremental changes in thyroid hormones and subsequent neurodevelopmental impacts. Study results related to cognitive outcomes are summarized in [ REF \_Ref424646544 \h ], behavioral outcomes (including ADHD) are including in [ REF \_Ref517280585 \h ], and studies related to autism are summarized in [ REF \_Ref517280594 \h ] (studies may be repeated between or within tables if they considered a variety of endpoints).

### Table [ SEQ Table \\* ARABIC ]. Overview of Studies Categorized as Group 2 - Cognition

	Hypothyro   Populati xinemia   on (pmol/L,   Analyze unless Endpoint( Summary d for Key otherwise s) of Key   Findings   noted) <sup>a</sup>   Test(s) Findings	Stage of Thyroid Hormon e Measur ement <sup>b</sup>
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ExcludeAuth="1"> <author>Grau</author> Year>2015RecNum>13CDisplayText>(2015) IDISPLAY (2015) I	455 Spanish mother- child pairs at age 1 assessm ent 289 Spanish mother- child pairs at age 6-8 assessm ent (1 yr 1st trimester cases = 47; 1 yr 2nd trimester cases = 42; 6-8 yr 1st trimester cases = 33; 6-8 yr 2nd trimester cases = 28) 331	fT4 < 10 <sup>th</sup> percentile (< 13.7) (first trimester) fT4 < 10 <sup>th</sup> percentile (11.5 pmol/L) (second trimester)	Child Intelligenc e Wechsler Intelligenc e Scale for Children (WISC) IV	Children of mothers with hypothyroxi nemia in the first and second trimester from an iodine-sufficient area did not have adverse neurodevel opmental outcomes (assessed at age 6 to 8 with the WISC) associated with levels of fT4.	Pregnan t woman
	mother-	percentile <sup>c</sup>	Simula	no	t woman

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Controlled Antenatal Thyroid Screening II: Effect of Treating Maternal Sub-Optimal Thyroid Function on Child Cognition	child pairs from the UK (ncases = 81; 40 of which were treated with thyroxine )		Wechsler Intelligenc e Scale for Children IV	difference between mean IQ scores of the children born to hypothyroxi nemic mothers and those born to mothers with normal gestational thyroid function (p = 0.875).	

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Päkkilä et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Pakkila</author><year>2015</year><recnum>323<displaytext>(2015)</displaytext><record><rec-number>323</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832494">323</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Päkkilä, F</author><author>Männistö, T</author><author>Hartikainen, A-L</author><author>Ruokonen, A. </author><author>et al.,</author></contributors><tittles><tittle>Maternal and child's thyroid function and child's intellect and scholastic performance<secondary-title>Thyroid<periodical><full-title>Thyroid</full-title></periodical><pages>1363- 1374</pages><volume>25</volume><number>12</number><dates><year>2015</year></dates><urls><urls><electronic-resource-num>10.1787/9789264091450- e</electronic-resource-num></urls></urls></secondary-title></tittle></tittles></record><td>5,295 Finnish mother- child pairs with sufficient maternal serum samples (n<sub>cases</sub> = 67)</td><td>fT4 &lt; 11.4 in the first trimester fT4 &lt; 11.1 in the second trimester<sup>d</sup></td><td>Child IQ categorize d as having mild cognitive limitation if IQ was between 50 and 85 points or severe intellectual deficiency if IQ was less than 50 points Wechsler Intelligenc e Scale for Children- Revised</td><td>Maternal thyroid dysfunction was not associated with the prevalence and odds of the child's severe intellectual deficiency or mild cognitive limitation.</td><td>Pregnan t woman, child</td></recnum></cite></endnote>	5,295 Finnish mother- child pairs with sufficient maternal serum samples (n <sub>cases</sub> = 67)	fT4 < 11.4 in the first trimester fT4 < 11.1 in the second trimester <sup>d</sup>	Child IQ categorize d as having mild cognitive limitation if IQ was between 50 and 85 points or severe intellectual deficiency if IQ was less than 50 points Wechsler Intelligenc e Scale for Children- Revised	Maternal thyroid dysfunction was not associated with the prevalence and odds of the child's severe intellectual deficiency or mild cognitive limitation.	Pregnan t woman, child

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															d for Key	otherwise	е	s)	of Key	Measur
						St	udy								Findings	noted)a		Test(s)	Findings	ement <sup>b</sup>

Costeira et al. (2011)	86	fT4 < 10 <sup>th</sup>	Mental	Children	Neonate
Costeira et al. (2011) Psychomotor Development of Children from an Iodine-Deficient Region	86 Portugue se mother- child pairs from iodine- deficient region (ncases = 8)	percentile (< 9.0 pg/mL or < 11.6)	Mental Developm ent Index Psychomot or Developm ent Index Bayley Scales of Infant Developm ent	born to mothers with low fT4 had increased odds of mild-to- severe delay in psychomot or developme nt compared to children born to mothers with fT4 within the population reference range. Neonatal thyroid status did not influence developme	neonate , pregnan woman
Júlvez et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Julvez</author><year>2013</year><recnum>1: Num&gt;<displaytext>&lt;(2013)</displaytext><rec-number>18<foreign-keys><key app="EN" db-<="" td=""><td>8Spanish mother-</td><td>Low fT4 (less than 10<sup>th</sup> (8.89), 5<sup>th</sup> (8.39)</td><td>Mental Developm ent Index</td><td>nt.  Children born to mothers with lower fT4 (&lt;10th</td><td>Pregnan t woman</td></key></foreign-keys></rec-number></recnum></cite></endnote>	8Spanish mother-	Low fT4 (less than 10 <sup>th</sup> (8.89), 5 <sup>th</sup> (8.39)	Mental Developm ent Index	nt.  Children born to mothers with lower fT4 (<10th	Pregnan t woman

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)a	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"	child	or 2.5 <sup>th</sup>	Psychomot	percentile	
timestamp="1432047637">18 <ref-type <="" name="Journal" td=""><td>pairs</td><td>(8.10)</td><td>or</td><td>(&lt;8.89</td><td></td></ref-type>	pairs	(8.10)	or	(<8.89	
Article">17 <contributors><author>&gt;Júlvez,</author></contributors>	(n <sub>cases</sub> =	percentile)	Developm	pmol/L))	
J <author>Alvarez-Pedrerol, M</author> <author>Rebagliato,</author>	164)		ent Index	show a decrease in	
M <author>Garcia-</author>			Bayley Scales of	Bayley	
Esteban, R <author>Lertxundi, N</author> <author>Espada,</author>			Infant	score	
M <author>Cauthor&gt;Caut</author>			Developm	compared	
J <titles><title>Thyroxine levels during pregnancy&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;td&gt;ent&lt;/td&gt;&lt;td&gt;to children&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;/tr&gt;&lt;tr&gt;&lt;td&gt;in healthy women and early child neurodevelopment</title><secondary-< td=""><td></td><td></td><td></td><td>born to</td><td></td></secondary-<></titles>				born to	
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Thyroxine Levels During Pregnancy in Healthy Women and Early Child Neurodevelopment					

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Li et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1">ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="1"&gt;ExcludeAuth="2!   Author&gt;<foreign-keys><key <="" app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" td="">   timestamp="1432047639"&gt;21 Ikey&gt;<ref-type name="Journal Article">Include Article"&gt;Include Exclude Exc</ref-type></key></foreign-keys></cite></endnote>	57 Chinese mother-child pairs (ncases = 19)	fT4 < 5 <sup>th</sup> percentile (< 11.9)	Mental Developm ent Index Psychomot or Developm ent Index Bayley Scales of Infant Developm ent	Children of hypothyroxi nemic women had significantly lower mean mental and psychomot or developme nt (Bayley) scores at 25-30 months compared to children of non-hypothyroxi nemic women.	Pregnan t woman

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Other Cognitive Tests					
Berbel et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Berbel</author><year>2009</year><recnum>4<displaytext>(2009)</displaytext><record><rec-number>4<foreign-keys><key app="EN" db-="" id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047627">4</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Berbel, P</author><author>Mestre, J L</author><author>Santamaría, A</author><author>Palazón, I</author><author>Franco, A</author><author>Graells, M</author><author>González-Torga, A</author><author>Morreale de Escobar, G</author></contributors><titiles><titile>Delayed neurobehavioral development in children born to pregnant women with mild hypothyroxinemia during the first month of gestation: the importance of early iodine supplementation<secondary-title>Thyroid<periodical><full-title>Thyroid</full-title></periodical><pages>511- 519</pages><volume>19 yolume&gt;<number>5 /number&gt;<section>511  /section&gt; dates&gt;<year>2009 /year&gt;<urls> /urls&gt;<electronic-resource- num="">10.1089/thy2008.0341 /electronic-resource-num&gt;/record&gt; /Cite&gt; /EndNote&gt;] Delayed Neurobehavioral Development in Children Born to Pregnant Women with Mild Hypothyroxinemia During the First Month of Gestation: The Importance of Early Iodine Supplementation</electronic-resource-></urls></year></section></number></volume></secondary-title></titile></titiles></rec-number></record></recnum></cite></endnote>	43 Spanish mother- child pairs in 3 groups based on fT4 level and timing of iodine suppleme ntationf	fT4 < 10 <sup>th</sup> percentile (< 10.6)	Neurobeha vioral performan ce (includes gross and fine motor coordinatio n, language skills, and socializatio n) Brunet- Lézine scale	A delay of 6–10 weeks in iodine supplement ation of hypothyroxi nemic mothers at the beginning of gestation increased the risk of neurodevel opmental delay.	Pregnan t woman, child

ExcludeAuth="1"> <author>Grau</author> Year>2015 <recnum>13</recnum> ClisplayText>(2015) <record><rec-number>13</rec-number><foreign-keys><key app="EN" db-id="z910avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047634">13</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Grau, G</author><author>Aguayo, A</author><author>Vela, A</author><author>Aniel-Quiroga, A</author><author>Afulor&gt;Aguayo, A</author><author>Martina, G</author><author>Martinez-Indart, L</author><author>Martul, P</author><author>Castaño, L</author><author>Rica, I</author><author>Castaño, L</author><author>Rica, I</author><author>Martinez-Indart, L</author>Rica, II<author>Martinez-Indart, L</author>Rica, II<author>Rica, II</author><author>Rica, II</author><author><auth< th=""><th>455 Spanish mother- child pairs at age 1 assessm ent 289 Spanish mother- child pairs at age 6-8 assessm ent (1 yr 1st trimester cases = 47; 1 yr 2nd trimester cases = 42; 6-8 yr 1st trimester cases = 33; 6-8 yr 2nd trimester cases = 28) 331</th><th>fT4 &lt; 10<sup>th</sup> percentile (&lt; 13.7) (first trimester) fT4 &lt; 10<sup>th</sup> percentile (11.5) (second trimester)</th><th>Neurocogn itive performan ce Brunet-Lézine scale</th><th>Children of mothers with hypothyroxi nemia in the first and second trimester from an iodine-sufficient area did not have adverse neurodevel opmental outcomes (assessed at age 1 with the Brunet-Lézine scale) associated with levels of fT4.</th><th>Pregnan</th></auth<></author></contributors></record>	455 Spanish mother- child pairs at age 1 assessm ent 289 Spanish mother- child pairs at age 6-8 assessm ent (1 yr 1st trimester cases = 47; 1 yr 2nd trimester cases = 42; 6-8 yr 1st trimester cases = 33; 6-8 yr 2nd trimester cases = 28) 331	fT4 < 10 <sup>th</sup> percentile (< 13.7) (first trimester) fT4 < 10 <sup>th</sup> percentile (11.5) (second trimester)	Neurocogn itive performan ce Brunet-Lézine scale	Children of mothers with hypothyroxi nemia in the first and second trimester from an iodine-sufficient area did not have adverse neurodevel opmental outcomes (assessed at age 1 with the Brunet-Lézine scale) associated with levels of fT4.	Pregnan
	mother-	percentile <sup>c</sup>	term	with	t woman

Controlled Antenatal Thyroid Screening II: Effect of Treating Maternal Sub-Optimal Thyroid Function on Child Cognition	child pairs from the UK (n <sub>cases</sub> = 81; 40 of which were treated with thyroxine )	memory and motor function Developm ental Neuropsyc hological Assessme nt (NEPSY) second edition	isolated hypothyroxi nemia, there were no differences in mean scores were identified from NEPSY assessmen ts (memory for designs; memory for designs delayed; fingertip tapping dominant hand; fingertip tapping non- dominant hand; narrative memory) (thyroxine treated n=38
			(thyroxine

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
			•	and NEPSY - List Memory (treated n=30, untreated n=26, p=0.361).	

Päkkilä et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"> ExcludeAuth="1"&gt; ExcludeAuth="1"&gt; ExcludeAuth="1"&gt; ExcludeAuth="1"&gt; ExcludeAuth="1"&gt; ExcludeAuth="1"  ExcludeAuth="1"&gt; ExcludeAuth="1"  ExcludeAuth="2015)  ExcludeAuth="2015)  ExcludeAuth="2015]  ExcludeAuth=="2015]  Ex</cite></endnote>	5,295 Finnish mother-child pairs with sufficient maternal serum samples (n <sub>cases</sub> = 67)	fT4 < 11.4 in the first trimester fT4 < 11.1 in the second trimester <sup>d</sup>	Child School Performan ce Teacher evaluation questionna ire and postal questionna ire (self- evaluation including the Youth Self- Report)	Female offspring of mothers with subclinical hypothyroid ism had more self-evaluated difficulties in mathematic s than did females of mothers with thyroid function within the population reference range. Male offspring of hypothyroxi nemic mothers repeated a school class more often than did males of mothers with normal thyroid function.	Pregnan t woman, child

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
				Male offspring of	
				hypothyroxi	
				nemic	
				mothers	
				also had a	
				higher odds	
				of having	
				difficulties	
				in Finnish	
				and/or	
				math.	

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
van Mil et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>van Mil</author><year>2012</year><recnum>45</recnum><displaytext>(2012)<record><rec-number>45</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437061977">45</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>van Mil, N.H.</author><author>Steegers-Theunissen, Regine P.M.</author><author>Bongers-Schokking, Jacoba J.</author><author>El Marroun, Hanan</author><author>Ghassabian, Akhgar</author><author>Hofman, Albert</author><author>Jaddoe, Vincent W.V.</author><author>Verhulst, Frank C.</author><author>de Rijke, Yolanda B.</author><author>Steegers, Eric A.P.</author><author>Tiemeier, Henning</author><author>title&gt;<citile><author>title&gt;<contributors><titile><author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>title&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totle&gt;<author>totl</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></titile></contributors></author></citile></author></contributors></record></displaytext></cite></endnote>	2,621 Dutch mother- child pairs (ncases = 476)	fT4 < 10 <sup>th</sup> percentile (11.82)	Fetal and infant head size  Sonograph y Child cognitive developme nt Parent Report of Children Abilities	Maternal hypothyroxi nemia was associated with an increase in fetal head and infant head size. In addition, maternal hypothyroxi nemia was associated with a statistically significant increased risk of cognitive delay.	Pregnan t woman
				•	

	Definition			
	of			Life
	Hypothyro			Stage of
Populati	xinemia			Thyroid
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on	(pmol/L,			Hormon
Analyze	unless	Endpoint(	Summary	e
d for Key	otherwise		of Key	Measur
		_ S)		
Study Findings	noted) <sup>a</sup>	Test(s)	Findings	ement <sup>b</sup>
			-	
Meta-Analysis				

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)a	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Thompson et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Thompson</author><year>2018</year><recnum>1956</recnum><displaytext>(2018)</displaytext><record><rec-number>1956</rec-number><foreign-keys><key app="EN" db-id="2910avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1516819485">1956</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Thompson, William</author><author>Russell, Ginny</author><author>Baragwanath, Genevieve</author><author>Matthews, Justin</author><author>Vaidya, Bijay</author><author>Thompson-Coon, Jo</author><tittles><tittle>Maternal thyroid hormone insufficiency during pregnancy and risk of neurodevelopmental disorders in offspring: A systematic review and meta-analysis<secondary-title>Clinical endocrinology</secondary-title><periodical><full-title>Clinical Endocrinology</full-title></periodical><dates><year>2018 //ear&gt; //author&gt; <pre>//author&gt; <pre>//author&gt; <pre>//author&gt; <pre>//author&gt;</pre> <pre>//aut</pre></pre></pre></pre></year></dates></tittle></tittles></contributors></record></cite></endnote>	15,147 mother- child pairs identified through multiple studiesh	Multiple studies evaluated, no set definition <sup>i</sup>	Child intellectual disability Various - Multiple studies evaluated	Compared to those children born to euthyroid mothers, children born to hypothyroxi nemic mothers were significantly more likely to show signs of intellectual impairment (odds ratio (OR) 1.63, 95% CI 1.03 to 2.56, p=0.04).	Pregnan t woman

Study	Populati on Analyze d for Key Findings	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)a	Endpoint( s)   Test(s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Wang et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]  Maternal Thyroxine Levels During Pregnancy and Outcomes of Cognitive Development in Children	8,273 mother- child pairs (n <sub>cases</sub> = 702) <sup>i</sup>	Multiple studies evaluated, no set definition	Cognitive developme nt Various – multiple studies evaluated	Meta- analysis showed low maternal fT4 level was significantly associated with increase in risk of delayed cognitive developme nt in offspring.	Pregnan t woman

- <sup>a</sup> Some studies did not explicitly define hypothyroxinemia, and instead examined the impact of altered thyroid hormone levels below a defined percentile cut point and of altered neurodevelopment. For the purposes of this review, these percentile cut points are considered to be analogous to a definition for hypothyroxinemia.
- <sup>b</sup> Pregnant woman: thyroid hormone measurement taken anytime during pregnancy. Fetus: thyroid hormone levels were measured from cord blood at birth. Neonate: thyroid hormone levels were measured from a source other than cord blood between birth and 28 days of age. Child: thyroid hormone levels were measured after 1 year of age.
- <sup>c</sup> The value of the hypothyroxinemic cut point was not identified in the paper.
- d Päkkilä et al. [ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Pakkila</Author><Year>2015</Year><RecNum>323</RecNum><DisplayText>(2015)</DisplayText><record><rec-number>323</rec-number><5oreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832494">323</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Päkkilä, F</author><author>Männistö, T</author><author>Hartikainen, A-L</author><author>Full-title><author><author>+title></title><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical><periodical></pr>
- 1374</pages><volume>25</volume>cnumber>12</number><dates><year>2015</year></dates><urls></urls><electronic-resource-num>10.1787/9789264091450-e</electronic-resource-num>(p. 1365), but low fT4 is not defined. The lower range of the reference interval is 11.4 pmol/L for fT4 for the first trimester, and 11.09 pmol/L is the lower bound of the reference interval for the second trimester. These are assumed to be the cut points used to define low fT4.
- f Group 1: fT4> 20th percentile at 4-6 gestational weeks and term, supplemented with 200 μg/day potassium iodine through term and lactation. Neurocognitive evaluation was done on children with mothers fT4>20th percentile at 4-6 weeks and at term. (n=13) Group 2: mildly hypothyroxinemic at 12-14 gestational weeks, supplemented with 200 μg/day potassium iodine through term and lactation. Neurocognitive evaluation done on children with maternal fT4 <10th percentile at 12-14 weeks and fT4>20th percentile at term (n=12) Group 3: enrolled at full term at which time they were supplemented with 200 μg/day potassium iodine throughout lactation. Neurocognitive evaluations were done on children born to a mother who had fT4<10th percentile at term. (n=18)

  9 The number of cases of hypothyroxinemia included in the Fan and Wu [ ADDIN EN.CITE <EndNote><Cite
- ExcludeAuth="1"><Author>Fan</Author>Year>2016</Year><RecNum>307</RecNum><DisplayText>(2016)</DisplayText><record><rec-number>307</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832068">307</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors><authors>
- h The number of cases of hypothyroxinemia included in the Thompson et al. (2018) meta-analysis are summed from Supplementary Table 1 of the paper. Cases in the pooled analysis were not identified.
- Where multiple cut points of TSH and fT4 were measured, the most extreme cutoff was used, and when an outcome was measured both against a continuous thyroid hormone measure and against a cut point (for example, cut point of the 10th percentile fT4 for hypothyroxinemia), the cut point was used.
- i The number of cases with hypothyroxinemia was determined from Table 1 of the paper and by deriving values from within individual studies if the number of cases was not indicated.

Table [ SEQ Table \\* ARABIC ]. Overview of Studies Categorized as Group 2 – Behavior (Including ADHD)

			·		
	Popul	Definition			Life
	ation	of			Stage
	Analy	Hypothyr			of
	zed	oxinemia			Thyroid
	for	(pmol/L,	Endpo		Hormon
	Key	unless	int(s)	Summary	е
	Findi	otherwise	Test(	of Key	Measur
Study	ngs	noted) <sup>a</sup>	s)	Findings	ementb

Ghassabian, Henrichs, and Tiemeier [ ADDIN EN.CITE < EndNote > < Cite	3,903	Severe:	ADHD	This paper	Fetus,
ExcludeAuth="1"> <author>Ghassabian</author> <year>2014</year> <recnum>11<td>Dutch</td><td>fT4 &lt; 5th or</td><td>Proble</td><td>reviews</td><td>Pregnan</td></recnum>	Dutch	fT4 < 5th or	Proble	reviews	Pregnan
um> <displaytext>(2014)</displaytext> <record><rec-number>11</rec-number><foreign-< td=""><td>mothe</td><td>10<sup>th</sup></td><td>ms</td><td>the</td><td>t woman</td></foreign-<></record>	mothe	10 <sup>th</sup>	ms	the	t woman
keys> <key <="" app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" td=""><td>r-child</td><td>percentile</td><td>Child</td><td>literature</td><td></td></key>	r-child	percentile	Child	literature	
timestamp="1432047632">11 <ref-type <="" name="Journal" td=""><td>pairs</td><td>of sample</td><td>Behavi</td><td>from the</td><td></td></ref-type>	pairs	of sample	Behavi	from the	
Article">17 <contributors><author>Ghassabian,</author></contributors>	(n <sub>cases</sub>	(exact	or	Generatio	
A <author>Henrichs, J</author> <author>Tiemeier,</author>	= 129)	value not	Checkli	n R cohort and	
HImpact of mild thyroid hormone		defined)	st for	presents	
deficiency in pregnancy on cognitive function in children: lessons from the Generation R			Toddle	novel	
Study <secondary-title>Best Practice &amp; Discourse and Clinical Endocrinology</secondary-title>			rs	results	
& Metabolism>eriodical> <full-title>Best Practice &amp;</full-title>				related to	
Research and Clinical Endocrinology & Detabolism / full-				ADHD and	
title> <pages>221-</pages>				maternal	
232 <volume>28</volume> <number>2</number> <section>221</section> <dates><y< td=""><td></td><td></td><td></td><td>hypothyro</td><td></td></y<></dates>				hypothyro	
ear>2014 <urls></urls> <electronic-resource-< td=""><td></td><td></td><td></td><td>xinemia.</td><td></td></electronic-resource-<>				xinemia.	
num>10.1016/j.beem.2013.04.008]				They	
Impact of Mild Thyroid Hormone Deficiency in Pregnancy on Cognitive Function in Children: Lessons from the				found	
Generation R Study				maternal	
oonoralion in Glady				hypothyro xinemia to	
				he	
				associated	
				with ADHD	
				problems	
				with some	
				of the	
				associatio	
				n	
				attenuated	
				when	
				considerin	
				g maternal factors	
				such as	

Study	Popul ation Analy zed for Key Findi ngs	Definition of Hypothyr oxinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpo int(s)   Test( s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
				age and education level. The review also summarize s results presented in Henrichs et al. (2010), van Mil et al. (2012), Ghassabia n et al. (2011), and Ghassabia n et al. (2012).	

		7	· <sub>T</sub>	γ	7
Kooistra, Crawford, van Baar, Brouwers, and Pop [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Kooistra</author>Year&gt;2006<recnum>47</recnum><displaytext>(2006)</displaytext><record><rec-number>47</rec-number><forcign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437065646">&lt;47<td>204 Dutch mothe r-child pairs (ncases = 108)</td><td>fT4 &lt; 10<sup>th</sup> percentile (&lt; 10.4)</td><td>Neonat al Develo pment Neonat al Behavi oral Assess ment Scale</td><td>Infants of women with hypothyro xinemia at GW 12 had a lower neonatal Behavioral Assessme nt Scale orientation index compared with infants of non-hypothyro xinemic women. Neonatal thyroid hormone levels were not assessed in regard to neurodeve lopment.</td><td>Neonate , pregnan t woman</td></key></forcign-keys></record></cite></endnote>	204 Dutch mothe r-child pairs (ncases = 108)	fT4 < 10 <sup>th</sup> percentile (< 10.4)	Neonat al Develo pment Neonat al Behavi oral Assess ment Scale	Infants of women with hypothyro xinemia at GW 12 had a lower neonatal Behavioral Assessme nt Scale orientation index compared with infants of non-hypothyro xinemic women. Neonatal thyroid hormone levels were not assessed in regard to neurodeve lopment.	Neonate , pregnan t woman
Päkkilä et al. [ ADDIN EN.CITE < EndNote > < Cite	5,131	fT4 < 11.4	Child	Maternal	Pregnan
ExcludeAuth="1"> <author>Päkkilä</author> <year>2014</year> <recnum>23</recnum>	Finnis	in the first	ADHD	hypothyro	t woman
<pre><displaytext>&lt;(2014)</displaytext></pre> /DisplayText> <record><rec-number>23</rec-number><foreign-< pre=""></foreign-<></record>	h mothe	trimester		xinemia was not	

Study	Popul ation Analy zed for Key Findi ngs	Definition of Hypothyr oxinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpo int(s)   Test( s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
keys> <key <="" app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" td=""><td>r-child</td><td>fT4 &lt; 11.1</td><td>Finnish</td><td>significantl</td><td></td></key>	r-child	fT4 < 11.1	Finnish	significantl	
timestamp="1432047640">23 <ref-type <="" name="Journal" td=""><td>pairs</td><td>in the</td><td>translat</td><td>y</td><td></td></ref-type>	pairs	in the	translat	y	
Article">17 <contributors><author>Päkkilä,</author></contributors>	(n <sub>cases</sub>	second	ions of	associated	
F <author>Männistö, T</author> <author>Pouta, A</author> <author>Hartikainen,</author>	= 66)	trimester	the Rutter	with teacher	
A <author>Ruokonen, A</author> <author>Surcel, H</author> <author>Bloigu,</author>			scale	ratings of	
A <author>Vääräsmäki, M</author> <author>Järvelin,</author>			B2	ADHD	
M <author>Moilanen, I</author> <author>Suvanto,</author>			52	symptoms	
E <titles><title>The impact of gestational thyroid&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;td&gt;in children&lt;/td&gt;&lt;td&gt;&lt;/td&gt;&lt;/tr&gt;&lt;tr&gt;&lt;td&gt;hormone concentrations on ADHD symptoms of the child</title><secondary-title>Journal of</secondary-title></titles>				at 8 years	
Clinical Endocrinology and Metabolism <periodical><full-< td=""><td></td><td></td><td></td><td>of age.</td><td></td></full-<></periodical>				of age.	
title>Journal of Clinical Endocrinology and Metabolism <pages>E1-</pages>					
E8 <volume>99</volume> <number>1</number> <section>E1</section> <dates><yea< td=""><td></td><td></td><td></td><td></td><td></td></yea<></dates>					
r>2014 <urls></urls> <electronic-resource-num>10.1210/jc.2013-</electronic-resource-num>					
2943]					
The Impact of Gestational Thyroid Hormone Concentration on ADHD Symptoms of the Child					

Study	Popul ation Analy zed for Key Findi ngs	Definition of Hypothyr oxinemia (pmol/L, unless otherwise noted) <sup>a</sup>	Endpo int(s)   Test( s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measur ement <sup>b</sup>
Meta-Analysis					
Thompson et al. [ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Thompson</author><year>2018</year><recnum>1956<displaytext>(2018)</displaytext><record><rec-number>1956<foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1516819485">1956</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Thompson, William</author><author>Russell, Ginny</author><author>Baragwanath, Genevieve</author><author>Matthews, Justin</author><author>Vaidya, Bijay</author><author>Thompson-Coon, Jo</author><futiles><title>Maternal thyroid hormone insufficiency during pregnancy and risk of neurodevelopmental disorders in offspring: A systematic review and meta-analysis</title><secondary-title>Clinical endocrinology<periodical><full-title>Clinical Endocrinology</full-title></periodical><dates><year>2018 <li>Auternal Thyroid Hormone Insufficiency during Pregnancy and Risk of Neurodevelopmental Disorders in Offspring: A Systematic Review and Meta-Analysis</li></year></dates></secondary-title></futiles></contributors></rec-number></record></recnum></cite></endnote>	15,14 7 mothe r-child pairs identifi ed throug h multipl e studie s°	Multiple studies evaluated, no set definition <sup>d</sup>	ADHD Variou s - Multipl e studies evaluat ed	No associatio n was found between maternal hypothyro xinemia and ADHD in children (OR 1.34, 95% CI 0.17 to 10.47, p=0.78).	Pregnan t woman

	Popul	Definition			Life
	ation	of			Stage
	Analy	Hypothyr			of
	zed	oxinemia			Thyroid
	for	(pmol/L,	Endpo		Hormon
	Key	unless	int(s)	Summary	e
	Findi	otherwise	Test(	of Key	Measur
Study	ngs	noted) <sup>a</sup>	s)	Findings	ement <sup>b</sup>

<sup>&</sup>lt;sup>a</sup> Some studies did not explicitly define hypothyroxinemia, and instead examined the impact of altered thyroid hormone levels below a defined percentile cut point and of altered neurodevelopment. For the purposes of this review, these percentile cut points are considered to be analogous to a definition for hypothyroxinemia.

<sup>&</sup>lt;sup>b</sup> Pregnant woman: thyroid hormone measurement taken anytime during pregnancy. Fetus: thyroid hormone levels were measured from cord blood at birth. Neonate: thyroid hormone levels were measured from a source other than cord blood between birth and 28 days of age. Child: thyroid hormone levels were measured after 1 year of age.

<sup>&</sup>lt;sup>c</sup> The number of cases of hypothyroxinemia included in the Thompson et al. (2018) meta-analysis are summed from Supplementary Table 1 of the paper. Cases in the pooled analysis were not identified.

<sup>&</sup>lt;sup>d</sup> Where multiple cut points of TSH and fT4 were measured, the most extreme cut point was used, and when an outcome was measured both against a continuous thyroid hormone measure and against a cut point (for example, cut point of the 10<sup>th</sup> percentile fT4 for hypothyroxinemia), the cut point was used.

## Table [ SEQ Table \\* ARABIC ]. Overview of Studies Categorized as Group 2 – Autism

Study	Popul ation Analy zed for Key Findin gs	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)	Endpoi nt(s)   Test( s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measure ment <sup>b</sup>
Brown et al. (2015)  Maternal Thyroid Autoantibody and Elevated Risk of Autism in a National Birth Cohort	1,916 Finnis	None	Autism Record	There was	Pregnant woman
	h mother -child		ed diagno sis	associatio n found between	
	pairs mother		from psychi	maternal fT4 and	
	s with fT4 (n <sub>cases</sub>		atric hospita	autism (cases, mean	
	= 958; cases		admissi ons	maternal fT4 (SD)	
	define d as		and outpati	14.7 (2.0); compariso	
	being diagno		ent visits of	n subjects, mean (SD)	
	sed with autism		childho od	= 14.7 (1.92)) OR = 1.09	
	)		autism	(95%Cl 0.52-	
				12.26, <i>p</i> =0.83)	

for Key Findin	(pmol/L, unless otherwise	Endpoi nt(s)	Summary	Thyroid Hormon e Measure
Findin	otherwise	Test(	of Key	Measure
Study gs	noted) <sup>a</sup>		Findings	ment <sup>b</sup>

Study	Popul ation Analy zed for Key Findin gs	Definition of Hypothyro xinemia (pmol/L, unless otherwise noted)a	Endpoi nt(s)   Test( s)	Summary of Key Findings	Life Stage of Thyroid Hormon e Measure ment <sup>b</sup>
Thompson et al. [ ADDIN EN.CITE <endnote><cite excludeauth="1"><author>Thompson</author>Year&gt;2018<recnum>1956</recnum>CDisplayText&gt;(2018)<record><rec-number>1956</rec-number><forcign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1516819485">1956</key></forcign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Author&gt;Matthews, Justin</author>Author&gt;Vaidya, Bijay<author>Thompson-Coon, Jo</author><!--</td--><td>15,147 mother -child pairs identifi ed throug h multipl e studie s<sup>c</sup></td><td>Multiple studies evaluated, no set definition<sup>d</sup></td><td>Autism Variou s - Multipl e studies evaluat ed</td><td>Of two studies identified, a study that used a continuous predictor (i.e., maternal fT4) found no associatio n between maternal hypothyrox inemia and autism, but the other study that used a binary predictor (i.e., maternal hypothyrox inemia) did.</td><td>Pregnant</td></contributors></record></cite></endnote>	15,147 mother -child pairs identifi ed throug h multipl e studie s <sup>c</sup>	Multiple studies evaluated, no set definition <sup>d</sup>	Autism Variou s - Multipl e studies evaluat ed	Of two studies identified, a study that used a continuous predictor (i.e., maternal fT4) found no associatio n between maternal hypothyrox inemia and autism, but the other study that used a binary predictor (i.e., maternal hypothyrox inemia) did.	Pregnant

Po	opul	Definition			
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2	zed	xinemia			Thyroid
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Fi	indin	otherwise	Test(	of Key	Measure
Study	gs	noted) <sup>a</sup>	s)	Findings	mentb

<sup>&</sup>lt;sup>a</sup> Some studies did not explicitly define hypothyroxinemia, and instead examined the impact of altered thyroid hormone levels below a defined percentile cut point and of altered neurodevelopment. For the purposes of this review, these percentile cut points are considered to be analogous to a definition for hypothyroxinemia.

<sup>&</sup>lt;sup>b</sup> Pregnant woman: thyroid hormone measurement taken anytime during pregnancy. Fetus: thyroid hormone levels were measured from cord blood at birth. Neonate: thyroid hormone levels were measured from a source other than cord blood between birth and 28 days of age. Child: thyroid hormone levels were measured after 1 year of age.

<sup>&</sup>lt;sup>c</sup> The number of cases of hypothyroxinemia included in the Thompson et al. (2018) meta-analysis are summed from Supplementary Table 1 of the paper. Cases in the pooled analysis were not identified.

<sup>&</sup>lt;sup>d</sup> Where multiple cut points of TSH and fT4 were measured, the most extreme cut point was used, and when an outcome was measured both against a continuous thyroid hormone measure and against a cut point (for example, cut point of the 10<sup>th</sup> percentile fT4 for hypothyroxinemia), the cut point was used.

Of the 15 papers identified as Group 2, 11 evaluated cognitive outcomes [ADDIN EN.CITE ADDIN EN.CITE ADDIN EN.CITE.DATA ], four evaluated behavioral outcomes [ADDIN EN.CITE ADDIN EN.CITE.DATA ] and two evaluated autism [ADDIN EN.CITE ADDIN EN.CITE.DATA ] as they relate to maternal thyroid hormone dysfunction, most often defined as maternal hypothyroxinemia (Thompson et al. (2018) evaluated all three categories of endpoints and is thus included in each count listed and in each table). Of the cognition studies, there were consistent results associating maternal hypothyroxinemia with decreased performance on the Bayley Scales. None of the Group 2 studies evaluated found a significant difference in IQ when comparing results in offspring of mother with hypothyroxinemia to those of mothers without hypothyroxinemia. However, three meta-analyses (including one full systematic review, Thompson et al., 2018) conclude that maternal hypothyroxinemia is associated with increased risk of cognitive delay, intellectual impairment, or lower scores on performance tests [ADDIN EN.CITE ADDIN EN.CITE.DATA]. These review papers evaluated studies that are included in Groups 1, 2, and 3 of this analysis and

]. These review papers evaluated studies that are included in Groups 1, 2, and 3 of this analysis and subsequently provide a more comprehensive understanding of the relationship between maternal hypothyroxinemia and altered cognitive development in offspring.

The three non-review studies that evaluated behavioral outcomes (Ghassabian, Henrichs, and Tiemeier, 2014; Kooistra, Crawford, van Baar, Brouwers, and Pop, 2006; Päkkilä et al., 2014) found mixed evidence relating maternal hypothyroxinemia with increased risk of ADHD or decreased scores on behavioral assessment tests. Thompson et al. (2018) did not find an association between maternal hypothyroxinemia and ADHD.

Lastly, one independent study (Brown et al., 2015) and one review paper (Thompson et al., 2018) were categorized as Group 2 and evaluated maternal hypothyroxinemia as associated with autism. Brown et al. (2015) did not find an association. Thompson et al. (2018) concluded that the association between maternal hypothyroxinemia and autism was uncertain.

Brief descriptions of all studies are presented in Appendix G.

## **Summary of Group 1 Studies**

After categorizing 15 studies in Group 2, 16 remained to be categorized as Group 1. The EPA focused on this set of 16 studies due to the presence of an analysis that evaluated maternal fT4 as a continuous variable, as it related to offspring neurodevelopment. Group 1 includes studies that found statistically significant changes and those that did not. In many but not all of these studies the general results, though not statistically significant, were consistent with the studies used for further quantitative analyses by the EPA. The EPA presents and characterizes all of the Group 1 studies below.

Studies categorized in Group 1 present data that may allow for the extraction of a function relating a continuous measure of thyroid hormone levels to some measure of neurodevelopment. In combination with the BBDR model, this function may be used to identify the dose of perchlorate associated with a defined change in neurodevelopment. Several of the studies in this group did not include an estimated regression function, but instead presented a scatterplot that depicts a regression or correlation analysis.<sup>11</sup> The graphical data presented in these studies was digitized to allow for the estimation of a

5-71

The authors of each of these studies were contacted from July 2015 to February 2017 to determine if they have the underlying data for these figures. The EPA has not received responses from many of the authors as of the date of release of this document.

function. The papers categorized as Group 1 are summarized in [ REF \_Ref491174324 \h ] in Section [ REF \_Ref492391106 \n \h ], including findings related to the continuous relationship presented in the paper and information necessary to evaluate the papers as outlined above in Step 3. Many of these papers present regression analyses on more than one endpoint or age group. The majority of the continuous relationships presented pertain to the entire range of maternal fT4 and are not specific to just the offspring of hypothyroxinemic mothers. However, many of the Group 1 studies also present a categorical analysis comparing neurodevelopmental outcomes in the offspring of hypothyroxinemic and non-hypothyroxinemic mothers; this is discussed further in Section [ REF \_Ref482275380 \r \h ]. To facilitate a better understanding of the relationship between maternal thyroid hormones and the neurodevelopmental outcomes, the study summaries have been grouped into four categories: cognitive (including IQ, Bayley Scales, and other measures), behavior (including ADHD and others), autism, and other endpoints not otherwise captured.

## Cognitive (IQ)

Detailed Summary of Ghassabian et al. [ADDIN EN.CITE <EndNote><Cite
ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum>
<DisplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key
app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
timestamp="1432047632">10</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Ghassabian, A</author><author>Marroun, H
E</author><author>Peeters, R P</author><author>Jaddoe, V W</author><author>Hofman,
A</author><author>Verhulst, F C</author><author>Tiemeier, H</author><author>White,
T</author></authors></contributors><titles><title>Downstream effects of maternal
hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age
children</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondarytitle></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</fulltitle></periodical><pages>23832390</pages><volume>99</wolume><number>7</number><section>2383
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Downstream Effects of Maternal Hypothyroxinemia in Early Pregnancy: Nonverbal IQ and Brain Morphology in School-Age Children

## Ghassabian et al. [ ADDIN EN.CITE < EndNote > < Cite

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volume>99
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2014</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2013-4281</electronicresource-num></record></Cite></EndNote>] aimed to examine whether children born to mothers with hypothyroxinemia during early pregnancy would show decreased nonverbal IQ at 6 years of age. Along with reductions in IQ, Ghassabian et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum><D isplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047632">10</key></foreign-keys></ref-type name="Journal Article">17</reftype><contributors><authors><author>Ghassabian, A</author><author>Marroun, H E</author><author>Peeters, R P</author><author>Jaddoe, V W</author><author>Hofman, A</author><author>Verhulst, F C</author><author>Tiemcier, H</author><author>White, T</author></authors></contributors><title>Downstream effects of maternal hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age children</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondarytitle></titles><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</fulltitle></periodical><pages>2383-2390</pages><volume>99</volume><number>7</number><section>2383</section><dates><year> 2014</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2013-4281</electronicresource-num></record></Cite></EndNote>] hypothesized that children of hypothyroxinemic mothers would display corresponding structural changes in the size of several key areas of the brain: the cortex, hippocampus, and corpus callosum. An additional aim was to investigate the relationship between subclinical hypothyroidism (normal fT4 and high TSH) in mothers and IQ in children. Participants (n = 3.727) were recruited from the Generation R birth cohort in the Netherlands; all singleton births with complete information on maternal thyroid levels and children's nonverbal IQ at age 6 were included in the study. A subset of children participating in the study (n = 652) underwent magnetic resonance imaging (MRI) testing to assess volume of various brain regions. Maternal thyroid function (fT4, TSH, and TPO Ab) were measured at a mean of 13.5 GW. Hypothyroxinemia was defined as fT4 levels below 10.99 pmol/L (the 5th percentile of the sample) and subclinical hypothyroidism as normal fT4 and high TSH (defined as TSH > 2.5 mIU/L or > 3.0 mIU/L). Children completed a well-validated Dutch nonverbal intelligence test (the Snijders-Oomen Niet-Verbale Intelligentie test), which consisted of Mosaics to assess spatial visualization skills and Categories to assess abstract reasoning skills. Linear regression models were used to investigate the association between maternal thyroid function and children's nonverbal IQ and brain volumetric measures. Maternal fT4 and TSH were divided by their SD in the regression analyses to determine the impact of SD increments or decrements in these parameters. Regression analyses with IQ as the outcome were adjusted for ethnicity, birth order, maternal age, BMI, marital status, maternal history of smoking, educational levels, maternal psychopathology during pregnancy, household income, and gestational age at time of thyroid sampling.

Ghassabian et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum><D isplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047632">10</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Ghassabian, A</author><author>Marroun, H E</author><author>Peeters, R P</author><author>Jaddoe, V W</author><author>Hofman, A</author><author>Verhulst, F C</author><author>Tiemeier, H</author><author>White,

T</author></authors></contributors><titles><title>Downstream effects of maternal hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age children</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>2383-

2390</pages><volume>99</volume><number>7</number><section>2383</section><dates><year>2014</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2013-4281</electronic-resource-num></record></Cite></EndNote>] did not find an association between continuous measures of maternal thyroid function (fT4, TSH, and TPO Ab positivity) and children's IQ at the age of 6 years. An analysis of continuous fT4 and its relationship with IQ in just the children of hypothyroxinemic mothers was not conducted. However, Ghassabian et al. [ ADDIN EN.CITE <EndNote><Cite

title></titles><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>2383- 2390</pages><volume>99</volume><number>7</number><section>2383</section><dates><year> 2014</year></dates><urls><urls><lectronic-resource-num>10.1210/jc.2013-4281</lectronic-resource-num></record></Cite></EndNote>] did find that children of mothers with hypothyroxinemia scored an average of 4.3 IQ points lower (p=0.001, 95% CI: -6.68, -1.81) on IQ tests than children of mothers with thyroid function within the population reference range. No

significant associations were found between maternal hypothyroxinemia and any of the brain volume

measures assessed. Maternal subclinical hypothyroidism did not show a significant relationship with children's IQ. A summary of results can be found in [ REF Ref517338346 \h ].

Table [ SEQ Table \\* ARABIC ]. Summary of Results from Ghassabian et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum><DisplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key app="EN" db-

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Article">17</ref-type><contributors><authors><author>Ghassabian,

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P</author><author>Jaddoe, V W</author><author>Hofman,

A</author><author>Verhulst, F C</author><author>Tiemeier,

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num>10.1210/jc.2013-4281</electronic-resource-num></record></Cite></EndNote>]

		Child's IQ at 6 years of ag	e
Maternal Thyroid Function	β	95% CI	p-value
TSH per SD	0.04	-0.17, 0.26	0.70
fT4 per SD	-0.24	-0.61, 0.14	0.21
Hypothyroxinemia	-4.32	-6.68, -1.81	0.001

Source: Ghassabian et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum><DisplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047632">10</key></foreign-keys><ref-type name="Journal Article">17</ref-

type><contributors><author>Ghassabian, A</author>Ghassabian, A</author>Marroun, H E</author>Guthor>Peeters, R P</author>Jaddoe, V W</author>Cauthor>Hofman, A</author>Cauthor>Verhulst, F C</author>Cauthor>Tiemeier, H</author>Cauthor>White, T</author>Cauthors>Contributors>Citiles>Citile>Downstream effects of matemal hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age children</title>Clinical Endocrinology and Metabolism</school-septimes of Clinical Endocrinology and Metabolism</school-septimes of Clini

2390</pages><volume>99</volume><number>7</number><section>2383</section><dates><year>2014</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2013-4281</electronic-resource-num></record></Cite></EndNote>], Table 2.

Models were adjusted for child's ethnic background and birth order and maternal age, BMI, marital status, maternal history of smoking, educational levels, maternal psychopathology in pregnancy, household income, and time of blood sampling in pregnancy. Hypothyroxinemia (n = 129): 0.03 < TSH < 2.5 mIU/L and fT4 < 10.99 pmol/L (less than the 5<sup>th</sup> percentile).

**Key Finding:** Maternal hypothyroxinemia during early pregnancy (mean GW = 13.5) was associated with an average reduction of 4.3 points (p = 0.001, 95% CI: -6.68, -1.81) in children's nonverbal IQ at age 6. Continuous measures of maternal thyroid function (TSH, fT4) across the entire range (not the hypothyroxinemic range) and subclinical hypothyroidism were not significantly associated with nonverbal IQ in 6-year-old children.

Detailed Summary of Korevaar et al. [ADDIN EN.CITE < EndNote > < Cite

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M.</author><author>Muetzel, Ryan</author><author>Medici, Marco</author><author>Chaker,
Layal</author><author>de Rijke, Yolanda

B. </author><author>Steegers, Eric A. P. </author><author>Visser, Theo

J.</author><author>White, Tonya</author><author>Tiemeier,

Henning</author><author>Peeters, Robin

P.</author></contributors><titles><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabe

Endocrinology</secondary-title></titles><periodical><full-title>The Lancet Diabetes & amp; Endocrinology</full-title></periodical><pages>35-

43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>22138587</isbn><urls><electronic-resource-num>10.1016/s2213-8587(15)00327-7</electronic-resource-num></cite></EndNote>]

Association of Maternal Thyroid Function during Early Pregnancy with Offspring IQ and Brain Morphology in Childhood: A Population-Based Prospective Cohort Study

## Korevaar et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Korevaar</Author><Year>2016</Year><RecNum>313</RecNum><Di splayText>(2016)</DisplayText><record><rec-number>313</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1491832236">313</key>< key app="ENWeb" db-id="">0</key></foreign-keys>< reftype name="Journal Article">17</ref-type>< contributors>< author> Xorevaar, Tim I.

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B.</author><author>Visser, Theo J.</author><author>White, Tonya</author><author>Tiemeier, Henning</author><author>Peeters, Robin

P.</author></authors></contributors><titles><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabet

Endocrinology</secondary-title></title>><periodical><full-title>The Lancet Diabetes & Endocrinology</full-title></periodical><pages>35-

7</electronic-resource-num></record></EndNote>] examined the relationship between maternal thyroid function, child IQ, and brain morphology. Maternal thyroid function data, including TSH, fT4, and TPO Ab, were assessed at 9-18 weeks of pregnancy. Child IQ was assessed between 5.6 and 7.9 years of age using a Dutch non-verbal intelligence test with high reliability (the Snijders-Oomen Niet-Verbale Intelligentie Test). The IQ test assesses a broad range of language-independent intelligence functions such as spatial visualization abilities and abstract reasoning abilities. Child brain morphology was assessed using structural MRI brain scans. Pregnant mothers were recruited if

they lived in the municipality of Rotterdam, Netherlands, and were enrolled as participants in the Generation R Study. The main exclusion criteria for mother-child pairs were abortion, fetal loss, or gestational age of 24 weeks or more at the start of the study. Additional exclusion criteria for motherchild pairs undergoing the IQ assessment were twin pregnancies, pre-existing thyroid disease, concomitant medication that affects thyroid function, fertility treatment, or IQ scores  $\leq 50$ . After exclusion criteria were applied, 3,839 mother-child pairs constituted the IQ study population, and 646 were included in the brain morphology study population (598 mother-child pairs overlapped between the two analyses). TSH and fT4 were log-transformed and then evaluated for associations with child IQ or brain morphology MRI outcomes using ordinary least-squares linear regression models with restricted cubic splines with knots at the 10<sup>th</sup>, 50<sup>th</sup>, and 90<sup>th</sup> percentiles. Effect estimates were reported from multivariate linear regression models with a quadratic term that approximated the model curves developed in the spline analysis. In analyses of IQ, models were adjusted for potential confounders including gestational age, maternal age, smoking, BMI, parity, education level, ethnic origin, fetal sex, and birthweight. Analyses of brain morphology were adjusted for gestational age, maternal age, BMI, child age, sex, birthweight, and gestational age at birth. All analyses were adjusted for hCG, TPO Ab, and child thyroid function.

Selected study results are included in [ REF \_Ref474405694 \h \\* MERGEFORMAT ]. Overall, analysis results between maternal fT4 and offspring IQ showed an inverted U-shaped association (p=0.0044 in the whole population ([ REF \_Ref474405694 \h \\* MERGEFORMAT ] A); p=0.016 after exclusion of women with overt hypothyroidism or hyperthyroidism ([ REF \_Ref474405694 \h \\* MERGEFORMAT ] C)). All associations remained similar after adjusting for concentrations of hCG, child TSH and fT4, or maternal TPO Ab. Associations remained significant after the exclusion of cases of overt hypothyroidism and overt hyperthyroidism, which implies that changes in fT4 in the subclinical range can affect IQ (Korevaar et al., 2016). Korevaar et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Korevaar</Author><Year>2016</Year><RecNum>313</RecNum><Di splayText>(2016)</DisplayText><record><rec-number>313</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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B.</author><author>Steegers, Eric A. P.</author><author>Visser, Theo J.</author><author>White

B.</author><author>Steegers, Eric A. P.</author><author>Visser, Theo J.</author><author>White, Tonya</author><author>Tiemeier, Henning</author><author>Peeters, Robin

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Endocrinology</secondary-title></title>><periodical><full-title>The Lancet Diabetes & mp; Endocrinology</full-title></periodical><pages>35-

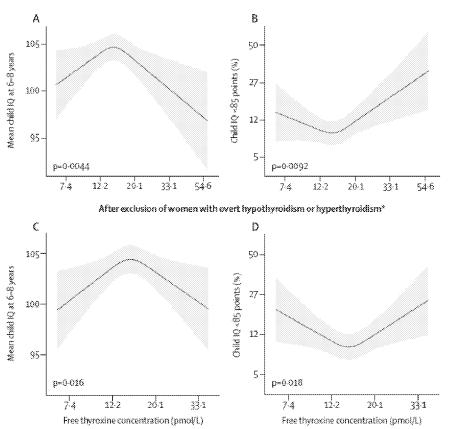
43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>2 2138587</isbn><urls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-

7</electronic-resource-num></record></Cite></EndNote>] found that the relationship demonstrated in the spline model ([ REF \_Ref474405694 \h ]) was well explained by the multivariate linear regression analysis with a quadratic term, as summarized in [ REF \_Ref479341381 \h ].

Further, high maternal fT4 (range of cut points:  $> 88^{th}$  to  $> 97^{th}$  percentile) corresponded to a 1.4 to 3.7 point reduction in mean offspring IQ when compared to the reference group (fT4 between  $10^{th}$  and  $90^{th}$  percentile). Low maternal fT4 (range of cut points:  $< 3^{rd}$  to  $< 11^{th}$ ) corresponded to a 1.5 to 3.8 point loss when compared to the reference group.

For the MRI portion of the study, associations with maternal fT4 showed a U-shaped association for mean child gray matter volume (p = 0.0062 in the whole population; p = 0.011 after exclusion of women with overt hypothyroidism or hyperthyroidism) and for mean cortex volume (p = 0.0011 in the whole population; p = 0.0016 after exclusion of women with overt hypothyroidism or hyperthyroidism).

# Figure [ SEQ Figure \\* ARABIC ]. Association between Maternal Free Thyroxine and Offspring IQ



(A, C) Association between maternal fT4 concentrations during pregnancy and child IQ as predicted mean and 95% CI in the whole population (A) and after the exclusion of women with overt thyroid disease (C). (B, D) Association of maternal fT4 levels during pregnancy and the risk of child IQ below 85 as predicted by back-transformed log odds with 95% CI in the whole population (B) and after the exclusion of women with overt thyroid disease (D). Scales for the x-axis might differ between the upper and lower graphs because of exclusions of women with overt, but not subclinical, disease entities. \*Overt hypothyroidism and hyperthyroidism are defined as the biochemical diagnosis made during pregnancy, based on the central 95% reference range, as advocated in international guidelines (Korevaar et al., 2016).

# Table [ SEQ Table \\* ARABIC ]. Summary of Results Evaluating the Relationship Between Maternal fT4 Levels with Mean Child IQ

Variables in Model	IQ: β (95% CI) <sup>b</sup>
fT4	33.81 (9.8, 57.82) a
fT4 <sup>2</sup>	-6.235 (-10.567, -1.903) <sup>a</sup>

Source: Korevaar et al. [ ADDIN EN.CITE < EndNote > < Cite

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id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Korevaar, Tim I. M.</author><author>Muetzel, Ryan</author><author>Medici, Marco</author><author>Chaker,

Layal</author><author>de Rijke, Yolanda B.</author><author>Steegers, Eric A. P.</author>tor>Visser, Theo J.</author>de Rijke, Yolanda B.</author>de Rijke, Yolanda B.</author>de Rijke, Yolanda B.</author>de Rijke, Yolanda B.</author>de Rijke, Yolanda B.</arthor>de Rijke, Yolanda B.</author>de Rijke, Yolanda B.</a>

Henning</author>>eauthor>Peeters, Robin P.</author></authors></contributors><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & prospective cohort study</title><secondary-title>The Lancet Diabetes & prospective

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43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>22138587ls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-7</electronic-resource-num></record></Cite></EndNote>].</ur>

*Key finding:* The authors concluded that both low and high maternal fT4 concentrations during pregnancy were associated with lower child IQ and lower gray matter and cortex volume.

Detailed Summary of Moleti et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ]

Effects of Maternal Iodine Nutrition and Thyroid Status on Cognitive Development in Offspring: A Pilot Study

Moleti et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] examined the effects of maternal iodine and/or levothyroxine (LT4) supplementation on the IQ of their offspring. This prospective, observational study included four study groups of 15 mother-child pairs from a small, rural area in northeastern Sicily. The groups were iodine (I); no iodine (no-I); iodine + LT4 (I + T4); and no iodine + LT4 (no-I + T4). Pregnant mothers were sampled for fT3, fT4, TSH, TPO Ab (considered positive if values > 34 IU/mL), and antithyroglobulin (TgAb) starting at GW < 12 weeks and continuing every 6 weeks throughout pregnancy. To be included in the iodine groups, women had to have regularly consumed iodized salt up to 2 years prior to pregnancy and continued consumption during pregnancy. For the no-iodine groups, women were excluded if they had regular iodized salt consumption prior to pregnancy or consumed iodized salt upon becoming pregnant. Other inclusion criteria for mothers included age greater than 18 years, an uncomplicated singleton pregnancy, (full) term delivery, no severe or chronic diseases (including thyroid autoimmune diseases), no major post-partum complications, and full disclosure of diet and lifestyle information during and after pregnancy.

<sup>&</sup>lt;sup>a</sup> Statistically significant at p<0.05.

<sup>&</sup>lt;sup>b</sup> Reported beta and 95% confidence interval are increase in IQ per log increase in fT4. 95% confidence intervals were calculated by multiplying standard error of 12.25 for fT4 and 2.210 for fT4<sup>2</sup> presented in Korevaar et al. (2016) by 1.96. fT4<sup>2</sup> refer to addition of a squared fT4 variable in the model. Analyses were performed using linear regression models in N=3,839 mother-child pairs. fT4 values were transformed by the natural logarithm.

Inclusion criteria for children included age 6 to 14 years, no major neonatal complications, no congenital hypothyroidism, no severe/chronic diseases or cognitive deficits, and regular school attendance. Children ages 6 to 12 years of age were administered IQ tests blind by trained psychologists with the WISC-III, full-scale IQ (FSIQ), verbal IQ, and performance IQ (PIQ). FSIQ scores of 70-85 were considered suggestive of borderline defective cognitive function, and scores less than 70 were considered suggestive of defective cognitive function.

Statistical analyses were performed using the Mann-Whitney (comparison of independent samples), Wilcoxon (comparison of dependent samples), and Friedman (comparison over the observation period of several dependent groups) tests. A chi-squared ( $\chi^2$ ) test was used to evaluate the association between categorical variables. Logistic regression models were used to evaluate the dependence of suboptimal cognitive outcomes (IQ scores less than 85) on various explanatory variables and confounders. Covariates assessed in the models included maternal age; gestational age at birth; birthweight; child age at time of IQ test; maternal fT4, and TSH at various times during gestation; maternal urinary iodide excretion; family socioeconomic status; maternal parity; sex of child; breastfeeding; major maternal stressful life events; and maternal/parental education.

Overall, mothers in the iodine groups had significantly higher fT4 concentrations than those in the noiodine groups. In this study, fT4 was evaluated across the entire range of pregnancy and fT4 concentrations, and not solely the hypothyroxinemic range in early pregnancy. The overall prevalence of borderline or defective cognitive function was more than three-fold higher in offspring of mothers not using iodized salt compared to mothers using it (76.9% vs. 23.1%, OR = 7.667, 95% CI: 2.365-24.856;  $\chi^2 = 12.65$ ; p = 0.0001). Children born to mothers in the iodine group had significantly higher FSIQ and verbal IQ (VIQ) scores compared to the no-iodine offspring (FSIQ: (I)  $93.1 \pm 11.8$  vs. (no-I)  $81.7 \pm 13.5$ , p = 0.028; VIQ (I)  $90.1 \pm 14.2$  vs. (no-I)  $80.3 \pm 12.3$ , p = 0.034). Similar differences were found between I + T4 and no-I + T4 offspring. The relationship between fT4 at various periods of gestation was consistently associated with verbal, performance, and FSIQ, but failed to reach statistical significance ([ REF Ref475454870 \h \\* MERGEFORMAT ]). Given the small sample size in the study, the authors caution against firm conclusions being drawn from their analysis. Additionally, the design of the study did not include parental IQ testing (a possible confounder for child IQ). The hypothesis set forward by the authors for the results of their analysis, which show iodine to be a more important predictor of neurodevelopmental outcomes compared to thyroid hormone levels, is that the fetus is exposed to insufficient amounts of thyroid hormone as a result of not having enough iodine from the mother to develop its own thyroid hormones.

# Table [ SEQ Table \\* ARABIC ]. Relationship Between Cognitive Outcomes and fT4 (Univariate Logistic Regression Models)

	VIQ: exp β (95% CI)	PIQ: exp β (95% CI)	FSIQ: exp β (95% CI)
	Costo (Ingala Preciso o	To reconstruction	
≤12 weeks	1.192 (0.980-1.450)	1.053 (0.881-1.259)	1.103 (0.928-1.311)
13 - 18 weeks	1.217 (1.015-1.459) <sup>a</sup>	1.010 (0.855-1.194)	1.165 (0.980-1.385)
19 - 24 weeks	1.234 (0.991-1.537)	1.125 (0.897-1.411)	1.155 (0.937-1.425)
25 - 30 weeks	1.137 (0.928-1.393)	1.003 (0.810-1.243)	1.031 (0.847-1.254)
Week 31 - term	1.191 (0.982-1.445)	1.010 (0.834-1.224)	1.070 (0.895-1.280)
	film (exam) a film may less	ata Pelaterangangan	
	1.017 (1.001-1.033) <sup>b</sup>	1.012 (0.995-1.028)	1.013 (0.998-1.029)

Source: Moleti et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ].

**Key finding:** The authors found that maternal iodine status is predictive of neurodevelopmental outcomes in offspring. The relationship between fT4 at various periods of gestation was consistently associated with verbal, performance, and FSIQ, but failed to reach statistical significance.

## **Cognitive (Bayley Scales)**

Detailed Summary of Pop et al. [ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayText>(1999)</DisplayText><record><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z,9t0avxvz,dfermedxe5vxfpkax2vzp0ftv29"

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L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H

L</author></authors></contributors></title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy</title><secondary-title>Clinical Endocrinology</secondary-

title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><urls></urls></record></Cite></EndNote>]

Low Maternal Free Thyroxine Concentrations During Early Pregnancy Are Associated with Impaired Psychomotor Development in Infancy

## Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayText>(1999)</DisplayText><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062208">40</key></foreign-

<sup>&</sup>lt;sup>a</sup> 95% CI interval suggests statistical significance, but significance for this result and most results in this table were not denoted in the study.

<sup>&</sup>lt;sup>b</sup> Statistically significant at *p* < 0.05.

keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Pop, V J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H L</author></authors></contributors><title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy</title><secondary-title>Clinical Endocrinology</secondary-title></title>>cperiodical><fulltitle>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><url s></urls></record></Cite></EndNote>] conducted a prospective cohort study to examine the association between maternal thyroid hormones during early gestation and children's neurodevelopment in an iodine-sufficient region. Women were recruited from a medical center and from community midwife practices at 12 GW. In the 291 healthy women who consented to participate, assessments of maternal thyroid function—fT4, TSH, and TPO Ab levels—were completed at 12 GW and 32 GW. At various stages of follow-up, women were excluded from the study due to neonatal deaths, moving outside of the area, and other pregnancy complications. Children of women remaining in the study at 10 months post-partum (n = 220) completed the Dutch version of the Bayley Scales of Infant Development (BSID) to evaluate mental and psychomotor development using the MDI and PDI, respectively. Pop et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayT ext>(1999)</DisplayText><record><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062208">40</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H L</author></authors></contributors><title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy</title><secondary-title>Clinical Endocrinology</secondary-title></title>><periodical><fulltitle>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><url s></urls></record></Cite></EndNote>] conducted t-tests to investigate maternal thyroid levels in the lowest 5th (< 9.8 pmol/L) and 10th (< 10.4 pmol/L) percentiles of fT4 levels at 12 and 32 GW. The authors also performed linear and logistic regression analyses, adjusted for confounders including maternal depression, psychosocial factors, tobacco and alcohol use during pregnancy, and demographic variables. Data analyses were conducted using t-tests, linear regression, and logistic regression.

Children of the 22 women in the lowest 10<sup>th</sup> percentile of fT4 at 12 GW had significantly lower scores on the PDI than the rest of the children (mean difference: 7.4, 95% CI: 1.1, 13.9). No significant differences were found when investigating PDI scores at 32 GW. MDI scores showed differences associated with fT4 levels, but these failed to reach significance at both 12 and 32 GW. Pop et al. [ ADDIN EN.CITE <EndNote><Cite

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ExcludeAuth="1">ExcludeAuth="1">Author>Pop</Author><Pop</P>
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J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H L</author></authors></contributors></title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy</title><secondary-title>Clinical Endocrinology</secondary-title></title><periodical><fulltitle>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><vear>1999</year></dates><url s></urls></record></Cite></EndNote>] also found a strong correlation between PDI scores and maternal fT4 levels (r = 0.46, p = 0.03) (see [ REF Ref457471597 \h ]). Although the authors did not present the equation representing the relationship in [ REF Ref457471597 \h l. 12 their Figure 2 was digitized using the WebPlotDigitizer Extension for Google Chrome and using the linear regression function in Excel on the extracted data. By doing so it was determined that for every 1 pmol/L change in fT4 there is an estimated 8.5 point change in PDI (95% CI: 0.01 - 17.04). 13 Children of mothers with fT4 concentrations in the lowest fT4  $5^{th}$  percentile (n = 11) and  $10^{th}$  percentile also had lower scores on the MDI subscale, although these differences were not significant. The results suggest that the timing of hypothyroxinemia during pregnancy may be important as it relates to performance on the BSID.

## Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

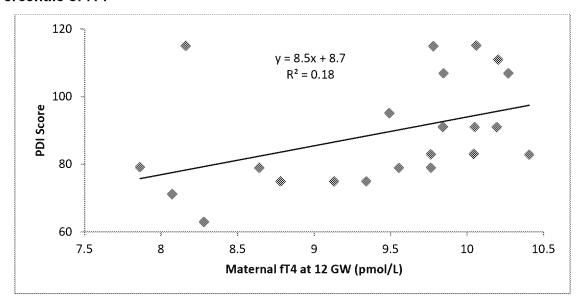
ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayText>(1999)</DisplayText><record><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmedxc5vxfpkax2vzp0ftv29" timestamp="1432062208">40</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H L</author></author></author></author><br/>Contributors><titles><title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in

The EPA attempted to contact study authors on July 20, 2015. To date, no response has been received.

R<sup>2</sup> value based on Excel analysis = 0.18; R<sup>2</sup> value reported by Pop et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1">Ex

infancy</title><secondary-title>Clinical Endocrinology</secondary-title></title>>cperiodical><fulltitle>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><url s></urls></record></Cite></EndNote>] additionally performed logistic regression analyses to investigate associations between maternal variables and low PDI scores (defined as below 84 points). In univariate regression analyses, fT4 in the lowest 10th percentile at 12 GW was significantly associated with a low PDI score (relative risk (RR) = 3.6; 95% CI: 1.1-12.1). To control for the effect of possible confounders, Pop et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Pop</Author><Year>1999</Year><RecNum>40</RecNum><DisplayT ext>(1999)</DisplayText><record><rec-number>40</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432062208">40</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Kuijpens, J L</author><author>van Baar, A L</author><author>Verkerk, G</author><author>van Son, M M</author><author>de Vijlder, J J</author><author>Vulsma, T</author><author>Wiersinga, W M</author><author>Drexhage, H A</author><author>Vader, H L</author></authors></contributors></title>Low maternal free thyroxine concentrations during early pregnancy are associated with impaired psychomotor development in infancy</title><secondary-title>Clinical Endocrinology</secondary-title></title>><periodical><fulltitle>Clinical Endocrinology</full-title></periodical><pages>149-155</pages><volume>50</volume><section>149</section><dates><year>1999</year></dates><url s></urls></record></Cite></EndNote>] performed a multiple logistic regression analysis that included a host of additional pregnancy-related factors, maternal mood indicators, and demographic features. The relative risk estimate for the association between fT4 in the lowest 10<sup>th</sup> percentile at 12 GW and low PDI scores increased (RR = 5.8; 95% CI: 1.3-12.6) in the multivariate regression analysis. No significant associations were found in univariate or multivariate regression analyses between fT4 in the lowest 10<sup>th</sup> percentile at 32 GW and low PDI scores.

Figure [ SEQ Figure \\* ARABIC ]. Digitization of Figure 2 from Pop et al. (1999): Correlation between PDI Score and fT4 at 12 GW for Women in the Lowest 10<sup>th</sup> Percentile of fT4



*Key finding*: Maternal fT4 levels in the lowest 10<sup>th</sup> percentile in the first trimester among an iodine-sufficient population were significantly associated with reduced psychomotor development in 10-month-old infants (a drop in the PDI score of 8.5 points was associated with every 1 pmol/L reduction in fT4).

Detailed Summary of Pop et al. [ADDIN EN.CITE <EndNote><Cite
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P</author><author>Valer, H L</author><author>Vulsma, T</author><author>van Baar, A
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hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up
study</title><secondary-title>Clinical Endocrinology</secondary-title></title></periodical><pages><282288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url>
rls></urls></record></Cite></EndNote>]

Maternal Hypothyroxinemia during Early Pregnancy and Subsequent Child Development: A 3-Year Follow-Up Study

In a prospective 3-year study, Pop et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayText><(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J J</author></author></authors></contributors><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title><periodical><full-title>Clinical Endocrinology

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] measured thyroid hormone levels (TSH and fT4) in randomly selected pregnant women at 12, 24, and 32 GW in the Netherlands. Pop et al. initially approached 1,361 women to join their cohort. Eight of these women were excluded due to the presence of hyper- or hypothyroidism. From the remaining 1,353 women, Pop et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayT ext>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors></title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title><periodical><full-title>Clinical

Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>csection>282</section>cdates><year>2003</year>c/dates><url s></urls></record></EndNote>] recruited 135 hypothyroxinemic cases (fT4 levels in the lowest 10th percentile (< 12.4 pmol/L)) and matched them based on parity and gravidity with an equal number of controls (fT4 levels in the 50th-90th percentiles (15.6–19.1 pmol/L)). TSH, fT4, and TPO Ab were assessed at 12 GW. At various stages of follow-up, women were excluded from the cohort due to a variety of exclusion criteria such as subclinical thyroid dysfunction (high TSH, normal fT4), fertility problems, autoimmune diseases, obstetrical complications, major depressive episodes, and prolonged or repeated hospitalizations after birth. This resulted in 63 cases and 62 controls at the age of 1 year and 57 cases and 58 controls at the age of 2 years. (For complete details on inclusion and exclusion of women in the cohort, see Figure 1 of Pop et al., 2003).

Infant development was assessed at 1 and 2 years of age using the Dutch version of the BSID to evaluate MDI and PDI scores. The mean scores for cases and controls are presented in [ REF \_Ref457848210 \h ]. This table demonstrates that statistically significant differences in MDI and PDI existed between the cases and controls at both 1 and 2 years of age.

Table [ SEQ Table \\* ARABIC ]. Differences on Bayley Mental (MDI) and Motor (PDI) Subscales Assessed at 1 and 2 Years<sup>a</sup>

	Cases <sup>b</sup>	Controls	p-value	95% CI
		TC COL		
Mental, mean (SD)	95 (15)	105 (14)	0.004	4-16
Motor, mean (SD)	91 (15)	99 (14)	0.02	3-12
	Ti.	ro feets		
Mental, mean (SD)	98 (15)	106 (14)	0.02	4-12
Motor, mean (SD)	92 (16)	102 (16)	0.005	6-16

Source: Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><Suffix>`, Table

2</Suffix><DisplayText>(2003, Table 2)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>vauthor>de Vijlder, J

J</author></authors></contributors></title>>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title></periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><urls></record></Cite></End Note>].

- a Based on a two-tailed t-test.
- <sup>b</sup> fT4 levels in the lowest 10<sup>th</sup> percentile (< 12.4 pmol/L).
- ° fT4 levels in the 50th-90th percentiles (15.6-19.1 pmol/L).

# Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayText>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma,

T</author><author>van Baar, A L</author><author>de Vijlder, J J</author></authors></contributors></title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondarytitle>Clinical Endocrinology</secondary-title></title>>epriodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] also evaluated the relationship between maternal fT4 levels at 12 GW with MDI and PDI at 2 years of age. They determined that there was a significant correlation between maternal fT4 levels at 12 GW and both MDI and PDI scores (MDI: r = 0.48, p =0.001; PDI: r = 0.38, p = 0.006) (see [ REF Ref457849041 \h ]) among the women with hypothyroxinemia. Although Pop et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayT ext>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J J</author></authors></contributors></title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondarytitle>Clinical Endocrinology</secondary-title></title><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-288</pages><volume>59</volume><section>282</section><dates><vear>2003</vear></dates><url s></urls></record></Cite></EndNote>] did not present the equation representing the relationship in [ REF Ref457849041 \h ], their Figure 3 was digitized using the WebPlotDigitizer Extension for Google Chrome and using the linear regression function in Excel on the extracted data. <sup>14</sup> By doing so,

it was determined that for every 1 pmol/L change in fT4 there is an estimated 8.4 point change in PDI

The EPA attempted to contact study authors on July 20, 2015. To date, no response has been received.

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(95% CI: 4.0–12.8)<sup>15</sup> and 6.3 point change in MDI (95% CI: 1.92–10.62).<sup>16</sup> Analogous analyses were not presented for individuals assessed at 1 year of age or for controls.

In addition, Pop et al. [ ADDIN EN.CITE <EndNote><Cite

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J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title>><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] examined the pattern of maternal fT4 levels throughout pregnancy and the distribution of Bayley Scales scores in the cases and controls. The authors found that the lowest Bayley scores were associated with the offspring of first-trimester hypothyroxinemic women (cases) whose fT4 levels continued to decrease as pregnancy progressed. Other women who

 $R^2$  value based on Excel analysis = 0.22;  $R^2$  value reported by Pop et al. [ ADDIN EN.CITE EndNote

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J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

<sup>288 &</sup>lt; pages > volume > 59 < volume > section > 282 < section > < dates > vgear > 2003 < ygear > < / dates > vurls > < / record > < / Cite > < / EndNote > ] = 0.23.

R² value based on Excel analysis and reported by Pop et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayText>( 2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

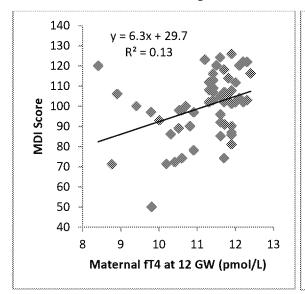
J</author></authors></contributors><title>>dernal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title></periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

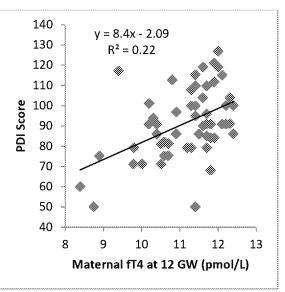
started out as hypothyroxinemic, but experienced increases in fT4 levels in either the second or third trimesters, had children with similar MDI and PDI scores to women who were not hypothyroxinemic in the first trimester. Additionally, the authors noted that children of women who had fT4 in the 50<sup>th</sup> to 90<sup>th</sup> percentile range during the first trimester (controls), but later experienced a drop in fT4, did not exhibit an association between neurodevelopment and fT4.

# Figure [ SEQ Figure \\* ARABIC ]. Digitization of Figure 3 from Pop et al. [ ADDIN EN.CITE <EndNote><Cite

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<DisplayText>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"<br/>
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Vijlder, J J</author></author></author></title>Maternal<br/>
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title></periodical><pages>282-

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><urls></urls></record></EndNote>]: Linear Regression of Maternal fT4 Concentrations of Cases with fT4 in the Lowest 10<sup>th</sup> Percentile at 12 GW with MDI and PDI Scores at 2 Years of Age





Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title>>cperiodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>csection>282</section>cdates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] note several potential shortcomings of their analysis. One is that the subgroups they assessed when evaluating the impact of the pattern of fT4 on PDI and MDI became very small because they excluded cases with potentially confounding data (e.g., mothers with episodes of depression). This may be why a statistical analysis of the individuals just in the group with low fT4 throughout pregnancy was not included in Pop et al. [ ADDIN EN.CITE <EndNote><Cite

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288</pages><volume>59</volume>582</section><dates><ycar>2003
s</urls></record></Cite></EndNote>]. Further, Pop et al. [ ADDIN EN.CITE <EndNote><Cite
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J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>59</volume>3282</section><dates><year>2003</year></dates><url s></urls></record></EndNote>] noted that an assessment of a child's abilities at a very young age (i.e., less than 1 year of age) may have limited value once the child reaches school. However, Bayley scores at 2 years of age have been found capable of predicting a child's ability at 5 years, particularly if the child is at risk of developmental delays (Van Baar & De Graff, 1994, as cited in Pop et al., 2003; also see Appendix H). Additionally, no data on daily food or iodine intake in the population are available. The authors point out that previous studies in the same general population have reported sufficient iodine intake. However, according to the study authors, the sufficiency of this iodine intake specifically for pregnant women has been called into question (see van Rees-Wortelboer et al., 1987). Lastly, Pop et al. ([ ADDIN EN.CITE <EndNote><Cite

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db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></title><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] note that the trend in norms for the developmental tests are representative of the general population, since the mean MDI and PDI of controls in this study are similar to those reported in their prior 1999 study.

## The Pop et al. [ ADDIN EN.CITE < EndNote > < Cite

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J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>59</volume>59</volume>59</volume>59</volume>59</volume>59</volume>59</volume>59</volume>60</volume>60</vol>
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**Key finding:** Maternal fT4 levels in the lowest 10<sup>th</sup> percentile in the first trimester were significantly associated with a clinically relevant difference in psychomotor and mental development among

2-year-olds (a drop in the PDI score of 8.4 points and in the MDI score of 6.3 points was associated with every 1 pmol/L reduction in fT4).

## Cognitive (Other)

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Maternal Hypothyroxinemia in Early Pregnancy Predicts Reduced Performance in Reaction Time Tests in 5- to 6-Year-Old Offspring

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ethnicity), and pregnancy characteristic (e.g., BMI and diabetes). Finken et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum>6Displa yText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></authors></contributors><titles><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6-year-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1417-

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In all regression models tested, maternal fT4 levels were inversely associated with reaction time SD within an individual in a reaction time task. Maternal hypothyroxinemia was significantly associated with higher baseline reaction time when compared to non-hypothyroxinemic mothers (fT4  $\geq$  10<sup>th</sup> percentile at the first pre-natal visit). Specifically, there was a 39.5 ms increase in mean reaction time and a 41.2 ms increase in SD of reaction time, when comparing the offspring of hypothyroxinemic to non-hypothyroxinemic mothers. Finken et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum><Displa vText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></authors></contributors><titles><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6vear-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title>>eriodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1417-1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year>

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between hypothyroxinemia and hyperthyrotropinemia; children of mothers with both conditions had a significantly greater reaction time mean and SD. Associations between maternal fT4 or TSH and other ANT outcomes (e.g., visuomotor coordination) were not observed. Results from the study are summarized in [ REF \_Ref457841191 \h ]. Results remained similar in the additional analyses after excluding subjects.

Table [ SEQ Table \\* ARABIC ]. Summary of Results Evaluating the Relationship between Thyroid Hormones and Developmental Outcomes in Finken et al. [ ADDIN EN.CITE <EndNote><Cite

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1426</pages><volume>98</volume><number>4</number><section>1417</section><d ates><year>2013</year></dates><urls><lelectronic-resource-

num>10.1210/jc.2012-3389</electronic-resource-num></record></Cite></EndNote>]

Test	Predictor	β (95% CI)	Р
Bas	eline Specia		
Mean reaction time, milliseconds	fT4 (pmol/L)	-4.5 (-9.0; 0.1)	0.06
	TSH (mU/L)	2.3 (-1.3; 5.9)	0.21
Reaction time, SD, milliseconds	fT4 (pmol/L)	-4.9 (-9.5; -0.2)	0.04
	TSH (mU/L)	1.0 (2.7; 4.6)	0.61
Pursuit			
Deviation, millimeters	fT4 (pmol/L)	-0.01 (-0.22; 0.21)	0.95
	TSH (mU/L)	0 (-0.17; 0.17)	0.99
SD of deviation, millimeters	fT4 (pmol/L)	0 (-0.20; 0.21)	0.97
	TSH (mU/L)	0.04 (-0.12; 0.20)	0.65
Mean distance, millimeters	fT4 (pmol/L)	-0.01 (-0.11; 0.09)	0.78
	TSH (mU/L)	0.04 (-0.05; 0.11)	0.40
Accuracy stability, millimeters	fT4 (pmol/L)	0.04 (-0.06; 0.15)	0.42
	TSH (mU/L)	0.01 (-0.08; 0.09)	0.84
Movement speed, seconds	fT4 (pmol/L)	-0.08 (-0.34; 0.18)	0.55
	TSH (mU/L)	0.10 (-0.11; 0.31)	0.34
Response O			
Part 1 (compatible)			
Mean reaction time, milliseconds	fT4 (pmol/L)	-5.1 (-12.1; 1.8)	0.15
	TSH (mU/L)	1.4 (-4.0; 7.0)	0.61
Reaction time, SD, milliseconds	fT4 (pmol/L)	-1.7 (-7.4; 3.9)	0.55
	TSH (mU/L)	4.6 (0.1; 9.1)	0.05
Part 2 (incompatible)			
Mean reaction time, milliseconds	fT4 (pmol/L)	2.0 (-7.6; 11.6)	0.69
	TSH (mU/L)	4.6 (-3.1; 12.3)	0.24
Reaction time, SD, milliseconds	fT4 (pmol/L)	0.9 (-6.7; 8.6)	0.82
	TSH (mU/L)	7.3 (1.2; 13.3)	0.02

Source: Finken et al. [ ADDIN EN.CITE < EndNote > < Cite

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1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year>2013</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2012-3389</electronic-resource-num></record></Cite></EndNote>].

Note: Results adjusted for gender and age at assessment, family background variables, including computer gaming (< 1 or  $\geq$  1 hr/day); birth order; maternal education (in 4 categories); ethnicity (Western or non-Western); and pregnancy characteristics, including maternal age, BMI, diabetes (yes or no), and hypertension (yes or no).

*Key finding:* Maternal fT4 levels at a median of 12.9 weeks were inversely associated with the reaction time variability among 5- to 6-year-old children (an increase in the reaction time SD of 4.9 ms was associated with every 1 pmol/L reduction in maternal fT4). Considering fT4 as a

categorical variable demonstrated that there was a 39.5 ms increase in mean reaction time and a 41.2 ms increase in SD of reaction time when comparing the offspring of hypothyroxinemic vs. non-hypothyroxinemic mothers.

Detailed Summary of Henrichs et al. [ ADDIN EN.CITE < EndNote > < Cite

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Study</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title></periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>4227-

4234</pages><volume>95</volume><number>9</number><section>4227</section><dates><year >2010</year></dates><urls><lelectronic-resource-num>10.1210/jc.2010-0415</electronic-resource-num></record></Cite></EndNote>]

Maternal Thyroid Function during Early Pregnancy and Cognitive Functioning in Early Childhood: The Generation R Study

Henrichs et al. [ ADDIN EN.CITE < EndNote > < Cite

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W</author><author>Visser, W</author><author>Steegers, E A P</author><author>Verhulst, F C</author><author>de Rijke, Y B</author><author>Tiemeier,

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4234</pages><volume>95</volume>91/number>9</number><section>4227</section><dates><year>2010</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2010-0415</electronic-resource-num></record></Cite></EndNote>] evaluated the relationship between early pregnancy TSH and fT4 levels in mothers and cognitive functioning of their children in a Netherlands-based study. Participants were recruited from a birth cohort and excluded if T4 treatment was received during pregnancy. Maternal blood was analyzed for fT4 and TSH at a mean GW of 13.3. TPO Abs

were not measured. Mothers were classified as having mild or severe hypothyroxinemia based on T4 levels  $< 10^{th}$  (< 11.76 pmol/L) and  $< 5^{th}$  percentiles (< 10.96 pmol/L), respectively. Parents completed several questionnaires to assess verbal development at age 18 months and verbal and nonverbal development at 30 months.

Verbal assessments at 18 and 30 months (the MacArthur Communicative Development Inventory and the Language Development Inventory, respectively) consisted of checklists of words; parents identified words spontaneously spoken by children, which were then summed to derive age- and gender-specific percentiles. Expressive language delay at age 18 or 30 months was operationalized as scores below the 15<sup>th</sup> percentile. Delay at 30 months was also indicated by a lack of word combinations (i.e., the child had not begun to construct phrases).

Nonverbal cognitive development was assessed using the Parent Report of Children's Abilities, which consists of questions to parents (e.g., on spatial ability, memory, and planning skills) and parent-administered tasks (e.g., building blocks). Cognitive delay was similarly defined as scores below the 15<sup>th</sup> age- and gender-specific percentiles. Logistic regression analyses were conducted with adjustment for maternal age, education, pre-natal distress, pre-natal smoking, birthweight, gestational age at blood sampling, and child ethnicity.

Results indicated that, while maternal TSH was not found to be related to the cognitive outcomes measured, maternal fT4 was found to be inversely related to expressive language delay at 30 months of age. Both mild and severe maternal hypothyroxinemia were found to be associated with a higher risk of expressive language delay across the age range of 18 months (OR = 1.44; 95% CI: 1.09, 1.91; p = 0.010) and 30 months of age (OR = 1.80; 95% CI: 1.24, 2.61; p = 0.002), respectively. Henrichs et al. [ ADDIN EN.CITE <EndNote><Cite

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Study authors were contacted on November 18, 2016, for this information and have not yet responded.

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Table [ SEQ Table \\* ARABIC ]. Summary of Results from Henrichs et al. [ ADDIN EN.CITE <EndNote><Cite

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Maternal Expressive Languag Thyroid Delay at 18 Months			Expressive Language Delay at 30 Months		Expressive Language Delay at 18 and 30 Months		Nonverbal Cognitive Delay at 30 Months	
Function Measure	n	OR (95% CI) p-value	n	OR (95% CI) p-value	n	OR (95% CI) p-value	n	OR (95% CI) p-value
TSH, per SD	3384	0.91 (0.81; 1.03) 0.136	2757	0.92 (0.81; 1.06) 0.249	3614	0.92 (0.84; 1.02) 0.100	2588	0.98 (0.88; 1.10) 0.759
fT4, per SD	3409	0.95 (0.83; 1.09) 0.430	2779	0.84 (0.71; 0.99) 0.039	3643	0.90 (0.80; 1.01) 0.069	2606	0.85 (0.72; 1.01) 0.057
Mild hypothyroxi- nemia	2736	1.33 (0.91; 1.94) 0.143	2225	1.47 (1.00; 2.17) 0.051	2926	1.44 (1.09; 1.91) <b>0.010</b>	2086	1.37 (0.90; 2.07) 0.139
Severe hypothyroxi- nemia	2736	1.77 (1.10; 2.84) <b>0.018</b>	2225	1.78 (1.07; 2.94) 0.024	2926	1.80 (1.24; 2.61) <b>0.002</b>	2086	2.03 (1.22; 3.39) <b>0.007</b>

**Key Finding:** Severe maternal hypothyroxinemia was associated with expressive language delay at

18 and 30 months and nonverbal cognitive delay at 30 months. Further, there was an increase in the

odds of expressive language delay and nonverbal cognitive delay when comparing the offspring of children exposed to mild versus severe maternal hypothyroxinemia, suggesting a potential doseresponse between the severity of hypothyroxinemia and expressive language delay and nonverbal cognitive delay. Additionally, maternal fT4 levels per SD were associated with a reduction in odds of expressive language delay at 30 months (OR = 0.84 per SD of fT4).

# Detailed Summary of Kasatkina et al. (2006)

Gestational Hypothyroxinemia and Cognitive Function in Offspring

Kasatkina et al. (2006) investigated the effect of maternal hypothyroxinemia on the cognitive function of offspring with maternal iodine deficiency. Maternal thyroid hormones (TSH, fT4) and TPO Ab were measured during the first trimester (5-9 GW) in 24 pregnant Russian women, and 13 of them were hypothyroxinemic. Hypothyroxinemia was defined as fT4 levels less than the 10th percentile. Children underwent neurological assessments at 6, 9, or 12 months of age using a standardized clinical-psychological method for evaluating neurological development called the Gnome method. From this, a calculation of the coefficient of mental development (CMD) was obtained for children up to 3 years of age. Normal neurological development was defined as CMD of 90-100 points; risk of delayed neuropsychological development was defined as CMD of 80-89 points; and delayed neurological development was identified as CMD of less than 80 points. Covariates, confounders, and any model adjustments were not displayed in the study. Fisher's, Student, and Mann-Whitney tests were used to compare relative values. Relationships between parameters were assessed by a Spearman correlation analysis.

Kasatkina et al. (2006) performed a Spearman correlation analysis of the relationship between maternal fT4 at 5-9 GW and CMD. The authors found a strong, direct association between the level of maternal fT4 in weeks 5–9 of pregnancy and CMD in children at age 6 months (r = 0.684, p = 0.020), 9 months (r = 0.629, p = 0.038), and 12 months (r = 0.708, p = 0.014). The authors also did a comparative analysis of CMD elements at 6, 9, and 12 months comparing children born to mothers with hypothyroxinemia (which was corrected with iodine treatment) to mothers with normal fT4 levels. The children born to mothers with corrected hypothyroxinemia had consistently lower scores for CMD, motor function, and cognitive function at 6, 9, and 12 months (with one exception that at 12 months cognitive function was slightly higher in the children born to mothers with corrected hypothyroxinemia). However, none of the differences were statistically significant results.

**Key Finding:** The authors found that maternal fT4 levels in early pregnancy correlated significantly with CMD in children at ages 6, 9, and 12 months. When comparing CMD scores with the offspring of mothers with corrected hypothyroxinemia to the offspring of mothers with normal fT4, consistently lower scores for CMD, motor function, and cognitive function at 6, 9, and 12 months were seen, though the difference did not reach statistical significance.

# Detailed Summary of Noten et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ]

Maternal Hypothyroxinaemia in Early Pregnancy and School Performance in 5-Year-Old Offspring

Noten et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] examined the effect of hypothyroxinemia during pregnancy on school performance of offspring at 5 years of age. In line with other studies, hypothyroxinemia was defined as an fT4 level less than the 10<sup>th</sup> percentile of distribution (< 8.06 pmol/L in their sample). Participants were recruited from the Amsterdam Born Children and Their Development study. After exclusion criteria that included mothers who were lost

to follow-up or who did not give consent for their child's school test results, 1,196 mother-child pairs were included in the study population. Maternal fT4, TSH, and TPO Ab (> 80 kU/L was considered positive) were measured around the beginning of the second trimester (mean gestational age of 12.9 weeks, interquartile range 11.9-14.3 weeks).

School records of offspring were based on scores obtained in arithmetic and language tests from the Central Institute for Test Development, a national monitoring and evaluation system in the Netherlands. Multivariable logistic regression was used to assess the effects of fT4 and TSH on the range of poor or subnormal school performance (defined as a rank score below C or B, respectively). Models were adjusted for maternal education, ethnicity, BMI, and depressive symptomatology, as these were related to both the exposure to maternal hypothyroxinemia and outcome of school performance (p < 0.1). Maternal age, hypertension, diabetes, smoking, timing of blood sampling, TPO Ab positivity, child gender, gestational age, and birthweight were evaluated as potential confounders, but were found to not be associated with the predictor and outcome and thus not included in the regression analyses. Additional analyses were repeated after excluding pre-term births (< 32 weeks of gestation) or very low birthweight offspring (< 1,500 grams).

The distribution of school test results in terms of rank scores differed between hypothyroxinemic and non-hypothyroxinemic groups, with hypothyroxinemia being associated with lower rank scores in both arithmetic and language tests. Maternal hypothyroxinemia was significantly associated with 1.61-fold (95% CI: 1.05-2.47; p = 0.03) increased odds of subnormal arithmetic school performance after adjusting for confounders. However, this association weakened, reaching non-significance, after inverse-probability weighting was applied to account for potential non-response bias. No associations were found with TSH, even before inverse probability weighting. The continuous measure maternal fT4 was inversely associated with the odds of subnormal arithmetic performance, but lost statistical significance after adjusting for confounders. Additional exclusion of pre-term births or very low birthweight offspring did not change these associations.

**Key finding:** Maternal hypothyroxinemia at the end of the first trimester was associated with reduced arithmetic test performance in 5-year-old offspring. Inverse probability weighting demonstrates that non-response bias may be present in this study, and thus the results should be interpreted with caution.

Detailed Summary of Oken et al. [ ADDIN EN.CITE < EndNote > < Cite

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Neonatal Thyroxine, Maternal Thyroid Function, and Child Cognition

## Oken et al. [ ADDIN EN.CITE < EndNote > < Cite

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N</author></contributors><titles><title>Neonatal thyroxine, maternal thyroid function, and child cognition</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>497-

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(record></cite></EndNote>]. The WRAVMA assesses visual-spatial analysis, visual-motor ability, and fine motor skills. Linear regression models, adjusted for other variables such as gender and age at delivery, were used to investigate the relationship between thyroid hormone levels and neurodevelopmental outcomes, with adjustment for covariates including gestation length, age at testing, sex, maternal education, and ethnicity.

Neonatal total T4 was not associated with either the PPVT or WRAVMA scores. Contrary to expectations, it was associated with lower scores on the VRM test, both as a continuous variable ( $\beta$  = -0.5; 95% CI -0.9, -0.2) and as a categorical variable. Maternal thyroid function (as assessed by total T4, TSH, TPO Ab, and history of thyroid disease) was not associated with children's scores on any of the neurodevelopmental outcomes studied. Further, maternal and neonatal total T4 were evaluated, as opposed to fT4, which is the form of the hormone that is available to cross the placenta and may be a better indicator for an offspring's neurodevelopmental outcome. There was no significant association between maternal total T4 levels and neonatal total T4 levels.

*Key Finding:* Neonatal total T4 levels within the euthyroid range were not predictive of cognitive outcome, nor were they associated with maternal T4 levels. Maternal thyroid function (T4, TSH, and TPO Ab measured at approximately 10 GW) was not associated with any of the neurodevelopmental outcomes assessed. This study did not make an attempt to separately analyze hypothyroxinemic maternal-child pairs.

#### Behavior

# Detailed Summary of Endendijk et al. (2017)

Maternal Thyroid Hormone Trajectories during Pregnancy and Child Behavioral Problems

Endendijk et al. (2017) examined associations between maternal thyroid hormone trajectories and child behavior problems based on thyroid assessments at separate trimesters of pregnancy. The authors assessed 442 pregnant Dutch mothers at 12, 24, and 36 GW for thyroid hormone levels (TSH and fT4) and TPO Abs in blood serum. Mothers and fathers reported on their child's behavior problems between 23 and 60 months of age by individually completing the Child Behavior Check List (CBCL) (for more information on the CBCL, please refer to Appendix H). The version of the CBCL the authors used was developed for children between 1.5 and 5 years old and contains 100 behavioral and emotional problem items. For each item, parents indicated if a problem was seen frequently or not. The results were differentiated into internalizing (anxiety, depression, withdrawn behavior) or externalizing (attention difficulties, aggression) problems. Potential covariates considered for this analysis included gestational age and weight at birth, child age at time of CBCL assessment, parity, breastfeeding, mothers' age, maternal psychopathology symptoms (pre-natal and at time of CBCL assessment), smoking and alcohol use during pregnancy, educational level, and TPO

Ab positivity. Analysis of variance (ANOVA) or chi-squared tests were used to compare the different thyroid trajectory groups with regard to the covariates, and internalizing and externalizing problems.

Endendijk et al. (2017) reported a significant effect of first-trimester fT4 on anxiety/depression ( $\beta$  = -0.12, p < 0.05). The authors also found that maternal thyroid hormones over the course of pregnancy predicted child anxiety/depression better than first-trimester fT4 levels alone. Further, the authors looked for differences in sex with correlations between maternal thyroid hormones and internalizing and externalizing problems. Lower fT4 in the first trimester was associated with more internalizing problems with girls only, primarily driven by anxiety/depression (r = 0.17, p < 0.05). However, between girls and boys, the association between first-trimester fT4 and anxiety/depression was not statistically different.

**Key finding:** The authors reported a significant effect of first-trimester fT4 on anxiety/depression. They did not do an analysis specific to the hypothyroxinemic population. The authors also found that maternal thyroid hormones over the course of pregnancy predicted child anxiety/depression better than first-trimester fT4 levels alone.

## Detailed Summary of Ghassabian et al. (2011)

Maternal Thyroid Function during Pregnancy and Behavioral Problems in the Offspring: the Generation R Study

Ghassabian et al. (2011) evaluated the association of maternal thyroid function with parent-reported problem behavior in offspring up to 3 years old. Maternal thyroid function data, including TSH and fT4, were assessed at the first pre-natal visit (mean = 13.3 GW, SD = 1.7). Child behavior, or the degree of internalizing (anxiety, sadness, and withdrawn behavior) and externalizing (attention problems and aggressive behavior) problems, was assessed with the CBCL at ages 1.5 and 3 years of age. The CBCL contains a standardized rating of behavioral and emotional problems based on 99 items that, in this study, was completed by both mothers and fathers. Pregnant mothers were recruited from the Generation R cohort, a population-based cohort in Rotterdam, Netherlands. The main exclusion criteria was if women were using thyroid medication. In total, 3,736 children of mothers with maternal thyroid hormone data and behavioral data (from either mothers' or fathers' responses to the CBCL) were included in the study. CBCL scores were analyzed as continuous dependent variables. Multiple linear regression analysis was used to assess maternal thyroid hormone function associations with their child's internalizing and externalizing scores at 1.5 and 3 years of age. Confounders included in the final model were maternal age, educational level and psychopathology, child's gender and ethnicity, mode of delivery, and gestational age at the time of thyroid sampling.

Maternal hypothyroxinemia was mildly associated with higher internalizing scores of children at 1.5 and 3 years of age ( $\beta$  = -0.19 for hypothyroxinemic mothers; 95% CI, -0.75 and 0.37; p = 0.51). In addition, there was no association between maternal hypothyroxinemia and externalizing scores ( $\beta$  = 0.17 for hypothyroxinemic mothers; 95% CI, -0.53 and 0.87; p = 0.64). There was also no association between maternal fT4 and TT4 with the children's externalizing scores at 1.5 and 3 years of age.

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Personal communication with study authors revealed that anxiety/depression scores were inverse transformed.

*Key finding:* The authors did not find maternal fT4 to be associated with internalizing or externalizing scores of children. Maternal hypothyroxinemia was mildly associated (though not statistically significant) with higher internalizing or externalizing scores of children at 1.5 and 3 years of age.

#### Detailed Summary of Modesto et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ]

Maternal Mild Thyroid Hormone Insufficiency in Early Pregnancy and Attention-Deficit/Hyperactivity Disorder Symptoms in Children

Modesto et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] examined the relationship between exposure to maternal mild thyroid hormone insufficiency in early pregnancy and ADHD in offspring at 8 years of age. Maternal thyroid function data, including TSH, fT4, and TPO Abs, were assessed during the first or second trimester at a mean of 13.6 (range 6.6–17.9) weeks of gestation. Mothers using the Conners' Parent Rating Scale-Revised Short Form assessed children's ADHD at 8 years of age. This assessment includes the four scales: Cognitive Problems/Inattention, Hyperactivity, Oppositional, and the ADHD index (ADHDi). The possible range of scores for this test is 0–36 with higher scores indicating more ADHD symptoms. Maternal hypothyroxinemia was defined as TSH levels within the population reference range (0.1 to 2.5 mIU/L) and low fT4 levels below the 5<sup>th</sup> percentile of the sample (< 0.85 ng/dL = 10.94 pmol/L). Positive results for TPO Abs were defined as plasma concentrations of 100 IU/mL or greater. Pregnant mothers were recruited if they lived in the municipality of Rotterdam, Netherlands, and were in the Generation R Study. The main exclusion criteria for mother-child pairs were birth of twins, availability of thyroid hormone data, and if no return assessment occurred of child with mother (or caregiver) by 6 years of age of the child.

Continuous measures of TSH and fT4 were divided by their standard deviations to easily compare their associations with outcomes ([ REF \_Ref475433275 \h \\* MERGEFORMAT ]). The authors examined associations between maternal thyroid function (across the entire range of fT4, not just the hypothyroxinemic range) with a child's ADHDi and Oppositional scores using linear regression. To satisfy the assumption of normality, these scores were transformed using natural logarithms. Regression coefficients were subsequently exponentiated and converted to percentage differences. Models were adjusted for the possible confounders of child age, sex, ethnic background, and parity, and maternal education level, smoking history, psychopathologic symptoms in pregnancy, age, marital status, household income, BMI, and gestational age at the time of blood sampling for maternal thyroid function data. Models were additionally adjusted for the following potential confounders in separate steps to test specificity of the findings: child's autistic symptoms, nonverbal IQ, Oppositional scores, or ADHDi scores.

In general, children of mothers with maternal hypothyroxinemia (127 mothers) in early pregnancy had higher ADHDi scores at 8 years old when compared to children without exposure to maternal hypothyroxinemia, after adjusting for the factors described above (7% increase in ADHDi scores (95% CI: 0.3%-15%)). Additional adjustments for a child's autistic symptoms or Oppositional scores did not affect this association. The results remained mostly unchanged when women with elevated TPO Abs were excluded. A separate analysis focusing on a fetal gestational age of less than 13 weeks at the time of sampling found statistically significant associations, after adjusting for age and sex, between maternal hypothyroxinemia and ADHDi scores ( $\beta = 0.15$ ; 95% CI: 0.01, 0.28; p = 0.04) and between fT4 and ADHDi scores ( $\beta = -0.02$ ; 95% CI: -0.04, -0.001; p = 0.04) at 8 years of age. However, when fully adjusted (for child age, sex, ethnic background, maternal educational level, age,

history of smoking, psychopathologic symptoms during pregnancy, parity, marital status, household income, BMI, and time of blood sampling in pregnancy), the relationship between fT4 and ADHDi scores was no longer statistically significant ( $\beta = -0.02$ ; 95% CI: -0.03, 0.003; p = 0.10). No associations were found between maternal subclinical hypothyroidism and ADHDi score in children; and maternal thyroid function in early pregnancy and children's Oppositional scores at 8 years of age.

# Table [ SEQ Table \\* ARABIC ]. Summary of Results Evaluating the Relationship Between Maternal Thyroid Function in Early Pregnancy and Children's ADHDi Scores at 8 Years of Age

Thyroid Function Variable	Parent-Reported ADHDi Score: β (95% CI)	Changes in Scores, % (95% CI)
	Maternal Hypothyroxinemia	
Unadjusted	0.08 (0.01, 0.15) <sup>d</sup>	8 (1, 16)
Adjusted for age and sex	0.08 (0.01, 0.15) <sup>d</sup>	8 (1, 16)
Fully adjusted model <sup>c</sup>	0.07 (0.003, 0.14) <sup>d</sup>	7 (3, 15)
	Free T4 Level per Standard Deviation	
Unadjusted	-0.01 (-0.02, 0.01)	-1 (-2, 1)
Adjusted for age and sex	-0.01 (-0.02, 0.01)	-1 (-2, 1)
Fully adjusted modelc	-0.01 (-0.02, 0.01)	-1 (-2, 1)
	TSH Level per Standard Deviation	
Unadjusted	-0.01 (-0.02, 0.01)	-1 (-2, 1)
Adjusted for age and sex	-0.01 (-0.02, 0.003)	-1 (-2, 0.3)
Fully adjusted model <sup>c</sup>	-0.01 (-0.02, 0.01)	-1 (-2, 1)
Maternal Fi	pothyroxinemia (TSH aut point 0.03-4.0	4 mIU/LP
Unadjusted	0.05 (-0.01, 0.11)	5 (-1, 12)
Adjusted for age and sex	0.05 (-0.01, 0.11)	5 (-1, 12)
Fully adjusted model <sup>c</sup>	0.05 (-0.01, 0.11)	5 (-1, 12)
Maletaal Hyperbyre	diremia (gestaliene) ago < 16 vest sat	ime of sampling)!
Unadjusted	0.17 (0.03, 0.31) <sup>d</sup>	18 (3, 36)
Adjusted for age and sex	0.15 (0.01, 0.28) <sup>d</sup>	16 (1, 32)
Fully adjusted model <sup>c</sup>	0.13 (-0.01, 0.26)	14 (-1, 29)
Free Li Level per Sissuis	to Bevioline (secretion legs 11 700	s at time of sampling)
Unadjusted	-0.02 (-0.04, -0.004) <sup>d</sup>	-2 (-4, -0.4)
Adjusted for age and sex	-0.02 (-0.04, -0.001) <sup>d</sup>	-2 (-4, -0.1)
Fully adjusted model <sup>c</sup>	-0.02 (-0.03, 0.003)	-2 (-3, 0.3)
TSH Local per Standar	Deviation (gestational age < 16 vecto	at time of sampling)
Unadjusted	-0.004 (-0.02, 0.01)	-0.4 (-2, 1)
Adjusted for age and sex	-0.01 (-0.02, 0.01)	-1 (-2, 1)
Fully adjusted model <sup>c</sup>	-0.004 (-0.02, 0.01)	-0.4 (-2, 1)

Source: Modesto et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ].

<sup>&</sup>lt;sup>a</sup> Results are presented in ng/dL unless otherwise noted.

<sup>&</sup>lt;sup>b</sup> Categorical analysis.

<sup>&</sup>lt;sup>c</sup> Adjusted for child age, sex, ethnic background, maternal educational level, age, history of smoking, psychopathologic symptoms during pregnancy, parity, marital status, household income, BMI, and time of blood sampling in pregnancy.

<sup>&</sup>lt;sup>d</sup> Statistically significant at p < 0.05.

e The SD for fT4 is 0.29 ng/dL.

<sup>&</sup>lt;sup>f</sup>The SD for TSH is 1.3 mIU/L.

**Key finding:** Maternal hypothyroxinemia in early pregnancy was associated with higher ADHDi scores in 8-year-old children compared with non-exposed children, after adjusting for confounders (7% increase in ADHDi scores (95% CI: 3%-15%)). When considering fT4 as a continuous variable, the association between maternal fT4 and offspring ADHDi scores was attenuated with the adjustment of confounders.

#### Detailed Summary of Oostenbroek et al. (2017)

Maternal Hypothyroxinaemia in Early Pregnancy and Problem Behavior in 5-Year-Old Offspring

Oostenbroek et al. (2017) investigated the association between maternal thyroid hormone function in early pregnancy and several types of problem behavior in offspring at age 5-6 years. The authors conducted a longitudinal study that used data from 2,000 mother-child pairs from the Amsterdam Born Children and their Development study. Primary exclusion criteria included children with congenital malformations and children of women who used anti-thyroid drugs during pregnancy. The median gestational age at maternal blood thyroid hormone (fT4 and TSH) measurement was 12.9 weeks (interquartile range: 11.9–14.1). Maternal hypothyroxinemia was defined at the cut points of fT4 level < 5th percentile (<7.75 pmol/L) and fT4 level < 10th percentile (<8.15 pmol/L).

Overall problem behavior, hyperactivity/ inattention, conduct problems, emotional problems, peer relationship problems, and prosocial behavior were measured at age 5–6 years using the Strengths and Difficulties Questionnaire (SDQ), which was filled out by both parents and teachers. The SDQ is a brief, validated questionnaire that can be filled out by the child's parents, caregivers, and/or teachers and is suitable for 3- to 16-year-old children. The questionnaire consists of 25 items subdivided into scales of hyperactivity/inattention, emotional problems, conduct problems, peer relationship problems, and prosocial behavior. Associations of continuous fT4 and TSH across the entire range of SDQ were assessed via multivariate logistic regression analysis. The adjusted model included ethnicity, years of education, pre-pregnancy BMI, hypertension, smoking during pregnancy of at least one cigarette per day, and anxiety level. Analyses were repeated after exclusion of children who were born very pre-term (i.e., <32 weeks) and/or with a very low birth weight (i.e., <1,500 g) and also after exclusion of children whose mothers were TPO Ab positive or on thyroid hormone supplements at their first visit.

[ REF \_Ref507166644 \h ] summarizes the associations between maternal thyroid hormone function parameters and parent or teacher report of SDQ as assessed by Oostenbroek et al. (2017). None of the results presented reached statistical significance at p < 0.05. The authors also analyzed the same outcomes listed in [ REF \_Ref507166644 \h ] with hypothyroxinemia at the <5th percentile and <10th percentile cut points. Significant results were seen for teacher reported hyperactivity/inattention in both the crude and adjusted model for the <5th percentile cut point (crude model: OR = 1.97, 95% CI: 1.19, 3.33, p = 0.008; adjusted model: OR = 1.70, 95% CI: 1.01, 2.86, p = 0.04) and in the crude model for the <10th percentile cut point (OR = 1.68, 95% CI: 1.15, 2.47; p = 0.008). Additionally, there was an association of borderline significance found between decreasing maternal fT4 and increased teacher-reported problem behavior.

# Table [ SEQ Table \\* ARABIC ]. Associations between Maternal Thyroid Function Parameters and Parent or Teacher Report of SDQ

Outcomes (fT4, Continuous, pmol/L)	Crude Model OR (95% CI)	Adjusted Model OR (95% CI) <sup>a</sup>
	otal Problems	
Parent reported	1.05 (0.91, 1.22)	1.10 (0.96, 1.27)
Teacher reported	0.85 (0.76, 0.96)	0.89 (0.79, 1.00)
Нурег	de l'antral de l'antra	
Parent reported	0.97 (0.86, 1.09)	1.01 (0.90, 1.13)
Teacher reported	0.91 (0.82, 1.01)	0.95 (0.85, 1.05)
Sitt	fire Control	
Parent reported	1.03 (0.90, 1.18)	1.07 (0.93, 1.22)
Teacher reported	0.96 (0.83, 1.12)	0.95 (0.82, 1.12)
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Parent reported	0.92 (0.82, 1.03)	0.94 (0.84, 1.05)
Teacher reported	1.2 <sup>b</sup> (0.92, 1.13)	1.04 (0.94, 1.15)
Peer Re	Property Control of the Control of t	-
Parent reported	0.99 (0.89, 1.10)	1.02 (0.92, 1.14)
Teacher reported	1.04 (0.93, 1.18)	1.07 (0.95, 1.20)
Prosec	d Creston School	-
Parent reported	1.02 (0.93, 1.13)	1.03 (0.93, 1.14)
Teacher reported	1.02 (0.94, 1.11)	1.03 (0.95, 1.12)

Source: Adapted from Oostenbroek et al. (2017, Tables 3 and 4).

*Key finding*: An association of borderline significance was found between decreasing maternal fT4 and increasing teacher-reported problem behavior. Significant odds of increased hyperactivity/ inattention were seen in children of hypothyroxinemic mothers compared to mothers without hypothyroxinemia.

<sup>&</sup>lt;sup>a</sup> The Adjusted Model was adjusted for ethnicity, years of education, pre-pregnancy BMI, hypertension, smoking during pregnancy of at least 1 cigarette per day, and anxiety level.

<sup>&</sup>lt;sup>b</sup> Result reported as "10.2" but this is assumed to be an error based on the 95% CI reported and other SDQ results presented.

#### **Autism**

Detailed Summary of Román et al. [ADDIN EN.CITE <EndNote><Cite
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Association of Gestational Maternal Hypothyroxinemia and Increased Autism Risk

### Román et al. [ ADDIN EN.CITE < EndNote > < Cite

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V</author><author>Hofman, A </author><author>de Rijke, Y B</author><author>Verhulst, F C</author><author>Tiemeier, H</author></author></contributors><titles><title>Association of gestational maternal hypothyroxinemia and increased autism risk</title><secondary-title>Annals of Neurology</secondary-title></title><periodical><full-title>Annals of Neurology</full-title></periodical><periodical><periodical><periodical><periodical>

742</pages><volume>74</volume><number>5</number><section>733</section><dates><year>20 13</verr></dates><urls></urls><electronic-resource-num>10.1002/ana.23976</electronic-resourcenum></record></Cite></EndNote>] conducted a study to examine the role of maternal hypothyroxinemia in autistic symptoms in children at 6 years of age. Participants were recruited from a population-based cohort in the Netherlands and included in the study complete biomarker and assessment data. Maternal TSH, fT4, and TPO Ab were measured near the beginning of the second trimester at a mean of 13.4 GW. Autistic symptoms were assessed via two questionnaires administered to parents: the Pervasive Developmental Problems Scale (PDP) and the Social Responsiveness Scale (SRS). The PDP is a subscale of the Child Behavior Checklist for Toddlers, which is an assessment of problem behaviors in children that is typically completed by parents or caregivers. The PDP measures autistic symptoms and has good predictive validity for identifying preschoolers at risk of ASD. The SRS is based on parent observations of social, language, and repetitive behaviors in a natural setting. Román et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Román</Author><Year>2013</Year><RecNum>26</RecNum><Displ ayText>(2013)</br>

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14 levels and PDP and SRS variables using linear and logistic regression models. Maternal fT4 levels were analyzed as a continuous variable (divided by the SD to facilitate interpretation) and categorically using definitions of mild and severe hypothyroxinemia (< 10th and 5th percentile of the sample, respectively). Children with borderline and clinical behavioral problems were defined as those above the 93rd and 98th percentiles on the PDP, respectively. A probable autistic child was defined by a PDP score > 98th percentile and an SRS score in the top 5% of the sample. Regression models were adjusted for confounders including sex, ethnicity, gestation age, birthweight, parental age, education, smoking history, and intelligence. In the linear regression analysis PDP and SRS scores were square root-transformed to satisfy the normality assumption.

## Román et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Román</Author><Year>2013</Year><RecNum>26</RecNum><Displ ayText>(2013)</DisplayText><record><rec-number>26</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047642">26</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Román, G C</author><author>Ghassabian, A</author><author>Bongers-Schokking, J</author><author>Jaddoe, V W V</author><author>Hofman, A </author><author>de Rijke, Y B</author><author>Verhulst, F C</author>Tiemeier, H</author></authors></contributors><title>Association of gestational maternal hypothyroxinemia and increased autism risk</title><secondary-title>Annals of Neurology</secondary-title></title>

title></periodical><pages>733-742</pages><volume>74</volume><number>5</number><section>733</section><dates><year>20 13</pages><urls><urls><lelectronic-resource-num>10.1002/ana.23976</electronic-resource-num></record></cite></EndNote>] found no associations between the continuous measures of maternal TSH and fT4 per SD and children's borderline or clinical behavioral problems. However, the authors found that severe maternal hypothyroxinemia was associated with almost four-fold increased odds of having a probable autistic child ( $OR_{adj} = 3.89, 95\%$  CI = 1.83, 8.20). Based on linear regression analysis, severe maternal hypothyroxinemia was also associated with a higher score of child autistic symptoms based on the PDP ( $\beta = 0.23, 95\%$  CI = 0.07, 0.37) and SRS ( $\beta = 0.05, 95\%$  CI = 0.01, 0.10) scales. However, the authors did not find this relationship to be concentration dependent. This is demonstrated by the negative associations seen between fT4 per SD and only mild hypothyroxinemia and the outcomes of lower PDP and SRS scale scores, as shown in [ REF \_Ref457851152 \h ]. It is unknown if there is a concentration-dependent relationship between fT4 and scores on the PDP or SRS scale (or risk of autism) when considering only individuals with severe

hypothyroxinemia, because this analysis was not conducted. No increased risk of autism was found for children of mothers positive for TPO Ab.

# Table [ SEQ Table \\* ARABIC ]. Summary of Key Findings from Román et al. [ ADDIN EN.CITE <EndNote><Cite

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C</author><author>Ghassabian, A</author><author>Bongers-Schokking,

H</author></authors></contributors><title>>Association of gestational maternal hypothyroxinemia and increased autism risk</title><secondary-title>Annals of Neurology</secondary-title></title><periodical><full-title>Annals of Neurology</full-title></periodical><pages>733-

742</pages><volume>74</volume><number>5</number><section>733</section><date>><year>2013</year></dates><urls></urls><electronic-resource-

num>10.1002/ana.23976</electronic-resource-num></record></Cite></EndNote>1

Variable	Pervasive Developmental Problems (PDP) β (95% CI), <i>P</i>	Social Responsiveness Scale (SRS) β (95% CI), P
TSH per SD	0.01 (-0.02, 0.03), 0.68	0.01 (-0.001, 0.01), 0.12
fT4 per SD	-0.03 (-0.06, -0.01), 0.04	-0.002 (-0.01, 0.01), 0.67
Mild hypothyroxinemia	0.09 (-0.02, 0.19), 0.62	0.02 (-0.01, 0.04), 0.33
Only mild hypothyroxinemia	-0.04 (-0.17, 0.004)	-0.02 (-0.06, 0.02), 0.28
Severe hypothyroxinemia	0.23 (0.07, 0.37), 0.04	0.05 (0.01, 0.10), 0.01

Source: Román et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Román</Author>Year>2013</Year><RecNum>26</RecNum><Suffix>`, Table 3</Suffix><DisplayText>(2013, Table 3)</DisplayText><record><rec-number>26</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047642">26</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author><author>C</author><author>Ghassabian, A</author><author>Bongers-Schokking, J</author><author>Jaddoe, V W V</author><author>Hofman, A </author><author>de Rijke, Y B</author><author>Verhulst, F C</author><author>Tiemeier,

H</author></authors></contributors><titles><title>Association of gestational maternal hypothyroxinemia and increased autism risk</title><secondary-title>Annals of Neurology</full-title>Annals of Neurology</full-title></periodical><pages>733-

742</pages><volume>74</volume><number>5</number><section>733</section><dates><year>2013</year></dates><urls></urls><electronic-resource-num>10.1002/ana.23976</electronic-resource-num></record></Cite></EndNote>].

Notes: Mild hypothyroxinemia (n = 136): 0.03 < TSH < 2.5 mIU/L and  $fT_4 < 11.82$ .

Only mild hypothyroxinemia (n = 295): 0.03 < TSH < 2.5 mIU/L and  $10.99 < fT_4 < 11.82$ .

Severe hypothyroxinemia (n = 159): 0.03 < TSH < 2.5 mIU/L and  $fT_4 < 10.99 \text{ pmol/L}$ .

The  $\beta$ 's are not interpretable, as the mathematically transformed scores were used in the analysis.

**Key finding:** Severe maternal hypothyroxinemia at a mean gestational age of 13.4 weeks was significantly associated with an almost four-fold increased risk of being a probable autistic child and a higher score on a scale of autistic symptoms among 6-year-old children.

#### **Other Endpoints**

Detailed Summary of Gyllenberg et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ]

Hypothyroxinemia During Gestation and Offspring Schizophrenia in a National Birth Cohort

Gyllenberg et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA] examined the relationship between maternal thyroid deficiency during gestation and schizophrenia in offspring with a nested case-control study. Retrospective study data were obtained from the Finnish Prenatal Study of Schizophrenia (FiPS-S), a subset within the Finnish Maternity Cohort (FMC). The FMC contains archived maternal blood samples from most pregnancies in Finland since 1983. Offspring in the cohort with schizophrenia or schizoaffective disorder (both referred to as schizophrenia hereafter) were identified using the Finnish Hospital and Outpatient Discharge Register and were included if they were within the age of risk for schizophrenia. For this study, records were included if the individual was diagnosed at 26 years old or less before December 31, 2009. The national hospital and outpatient information recorded diagnoses for psychiatric hospital admissions and psychiatric outpatient treatment visits and was obtained for FiPS-S participants. Maternal fT4 and TSH were measured via blood samples drawn during the first trimester or early second trimester (2-4 months of pregnancy). TPO Ab was not measured. Blood samples were excluded if they contained HIV, syphilis, or hepatitis.

After cases of schizophrenia were identified, they were matched 1:1 with control subjects. Control subjects were drawn randomly from the FMC if they did not have schizophrenia or other psychotic disorders. Maternal blood serum samples of 1,010 case-control pairs were assessed for fT4, and 948 case-control pairs were assessed for TSH. Maternal thyroid deficiencies measured included hypothyroxinemia (classified as  $fT4 \le 10^{th}$  percentile and TSH outside of the  $5^{th}$ - $95^{th}$  percentile), hypothyroidism, subclinical hypothyroidism, hyperthyroidism, and subclinical hyperthyroidism.

Chi squared, Fisher's exact, and *t*-tests were used to evaluate relationships between covariates, thyroid hormones, and schizophrenia. fT4 levels were log-transformed before being analyzed as continuous variables to correct the skewness of the data. Categorical transformation of continuous variables allowed maternal thyroid disorder data to be interpreted using logistic regression models.

A summary of results from Gyllenberg et al. (2016) regarding the association between maternal hypothyroxinemia and offspring schizophrenia is presented in [ REF \_Ref512854775 \h ]. Maternal hypothyroxinemia was associated with an increased odds of schizophrenia in offspring (OR = 1.75; 95% CI: 1.22-2.50, p = 0.002). The proportion of subjects with maternal hypothyroxinemia was 11.8% among schizophrenic subjects compared with 8.6% among control subjects. This association remained significant after adjusting for maternal psychiatric history, province at birth, and maternal smoking status (OR = 1.70; 95% CI: 1.13-2.55; p = 0.010).

Defining maternal fT4 as a continuous variable, there was a linear association between log units of maternal fT4 and log odds of schizophrenia in offspring. A conditional logistic regression analysis revealed that the odds of schizophrenia decreased by almost 50% per log unit increase of maternal fT4 (OR = 0.54; 95% CI: 0.31-0.94; p = 0.028). This relationship remained significant after adjusting for maternal psychiatric history and province of birth. However, after individually adjusting for

maternal smoking during early gestation and collectively adjusting for maternal psychiatric history, province of birth, and maternal smoking, the association between maternal hypothyroxinemia and offspring schizophrenia was no longer significant. The adjustment for maternal smoking resulted in a greater than 20% change in the odds ratio. According to the authors, "this finding suggests that the relationship between maternal fT4 and schizophrenia is present only when fT4 levels are below a particular threshold" (p. 966, defined in this study as under the  $10^{th}$  percentile of the distribution). The association between log-transformed TSH and schizophrenia was not significant. In analyzing for potential mediators in the association between hypothyroxinemia and schizophrenia, pre-term birth and birthweight both decreased the odds ratios, and the p-value was no longer significant.

Table [ SEQ Table \\* ARABIC ]. Summary of Results Regarding the Association Between Maternal Hypothyroxinemia and Offspring Schizophrenia

Model	Odds Ratio (95% CI)	p-Value
Categorical Hypothyroxinemia <sup>a</sup> During Ear	V Costalion	
Unadjusted	1.75 (1.22-2.50)	0.002
Individually adjusted for maternal psychiatric history	1.92 (1.30-2.83)	0.001
Individually adjusted for province of birth	1.78 (1.23-2.57)	0.002
Individually adjusted for maternal smoking during early gestation <sup>b</sup>	1.62 (1.12-2.37)	0.012
Adjusted for maternal psychiatric history, province of birth, and maternal smoking	1.70 (1.13-2.55)	0.010
Log-Transformed fT4 During Early Ge	Station	
Unadjusted	0.54 (0.31-0.94)	0.028
Individually adjusted for maternal psychiatric history	0.53 (0.30-0.94)	0.029
Individually adjusted for province of birth	0.57 (0.33-0.999)	0.049
Individually adjusted for maternal smoking during early gestation <sup>b</sup>	0.71 (0.39-1.27)	0.243
Adjusted for maternal psychiatric history, province of birth, and maternal smoking	0.74 (0.40-1.38)	0.342

Source: Gyllenberg et al. [ADDIN EN.CITE ADDIN EN.CITE.DATA ].

Key finding: An association between fT4 as a continuous variable across the entire fT4 range (i.e., not just the hypothyroxinemic range) and odds of schizophrenia diagnosis was found using conditional logistic regression. However, this relationship was attenuated after controlling for smoking. Considering fT4 as a categorical variable demonstrated that hypothyroxinemia during early to mid-gestation was associated with 70% increased odds of schizophrenia diagnosis in offspring of hypothyroxinemic mothers compared to the offspring of non-hypothyroxinemic mothers.

#### 5.3.3 Results of Step 3

[ REF \_Ref512437248 \h ] presents the ROB ratings and tier rank for each study identified. In the overall tier ranking, all studies except for three (Kasatkina et al., 2006; Oken et al., 2003; Pop et al., 1999) were rated as Tier 1, indicating that the majority of identified studies were of high quality. The Oken et al. (2003) and Pop et al. (1999) papers were rated as Tier 2. Both of these studies were rated

<sup>&</sup>lt;sup>a</sup> Defined as fT4  $\leq$  14.03 pmol/L ( $\leq$ 10<sup>th</sup> percentile) and TSH > 0.24-3.38 mlU/L (> 5<sup>th</sup>-95<sup>th</sup> percentile).

<sup>&</sup>lt;sup>b</sup> Based on log-transformed levels of cotinine, a metabolite of nicotine and biomarker of smoking, in maternal sera from FMC.

as probably high ROB for confounding and for attrition, which was also true for several other studies. The Pop et al. (1999) paper was additionally rated as probably high ROB for exposure assessment because the interassay coefficients of variation for the assay technique were slightly above the EPA's specified maximum of 10%, ranging from 11.1% to 12.2%. Because of this one additional probably high ROB rating, the Pop et al. (1999) paper received a Tier 2 ranking instead of a Tier 1 ranking. Similarly, the Oken et al. (2003) paper was rated as probably high ROB for outcome assessment because, in addition to issues with confounding and attrition, it did not provide evidence of blinding, and thus was ranked as Tier 2. The Kasatkina et al. (2006) paper was the only study rated as Tier 3 (i.e., low quality). This study was rated as definitely high ROB for confounding because it did not adjust for confounders. It was also rated as probably high ROB for three components—attrition, exposure assessment, and outcome assessment—since very few details on methodology were provided in the paper.

In the individual ROB ratings, there were several notable similarities across studies. No studies accounted for all of the primary covariates, and thus none were rated as definitely low ROB for confounding. Similarly, none used the gold standard for fT4 measurement; and the EPA did not assign any definitely low ROB ratings for exposure assessment. The majority of studies were rated as probably high ROB for attrition/exclusion. This is because most studies examined a subset of individuals drawn from a larger cohort, and there were either significant differences in characteristics of respondents and non-respondents or an incomplete non-response analysis. Details on the individual ROB ratings for each study can be found in Appendix E.

# Table [ SEQ Table \\* ARABIC ]. Risk-of-Bias Ratings and Overall Tier for Each Study

	y is rated as "definitely low" or			ROB Ratings			nse an?	ake Eke
questions and i risk of bias for a • Tier 2: Study i 3rd tiers. • Tier 3: A study	• Tier 3: A study is rated as "definitely high" or "probably high" risk of bias for the majority of		Were outcome data complete without attrition or exclusion from analysis?	Can we be confident in the exposure characterization?	Can we be confident in the outcome assessment?	Were all measured outcomes reported?	Did the study provide dose-response information specific to children of hypothyroxinemic pregnant women?	Did the study measure iodine intake in participants?
Ranking	Study	Did the study design or analysis account for important confounding and modifying variables?	Were witho from a	Can v	Can v	Were all m reported?	Did th inforn hypot	Did th
	Endendijk et al. (2017)	-	-	+	+	++	N	N
	Finken et al. (2013)	-	-	+	++	117	N	N
	Ghassabian et al. (2011)	+	-	+	+	++	N	N
	Ghassabian et al. (2014)	+	-	+	-	++	N	N
	Gyllenberg et al. (2016)	+	++	+	**	**	N	N
	Henrichs et al. (2010)	-	-	+	+	+	N	N
Tier 1	Korevaar et al. (2015)	+	-	+	+++	++	N	N
	Modesto et al. (2015)	+	-	+	+	**	N	N
	Moleti et al. (2015)	+	++	+	+++	-	N	Y
	Noten et al. (2015)	+	++	+	**	++	N	N
	Oostenbroek et al. (2017)	+	-	+	-	++	N	N
	Pop et al. (2003)	-	++	+	**	++	Y	N
	Roman et al. (2013)	+	-	+	+	++	N	N

	s rated as "definitely low" or k of bias for the majority of			ROB Ratings			onse if ien?	ake	
questions and is not rated as "definitely high" risk of bias for any question.  • Tier 2: Study meets neither criteria for 1st or 3rd tiers.  • Tier 3: A study is rated as "definitely high" or "probably high" risk of bias for the majority of questions		Did the study design or analysis account for important confounding and modifying variables?	Were outcome data complete without attrition or exclusion from analysis?	Can we be confident in the exposure characterization?	e be confident in the ne assessment?	Were all measured outcomes reported?	Did the study provide dose-response information specific to children of hypothyroxinemic pregnant women?	Did the study measure iodine intake in participants?	
Ranking	Study	Did the straccount fo confoundir variables?	Were withou	Can we be exposure c	Can we be outcome as	Were all π reported?	Did the informat hypothy	Did the	
Tier 2	Oken et al. (2009)	-	-	+	-	++	N	N	
	Pop et al. (1999)	-	-	-	++	++	Y	N	
Tier 3	Kasatkina et al. (2006)		1	-	-	+	N	N	
Key for ROB Rat	tings:						2		
Definitely Low	Probably Low	Probably High	Definitely High						
++	+	-							

#### 5.3.4 Results of Step 4

After identifying 15 Group 1 studies that are Tier 1 or Tier 2, the EPA evaluated the feasibility of deriving functions for one or more studies to assess the impact that incremental changes in fT4 have on neurodevelopmental outcomes. As explained in the subsequent sections, the studies analyzed were too disparate to conduct a meta-analysis. Ultimately, the EPA focused on five studies, four related to cognition (Pop et al., 1999, 2003, Finken et al., 2013, and Korevaar et al., 2016) and one related to behavior (Endendijk et al., 2017), that could be used to describe how incremental changes in thyroid hormone levels in early pregnancy could result in subsequent changes in neurodevelopment.

#### **Group 1 Studies Not Evaluated Further**

## Cognition

As described in [ REF \_Ref491174324 \h ], of the 16 Group 1 papers identified, 10 of them evaluate cognition using a variety of tests including various IQ tests, Bayley Scales of Infant Development (mental and psychomotor developmental indices), school performance measures, and other validated tests associated with child cognition (Finken et al., 2013; Ghassabian et al., 2014; Henrichs et al., 2010; Kasatkina et al., 2006; Korevaar et al., 2016; Moleti et al., 2016; Noten et al., 2015; Oken et al., 2009; Pop et al., 1999, 2003). Of these 10 studies, two pairs of studies evaluated the same outcomes with the same test. That is, both Korevaar et al. (2016) and Ghassabian et al. (2014) evaluated child IQ using a non-verbal IQ test (Snijders-Oomen Niet-Verbale Intelligentic Test), and both Pop et al. (1999, 2003) papers used the BSID to conduct their neurodevelopmental assessment. The EPA has determined that dose-response analyses will be performed on Korevaar et al. (2016) and the associated data, and both Pop et al. papers.

The EPA has elected not to conduct dose-response analysis on Ghassabian et al. (2014) because it uses the same dataset used in Korevaar et al. (2016) (i.e., the Generation R Cohort). Korevaar et al. (2016) provide a more comprehensive analysis of the fT4/IQ relationship, and little would be gained by conducting dose-response analysis with the effect estimates presented in the Ghassabian et al. (2014) paper. Further, given that Ghassabian et al. (2014) and Korevaar et al. (2016) are both papers evaluating results from the Generation R cohort, it would be inappropriate to combine their effect estimates given the overlap in the populations evaluated [ ADDIN EN.CITE <EndNote><Cite><Author>Dickersin</Author><Year>1992</Year><RecNum>1977</RecNum>< DisplayText>(Dickersin & DisplayText>(Dickersin & DisplayText) = 1977</recnumber><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1523366779">1977</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Dickersin, K </author><author>Berlin, JA</author></authors></contributors><titles><title>Meta-analysis: State of the science</title><secondary-title>Epidemiologic Reviews</secondary-title></title><periodical><fulltitle>Epidemiologic Reviews</full-title></periodical><pages>154-176</pages><volume>14</volume><dates><year>1992</year></dates><urls></urls></record></Cit e></EndNote>].

Moleti et al. (2016) also evaluated fT4's association with IQ but used a different IQ test than Korevaar et al. (2016) did (WISC-IQ vs. Snijders-Oomen Niet-Verbale Intelligentie Test). Moleti et al. (2016) found a consistent positive association between maternal fT4 and VIQ, PIQ, and full-scale IQ in a univariate model though these relationships failed to reach statistical significance (this study had a sample size of less than 60 and is considered a pilot study by the authors). After a critical evaluation of the Moleti et al. (2016) paper, the EPA found it is not possible to conduct dose-response

analysis on this paper because the statistical approach used to estimate the effect estimates in Table 4 of Moleti et al. (2016) is not clear in the paper's methods. Specifically, the authors state "logistic regression models were used to assess the dependence of suboptimal cognitive outcomes (IQ scores < 85, i.e., -1SD) on various explanatory variables and confounders" (Moleti et al., 2016 p. 6). However, the results in Table 4 of Moleti et al. (2016) appear to be an analysis of fT4 and continuous IQ scores, not fT4's effect on suboptimal cognitive outcomes. This is based on the direction of the odds ratios explaining the relationship between both fT4 and urinary iodine with IQ and the conclusion stated by Moleti et al. (2016) that there is a positive association between maternal urinary iodine and full-scale IQ. In addition to the lack of clarity regarding the regression analysis performed in Moleti et al. (2016), there is also a lack of information on any fT4 transformations and functional form. Therefore, the EPA is unable to translate the exp  $\beta$  presented in Table 4 of Moleti et al. (2016) into an effect estimate that can be used to inform the perchlorate MCLG.<sup>19</sup>

It is possible to derive dose-response functions to explain the relationship between maternal fT4 estimates with the CMD at various infant ages based on the data in Kasatkina et al. (2006). However, the EPA has opted to not derive functions from this study because it demonstrated the highest risk of bias of any Group 1 study assessed in the data quality evaluation (Appendix E). There are several reasons for this. First, there was no attempt to evaluate any covariates in the population. It is also unclear how valid and reliable the fT4 measurements were, because no interassay coefficients of variation were reported for the assays used. Similarly, few details were provided on the outcome assessment. The EPA was unable to locate any further information on the methods used to calculate the CMD. Additionally, the results from these analyses are based on fewer than 10 individuals in each age group, and it is unclear if the women represented by the data in the figure were treated with iodine supplementation or not. Lastly, the underlying data for the dose-response function are based on the digitization of figures in the paper, which was done for other analyses. However, the figure's axes were flipped in the original publication, with fT4 as the dependent variable and CMD as the independent variable (i.e., CMD is predicting fT4 instead of fT4 predicting CMD). Given the uncertainties in the analyses, the EPA did not pursue this paper any further.

There are two other papers, Noten et al. (2015) and Oken et al. (2009), that evaluated cognition and were not evaluated for dose-response functions. Noten et al. (2015) evaluated school performance, and Oken et al. (2009) evaluated VRM, PPVT, and WRAVMA. Given that neither of these studies found a significant relationship between maternal thyroid hormone parameters and the cognitive outcomes evaluated in their studies, and that no other study (with which these studies could be combined in a meta-analysis) evaluated the same outcome, it would be difficult to justify using either study to inform an MCLG for perchlorate. Subsequently, neither study was pursued further.

Lastly, analysis was not conducted on Henrichs et al. (2010) given the lack of data needed to conduct the analysis.  $^{20}$  [ REF \_Ref491174324 \h ] (at the end of Section [ REF \_Ref517449014 \r \h ]) summarizes the EPA's conclusions related to the Group 1 – cognition papers.

Attempts were made to contact the study authors, and no response was received.

Attempts were made to contact the study authors, and no response was received.

#### **Behavior**

In total four papers were located in Group 1 that evaluated maternal fT4 and its relationship with behavioral measures (Endendijk et al., 2017; Ghassabian et al., 2011; Modesto et al., 2015; Oostenbroek et al., 2017). Of these four papers only one was considered for further analysis (Endendijk et al., 2017). Modesto et al. (2015) was not evaluated due to a lack of an association between maternal thyroid hormone levels and ADHD. Given that no other study located evaluated maternal thyroid hormone levels as they relate to ADHD, it would be inappropriate to inform the MCLG based on this singular null finding, and subsequently the EPA did not analyze this paper any further.

Ghassabian et al. (2011) and Endendijk et al. (2017) both used the CBCL to evaluate internalizing and externalizing behaviors. However, the two studies reported their results differently, with Endendijk et al. (2017) noting differences at the subscale level (i.e., anxiety/depression, which are subscales of internalizing behavior) and Ghassabian et al. (2011) evaluating results at the higher level (i.e., internalizing or externalizing). Given the differences in the endpoints evaluated when considering Endendijk et al. (2017) and Ghassabian et al. (2011), it would not be appropriate at this time to combine the effect estimates from these papers to evaluate the relationship between maternal fT4 and child behavioral outcomes. Subsequently, the EPA did not conduct any dose-response modeling based on the Ghassabian et al. (2011) paper.

Oostenbroek et al. (2017) also evaluated maternal thyroid hormones as they relate to behavior using the SDQ. There is potential to derive a dose-response function based on the analysis presented in Oostenbroek et al. (2017). However, the analysis in Oostenbroek et al. (2017) is based on a logistic regression analysis informing an odds ratio. Therefore, to conduct an analysis evaluating how a change in fT4 would impact the change in odds of teacher-reported problem behavior, the baseline odds of teacher-reported problem behavior in the United States are needed [ ADDIN EN.CITE <EndNote><Cite><Author>RTI

International</author><Year>2015</Year><RecNum>2024</RecNum><DisplayText>(RTI International, 2015)</DisplayText><record><rec-number>2024</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

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International,</author></authors></contributors><titles><title>User&apos;s manual - Environmental Benefits Mapping and Analysis Program - Community

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face="underline" font="default" size="100%">https://19january2017snapshot.epa.gov/sites/production/files/2015-

04/documents/benmap-ce\_user\_manual\_appendices\_march\_2015.pdf</style></url></related-urls></url></record></Cite></EndNote>]. The EPA has not identified data on the baseline rate of teacher-reported problem behavior in the United States.<sup>21</sup> Subsequently, the necessary data to derive a

5-121

The National Health Interview Survey did include a short version of the SDQ and evaluated outcomes in more than 10,000 U.S. children; however, these are based on parent reports [ ADDIN EN.CITE <EndNote><Cite><Author>YouthInMind</Author><Year>2004</Year><RecNum>1978</RecNum><Dis playText>(YouthInMind, 2004)</DisplayText><record><rec-number>1978</rec-number><forcign-

function from Oostenbroek et al. (2017) are not available. A summary of conclusions related to each behavior paper is provided in [ REF Ref512354169 \h ].

#### Autism

Only one Group 1 paper evaluated continuous maternal fT4 as associated with autism (Román et al., 2013). The data needed to inform a dose-response function from this paper (SD) are not available, and subsequently the EPA did not conduct additional modeling based on this paper. The EPA's evaluation of this paper is summarized in [ REF Ref512354178 \h ].

### Other Endpoints

Gyllenberg et al. (2016) evaluated the relationship between maternal fT4 and schizophrenia. Gyllenberg et al. (2016) found 70% increased odds of schizophrenia diagnosis with maternal hypothyroxinemia. Additionally, they found continuous log maternal fT4 to be associated with schizophrenia (OR = 0.57 (95% CI: 0.33-0.999) when individually adjusted for province of birth and OR = 0.53 (95% CI: 0.30-0.94) when individually adjusted for maternal psychiatric history). However, after adjusting for maternal smoking and psychiatric history, this relationship was attenuated (OR = 0.74 (95% CI = 0.40 -1.38)). Given that no other papers evaluated maternal fT4 and schizophrenia, the EPA has deemed it inappropriate to inform an MCLG based on this study alone given the attenuation of statistical significance with the adjustment for smoking. Therefore, additional analyses were not performed on this paper.

## **Group 1 Studies Considered for Further Analysis**

Five of the 16 Group 1 studies (Pop et al., 1999, 2003; Finken et al., 2013; and Korevaar et al., 2016; Endendijk et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Endendijk</Author><Year>2017</Year><RecNum>1915</RecNum>< DisplayText>(2017)</DisplayText><rccord><rcc-number>1915</rcc-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1503500102">1915</key></foreign-keys>< ref-type name="Journal Article">17</ ref-type>< contributors>< author> Endendijk, J.J.</ author> < author> Wijnen,

H.A.</author><author>Pop, V.J.</author><author>van Baar,

A.L.</author></authors></contributors><titles><title>Maternal thyroid hormone trajectories during pregnancy and child behavioral problems</title><secondary-title>Hormones and Behavior</secondary-title></title></periodical><full-title>Hormones and Behavior</full-title></periodical><pages>84-

92</pages><volume>94</volume><dates><ycar>2017</year></dates><urls></urls></record></Cite ></EndNote>] included data that could be used to quantitatively describe the relationship between thyroid hormone levels in early pregnancy and changes in neurodevelopment. Neurodevelopmental

keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1523369849">1978</key></foreign-keys><ref-type name="Web Page">12</ref-type><contributors><author>YouthInMind</author></authors></contributors><tittles><tittle>No rmative SDQ Data from the USA</title></title><volume>2018</volume><number>April</number><dates><year>2004</year></dates><urls><related-urls><url><tityle face="underline" font="default" size="100%">http://www.sdqinfo.org/norms/USNorm.html</rr>

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Attempts were made to contact the study authors, and no response was received.

outcomes associated with these studies included assessment of the BSID, a Dutch non-verbal intelligence test (the Snijders-Oomen Niet-Verbale Intelligentie Test), and the standard deviation of reaction time. These studies can be identified in the "EPA Evaluation Column" of [ REF \_Ref491174324 \h ] through [ REF \_Ref512805292 \h ], which states that further analysis was conducted.

# Table [ SEQ Table \\* ARABIC ]. Summary of Group 1 Study Results Based on Continuous Measure of fT4 - Cognition

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) <u>Underlined &amp; Bold indicate</u> statistically significant at p < 0.05	Study Summary	EPA Evaluation
Ghassabian et al. [ ADDIN EN.CITE <endnote><c excludeauth="1" ite=""><author> Ghassabian</author><year>2014</year><recnum>10 </recnum>&lt; DisplayText&gt;( 2014)<recrumber>10 recnumber&gt;10 number&gt;fore ign-keys&gt;<key app="EN" db-id="z9t0avxvz dfermedxe5v xfpkax2vzp0ft v29" timestamp="1 432047632">1 0</key><ref-medyseys><ref-gn-keys><ref-gn-keys><ref-gn-keys></ref-gn-keys></ref-gn-keys></ref-gn-keys></ref-medyseys></recrumber></c></endnote>	3,727 Dutch mother-child pairs	No specific data on iodine status provided.	One time during pregnancy (mean: 13.5 GW ± 2 GW, range: 5.7-17.9 GW) (pmol/L)	Nonverbal IQ at 6 years	-0.24 (-0.61, 0.14), fT4 per SD <sup>a</sup>	Maternal hypothyroxinemia during early pregnancy (mean GW = 13.5) was associated with an average reduction of 4.3 points (p = 0.001, 95% CI: -6.68, -1.81) in children's nonverbal IQ at age 6. Continuous measures of maternal thyroid function (TSH, fT4) across the entire range (not the hypothyroxinemic range) and subclinical hypothyroidism were not significantly associated with nonverbal IQ in 6- year-old children.	Further analysis was not performed because this study is based on the same dataset as Korevaar et al. (2016), which provided a more comprehensive analysis of the fT4/IQ relationship. Subsequently, efforts to evaluate the fT4/IQ relationship from the Generation R cohort data are based on Korevaar et al. (2016) or the dataset Dr. Korevaar provided to the EPA.

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Korevaar et al.	3,839 Dutch mother-child	Yes, urine samples were	One time in pregnant	Child IQ at 6 years <sup>b</sup>	Variable	β (SE)	The authors concluded that both	Further analysis was performed to
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
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Moleti et al. [ ADDIN EN.CITE ADDIN EN.CITE.DAT A ] [Tier 1]	59 Sicilian mother-child pairs	Yes, urinary iodide excretion measured in random urine samples throughout gestation and from children at time of cognitive evaluation.	Every 6 weeks, from < 12 GW to term (pmol/L)	Child IQ at 6-12 years	fT4 measured ≤ 12 GW <sup>c</sup> 1.192 (0.980, 1.450) [exp β Verbal IQ] 1.053 (0.881, 1.259) [exp β Performance IQ]] 1.103 (0.928, 1.311) [exp β Full-Scale IQ] fT4 measured 13-18 GW 1.217 (1.015 -1.459) [exp β Verbal IQ] 1.010 (0.855 -1.194)	The authors found that maternal iodine status is predictive of neurodevelopmental outcomes in offspring. The relationship between fT4 at various periods of gestation was consistently associated with verbal, performance, and full-scale IQ, but	Further analysis was not performed because the description of the statistical methods is not sufficient to allow for the derivation of a dose-response function.d

Timing of Maternal Thyroid Hormone Study Study Iodine Status Sampling [ROB Tier] Population Assessed (Units) Endpoints	Effect Estimates  [β based on untransformed fT4, unless otherwise noted] (95 % CI)  Underlined & Bold indicate statistically significant at p < 0.05  [exp β Performance IQ]]  1.165 (0.980-1.385)  Effect Estimates  Study Summary  EPA Evaluation failed to reach statistical significance.
	[exp β Full-Scale IQ] fT4 measured 19-24 GW 1.234 (0.991-1.537) [exp β Verbal IQ] 1.125 (0.897-1.411) [exp β Performance IQ] 1.155 (0.937-1.425)
	[exp β Full-Scale IQ] fT4 measured 25-30 GW 1.137 (0.928-1.393) [exp β Verbal IQ] 1.003 (0.810-1.243) [exp β Performance IQ] 1.031 (0.847-1.254)
	[exp β Full-Scale IQ] fT4 measured GW 31-term 1.191 (0.982-1.445) [exp β Verbal IQ] 1.010 (0.834-1.224) [exp β Performance IQ] 1.070 (0.895-1.280) [exp β Full-Scale IQ]

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints Boyley Scales (MB) au	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Pop et al. (1999) [Tier 2]	220 children of Dutch women, 22 of which had fT4 < 10 <sup>th</sup> percentile from iodine- sufficient area	No specific data on iodine status provided.	First and third trimesters of pregnancy (result presented is specific to first trimester (12 GW) of pregnancy) (pmol/L)	Bayley Scales score (PDI) at 10 months  Bayley Scales score (MDI) at 10 months	8.5 (0.01, 17.04) e.f  Not reported	Maternal fT4 levels in the lowest 10 <sup>th</sup> percentile in the first trimester among an iodine-sufficient population were significantly associated with reduced psychomotor development in 10- month-old infants.	Further analysis was performed to estimate the incremental impact that doses of perchlorate (using the BBDR model) have on PDI.  Data were not presented in the paper to allow for analogous analysis of MDI.

Pop et al. [	57 children (2	No specific	Repeated	Bayley Scales score (PDI)	8.4 (4.0, 12.8) <sup>e</sup>	Maternal fT4 levels in	Further analysis
ADDIN EN.CITE	years old) of Dutch women	data on iodine status	each trimester of	at age 2 years Bayley Scales score (MDI)	6.3 (1.9, 10.62)	the lowest 10 <sup>th</sup> percentile in the first	was performed to estimate the
<endnote><ci< td=""><td>with hypothy-</td><td>provided.</td><td>pregnancy</td><td>at age 2 years</td><td>0.0 (1.0, 10.02)</td><td>trimester were</td><td>incremental impact</td></ci<></endnote>	with hypothy-	provided.	pregnancy	at age 2 years	0.0 (1.0, 10.02)	trimester were	incremental impact
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
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				Charles and C			
Finken et al. [ ADDIN	1,765 Dutch mother-child	No specific data on	One time during	Age 5-6 years Baseline Speed		Continuous maternal fT4 levels at a median	Further analysis was performed to
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ExcludeAuth=" 1"> <author>Fi nken</author>			interquartil e range (IQR):	Reaction time, SD, milliseconds	<u>-4.9 (-9.5, -0.2)</u>	variability among 5- to 6-year-old children (an increase in the	perchlorate (using the BBDR model) have on the
<year>2013<!--</td--><td></td><td></td><td>11.9–14.3)</td><td>Pursuit</td><td></td><td>reaction time SD of</td><td>standard deviation</td></year>			11.9–14.3)	Pursuit		reaction time SD of	standard deviation
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d> <rec- number&gt;9<td></td><td></td><td></td><td>Accuracy stability, millimeters</td><td>0.04 (-0.06, 0.15)</td><td>Considering fT4 as a categorical variable</td><td></td></rec- 				Accuracy stability, millimeters	0.04 (-0.06, 0.15)	Considering fT4 as a categorical variable	
C-				Movement speed, seconds	-0.08 (-0.34, 0.18)	demonstrated that	
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
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Study	Study	Iodine Status	Timing of Maternal Thyroid Hormone Sampling	Fallschatz	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate	Ob. Jr. O	FD4 F. J. J. J.
[ROB Tier] performance in	Population	Assessed	(Units)	Endpoints	statistically significant at p < 0.05	Study Summary	EPA Evaluation
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
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Henrichs et al. [ ADDIN EN.CITE <endnote><ci excludeauth=" 1" te=""><author>H enrichs</author><td>2,779 Dutch mother-child pairs</td><td>No specific data on iodine status provided.</td><td>One time during pregnancy (mean: 13.3 GW ± 1.7 GW) (pmol/L)</td><td>Odds of expressive language delay at 18 and 30 monthsh  Expressive Language Delay 18 months 30 months  18 and 30 months  Nonverbal cognitive delay</td><td>0.95 (0.83, 1.09), fT4 per SD 0.84 (0.71, 0.99), fT4 per SD 0.90 (0.80-1.01), fT4 per SD</td><td>Severe maternal hypothyroxinemia was associated with expressive language delay at 18 and 30 months and noncognitive delay at 30 months. Further, there was an increase in the odds of expressive language delay and noncognitive delay</td><td>Further analysis was not performed because the SD needed to translate this result for a regression analysis was not available.</td></ci></endnote>	2,779 Dutch mother-child pairs	No specific data on iodine status provided.	One time during pregnancy (mean: 13.3 GW ± 1.7 GW) (pmol/L)	Odds of expressive language delay at 18 and 30 monthsh  Expressive Language Delay 18 months 30 months  18 and 30 months  Nonverbal cognitive delay	0.95 (0.83, 1.09), fT4 per SD 0.84 (0.71, 0.99), fT4 per SD 0.90 (0.80-1.01), fT4 per SD	Severe maternal hypothyroxinemia was associated with expressive language delay at 18 and 30 months and noncognitive delay at 30 months. Further, there was an increase in the odds of expressive language delay and noncognitive delay	Further analysis was not performed because the SD needed to translate this result for a regression analysis was not available.
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Study	Study	Iodine Status	Timing of Maternal Thyroid Hormone Sampling	Fadadista	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate	Ohrahi Orimanani	FDA Fredrication
imestamp="14" 32047636">16 <ref- name="Journal Article" type="">17<contribu tors=""><author>Hen richs, J</author><au thor="">Bongers- Schokking, J J<au thor="">Schenk, J J<au thor="">Ghassabi an, A<au thor="">Schmidt, H G<au thor="">Visser, T J<au thor="">Visser, T J<au thor="">Hooijkaas</au></au></au></au></au></au></au></contribu></ref->	Population	Assessed	(Units)	Endpoints	statistically significant at p < 0.05	non-cognitive delay. Additionally, maternal fT4 levels per SD were associated with a reduction in odds of expressive language delay at 30 months (OR = 0.84 per SD of fT4).	EPA Evaluation
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate	Study Summary	EPA Evaluation
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dNote>]							
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Noten et al. [	1,196 Dutch	No specific	One time in	Odds of poor school	Odds Ratio	Maternal	Further analysis
ADDIN	mother-child	data on	pregnant	performance at 5 years	Cadoridad	hypothyroxinemia at	was not performed.
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
ADDIN EN.CITE.DAT A ] [Tier 1]		iodine status provided.	(mean: 12.9 GW; range 11.9- 14.3) (pmol/L)	Arithmetic Test Poor school performance fT4, continuous -Unadjusted -Adjusted <sup>k</sup>	0.88 (0.77, 1.02) 0.92 (0.80, 1.07)	trimester was associated with reduced arithmetic test performance in 5- year-old offspring. Inverse probability weighting	relationship between maternal fT4 as a continuous variable and odds of subnormal arithmetic performance was
				Arithmetic Test Subnormal school performance  T4, continuous -Unadjusted -Adjustedk	0.89 (0.81, 0.98) 0.92 (0.83, 1.02)	demonstrates that non-response bias may be present in this study, and thus the authors caution that the results should be interpreted with	observed, this relationship diminished with the adjustment of confounders and with inverse probability
			Language Test Poor school performance fT4, continuous -Unadjusted -Adjustedk	0.95 (0.84, 1.08) 1.02 (0.90, 1.16)	caution.	weighting. No other studies related maternal fT4 as a continuous variable to subpar school performance, thus,	
				Language Test Subnormal school performance fT4, continuous -Unadjusted -Adjusted <sup>k</sup>	0.98 (0.90, 1.08) 1.02 (0.93, 1.13)		it would not be feasible to conduct a meta-analysis on this outcome to inform an MCLG for perchlorate.

Oken et al. [ ADDIN	500 mother- child pairs in	No specific data on	One time in pregnant	Age 6 months Visual Recognition Memory	-0.04 (-0.8, 0.7) [T4] <sup>i</sup>	Neonatal total T4 levels within the	Further analysis was not performed
		•		~ · · · · · · · · · · · · · · · · · · ·	-0.04 (-0.8, 0.7) [T4] <sup>1</sup> -0.1 (-0.7, 0.5) [T4] <sup>1</sup> 0.004 (-0.6, 0.6) [T4] <sup>1</sup>		_
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at ρ < 0.05	Study Summary	EPA Evaluation
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- a Model adjusted for ethnicity, birth order, maternal age, BMI, marital status, maternal history of smoking, educational levels, maternal psychopathology during pregnancy, household income, and gestational age at time of thyroid sampling.
- b Model adjusted for gestational age, maternal age, smoking, BMI, parity, education level, ethnic origin, fetal sex, and birthweight.
- <sup>c</sup> Model adjusted for maternal age; gestational age at birth; birthweight; child age at time of IQ test; maternal fT4, fT4, and TSH various times during gestation; maternal urinary iodide excretion; family socioeconomic status; maternal parity; sex of child; breastfeeding; major maternal stressful life events; and maternal/parental education.
- d Attempts have been made to contact study authors with no success.
- <sup>e</sup> Based on digitization of figures representing correlation analysis presented in the paper.
- f Model adjusted for maternal depression, psychosocial factors, tobacco and alcohol use during pregnancy, and demographic variables.
- 9 Model adjusted for age, gender, demographic characteristics (e.g., birth order, ethnicity) and pregnancy characteristic (e.g., BMI and diabetes).
- h Expressive language delay at 30 months was defined as an expressive vocabulary score below the 15th age- and gender-specific percentile or no word combinations
- Model adjusted for maternal age, education, pre-natal distress, pre-natal smoking, birthweight, gestational age at blood sampling, and child ethnicity.
- <sup>1</sup> Covariates, confounders, and any model adjustments were not displayed in the study.
- <sup>k</sup> Model adjusted for maternal education, ethnicity, BMI, and depressive symptomatology.
- <sup>1</sup> Model adjusted for gestation length, age at testing, sex, maternal education, and ethnicity.

Table [ SEQ Table \\* ARABIC ]. Summary of Group 1 Study Results Based on Continuous Measure of fT4 – Behavior (including attention-related and problem behavior)

			Maternal Thyroid Hormone		Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI)		
Study [ROB Tier]	Study Population	lodine Status Assessed	Sampling (Units)	Endpoints	Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation

Endendijk et al. [ ADDIN EN.CITE <endnote><c excludeauth="1" ite=""><author> Endendijk<year> 2017</year>&lt; RecNum&gt;191 5&lt; DisplayText&gt;( 2017)<recor d=""><rec-number>1915</rec-number>19151915<fore ign-keys=""><key app="EN" db-id="29t0avxvz dfermedxe5v xfpkax2vzp0ft v29" timestamp="1 503500102">1 915</key><ref-type name="Journ al Article">17</ref-type><contrib utors=""><author>E</author></contrib></fore></recor></author></c></endnote>	442 Dutch pregnant mother-child pairs	No specific data on iodine status provided.	12, 24, 36 GW (pmol/L)	Child behavior problems as measured by internalizing scores (anxiety/depression) from the CBCL assessment (fT4 measured at 12GW)	- <u>0.12</u> a,b	The authors reported a significant effect of first-trimester fT4 on anxiety/depression or internalizing behaviors. The authors also evaluated the trajectory of thyroid hormone levels as they related to the prediction of anxiety/depression and determined that including the trajectory of fT4 in the first-trimester model improved the prediction of child anxiety/depression scores.	Further analysis was performed to evaluate first- trimester fT4 concentration and incremental changes in anxiety/depression as measured by the CBCL
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Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at ρ < 0.05	Study Summary	EPA Evaluation
olume>94lume> <dates> <year>2017<!--<br-->year&gt;</year></dates> <urls></urls> te>>] [Tier 1]							

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Ghassabian et 3,736 Dutch al. [ ADDIN mother-child EN.CITE pairs	mother-child	Dutch No specific r-child data on iodine status	Mean 13.3 GW	Child behavior problems as measured by the CBCL assessment	Effect estimates are beta per SD <sup>c</sup>	Neither maternal fT4 nor maternal hypothyroxinemia was associated with higher child internalizing and	Further analysis was not performed because no
ADDIN EN.CITE.DAT		provided.		Internalizing Scores			relationship was found between
A ] [Tier 1]				fT4 (mother report; 1.5 and 3 years)	0.03 (-0.11, 0.7)	externalizing scores at 1.5 and 3 years of age.	maternal fT4 and internalizing and externalizing
				fT4 (father report; 3 years)	-0.07 (-0.025, 12)	age.	behaviors.
			fT4 (both parents report; 1.5 and 3 years)	0.01 (-0.13, 0.15)		Although Endendijk et al. (2017) also utilized the CBCL,	
				Hypothyroxinemia	-0.19 (-0.75, 0.37)		the effect estimates reported in the
				Externalizing Scores			paper varied from
			fT4 (mother report; 1.5 and 3 years)	0.02 (-0.16, 0.20)		those reported in Ghassabian et al. (2011), and	
			fT4 (father report; 3 years)	-0.14 (-0.39, 0.11)		therefore the estimates could not	
				fT4 (both parents report; 1.5 and 3 years)	-0.02 (-0.16, 0.20)		be combined.
				Hypothyroxinemia	0.17 (-0.53, 0.87)		

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Modesto et al. [ ADDIN EN.CITE ADDIN EN.CITE.DAT A ] [Tier 1]	127 Dutch mother-child pairs	No specific data on iodine status provided.	One time in pregnant mother (mean: 13.6 GW; range 6.6-17.9) (result presented is specific to mothers with TH measured < 13 GW) (ng/dL)	Parent-reported ADHD index scores of offspring at 8 years old Parent Reported ADHDi Score fT4 Level per SD -Unadjusted -Adjusted for age and sex -Fully adjusted modeld fT4 per SD (gestational age < 13 weeks -Unadjusted -Adjusted for age and sex -Fully adjusted modeld	-0.01 (-0.02, 0.01) -0.004 (-0.02, 0.01) -0.01 (-0.02, 0.01) -0.02 (-0.04, -0.004) -0.02 (-0.04, -0.001) -0.02 (-0.03, 0.003)	Maternal hypothyroxinemia in early pregnancy was associated with higher ADHDi scores in 8- year-old children compared with non- exposed children, after adjusting for confounders (7% increase in ADHDi scores (95% CI: 3%- 15%)). When considering fT4 as a continuous variable, the association between maternal fT4 and offspring ADHDi scores was attenuated with the adjustment of confounders.	Further analysis was not performed because no relationship was found between a continuous measure of maternal fT4 and neurodevelopment. No other studies related maternal fT4 as a continuous variable to ADHD or parent-reported oppositional scale; thus, it would not be feasible to conduct a meta- analysis on this outcome to inform an MCLG for perchlorate.

Oostenbroek	200 Dutch	No specific	Median	Total problems		Maternal	Further analysis
et al. [ ADDIN	hypothyroxi-	data on	12.9 GW	Parent reported	4.05 (0.04.4.00)	hypothyroxinemia <5th	was not performed.
EN.CITE	nemic	iodine status	(IQR: 11.9-	- Unadjusted OR	1.05 (0.91, 1.22)	percentile was	Although a
<endnote><ci< td=""><td>mother-child</td><td>provided.</td><td>14.1 GW)</td><td><ul> <li>Fully adjusted OR<sup>e</sup></li> </ul></td><td>1.10 (0.96, 1.27)</td><td>associated with a 1.70</td><td>borderline</td></ci<></endnote>	mother-child	provided.	14.1 GW)	<ul> <li>Fully adjusted OR<sup>e</sup></li> </ul>	1.10 (0.96, 1.27)	associated with a 1.70	borderline
te	pairs		fT4<5th			(95% CI: 1.01–2.86)	statistically
ExcludeAuth="				Teacher reported	0.05 (0.70, 0.00)	increased odds of	significant
1"> <author>O</author>			percentile	- Unadjusted OR	0.85 (0.76, 0.96)	teacher-reported	association was
ostenbroek <td></td> <td></td> <td>(&lt;7.75</td> <td><ul> <li>Fully adjusted ORe</li> </ul></td> <td>0.89 (0.79, 1.00)</td> <td>hyperactivity/</td> <td>found between</td>			(<7.75	<ul> <li>Fully adjusted ORe</li> </ul>	0.89 (0.79, 1.00)	hyperactivity/	found between
uthor> <year>2</year>			pmol/L)			inattention after	teacher-reported
017 <r< td=""><td></td><td></td><td>and</td><td>Hyperactivity/inattention</td><td></td><td>adjustment for</td><td>problem behavior</td></r<>			and	Hyperactivity/inattention		adjustment for	problem behavior
ecNum>1914<			fT4<10th	Parent reported		confounders. By	and maternal fT4
/RecNum> <di< td=""><td></td><td></td><td>percentile</td><td><ul> <li>Unadjusted OR</li> </ul></td><td>0.97 (0.86, 1.09)</td><td>increasing the cut-off</td><td>additional data</td></di<>			percentile	<ul> <li>Unadjusted OR</li> </ul>	0.97 (0.86, 1.09)	increasing the cut-off	additional data
splayText>(20			(<8.15	<ul> <li>Fully adjusted OR<sup>e</sup></li> </ul>	1.01 (0.90, 1.13)	level to<10th	needs, the baseline
17) <td></td> <td></td> <td>pmol/L)</td> <td></td> <td></td> <td>percentile, the odds</td> <td>rate of teacher-</td>			pmol/L)			percentile, the odds	rate of teacher-
ext> <record>&lt;</record>				Teacher reported		ratio became 1.47	reported problem
rec-				- Unadjusted OR	0.91 (0.82, 1.01)	(95% CI: 0.99-2.20).	behavior in the
number>1914				- Fully adjusted ORe	0.95 (0.85, 1.05)	There were no	United States is
<td></td> <td></td> <td></td> <td></td> <td></td> <td>associations between</td> <td>needed.</td>						associations between	needed.
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gn-keys> <key< td=""><td></td><td></td><td></td><td>Parent reported</td><td></td><td>function parameters</td><td></td></key<>				Parent reported		function parameters	
app="EN" db-				- Unadjusted OR	1.03 (0.90, 1.18)	as continuous	
id="z9t0avxvzd				- Fully adjusted ORe	1.07 (0.93, 1.22)	variables and	
fermedxe5vxfp						hyperactivity/	
kax2vzp0ftv29				Teacher reported		inattention as reported	
н				<ul> <li>Unadjusted OR</li> </ul>	1.03 (0.90, 1.18)	by parents, nor with	
timestamp="15				- Fully adjusted ORe	1.07 (0.93, 1.22)	teacher or parent	
03499994">19						reports of other types	
14 <td></td> <td></td> <td></td> <td>Conduct problems</td> <td></td> <td>of problem behavior.</td> <td></td>				Conduct problems		of problem behavior.	
ign-keys> <ref-< td=""><td></td><td></td><td></td><td>Parent reported</td><td></td><td>·</td><td></td></ref-<>				Parent reported		·	
type				- Unadjusted OR	0.92 (0.82, 1.03)		
name="Journal				- Fully adjusted ORe	0.94 (0.84, 1.05)		
Article">17 <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>							
f-				Teacher reported			
type> <contribu< td=""><td></td><td></td><td></td><td>- Unadjusted OR</td><td>1.2<sup>p</sup> (0.92, 1.13)</td><td></td><td></td></contribu<>				- Unadjusted OR	1.2 <sup>p</sup> (0.92, 1.13)		
tors> <authors< td=""><td></td><td></td><td></td><td>- Fully adjusted ORe</td><td>1.04 (0.94, 1.15)</td><td></td><td></td></authors<>				- Fully adjusted ORe	1.04 (0.94, 1.15)		
> <author>Oost</author>				• •			
enbroek,							
M.H.W. <td></td> <td></td> <td></td> <td>Peer relationship problems</td> <td></td> <td></td> <td></td>				Peer relationship problems			
r> <author>Ker</author>				Parent reported			

sten,	- Unadjusted OR	0.99 (0.89, 1.10)
R.H.J. <td><ul> <li>Fully adjusted ORe</li> </ul></td> <td>1.02 (0.92, 1.14)</td>	<ul> <li>Fully adjusted ORe</li> </ul>	1.02 (0.92, 1.14)
> <author>Tros</author>	T l	
, B. <a< td=""><td><u>Teacher reported</u> - Unadjusted OR</td><td>1.04 (0.93, 1.18)</td></a<>	<u>Teacher reported</u> - Unadjusted OR	1.04 (0.93, 1.18)
uthor>Kunst,	- Fully adjusted ORe	1.07 (0.95, 1.10)
A.E.	- I uny adjusted OTC	1.07 (0.55, 1.20)
<author>Vrijko</author>	Prosocial problem behavior	
tte,	Parent reported	
T.G.M. <td>- Unadjusted OR</td> <td>1.02 (0.93, 1.13)</td>	- Unadjusted OR	1.02 (0.93, 1.13)
r> <author>Fin</author>	- Fully adjusted ORe	1.03 (0.93, 1.14)
ken,	, ,	
M.J.J. <td>Teacher reported</td> <td></td>	Teacher reported	
> </td <td>- Unadjusted OR</td> <td>1.02 (0.94, 1.11)</td>	- Unadjusted OR	1.02 (0.94, 1.11)
contributors> <t< td=""><td>- Fully adjusted ORe</td><td>1.03 (0.95, 1.12)</td></t<>	- Fully adjusted ORe	1.03 (0.95, 1.12)
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[ROB Tier]	Population	Assessed	(Units)	Endpoints	Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Study	Study	Iodine Status	Sampling		-, ,		
			Hormone		otherwise noted] (95 % CI)		
			Thyroid		untransformed fT4, unless		
			Maternal		[β based on		
			Timing of		Effect Estimates		

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[Tier 1]

<sup>&</sup>lt;sup>a</sup> The study authors were contacted about the sign on this beta because they state that lower fT4 is associated with more anxiety/depression, but the beta in the paper is positive (higher fT4 is related to higher CBCL scores). The opposite is expected, as indicated here.

b Model adjusted for gestational age and weight at birth, child age at time of CBCL assessment, parity, breastfeeding, mother's age, maternal psychopathology symptoms (pre-natal and at time of CBCL assessment), smoking and alcohol use during pregnancy, educational level, and TPO Ab positivity.

<sup>&</sup>lt;sup>c</sup> Model adjusted for maternal age, educational level and psychopathology, child's gender and ethnicity, mode of delivery and gestational age at the time of thyroid sampling.

d Model adjusted for child age, sex, ethnic background, and parity, and maternal education level, smoking history, psychopathologic symptoms in pregnancy, age, marital status, household income, BMI, and gestational age at the time of blood sampling for maternal thyroid function data.

<sup>&</sup>lt;sup>e</sup> Model adjusted for ethnicity, years of education, pre-pregnancy BMI, hypertension, smoking during pregnancy of at least one cigarette per day, and anxiety level.

### Table [ SEQ Table \\* ARABIC ]. Summary of Group 1 Study Results Based on Continuous Measure of fT4 - Autism

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Román et al. [ ADDIN EN.CITE <endnote><ci excludeauth=" 1" te=""><author>R omán</author><year>2013</year><recnu m="">26</recnu>262013)(2013)<rec ord=""><rec- number="">26<forei gn-keys=""><key app="EN" db-="" id="z9t0avxvzd fermedxe5vxfp kax2vzp0ftv29" timestamp="14 32047642">26 </key><ref- name="Journal Article" type="">17<td>4,039 children of mothers in the Netherlands</td><td>No specific data on iodine status provided.</td><td>One time during pregnancy (mean: 13.4 weeks) (pmol/L)</td><td>Age 6 years Pervasive Developmental Problems Borderline fT4 per SD Clinical fT4 per SD</td><td>0.93 (0.80, 1.09)<sup>a</sup> 0.95 (0.77, 1.17)<sup>a</sup></td><td>Severe maternal hypothyroxinemia at a mean gestational age of 13.4 weeks was significantly associated with an almost four-fold increased risk of being a probable autistic child and a higher score on a scale of autistic symptoms among 6-year-old children.</td><td>Further analysis was not performed because the authors cautioned that the betas relating the continuous measure of maternal fT4 and PDP or SRS scale scores were not interpretable because mathematically transformed scores were used in the analyses and the SD needed for analysis was not provided.</td></ref-></forei></rec-></rec></ci></endnote>	4,039 children of mothers in the Netherlands	No specific data on iodine status provided.	One time during pregnancy (mean: 13.4 weeks) (pmol/L)	Age 6 years Pervasive Developmental Problems Borderline fT4 per SD Clinical fT4 per SD	0.93 (0.80, 1.09) <sup>a</sup> 0.95 (0.77, 1.17) <sup>a</sup>	Severe maternal hypothyroxinemia at a mean gestational age of 13.4 weeks was significantly associated with an almost four-fold increased risk of being a probable autistic child and a higher score on a scale of autistic symptoms among 6-year-old children.	Further analysis was not performed because the authors cautioned that the betas relating the continuous measure of maternal fT4 and PDP or SRS scale scores were not interpretable because mathematically transformed scores were used in the analyses and the SD needed for analysis was not provided.

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<sup>a</sup> Model was adjusted for child's sex, ethnicity, gestational age at birth, and birthweight, maternal age, educational level, smoking history, pre-natal psychopathology, thyroid medication during pregnancy, parity, marital status, folate and C-reactive levels in early pregnancy, time of thyroid sampling, and paternal age. According to the study authors "the betas are not interpretable, as mathematically transformed scores were used in the analyses" (p.6).

### Table [ SEQ Table \\* ARABIC ]. Summary of Group 1 Study Results Based on Continuous Measure of fT4 – Other Outcomes

Study [ROB Tier]	Study Population	lodine Status Assessed	Timing of Maternal Thyroid Hormone Sampling (Units)	Endpoints	Effect Estimates [β based on untransformed fT4, unless otherwise noted] (95 % CI) Underlined & Bold indicate statistically significant at p < 0.05	Study Summary	EPA Evaluation
Gyllenberg et	1,010 Finnish	No specific	One time	Odds of diagnosis of	Various Agesa	An association between fT4	Further analysis was
al. [ ADDIN	mother-child	data on	during	schizophrenia/		as a continuous variable	not performed. While
EN.CITE	pairs	iodine status	pregnancy	schizoaffective		across the entire fT4 range	the relationship
ADDIN		provided.	(2-4	disorder		(i.e., not just the	between maternal
EN.CITE.DAT A]			months pregnancy)	Log-Transformed fT4		hypothyroxinemic range) and odds of schizophrenia	fT4 as a continuous
[Tier 1]			(pmol/L)	During Early		diagnosis was found using	variable and odds of
[]			(P1110#2)	Gestation		conditional logistic	offspring
				Unadjusted	<u>0.54 (0.31, 0.94)</u>	regression. However, this	schizophrenia
						relationship was attenuated	diagnosis was
				Individually adjusted		after controlling for smoking.	observed, statistical
				for maternal	0.52 (0.20, 0.04)	Considering fT4 as a	significance was lost
				psychiatric history	0.53 (0.30, 0.94)	categorical variable demonstrated that	after adjustment for confounders. No
				Individually adjusted		hypothyroxinemia during	other studies related
				for province of birth	0.57 (0.33, 0.999)	early to mid-gestation was	maternal fT4 as a
						associated with 70%	continuous variable
				Individually adjusted		increased odds of	to risk of
				for maternal smoking during early gestation	0.71 (0.39, 1.27)	schizophrenia diagnosis in offspring of	schizophrenia, thus,
				during early gestation	0.71 (0.53, 1.27)	hypothyroxinemic mothers	it would not be
				Adjusted for maternal		compared to the offspring of	feasible to conduct a
				psychiatric history,		non-hypothyroxinemic	meta-analysis on
				province of birth, and		mothers.	this outcome to
				maternal smoking	0.74 (0.40, 1.38)		inform an MCLG for
							perchlorate.

### 5.4 Findings Related to Categorical Analyses Supporting MCLG Development Based on Avoiding Hypothyroxinemia

Given the SAB's recommendation to concentrate on sensitive populations, one of which is the fetuses of hypothyroxinemic mothers, it is also useful to evaluate the impact of maternal hypothyroxinemia as evaluated by the studies identified in both Group 1 and Group 2 (keeping in mind that some Group 1 studies included analysis on both continuous and categorical measures of fT4). That is, in Section [ REF \_Ref482192446 \n \h ], the evaluation of the papers was specific to the concentration-response functions present in the analyses based on a continuous measure of thyroid hormone levels. However, of specific interest may be how neurodevelopmental outcomes vary when comparing offspring of hypothyroxinemic mothers to those of non-hypothyroxinemic mothers. [ REF \_Ref482284683 \h ] presents a summary of the findings of studies that evaluate the impact of maternal hypothyroxinemia, as a categorical variable, on adverse neurodevelopmental outcomes in offspring, including IQ score and performance on the BSID.

Evaluating these results as a whole demonstrates that in different populations, at different ages for neurodevelopmental assessment, and at various cut points for fT4, there is a significant difference in performance on global cognitive tests when comparing the offspring of hypothyroxinemic women to those of non-hypothyroxinemic women (Costeira et al., 2011; Ghassabian et al., 2014; Júlvez et al., 2013; Korevaar et al., 2016; Li et al., 2010; Pop et al., 1999, 2003). These findings are supported by several systematic reviews and meta-analyses including Fan and Wu [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Fan</Author><Year>2016</Year><RecNum>307</RecNum><Display Text>(2016)</DisplayText><record><rec-number>307</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832068">307</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Fan, X.</author><author>Wu, L.</author></author></author></author>><auth-address>a General Surgery

Department and.
b Department of Obstetrics and Gynecology , Central People's Hospital of Siping , Siping , China.</auth-address><title>>title>The impact of thyroid abnormalities during pregnancy on subsequent neuropsychological development of the offspring: a meta-analysis</title><secondary-title>Journal of Maternal-Fetal & Department of Medicine

#xD;b Department of Obstetrics and Gynecology , Central People&apos;s Hospital of thyroid abnormalities during pregnancy on subsequent neuropsychological development of the offspring: a meta-analysis

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6</pages><volume>29</volume><number>24</number><keywords><keyword>Hypothyroxinaemi a</keyword><keyword>meta-analysis</keyword><keyword>neuropsychological development</keyword><keyword>pregnancy</keyword><keyword>subclinical hypothyroidism</keyword><keyword>thyroid peroxidase

antibodies</keyword></keywords><dates><year>2016</year><pub-dates><date>Dec</date></pub-dates></dates><isbn>1476-4954 (Electronic)&#xD;1476-4954 (Linking)</isbn><accession-num>26988121</accession-num><urls><related-urls><url><style face="underline" font="default" size="100%">http://www.ncbi.nlm.nih.gov/pubmed/26988121</style></url></related-urls></urls><electronic-resource-num>10.3109/14767058.2016.1152248</electronic-resource-num></re></ri>/record></Cite></EndNote>], Wang et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ], and Thompson et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Thompson</Author><Year>2018</Year><RecNum>1956</RecNum>< DisplayText>(2018)</DisplayText><rec-number>1956</rec-number><foreign-keys><key

app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1516819485">1956</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Thompson, William</author><author>Russell,

Ginny</author><author>Baragwanath, Genevieve</author><author>Matthews,

Justin</author><author>Vaidya, Bijay</author><author>Thompson-Coon,

Jo</author></authors></contributors></title>Maternal thyroid hormone insufficiency during pregnancy and risk of neurodevelopmental disorders in offspring: A systematic review and meta-analysis</title><secondary-title>Clinical endocrinology</secondary-title></title><periodical><full-title>Clinical Endocrinology</full-

title></periodical><dates><year>2018</year></dates><isbn>1365-

2265</isbn><urls></record></Cite></EndNote>]. Fan and Wu (2016) and Wang et al. (2016) found that hypothyroxinemia was associated with a 5.7-point lower score on intelligence tests and a three-fold increased risk of delayed cognitive development in children, respectively. Thompson et al. (2018) found that maternal hypothyroxinemia is associated with increased risk of cognitive delay, intellectual impairment, or lower scores on performance tests, but they did not find this association with ADHD or autism.

The only studies that did not find a statistically significant effect on any of the evaluated endpoints in [ REF\_Ref482284683 \h ] are Grau et al. (2015) and Hales et al. (2018). As previously discussed Grau et al. (2015) has a high hypothyroxinemic cut point (13.7 pmol/L), and individuals are not iodine deficient. Hales et al. (2018) hypothesizes that the lack of congruence in their findings may be related to varying definitions of suboptimal thyroid function, lack of universal pregnancy-specific reference ranges for thyroid function tests, and the application of various tools to measure cognition in children across the age spectrum.

Additionally, studies identified in the literature review that are not summarized in [ REF \_Ref482284683 \h ] also associated maternal hypothyroxinemia with an offspring's increased risk of schizophrenia (Gyllenberg et al., 2016), ADHD (Modesto et al., 2015), expressive language delay (Henrichs et al., 2010), and other outcomes (Finken et al., 2013; Kooistra et al., 2006; Noten et al., 2015; Päkkilä et al., 2015; Román et al., 2013; van Mil et al., 2012; Oostenbroek et al., 2017). These studies demonstrate the sensitivity of the offspring of hypothyroxinemic mothers to adverse neurodevelopmental effects. An approach to inform the derivation of an MCLG based on avoiding hypothyroxinemia is presented in Section [ REF Ref517616944 \r \h ].

# Table [ SEQ Table \\* ARABIC ]. Summary of Studies That Compared Neurodevelopment Outcomes (BSID or IQ) in the Offspring of Hypothyroxinemic (hX) and Non-Hypothyroxinemic Mothers

Study	Population	Definition of hX for result (pmol/L unless otherwise noted)	Definition of Reference Group	Statistical method	Age of Neurodevelopmental Assessment	Difference in Score When Comparing hX Population to Reference (95% CI)	<i>p</i> -value	Gestational Week of Thyroid Hormone Measurement
Costeira et al. 2011	Portuguese mother-child pairs from iodine-deficient region	fT4 < 10 <sup>th</sup> percentile (< 11.6) (n = 8)	tT4, fT4, tT3, and fT3 between 50 <sup>th</sup> to 90 <sup>th</sup> percentile (n = 36)	t-test on difference between mean scores <sup>a</sup>	24 months	-10.8 (CI: NR) points lower MDI in hX group	> 0.05	< 15
Júlvez et al. 2013	Spanish mother-child pairs	fT4 < 10 <sup>th</sup> percentile (< 8.89) (n = 164) <sup>b</sup>	fT4 > 10 <sup>th</sup> percentile (n = 1,643)	multivariate linear regression <sup>c</sup>	14 months	-2.2 (-4.47, -0.01) points lower MDI in hX group	NR	Median 13, range 8-20
Li et al. 2010	Chinese mother-child pairs	tT4 < 101.79 nmol/L (n = 19)	matched controls with tT4 > 101.79 nmol/L (n = 38)	t-test on difference between mean scores	25-30 months	-9.3 points lower MDI in hX group	0.004	16-20
Pop et al. 1999	Dutch mother- child pairs	fT4 < 10 <sup>th</sup> percentile (10.4) (n = 22) at 12 GW	fT4 >10 <sup>th</sup> percentile (10.4 pmol/L) (n = 198)	t-test on difference between mean scores <sup>d</sup>	10 months	NR because did not find a statistically significant difference	NR	12
Pop et al. 2003	Dutch mother- child pairs	fT4 < 10 <sup>th</sup> percentile (12.4 pmol/L) (n = 57)	fT4 > 10 <sup>th</sup> percentile (12.4 pmol/L) (n = 58)	t-test on difference between mean scores	12 months	-10 (-4, -16) points lower MDI in hX group	0.004	12

Study	Population	Definition of hX for result (pmol/L unless otherwise noted)	Definition of Reference Group	Statistical method	Age of Neurodevelopmental Assessment	Difference in Score When Comparing hX Population to Reference (95% CI)	<i>p</i> -value	Gestational Week of Thyroid Hormone Measurement
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Costeira et al. 2011	Portuguese mother-child pairs from iodine-deficient region	fT4 < 10 <sup>th</sup> percentile (< 11.6) (n = 8)	tT4, fT4, tT3, and fT3 between 50 <sup>th</sup> to 90 <sup>th</sup> percentile (n = 36)	t-test on difference between mean scores <sup>a</sup>	24 months	-13.6 (CI: NR) points lower PDI score in hX group	< 0.05	< 15
Júlvez et al. 2013	Spanish mother-child pairs	fT4 < 10 <sup>th</sup> percentile (< 8.89) (n = 164) <sup>b</sup>	fT4 > 10 <sup>th</sup> percentile (n = 1,643)	multivariate linear regression <sup>c</sup>	14 months	0.28 (-1.94 to 2.51) points higher in PDI score in hX group	NR	Median 13, range 8-20
Li et al. 2010	Chinese mother-child pairs	tT4 < 101.79 nmol/L (n = 19)	matched controls with tT4 > 101.79 nmol/L (n = 38)	t-test	25-30 months	-7.6 (CI: NR) points lower PDI score in hX group	0.007	16-20
Pop et al. 1999	Dutch mother- child pairs	fT4 < 10 <sup>th</sup> percentile (10.4) (n = 22) at 12 GW	fT4 >10 <sup>th</sup> percentile (10.4 pmol/L) (n = 198)	t-test on difference between mean scores <sup>d</sup>	10 months	-7.4 (-1.1, -13.9) points lower PDI score in hX group	NR	12
Pop et al. 2003	Dutch mother- child pairs	fT4 < 10 <sup>th</sup> percentile (12.4 pmol/L) (n = 57)	fT4 > 10 <sup>th</sup> percentile (12.4 pmol/L) (n = 58)	t-test on difference between mean scores	24 months	-10 (-6, -16) points lower PDI score in hX group	0.005	12
		,						
Ghassabian et al. 2014	3,727 Dutch mother-child pairs	fT4 < 5 <sup>th</sup> percentile (< 10.99)	fT4 > 5 <sup>th</sup> percentile (> 10.99 pmol/L) (n = 3,598)	multivariate linear regression	6 years	-4.32 (-6.68, -1.81) IQ points lower in hX group <sup>e</sup>	0.001	mean 13.5, range 5.1– 17.9

Study	Population	Definition of hX for result (pmol/L unless otherwise noted)	Definition of Reference Group	Statistical method	Age of Neurodevelopmental Assessment	Difference in Score When Comparing hX Population to Reference (95% CI)	<i>p</i> -value	Gestational Week of Thyroid Hormone Measurement
Grau et al. 2015	289 Spanish mother-child pairs	fT4 < 10 <sup>th</sup> percentile (< 13.7) (first trimester) (n = 42)	Urinary iodide concentration >150 g/L, fT4 > 10 <sup>th</sup> percentile and TPO Ab negative (n = 39)	Multivariate ANOVA	1 year	2.7 (CI: NR) IQ points higher in the hX mothers <sup>f</sup>	0.419	end of the 1st trimester
Hales et al. 2018	331 UK mother- child pairs	fT4 < 2.5 <sup>th</sup> percentile <sup>9</sup> (n = 81)	fT4 > 2.5 <sup>th</sup> percentile (n = 232)	Univariate and multivariate logistic regression	7.00–10.92 years	No difference between mean IQ scores of hX children, and those born to mothers with normal thyroid hormone functionh	0.875	< 16 weeks
Korevaar et al. 2016 <sup>i</sup>	3,839 Dutch mother-child pairs	≤ 10 <sup>th</sup> percentile (n = 386)	fT4 between 10 <sup>th</sup> and 90 <sup>th</sup> percentile (n = 3,288) <sup>j</sup>	Multivariate ANOVA	Median 6 years (range 5.6-7.9 years)	-1.8 (0, -3.2) point loss in low fT4 group compared to control group <sup>k</sup>	0.05	9-18

NR = not reported; hX = hypothyroxinemia; GW = gestational weeks.

<sup>&</sup>lt;sup>a</sup> Model adjusted for maternal age, parity, socioeconomic-cultural status, length of breastfeeding, use of alcohol and tobacco during pregnancy, children's sex, gestational age, type of delivery, Apgar score, birthweight, body length, and head circumference.

<sup>&</sup>lt;sup>b</sup> n for each group is estimated based on the knowledge that analysis is for individuals with fT4 less than 10<sup>th</sup> percentile and there are 1,642 people in the regression analysis.

c Results adjusted for region, sex, psychologist administering IQ test, gestational age and weight at birth, circulating vitamin D during pregnancy and year –season of sampling, urine cotinine levels during pregnancy, TSH, maternal education, parental social class, quality of the test, and child neurologic pathology.

<sup>&</sup>lt;sup>d</sup> Model adjusted for maternal depression, psychosocial factors, tobacco and alcohol use during pregnancy, and demographic variables.

<sup>&</sup>lt;sup>e</sup> Model adjusted for ethnicity, birth order, maternal age, BMI, marital status, maternal history of smoking, educational levels, maternal psychopathology during pregnancy, household income, and gestational age at time of thyroid sampling.

<sup>&</sup>lt;sup>f</sup> Results from first trimester, total mean WISC, Table 4 of Grau et al. (2015).

<sup>&</sup>lt;sup>g</sup> The value of the hypothyroxinemic cut point was not identified in the paper.

	Definition of						Gestational
	hX for result				Difference in Score		Week of
	(pmol/L unless otherwise	Definition of	Statistical	Age of Neurodevelopmental	When Comparing hX Population to		Thyroid Hormone
Study Population	noted)	Reference Group	method	Assessment	Reference (95% CI)	p-value	Measurement

h Model adjusted for child sex, age of mothers at birth of offspring, child breastfeeding, schooling (Welsh- or English-medium school attended), place of assessment (home or research center), and socioeconomic status. The overall difference between groups is not reported in Supplementary Table 2 of Hales et al. (2018).

Data are taken from Supplemental Figure 2 of Korevaar et al. (2016).

Reference group is calculated as the total population without the mothers less than or equal to the 10<sup>th</sup> percentile, and after exclusion of women with *in vitro* fertilization treatment (N = 76) or women with known thyroid disorders or thyroid-interfering medication usage (N = 89). Data are taken from Supplemental Figure 2.

<sup>&</sup>lt;sup>k</sup> Model adjusted for gestational age, maternal age, smoking, BMI, parity, education level, ethnic origin, fetal sex, and birthweight.

# 5.5 Summary of Literature Review Findings to Connect Alterations in Thyroid Hormone Levels to Alternations in Neurodevelopment

In this literature review, 16 papers were identified as potentially having data or presenting analysis that may inform the quantitative understanding of the incremental changes in thyroid hormone levels and subsequent adverse neurodevelopmental outcomes. Of these 16 papers, 5 were deemed useful for further quantitative analysis (see [ REF \_Ref491174324 \h ]). Taken as a whole, the majority of the Group 1 studies imply that neurodevelopmental impacts due to altered maternal fT4 levels or presence of maternal hypothyroxinemia occur. In addition, the Group 2 studies (along with several Group 1 studies) also demonstrate adverse impact of maternal hypothyroxinemia on neurodevelopmental outcomes of the offspring. Overall, the results of this literature review lend support to the concept that maternal fT4, especially in the hypothyroxinemic range, is critical to the offspring's proper neurodevelopment. Across different age ranges and neurodevelopment indices, the impact of altered fT4 is seen even with small incremental changes in fT4 (and in populations with fT4 across the "normal" range). Further information on Group 1 and Group 2 studies that examined associations of thyroid hormone/hypothyroxinemia is detailed in Appendix I and is organized by neurodevelopmental endpoint.

In subsequent sections, two approaches are presented to link perchlorate doses to alterations in neurodevelopment, via alterations in thyroid hormone levels. Section [ REF \_Ref456208917 \r \h ] presents the approach to connect changes in fT4 with changes in neurodevelopment as estimated by the five Group 1 studies identified as including data that could be used to quantitatively describe the relationship between thyroid hormone levels in early pregnancy and changes in neurodevelopment in offspring. Section [ REF \_Ref481160968 \r \h ] presents an alternative option using the BBDR model to evaluate a shift in the proportion of the population that will fall below a hypothyroxinemic cut point, given exposure to perchlorate.

# 6. Informing the Derivation of an MCLG: Neurodevelopmental Effects

Based on the studies identified for additional quantitative analysis in Section [ REF\_Ref488161860 \n \h ], this section presents potential approaches to connect perchlorate exposure and alterations in fT4 levels in pregnant women as predicted by the BBDR model to potential adverse neurodevelopmental impacts in the offspring. This section demonstrates how the EPA could inform the derivation of an MCLG for perchlorate by evaluating how various doses of perchlorate to a pregnant woman may result in changes in neurodevelopment of her child.

The BBDR model is capable of predicting fT4 levels at various gestational weeks of pregnancy for various iodine intake levels. For the purposes of this report, one gestational week was selected from each study of interest for analysis based on when fT4 values were collected. Given that none of the studies identified for additional quantitative analysis are specific to an iodine intake level, analyses were conducted to evaluate the impact of perchlorate on the potentially sensitive population of individuals with low iodine intake. That is, the analyses were conducted for individuals with an assumed iodine intake of 75  $\mu$ g/day with pTSH = 0.398, as discussed in Section [ REF \_Ref482972870 \r \h ]. Further, to evaluate the impact of fT4 at various percentiles, the EPA used the derived fT4 distributions (as described in Section [ REF \_Ref482787773 \r \h ]) and calculated the resulting changes in the outcome of interest based on effect measures from the study of interest. Central, lower, and upper-bound effect estimates from the 95% CIs were used in all analyses.

An overview of the approach is presented in [ REF \_Ref482890788 \h ]. While [ REF \_Ref482890788 \h ] is specific for estimating neurodevelopmental effects for the low-iodine intake population, the same process could be followed for any iodine intake level that is output from the BBDR model. Analyses based on the results from each study are presented in alphabetical order.

# Figure [ SEQ Figure \\* ARABIC ]. Outline of Approach to Assess the Effect of Doses of Perchlorate on Potential Neurodevelopmental Outcomes

[EMBED Visio.Drawing.15]

#### 6.1 Analysis Using Endendijk et al. (2017)

As described in Section [ REF \_Ref512610865 \r \h ], Endendijk et al. (2017) found an association between maternal fT4 in the first trimester and anxiety/depression. The authors reported an effect estimate of 0.12 (SE = 0.005) when controlling for maternal pre-natal psychopathology symptoms and TPO Ab in a linear regression analysis. Through personal communication with the author, the EPA determined that the anxiety/depression scores were inverse-transformed<sup>23</sup> in the model, and subsequently the relationship between fT4 and anxiety/depression can be described as:

$$\Delta AD = \left(\frac{1}{\beta * fT4_2}\right) - \left(\frac{1}{\beta * fT4_1}\right)$$

Where,

 $\Delta AD$  = the change in the anxiety/depression score

 $\beta$  = 0.12 (95% CI: 0.11, 0.13)

 $fT4_1$  = fT4 concentration with no perchlorate dose

 $fT4_2$ = fT4 with increased dose of perchlorate

This function was used to evaluate how incremental changes in first-trimester fT4 output by the BBDR model impact an individual's anxiety/depression score as measured by the CBCL. To evaluate the impact of changes in fT4 levels due to perchlorate on anxiety/depression using the function estimated in Endendijk et al. (2017), GW 12 fT4 output from the BBDR model was used (see Section [ REF \_Ref482972870 \r \h ]). Estimated fT4 levels at various percentiles were also used (see Section [ REF \_Ref485386250 \r \h ]). GW 12 was selected because this was the week during the first trimester that fT4 measurements were taken from pregnant women in the Endendijk et al. (2017) study.

Results of an analysis demonstrating changes in anxiety/depression score as a result of increasing perchlorate doses are presented in [ REF  $\_$ Ref512612375  $\h$  ].

The EPA has made an assumption that by inverse-transform, the authors meant that they used the reciprocal of each anxiety/depression score as the dependent variable in their models. The Agency has reached out to the study authors to confirm this assumption.

6-3

Table [ SEQ Table \\* ARABIC ]. Estimated Change in Anxiety/Depression Score Based on Endendijk et al. (2017) Central, Lower, and Upper Beta Estimates due to a Change in Perchlorate Dose from 0 μg/kg/day at Various Percentiles of fT4

			fT	4 Percentile		
		2.5 <sup>th</sup>		10 <sup>th</sup>		50 <sup>th</sup>
Dose of		ΔAD		ΔAD		ΔAD
Perchlorate (μg/kg/day)	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>
			100	100		
0	6.72	N/A	8.09	N/A	10.67	N/A
0	5.58	N/A	6.71	N/A	8.85	N/A
1	5.51	0.02 (0.02, 0.02)	6.64	0.01 (0.01, 0.01)	8.78	0.01 (0.01, 0.01)
2	5.44	0.04 (0.04, 0.04)	6.57	0.03 (0.02, 0.03)	8.71	0.01 (0.01, 0.02)
3	5.38	0.06 (0.05, 0.06)	6.51	0.04 (0.04, 0.04)	8.65	0.02 (0.02, 0.02)
4	5.32	0.07 (0.07, 0.08)	6.45	0.05 (0.05, 0.06)	8.59	0.03 (0.03, 0.03)
5	5.26	0.09 (0.08, 0.10)	6.39	0.06 (0.06, 0.07)	8.53	0.04 (0.03, 0.04)
10	5.00	0.17 (0.16, 0.19)	6.13	0.12 (0.11, 0.13)	8.27	0.07 (0.06, 0.07)

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

### 6.2 Analysis Using Finken et al. (2013)

As described in Section [ REF \_Ref485383673 \r \h \\* MERGEFORMAT ], Finken et al. (2013) found an association between decreased fT4 levels and increased standard deviation of reaction time. This association, according to Finken et al. (2013) "...has been found to be related more strongly to individual difference in intelligence, cognitive aging, and neurological disorders" (Finken et al., 2013, p. 1419). This relationship was described in [ REF \_Ref457841191 \h \\* MERGEFORMAT ], from Finken et al.'s (2013) multivariate regression analysis, which found that for every pmol/L decrease in fT4 there is a subsequent increase in the SD of reaction time with  $\beta$  = -4.9 (-9.5; -0.2), holding all else equal. This function was derived from and is thus applicable to the entire range of fT4 values in Finken et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum>6Displa yText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></author></contributors><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6-year-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></periodical><full-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></periodical><pages>1417-

1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year>2013</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2012-3389</electronic-resource-num></re></ri>It is unclear exactly what the entire range

encompasses, but the 2.5<sup>th</sup> percentile is reported as 6.5 pmol/L, and the interquartile range presented in the paper is 8.7 to 10.5 pmol/L.

Using the predicted changes in fT4 for increasing doses of perchlorate output by the BBDR model and estimating different distributional baselines, the following function can be used to estimate changes in the SD of reaction time as associated with changes in fT4:

$$\Delta$$
SD Reaction Time (ms) =  $\beta \times \Delta$  fT4

To evaluate the impact of changes in fT4 levels due to perchlorate on the SD of reaction time using

Where:

 $\beta$  = -4.9 (95% CI: -9.5, -0.2)

 $\Delta fT4 = fT4_{increasedClO4} - fT4_{baselineClO4}$ 

the function estimated in Finken et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum><Displa vText>(2013)</DisplayText><rccord><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></authors></contributors><titles><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6year-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title>>eriodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1417-1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year> 2013</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2012-3389</electronicresource-num></record></Cite></EndNote>], gestational week 13 fT4 output from the BBDR model was used (see Section [ REF Ref482972870 \r \h ]). Estimated fT4 levels at various percentiles were also used (see Section [ REF Ref485386250 \r \h ]). Gestational week 13 was selected because the median week of fT4 collection in the Finken et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum><Displa vText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></authors></contributors><titles><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6year-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1417-1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year>

2013</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2012-3389</electronic-resource-num></record></Cite></EndNote>] study was 12.9 (IQR: 11.9–14.3).

Results of an analysis demonstrating how increasing perchlorate dosage increases the SD of reaction time are presented in [ REF Ref485385829 \h ].

Table [ SEQ Table \\* ARABIC ]. Estimated Change in the Standard Deviation of Reaction Time (milliseconds) based on Finken et al. (2013) Central, Lower, and Upper Beta Estimates due to a Change in Perchlorate Dose from 0 μg/kg/day at Various Percentiles of fT4

			ſſ	4 Percentile		
		2.5 <sup>th</sup>		10 <sup>th</sup>		50 <sup>th</sup>
Dose of		ΔRTSD		ΔRTSD		ΔRTSD
Perchlorate (µg/kg/day)	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>2</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>2</sup>
		0.0	ne li take = 1			
0	6.70	N/A	8.07	N/A	10.64	N/A
		100	100	75 december		
0	5.57	N/A	6.70	N/A	8.84	N/A
1	5.50	0.34 (0.01, 0.66)	6.63	0.34 (0.01, 0.66)	8.77	0.34 (0.01, 0.66)
2	5.43	0.67 (0.03, 1.30)	6.56	0.67 (0.03, 1.30)	8.71	0.67 (0.03, 1.30)
3	5.37	0.98 (0.04, 1.90)	6.50	0.98 (0.04, 1.90)	8.64	0.98 (0.04, 1.90)
4	5.31	1.28 (0.05, 2.48)	6.44	1.28 (0.05, 2.48)	8.58	1.28 (0.05, 2.48)
5	5.25	1.56 (0.06, 3.03)	6.38	1.56 (0.06, 3.03)	8.52	1.56 (0.06, 3.03)
10	4.99	2.83 (0.12, 5.48)	6.12	2.83 (0.12, 5.48)	8.27	2.83 (0.12, 5.48)

RTSD = Standard deviation of reaction time.

# 6.3 Analysis Using Korevaar et al. (2016) and an EPA Analysis of Korevaar et al. (2016)

As summarized in Section [ REF \_Ref482388672 \r \h ], Korevaar et al. (2016) presented a spline model and a multivariate regression with a quadratic term ([ REF \_Ref479341381 \h ]) to explain the relationship between maternal fT4 and child IQ, both of which they found to have an inverted U-shaped association. Korevaar provided the EPA with complete study data, <sup>24</sup> and subsequently the EPA confirmed the authors' conclusion that the regression analysis with the quadratic term sufficiently explained the relationship between maternal fT4 and child IQ assessed with the spline model.

#### 6.3.1 Analysis with Quadratic Function from Korevaar et al. (2016)

Given the adequate fit of the quadratic regression model, the EPA used the quadratic equation coefficients put forth in [ REF \_Ref479341381 \h ] to determine the changes in child IQ for given changes in maternal fT4. The following function was used to determine the change in child IQ for a given change in maternal fT4 due to perchlorate:

$$\Delta IQ = (\beta_1 \times lnfT4_2 + \beta_2 \times ln(fT4_2)^2) - (\beta_1 \times lnfT4_1 + \beta_2 \times ln(fT4_1)^2)$$

Where:

Data were provided by Dr. Tim Korevaar on February 6, 2017.

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

 $\beta_1 = 33.81 \ (95\% \ CI: 9.8, 57.82)$ 

 $\beta_2 = -6.235 (95\% \text{ CI: } -10.567, -1.903)$ 

 $fT4_1 = fT4$  when perchlorate dose = 0  $\mu$ g/kg/day at a given fT4 percentile for the low-iodine intake population.

 $fT4_2 = fT4$  when perchlorate dose is at a defined dose, X, which is greater than 0  $\mu$ g/kg/day, for the low-iodine intake population.

To evaluate the impact of perchlorate induced changes in fT4 levels on IQ using the estimated function from Korevaar et al. [ ADDIN EN.CITE < EndNote > < Cite

ExcludeAuth="1"><Author>Korevaar</Author><Year>2016</Year><RecNum>313</RecNum><Di splayText>(2016)</DisplayText><record><rec-number>313</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmcdxe5vxfpkax2vzp0ftv29"

timestamp="1491832236">313</key><key app="ENWeb" db-id="">0</key></foreign-keys><reftype name="Journal Article">17</ref-type><contributors><author>><author>Korevaar, Tim I.

M.</author><author>Muetzel, Ryan</author><author>Medici, Marco</author><author>Chaker, Layal</author><author>thor>Jaddoe, Vincent W. V.</author><author>de Rijke, Yolanda

B.</author><author>Steegers, Eric A. P.</author><author>Visser, Theo J.</author><author>White, Tonya</author><author>Tiemeier, Henning</author><author>Peeters, Robin

P.</author></authors></contributors><titles><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabet

Endocrinology</secondary-title></title>><periodical><full-title>The Lancet Diabetes & Endocrinology</full-title></periodical><pages>35-

43</pages><volume>4</volume>1</number>dates><year>2016</year></dates><isbn>2 2138587</isbn><urls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-

7</electronic-resource-num></record></Cite></EndNote>], GW 13 fT4 output from the BBDR model was used (see Section [ REF \_Ref482972870 \r \h ]), along with the estimated fT4 levels at various percentiles (see Section [ REF \_Ref485386250 \r \h ]). Gestational week 13 was selected because the median week of fT4 collection in this study was 13.2 weeks (range 9–18 GW). The results of this analysis are presented in [ REF \_Ref490658761 \h ].

# Table [ SEQ Table \\* ARABIC ]. Estimated Change in Nonverbal IQ, based on Korevaar et al. (2016) Central, Lower, and Upper Quadratic-Beta Estimates due to a Change in Perchlorate Dose from 0 μg/kg/day at Various Percentiles of fT4

			fT4	Percentile		
		2.5 <sup>th</sup>		10 <sup>th</sup>		50 <sup>th</sup>
Dose of		ΔIQ		ΔIQ		ΔIQ
Perchlorate	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4	Central	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>
(µg/kg/day)	(pinont)	(Lower, Opper)	(pmol/L)	(Lower, Upper) <sup>a</sup>	(pilloirt)	(Lower, Opper)
0	6.70	N/A	8.07	N/A	10.64	N/A
		let di		As paren		
0	5.57	N/A	6.70	N/A	8.84	N/A
1	5.50	-0.16 (0.33, -0.65)	6.63	-0.11 (0.32, -0.53)	8.77	-0.05 (0.29, -0.39)
2	5.43	-0.31 (0.65, -1.27)	6.56	-0.21 (0.62, -1.04)	8.71	-0.10 (0.56, -0.77)
3	5.37	-0.46 (0.95, -1.88)	6.50	-0.31 (0.91, -1.53)	8.64	-0.15 (0.82, -1.13)
4	5.31	-0.61 (1.25, -2.46)	6.44	-0.41 (1.19, -2.01)	8.58	-0.20 (1.08, -1.48)
5	5.25	-0.75 (1.52, -3.03)	6.38	-0.51 (1.46, -2.47)	8.52	-0.25 (1.32, -1.82)
10	4.99	-1.43 (2.77, -5.63)	6.12	-0.96 (2.65, -4.57)	8.27	-0.48 (2.40, -3.35)

<sup>&</sup>lt;sup>1</sup> Result based on central, lower and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

#### 6.3.2 EPA Independent Analysis of Korevaar et al. (2016) Data

The EPA conducted an additional analysis of the Korevaar et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Korevaar</Author><Year>2016</Year><RecNum>313</RecNum><Di splayText>(2016)</DisplayText><rec-number>313</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfcrmedxe5vxfpkax2vzp0ftv29"

timestamp="1491832236">313</key><key app="ENWeb" db-id="">0</key></foreign-keys><reftype name="Journal Article">17</ref-type><contributors><author>><author>Korevaar, Tim I.

M.</author><author>Muetzel, Ryan</author><author>Medici, Marco</author><author>Chaker,

Layal</author><author>Iaddoe, Vincent W. V.</author><author>de Rijke, Yolanda

B.</author><author>Steegers, Eric A. P.</author><author>Visser, Theo J.</author><author>White, Tonya</author><author>Tiemeier, Henning</author><author>Peeters, Robin

P.</author></authors></contributors><titles><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabet

Endocrinology</secondary-title></title><periodical><full-title>The Lancet Diabetes & Endocrinology</full-title></periodical>pages>35-

43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>2 2138587</isbn><urls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-

7</electronic-resource-num></record></Cite></EndNote>] data to account for uncertainties related to the original Korevaar et al. (2016) analysis. The study authors agreed to share their original dataset, and the EPA was thus able to further refine the model relating maternal fT4 and child IQ.

The first step in this reanalysis was to develop a causal model for the effect of maternal fT4 and child IQ in order to identify the minimum set of confounders. The EPA accomplished this by examining the literature for documented causal relationships between each of the variables that were originally

included in the Korevaar et al. (2016) model. This process allowed for the identification of the minimum set of variables that adequately controlled for confounding of the relationship between maternal fT4 and child IQ.

#### **Identification of Proper Functional Form**

After identifying the minimum set of confounders, the EPA tested models using ln(fT4) as the independent variable, as well as using raw, untransformed fT4 as the independent variable of interest. Three different functional forms were tested: a cubic restricted spline as was used in the original Korevaar et al. (2016) manuscript, a quadratic equation as was also estimated in the original Korevaar et al. (2016) manuscript, and a linear spline model. The linear spline form would allow for the use of a spline function with a more straightforward estimation of betas over the range of the function than the cubic restricted spline. The EPA opted to use the same knots in the spline functions that the study authors used in the original manuscript for both the linear and the cubic restricted splines: the 10<sup>th</sup> percentile, the 50<sup>th</sup> percentile, and the 90<sup>th</sup> percentile. Overall, six models using the minimum set of confounders would be estimated, three using ln(fT4) as the independent variable, and three using raw fT4 as the independent variable. The cubic restricted spline models and the linear spline models were estimated using the "mkspline" command in Stata [ ADDIN EN.CITE

<EndNote><Cite><Author>StataCorp</Author><Year>2015</Year><RecNum>1984</RecNum><
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DisplayText>(StataCorp, 2015)</DisplayText><record><rec-number>1984</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1524593247">1984</key></foreign-keys><ref-type name="Computer Program">9</ref-

type><contributors><author>>StataCorp</author></contributors><titles><title>S tata Statistical Software: Release 14</title></title></dates><year>2015</year></dates><publication>College Station, TX</pub-location><publisher>StataCorp LP</publisher><urls></urls></record></EndNote>]. The "mkspline" syntax used for each model is reported in Appendix J.

#### Additional Adjustments to the Analysis Sample

It was determined that there were many values of the independent variable of interest, fT4, which were imputed using multiple imputation by the study authors. The EPA opted to drop all data points that did not have a measured value for fT4 in the original data to avoid impacting the estimation of the effect estimate of the independent variable of interest with data that were not directly measured. However, the EPA continued to use the multiply imputed dataset for all other variables. The EPA also decided *a priori* to drop a single data point with an fT4 value above 90 pmol/L, as it was suspected that this value had a large influence on the estimation of the EPA's models, and may have been an invalid data point.

#### **Choosing a Preferred Model**

The causal model building process identified two minimum sets of confounders that would effectively isolate the total effect of maternal fT4 on child IQ. The first set includes:

- Continuous maternal age in years;
- Categorical maternal smoking status (non-smoker, previous smoker, or current smoker); and

• Categorical maternal ethnicity (Dutch, Moroccan, Turkish, Surinamese, Cape Verdian, Dutch Antilles, Indonesian, Asian, "other European, North American, or Australian," or "other Asian, African, or South American").

The second set of confounding variables includes:

- Categorical maternal education (none or primary only, secondary phase (3-4 years), secondary phase (4-5 years), higher phase 1 (6-8 years), or higher phase 2 (>8 years));
- Categorical maternal smoking status (previously defined); and
- Categorical maternal ethnicity (previously defined.)

The EPA moved forward with models utilizing both of these minimum sets of confounders, and opted to use model fit statistics to eventually choose the most appropriate set.

Since there were two minimum sets of confounders identified by the EPA's causal model, 12 models were estimated in total (six for each minimum set of confounders). Each model was estimated as previously described, and beta estimates were compared for effect plausibility, as were overall model fit estimates using adjusted  $R^2$  statistics and F-statistics. Since the data were multiply imputed, adjusted  $R^2$  statistics were calculated as the mean adjusted  $R^2$  statistic when estimating the model on each multiply imputed dataset separately. This was done using the "mibeta" command in Stata [ ADDIN EN.CITE

<EndNote><Cite><Author>StataCorp</Author><Year>2015</Year><RecNum>1984</RecNum>< DisplayText>(StataCorp, 2015)</DisplayText><record><rec-number>1984</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1524593247">1984</key></foreign-keys><ref-type name="Computer Program">9</ref-

type > < contributors > < authors > < author > < / authors > < / contributors > < title > < title > < title > < title > < / contributors > < / contributors > < title > < / contributors > < title > < / contributors > < / contributors > < title > < / contributors > < / contributors > < / contributors > < title > < / contributors > < / contribu

LP</publisher><urls></urls></record></Cite></EndNote>]. Adjusted  $R^2$  statistics and F-statistics for each model are presented in [ REF Ref512353025 \h ].

### Table [ SEQ Table $\$ ARABIC ]. Adjusted $R^2$ Statistics and F-Statistics for Each Estimated Model

Model Specification  Sontrolling for age, smoking, attnibity	Adjusted R <sup>2</sup>	<i>F</i> -Statistic (Model DF, Residual DF)
Linear spline; untransformed fT4	0.102	25.98 (16, 3512.2)
Linear spline; natural log fT4	0.102	26.06 (16, 3513.1)
Cubic restricted spline; untransformed fT4	0.101	29.37 (14, 3473.8)

Model Specification	Adjusted R <sup>2</sup>	F-Statistic (Model DF, Residual DF)
Cubic restricted spline; natural log fT4	0.102	29.49 (14, 3472.4)
Quadratic; untransformed fT4	0.099	28.65 (14, 3480.8)
Quadratic; natural log fT4	0.101	29.23 (14, 3478.7)
Controlling for education, smoking, ethnicity		
Linear spline; untransformed fT4	0.132	28.53 (19, 3285.2)
Linear spline; natural log fT4	0.132	28.60 (19, 3285.2)
Cubic restricted spline; untransformed fT4	0.131	31.49 (17, 3188.1)
Cubic restricted spline; natural log fT4	0.132	31.64 (17, 3186.0)
Quadratic; untransformed fT4	0.129	31.05 (17, 3201.5)
Quadratic; natural log fT4	0.131	31.42 (17, 3190.5)

By comparing adjusted  $R^2$  statistics, the minimum set of confounders was determined for which the set that included mother's education, smoking status, and mother's ethnicity had a superior fit than with the other minimum set, regardless of the functional form utilized for the model. The EPA moved forward using this minimum set of confounders only. The Agency also determined that the linear spline function had the highest adjusted  $R^2$  value among the different functional forms. This form was preferred *a priori* as it was more flexible than the quadratic equation, and also easier to estimate betas from than the cubic restricted spline for use in a concentration-response function. Although the cubic restricted spline models had a slightly higher F-statistic than the linear spline models, the EPA chose to move forward with the linear spline models, as the improvement in fit was not large enough to warrant the much greater complexity of using the cubic restricted spline models for predictive modeling. The choice between the use of  $\ln(fT4)$  and untransformed fT4 was difficult, as each function had a very similar adjusted  $R^2$  value, although the function utilizing  $\ln(fT4)$  had a marginally higher adjusted  $R^2$ . The EPA chose to assess the model assumptions for both  $\ln(fT4)$  and untransformed fT4 using the linear spline function that controlled for education, smoking status, and mother's ethnicity.

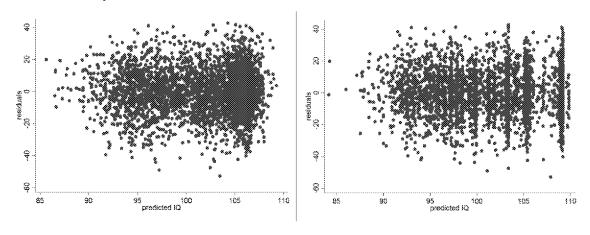
The EPA also tested models that included an interaction term between gestational age at blood draw and ln(fT4) (or fT4), given the natural variability of fT4 throughout the course of pregnancy, and thus the changing definition of low or high fT4 by gestational week. The fit of these models was similar to those models without the interaction term presented above, but the F-statistics of the models were

substantially lower due to the increased model degrees of freedom from the increased number of variables included in the model. The results of this analysis are presented in Appendix J.

#### **Post Hoc Model Assumption Tests**

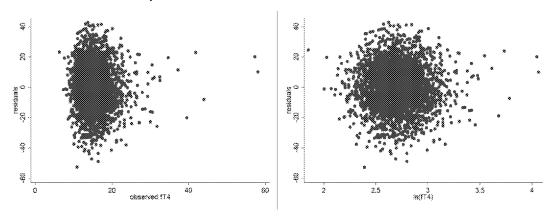
After estimating these two linear spline functions, residuals were examined for patterns that would violate the assumptions of the regression models. Neither model displayed convincing evidence of a relationship between residuals and model predicted values; however, there was some evidence of heteroscedasticity between residuals and the EPA's independent variable of interest. Scatterplots examining these potential assumption violations are presented in [ REF \_Ref512353059 \h ] and [ REF \_Ref512353068 \h ].

Figure [ SEQ Figure \\* ARABIC ]. Plots of Model Residuals versus Predicted IQ for Both Linear Spline Models



Note: Model using untransformed fT4 on left, model using natural log fT4 on right.

Figure [ SEQ Figure \\* ARABIC ]. Plots of Model Residuals versus fT4 and Natural Log fT4 for Both Linear Spline Models



Note: Model using untransformed fT4 on left, model using natural log fT4 on right.

To test for heteroscedasticity in its models, the EPA ran the Breusch-Pagan and the Cook-Weisberg tests [ ADDIN EN.CITE

<EndNote><Cite><Author>Breusch</Author><Year>1979</Year><RecNum>1981</RecNum><Di

```
splayText>(Breusch & Dagan, 1979; Cook & Dagan, Weisberg,
1983)</br>

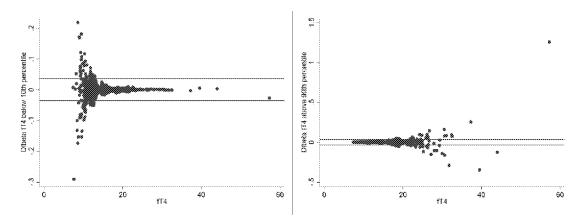
1981
/rec-number>
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keys><ref-type name="Journal Article">17</ref-type><contributors><author>Breusch, T.
S.</author><author>Pagan, A. R.</author></authors></contributors><titles><title>A simple test for
heteroscedasticity and random coefficient variation</title><secondary-
title>Econometrica</secondary-title></title>>eriodical><full-title>Econometrica</full-
title></periodical><pages>1287</pages><volume>47</volume><number>5</number><dates><year
>1979</year></dates><isbn>00129682</isbn><urls><related-urls><url><style face="underline"
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size="100%">https://www.jstor.org/stable/1911963?origin=crossrcf</style></url></related-
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num><language>en</language><access-date>2018-04-13 15:39:41</access-
date></record></Cite><Cite><Author>Cook</Author><Year>1983</Year><RecNum>1982</RecN
um><record><rec-number>1982</rec-number><foreign-keys><key app="EN" db-
id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1524592933">1982</key></foreign-
keys><ref-type name="Journal Article">17</ref-type><contributors><author>>cauthor>Cook, R.
Dennis</author><author>Weisberg,
Sanford</author></authors></contributors><title>Diagnostics for heteroscedasticity in
regression</title><secondary-title>Biometrika</secondary-title></title><periodical><full-
title>Biometrika</full-title></periodical><pages>1-
10</pages><volume>70</volume><number>1</number><dates><year>1983</year></dates><isbn>
0006-3444</isbn><urls></urls><electronic-resource-num>10.1093/biomet/70.1.1</electronic-
resource-num><language>English (US)</language></record></Cite></EndNote>], as well as
White's general test for heteroscedasticity [ ADDIN EN.CITE
<EndNote><Cite><Author>White</Author><Year>1980</Year><RecNum>1980</RecNum><Disp
layText>(White, 1980)</DisplayText><record><rec-number>1980</rec-number><foreign-
keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"
timestamp="1524592668">1980</key></foreign-keys><ref-type name="Journal Article">17</ref-
type><contributors><author>White,
Halbert</author></authors></contributors><titles><title>A heteroskedasticity-consistent covariance
matrix estimator and a direct test for heteroskedasticity</title><secondary-
title>Econometrica</secondary-title></titles><periodical><full-title>Econometrica</full-
title></periodical><pages>817-
838</pages><volume>48</volume><number>4</number><dates><year>1980</year></dates><isbn
>00129682, 14680262</isbn><urls><related-urls><url><style face="underline" font="default"
size="100%">http://www.jstor.org/stable/1912934</style></url></related-
urls></urls></record></Cite></EndNote>]. None of these tests suggested evidence of
heteroscedasticity, and thus it was concluded that the models conformed to the regression assumption
of homoscedastic residuals.
```

The EPA also tested the estimated models for influential observations using the DFBETA technique to see how much influence each data point had on effect estimates. DFBETA essentially removes each point one at a time from the regression analysis and measures the effect removing the point has on each beta estimate. A "rule of thumb" is that a point that affects betas by more than 2/sqrt(n) is worth investigating further. DFBETA plots for the estimated beta for fT4 values below the 10<sup>th</sup>

percentile, as well as for the estimated beta for fT4 values above the  $90^{th}$  percentile, are shown in [ REF Ref512353091 \h ].

Figure [ SEQ Figure \\* ARABIC ]. Plots of DFBETAs versus fT4 for Estimated Beta Representing fT4 Values below the 10<sup>th</sup> Percentile and Estimated Beta Representing fT4 Values above the 90<sup>th</sup> Percentile



Note: Beta for fT4 values under 10<sup>th</sup> percentile on left, beta for fT4 values above 90<sup>th</sup> percentile on right. Red lines represent +/- 2/sqrt(n) for this model, or +/- 0.36.

From these analyses, the EPA opted to drop one data point because of its strong influence on regression betas ([ REF \_Ref512353091 \h ] and [ REF \_Ref512353166 \h ]). This data point had a very high fT4 value, coupled with a high IQ value. The data point is represented by the outlier on the right scatter plot in [ REF \_Ref512353091 \h ]. The IQ and fT4 values of the omitted data point are shown in [ REF \_Ref512353166 \h ].

Table [ SEQ Table \\* ARABIC ]. Data Point Dropped from Analysis due to Large Influence on Results

Data point	fT4 (pmol/L)	IQ	
1	57.318	119	

Model results including this influential data point are included in Appendix J.

## **Final Model Results**

Once this data point was removed, the two linear spline models were re-estimated. The betas, adjusted  $R^2$  statistics, and F-statistics are reported in [REF\_Ref512353213 \h].

Model Specification	Child IQ β (95% CI)	P-Value	Adjusted R <sup>2</sup>	F-Statistic (Model DF, Residual DF)	
Untransformed IT4					
Beta below 10 <sup>th</sup> percentile fT4	1.73 (0.44, 3.02)*	0.009		28.58 (19, 3281.1)	
Beta between 10th and 50th percentile fT4	-0.02 (-0.60, -0.56)	0.944	0.422		
Beta between 50th and 90th percentile fT4	-0.09 (-0.50, 0.32)	0.658	0.132		
Beta above 90th percentile fT4	-0.37 (-0.70, -0.03)*	0.033			
Log-transformed fl 4					
Beta below 10 <sup>th</sup> percentile ln(fT4)	17.26 (3.77, 30.75)*	0.012			
Beta between 10th and 50th percentile In(fT4)	-0.16 (-7.91, 7.59)	0.968	0.132	28.64	
Beta between 50th and 90th percentile In(fT4)	-0.61 (-7.57, 6.35)	0.863	0.132	(19, 3282.2)	
Beta above 90th percentile In(fT4)	-10.98 (-19.69, -2.26)*	0.014			

Note: 10<sup>th</sup> percentile fT4 = 11.77 pmol/L. 50<sup>th</sup> percentile fT4 = 14.76 pmol/L. 90<sup>th</sup> percentile fT4 = 18.94 pmol/L. 10<sup>th</sup> percentile log-transformed fT4 = 2.47. 50<sup>th</sup> percentile log-transformed fT4 = 2.69. 90<sup>th</sup> percentile log-transformed fT4 = 2.94. \* signifies beta estimates statistically significant at the 95% confidence level.

Given the slightly better fit of the log-transformed fT4 model, the EPA designated this model as the preferred model. Using the predicted changes in fT4 for increasing doses of perchlorate from the BBDR model, the following function can be used to estimate changes in IQ as a result of changes in fT4:

When fT4<sub>1</sub> and fT4<sub>2</sub> are both less than or equal to 11.76<sup>25</sup> pmol/L:

$$\Delta IQ = (\beta_1 \times \ln(fT4_2)) - (\beta_1 \times \ln(fT4_1))$$

Where:

 $\beta_1 = 17.26 (3.77, 30.75)$ 

 $fT4_1 = fT4$  when perchlorate dose = 0 µg/kg/day at a given fT4 percentile

 $fT4_2$  = fT4 when perchlorate dose is at a defined dose, X, which is greater than 0 µg/kg/day

Note that when the fT4 values are greater than 11.76 the function and beta effect estimates will vary. However, given that the BBDR model results estimate the median fT4 to be 10.7 pmol/L with adequate iodine (i.e., iodine intake =  $170 \mu g/day$ ), using different effect estimates is unlikely.

Results of an analysis demonstrating how increasing perchlorate dosage decreases IQ based on the EPA's analysis of the Korevaar et al. (2016) data are presented in [ REF \_Ref485387377 \h ] using the beta estimates reported in [ REF \_Ref512353213 \h ].

This and the other fT4 values presented in this key for the above function represent untransformed values of the ln(fT4) values at each knot of the spline. The values were obtained by calculating exp(ln(fT4)).

Table [ SEQ Table \\* ARABIC ]. Estimated Change in IQ, Based on Independent Analysis of Korevaar et al. (2016) Central, Lower, and Upper Multivariate-Beta Estimates due to a Change in Perchlorate Dose from 0  $\mu$ g/kg/day at Various Percentiles of fT4

			fT4	Percentile		
	2.5 <sup>th</sup>		10 <sup>th</sup>		50 <sup>th</sup>	
Dose of		ΔIQ		ΔΙQ		ΔIQ
Perchlorate (μg/kg/day)	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>
		10.00		Thursday		
0	6.70	N/A	8.07	N/A	10.64	N/A
		0.01	ne intake = 1	Sugary		
0	5.57	N/A	6.70	N/A	8.84	N/A
1	5.50	-0.22		-0.18	8.77	-0.14
<u>'</u>	3.30	(-0.05, -0.39)	6.63	(-0.04, -0.32)		(-0.03, -0.24)
2	2	-0.43		-0.36	8.71	-0.27
	5.43	(-0.09, -0.76)	6.56	(-0.08, -0.63)		(-0.06, -0.48)
3		-0.63		-0.52	8.64	-0.39
<u> </u>	5.37	(-0.14, -1.12)	6.50	(-0.11, -0.93)	0.04	(-0.09, -0.70)
4		-0.83		-0.68	8.58	-0.52
<b>T</b>	5.31	(-0.18, -1.47)	6.44	(-0.15, -1.22)	0.00	(-0.11, -0.92)
5		-1.02		-0.84	8.52	-0.63
J	5.25	(-0.22, -1.81)	6.38	(-0.18, -1.50)	0.02	(-0.14, -1.13)
10		-1.89		-1.55	8.27	-1.16
10	4.99	(-0.41, -3.36)	6.12	(-0.34, -2.77)	0.27	(-0.25, -2.07)

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

# 6.4 Analysis using Pop et al. (1999, 2003)

As summarized in Section [ REF \_Ref482388672 \r \h ], two Pop et al. papers found a relationship between fT4 and neurodevelopmental outcomes measured using the BSID. Specifically, Pop et al. (1999, 2003) found a relationship between changes in fT4 and PDI. Pop et al. (2003) also found a relationship between fT4 and MDI. Using the WebPlotDigitizer Extension for Google Chrome and using the linear regression function in Excel (2013), effect estimates were estimated based on a bivariate linear regression analysis. The effect estimates and their associated 95% CIs are summarized in [ REF \_Ref485388464 \h ].

Table [ SEQ Table \\* ARABIC ]. Pop et al. (1999, 2003) Effect Estimates<sup>a</sup>

Study/Endpoint	β (95% CI)					
Pop et al. (1999)/PDI	8.5 (95% CI: 0.01, 17.0)					
Pop et al. (2003)/PDI	8.4 (95% CI: 4.0, 12.8)					
Pop et al. (2003)/MDI	6.3 (95% CI: 1.9, 10.6)					
<sup>a</sup> All estimates are based on bivariate regression analysis of digitized data.						

These Pop et al. (1999, 2003) functions are specific to the individuals in their sample who were hypothyroxinemic. In Pop et al. (1999) the  $10^{th}$  percentile hypothyroxinemic cut point corresponded to an fT4 level of 10.4 pmol/L, and in Pop et al. (2003) this corresponded to an fT4 level of 12.4 pmol/L. The 10.4 pmol/L level is between the  $43^{rd}$  and  $44^{th}$  percentile of the median iodine intake and between the  $87^{th}$  and the  $88^{th}$  percentile of the 75  $\mu$ g/day iodine intake distributions predicted by the BBDR model and estimated distribution as explained in Section [ REF \_Ref482274870 \r \h ]. The lowest fT4 level evaluated in Pop et al. (1999) was 7.86 pmol/L<sup>26</sup>. The 12.4 pmol/L level is equivalent to between the  $84^{th}$  and  $85^{th}$  percentile of the median iodine intake and between the  $99^{th}$  and  $99.5^{th}$  percentile of the 75  $\mu$ g/day iodine intake distributions predicted by the BBDR model and estimated distribution as explained in Section [ REF \_Ref482274870 \r \h ]. The lowest fT4 level evaluated in Pop et al. (2003) was 8.39 pmol/L.[ NOTEREF \_Ref488142491 \h \\*

To evaluate the impact of the changes in fT4 levels due to perchlorate on MDI and PDI using the Pop et al. (1999, 2003) studies, the EPA used GW 12 fT4 output from the BBDR model (see Section [ REF \_Ref482972870 \r \h ]), along with the estimated fT4 levels at various percentiles (see Section [ REF \_Ref485386250 \r \h ]). This is because Pop et al. (1999, 2003) measured fT4 levels in their study populations at GW 12. The subsequent subsections outline the specific analysis and results for each study/endpoint combination from the Pop et al. (1999, 2003) analyses.

## 6.4.1 Analysis for Pop et al. (1999) - PDI

To evaluate the impact that a change in fT4 as a result of an increased dose of perchlorate will have on resulting PDI scores based on Pop et al. (1999), the following equation was utilized:

$$\Delta PDI = \beta \times \Delta fT4 (fT4 \text{ in pmol/L})$$

Where:

 $\beta = 8.5 (95\% \text{ CI: } 0.01, 17.0)$ 

 $\Delta fT4 = fT4_{increasedClO4} - fT4_{baselineClO4}$ 

Results of an analysis demonstrating how increasing perchlorate dosage decreases PDI based on Pop et al. (1999) are presented in [ REF \_Ref485389086 \h ].

<sup>6</sup> This value was extracted from the digitized datasets from the Pop et al. (1999, 2003) analyses.

6-17

Table [ SEQ Table \\* ARABIC ]. Estimated Change in PDI, Based on Pop et al. (1999) Central, Lower, and Upper Beta Estimates due to a Change in Perchlorate Dose from 0  $\mu$ g/kg/day at Various Percentiles of fT4

			fT4	Percentile		
	2.5 <sup>th</sup>			10 <sup>th</sup>		50 <sup>th</sup>
		ΔPDI		ΔPDI		ΔPDI
Dose of Perchlorate (µg/kg/day)	fT4 (pmol/L)	Central (Lower, Upper)ª	fT4 (pmol/L)	Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Central (Lower, Upper)ª
0	6.72	N/A	8.09	N/A	10.67	N/A
				egieri)		
0	5.58	N/A	6.71	N/A	8.85	N/A
1	5.51	-0.60 (0.00, -1.21)	6.64	-0.60 (0.00, -1.21)	8.78	-0.60 (0.00, -1.21)
2	5.44	-1.17 (0.00, -2.35)	6.57	-1.17 (0.00, -2.35)	8.71	-1.17 (0.00, -2.35)
3	5.38	-1.72 (0.00, -3.44)	6.51	-1.72 (0.00, -3.44)	8.65	-1.72 (0.00, -3.44)
4	5.32	-2.24 (0.00, -4.48)	6.45	-2.24 (0.00, -4.48)	8.59	-2.24 (0.00, -4.48)
5	5.26	-2.74 (0.00, -5.47)	6.39	-2.74 (0.00, -5.47)	8.53	-2.74 (0.00, -5.47)
10	5.00	-4.94 (-0.01, -9.88)	6.13	-4.94 (-0.01, -9.88)	8.27	-4.94 (-0.01, -9.88)

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

# 6.4.2 Analysis for Pop et al. (2003) - PDI

To evaluate the impact that a change in fT4 as a result of an increased dose of perchlorate will have on resulting PDI scores based on Pop et al. (2003), the following equation was utilized:

$$\Delta PDI = \beta \times \Delta fT4 (fT4 \text{ in pmol/L})$$

Where:

$$\beta = 8.4 (95\% \text{ CI: } 4.0, 12.8)$$

$$\Delta fT4 = fT4_{increasedClO4} - fT4_{baselineClO4}$$

Results of an analysis demonstrating how increasing perchlorate dosage decreases PDI based on Pop et al. (2003) are presented in [ REF Ref485389400 \h ].

Table [ SEQ Table \\* ARABIC ]. Estimated Change in PDI, Based on Pop et al. (2003) Central, Lower, and Upper Beta Estimates due to a Change in Perchlorate Dose from  $0 \mu g/kg/day$  at Various Percentiles of fT4

	fT4 Percentile							
	2.5 <sup>th</sup>		10 <sup>th</sup>		50 <sup>th</sup>			
Dose of Perchlorate (µg/kg/day)	fT4 (pmol/L)	Δ PDI Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Δ PDI Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Δ PDI Central (Lower, Upper) <sup>a</sup>		
		1	ine intake =	76 (1976) 		1		
0	6.72	N/A	8.09	N/A	10.67	N/A		
		6		(Constitution				
0	5.58	N/A	6.71	N/A	8.85	N/A		
1	5.51	-0.60 (-0.28, -0.91)	6.64	-0.60 (-0.28, -0.91)	8.78	-0.60 (-0.28, -0.91)		
2	5.44	-1.16 (-0.55, -1.77)	6.57	-1.16 (-0.55, -1.77)	8.71	-1.16 (-0.55, -1.77)		
3	5.38	-1.70 (-0.81, -2.59)	6.51	-1.70 (-0.81, -2.59)	8.65	-1.70 (-0.81, -2.59)		
4	5.32	-2.21 (-1.05, -3.37)	6.45	-2.21 (-1.05, -3.37)	8.59	-2.21 (-1.05, -3.37)		
5	5.26	-2.70 (-1.29, -4.12)	6.39	-2.70 (-1.29, -4.12)	8.53	-2.70 (-1.29, -4.12)		
10	5.00	-4.88 (-2.32, -7.44)	6.13	-4.88 (-2.32, -7.44)	8.27	-4.88 (-2.32, -7.44)		

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

# 6.4.3 Analysis for Pop et al. (2003) - MDI

To evaluate the impact that a change in fT4 as a result of an increased dose of perchlorate will have on resulting MDI scores based on Pop et al. (2003), the following equation was utilized:

$$\Delta MDI = \beta \times \Delta fT4 (fT4 \text{ in pmol/L})$$

Where:

$$\beta = 6.3 (95\% \text{ CI: } 1.92, 10.6)$$

$$\Delta fT4 = fT4_{increasedClO4} - fT4_{baselineClO4}$$

Results of an analysis demonstrating how increasing perchlorate dosage decreases MDI based on Pop et al. (2003) are presented in [ REF Ref485389508 \h ].

Table [ SEQ Table \\* ARABIC ]. Estimated Change in MDI, Based on Pop et al. (2003) Central, Lower, and Upper Beta Estimates due to a Change in Perchlorate Dose from 0 μg/kg/day at Various Percentiles of fT4

	fT4 Percentile							
	2.5 <sup>th</sup>			10 <sup>th</sup>		50 <sup>th</sup>		
Dose of Perchlorate (µg/kg/day)	fT4 (pmol/L)	Δ MDI Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Δ MDI Central (Lower, Upper) <sup>a</sup>	fT4 (pmol/L)	Δ MDI Central (Lower, Upper) <sup>a</sup>		
			ine intalie =	1/ ( 1 ( 1 ( 1 ( 1 ( 1 ( 1 ( 1 ( 1 ( 1 (	T			
0	6.72	N/A	8.09	N/A	10.67	N/A		
				Astronomic Control		,		
0	5.58	N/A	6.71	N/A	8.85	N/A		
1	5.51	-0.45 (-0.14, -0.75)	6.64	-0.45 (-0.14, -0.75)	8.78	-0.45 (-0.14, -0.75)		
2	5.44	-0.87 (-0.27, -1.46)	6.57	-0.87 (-0.27, -1.46)	8.71	-0.87 (-0.27, -1.46)		
3	5.38	-1.27 (-0.39, -2.14)	6.51	-1.27 (-0.39, -2.14)	8.65	-1.27 (-0.39, -2.14)		
4	5.32	-1.66 (-0.51, -2.79)	6.45	-1.66 (-0.51, -2.79)	8.59	-1.66 (-0.51, -2.79)		
5	5.26	-2.03 (-0.62, -3.41)	6.39	-2.03 (-0.62, -3.41)	8.53	-2.03 (-0.62, -3.41)		
10	5.00	-3.66 (-1.12, -6.16)	6.13	-3.66 (-1.12, -6.16)	8.27	-3.66 (-1.12, -6.16)		

<sup>&</sup>lt;sup>a</sup> Result based on central, lower, and upper 95% CI effect estimates; BBDR model output using pTSH = 0.398; calibrated for median population. Additional details can be found in Appendix A.

# 6.5 Uncertainties Related to the Approach

The approach presented in this report is dependent upon predictions from the BBDR model, the derivation of the distribution of fT4, and the evaluations of the relationship between fT4 and neurodevelopment. Each of these steps has inherent uncertainties associated with it. Uncertainties associated with the BBDR model and derivation of the distribution of fT4 have been discussed previously (see Sections [ REF \_Ref482965010 \r \h ] and [ REF \_Ref482965013 \r \h ], respectively). One global uncertainty in this analysis is that the state of the science on the relationship between maternal fT4 levels and offspring neurodevelopment is constantly evolving. For example, since the literature review for this report ceased, at least one additional article has been published evaluating the relationship between maternal fT4 and adverse neurodevelopmental outcomes [ ADDIN EN.CITE ADDIN EN.CITE.DATA ]. The impact of future research on this assessment is unknown.

This section will focus on the uncertainties related to the literature identified that connects the estimated maternal fT4 levels with offspring neurodevelopmental outcomes and the subsequent quantitative analysis. Specifically, the first sections (Section [ REF Ref492564492 \n \h ] through [

REF \_Ref492564536 \n \h ]) discuss uncertainties as they pertain to attributes of the body of literature evaluating incremental changes in fT4 and their potential association with incremental changes in neurodevelopmental outcomes. Section [ REF \_Ref492583568 \r \h ] discusses uncertainties related to the underlying quantitative analyses.

## 6.5.1 Population for Which Dose-Response Information Is Available

The SAB stated that fetuses of hypothyroxinemic women may be a potentially sensitive life stage (SAB, 2013). However, the majority of Group 1 studies presented analyses using the full range of maternal fT4 for the population to assess incremental changes as they relate to subsequent neurodevelopment of their offspring, compared to evaluating incremental changes in maternal fT4 only in the hypothyroxinemic range. This may help to explain why many of the studies identified in the literature review did not find a statistically significant relationship between continuous maternal fT4 and neurodevelopment. As demonstrated in Korevaar et al. (2016), there may be an ideal range of maternal fT4 values, and the greatest impact of change will occur when maternal fT4 values fall below or above that range. When comparing offspring of mothers who experienced hypothyroxinemia to offspring of those who did not, adverse neurodevelopmental outcomes were more consistently related to exposure to maternal hypothyroxinemia, compared to evaluating incremental changes in maternal fT4 across the full spectrum. Thus, uncertainty exists as to whether the studies evaluating the full spectrum of fT4 are underestimating the impact of fT4 on neurodevelopment for the sensitive population of the fetuses of hypothyroxinemic pregnant women.

Further, none of the current studies analyzed in Section [ REF Ref456208917 \r \h \\* MERGEFORMAT | are based on a U.S. population. The data from all papers are from the Netherlands and are being considered for use to inform the derivation of an MCLG for the United States. There is no reason to believe that the impact of fT4 on neurodevelopment would differ by country, unless there is a substantial difference in iodine intake. Unfortunately, exact data on iodine intake levels in the Pop et al. (1999, 2003) papers are not provided. However, the authors cite two studies on iodine levels in the general population of the Netherlands. Van Rees-Wortelboer et al. [ ADDIN EN.CITE < EndNote > < Cite Exclude Auth="1" > < Author > van Rees-Wortelboer</Author><Year>1987</Year><RecNum>298</RecNum><DisplayText>(1987)</Displa yText><record><rec-number>298</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470950295">298</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>>author>van Rees-Wortelboer, M. M., </author><author>Schröder-van der Elst, J. P., </author><author>Lycklama, A., </author> <author> Van der Heide, D. </author> </contributors > < title > Iodine and goiter in The Netherlands</title><secondary-title>Nederlands Tijdschrift voor Geneeskunde</secondary-title></title><periodical><full-title>Nederlands Tijdschrift voor Geneeskunde</full-title></periodical><pages>1821-1824</pages><volume>121</volume><number>41</number><dates><year>1987</year></dates>< urls></urls></record></Cite></EndNote>] studied a sample of 253 Dutch women aged 23–32 and found a mean urinary iodide excretion (corrected for creatinine) of 115 μg/g creatinine. Brussaard et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Brussaard</Author><Year>1997</Year><RecNum>55</RecNum><Di splayText>(1997)</DisplayText><record><rec-number>55</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1437143470">55</key></foreign-keys><ref-type name="Journal Article">17</ref-

the Netherlands was 204 µg/day.

type><contributors><author>Brussaard, J H</author><author>Hulshof, K F</author><author>Kistemaker, C</author><author>Lowik, M R</author></contributors></title>Adequcy of the iodine supply in the Netherlands</title><secondary-title>European Journal of Clinical Nutrition</secondarytitle></titles><periodical><full-title>European Journal of Clinical Nutrition</fulltitle></periodical><pages>5</pages><volume>51</volume><section>11</section><dates><year>19 97</year></dates><urls></urls></record></Cite></EndNote>] compiled studies on iodine supply after the introduction of iodized salt in the Netherlands in 1982. The researchers found that estimates of the mean iodine intake in women under the age of 65 ranged from 135 to 149 μg/day. An additional study of Dutch food consumption data from 1997 to 1998 by Verkaik-Kloosterman et al. ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Verkaik-Kloosterman</Author><Year>2010</Year><RecNum>299</RecNum><DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayText>(2010)</DisplayT ayText><record><rec-number>299</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470950359">299</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>Verkaik-Kloosterman, J., </author><author>van&apos;t Veer, P., </author><author>Ocké, M. C. </author></authors></contributors><title>Reduction of salt: will iodine intake remain adequate in The Netherlands?</title><secondary-title>British Journal of Nutrition</secondarytitle></title>>cperiodical><full-title>British Journal of Nutrition</fulltitle></periodical><pages>1712-1718</pages><volume>104</volume><number>11</number><dates><vear>2010</vear></dates>< urls></urls></record></Cite></EndNote>] estimated that the median iodine intake of adult women in

Korevaar et al. (2016) provided urinary iodide data for only 672 women in their studies. The mean reported urinary/creatinine ratio in this subsample was 227  $\mu$ g/g. Iodine intake was not assessed in Endendijk et al. (2017), although they did cite Wiersinga et al. (2001, as cited in Endendijk et al., 2017) as a study demonstrating that the study area in which the Endendijk analysis was conducted is iodine sufficient. In this study school children's median iodine levels were 154.4  $\mu$ g/L (reported as 15.44  $\mu$ g/dL). Estimated daily urinary output or iodine intake were not provided in this study.

Iodine intake levels appear to be similar or lower in the United States. In a recent analysis of NHANES data, Caldwell et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Caldwell</Author><Year>2013</Year><RecNum>110</RecNum><Di splayText>(2013)</DisplayText><record><rec-number>110</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1458165747">110</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Caldwell, K.L.</author><author>Yi, P.</author><author>Mortensen, M.E.</author><author>Makhmudov, A.</author><author>Merrill, L.\(\)author\(\)<author\(\)Moye, J.\(\)\(\)author\(\)\(\)\(\)author\(\)\(\)\(\)author\(\)\(\)\(\)author\(\)\(\)\(\) pregnant women in the National Children' Study and in U.S. women (15-44 Years), National Health and Nutrition Examination Study 2005-2010</title><secondary-title>Thyroid</secondarytitle></title></periodical><full-title>Thyroid</full-title></periodical><pages>927-937</pages><volume>23</volume><number>8</number><dates><year>2013</year></dates><url> ></urls></record></Cite></EndNote>] found that the median urinary iodide excretion level in women aged 20-39 was 113 μg/g creatinine, which is much lower than that seen in Korevaar et al. (2016)'s sample. Murray et al. [ ADDIN EN.CITE ADDIN EN.CITE.DATA ] estimated the

lower- and upper-bound mean iodine intakes for U.S. women aged 25–45 as approximately 145 to 197  $\mu$ g/day, which is in line but slightly higher than the studies of Dutch women by Brussaard et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Brussaard</Author><Year>1997</Year><RecNum>55</RecNum><DisplayText>(1997)</DisplayText><record><rec-number>55</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"

timestamp="1437143470">55</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Brussaard, J H</author><author>Hulshof, K

F</author><author>Kistemaker, C</author><author>Lowik, M

R</author></authors></contributors><title>Adequcy of the iodine supply in the Netherlands</title><secondary-title>European Journal of Clinical Nutrition</secondary-

title></titles><periodical><full-title>European Journal of Clinical Nutrition</full-

title></periodical><pages>5</pages><volume>51</volume><section>11</section><dates><year>19 97</year></dates><urls></record></Cite></EndNote>] and lower than Verkaik-Kloosterman et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Verkaik-

Kloosterman</author><Year>2010</Year><RecNum>299</RecNum><DisplayText>(2010)</DisplayText><record><rec-number>299</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1470950359">299</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>><author>Verkaik-Kloosterman, J., </author><author>van&apos;t Veer, P., </author><author>Ocké, M. C.</author></author></author></contributors><title>Reduction of salt: will iodine intake remain adequate in The Netherlands?</title><secondary-title>British Journal of Nutrition</secondary-

adequate in The Netherlands?</title><secondary-title>British Journal of Nutrition title></title><periodical><full-title>British Journal of Nutrition</full-title></periodical><pages>1712-

1718</pages><volume>104</volume><number>11</number><dates><year>2010</year></dates><urls></urls></record></Cite></EndNote>]. Thus, there is little reason to believe that the uncertainty in iodine intake levels would have a large impact on the resulting derivation of an MCLG.

The influence of nationality on endpoints and examinations used to evaluate them may produce uncertainty when comparing study populations. For example, in Pop et al. (1999, 2003) and other Dutch studies, the Dutch version of the BSID was used to evaluate neurodevelopmental endpoints. The Dutch version of the BSID is a translation of the U.S. version and uses the same cut points for impairment [ ADDIN EN.CITE

<EndNote><Cite><Author>Pop</Author><Year>2003</Year><RecNum><DisplayText>(Pop et al., 2003; van der Meulen & Smrkovsky, 1984)</DisplayText><record><recnumber><25</recnumber><foreign-keys><key app="EN" db-

id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288 < pages > <volume > 59 < /volume > <section > 282 < /section > <dates > <year > 2003 < /year > </dates > <url s > </url s > </crite > </crite > <url dates > <url dates

Meulen</Author><Year>1984</Year><RecNum>233</RecNum><record><rec-number>233</rec

number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468266018">233</key></foreign-keys><ref-type name="Book">6</reftype><contributors><author>van der Meulen, B.F. </author><author>Smrkovsky, M.</author></authors></contributors><title>Bayley Ontwikkelingsschalen</title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title></title> The Netherlands</publocation><publisher>Kinderstudies</publisher><urls></urls></record></Cite></EndNote>]. However, Dutch norms may differ from U.S. norms, and one study found that "for all subtests... these differences are clinically relevant" [ ADDIN EN.CITE <EndNote><Cite><Author>Steenis</Author><Year>2015</Year><RecNum>1911</RecNum><Dis playText>(Steenis, Verhoeven, Hessen, & Daar, 2015)</br> number>1911</rec-number><foreign-keys><key app="EN" dbid="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1503343665">1911</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><author>><author>>Steenis, L.J.P.</author><author>Verhoeven, M.</author><author>Hessen, D.J.</author><author>van Baar, A.L.</author></authors></contributors><title>Performance of Dutch children on the Bayley III: A comparison study of U.S. and Dutch norms</title><secondary-title>PLoS ONE</secondarytitle></title> PLoS ONE</full-title>PLoS ONE</fulltitle></periodical><volume>10</volume><number>8</number><dates><year>2015</year></dates> <url>></urls></record></Cite></EndNote>].

#### 6.5.2 Endpoint Evaluated

There are numerous indices by which to assess neurodevelopmental impacts. Some evaluate a single attribute of neurodevelopment (e.g., language delay), whereas others evaluate a broader set of attributes. The majority of neurodevelopmental indicators tests used in the cognitive studies measure indices of global neuropsychological performance. Only one study evaluated further for doseresponse was based on a behavioral endpoint (Endendijk et al., 2017). Animal data show that thyroid hormone decrement affects different brain areas depending on the stage of gestation when hormone deficiency is present [ ADDIN EN.CITE < EndNote > < Cite > < Author > Morreale de Escobar</Author><Year>2004</Year><RecNum>49</RecNum><DisplayText>(G Morreale de Escobar et al., 2004)</br/>
/DisplayText><record><rec-number>49</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1437077734">49</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Morreale de Escobar, G</author><author>Obregón, M J</author><author>Escobar del Rey, F</author></authors></contributors><title>Role of thyroid hormone during early brain development</title><secondary-title>European Journal of Endocrinology</secondary-title></titles><periodical><full-title>European Journal of Endocrinology</full-title></periodical><pages>U25-U37</pages><volume>151</volume><number>Suppl 3</number><dates><year>2004</year><pubdates><date>November 1, 2004</date></pub-dates></dates><urls><relatedurls><url>http://www.eje-online.org/content/151/Suppl 3/U25.abstract</url></relatedurls></urls><electronic-resource-num>10.1530/eje.0.151U025</electronic-resourcenum></record></Cite></EndNote>]. As such, global markers may not evaluate the most sensitive neurodevelopmental endpoint in regard to decreased maternal thyroid hormone levels depending on the phase of gestation in which a fetus may be exposed to low maternal thyroid hormone levels. Further, mothers with pre-existing thyroid disease were excluded from certain studies used in the

analysis (e.g., Korevaar et al., 2016; Finken et al., 2013), which may have potentially reduced the relationship between fT4 and neurodevelopmental outcomes in these studies. Additionally, considering the analysis for Finken et al. (2013), it is difficult to ascertain the true implications of a change in the standard deviation of reaction time, the endpoint with which there is a significant association with maternal fT4 levels.

## 6.5.3 Iodine Intake Data for the Study Population

Iodine intake status may impact neurodevelopmental outcomes, and the BBDR model presents dose-response information by iodine intake level. However, none of the five studies carried forward for additional quantitative analysis provided information specifically on iodine intake levels, though Korevaar et al. (2016) provided some data on urinary iodine as discussed in Section [ REF \_Ref492564492 \r \h ]. Therefore, uncertainty exists regarding how information culled from each study relates to a specific iodine intake rate, and the corresponding fT4 changes, as predicted by the BBDR model.

## 6.5.4 Size of the Study Population

A larger cohort may have more power to detect a subtle difference in an outcome compared to a smaller one. Additionally, a larger study may be more easily extrapolated compared to a smaller study. However, smaller studies have benefits as well. For example, some of the standard neurodevelopmental evaluations require one-on-one assessments, which may not be feasible in cohorts with thousands of individuals. Further, a small study that does detect an association is notable given the lack of power that is associated with a small sample size.

The analyses by Pop et al. (1999; 2003) have small (n < 30) populations when considering the results used for quantitative analysis. However, despite a small sample size and low statistical power, the studies found statistically significant relationships between fT4 and a measure of neurodevelopment. Additionally, the Pop et al. (1999, 2003) findings of an association between maternal hypothyroxinemic levels of fT4 and adverse neurodevelopmental outcomes are supported by many other papers (i.e., the majority of Group 1 and Group 2 papers). There were 442 women in the final sample in Endendijk et al. (2017). Finken et al. (2013) and Korevaar et al. (2016) both used larger cohorts (n > 1,700 for both) in the analyses the EPA used to evaluate incremental changes in neurodevelopment.

# 6.5.5 The Method and Timing of Maternal fT4 Measurement During Pregnancy

Repeated thyroid hormone measurements give a more complete picture as to how alterations in thyroid homeostasis impact neurodevelopment throughout gestation. Given that perchlorate exposure likely does not occur in just one phase of gestation or early infancy, having a more comprehensive evaluation of the impact of thyroid hormone levels on neurodevelopment may provide a better concept of how suppressed thyroid hormone levels impact neurodevelopment throughout pregnancy or early infancy. However, this must be considered alongside the fact that the BBDR model is focused on early pregnancy. Subsequently, studies with a more complete analysis of fT4 across the course of pregnancy may be informative but it may not be possible to connect them to the output of the BBDR model. In regard to maternal thyroid hormone levels, fT4 in early pregnancy (i.e., prior to fetal thyroid hormone development, which occurs between GW 12 and GW 16) is believed to be most important in regard to neurodevelopmental outcomes (Morreale de Escobar, 2001; Morreale de Escobar et al., 2004).

All studies identified for further quantitative analysis on the relationship between maternal fT4 and neurological effects in the offspring evaluated this relationship based on a one-time fT4 measurement during pregnancy with the exception of Endendijk et al. (2017). Endendijk et al. (2017) conducted growth mixture modeling, from which the authors determined that the trajectory of maternal thyroid hormones throughout the course of pregnancy was a better predictor of internalizing problems compared to just first-trimester fT4. However, given that the BBDR model concentrates on early pregnancy, results from the analysis on the trajectories of thyroid hormones could not be used in the further dose-response analysis to inform the MCLG for perchlorate.

From evaluating the results of the literature review it appears the relationship between maternal fT4 and fetal brain development has a temporal factor, with this influence likely being greatest in early pregnancy (i.e., prior to mid-gestation). While maternal hypothyroxinemia later in pregnancy appears to be less influential, it is not without effect. For example, Pop et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayT ext>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>59</volume>59</volume>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32</section>32

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayT ext>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume>59</volume>3282</section><dates><year>2003</year></dates><url s></urls></record></EndNote>]. Additionally, the authors found that individuals born to mothers with consistently low fT4 had the worst neurodevelopmental outcomes. Further, Endendijk et al. (2017) found that individuals in a group with the highest and non-increasing TSH levels with the lowest fT4 that decreased the least amount throughout the course of pregnancy showed more anxiety/depression symptoms compared to groups with either increasing TSH and decreasing fT4 at

intermediate levels or lowest and increasing TSH with highest and decreasing fT4 levels. Subsequently, how maternal thyroid hormones change throughout the course of pregnancy represents an uncontrolled source of variability given that what happens after GW 12 to maternal fT4 levels appears to have an impact on neurodevelopmental outcome.

If low fT4 later in pregnancy also causes further damage beyond the effects in the first trimester, it is likely that a steeper dose-response slope would have been found if only those women who were consistently hypothyroxinemic throughout pregnancy were evaluated (see Pop et al., 2003, Figure 3). This weakening of the correlation between hormone and outcome is likely to lessen the perchlorate dose-response slope on neurodevelopment because it relies on both the relationship between perchlorate's effect on hormones (BBDR model) and then hormones' effect on outcome.

Also related to the method of assessing fT4 is that uncertainties exist in regard to the analytical approach to evaluating fT4 levels. Although a working group was put together by the International Federation of Clinical Chemistry and Laboratory Medicine in 2005 to standardize thyroid function tests primarily for TSH and fT4, this standardization has not yet been achieved [ ADDIN EN.CITE <EndNote><Cite><Author>Thienpont</Author><Year>2015</Year><RecNum>276</RecNum><D isplayText>(Thienpont, Faix, &amp; Beastall, 2015)</DisplayText><record><rec-number>276</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468523238">276</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Faix, J. D.</author><author>Faix, J. D.</author><author>Faix, J. D.</author><author>Beastall, G.</author></author><author>Faix, Ghent, Belgium.</authoraddress>Mass Spectrometry Reference Laboratory LAC at Ghent University, Ghent, Belgium.</authoraddress><title>Standardization of FT4 and harmonization of TSH measurements - a request for input from endocrinologists and other physicians</a>/title><secondary-title>Endocrine Journal</a>/full-title></periodical><pages>855-

6</pages><volume>62</volume><number>10</number><edition>2015/07/28</edition><dates><ye ar>2015</year></dates><isbn>1348-4540 (Electronic)&#xD;0918-8959 (Linking)</isbn><accession-num>26211473</accession-num><urls></urls><electronic-resource-num>10.1507/endocrj.EJ15-0382</electronic-resource-num><remote-database-provider>NLM</remote-database-

provider><language>eng</language></record></Cite></EndNote>]. As such, assay results obtained at different times or in different countries may not be directly comparable due to differences in analytical procedures. fT4 could be as little as 1% of the circulating total T4 concentration [ ADDIN EN.CITE

<EndNote><Cite><Author>Koulouri</Author><Year>2013</Year><RecNum>275</RecNum><Dis playText>(Koulouri, Moran, Halsall, Chatterjee, & DisplayText><record><recnumber>275</rec-number><foreign-keys><key app="EN" db-

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& Description of the Research Clinical Endocrinology & Description of the Research Cl

title></titles><periodical><full-title>Best Practice & Dinical Endocrinology & Metabolism</full-title></periodical><pages>745-62</pages><volume>27</volume><number>6</number><edition>2013/11/28</edition><keywords ><keyword>Female</keyword><keyword>Humans</keyword><keyword>Hyperthyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Hypothyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Male</keyword><keyword>Pregnancy</keyword ><keyword>Thyroid Function Tests</keyword><keyword>Thyroid Gland/ physiopathology</keyword><keyword>Thyroid Hormones/ analysis</keyword></keywords><dates><year>2013</year><pub-dates><date>Dec</date></pubdates></dates><isbn>1878-1594 (Electronic)&#xD;1521-690X (Linking)</isbn><accessionnum>24275187</accession-num><urls></urls><custom2>3857600</custom2><electronic-resourcenum>10.1016/j.beem.2013.10.003</electronic-resource-num><remote-databaseprovider>NLM</remote-databaseprovider><language>eng</language></record></Cite></EndNote>], making it difficult to measure. The bulk of the enzyme-linked immunosorbent assays used for measurement of fT4 require no pretreatment or serum extraction procedures to be undertaken. Rather, they rely on competitive binding assays that could be affected by changes in endogenous serum protein ratios, displacing agents such as heparin, and the presence of other antibodies that may block the action of the assay antibody ADDIN EN.CITE <EndNote><Cite><Author>Koulouri</Author><Year>2013</Year><RecNum>275</RecNum><Dis playText>(Koulouri et al., 2013)</br> keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468523217">275</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Koulouri, O.</author><author>Moran, C.</author><author>Halsall, D.</author><author>Chatterjee, K.</author><author>Gurnell, M.</author></authors></contributors><auth-address>Metabolic Research Laboratories, Wellcome Trust - MRC Institute of Metabolic Science, University of Cambridge and National Institute for Health Research Cambridge Biomedical Research Centre, Addenbrooke's Hospital, Cambridge CB2 0QQ, UK.</auth-address><title>Pitfalls in the measurement and interpretation of thyroid function tests</title><secondary-title>Best Practice & Dinical Endocrinology & Metabolism</secondary-title></title>>eriodical><full-title>Best Practice & Description (Amp): Research Clinical Endocrinology & December 2015 Endocrinology & 62</pages><volume>27</volume><number>6</number><edition>2013/11/28</edition><keywords ><keyword>Female</keyword><keyword>Humans</keyword><keyword>Hyperthyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Hypothyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Male</keyword><keyword>Pregnancy</keyword ><keyword>Thyroid Function Tests</keyword><keyword>Thyroid Gland/ physiopathology</keyword><keyword>Thyroid Hormones/ analysis</keyword></keywords><dates><year>2013</year><pub-dates><date>Dec</date></pubdates></dates><isbn>1878-1594 (Electronic)&#xD;1521-690X (Linking)</isbn><accessionnum>24275187</accession-num><urls></urls><custom2>3857600</custom2><electronic-resourcenum>10.1016/j.beem.2013.10.003</electronic-resource-num><remote-databaseprovider>NLM</remote-databaseprovider><language>eng</language></record></Cite></EndNote>]. Correlation between liquid chromatographic tandem mass spectrometry (MS) values versus standard immunoassay techniques

showed that there was a correlation coefficient of approximately 0.75 for fT4, unlike total T4, which had a correlation coefficient of 0.91–0.95. This was likely due to the ability of tandem MS to better measure fT4 compared to immunoassay [ ADDIN EN.CITE ADDIN EN.CITE.DATA ].

In regards to the studies further evaluated for quantitative assessment, both Pop et al. studies used immunoassay techniques to assess the fT4 levels in their cohort. Korevaar et al. (2016) and Finken et al. (2013) used chemiluminescence assays, as did Endendijk et al. (2017). Additionally, there could be individual variation in fT4 measurements or interference from some over-the-counter medications, such as aspirin and other NSAIDs, which could displace thyroid hormones on TBG or TTR, causing artificial elevations in fT4 measurement [ ADDIN EN.CITE <EndNote><Cite><Author>Koulouri</Author><Year>2013</Year><RecNum>275</RecNum><Dis playText>(Koulouri et al., 2013)</br>

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275</rec-number><foreign-</td> keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1468523217">275</key></foreign-keys><ref-type name="Journal Article">17</reftype><contributors><author>Koulouri, O.</author><author>Moran, C.</author><author>Halsall, D.</author><author>Chatterjee, K.</author><author>Gurnell, M.</author></authors></contributors><auth-address>Metabolic Research Laboratories, Wellcome Trust - MRC Institute of Metabolic Science, University of Cambridge and National Institute for Health Research Cambridge Biomedical Research Centre, Addenbrooke's Hospital, Cambridge CB2 0QQ, UK.</auth-address><title>Pitfalls in the measurement and interpretation of thyroid function tests</title><secondary-title>Best Practice & Dinical Endocrinology & Metabolism</secondary-title></title>>eriodical><full-title>Best Practice & Description (Amp); Research Clinical Endocrinology & Detabolism </ full-title ></periodical >< pages > 745-62</pages><volume>27</volume><number>6</number><edition>2013/11/28</edition><keywords ><keyword>Female</keyword><keyword>Humans</keyword><keyword>Hyperthyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Hypothyroidism/blood/ diagnosis/physiopathology</keyword><keyword>Male</keyword><keyword>Pregnancy</keyword ><keyword>Thyroid Function Tests</keyword><keyword>Thyroid Gland/ physiopathology</keyword><keyword>Thyroid Hormones/ analysis</keyword></keywords><dates><year>2013</year><pub-dates><date>Dec</date></pubdates></dates><isbn>1878-1594 (Electronic)&#xD;1521-690X (Linking)</isbn><accessionnum>24275187</accession-num><urls></urls><custom2>3857600</custom2><electronic-resourcenum>10.1016/j.beem.2013.10.003</electronic-resource-num><remote-databaseprovider>NLM</remote-databaseprovider><language>cng</language></record></Cite></EndNote>]. Thus, the true fT4 levels of the subjects in the papers used for further analysis may differ from the values obtained in their analyses. However, because all individuals in the papers were assessed using the same techniques, any bias resulting from this source of uncertainty would likely be random and would tend to bias toward the null.

## 6.5.6 Uncertainties Related to the Quantitative Analyses

There is uncertainty regarding the true fT4 levels at various percentiles in the distribution around the median output from the BBDR model. In most analyses presented in this report, unit changes in endpoints due to changes in fT4 assuming a particular dose of perchlorate are the same across fT4 percentiles. This is due to the linear relationships assumed between fT4 and each endpoint modeled in the epidemiologic studies, along with the EPA's assumption of normality of errors around the predicted median fT4 values output by the BBDR model. The exceptions to this pattern of results are

the analysis results based on the Korevaar et al. (2016) data (both the original quadratic function and the EPA reanalysis). In these analyses, larger unit changes are seen with decreasing percentiles of fT4, which is due to the assumption of non-linear relationships between fT4 and IQ, and especially due to the log-transformation of fT4 in the models. This pattern of results agrees with the hypothesis that decreases in fT4 from lower base levels of fT4 have a larger effect on neurodevelopmental endpoints than at higher base levels of fT4.

One uncertainty consideration, specifically when evaluating the lower percentiles of fT4 from the derived distributions, is the potential for extrapolating beyond the data from which the function was derived. For example, [ REF \_Ref492567077 \h ] summarizes the lowest fT4 value observed in each of the studies that the EPA evaluated further. Given that a lower fT4 level is associated with worse neurodevelopmental outcomes, there is no reason to believe that predictions of fT4's effect on neurodevelopment are being overestimated by falling below the observed range.

Table [ SEQ Table \\* ARABIC ]. Lowest Observed fT4 Levels in Group 1 Studies Used for Analysis

Study	Lowest Observed fT4 (pmol/L)
Finken et al. (2013) <sup>a</sup>	< 6.5
Korevaar et al. (2016)	6.38
Pop et al. (1999) <sup>b</sup>	7.86
Pop et al. (2003) <sup>b</sup>	8.39
Endendijk et al. (2017) <sup>c</sup>	< 11.97

<sup>&</sup>lt;sup>a</sup> Full range not presented. 6.5 pmol/L represents the 2.5th percentile fT4.

Further, several of the dose-response functions presented in this report linking maternal fT4 to an offspring's neurodevelopment are based on the EPA's own digitization of data from a figure depicting a correlation or regression analysis (Pop et al., 1999, 2003). The EPA has not used multiple regression analysis to adjust for confounding or interacting variables because it does not have the raw data to do so. However, as discussed in Section [ REF \_Ref482192481 \r \h ], Pop et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Pop</Author><Year>2003</Year><RecNum>25</RecNum><DisplayT ext>(2003)</DisplayText><record><rec-number>25</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047641">25</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Pop, V J</author><author>Brouwers, E P</author><author>Vader, H L</author><author>Vulsma, T</author><author>van Baar, A L</author><author>de Vijlder, J

J</author></authors></contributors><titles><title>Maternal hypothyroxinemia during early pregnancy and subsequent child development: a 3-year follow-up study</title><secondary-title>Clinical Endocrinology</secondary-title></titles><periodical><full-title>Clinical Endocrinology</full-title></periodical><pages>282-

288</pages><volume>59</volume><section>282</section><dates><year>2003</year></dates><url s></urls></record></Cite></EndNote>] took steps to mitigate the potential for residual confounding

<sup>&</sup>lt;sup>b</sup> Range of data estimated from digitized dataset.

<sup>&</sup>lt;sup>c</sup> Full range not presented. 11.97 pmol/L represents the 2.5th percentile fT4.

during the development of the cohorts for which they conducted their evaluation. Specifically, women were excluded from analysis due to pregnancy and child outcome-related factors that could interfere with a child's neurodevelopmental outcome. For example, women were excluded if they had at least one episode of reported post-partum depression. Additionally, as outlined in the Pop et al. (1999) summary, in their logistic regression analysis they found a much higher relative risk after controlling for confounding variables (RR = 5.8; 95% CI: 1.3-12.6) compared to their univariate analysis (RR = 3.6; 95% CI: 1.1-12.1). This implies that the current effect estimates derived from the Pop et al. studies may be underestimating the true effect of fT4 level in the maternal hypothyroxinemic range and its impact on neurodevelopment. The analyses provided by Korevaar et al. [ ADDIN EN.CITE <EndNote><Cite

timestamp="1491832236">313</key><key app="ENWeb" db-id="">0</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Korevaar, Tim I.

M.</author><author>Muetzel, Ryan</author><author>Medici, Marco</author><author>Chaker,
Layal</author><author>Jaddoe, Vincent W. V.</author><author>de Rijke, Yolanda

B.</author><author> Steegers, Eric A. P.</author> author> Visser, Theo J.</author> author> White, Tonya</author> author> Tiemeier, Henning</author> author> Peeters, Robin

P.</author></contributors><titles><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabe

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43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>2 2138587</isbn><urls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-

7</electronic-resource-num></record></Cite></EndNote>] and Finken et al. [ ADDIN EN.CITE <EndNote><Cite

ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum>6Displa yText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></author></a></al></al>

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Additionally, some studies that were not used for further quantitative analysis found associations that were attenuated when adjusting models for potential confounders (e.g., Gyllenberg et al., 2016; Modesto et al., 2015) given the impact that variables such as smoking may have on both thyroid hormone levels and neurodevelopmental outcomes. For example, Gyllenberg et al. (2016) observed an association between fT4 and risk of schizophrenia in their unadjusted analysis, but after individually adjusting for maternal smoking during early gestation and collectively adjusting for maternal psychiatric history, province of birth, and maternal smoking, the association between maternal hypothyroxinemia and offspring schizophrenia was no longer significant. The adjustment for maternal smoking resulted in a greater than 20% change in the odds ratio. According to the authors, the finding that the continuous measure of fT4 lost statistical significance while there was still an association between maternal hypothyroxinemia after adjusting for confounders "suggests that the relationship between maternal fT4 and schizophrenia is present only when fT4 levels are below a particular threshold" (Gyllenberg et al., 2016, p. 966). Other factors that were not measured or controlled by the papers assessed, such as maternal lead or arsenic levels, are additional sources of uncontrolled variability that would tend to weaken the correlation between fT4 and neurodevelopment and thus perchlorate and neurodevelopment.

# 7. Informing the Derivation of an MCLG: Hypothyroxinemia

The previous section outlined a process by which the output of the BBDR model was connected to neurodevelopmental outcomes in the offspring of pregnant women who are exposed to perchlorate in pregnancy. This connection was made by using epidemiologic literature to connect shifts in fT4, as predicted by the BBDR model, to various indicators of neurodevelopment.

An alternative option to inform future decisions on the derivation of an MCLG using the BBDR model is to evaluate a shift in the proportion of the population that will fall below a hypothyroxinemic cut point, given exposure to perchlorate. This approach does not directly connect the BBDR output to neurodevelopment. However, for pregnant mothers in early pregnancy, this shift could be related to avoiding an increase in the offspring population's risk of adverse neurodevelopmental impacts given the preponderance of evidence located in the EPA's literature review that finds an association between hypothyroxinemia and adverse neurodevelopmental outcomes, as outlined in Section [ REF Ref482392166 \r \h ].

For example, many Group 2 studies found an increased risk of adverse neurodevelopmental outcomes in the offspring of mothers with maternal hypothyroxinemia, compared to non-hypothyroxinemic mothers. These effects included decreased performance on the Bayley Scales for Infant Development (Costeira et al., 2011; Júlvez et al., 2013; Li et al., 2010), behavioral assessment scales (Kooistra et al., 2006), and increased fetal and infant head size (van Mil et al., 2012). Additionally, Päkkilä et al. (2015) found increased odds of having repeated a school class as assessed in 16-year-olds. Further, two meta-analyses (Fan and Wu, 2016; Wang et al., 2016) concluded that low fT4 is associated with adverse neurodevelopmental outcomes including increased risk of delayed cognitive development and lower intelligence and motor scores. Thompson et al. (2018) found that compared to those children born to euthyroid mothers, children born to hypothyroxinemic mothers were significantly more likely to show signs of intellectual impairment.

In addition to presenting regression analyses that evaluated how thyroid hormone levels impact neurodevelopment, 11 Group 1 studies also found strong associations between maternal hypothyroxinemia and neurodevelopment outcomes in offspring when compared to non-hypothyroxinemic groups. That is, of the 13 studies in Group 1 that performed a categorical analysis, the only studies that did not find an adverse association between maternal hypothyroxinemia and a measure of neurodevelopment were Oken et al. (2009) and Kasatkina et al. (2006). As previously discussed, Oken et al. (2009) evaluated only T4, not fT4, which may be an explanation for the null finding. The findings from Kasatkina et al. (2006) were not assessed further, as the study was designated as Tier 3 (high ROB).

In evaluating neurodevelopment with the BSID, both Pop et al. studies (1999, 2003) found the relationship between reduced fT4 and reduced MDI or PDI only in hypothyroxinemic first-trimester pregnant mothers. Further, both Ghassabian et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Ghassabian</Author><Year>2014</Year><RecNum>10</RecNum><D isplayText>(2014)</DisplayText><record><rec-number>10</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047632">10</key></foreign-keys><ref-type name="Journal Article">17</ref-type><contributors><author>Ghassabian, A</author><author>Marroun, H E</author><author>Pcetcrs, R P</author><author>Jaddoc, V W</author><author>Hofman, A</author><author>Verhulst, F C</author><author>Tiemeier, H</author><author>White,

T</author></authors></contributors><title>Downstream effects of maternal

hypothyroxinemia in early pregnancy: nonverbal IQ and brain morphology in school-age children</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondarytitle></titles><periodical><full-title>Journal of Clinical Endocrinology and Metabolism</fulltitle></periodical><pages>2383-2390</pages><volume>99</volume><number>7</number><section>2383</section><dates><year> 2014</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2013-4281</electronicresource-num></record></Cite></EndNote>] and Korevaar et al. [ ADDIN EN.CITE <EndNote><Cite ExcludeAuth="1"><Author>Korevaar</Author><Year>2016</Year><RecNum>313</RecNum><Di splayTcxt>(2016)</DisplayTcxt><record><rec-number>313</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1491832236">313</key><key app="ENWeb" db-id="">0</key></foreign-keys><reftype name="Journal Article">17</ref-type><contributors><author>Korevaar, Tim I. 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P.</author><author>Visser, Theo J.</author><author>White, Tonya</author><author>Tiemeier, Henning</author><author>Peeters, Robin P.</author></authors></contributors><title>Association of maternal thyroid function during early pregnancy with offspring IQ and brain morphology in childhood: a population-based prospective cohort study</title><secondary-title>The Lancet Diabetes & Diabet Endocrinology</secondary-title></title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></secondary-title></seconda Endocrinology</full-title></periodical><pages>35-43</pages><volume>4</volume><number>1</number><dates><year>2016</year></dates><isbn>2 2138587</isbn><urls></urls><electronic-resource-num>10.1016/s2213-8587(15)00327-7</electronic-resource-num></record></EndNote>] found maternal hypothyroxinemia to be associated with lower non-verbal intelligence scores in offspring compared to IQ scores from offspring of non-hypothyroxinemic mothers. Finken et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Finken</Author><Year>2013</Year><RecNum>9</RecNum><Displa yText>(2013)</DisplayText><record><rec-number>9</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29" timestamp="1432047631">9</key></foreignkeys><ref-type name="Journal Article">17</ref-type><contributors><authors><author>Finken, M J J</author><author>van Eijsden, M</author><author>Loomans, E M</author><author>Vrijkotte, T G M</author><author>Rotteveel, J</author></authors></contributors><titles><title>Maternal hypothyroxinemia in early pregnancy predicts reduced performance in reaction time tests in 5- to 6year-old offspring</title><secondary-title>Journal of Clinical Endocrinology and Metabolism</secondary-title></title>>epriodical><full-title>Journal of Clinical Endocrinology and Metabolism</full-title></periodical><pages>1417-1426</pages><volume>98</volume><number>4</number><section>1417</section><dates><year> 2013</year></dates><urls></urls><electronic-resource-num>10.1210/jc.2012-3389</electronicresource-num></record></Cite></EndNote>] found increased baseline reaction time and standard deviation of reaction time in offspring of hypothyroxinemic mothers compared to nonhypothyroxinemic mothers. Henrichs et al. [ ADDIN EN.CITE < EndNote > < Cite ExcludeAuth="1"><Author>Henrichs</Author><Year>2010</Year><RecNum>16</RecNum><Dis playText>(2010)</DisplayText><record><rec-number>16</rec-number><foreign-keys><key app="EN" db-id="z9t0avxvzdfermedxe5vxfpkax2vzp0ftv29"